

## Electronic Supplementary Information

### Synergy of $Mg^{2+}$ and poly(aspartic acid) in additive-controlled calcium carbonate precipitation

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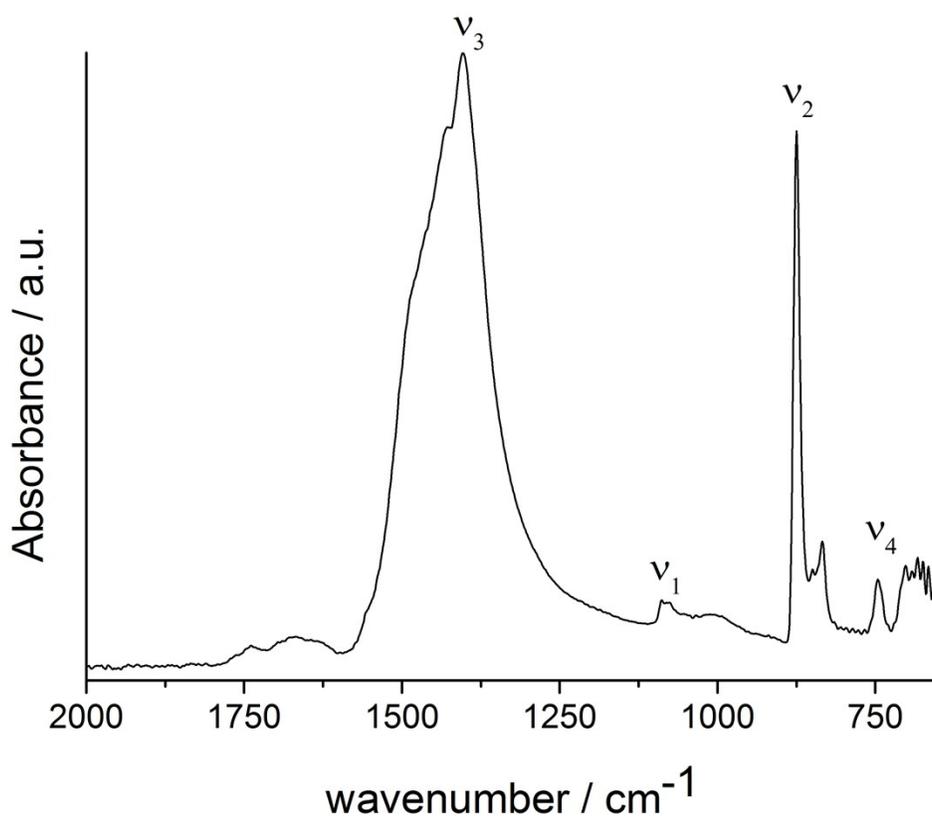


Figure S1: IR spectrum of a sample obtained with 10  $\mu\text{g/ml}$  PAsp at the end of the titration assay experiment. The bands are consistent with vaterite – ( $\nu_4 = 743 \text{ cm}^{-1}$ ).

