

Biomass, Bioenergy and Biofuels

Prof. Tony Bridgwater
Bioenergy Research Group
Aston University
Birmingham, UK



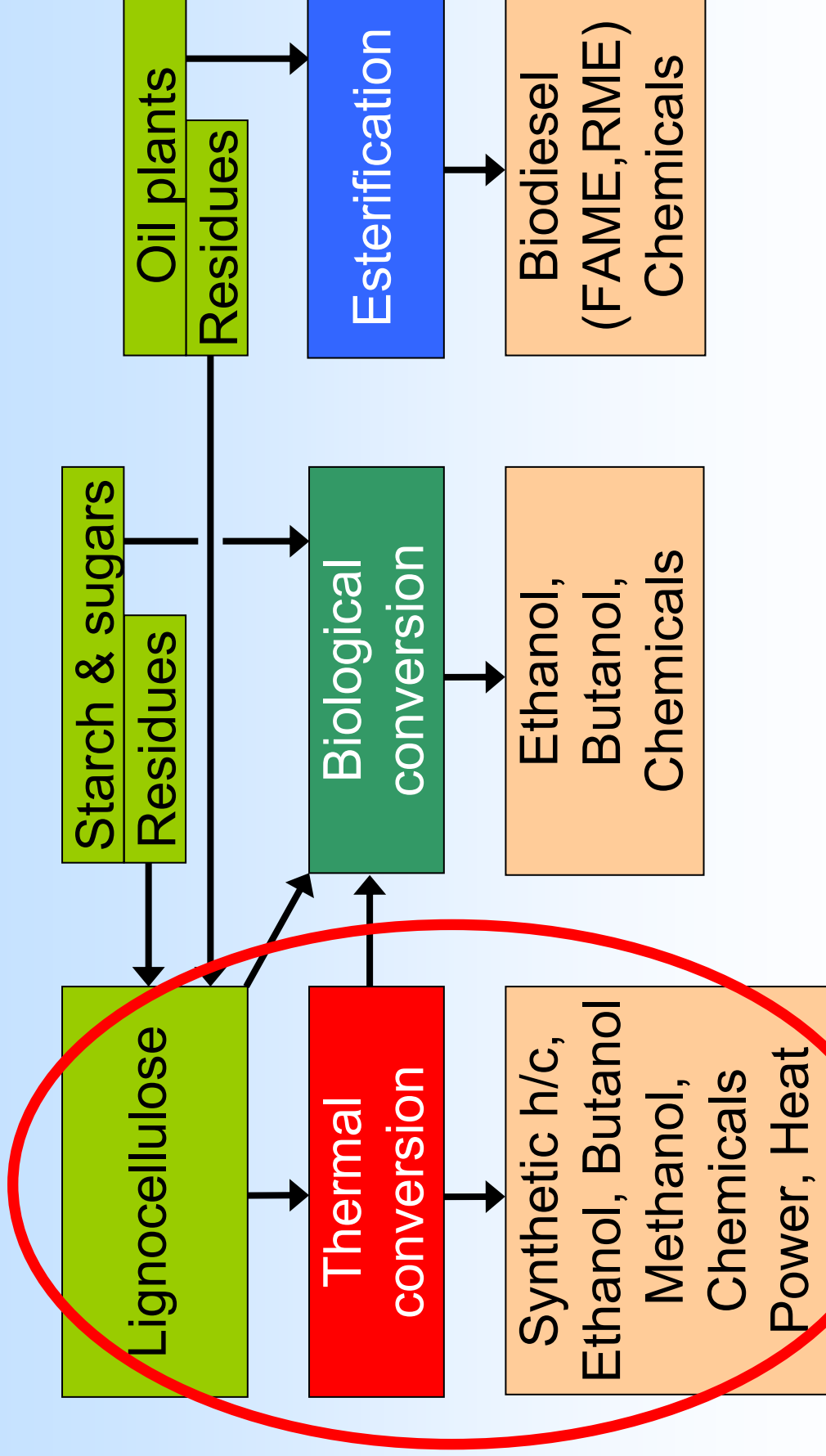
Biomass

- Energy crops
- Agricultural and forestry wastes
- Industrial & consumer wastes



- Biomass is widely dispersed, with few sites that can sustainably produce 100,000 dry t/y, about 20-25 MWe,

