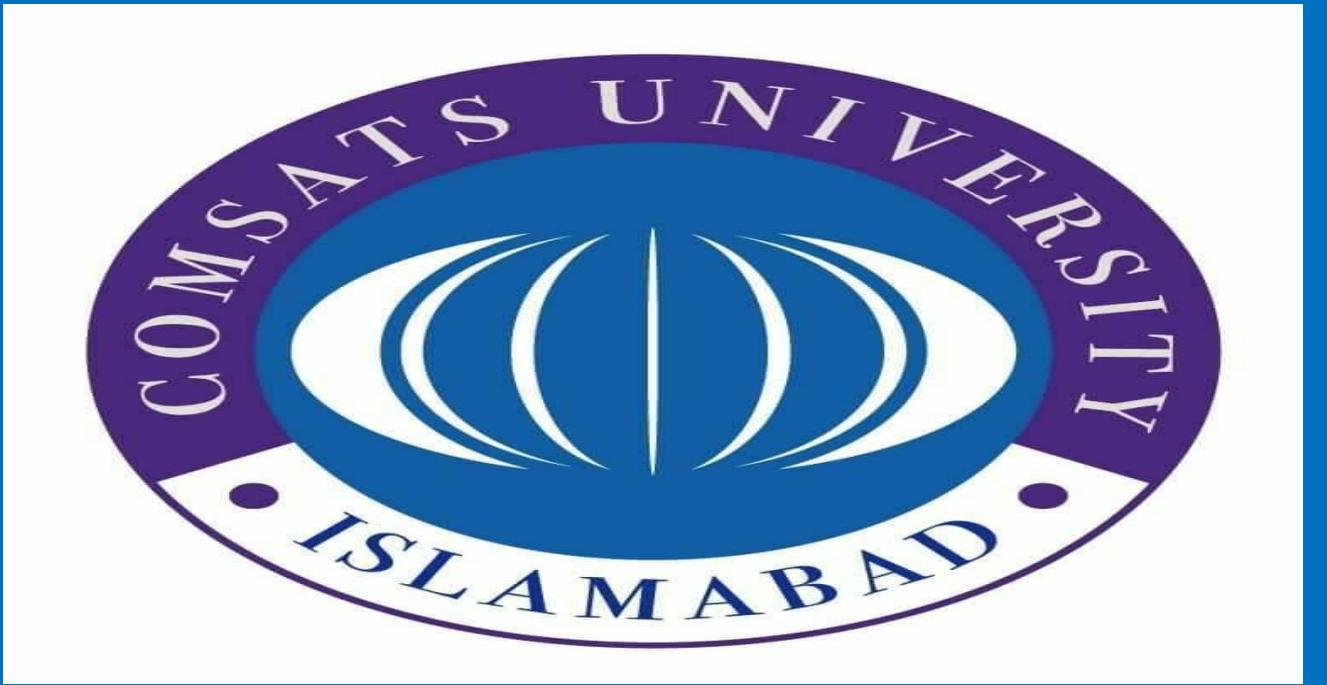


Dielectric and Transport Behavior of Dual Doping (Sm-Ce) in $\text{Bi}_2\text{Ca}_2\text{CoO}_6$ Perovskite Cobaltite



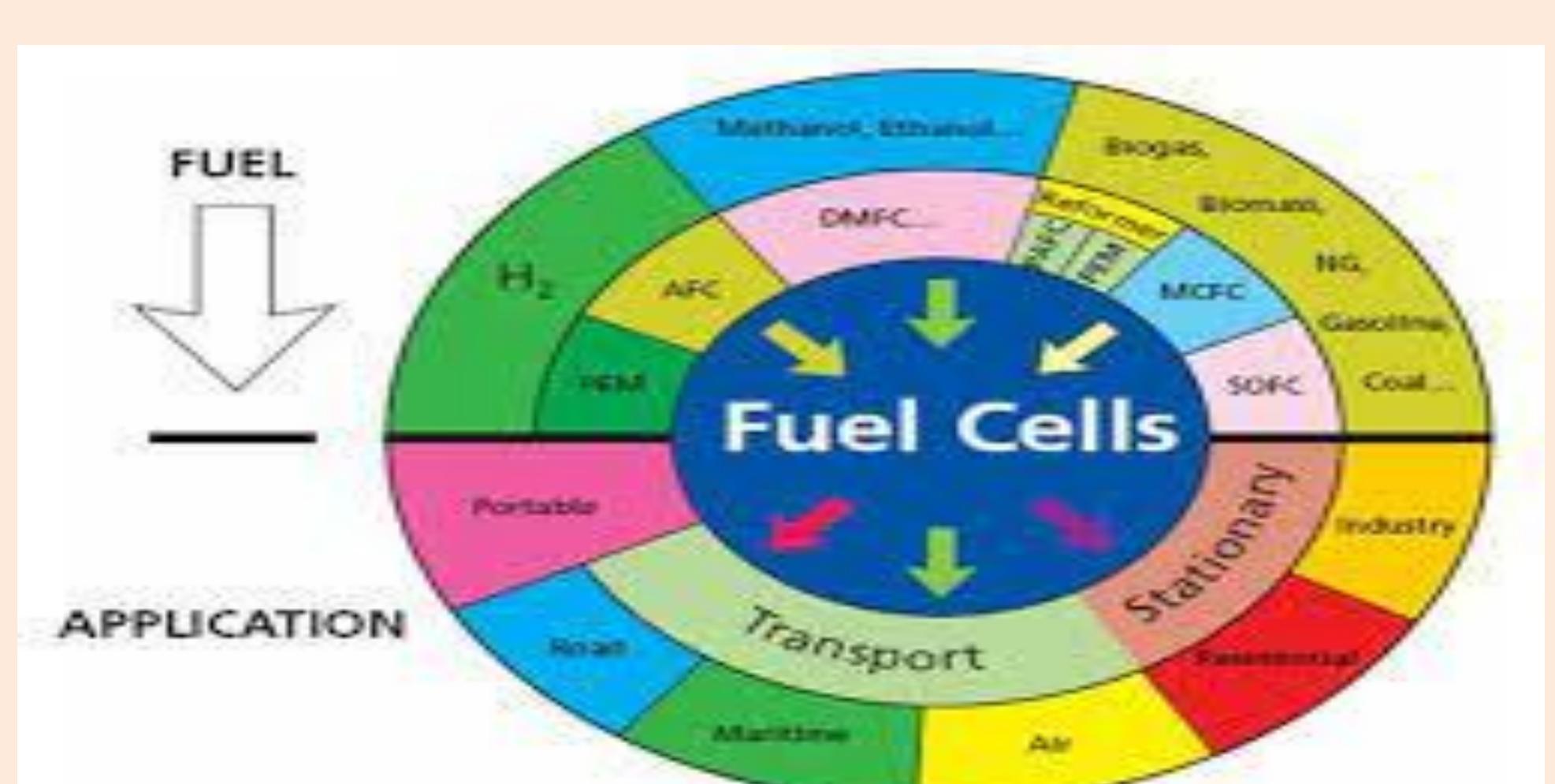
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COMSATS University Islamabad, 45550, Islamabad, Pakistan

Abstract

- $\text{Bi}_2\text{Ca}_{1.80}\text{Sm}_{0.10}\text{Ce}_{0.10}\text{CoO}_6$ (BCSCCO) perovskite were synthesized the using co-precipitation route
- XRD analysis confirms the monoclinic structure with space group P21/m
- The obtained results are validated using the Jonscher power law.

