

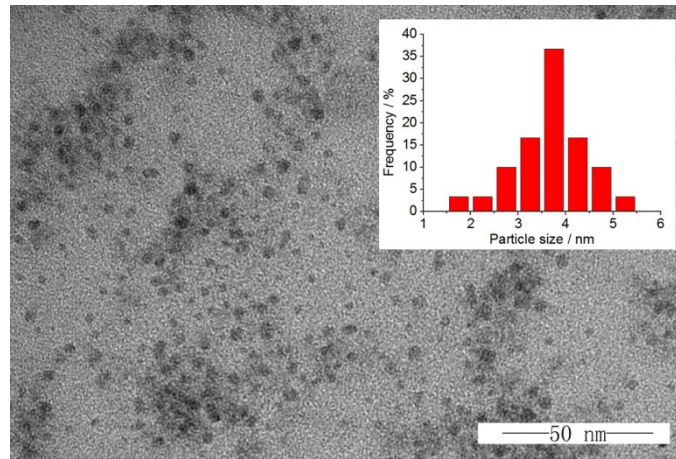
## Electronic Supplementary Information

### **Synthesis of magnetic core-shell carbon dots@MFe<sub>2</sub>O<sub>4</sub> (M = Mn, Zn and Cu) hybrid materials and their catalytic properties**