Feeds, processes, products

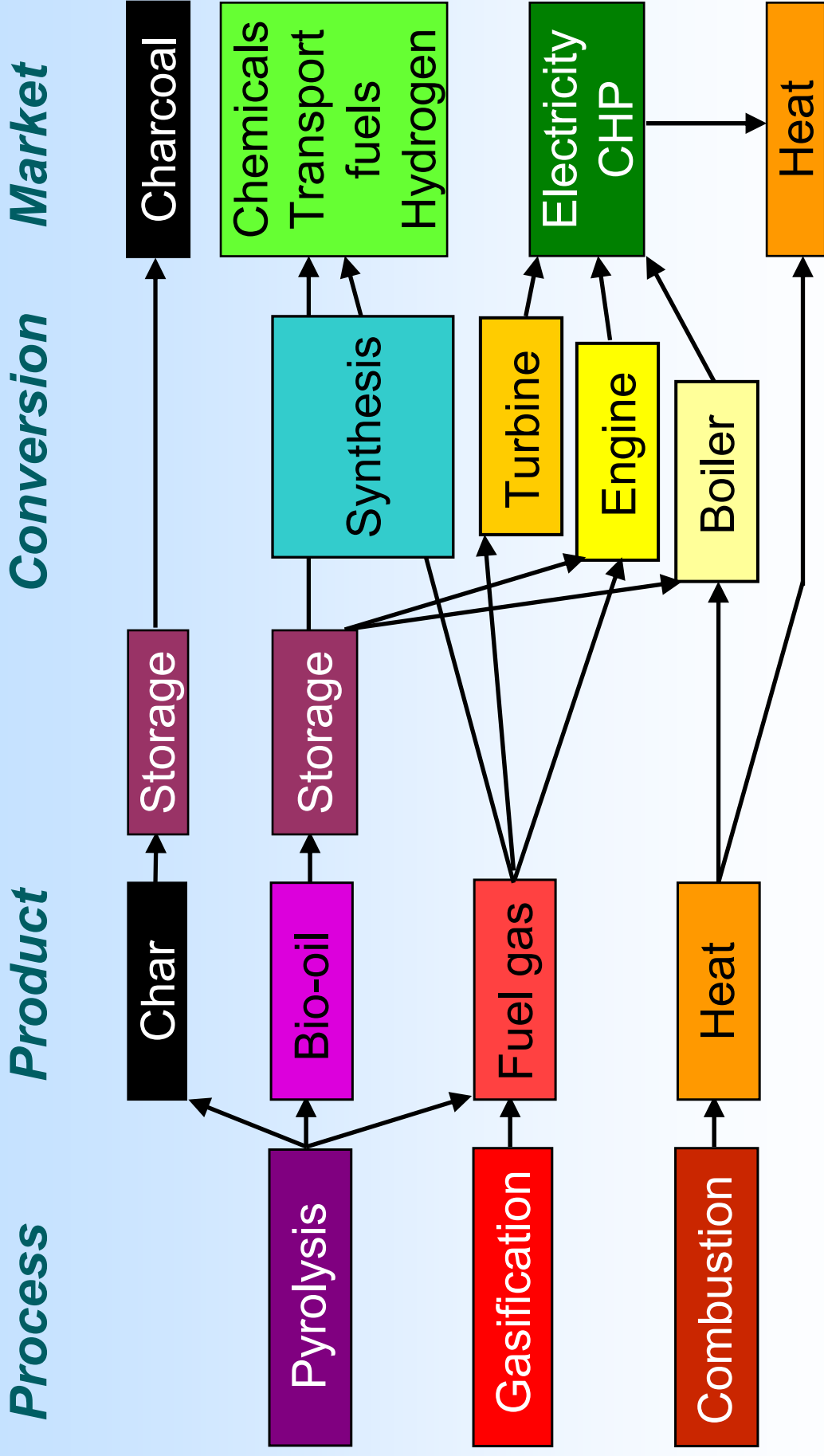


Thermal vs biological

- Thermal conversion
 - Dry feed needed
 - Fast processes
 - Less selective - Mixed products
 - More versatile in range of products and applications
- Biological conversion
 - Wet feed acceptable
 - Slow processes
 - More selective - Single product generally (e.g. ethanol)
 - Less versatile



