

## Two Alternative Approaches to Access Mixed Hydride-Amido Zinc Complexes: Synthetic, Structural and Solution Implications

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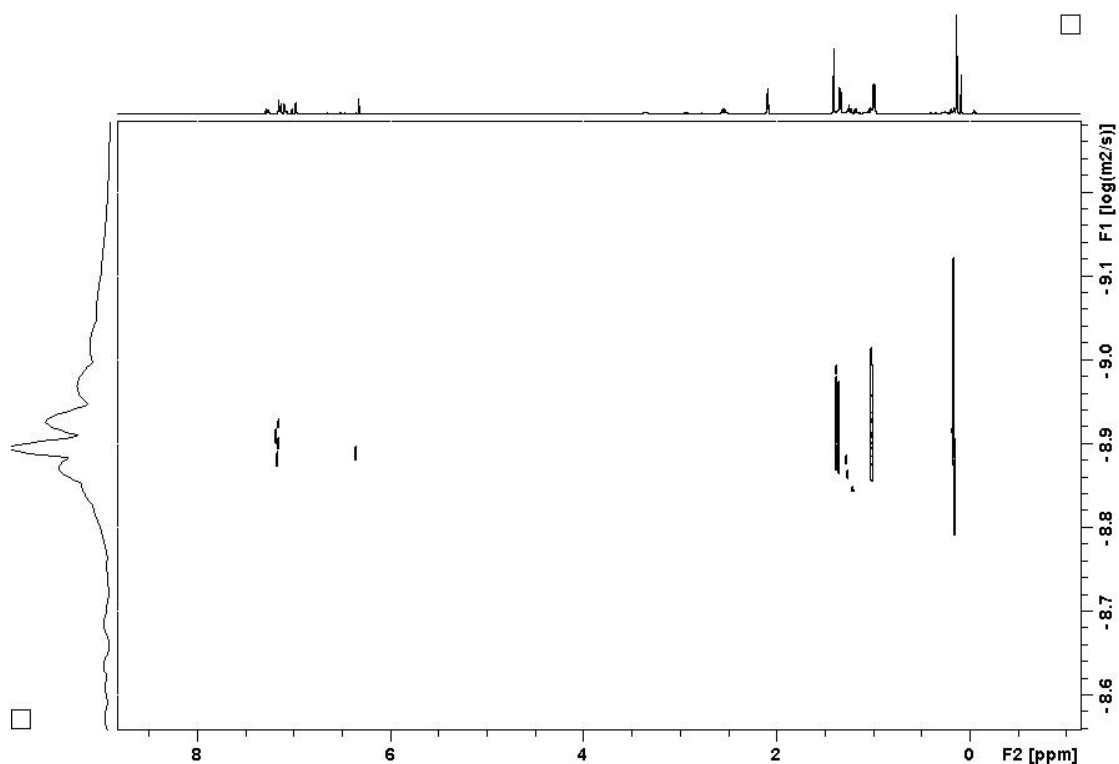
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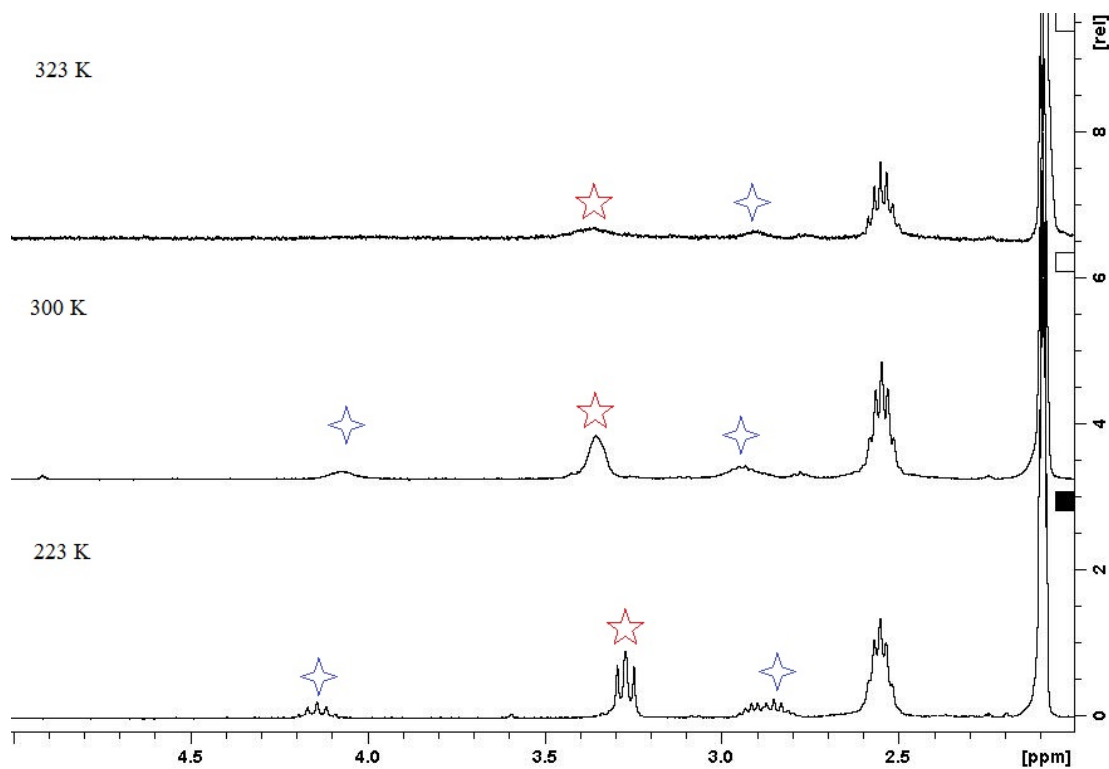
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**Figure S13** Aromatic region of <sup>1</sup>H NMR spectra of isolated crystals of **3** in d<sub>8</sub>-toluene highlighting the relevant resonances of the imidazole backbone of **3** (blue), IPr (red) and ZnH<sub>2</sub>·IPr species (green) at 220 K, 250K, 280K and 310 K.



