

# All Conducting Polymer Electrodes for Asymmetric Solid-State Supercapacitors

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## Electronic Supplementary Information (ESI)

Specific areal capacitance ( $C_A$ ) and cell capacitance ( $C_{cell}$ ) were calculated from the charge-discharge curves according to the following equations.

Specific areal capacitance ( $\text{mF/cm}^2$ ) was calculated by dividing the single PEDOT electrode capacitance by area.

Specific areal capacitance ( $C_A$ ) =  $(i/A_{\text{single}})(\Delta t/\Delta E)$  (for 3-electrode configuration).

$A_{\text{single}}$  is the area of single electrode,  $i$  is the current applied,  $\Delta t$  is the discharge time and  $\Delta E$  is the potential window.

Cell capacitance ( $C_{cell}$ ) =  $(i/A_{\text{two}})(\Delta t/\Delta V)$  (for 2-electrode configuration).

$A_{\text{two}}$  is the total area of both the electrodes.

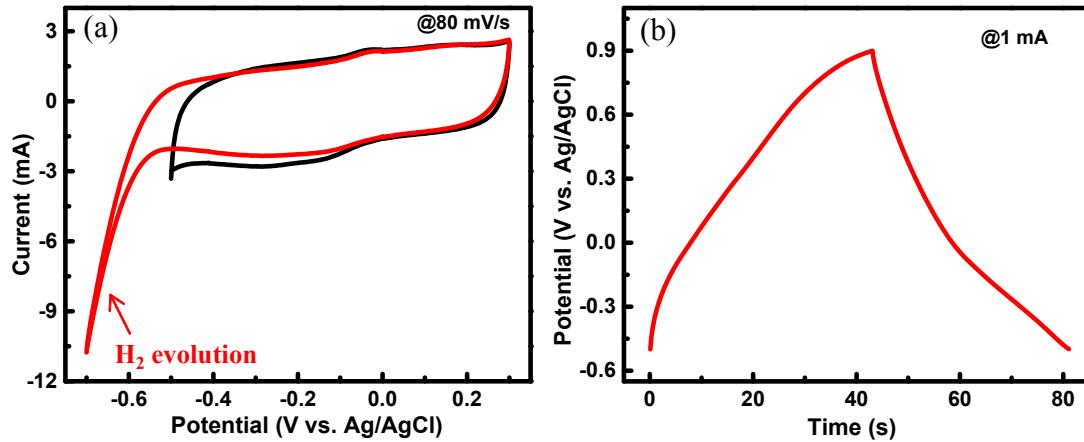
Volumetric stack capacitance ( $\text{F/cm}^3$ ) was calculated by considering the total volume of the both the electrodes.

Volumetric stack capacitance ( $C_{\text{vol}}$ ) =  $(i/v_t)(\Delta t/\Delta V)$

Energy density ( $E$ ) =  $\frac{1}{2}C_{\text{vol}}V^2$  (in  $\text{Wh/cm}^3$ )

