

## Supplementary Data to

# Polyfluorinated carba-*closo*-dodecaboranes with amino and ammonio substituents bonded to boron

*Szymon Z. Konieczka,<sup>a</sup> Michael Drisch,<sup>a</sup> Katharina Edkins,<sup>b</sup> Michael Hailmann<sup>a</sup>  
and Maik Finze<sup>\*a</sup>*

<sup>a</sup> Institut für Anorganische Chemie, Julius-Maximilians-Universität Würzburg, Am Hubland, 97074  
Würzburg, Germany

<sup>b</sup> School of Medicine, Pharmacy and Health, Durham University, University Boulevard,  
Stockton-on-Tees, TS17 6BH, UK

**Table S1** Crystal data and structure refinement parameters for 1-H-12-H<sub>3</sub>N-*closo*-1-CB<sub>11</sub>F<sub>10</sub>·H<sub>2</sub>O (**H1**·H<sub>2</sub>O), 1-H-7-H<sub>3</sub>N-*closo*-1-CB<sub>11</sub>F<sub>10</sub>·diglyme (**H2**·diglyme), 1-H-2-H<sub>3</sub>N-*closo*-1-CB<sub>11</sub>F<sub>10</sub>·0.5H<sub>2</sub>O (**H3**·0.5H<sub>2</sub>O), [Et<sub>4</sub>N][1-H-12-H<sub>2</sub>N-*closo*-1-CB<sub>11</sub>F<sub>10</sub>] ([Et<sub>4</sub>N]**1**) and 7-H<sub>3</sub>N-12-F-*closo*-1-CB<sub>11</sub>H<sub>10</sub>·(CH<sub>3</sub>)<sub>2</sub>CO (**H4**·(CH<sub>3</sub>)<sub>2</sub>CO).

Compound	H1·H <sub>2</sub> O	H2·diglyme	H3·0.5H <sub>2</sub> O	[Et <sub>4</sub> N] <b>1</b>	H4·(CH <sub>3</sub> ) <sub>2</sub> CO
Formula	CH <sub>6</sub> B <sub>11</sub> F <sub>10</sub> NO	C <sub>7</sub> H <sub>18</sub> B <sub>11</sub> F <sub>10</sub> NO <sub>3</sub>	CH <sub>5</sub> B <sub>11</sub> F <sub>10</sub> NO <sub>0.5</sub>	C <sub>9</sub> H <sub>23</sub> B <sub>11</sub> F <sub>10</sub> N <sub>2</sub>	C <sub>4</sub> H <sub>19</sub> B <sub>11</sub> FNO
Formula weight	356.98	473.13	348.11	468.20	235.11
<i>T</i> / K	100	100	100	100	100
Crystal system	orthorhombic	monoclinic	orthorhombic	monoclinic	monoclinic
Space group	<i>Iba2</i>	<i>P2<sub>1</sub>/n</i>	<i>Ccc2</i>	<i>P2<sub>1</sub>/n</i>	<i>P2<sub>1</sub>/c</i>
<i>a</i> / Å	12.356(4)	8.647(5)	14.684(4)	13.3754(10)	9.8577(8)
<i>b</i> / Å	20.211(6)	14.863(7)	22.575(7)	11.0807(8)	10.3265(8)
<i>c</i> / Å	9.735(3)	15.435(8)	14.269(4)	13.7559(10)	13.5715(11)
$\beta$ / deg		94.148(18)		102.726(3)	96.499(2)
<i>V</i> / Å <sup>3</sup>	2431.2(12)	1978.5(17)	4730(2)	1988.7(3)	1372.64(19)
<i>Z</i>	8	4	16	4	4
$\rho_{\text{calc.}}$ / g cm <sup>-3</sup>	1.951	1.588	1.9543	1.564	1.138
$\mu$ / mm <sup>-1</sup>	0.216	0.161	0.216	0.150	0.068
<i>F</i> (000) / e	1376	944	2674.9	944	488
Refl. collected/unique	19274/2384	10659/3626	31184/4659	25500/4066	14304/2726
Refl. observed [ <i>I</i> > 2 $\sigma$ ( <i>I</i> )]	1862	2056	4010	3225	2385
<i>R</i> <sub>int</sub>	0.0957	0.0801	0.0639	0.0492	0.0351
Restraints/parameters	4/227	0/294	0/426	0/300	0/217
$\theta$ range / deg	1.93–26.00	1.91–25.50	1.65–26.00	1.92–26.37	2.08–26.09
<i>R</i> 1 [ <i>I</i> > 2 $\sigma$ ( <i>I</i> )] <sup>a</sup>	0.0414	0.0740	0.0362	0.0375	0.0405
<i>wR</i> 2 (all data) <sup>b</sup>	0.0869	0.1797	0.0957	0.0969	0.1249
GOF on <i>F</i> <sup>2c</sup>	1.309	1.175	1.0782	1.075	1.073
Larg. diff. peak/hole / e Å <sup>-3</sup>	0.309/−0.413	0.492/−0.475	0.341/−0.243	0.263/−0.229	0.322/−0.197