**Figure S1**  $^1\text{H}$  DOSY NMR spectrum of complex **1** in  $\text{C}_7\text{D}_8$  solution.



**Figure S2** Region of  $^1\text{H}$  NMR spectra of **1** in  $d_8$ -toluene containing the hydride resonances at 323 K (top); 300K (middle) and 223 K (bottom).

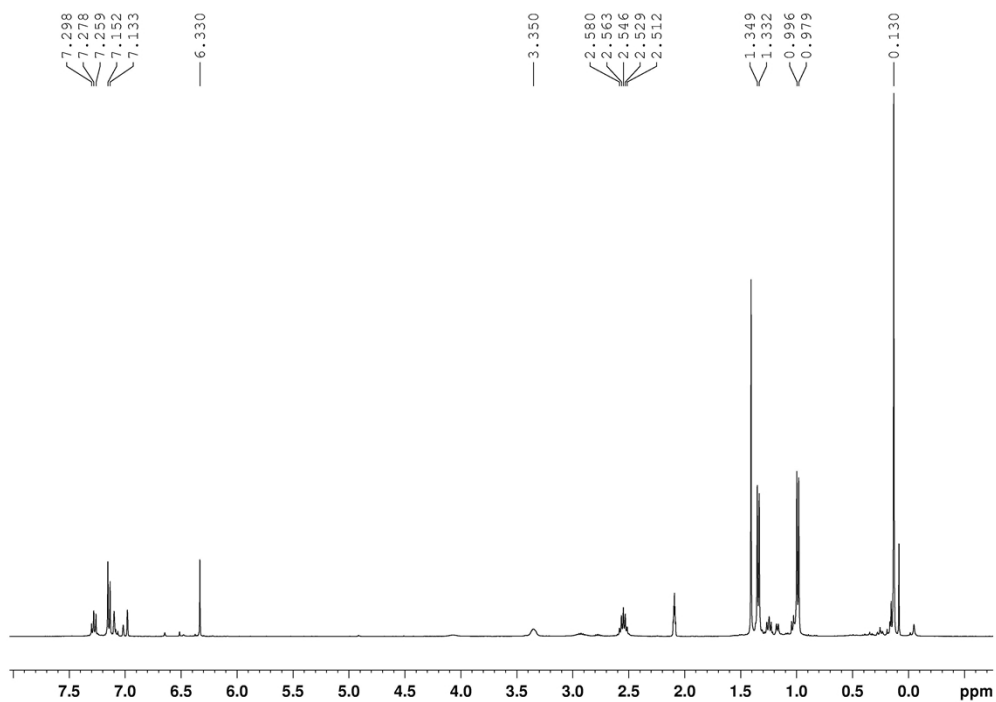


Figure S3  $^1\text{H}$  NMR spectrum of complex **1** in  $\text{C}_7\text{D}_8$  solution.