Thermal Biomass Conversion

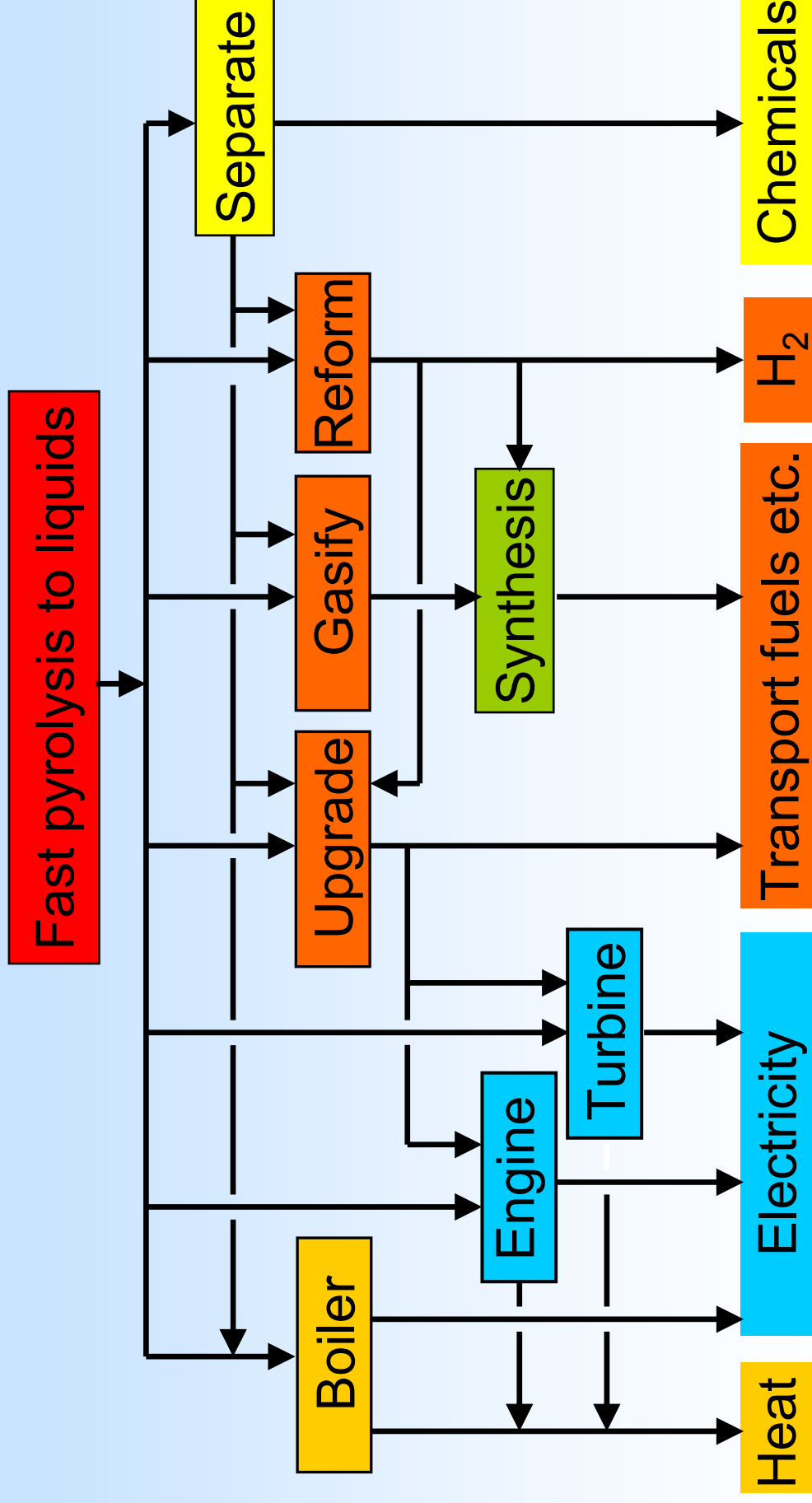


Pyrolysis

	Liquid	Char	Gas
FAST PYROLYSIS moderate temperature (~500C) short hot vapour residence time (<2 s)	75% 25% water	12%	13%
INTERMEDIATE PYROLYSIS Low-moderate temperature, moderate hot vapour residence times	50% 50% water	25%	25%
SLOW PYROLYSIS Low-moderate temperature, long residence times	30% 70% water	35%	35%
GASIFICATION high temperature (>800C), long vapour residence time	5% (tar) 5% water	10%	85%



Applications for bio-oil



Fast pyrolysis for liquids

- Biomass is heated as quickly as possible
- To a carefully controlled temperature ($\sim 500^{\circ}\text{C}$)
- Products are cooled as quickly as possible ($< 2\text{ s}$)
- The liquid has unique properties
 - Dark brown mobile liquid,
 - Combustible, but not flammable,
 - Heating value $\sim 17\text{ MJ/kg}$,
 - Not miscible with hydrocarbons,
 - Density $\sim 1.2\text{ kg/l}$,
 - Acid, pH ~ 2.5 ,
 - Pungent odour,



Process requirements

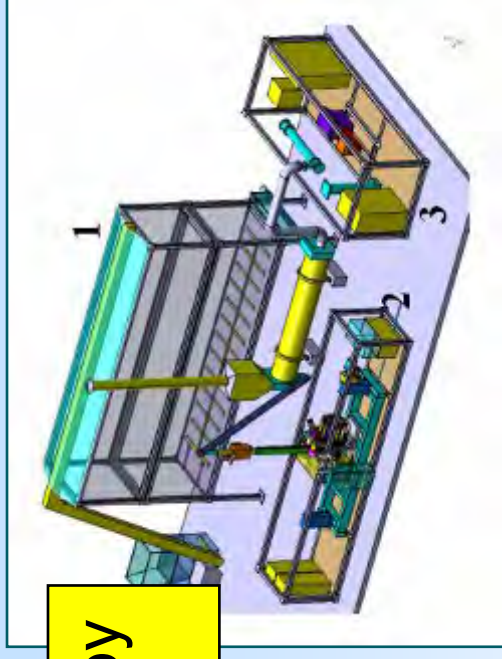
- Drying**