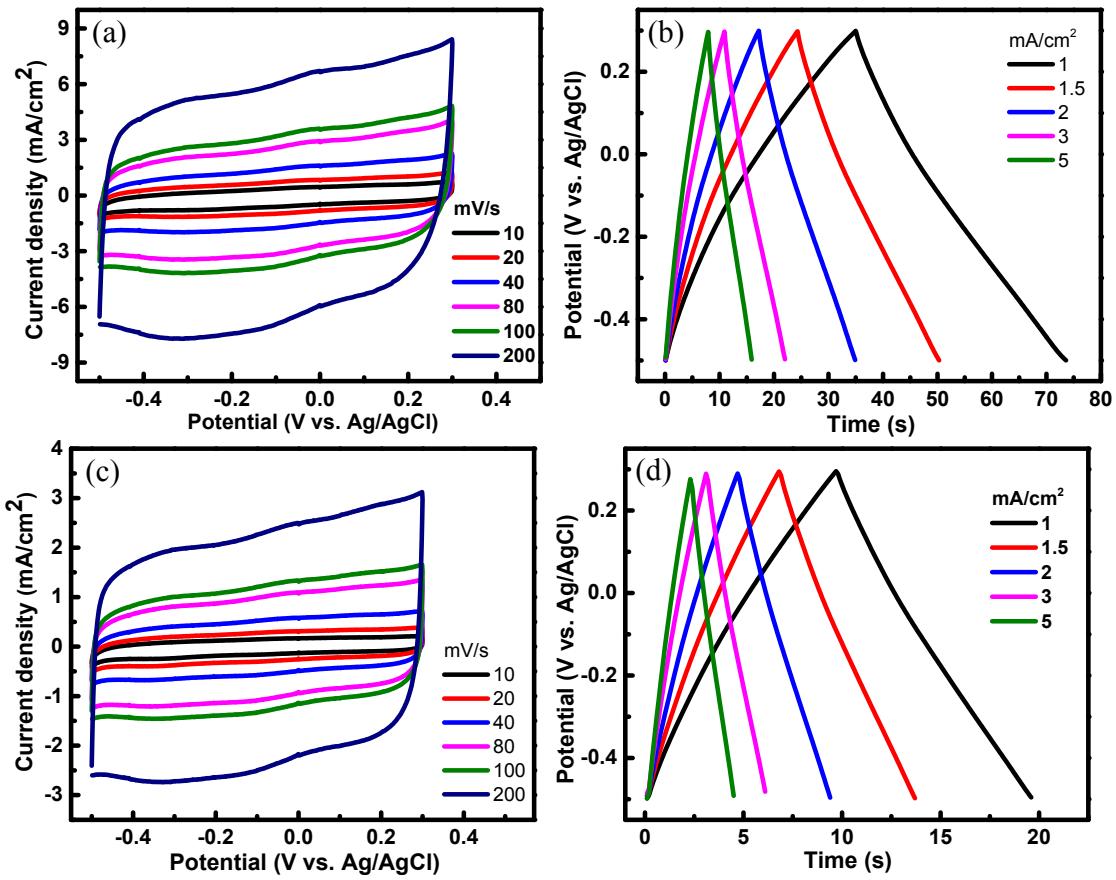
$$\text{Power density (P)} = \frac{E}{\Delta t} \text{ (in W/cm}^3\text{).}$$

Where  $i$  is the discharge current density,  $A_{\text{single}}$  and  $A_{\text{two}}$  are the areas of the single and two electrodes in  $\text{cm}^2$  respectively,  $C_{\text{cell}}$  is the cell capacitance,  $C_{\text{vol}}$  is the volumetric stack capacitance,  $v_t$  is the total volume of the electrodes.

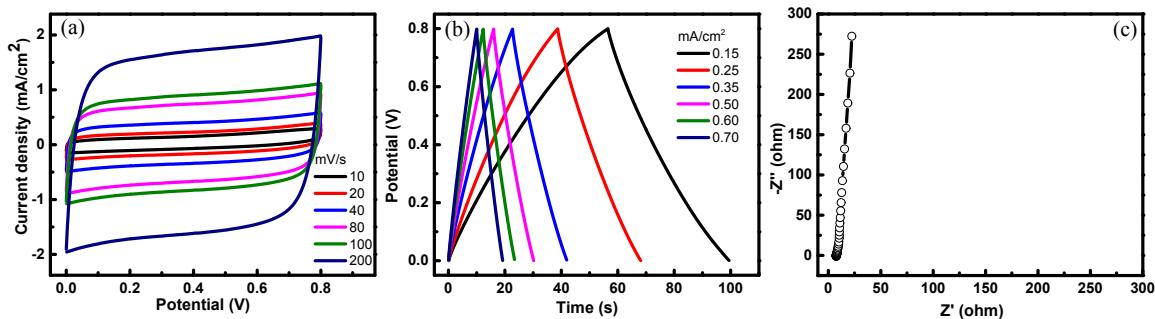


**Fig. S1** (a) CVs of PEDOT in different negative potential windows. (b) Charge-discharge curve for the PEDOT in a wide potential window of 1.4 V.

