

**Electronic Supplementary Information**

**Two step and one step preparation of porous nanocomposite cellulose membranes doped with TiO<sub>2</sub>**

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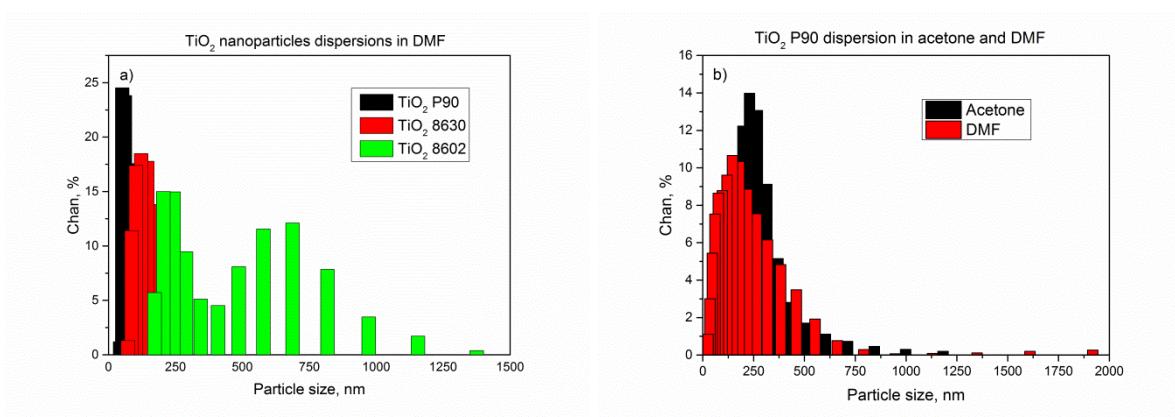


Figure SI-1. Particle size distribution (by number) for different TiO<sub>2</sub> nanoparticles in DMF (a) and for TiO<sub>2</sub> P90 in acetone and DMF (b)

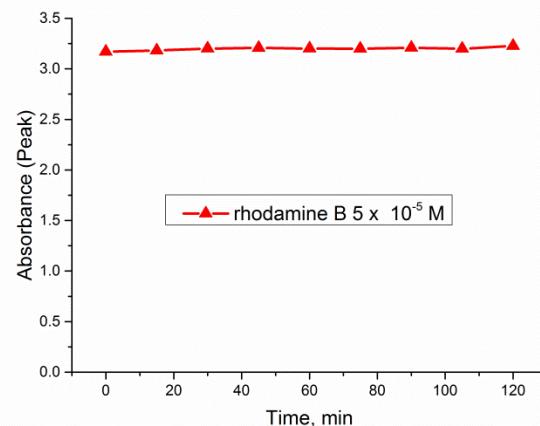


Figure SI-2. Rhodamine B degradation under UV excitation at 365 nm in the absence of catalyst

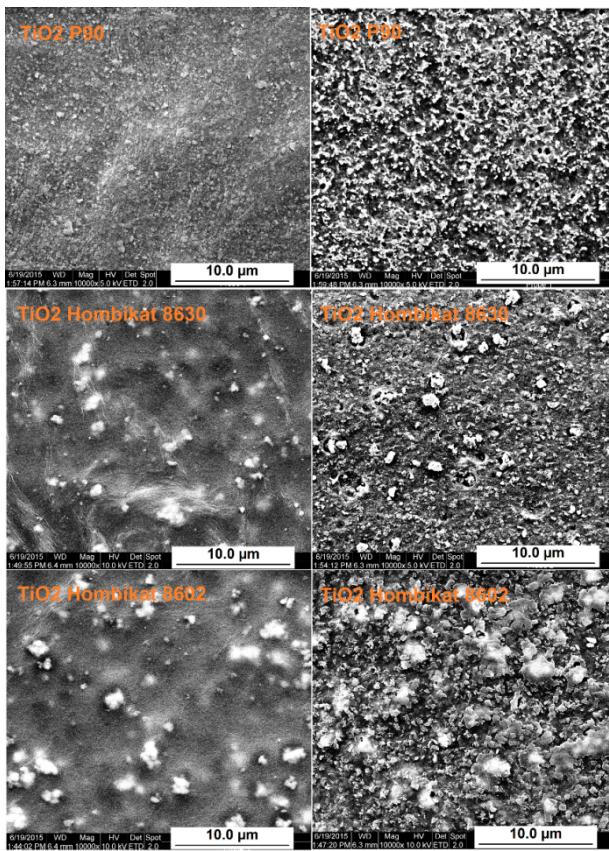


Figure SI-3. CA: $\text{TiO}_2$ =10:2 membranes prepared from DMF solution containing different  $\text{TiO}_2$  dopants: top surface (left) and glass surface (right),