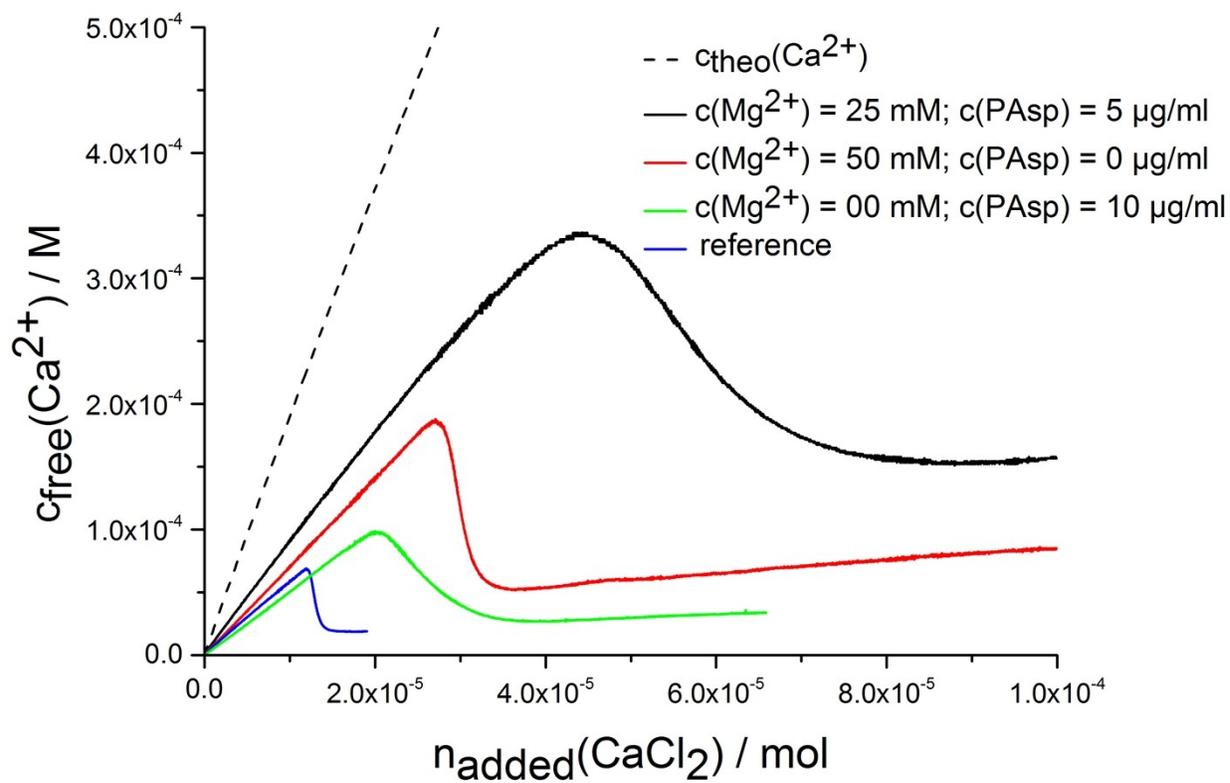


Figure S2 Development of free calcium concentration upon continuous addition of calcium solution with and without additives (as indicated) into 10 mM carbonate buffer (pH 9.75) at a rate of 10  $\mu\text{l/min}$ .

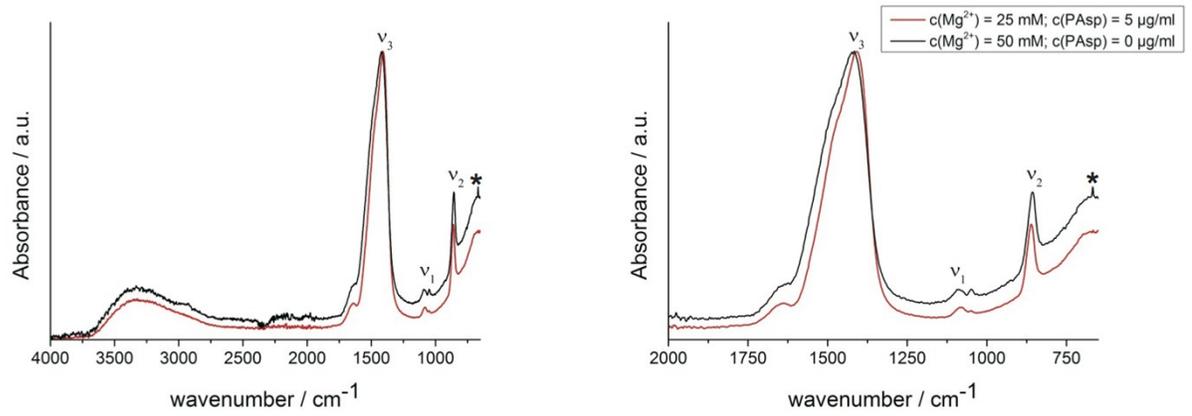


Figure S3: Complete spectral range of the IR-spectra of the ACCs formed in presence of Mg and both Mg and PAsp. The band marked by an asterisk is due to atmospheric CO<sub>2</sub> (668 cm<sup>-1</sup>; bending vibration).

