Electronic Supplementary Information

Facile synthesis of one Ag(I)-doped coordination polymer with enhanced catalytic performance in the photodegradation of azo dyes in water

Fan Wang,^a Fu-Ling Li,^a Miao-Miao Xu,^a Hong Yu,^{*a} Jian-Guo Zhang,^a Hai-Tao Xia^a and Jian-Ping Lang^{*a,b}

^a College of Chemistry, Chemical Engineering and Materials Science, Soochow University, Suzhou 215123, Jiangsu, People's Republic of China

^b State Key Laboratory of Organometallic Chemistry, Shanghai Institute of Organic Chemistry, Chinese Academy of Sciences, Shanghai 200032, People's Republic of China

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compound 2·3nMeCN	
Empirical Formula	$C_{66}H_{81}F_{24}N_{11}P_4Pb_2S_4$
Formula Weight	2150.92
Crystal System	monoclinic
Space Group	C2/c
<i>a</i> (Å)	29.1107(14)
<i>b</i> (Å)	10.0415(5)
<i>c</i> (Å)	31.3043(14)
β (°)	105.3360(10)
$V(\text{\AA}^3)$	8823.6(7)
Ζ	4
$\rho_{\rm calc} ({\rm g \ cm}^{-3})$	1.619
F(000)	4232
μ (MoK α , mm ⁻¹)	4.072
Total reflections	89835
Unique reflection	$8066 (R_{\rm int} = 0.0342)$
No. Observations	$6809 (I > 2.00\sigma(I))$
No. Parameters	523
R^{a}	0.0387
$R_w^{\ b}$	0.1063
GOF^{c}	1.125

 Table S1. Summary of crystallographic data for 2·3nMeCN.

^{a)} $R = \Sigma ||F_0| - |F_c|| / \Sigma |F_0|;$ ^{b)} $R_w = [\Sigma w (F_o^2 - F_c^2)^2 / \Sigma w (F_o^2)^2]^{1/2};$ ^{c)} $GOF = \{\Sigma w [(F_o^2 - F_c^2)^2] / (n-p)\}^{1/2}.$

Compound 2·3nMeCN				
Pb(1)-S(1)	2.6743 (16)	Pb(1)-S(2)	2.6287(17)	
Pb(1)-N(3)	2.636(5)	Pb(1)-N(4)	2.622(5)	
S(1)-Pb(1)-N(3)	89.62(14)	S(1)-Pb(1)-N(4)	84.70(13)	
S(2)-Pb(1)-N(3)	89.21(13)	S(2)-Pb(1)-N(4)	78.09(13)	
S(1)-Pb(1)-S(2)	89.72(6)	N(3)-Pb(1)-N(4)	166.1(2)	

Table S2. Selected bond lengths (Å) and angles (°) for 2·3nMeCN.



Fig. S1 The Raman spectra of 2 and $2 \cdot 3nMeCN$.



Fig. S2 The TGA curves for 2.



Fig. S3 The solid state emission spectra of 2 and 2.3nMeCN (excitation at 400 nm).



Fig. S4 The Raman spectrum of 2a.



Fig. S5 The XPS spectra for two samples. (a) The Pb 4f core level spectrum for 2; (b) The Pb 4f core level spectrum for 2a; (c) The S 2p core level spectrum for 2; (d) The S 2p core level spectrum for 2a; (e) The Ag 3d core level spectrum for 2a.













Fig. S6 Comparison of catalytic activity of the different catalysts (1, 2 and 2a) in photodegradation of acid orange 7 (a), orange I (b), orange IV (c), orange G (d), acid red 27 (e), sunset yellow (f), nigrosin (g), acid chrome blue K (h) and eriochrome black T (i) under UV light irradiation.



Fig. S7 Comparison of catalytic activity of 2a and $Pb(OAc)_2$ in photodegradation of MO under UV light irradiation.



Fig. S8 The PXRD patterns of 2a before and after photodegradation.



Fig. S9 Mechanism of the catalytic photodegradation by compound 2 (a) and 2a (b) under UV light irradiation.



Fig. S10 The effect of addition of quenchers on the catalytic activity of 2a under UV light irradiation in amido black 10B photodegradation.