 - <10%. Feed and reaction water report to bio-oil
- Comminution**
 - -2mm (fluid bed), -6 mm (CFB)
- Fast pyrolysis**
 - high heat rate, controlled T, short residence time,
- Char separation**
 - Efficient char separation needed
- Liquid recovery**
 - By condensation and coalescence.



Pyrolysis examples



100 & 200
t/d by
Dynamotive



50 t/d by
Pytec



50 t/d by
BTG



Mobile units



50 t/d by
Ensyn

Gasification

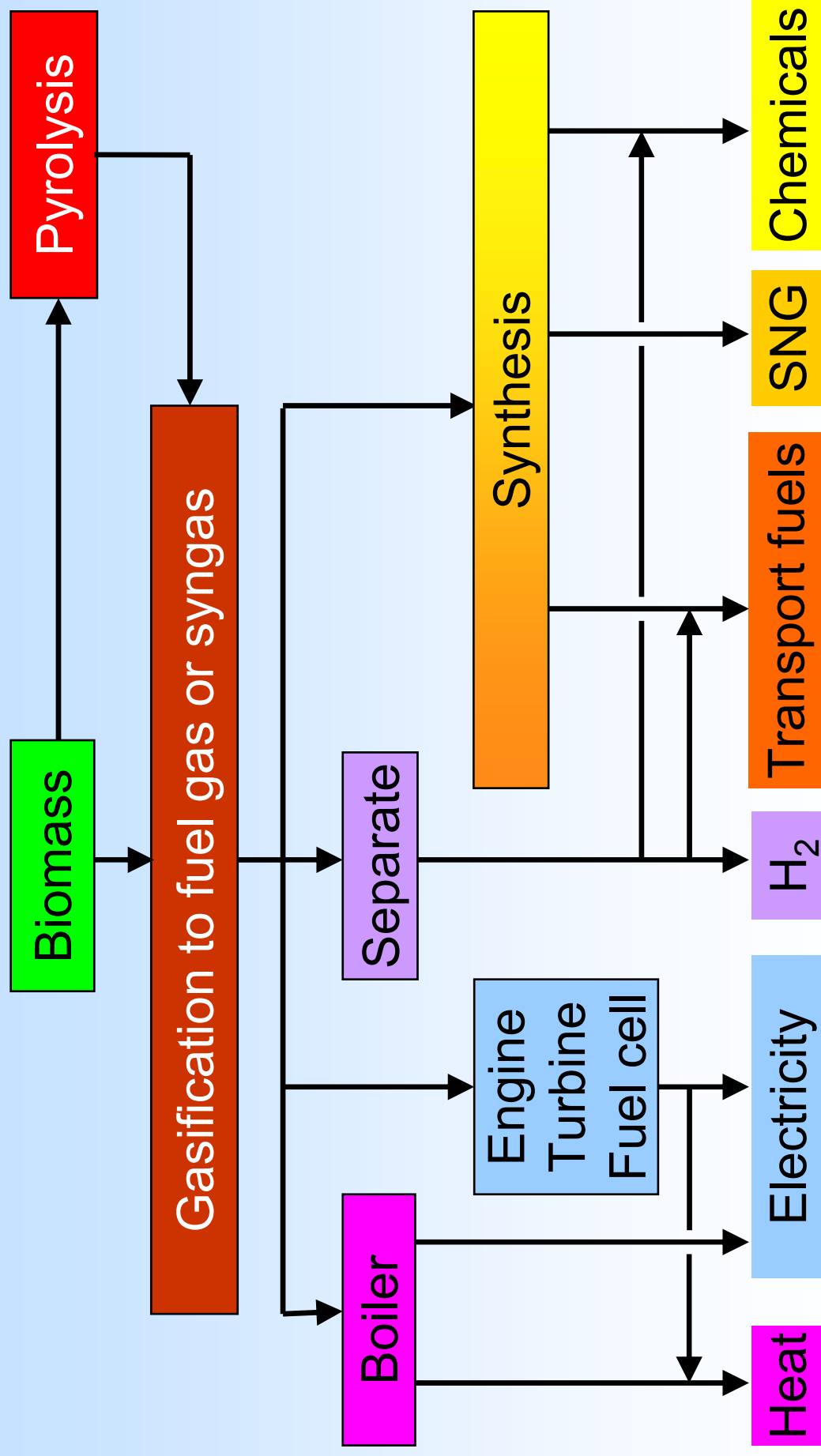
- Gasification is the substantial conversion of organic material into a combustible gas.
- This can be thermal:
 - Partial oxidation by incomplete combustion with air or oxygen to give a combustible fuel gas or synthesis gas. This is mostly carbon monoxide, carbon dioxide, hydrogen, and nitrogen if air is used.
 - Steam gasification or pyrolytic gasification using heat derived from combustion of byproduct char, without addition of oxygen or air. The product gas also contains significant quantities of methane.
- Or it can be biological:
 - Anaerobic digestion to methane and carbon dioxide.