<sup>a</sup>  $R1 = (\sum||F_o| - |F_c||) / \sum|F_o|$ . <sup>b</sup>  $wR2 = [\sum w(F_o^2 - F_c^2)^2 / \sum w(F_o^2)^2]^{0.5}$ , weight scheme:  $w = [\sigma^2 F_o + (aP)^2 + bP]^{-1}$ ;  $P = [\max(0, F_o^2) + 2F_c^2] / 3$ ; H1·H<sub>2</sub>O:  $a = 0.0031$ ,  $b = 4.6387$ ; H2·diglyme:  $a = 0.0169$ ,  $b = 3.6200$ ; H3·0.5H<sub>2</sub>O:  $a = 0.0528$ ,  $b = 0.5304$ ; [Et<sub>4</sub>N]**1**:  $a = 0.0455$ ,  $b = 0.4087$ ; H4·(CH<sub>3</sub>)<sub>2</sub>CO:  $a = 0.0815$ ,  $b = 0.2877$ ; <sup>c</sup> GOF:  $S = [\sum w(F_o^2 - F_c^2)^2 / (m - n)]^{0.5}$ ; ( $m$  = reflections,  $n$  = variables).

**Table S2** Energies, enthalpies and free energies as well as free energies of solvation<sup>a</sup> calculated at the B3LYP/6-311++G(d,p) level of theory.

Compound	$E^{ZPC}$ [au]	$H$ [au]	$G$ [au]	$G_s(\text{H}_2\text{O})$ [kJ mol <sup>-1</sup> ] gas phase model	$G_s(\text{H}_2\text{O})$ [kJ mol <sup>-1</sup> ] solute model <sup>b</sup>
[1-H-12-H <sub>2</sub> N- <i>closo</i> -1-CB <sub>11</sub> F <sub>10</sub> ] <sup>-</sup> ( <b>1</b> )	-1367.570024	-1367.548691	-1367.615252	-204.1	-207.0
1-H-12-H <sub>3</sub> N- <i>closo</i> -1-CB <sub>11</sub> F <sub>10</sub> ( <b>H1</b> )	-1368.032236	-1368.010643	-1368.079101	-135.4	-148.6
[1-H-7-H <sub>2</sub> N- <i>closo</i> -1-CB <sub>11</sub> F <sub>10</sub> ] <sup>-</sup> ( <b>2</b> )	-1367.570586	-1367.549108	-1367.618334	-204.4	-209.3
1-H-7-H <sub>3</sub> N- <i>closo</i> -1-CB <sub>11</sub> F <sub>10</sub> ( <b>H2</b> )	-1368.027809	-1368.006221	-1368.074760	-148.3	-169.9
[1-H-2-H <sub>2</sub> N- <i>closo</i> -1-CB <sub>11</sub> F <sub>10</sub> ] <sup>-</sup> ( <b>3</b> )	-1367.572853	-1367.551667	-1367.617578	-206.3	-211.8
1-H-2-H <sub>3</sub> N- <i>closo</i> -1-CB <sub>11</sub> F <sub>10</sub> ( <b>H3</b> )	-1368.014879	-1367.993338	-1368.060589	-179.3	-206.9

<sup>a</sup>  $G_s$  calculated using the conductor polarizable continuum model (CPCM). <sup>b</sup> The geometries were reoptimized within the CPCM.

**Table S3** Experimental<sup>a</sup> and calculated<sup>b</sup> chemical shifts of **1**, **H1**, **2**, **H2**, **3** and **H3**.