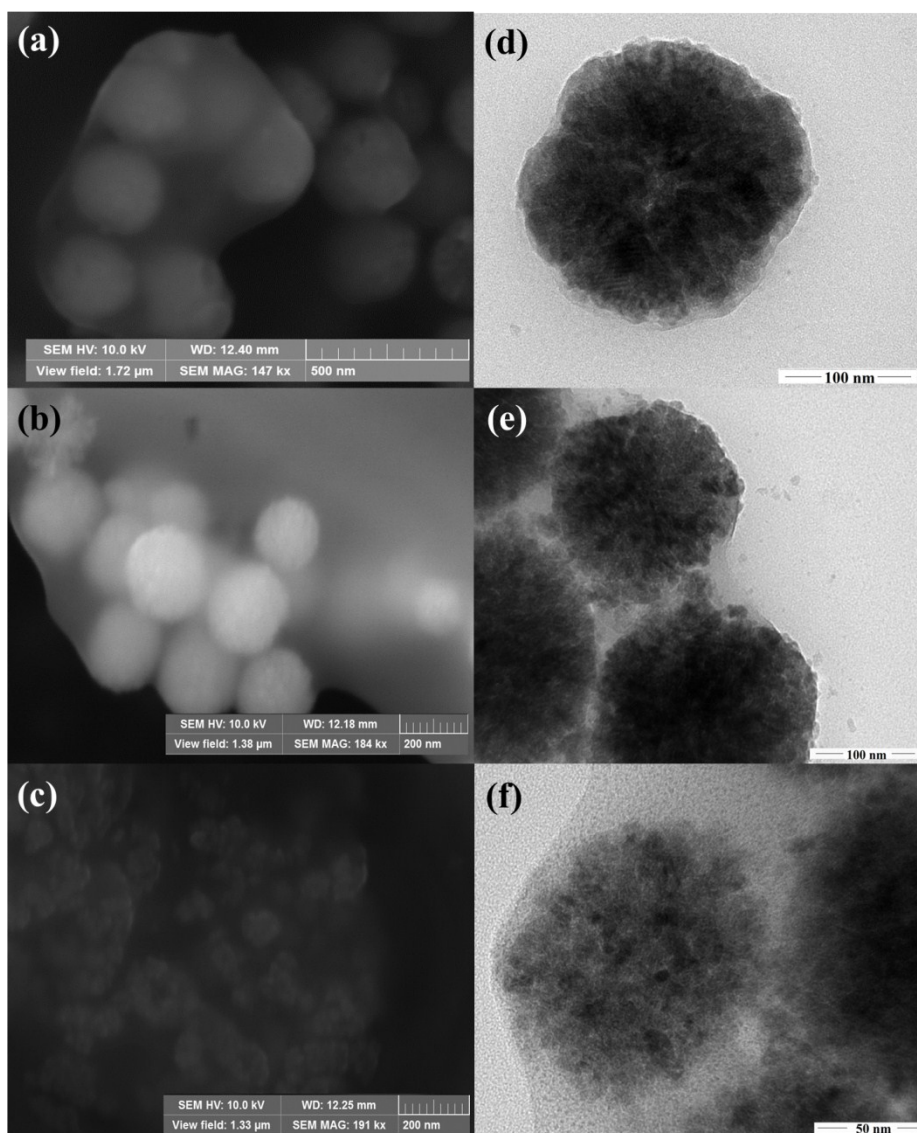
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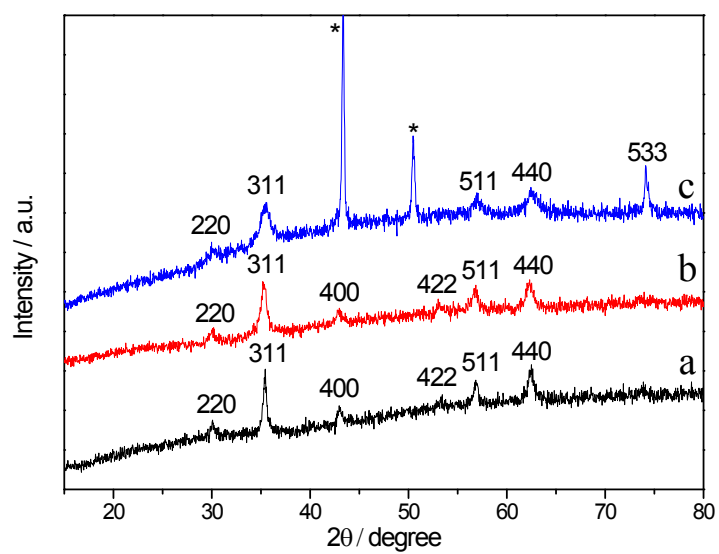
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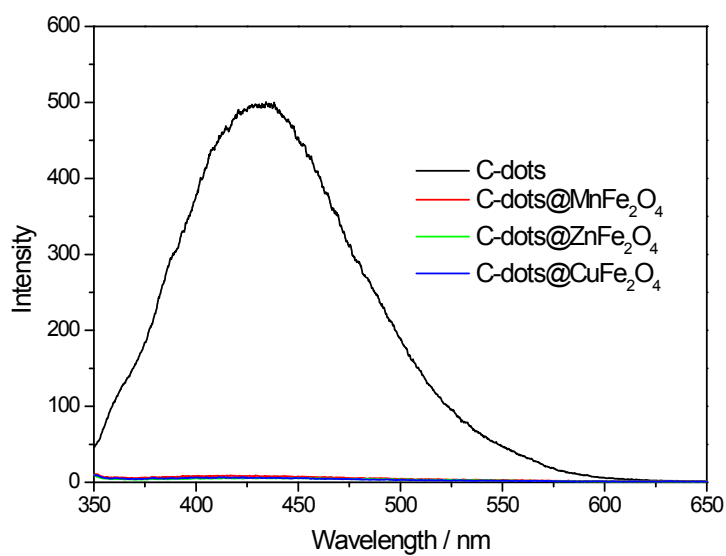
**Figure S1.** TEM image of as-prepared C-dots. (Inset: Size distribution histogram obtained from TEM measurements.)