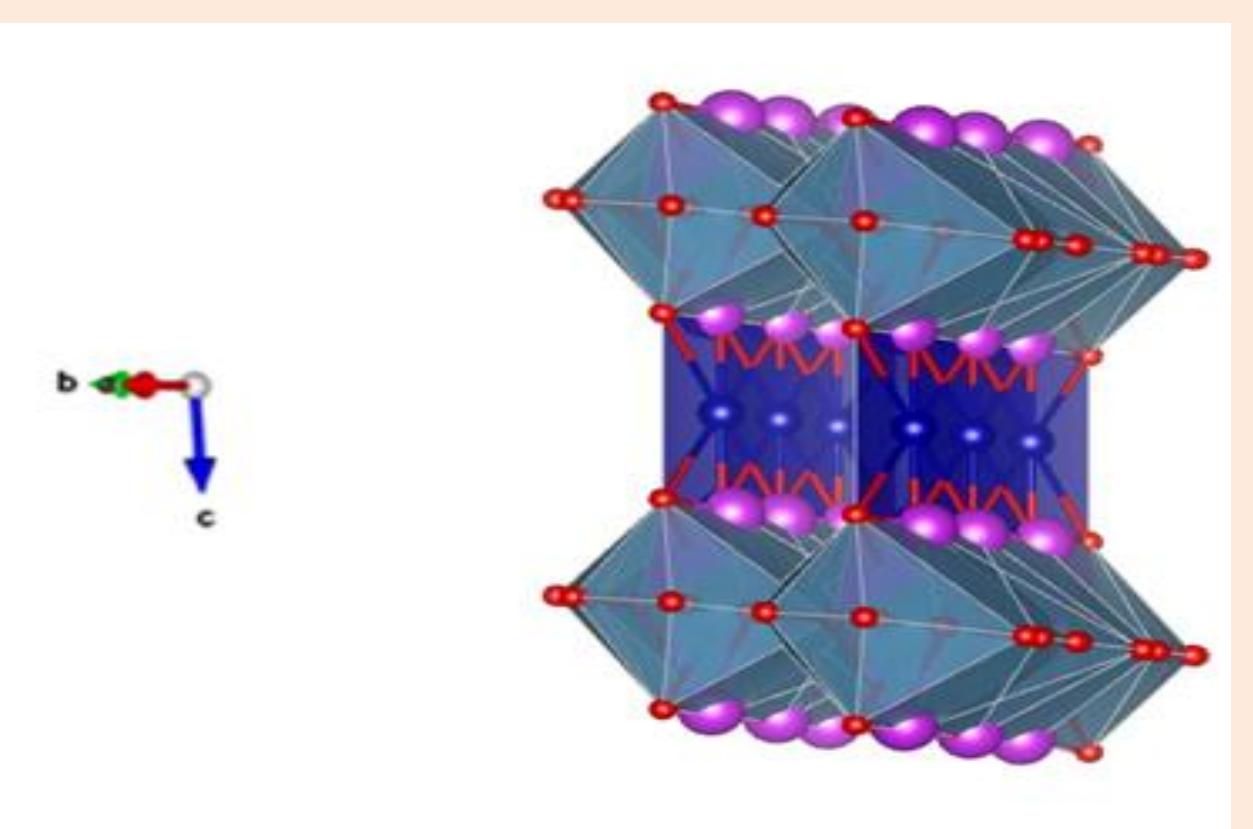
Introduction

- Fossil fuels are the major sources of our energy production in current era, which are limited by nature. Geologists are expecting the current reserves to last up-to 2042 [1].