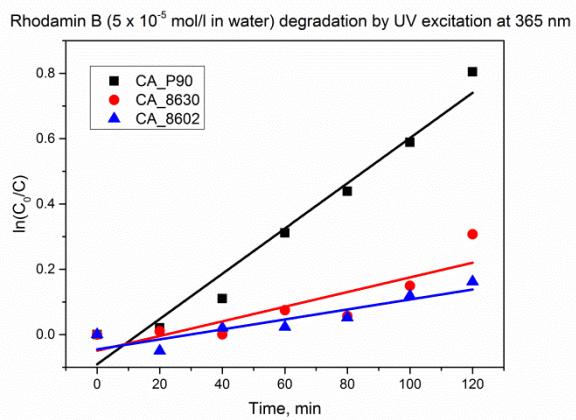


Figure SI-4. Rhodamine B degradation with CA: $\text{TiO}_2$  = 10:2 membranes (dried composite) containing different types of  $\text{TiO}_2$  nanoparticles

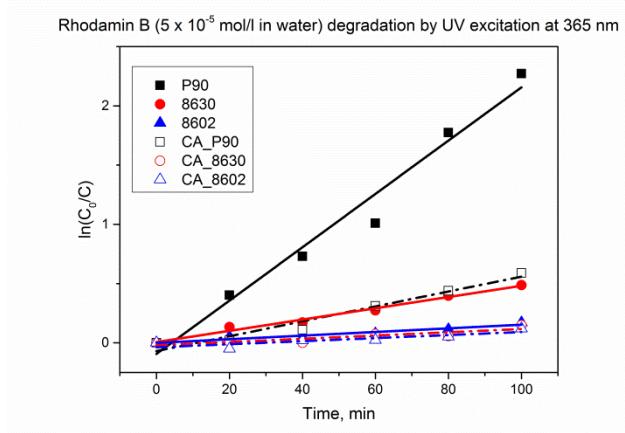


Figure SI-5. The reduction of the catalytic activity by embedment in the CA:TiO<sub>2</sub> = 10:2 membrane for all types of TiO<sub>2</sub>

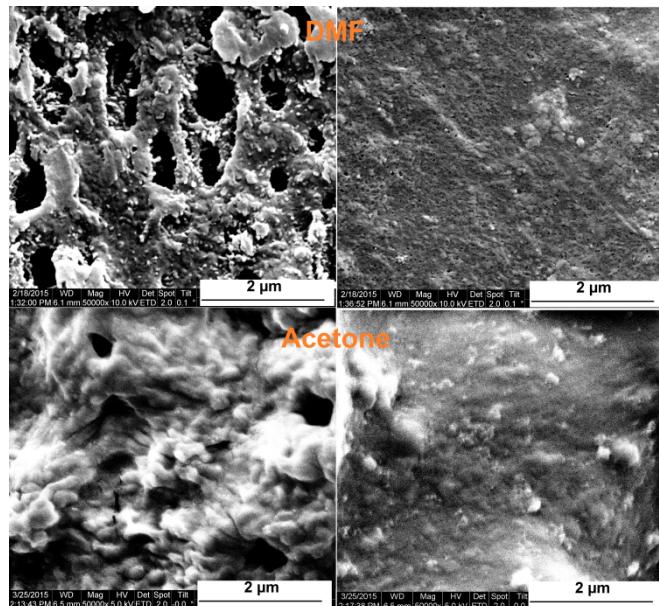


Figure SI-6. CA:TiO<sub>2</sub>=10:1 membranes prepared from DMF and acetone solution: glass surface (left) and top surface (right)

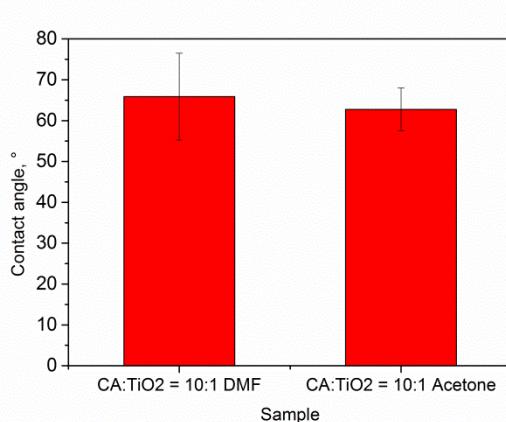


Figure SI-7. Water contact angles for the CA:TiO<sub>2</sub>=10:1 membranes prepared from DMF and acetone solutions

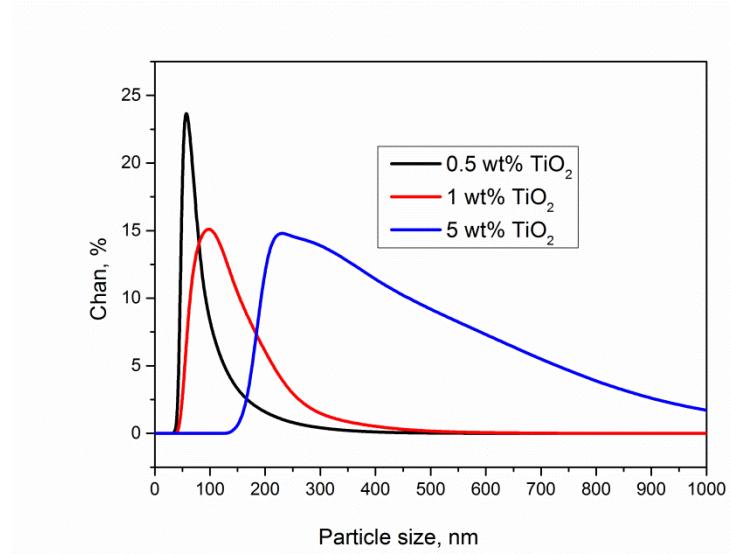


Figure SI-8. Particle size distribution of differently concentrated  $\text{TiO}_2$  P90 dispersions in DMF