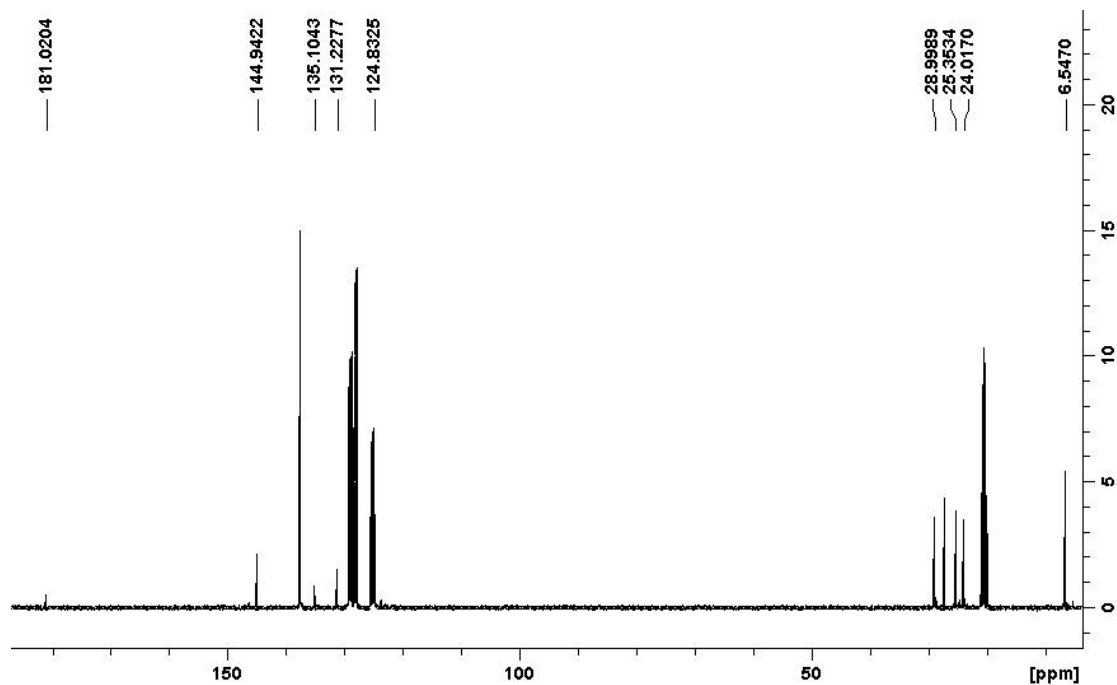
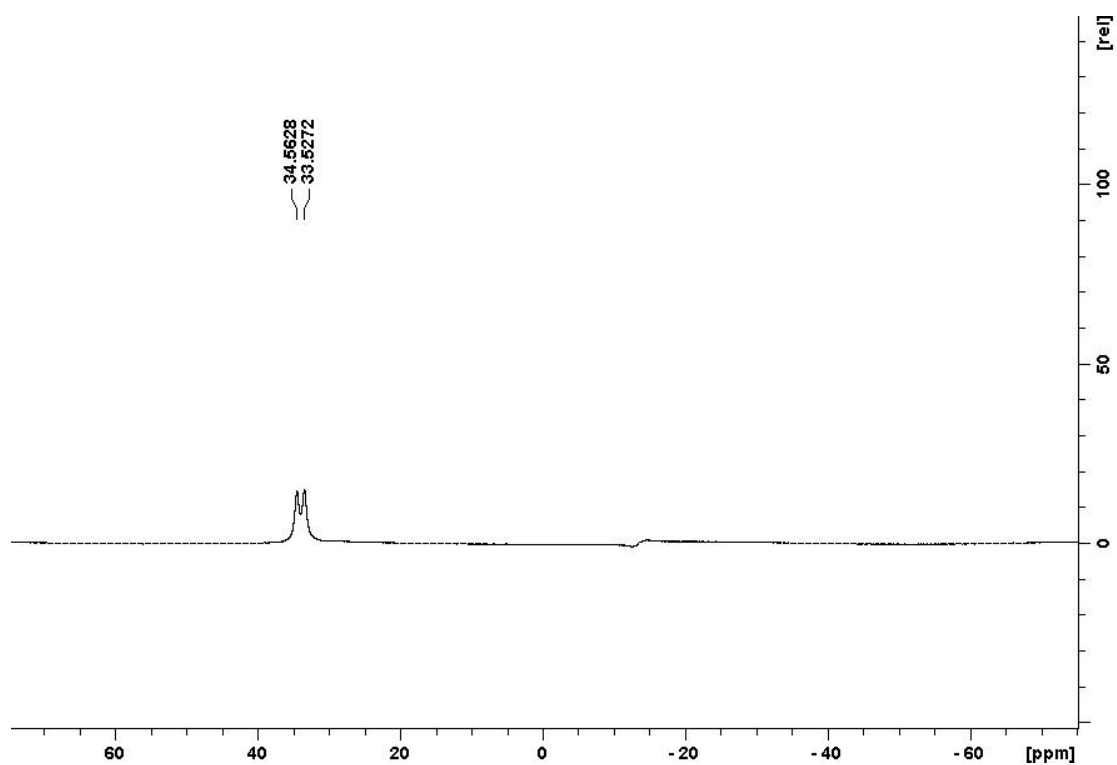
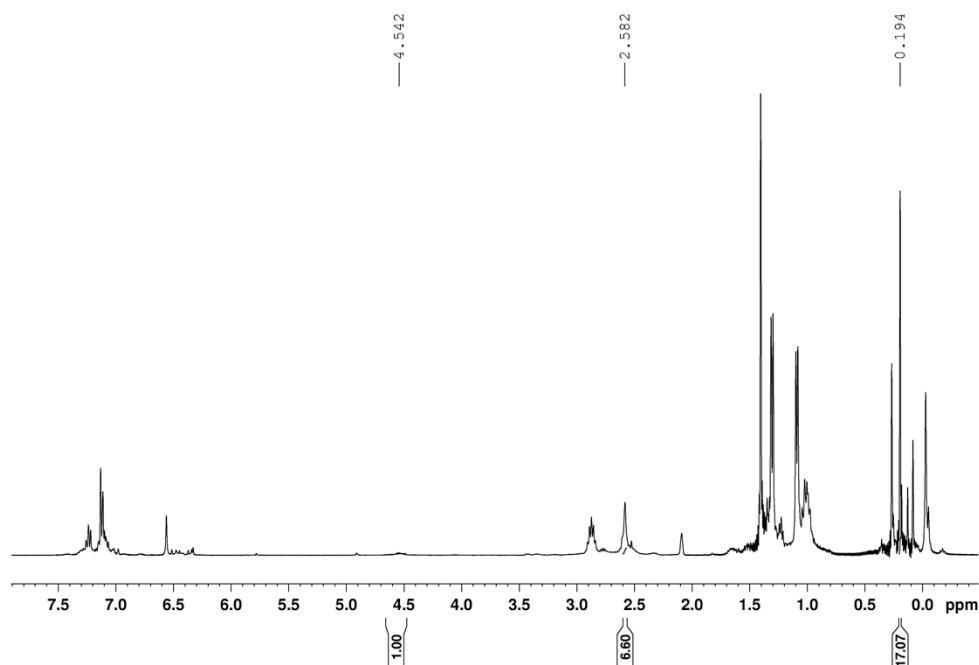