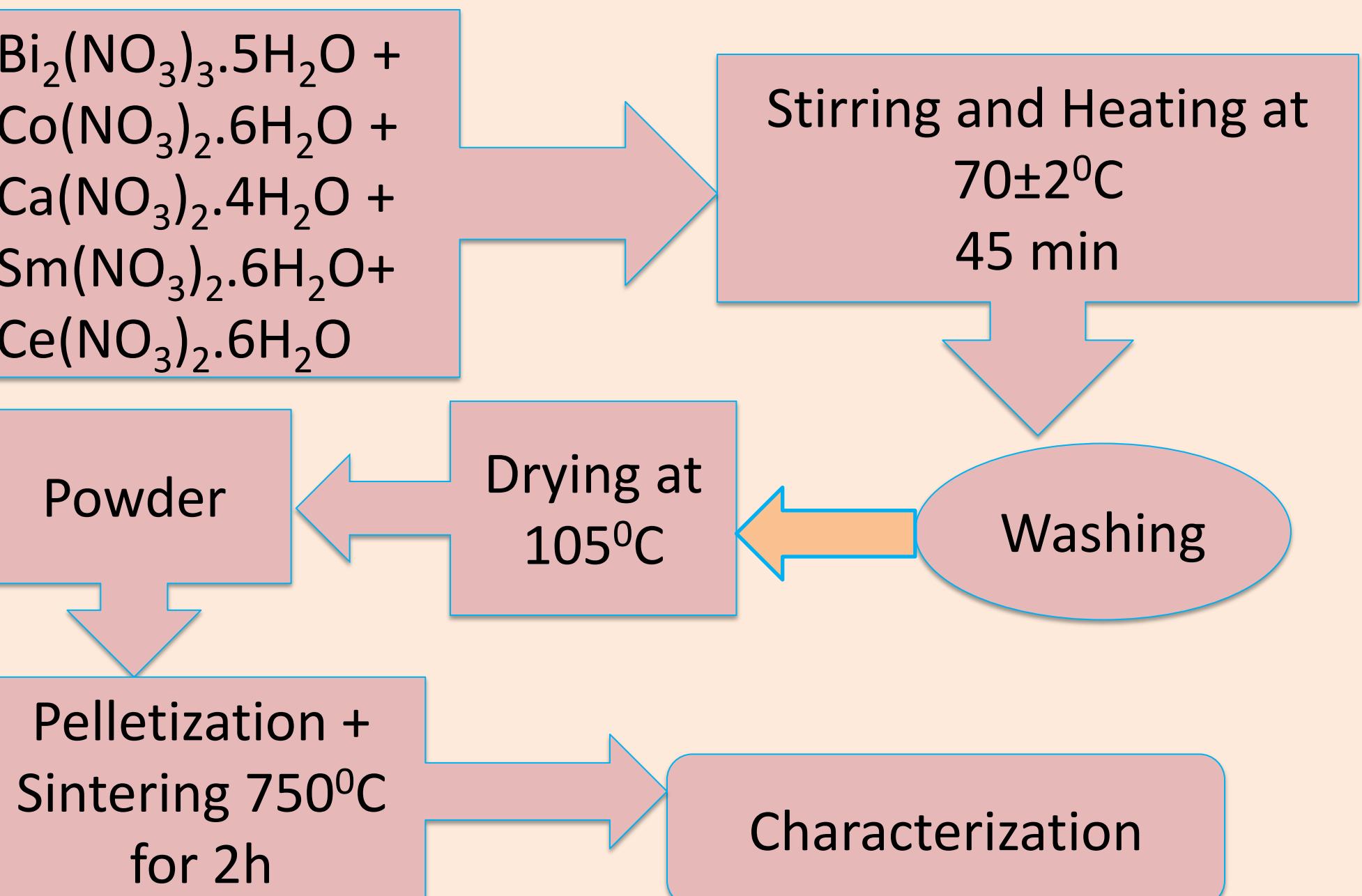


Crystal Structure

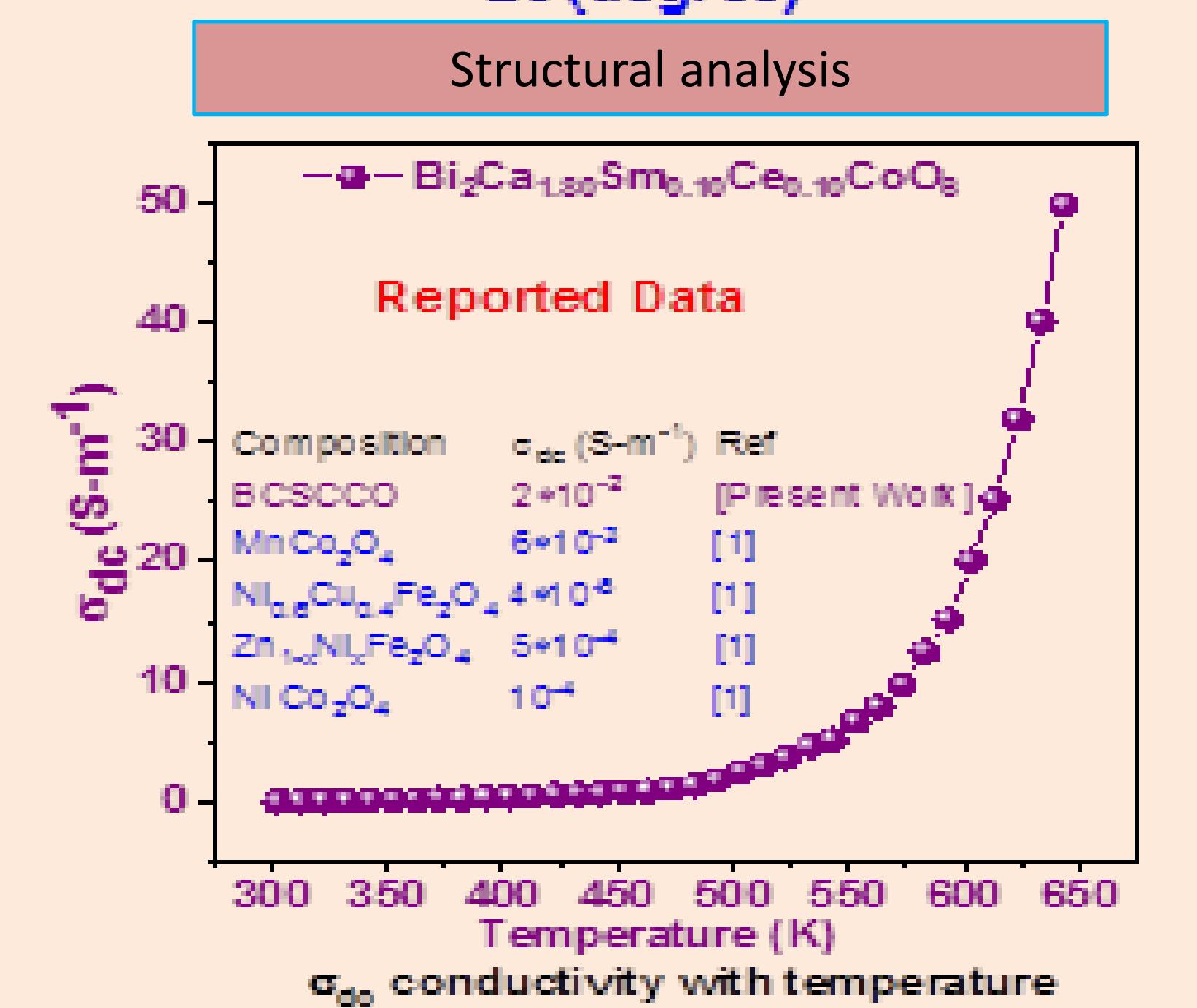
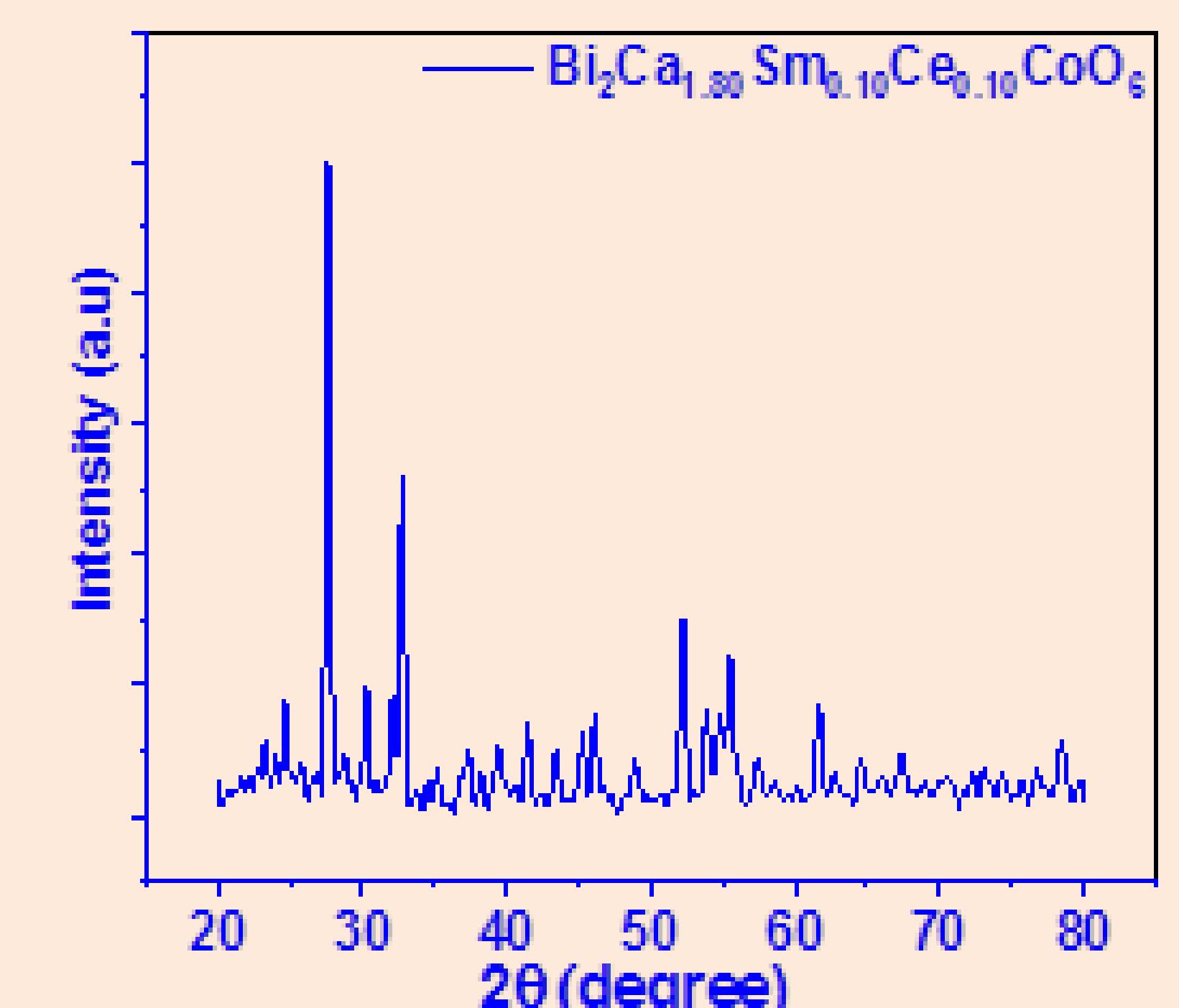
$\text{Bi}_2\text{Ca}_2\text{CoO}_6$
Crystal System