Compound	<sup>1</sup> H		<sup>13</sup> C	<sup>11</sup> B		<sup>19</sup> F										
	CH <sub>cluster</sub>	NH <sub>n</sub>	C <sub>cluster</sub>	B2–B6	B7–B11	B12	F2–F6	F7–F11								
[1-H-12-H <sub>2</sub> N- <i>closo</i> -1-CB <sub>11</sub> F <sub>10</sub> ] <sup>-</sup> ( <b>1</b> )	3.36	≈ 1	15.9	-17.6	-15.3	-15.3	-256.1	-254.8								
calculated	1.80	-0.55	21.9	-19.3	-16.6	-16.1	-289.5	-288.1								
1-H-12-H <sub>3</sub> N- <i>closo</i> -1-CB <sub>11</sub> F <sub>10</sub> ( <b>H1</b> )	4.14	6.49	22.5	-17.6	-16.0	-23.5	-250.8	-255.6								
calculated	2.70	3.60	31.6	-19.4	-18.5	-26.8	-275.5	-289.4								
	CH <sub>cluster</sub>	NH <sub>n</sub>	C <sub>cluster</sub>	B2+B3	B4+B6	B5	B7	B8+B11	B9+B10	B12	F2+F3	F4+F6	F5	F8+F11	F9+F10	F12
[1-H-7-H <sub>2</sub> N- <i>closo</i> -1-CB <sub>11</sub> F <sub>10</sub> ] <sup>-</sup> ( <b>2</b> )	3.27	n. o. <sup>c</sup>	18.0	-17.7	-17.4	-18.4	-22.5	-15.3	-16.2	-7.9	-257.3	-256.0	-257.4	-256.9	-256.3	-251.9
calculated	1.78	-0.47	23.8	-18.6	-18.9	-19.7	-24.3	-16.3	-17.1	-8.8	-291.5	-289.1	-290.1	-290.1	-287.8	-282.6
1-H-7-H <sub>3</sub> N- <i>closo</i> -1-CB <sub>11</sub> F <sub>10</sub> ( <b>H2</b> )	3.88	6.61	19.6	-16.9	-16.9	-15.5	-31.4	-15.5	-15.5	-8.1	n.a.p. <sup>d</sup>	n.a.p.	-247.9	n.a.p.	n.a.p.	-251.1
calculated	2.48	3.61	25.4	-19.7	-18.4	-14.9	-33.7	-17.6	-16.8	-8.9	-292.3	-279.1	-267.7	-289.9	-275.3	-282.3
	CH <sub>cluster</sub>	NH <sub>n</sub>	C <sub>cluster</sub>	B2	B3+B6	B4+B5	B7+B11	B8+B10	B9	B12	F3+F6	F4+F5	F7+F11	F8+F10	F9	F12
[1-H-2-H <sub>2</sub> N- <i>closo</i> -1-CB <sub>11</sub> F <sub>10</sub> ] <sup>-</sup> ( <b>3</b> )	3.48	n. o.	18.7	-25.4	-17.8	-17.6	-15.2	-16.2	-15.6	-7.4	-257.4	-255.5	-256.0	-255.6	-255.0	-248.8
calculated	1.66	0.00	26.7	-26.3	-20.0	-18.9	-16.0	-17.6	-16.6	-8.0	-294.1	-289.5	-289.4	-287.9	-287.8	-277.4
1-H-2-H <sub>3</sub> N- <i>closo</i> -1-CB <sub>11</sub> F <sub>10</sub> ( <b>H3</b> )	3.78	6.95	14.3	-32.0	-18.0	-17.1	-15.2	-15.9	-13.1	-6.9	-257.1	-252.9	-256.0	-253.1	-248.0	-245.0
calculated	1.89	3.90	17.4	-34.5	-20.9	-18.1	-17.0	-16.9	-11.5	-6.9	-294.3	-278.2	-289.9	-275.9	-261.7	-260.7

<sup>a</sup> Solvent: CD<sub>3</sub>CN. <sup>b</sup> B3LYP/6-311++G(2d,p)/B3LYP/6-311++G(d,p). <sup>c</sup> n.o. = not observed. <sup>d</sup> n.a.p. = no assignment possible.