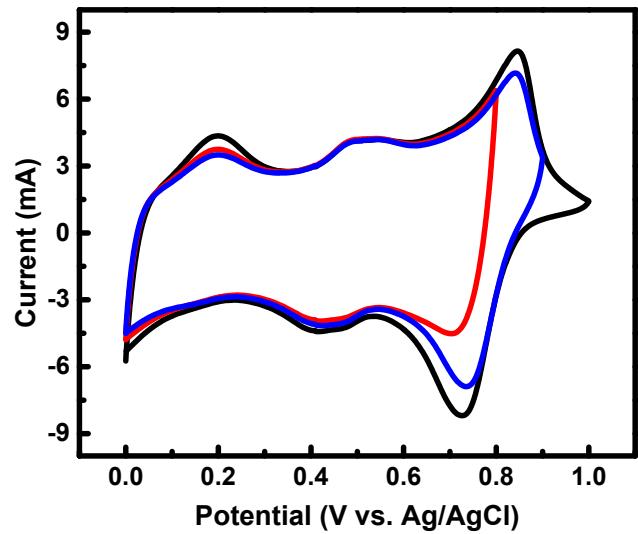
The sharp rise in the current below -0.6 V is due to hydrogen evolution reaction (see red curve in Figure S1a).



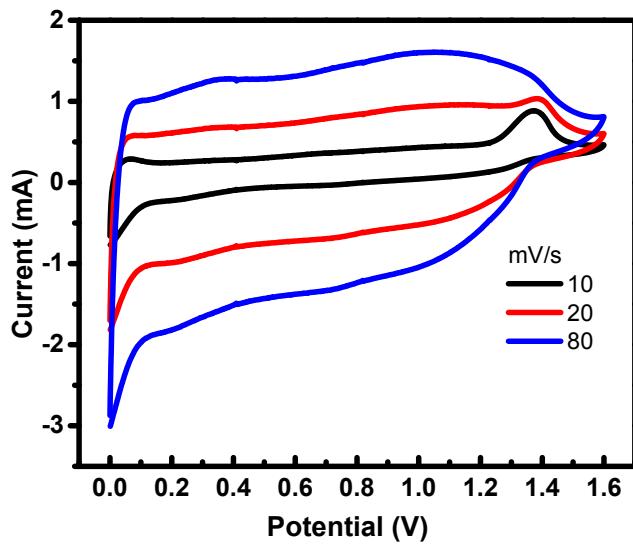
**Fig. S2** (a) and (b) CV and CD of 5 minute deposited PEDOT sample. (c) CV and (d) CD of 15 minute deposited PEDOT sample.