Synthesis Route



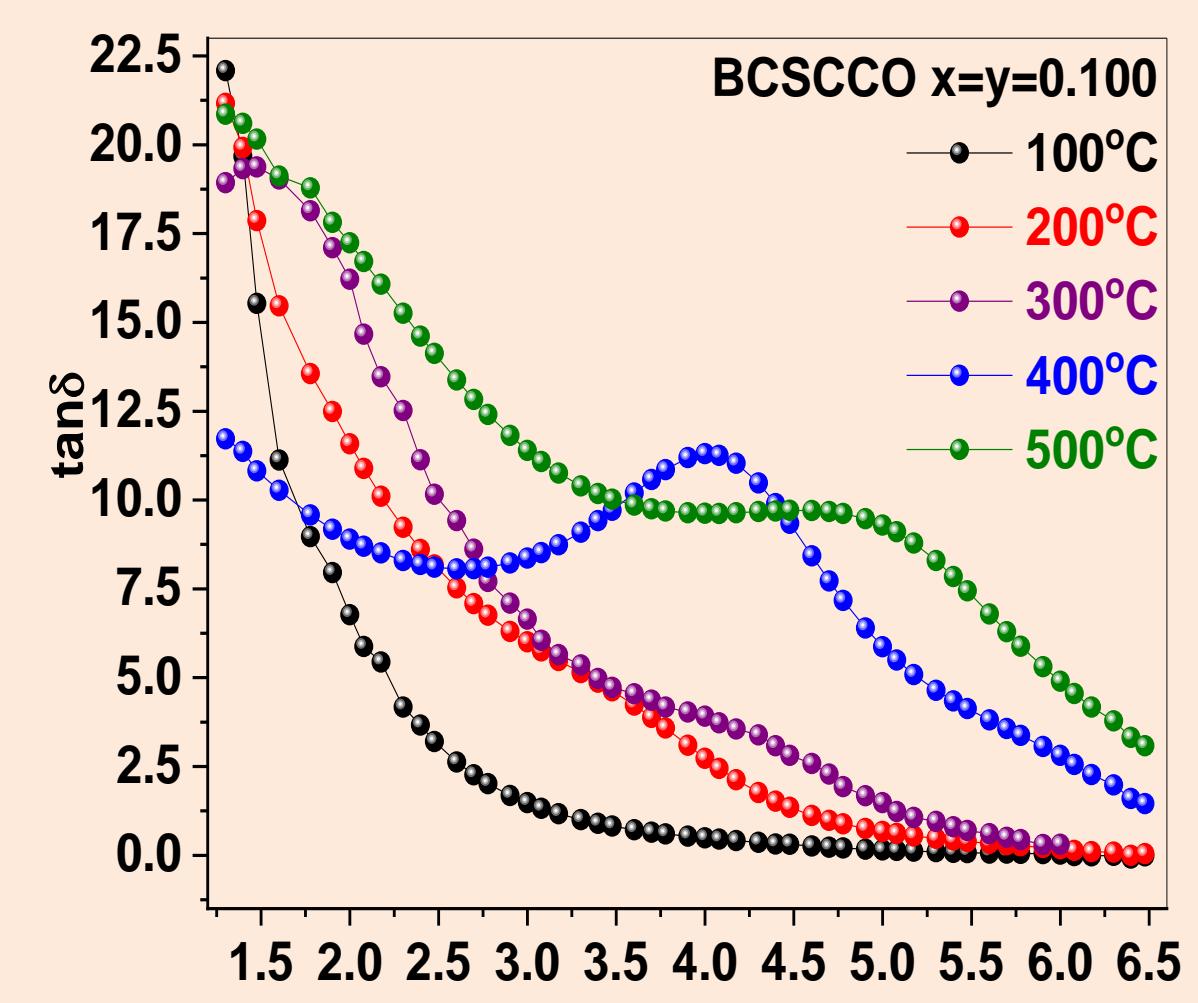
Results



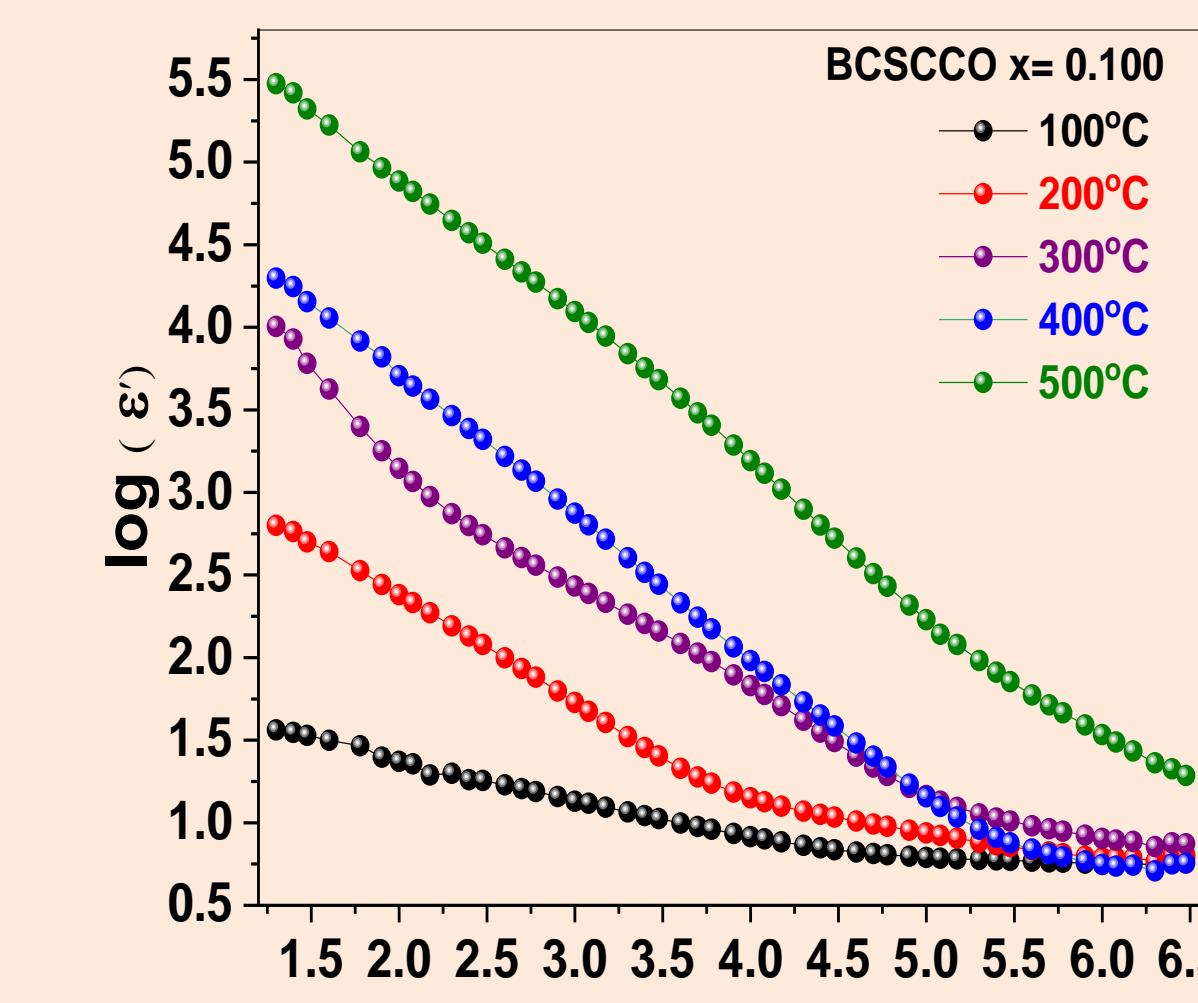
Acknowledgement



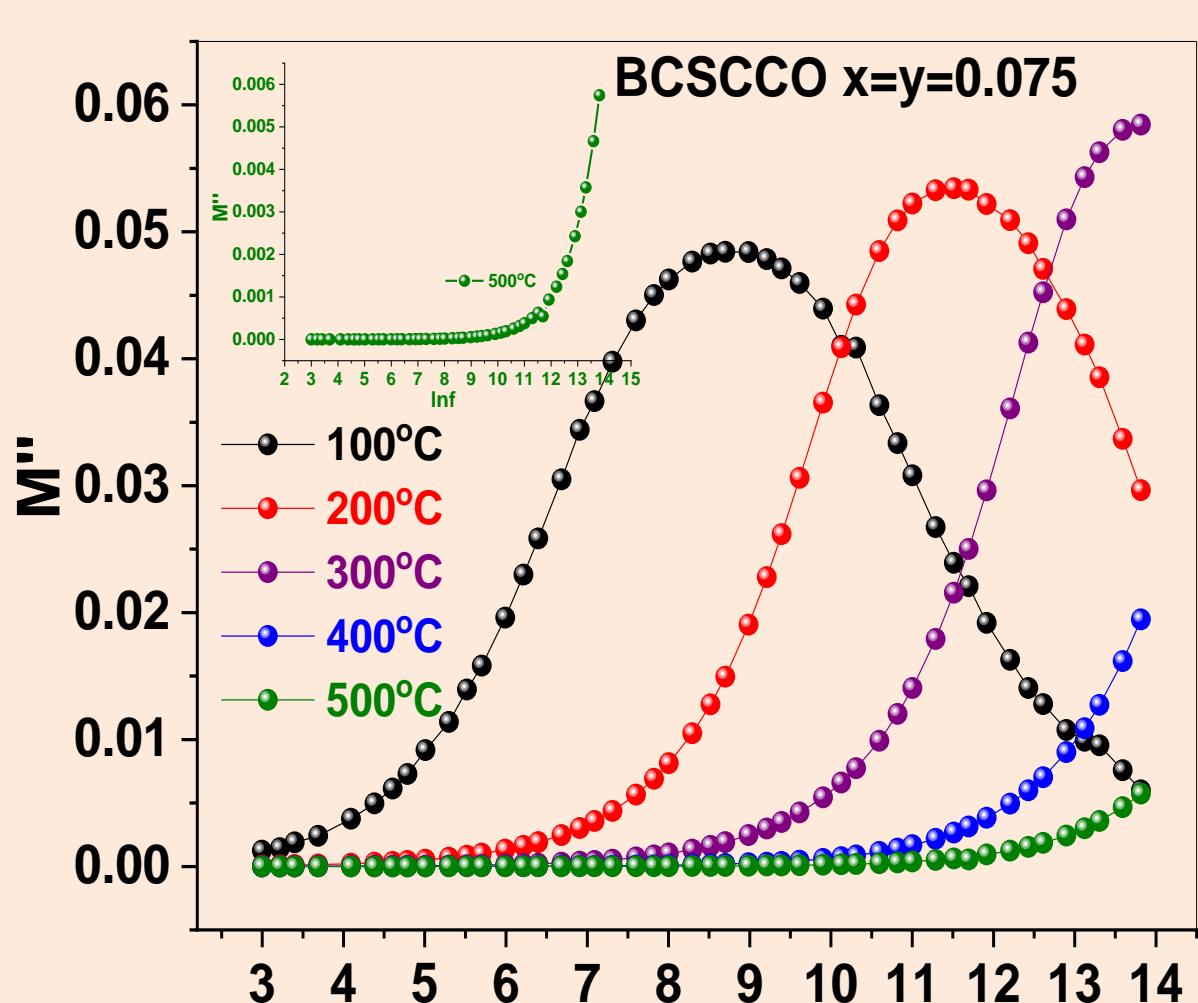
Results



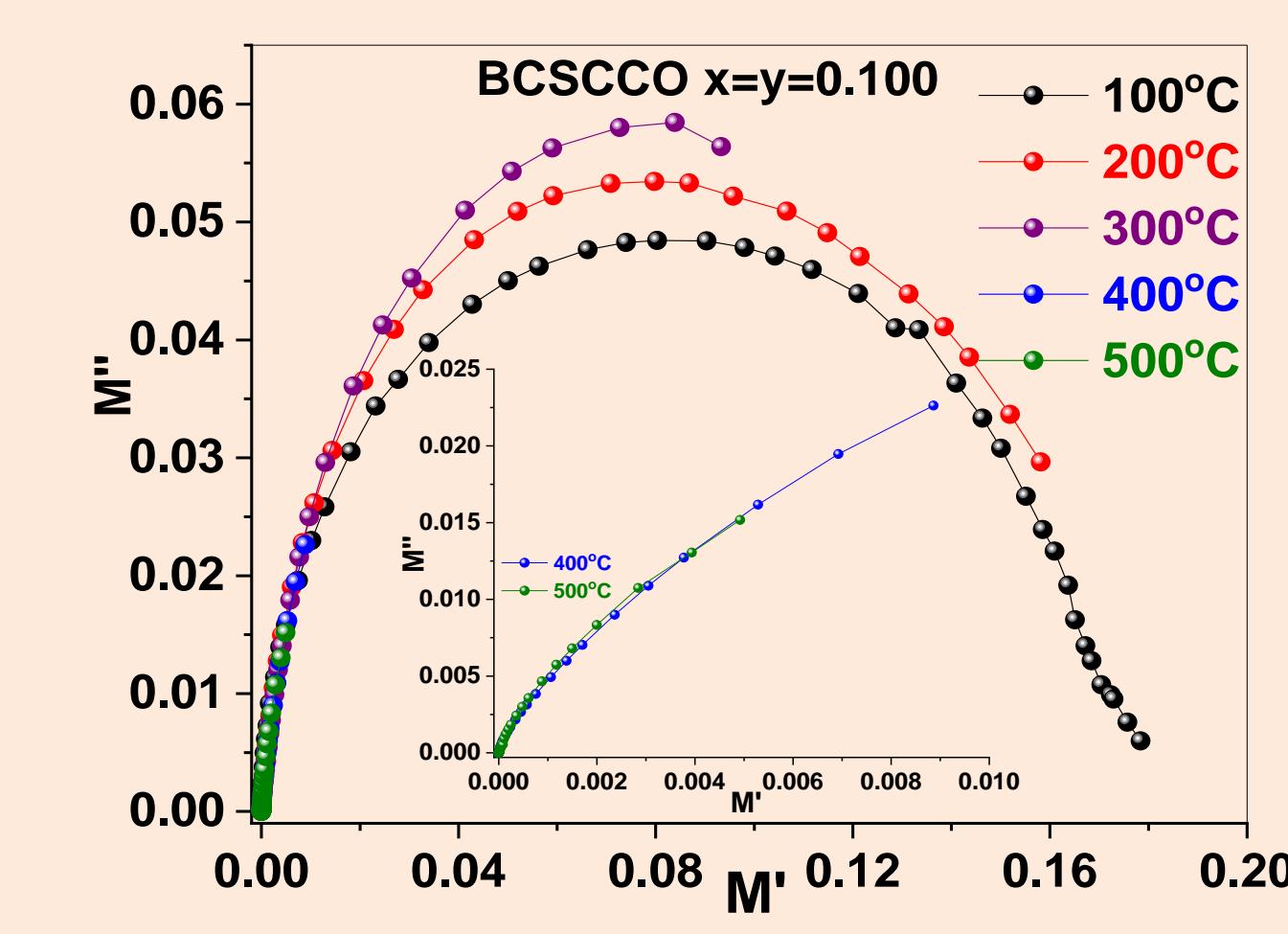
Loss Tangent with temperature



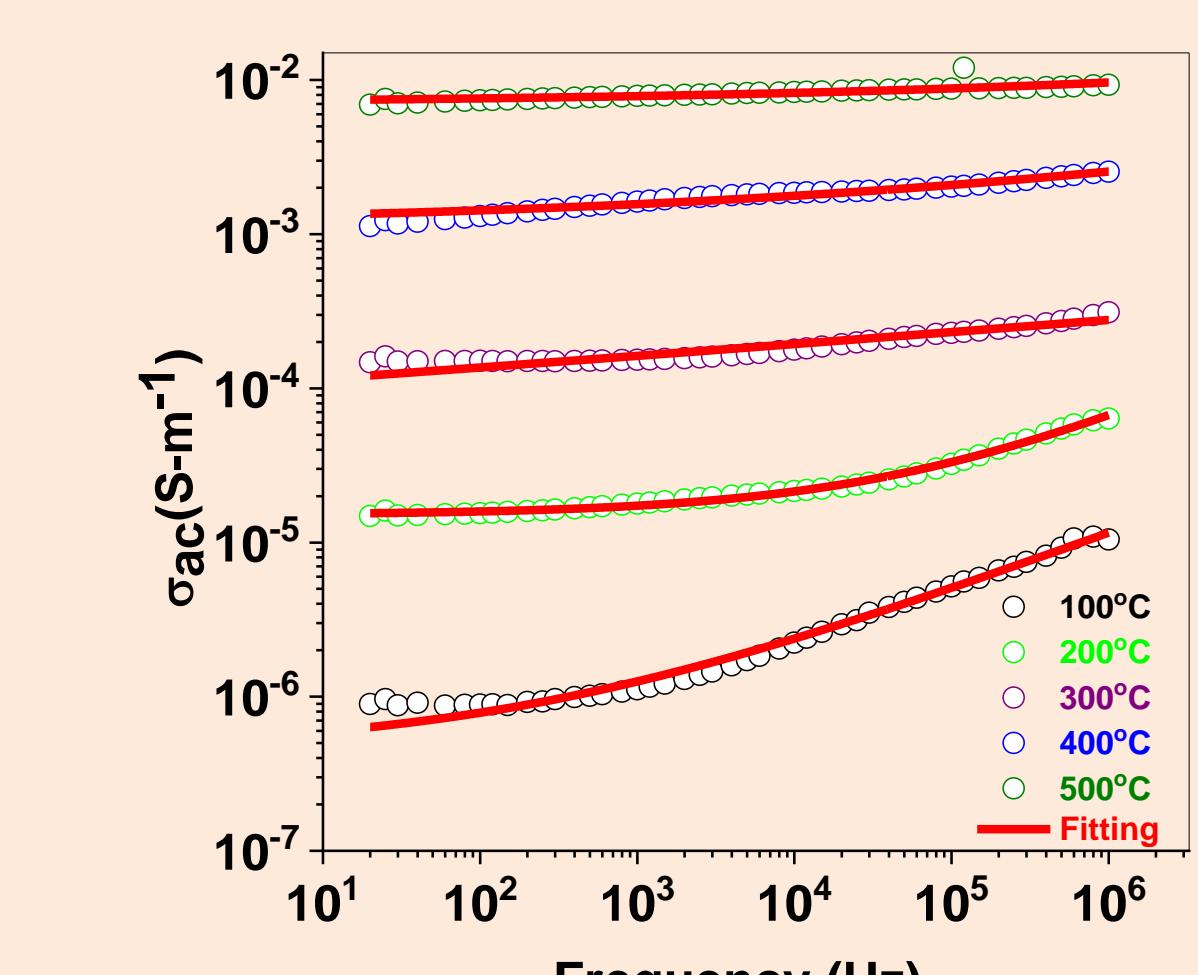