**Table S4** Experimental and calculated<sup>a</sup> atomic distances<sup>b</sup> of 1-H-7-H<sub>3</sub>N-*closo*-1-CB<sub>11</sub>F<sub>10</sub> (**H2**), [1-H-7-H<sub>2</sub>N-*closo*-1-CB<sub>11</sub>F<sub>10</sub>]<sup>−</sup> (**2**), 7-H<sub>3</sub>N-12-F-*closo*-1-CB<sub>11</sub>H<sub>10</sub> (**H4**), [7-H<sub>2</sub>N-12-F-*closo*-1-CB<sub>11</sub>H<sub>10</sub>]<sup>−</sup> (**4**) and related {*closo*-1-CB<sub>11</sub>} clusters.<sup>c</sup>

Compound/anion	<b>2</b> calc.	H2·diglyme exp.	H2 calc.	<b>4</b> calc.	H4·H <sub>2</sub> O exp.	H4 calc.	[7-H <sub>2</sub> N- <i>closo</i> -1-CB <sub>11</sub> H <sub>11</sub> ] <sup>−</sup> calc.	7-H <sub>3</sub> N- <i>closo</i> -1-CB <sub>11</sub> H <sub>11</sub> calc.
Ref.	<sup>d</sup>	<sup>d</sup>	<sup>d</sup>	1	<sup>d</sup>	1	1	1
<i>d</i> (B–N)	1.472	1.529(6)	1.562	1.493	1.546(2)	1.579	1.496	1.586
<i>d</i> (C–B2/B3)	1.718	1.717(8)	1.704	1.706	1.704(2)	1.698	1.702	1.697
<i>d</i> (C–B4/B6)	1.716	1.716(8)	1.736	1.706	1.715(2)	1.716	1.710	1.717
<i>d</i> (C–B5)	1.717	1.726(7)	1.728	1.704	1.714(2)	1.714	1.711	1.709
<i>d</i> (B2–B3)	1.788	1.803(8)	1.815	1.772	1.789(2)	1.797	1.786	1.796
<i>d</i> (B2–B6/B3–B4)	1.800	1.789(8)	1.792	1.779	1.776(2)	1.778	1.784	1.775
<i>d</i> (B5–B4/B6)	1.796	1.796(9)	1.808	1.784	1.785(2)	1.781	1.777	1.781
<i>d</i> (B2/B3–B7)	1.782	1.772(7)	1.752	1.788	1.760(2)	1.748	1.787	1.751
<i>d</i> (B2–B11/B3–B8)	1.772	1.778(7)	1.782	1.774	1.777(2)	1.779	1.768	1.782
<i>d</i> (B6–B11/B4–B8)	1.777	1.789(8)	1.775	1.768	1.772(2)	1.763	1.768	1.770
<i>d</i> (B6–B10/B4–B9)	1.779	1.788(8)	1.796	1.773	1.771(2)	1.780	1.769	1.780
<i>d</i> (B5–B9/B10)	1.779	1.774(8)	1.788	1.771	1.768(2)	1.775	1.776	1.777
<i>d</i> (B7–B8/B11)	1.823	1.798(8)	1.782	1.798	1.779(2)	1.761	1.814	1.759
<i>d</i> (B8–B9/B10–B11)	1.815	1.818(8)	1.814	1.794	1.787(2)	1.788	1.781	1.783
<i>d</i> (B9–B10)	1.817	1.794(9)	1.833	1.789	1.793(2)	1.800	1.792	1.796
<i>d</i> (B7–B12)	1.810	1.797(8)	1.775	1.813	1.772(2)	1.764	1.788	1.755
<i>d</i> (B8/B11–B12)	1.791	1.792(9)	1.808	1.783	1.787(2)	1.797	1.785	1.795
<i>d</i> (B9/B10–B12)	1.801	1.790(9)	1.794	1.780	1.779(2)	1.775	1.790	1.781
<i>d</i> (B2–F2/B3–F3)	1.367	1.361(6)	1.360	–	–	–	–	–
<i>d</i> (B4–F4/B6–F6)	1.366	1.359(7)	1.351	–	–	–	–	–
<i>d</i> (B5–F5)	1.366	1.363(7)	1.351	–	–	–	–	–
<i>d</i> (B8–F8/B11–F11)	1.376	1.365(7)	1.371	–	–	–	–	–
<i>d</i> (B9–F9/B10–F10)	1.373	1.377(6)	1.358	–	–	–	–	–
<i>d</i> (B12–F12)	1.378	1.367(7)	1.372	1.395	1.3865(15)	1.392	–	–

<sup>a</sup> B3LYP/6-311++G(d,p). <sup>b</sup> All *d*(CB), *d*(BB) and *d*(BF) distances are mean values for *C*<sub>5v</sub> symmetry of the carborane cage; *d*(NH) are mean values. <sup>c</sup> Atomic distances in Å. <sup>d</sup> This work.