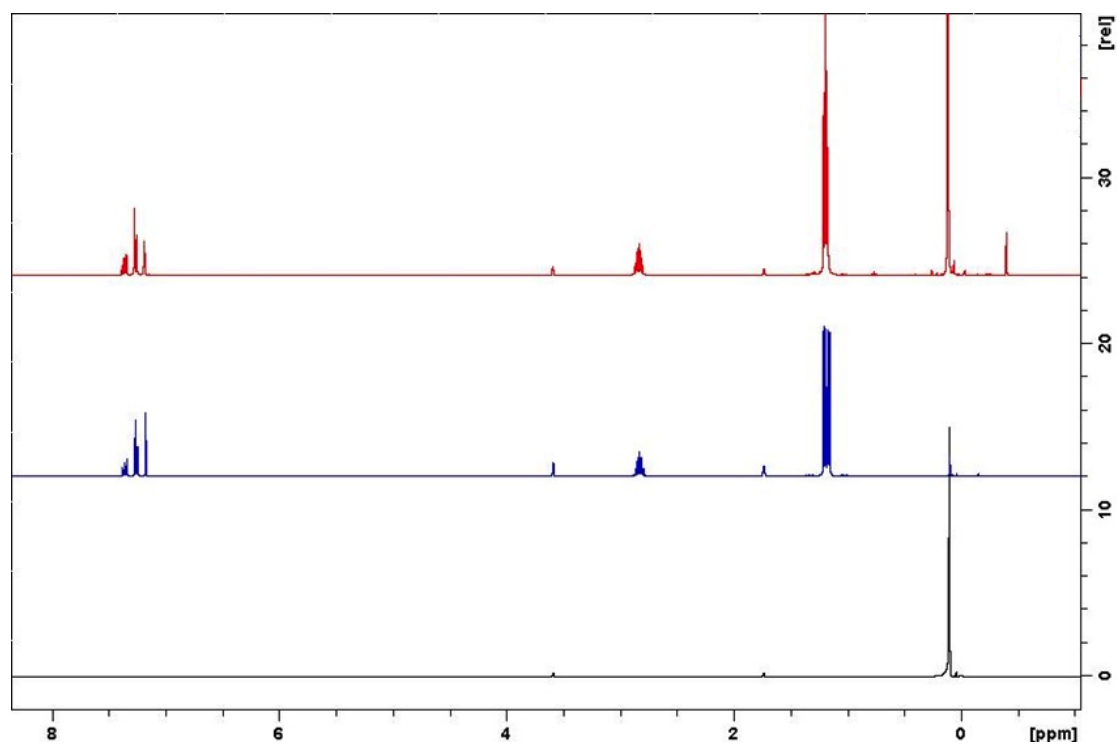
Figure S4  $^{13}\text{C}$  NMR spectrum of complex **1** in  $\text{C}_7\text{D}_8$  solution.