Gasification types

Type	Gas HV	Efficiency	Comments
AIR	~ 5 MJ/Nm ³	High	Simple
OXYGEN	~ 10 MJ/Nm ³	Moderate	Costly
STEAM (PYROLYSIS)	~ 15 MJ/Nm ³	Low	Complex process, Endothermic
PRESSURE AIR	~ 5 MJ/Nm ³	High	Costly, but high IGCC potential
OXYGEN	~ 10 MJ/Nm ³	Moderate	
BIOGAS	~ 18 MJ/Nm ³	Moderate	
Methane for comparison	35.5 MJ/Nm ³		

Applications for fuel gas or syngas



Gasifier examples



2 MWe by
Austrian
Energy at
Guessing

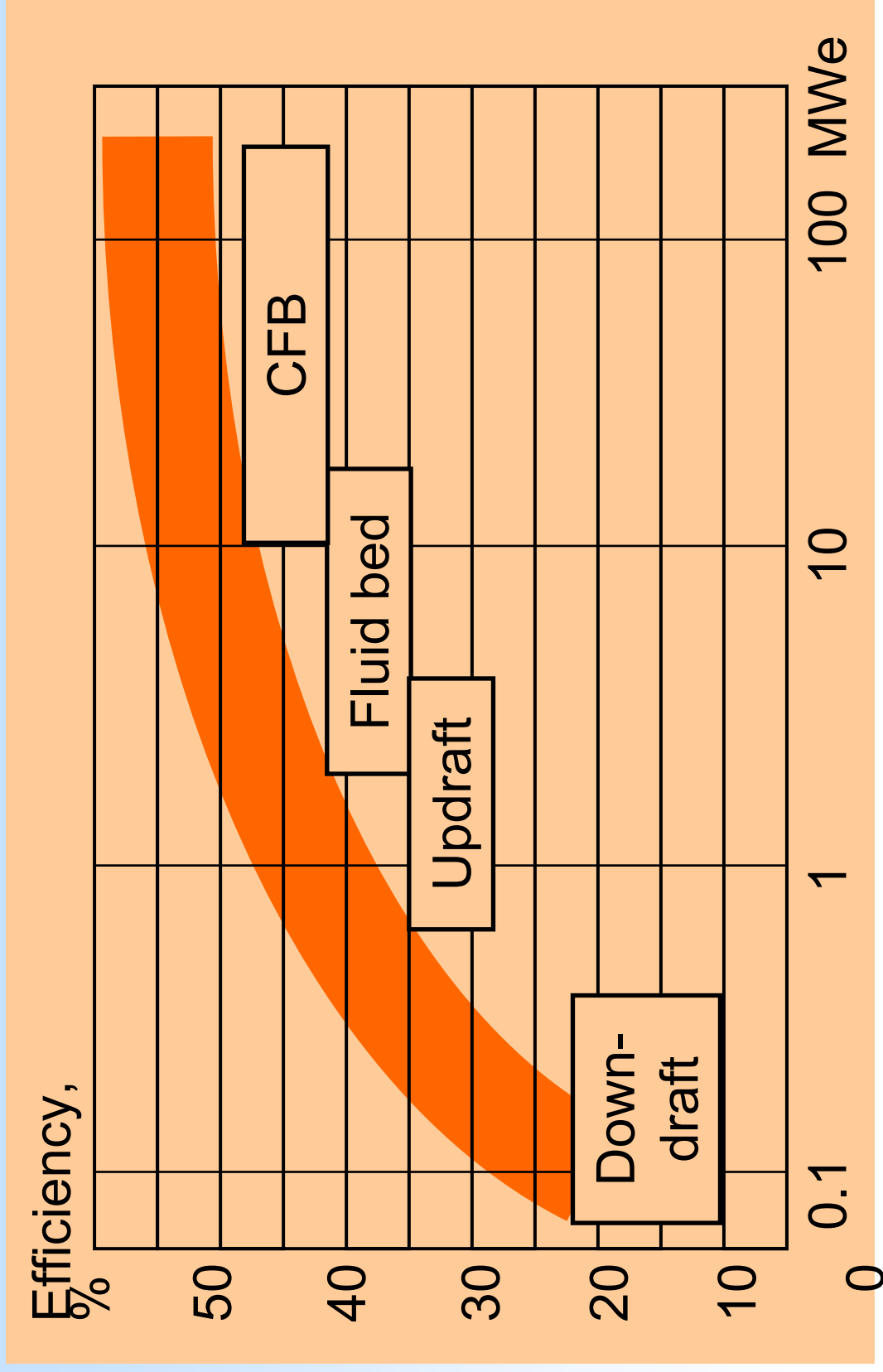


50 kWe by
Biomass
Eng'g in UK

45 MWe
co-firing
at Lahti



Efficiency, technology & scale



Main challenge - gas quality

Contaminant	Examples	Problem	Solution
Particulates	Ash, char, fluid bed material	Erosion	Cyclone, barrier filter.
Alkali metals	Na, K compounds	Hot corrosion	Temperature.
Fuel-bound nitrogen	Ammonia, HCN	NOx	Temperature, SCR.
Tars	Refractive aromatics	Clogs filters, Difficult to burn, Deposits.	Cracking, Removal, Combustion.
Sulphur, chlorine	HCl, H ₂ S	Corrosion Emissions	Capture, Washing.

Renewable transport fuels

Oxygenates	Generation	Process
● Methanol	2°	Thermal
● Ethanol	1° & 2°	Biological or Thermal
● Butanol	1° & 2°	Biological or Thermal
● Mixed alcohols	2°	Thermal
● Dimethyl ether	2°	Thermal
Hydrocarbons		
● Biodiesel	1°	Physical + chemical
● Synthetic diesel	2°	Thermal
● Synthetic gasoline	2°	Thermal
● Methane (CSNG)	1° & 2°	Thermal
Other		
● Hydrogen	1° & 2°	Thermal or Biological

Synthetic hydrocarbons

- Synthetic hydrocarbons are entirely compatible with conventional crude oil-derived fuels in all proportions, but much cleaner. They include diesel, gasoline, kerosene
- The process is biomass or bio-oil gasification to a clean syngas and synthesis of hydrocarbons.
- Concept proven from coal in South Africa and from gas in Far East
- Biomass gasification technology is unproven at large scale. Gas cleaning claimed to not be a problem, but there is no evidence and no large scale experience



Gasification + Fischer-Tropsch

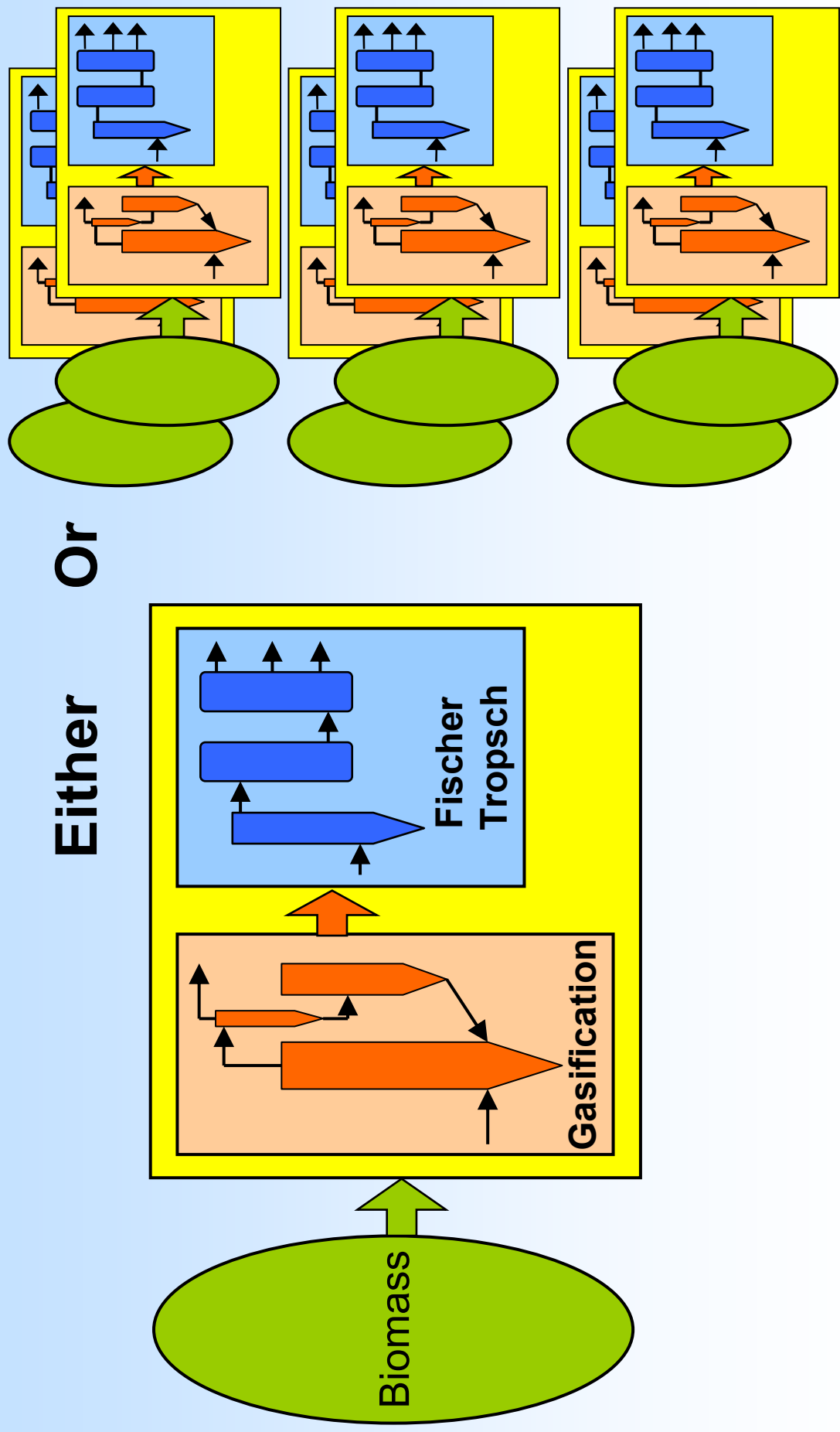
Fischer Tropsch is claimed to require a minimum size of 1 million t/y fuels to be economic