**Table S5** Experimental and calculated<sup>a</sup> atomic distances<sup>b</sup> of 1-H-2-H<sub>3</sub>N-*closo*-1-CB<sub>11</sub>F<sub>10</sub> (**H3**), [1-H-2-H<sub>2</sub>N-*closo*-1-CB<sub>11</sub>F<sub>10</sub>]<sup>-</sup> (**3**) and related {*closo*-1-CB<sub>11</sub>} clusters.<sup>c</sup>

Compound/anion	<b>3</b>	H3·0.5H <sub>2</sub> O	H3	[2-H <sub>2</sub> N- <i>closo</i> -1-CB <sub>11</sub> H <sub>11</sub> ] <sup>-</sup>	2-H <sub>3</sub> N- <i>closo</i> -1-CB <sub>11</sub> H <sub>11</sub> ·EtOH	2-H <sub>3</sub> N- <i>closo</i> -1-CB <sub>11</sub> H <sub>11</sub>
	calc.	exp.	calc.	calc.	exp.	calc.
Ref.	<sup>d</sup>	<sup>d</sup>	<sup>d</sup>	2	3	2
<i>d</i> (B–N)	1.460	1.530(4)	1.556	1.483	1.5396(10)	1.579
<i>d</i> (C–B2)	1.763	1.687(4)	1.671	1.744	1.6872(11)	1.672
<i>d</i> (C–B3/B6)	1.700	1.733(4)	1.729	1.699	1.7202(12)	1.723
<i>d</i> (C–B4/B5)	1.707	1.721(4)	1.724	1.701	1.7075(12)	1.710
<i>d</i> (B2–B3/B6)	1.793	1.788(4)	1.765	1.789	1.7697(12)	1.757
<i>d</i> (B3–B4/B5–B6)	1.803	1.804(4)	1.795	1.782	1.7782(12)	1.776
<i>d</i> (B4–B5)	1.788	1.807(4)	1.816	1.777	1.7885(13)	1.790
<i>d</i> (B2–B7/B11)	1.783	1.778(5)	1.756	1.785	1.7668(12)	1.744
<i>d</i> (B3–B7/B6–B11)	1.781	1.793(5)	1.789	1.774	1.7805(12)	1.783
<i>d</i> (B3–B8/B6–B10)	1.787	1.781(5)	1.778	1.774	1.7639(12)	1.767
<i>d</i> (B4–B8/B5–B10)	1.781	1.791(5)	1.793	1.776	1.7789(12)	1.781
<i>d</i> (B4/B5–B9)	1.779	1.790(5)	1.787	1.773	1.7731(12)	1.775
<i>d</i> (B7–B11)	1.795	1.811(4)	1.815	1.784	1.7985(12)	1.801
<i>d</i> (B7–B8/B10–B11)	1.806	1.809(4)	1.807	1.784	1.7876(12)	1.786
<i>d</i> (B9–B8/B10)	1.818	1.828(4)	1.829	1.792	1.7888(12)	1.791
<i>d</i> (B7/B11–B12)	1.803	1.803(4)	1.799	1.785	1.7799(12)	1.783
<i>d</i> (B8/B10–B12)	1.803	1.806(4)	1.823	1.784	1.7873(12)	1.792
<i>d</i> (B9–B12)	1.799	1.799(4)	1.811	1.783	1.7822(12)	1.784
<i>d</i> (B3–F3/B6–F6)	1.370	1.348(4)	1.366	–	–	–
<i>d</i> (B4–F4/B5–F5)	1.366	1.345(4)	1.352	–	–	–
<i>d</i> (B7–F7/B11–F11)	1.377	1.362(3)	1.371	–	–	–
<i>d</i> (B8–F8/B10–F10)	1.374	1.372(4)	1.358	–	–	–
<i>d</i> (B9–F9)	1.373	1.353(4)	1.357	–	–	–
<i>d</i> (B12–F12)	1.373	1.362(4)	1.358	–	–	–

<sup>a</sup> B3LYP/6-311++G(d,p). <sup>b</sup> All *d*(CB), *d*(BB) and *d*(BF) distances are mean values for C<sub>5v</sub> symmetry of the carborane cage; *d*(NH) are mean values. <sup>c</sup> Atomic distances in Å. <sup>d</sup> This work.

## References

- S1 S. Z. Konieczka, A. Himmelspach, M. Hailmann, M. Finze, *Eur. J. Inorg. Chem.*, 2013, 134–146.
- S2 M. Finze, *Chem. Eur. J.*, 2009, **15**, 947–962.
- S3 A. Himmelspach, G. J. Reiss, M. Finze, *Acta Crystallogr.*, 2011, **E67**, o704.