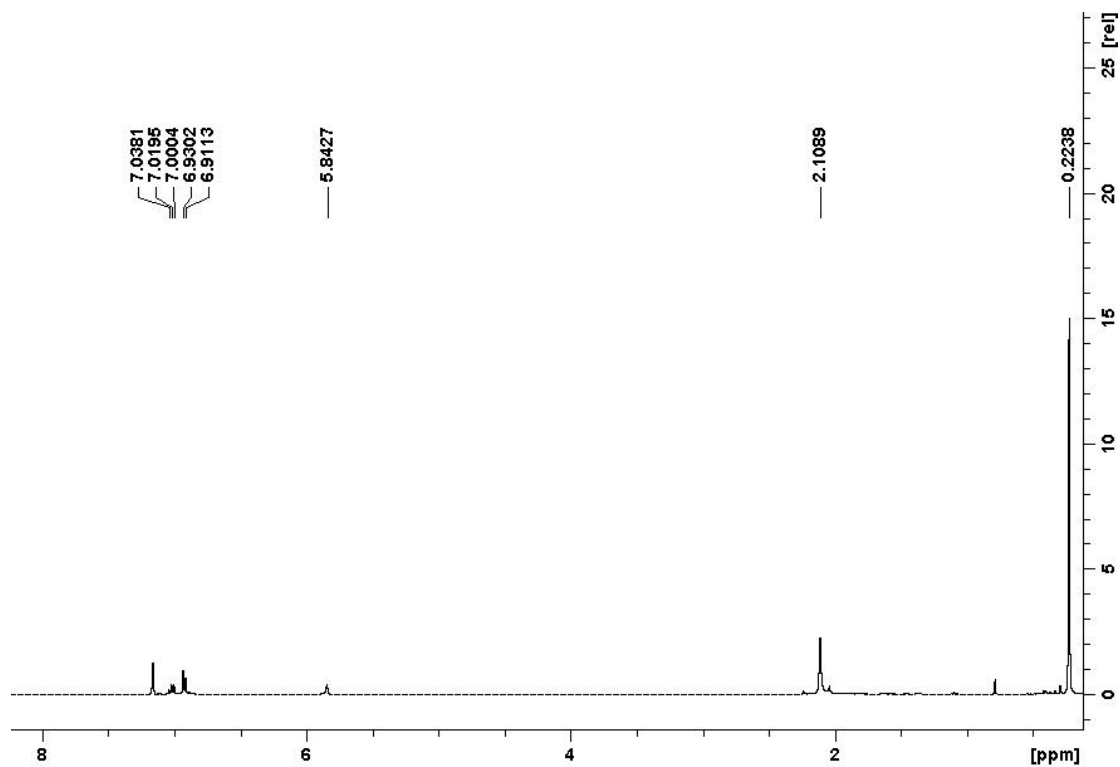
**Figure S5**  $^{11}\text{B}$  NMR spectrum of filtrate from reaction of DMAB,  $\text{Zn}(\text{HMDS})_2$  and IPr.



