

Direct chemical synthesis of well dispersed L1₀-FePt Nanoparticles with tunable size and coercivity

Jianghai He,^{a,‡} Baoru Bian,^{a,‡} Qiang Zheng,^b Juan Du,^{a,*} Weixing Xia,^a Jian Zhang,^a Aru Yan,^a and J. Ping Liu^c

^aKey Laboratory of Magnetic Materials and Devices, Ningbo Institute of Material Technology & Engineering,
Chinese Academy of Sciences, Ningbo, 315201, China

^bSchool of Materials Science and Engineering, Ningbo University of Technology, Ningbo, 315016, China

^cDepartment of Physics, University of Texas at Arlington, Arlington, TX, United States.

‡ These authors contributed equally to this work.

***Corresponding author: dujuan@nimte.ac.cn (J. Du)**

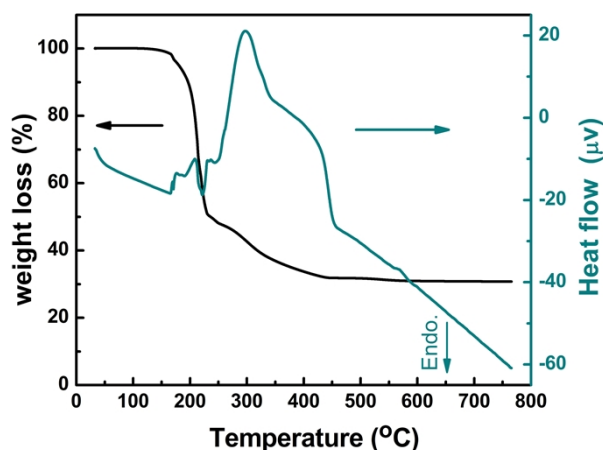


Figure S1a. TG/DTA curves illustrated the decomposition of the precursor.

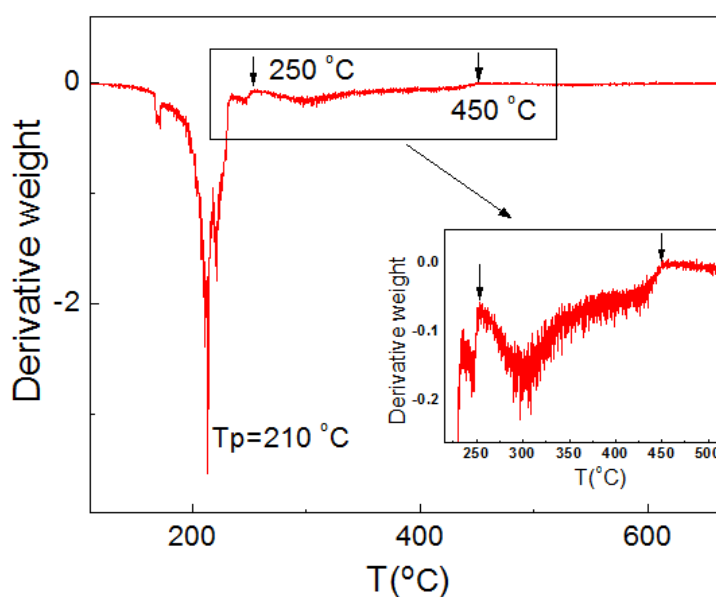


Figure S1b The derivative of weight loss.

DTA and TG were measured to know the decomposition process of the mixture of precursors ($\text{Fe}(\text{acac})_3$ and $\text{Pt}(\text{acac})_2$) in an alumina crucible with a heating rate of $5\text{ }^\circ\text{C}/\text{min}$ under Ar flow. As shown in Figure S1, the precursor has two main weight loss stages in the temperature range of $200\text{--}500\text{ }^\circ\text{C}$. The first weight loss occurs at around $200\text{ }^\circ\text{C}$, and the very sharp and strong peak at 210°C as clearly shown on the derivative curve of weight loss in Figure S2b is attributed to the decomposition of the precursors. The strong and obvious endothermic peak appeared at $300\text{ }^\circ\text{C}$ has a little bit delay compared with that of weight loss. The second weight loss occurs at $250\text{ }^\circ\text{C}$ and finished at $450\text{ }^\circ\text{C}$ as clearly shown in Figure S2b, which may be attributed to the loss of carbon. Since the decomposition of metal precursors of $\text{Fe}(\text{acac})_3$ and $\text{Pt}(\text{acac})_2$ gives out Fe and Pt metal atoms, carbon C and H_2O . At temperature around $250\text{ }^\circ\text{C}$,

the reaction of $C+H_2O = H_2+CO$ will happen, which leads to the weight loss and an endothermic kink. At a temperature around 450 °C, this reaction ends, and therefore weight loss stops and endothermic peak disappears. There is another possibility that the change of weight loss between 250 °C and 450 °C may due to a phase transition from fcc structure to $L1_0$ structure. At higher temperatures no weight loss and endothermic peak is observed.

