

ESI Table 1. The initial reactant concentrations and source references.

Reactant	Initial concentration	Unit	Source references
<i>SCFAcoeff</i>	(1.00 → 3.60)	μM	<i>N/A</i>
<i>TR</i>	0.05	μM	^{1,2}
<i>TRac</i>	0	μM	
<i>HST</i>	0.3	μM	^{1,2}
<i>HSTac</i>	0	μM	
<i>βeta</i>	1.1	μM	³
<i>MT</i>	5.5	μM	⁴⁻⁷
<i>tuGDP</i>	3	μM	
<i>tuGTP</i>	1.5	μM	

ESI Table 2. The kinetic reactions, rate constants, rate laws and ODEs used in the modeled system.

Equation; activator	Rate constant	Rate law	ODE
$TR \rightarrow TRac;$ $SCFAcoeff$ $TRac \rightarrow TR$	k1f k1b	irr. mass action, activated irr. mass action	$d[TRac]/dt = - (k1b[TRac])$ $+ (k1f [TR] [SCFAcoeff])$
$HST \rightarrow HSTac;$ $TRac$ $HSTac \rightarrow HST$	k2f k2b	irr. mass action, activated irr. mass action	$d[HSTac]/dt = - (k2b[HSTac])$ $+ (k2f [HST] [TRac])$
$\rightarrow \beta eta;$ $HSTac$	k3f	zero order, activated	$d[\beta eta]/dt = - (k3b[\beta eta])$ $+ (k3f [HSTac])$
$\beta eta \rightarrow$	k3b	irr. mass action	
$tuGTP \rightarrow MT$	k4; k4x	non-linear second order	$d[MT]/dt = - (k5[MT][\beta eta])$ $+ ([tuGTP] (k4x + k4 [MT]^2))$
$MT \rightarrow tuGDP;$ βeta	k5	irr. mass action, activated	$d[tuGDP]/dt = - (k6 [tuGDP])$ $+ (k5[MT][\beta eta])$
$tuGDP \rightarrow tuGTP$	k6	irr. mass action	$d[tuGTP]/dt = - ([tuGTP] (k4x + k4 [MT]^2)) + (k6 [tuGTP])$

irr: irreversible; f: denotes a forward reaction; b: denotes a backward reaction

ESI Table 3. The forward and reverse kinetic rate constants in the modelled pathway.

Reaction	Rate constant	Value	Units	Range	Source references
TR acetylation	k1f	0.30	$\mu\text{M}^{-1}\text{s}^{-1}$	0.2 to 5 s^{-1}	1,8
TR deacetylation	k1b	0.25	s^{-1}		
Histone acetylation	k2f	0.275	$\mu\text{M}^{-1}\text{s}^{-1}$	0.35 s^{-1} ; 0.3 s^{-1}	9
Histone deacetylation	k2b	0.05	s^{-1}		
<i>beta</i> synthesis	k3f	0.23	s^{-1}	0.01–0.6 s^{-1}	10,11
<i>beta</i> degradation	k3b	0.30	s^{-1}	0.06–2 s^{-1}	
MT rescue	k4	50			
	k4x	0.081	s^{-1}	upper & lower	12
MT catastrophe	k5	1	$\mu\text{M}^{-1}\text{s}^{-1}$	constraints	
<i>tuGDP</i> phosphorylation	k6	0.0065	s^{-1}		

f: denotes a forward reaction; b: denotes a backward reaction

ESI Table 4. Published experimental data for MT dynamicity parameters in **(A)** interphase cells and control cells for MT destabilising and stabilising treatments; **(B)** mitotic cells and cells undergoing MT-destabilising and MT-stabilising treatments.

(A)

MT dynamicity parameter	Interphase cells ^a	control cells (destabilising) ^b	control cells (stabilising) ^b
catastrophe frequency [s^{-1}]	0.023	0.004	0.012
rescue frequency [s^{-1}]	0.108	0.036	0.020
% time elongating	15.9	66.2	38.7
% time shortening	12.2	7.2	19.9
% time pausing	72.0	28.9	39.4
growth rate [$\mu m. min^{-1}$]	9.875	3.6	4.8
shrinkage rate [$\mu m. min^{-1}$]	14.50	9.5	6.4

(B)

MT dynamicity parameter	Mitotic cells ^a	MT-destabilised cells ^b	MT-stabilised cells ^b
catastrophe frequency [s^{-1}]	0.089	0.018	0.006
rescue frequency [s^{-1}]	0.033	0.021	0.038
% time elongating	50.5	34.1	63.8
% time shortening	38.1	29.0	7.6
% time pausing	11.4	38.7	29.0
growth rate [$\mu m. min^{-1}$]	13.12	6.0	4.0
shrinkage rate [$\mu m. min^{-1}$]	15.15	6.9	10.0

Mean values: ^a Rusan *et al.* (2001); ^b Kline-Smith & Walczak (2002) ^{6,13}

ESI Table 5. The sensitivities of the model outcomes to the initial parameters. The results show the summarised, scaled sensitivities: **(A)** the influences of the initial concentrations of the reactants on the reaction fluxes; **(B)** the influences of the kinetic rate constants on the temporal concentrations of the reactants.

A		B	
Variable: initial concentration	Target function: fluxes of reactions	Variable: kinetic rate constant	Target function: temporal concentrations
<i>TR</i>	46.73	k1f (<i>TR</i> -acetylation)	20.40
<i>HST</i>	54.14	k1b (<i>TR</i> -deacetylation)	19.13
<i>beta</i>	10.87	k2f (<i>HST</i> -acetylation)	46.70
<i>MT</i>	75.86	k2b (<i>HST</i> -deacetylation)	39.78
<i>tuGDP</i>	41.43	k3f (<i>beta</i> -synthesis)	54.09
<i>tuGTP</i>	21.03	k3b (<i>beta</i> -degradation)	48.44
		k4 (rescue)	68.51
		k4x (rescue-x)	30.62
		k5 (catastrophe)	67.14
		k6 (phosphorylation)	144.71

f: denotes a forward reaction; b: denotes a backward reaction

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