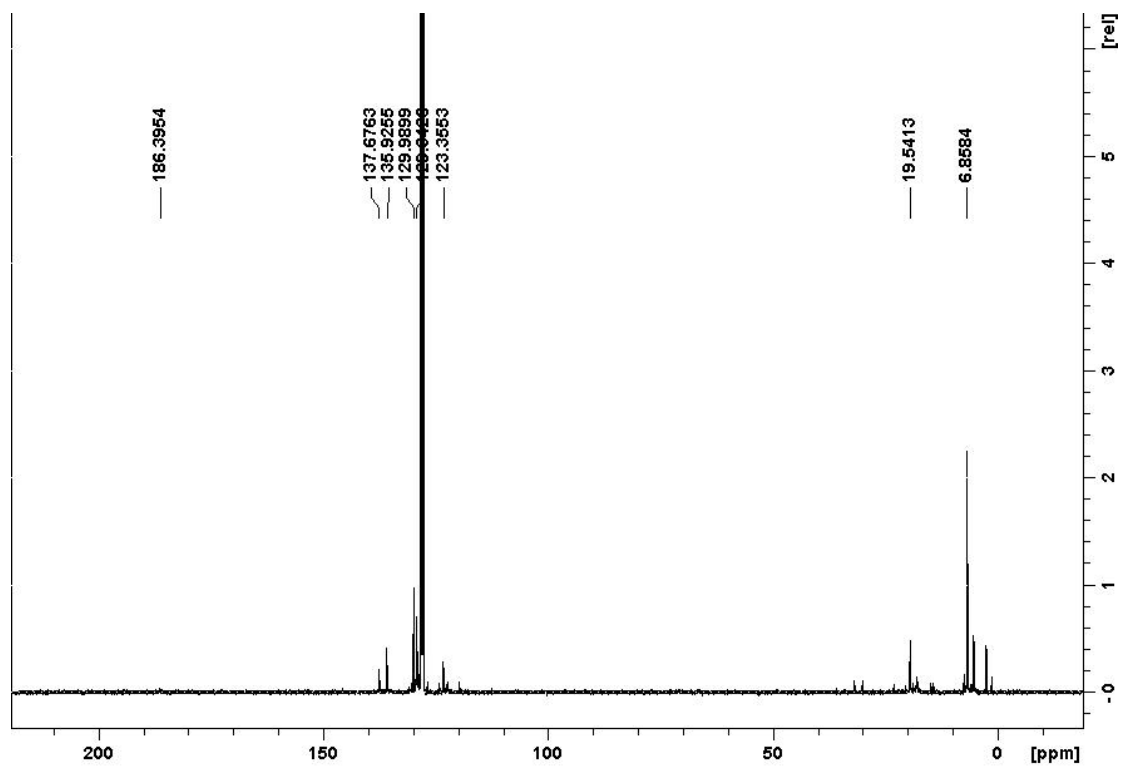
**Figure S6**  $^1\text{H}$  NMR spectrum of filtrate from reaction of DMAB,  $\text{Zn}(\text{HMDS})_2$  and IPr with resonances representing  $[\text{HB}(\text{NMe}_2)(\text{HMDS})]$  highlighted.



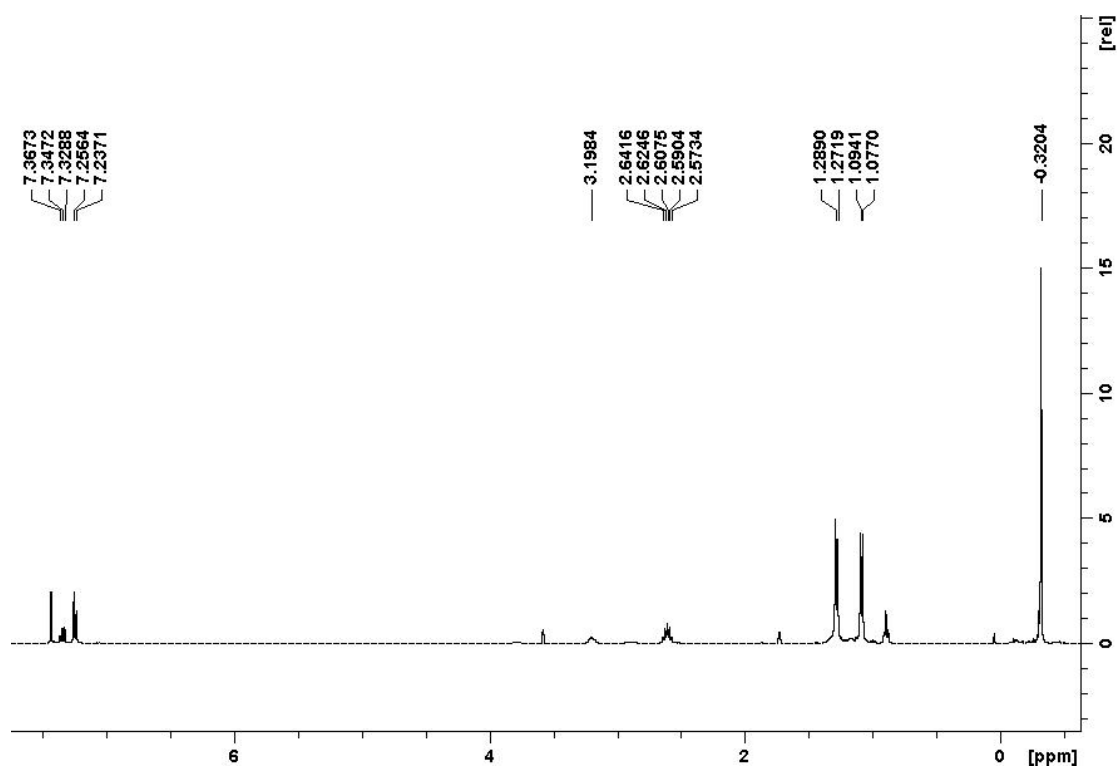
**Figure S7**  $^1\text{H}$  NMR spectra of  $\text{Zn}(\text{HMDS})_2$  (bottom), IPr (middle), and an equimolar amount of  $\text{Zn}(\text{HMDS})_2$  and IPr (top) confirming no interaction has occurred, in  $d_8$ -THF.