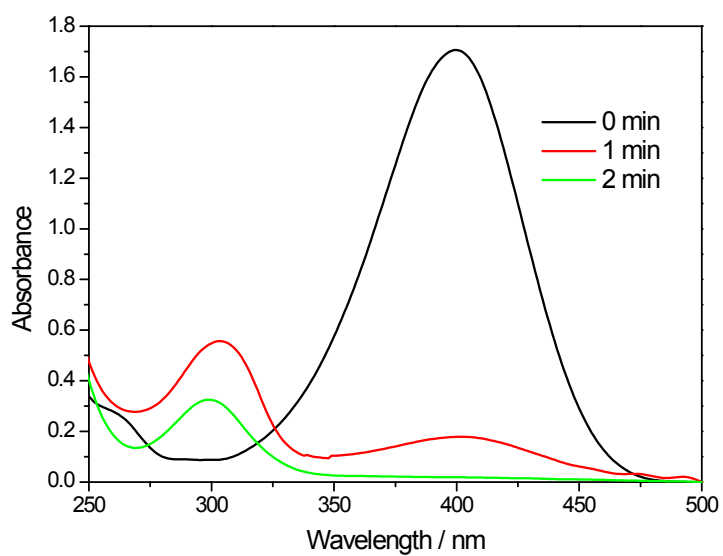
**Figure S2.** Magnified FLSEM and TEM images of (a, d) C-dots@MnFe<sub>2</sub>O<sub>4</sub>, (b, e) C-dots@ZnFe<sub>2</sub>O<sub>4</sub> and (c, f) C-dots@CuFe<sub>2</sub>O<sub>4</sub>.



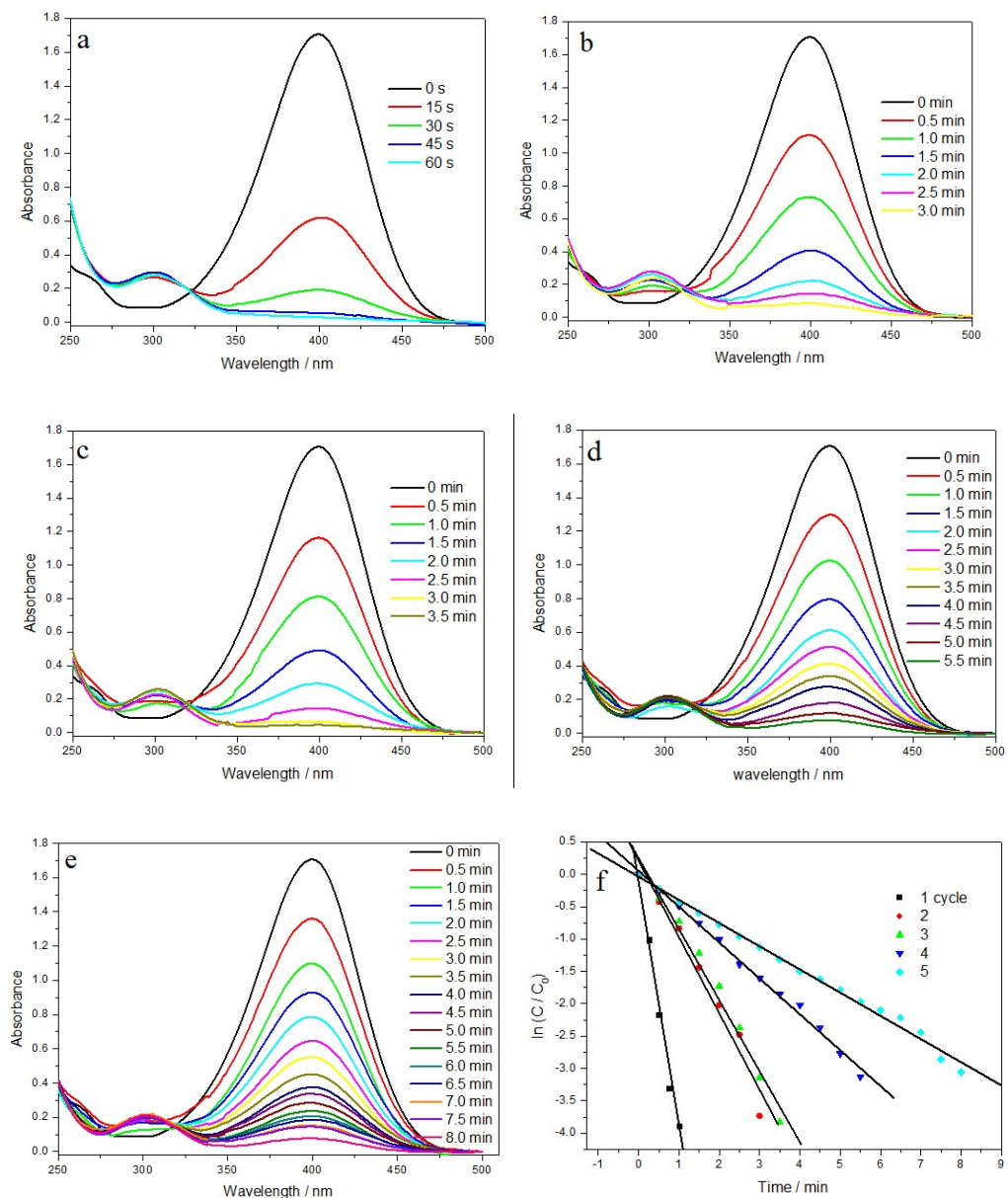
**Figure S3.** XRD patterns of (a) MnFe<sub>2</sub>O<sub>4</sub> MNPs, (b) ZnFe<sub>2</sub>O<sub>4</sub> MNPs and (c) CuFe<sub>2</sub>O<sub>4</sub> MNPs.