Biomass is a dispersed resource that has to be collected and transported over large distances. There are 2 options:

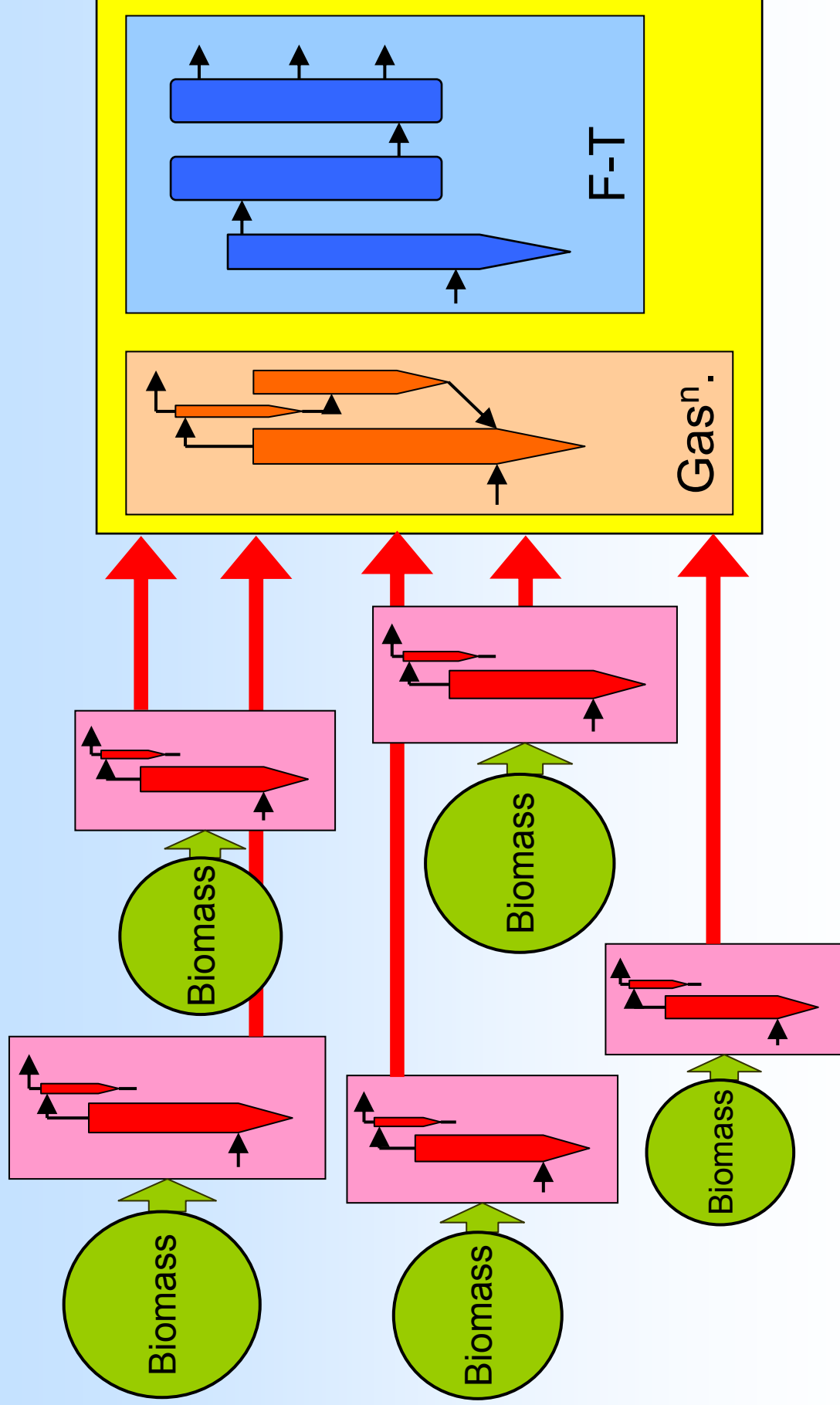
1. Large gasification plants integrated with FT synthesis.
This requires delivery of very large quantities of biomass
– around 5 million t/y to meet current views on economies of scale for FT plants.
2. Small multiple gasification plants integrated with downscaled Fischer-Tropsch (FT) hydrocarbon synthesis. This is costly due to absence of economies of scale and downscaling is unproven.



Gasification + Fischer-Tropsch



Pyrolysis + Fischer Tropsch



Comparison liquid vs solid feed

- Capital cost **increase** of ~10% due to diseconomies of scale in small pyrolysis plants
- Capital cost **reduction** of ~8% due to lower raw material handling costs
- Capital cost **reduction** of ~12% due to lower gasification costs in feeding a liquid at pressure compared to solid biomass
- Efficiency **loss** of ~6% due to additional processing step