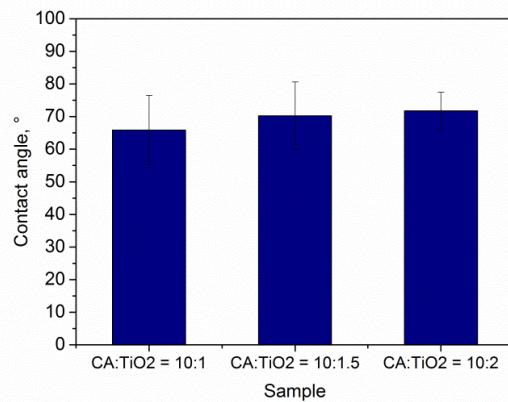


Figure SI-9. Water contact angles for CA: $\text{TiO}_2$  membranes with different  $\text{TiO}_2$  contents

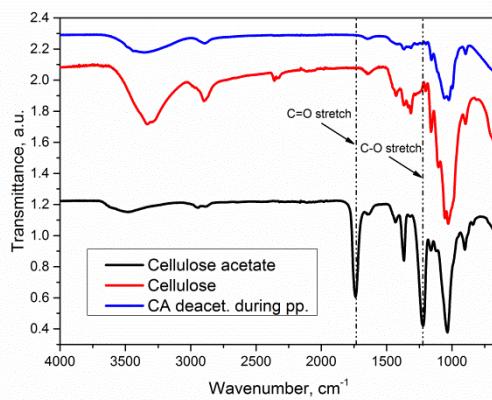


Figure SI-10. FT-IR Spectra of a CA: $\text{TiO}_2$  membrane before and after deacetylation

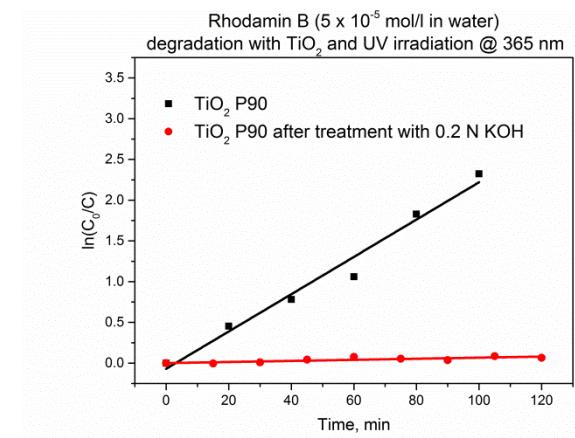


Figure SI-11. Catalytic activity of  $\text{TiO}_2$  P90 as received and after treatment with 0.2 N KOH in  $\text{H}_2\text{O}$

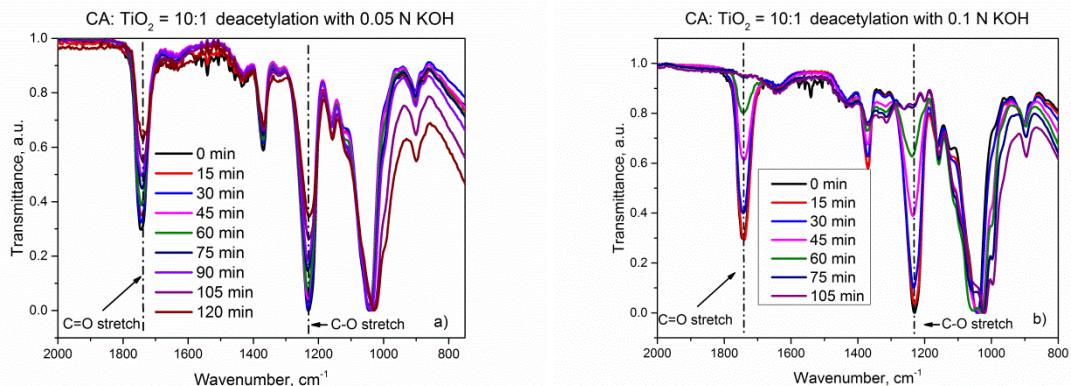


Figure SI-12. FT-IR spectra of a CA: $\text{TiO}_2$  =10:1 membrane after deacetylation for increasing amount of time with 0.05 N KOH (a) and 0.1 N KOH (b)