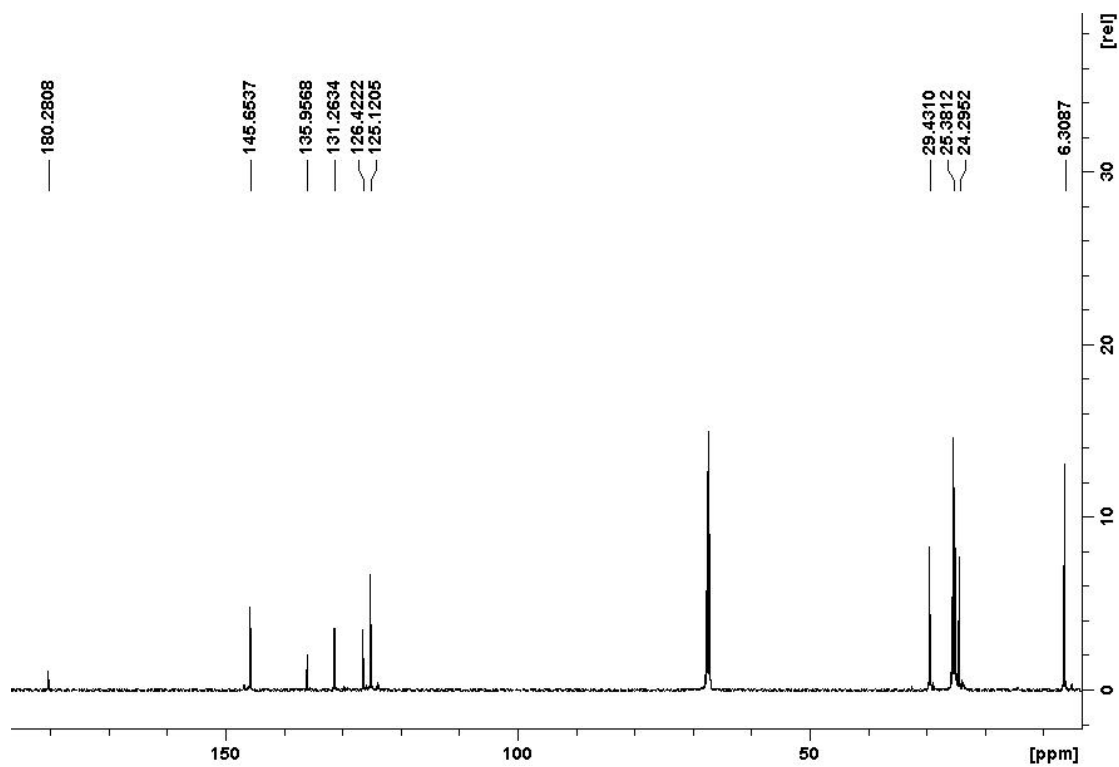
**Figure S8**  $^1\text{H}$  NMR spectrum of complex **2** in  $\text{C}_6\text{D}_6$  solution.



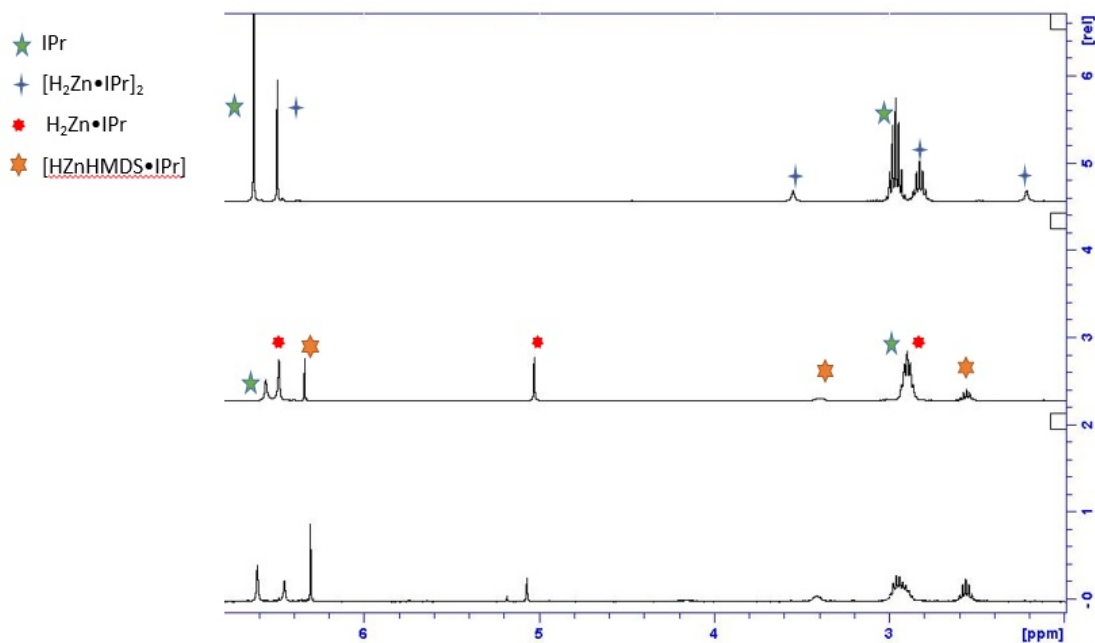
**Figure S9**  $^{13}\text{C}$  NMR spectrum of complex **2** in  $\text{C}_6\text{D}_6$  solution.