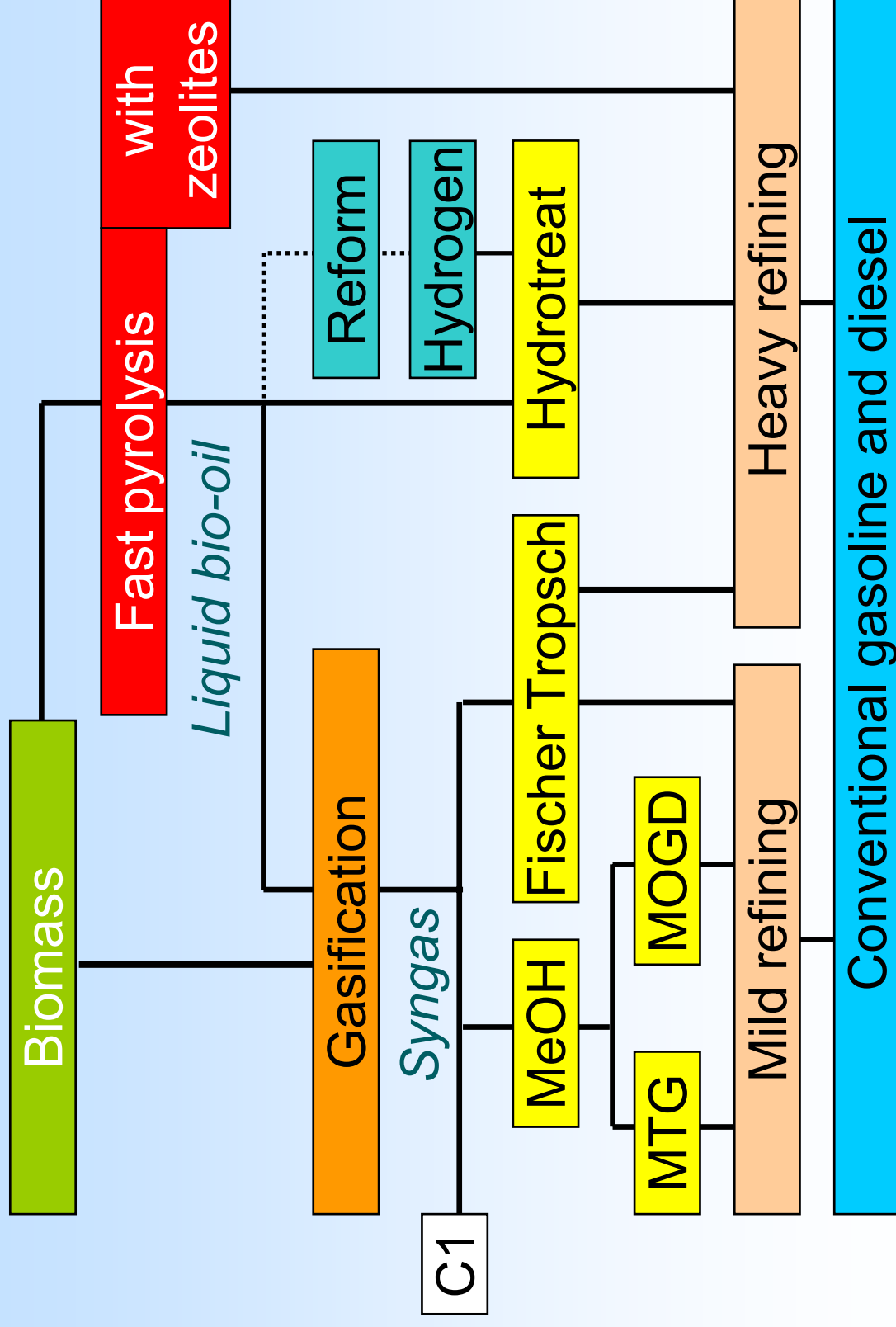


Routes to bio-hydrocarbon fuels

1. Thermal gasification + Fischer Tropsch
 - Solid biomass
 - Pyrolysis liquid as a preparation step
2. Thermal gasification + Methanol synthesis + upgrading by MTG or MOGD
3. Pyrolysis + upgrading by hydro-processing
4. Hydro-processing vegetable oil

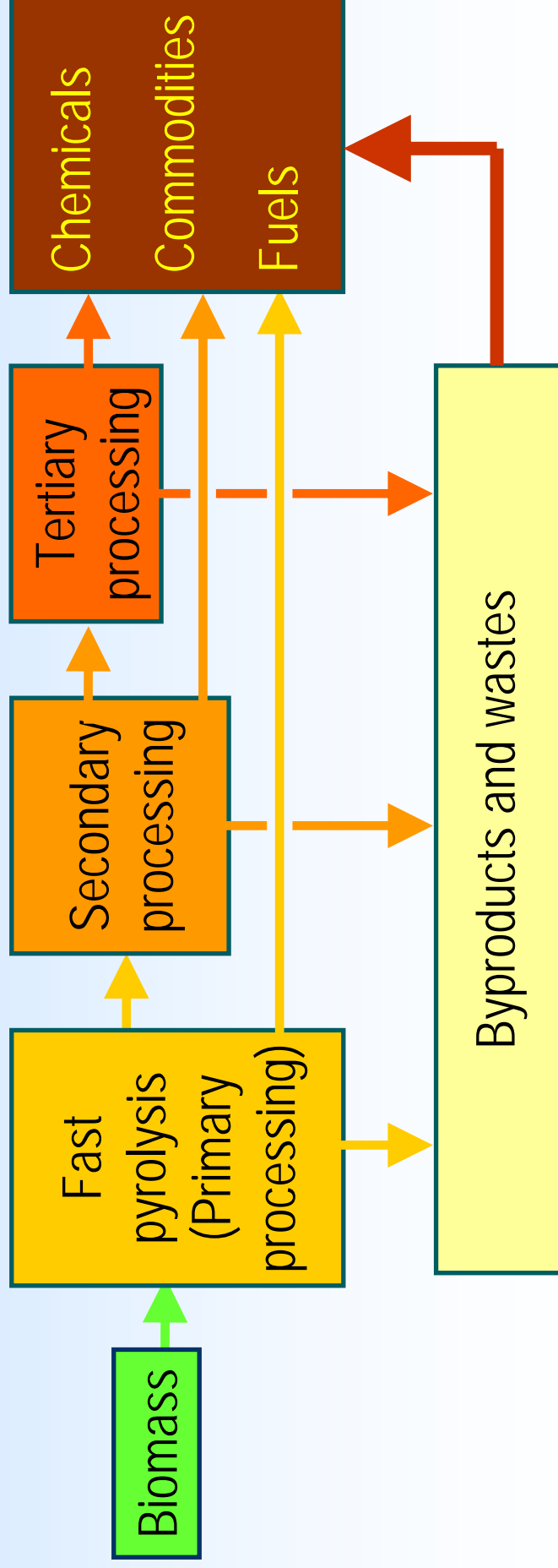


Transport fuels from biomass



Biorefinery concept

- Integrate production of higher value chemicals and commodities, as well as fuels and energy,
- Optimise use of resources, maximise profitability, maximise benefits, minimise wastes