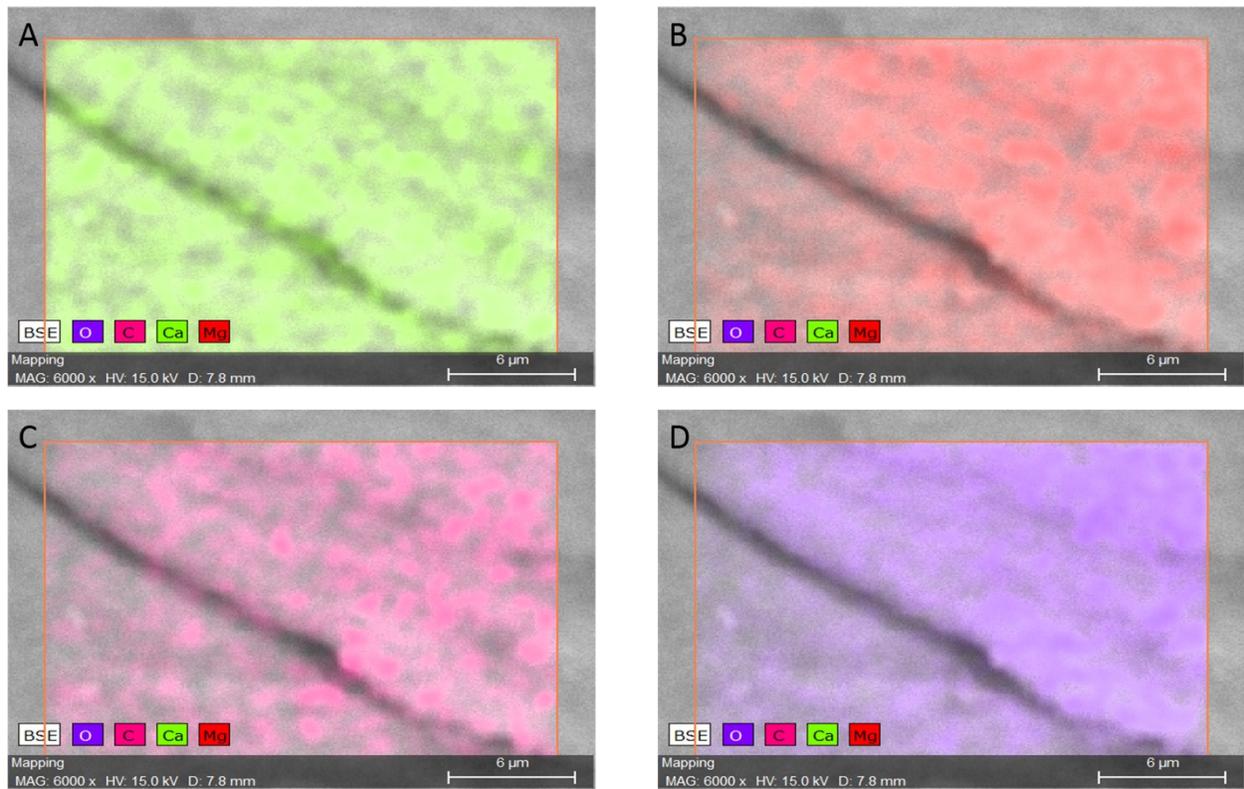
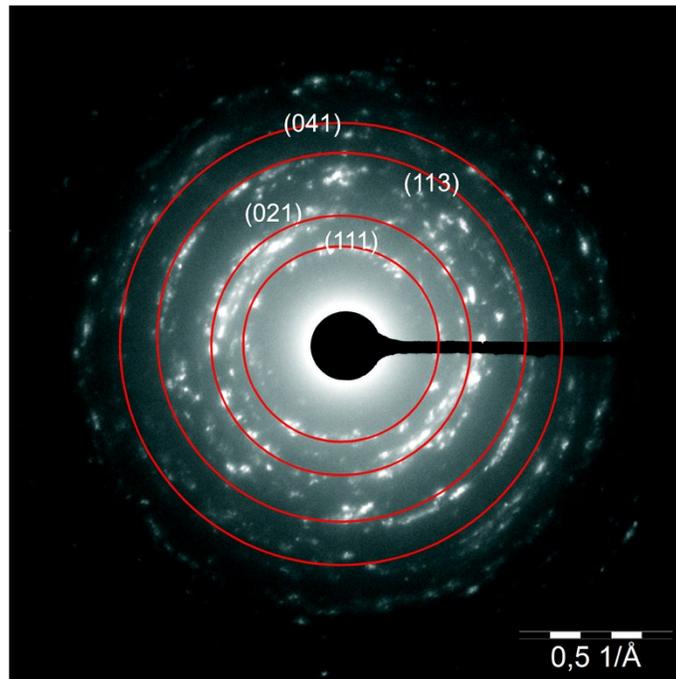


Figure S4: EDS mapping of the different Elements



*Figure S5: Indexed ED pattern of the crystalline part. The reflexes correspond to aragonite*