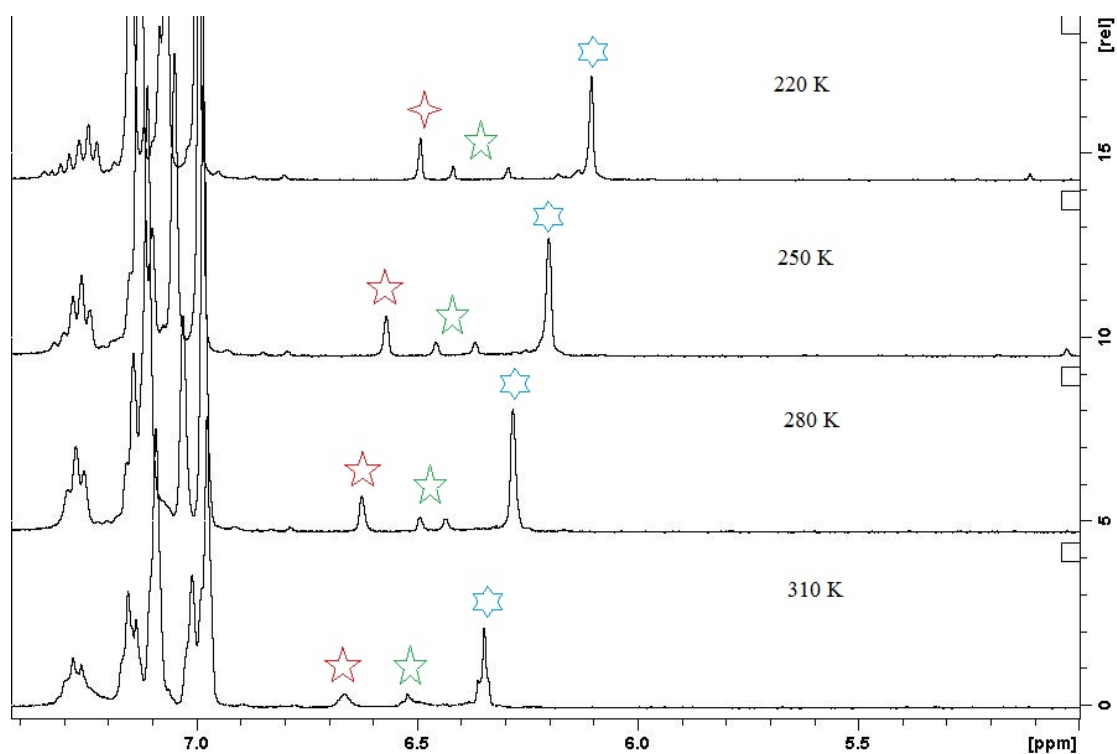
**Figure S10**  $^1\text{H}$  NMR spectrum of monomeric  $(\text{HMDS})\text{ZnH}\cdot\text{IPr}$  (**3**) in  $d_8\text{-THF}$



**Figure S11**  $^{13}\text{C}$  NMR spectrum of monomeric  $(\text{HMDS})\text{ZnH}\cdot\text{IPr}$  (**3**) in  $d_8$ -THF



**Figure S12**  $^1\text{H}$  NMR spectra in  $\text{C}_6\text{D}_6$  of (a, top) an authentic sample of  $[\text{ZnH}_2\cdot\text{IPr}]_2$  (B) with two molar equivalents of IPr added; (b, middle) the same sample with two molar equivalents of  $\text{Zn}(\text{HMDS})_2$  added; (c, bottom) an authentic sample of  $(\text{HMDS})\text{ZnH}\cdot\text{IPr}$  (**3**) for comparison



**Figure S13** Aromatic region of <sup>1</sup>H NMR spectra of isolated crystals of **3** in d<sub>8</sub>-toluene highlighting the relevant resonances of the imidazole backbone of **3** (blue), IPr (red) and ZnH<sub>2</sub>.IPr species (green)<sup>1</sup> at 220 K, 250K, 280K and 310 K.

<sup>1</sup> In this deuterated solvent monomeric and dimeric forms of ZnH<sub>2</sub>IPr appear also to coexist in solution.