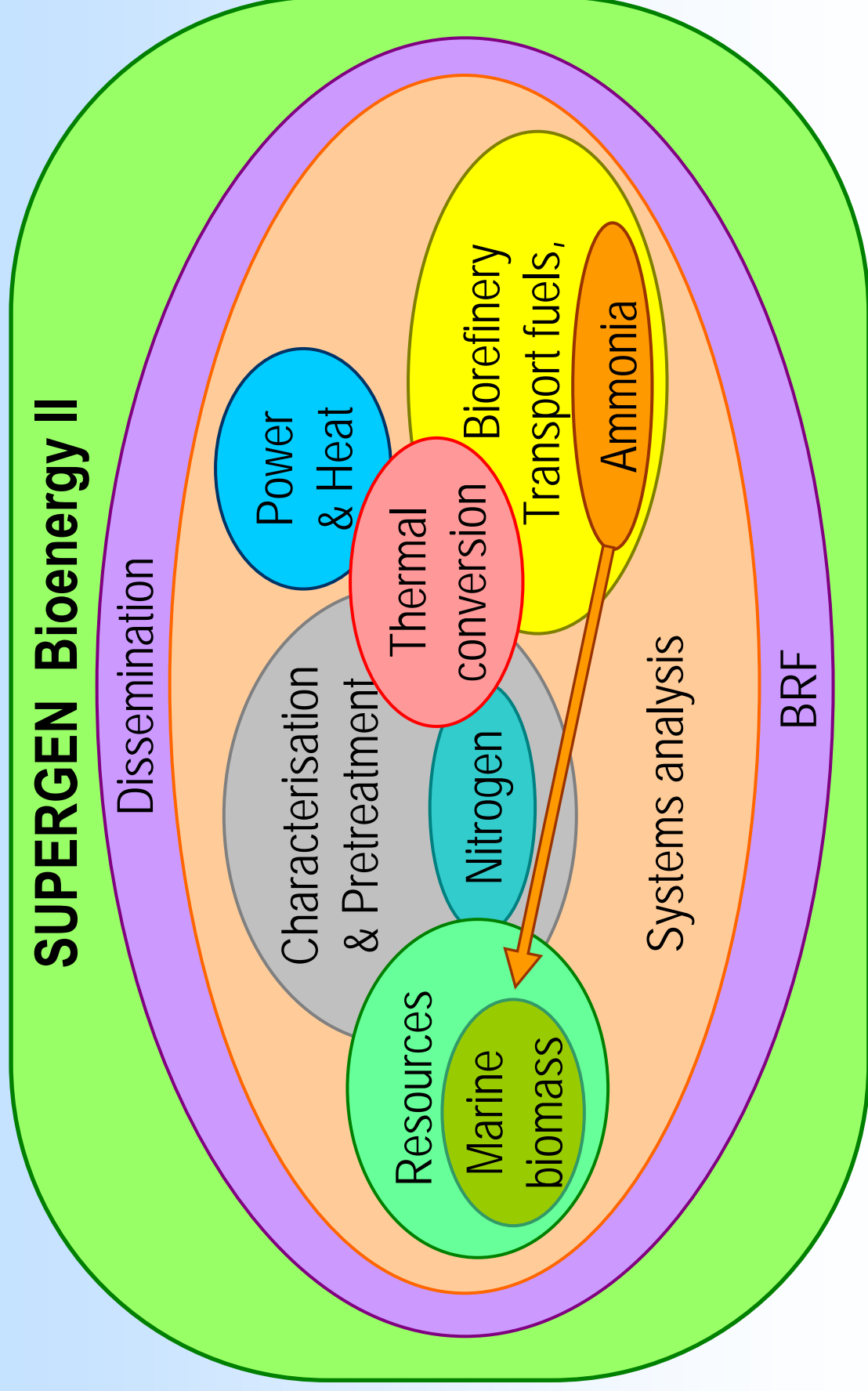


SUPERGEN Bioenergy

- SUPERGEN Bioenergy is one of 13 academically led consortia funded by EPSRC researching renewable energy
- A consortium of 14 research organisations and 11 companies
- Continuation of a 4 year phase 1 project with a second 4 year term and a £6.35 million budget
- Continuation of the focus on interfaces between biomass production, thermal conversion and product utilisation with added activities on:
 - marine biomass as a new resource,
 - nitrogen as a significant component of biomass,
 - second generation biofuels,
 - biorefineries and fertilisers



SUPERGEN Bioenergy structure



Partners

ACADEMIC

- Aston (Management)
- Cranfield
- Forest Research
- Imperial College
- IGER
- Leeds (Finance)
- Manchester
- Policy Studies Institute
- Rothamsted Research
- Sheffield

ASSOCIATE PARTNERS

- Irish Seaweed Centre, Oxford, SAMS, Ulster

INDUSTRIAL INVITATIONS

- Alstom
- AMEC
- Bical
- BIFFA
- Biomass Engineering
- BP
- Coppice Resources
- E.On
- Johnson Matthey
- RWE
- Rural Generation



Overall conclusions

- Bio-oil from fast pyrolysis:
 - Storable and transportable
 - Successfully used in boilers, gas turbines, diesel engines, and for chemicals,
 - Good economics and efficiency as an energy carrier for synthesis of transport fuels etc.
- Gasification
 - More developed commercially,
 - Successfully used for heat and power production, but has to be close coupled to applications,
- Biorefineries
 - Opportunity to improve costs and performance
- SUPERGEN Bioenergy is contributing to all aspects of the work described here