**Figure S4.** PL spectra of the C-dots and C-dots@MFe<sub>2</sub>O<sub>4</sub> (M= Mn, Zn and Cu) hybrid materials.



**Figure S5.** Time-dependent UV-Vis absorption spectra for the reduction of p-NP by  $\text{NaBH}_4$  in the presence of  $\text{CuCl}_2$ .



**Figure S6.** Time-dependent UV-visible absorption spectra for (a) 1<sup>st</sup>, (b) 2<sup>nd</sup>, (c) 3<sup>rd</sup>, (d) 4<sup>th</sup> and (e) 5<sup>th</sup> cycles of the catalytic reduction of p-NP with the C-dots@CuFe<sub>2</sub>O<sub>4</sub> catalyst. (f) Plots of  $\ln(C/C_0)$  versus reaction time for five successive cycle reactions with the C-dots@CuFe<sub>2</sub>O<sub>4</sub> catalyst.

**Table S1.** Elemental and ICP analyses for the C-dots@MFe<sub>2</sub>O<sub>4</sub> (M = Mn, Zn and Cu)

hybrid materials.

Sample	element (wt %)
C-dots@MnFe <sub>2</sub> O <sub>4</sub>	C, 19.08; H, 1.92; Mn, 14.78; Fe, 41.95
C-dots@ZnFe <sub>2</sub> O <sub>4</sub>	C, 21.44; H, 2.37; Zn, 15.76; Fe, 39.59
C-dots@CuFe <sub>2</sub> O <sub>4</sub>	C, 20.88; H, 2.16; Cu, 22.41; Fe, 34.27



**Table S2.** The proportion of different carbon in the total carbon at the surface of the C-dots@MFe<sub>2</sub>O<sub>4</sub> (M = Mn, Zn and Cu) calculated from XPS.

Sample	C-C	C-OH	C=O	O=C-OH
C-dots@MnFe <sub>2</sub> O <sub>4</sub>	65.68%	20.02%	7.92%	6.39%
C-dots@ZnFe <sub>2</sub> O <sub>4</sub>	67.16%	19.94%	6.82%	6.08%
C-dots@CuFe <sub>2</sub> O <sub>4</sub>	63.00%	22.15%	8.53%	6.32%

**Table S3.** Substituent and reducing agent effects on the catalytic reduction of nitrophenols with C-dots@CuFe<sub>2</sub>O<sub>4</sub>.

Reducing agent	Sample	Reaction time / min	Conversion / %	$k_{app} / (\text{min}^{-1})$
NaBH <sub>4</sub>	o-NP	1	97.2	3.75
	m-NP	0.75	98.0	5.30
	p-NP	1	98.0	4.03
AB	o-NP	12	5.1	0.0033
	m-NP	14	23.2	0.0171
	p-NP	14	18.4	0.0136