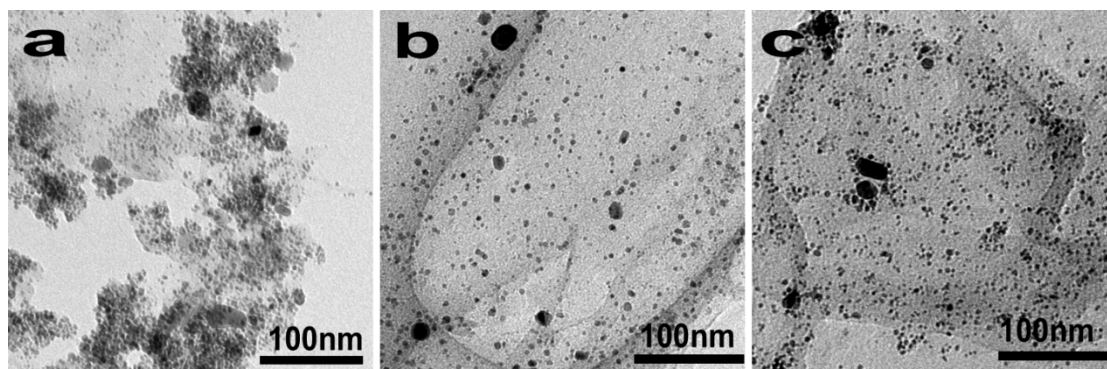
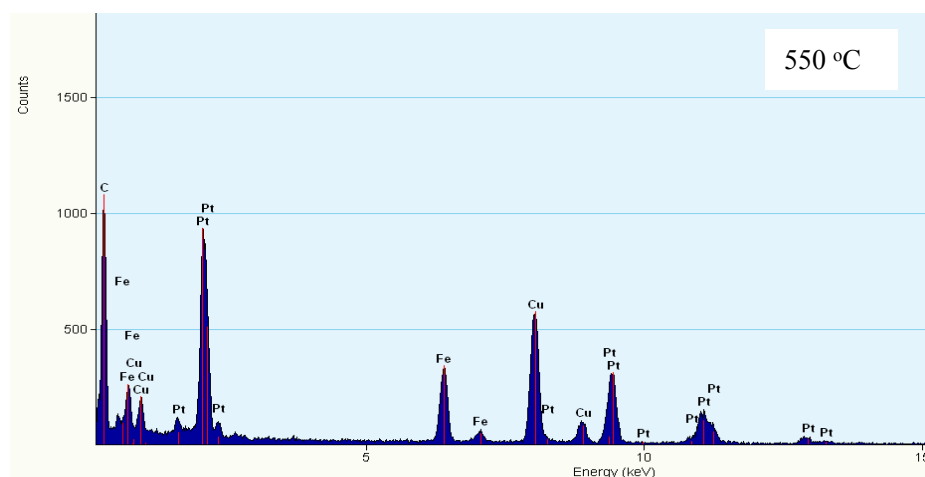
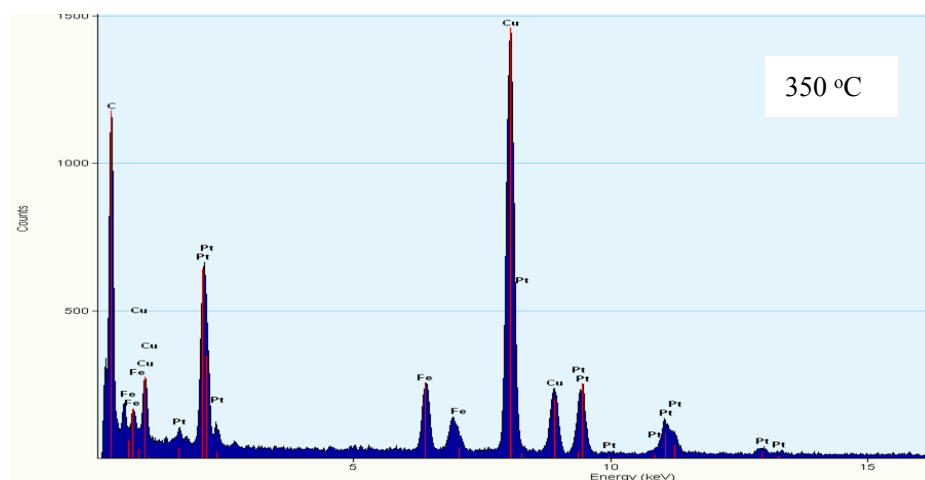


Figure S2 TEM images of the FePt NPs obtained at 300 °C (a), 400 °C (b), 450 °C (c).