Dielectric Constant



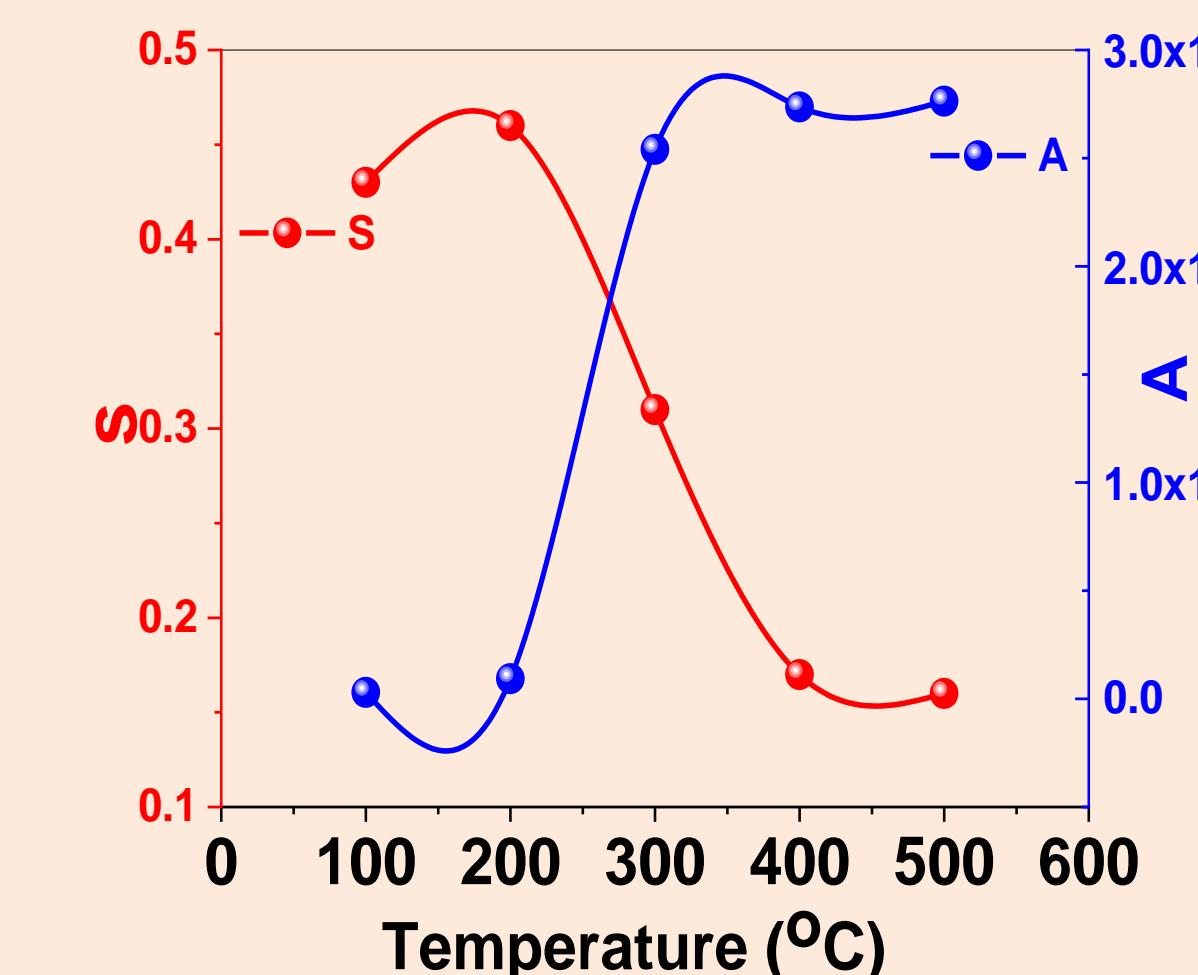
Imaginary part of modulus



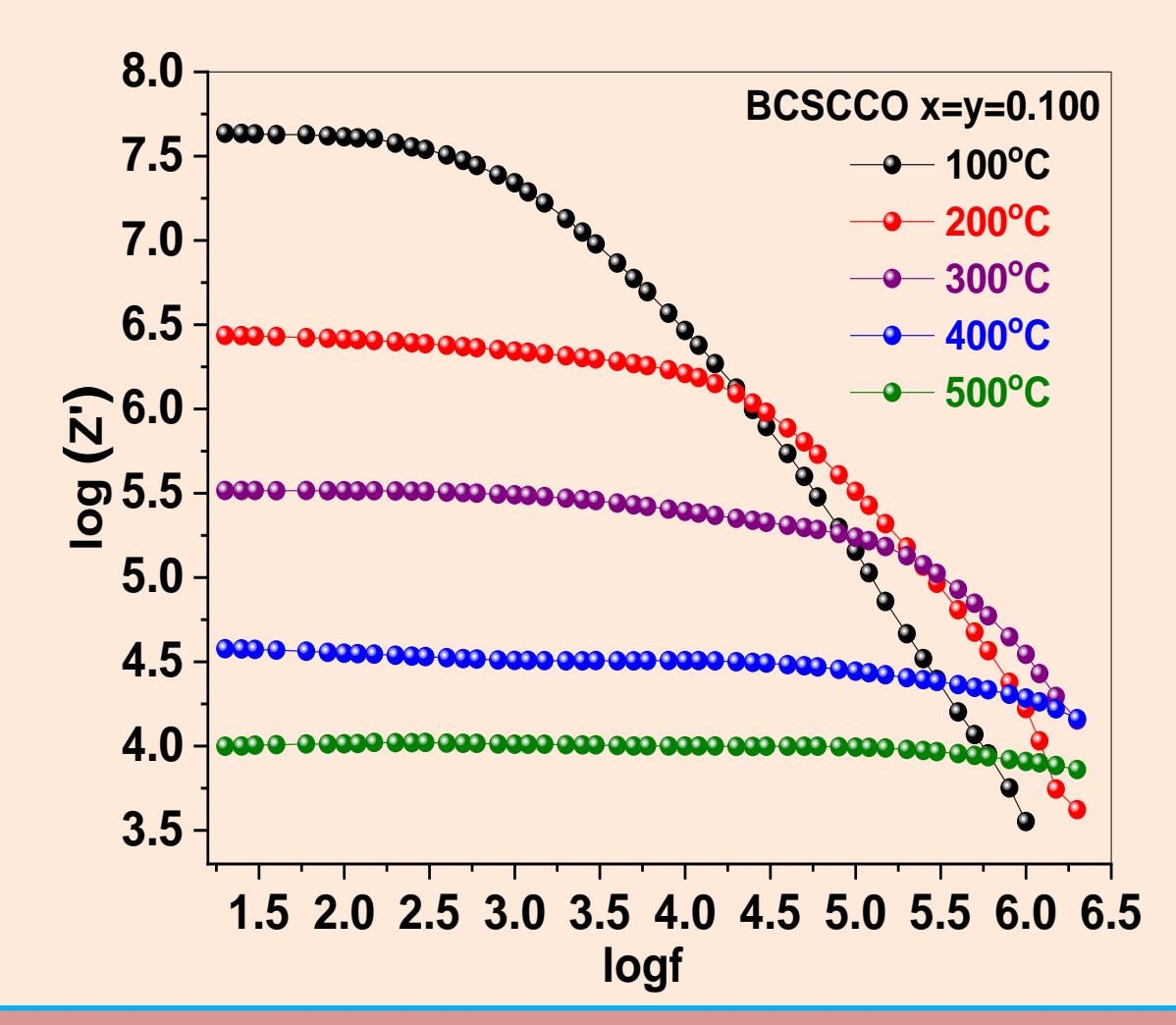
Modulus Spectroscopy



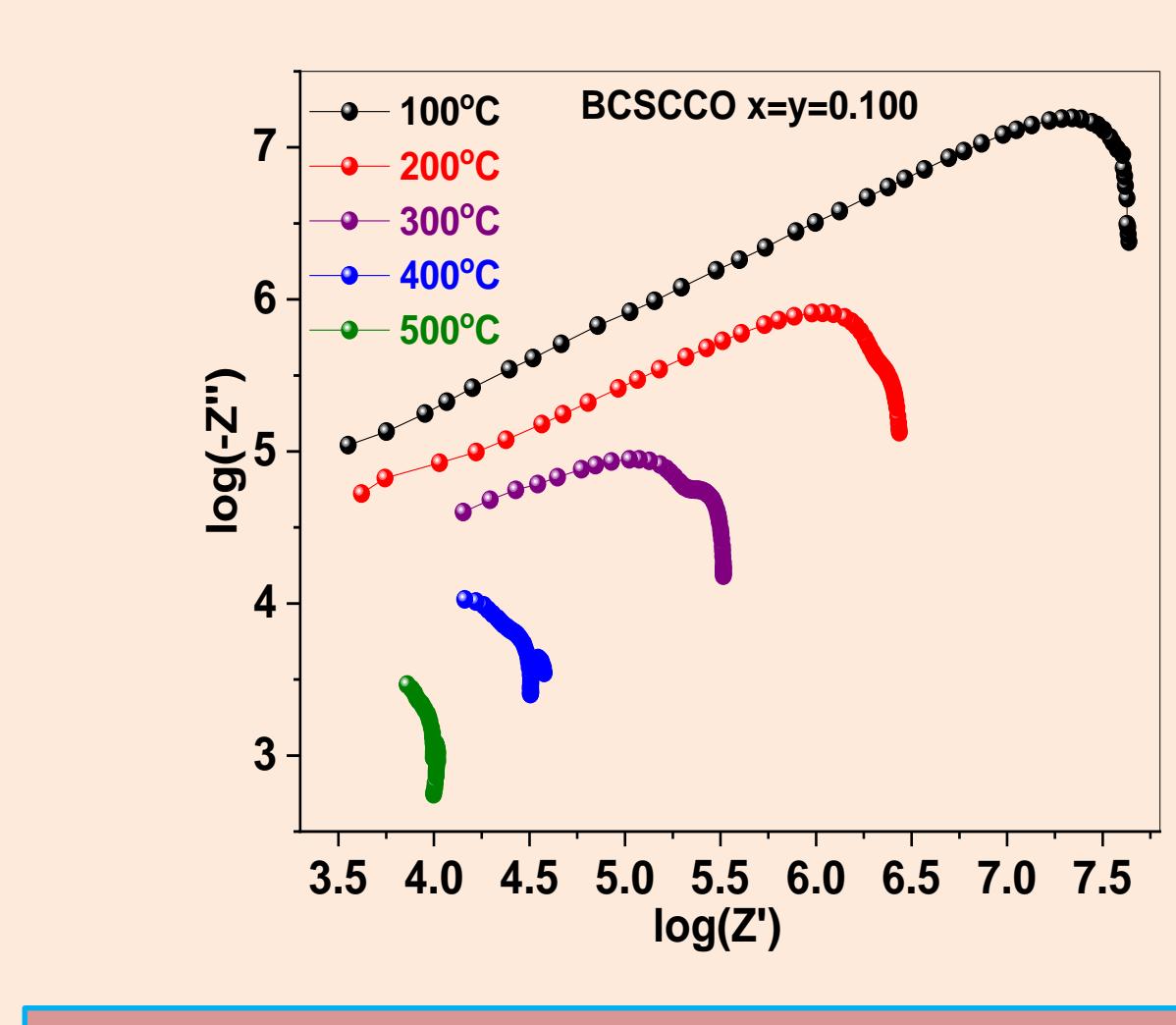
Jonscher Fitting



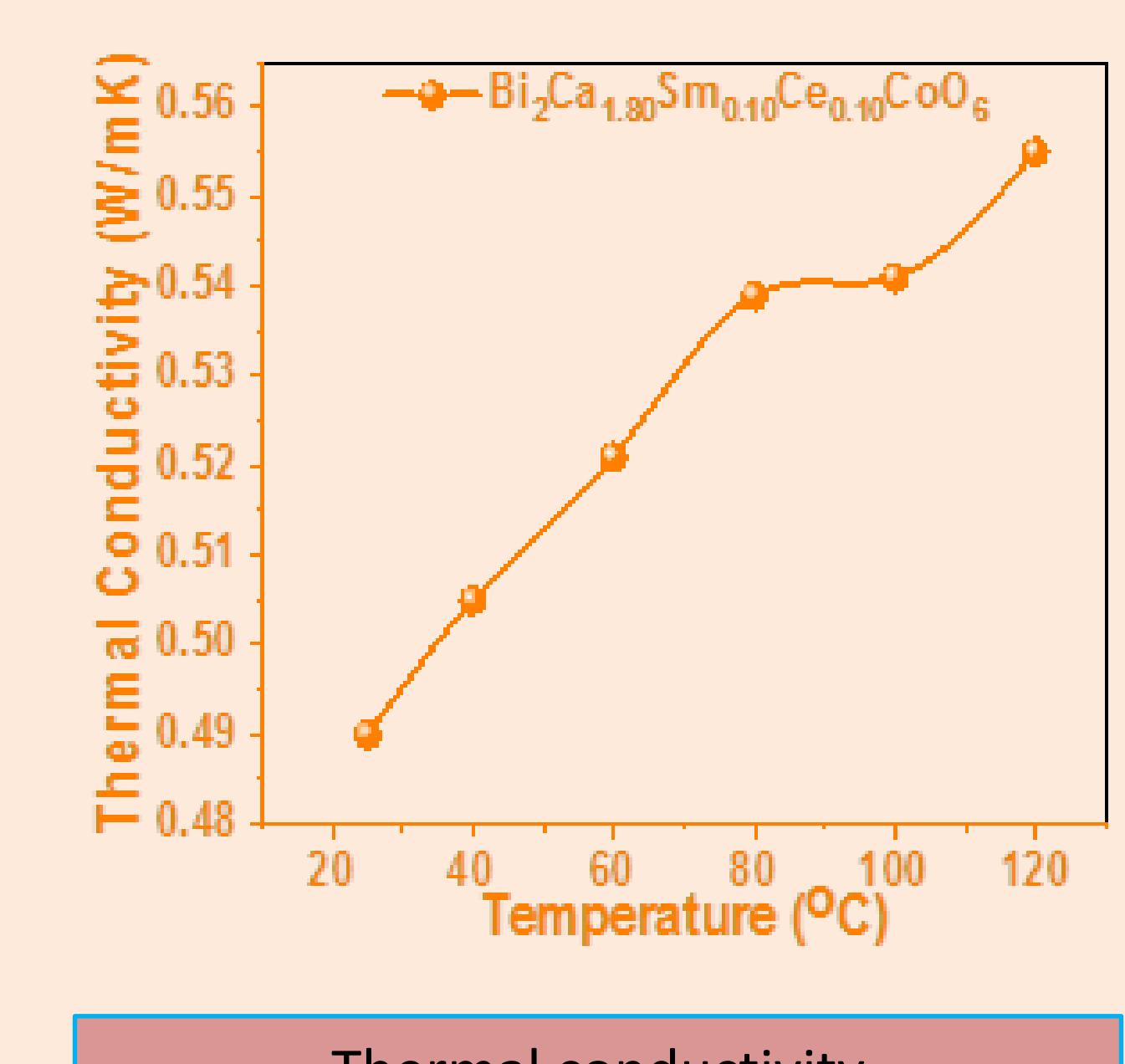
Variation of s and A with temperature



Resistive part of impedance



Cole-Cole plot



Thermal conductivity

Conclusion

- Synthesis of BCSCCO nanoparticles was done using co-precipitation route.
- Jonscher power law is used to investigate the conduction process.
- The prepared sample follow the Correlated barrier hopping conduction mechanism.
- The sample BCSCCO has a higher value of electrical conductivity.
- At 500°C the BCSCCO sample has been depicted with dielectric constant of 3.8×10^6 .
- The synthesized sample is suitable for renewable energy applications and energy storage.

References

- [1] Muraleedharan, S., Thiruvengatam, V., Karayi, S. O., Karuppiah, K., Jawahar, V., & Ashok, A. M. (2020). Investigation on temperature-dependent electrical properties of $\text{La}_{1-x}\text{A}_x\text{CoO}_3$ ($\text{A}=\text{La}, \text{Li}, \text{Mg}, \text{Ca}, \text{Sr}, \text{Ba}$). CrystEngComm, 22(1), 85-94