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Supporting information

EXPERIMENTAL

Materials

B^{III}SubPcCl (98%) was purchased from Aldrich. C_{60} of 99.9 % purity was received from MTR Ltd. Sodium fluorenone ketyl was obtained as described.¹ Solvents were purified in argon atmosphere. *o*-Dichlorobenzene ($C_6H_4Cl_2$) was distilled over CaH₂ under reduced pressure; hexane was distilled over Na/benzophenone. The solvents were degassed and stored in the glove box. All operations on the synthesis of **1** and the storage of crystals were carried out in a MBraun 150B-G glove box with controlled atmosphere and water and oxygen content less than 1 ppm. KBr pellets for IR- and UV-visible-NIR measurements were prepared in the glove box

General

UV-visible-NIR spectra were measured in KBr pellets on a Perkin Elmer Lambda 1050 spectrometer in the 250-2500 nm range. FT-IR spectra were obtained in KBr pellets with a Perkin-Elmer Spectrum 400 spectrometer (400-7800 cm⁻¹). EPR spectra were recorded for sealed polycrystalline samples of **1** at 295 K with a JEOL JES-TE 200 X-band ESR spectrometer.

X-ray crystal structure determination

Crystal data of **1** at 100(2) K: C₁₄₆H₄₈B₂Cl₄N₁₂O₂, M_r = 2165.38 g mol⁻¹, black plate, monoclinic, $P 2_1$, a = 15.4619(8), b = 16.9321(8), c = 19.6980(10) Å, $\beta = 110.386(5)^\circ$, V = 4834.0(4) Å³, Z = 2, $d_{calc} = 1.488$ g·cm⁻³, $\mu = 0.314$ mm⁻¹, F(000) = 2204, $2\theta_{max} = 69.50^\circ$, reflections measured 75952, unique reflections 21328, reflections with $I > 2\sigma(I) = 16669$, parameters refined 1708, restraints 2395, $R_1 = 0.0808$, $wR_2 = 0.2045$, G.O.F. = 1.005, CCDC 1046031.

The intensity data for 1 were collected on a MAR225 CCD detector using synchrotron radiation at the BESSY storage ring, BL 14.2 (λ =0.9050 Å, PSF of the Free University of Berlin, Germany). The structures were solved by direct method and refined by the full-matrix

least-squares method against F^2 using SHELX-97 package.² Non-hydrogen atoms were anisotropically refined. Positions of hydrogen atoms were included into refinement in a riding mode. Seefor crystallographic data in CIF. Structure of **1** contains two orientations of the C_{60} molecule having 0.537(5(/0.463(5) occupancies. One of two solvent $C_6H_4Cl_2$ molecules disordered between two orientations with the 0.660(6)/0.340(6) occupancies. To keep geometry of disordered C_{60} and one $C_6H_4Cl_2$ molecules close to ideal one restraints were applied for the refinement of the crystal structures of **1**.

IR- spectra.

	B ^{III} SubPcCl	C ₆₀	(Na ⁺)(Fluorenone ^{•-})	$C_6H_4Cl_2$	1
B ^{III} SubPc	444w				437w*
	631w				596w
	697w				696w
	750vs				737vs*
	758m				760m*
	797w				805w
	880w				881w
	951m				943w
	960m				-
	1088w				1084s*
	1130s				1130s*
	1150s				1154w*
	1196w				1196w
	1231m				1229w
	1281s				1289w
	1321w				1326w
	1386m				1388w*
	1432s				1431s*
	1440s				-
	1453s				1457s*
	1491w				1496w*
	2853w				2850w
	2925w				2920w
	3058w				3058w*
C ₆₀		526s			527m
- 00		576m			575w
		1182m			1181w
		1429s			1431s*
Fluorenone ⁻			439w		437w*
ridorenone			671w		673w
			723vs		075W
			735s		737vs*
			755w		
					760m*
			922s		-
			989m		-
			1080m		1084s*
			1151w		1154w*
			1197w		1196w*
			1392m		1388w*
			1400m		-
			1439m		1431s*
			1453s		1457s*
			1492m		1496w*
			1594s		
			1651m		1630m
			1675m		
			1714m		
			3058w		3058w*
C ₆ H ₄ Cl ₂				657w	657w
				748s	737vs*
				1030m	1034w
				1122m	1130s*
				1453m	1457s*
			I	1733111	173/5

 Table S1. IR-spectra (cm⁻¹ in KBr) of starting compounds and complex 1.

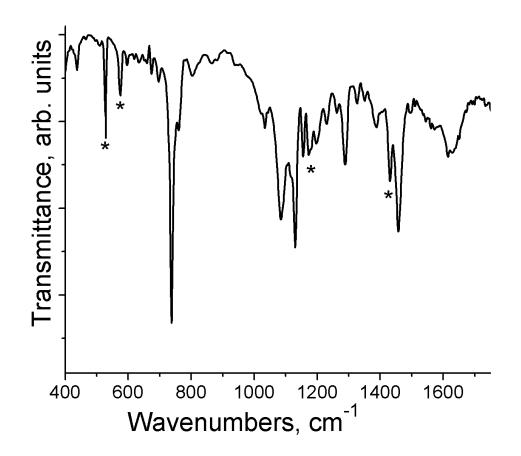


Figure S1. IR-spectrum of complex 1 in KBr pellet prepared in anaerobic conditions. Absorption bands of C_{60} are marked by asterisks.

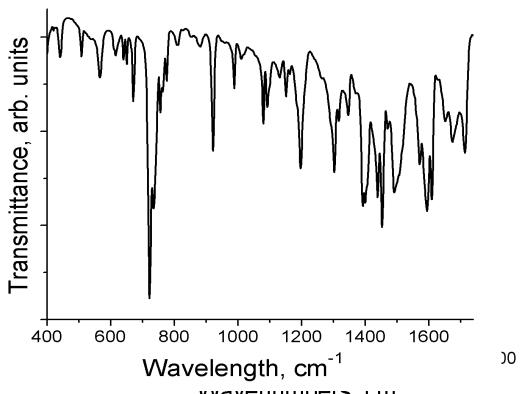


Figure S3. IR-spectrum of starting B^{III}SubPcCl in KBr pellet prepared in anaerobic **Figure S2.** IR-spectrum of starting B^{III}SubPcCl in KBr pellet.

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