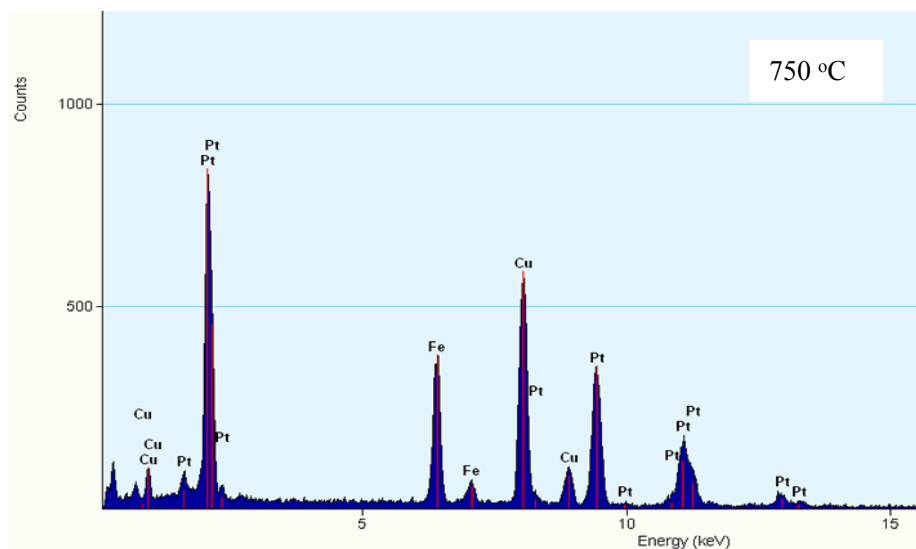


Figure S3. The composition of FePt NPs was characterized by EDS. The EDS image of FePt nanoparticles prepared at temperature of 350 °C, 550 °C and 750 °C.

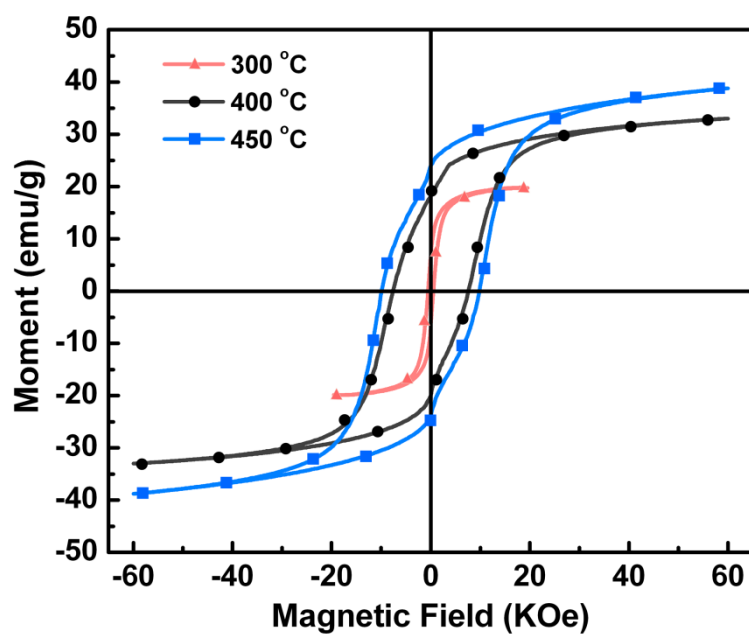


Figure S4. Room temperature magnetic hysteresis loops of the FePt NPs prepared at 300 °C, 400 °C and 450 °C.

Table 1 Temperature dependent of the synthesized FePt nanoparticles size, coercivity and chemical ordering degree.

Temperature (°C)	300	350	400	450	550	750
Average size (nm)	6.6	7.2	8.3	9.8	11.5	15
H_c (Oe)	530	3150	7600	9900	11200	21500
Ordering degree, S	-	-	0.57	0.64	0.66	0.72