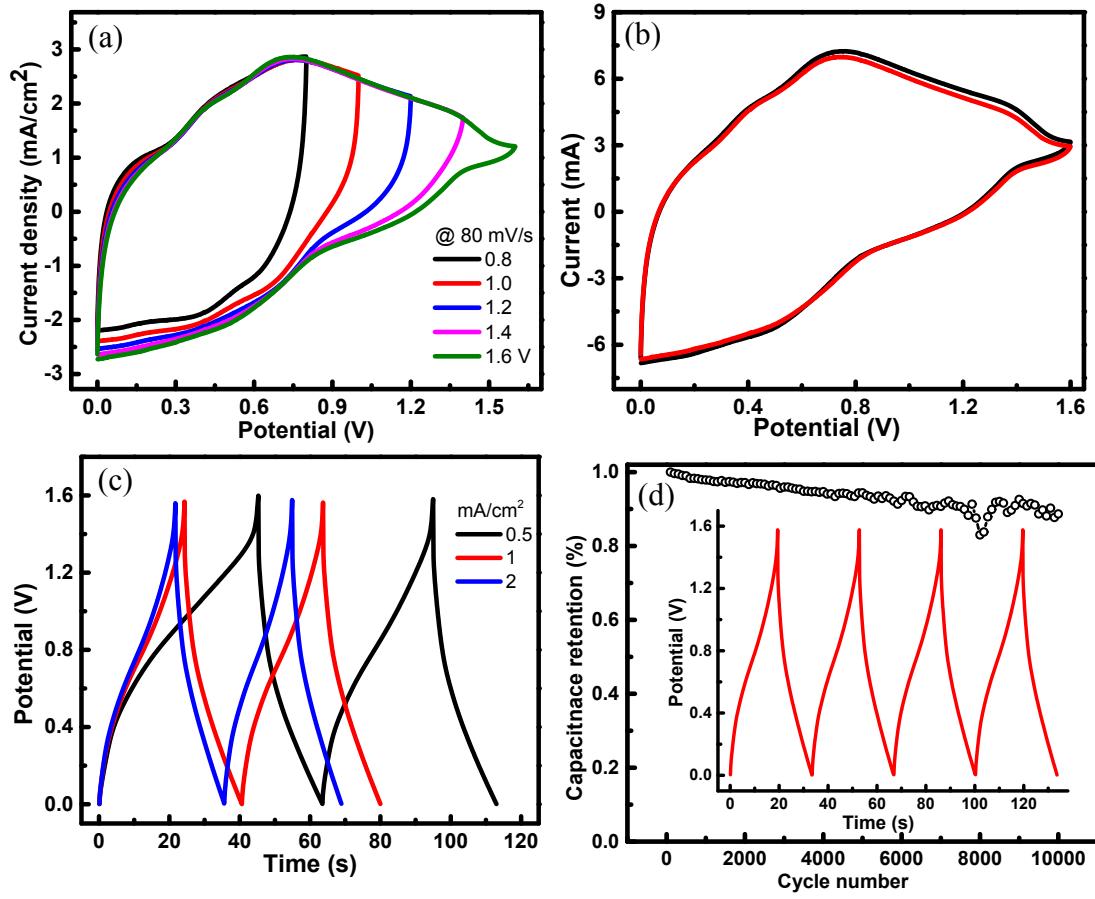
**Fig. S3** (a) CV and (b) CDs of symmetric PEDOT/Au/PEN//PEDOT/Au/PEN solid state device using PVA/ $\text{H}_2\text{SO}_4$  gel electrolyte. (c) Nyquist plot for the symmetric PEDOT solid state device.



**Fig. S4** CVs of PANI in different positive potential windows at a scan rate of 80 mV/s. CV is getting narrow down above the potential of 0.8 V.



**Fig. S5** Unoptimised CVS of the PANI//PEDOT ASC.



**Fig. S6** (a) CVs of ASC device at a scan rate of 80 mV/s in different potential windows. (b) CV scans at a scan rate of 80 mV/s and (c) CDs of ASC device at different current densities after 100 cycles of charging and discharging. (d) Cycling stability of the optimized PANI//PEDOT ASC solid state supercapacitor over 10,000 cycles. Inset shows the charge-discharge curves at a current density of 2  $\text{mA/cm}^2$ .

**Table S1.** Comparison of the electrochemical performance of the ASCs reported in the literature.

S. No.	ASC	Electrolyte	Potential window	Energy density (mWh/cm <sup>3</sup> )	Power density (W/cm <sup>3</sup> )	References
1.	VO <sub>x</sub> //VN	PVA/LiCl	1.8	0.61	0.85	Lu et al., Nano Lett. 2013, 13, 2628–2633
2.	PANI//MoO <sub>3</sub> /WO <sub>3</sub>	PVA/H <sub>3</sub> PO <sub>4</sub>	1.9	1.9	0.73	Xiao et al., Adv. Energy Mater. 2012, 2, 1328–1332
3.	MnO <sub>2</sub> NWs//Fe <sub>2</sub> O <sub>3</sub> NTs	PVA/KCl	1.6	0.55	0.139	Yang et al., Nano Lett. 2014, 14, 731–736.
4.	Co <sub>9</sub> S <sub>8</sub> // Co <sub>3</sub> O <sub>4</sub> @RuO <sub>2</sub>	PVA/KOH	1.6	1.44	0.89	Xu et al., ACS Nano, 2013, 7, 5453–5462
5.	PANI//PEDOT	PVA/H <sub>2</sub> SO <sub>4</sub>	1.6	9	2.8	This work