Supporting Information

Order-Disorder Phase Transition Coupled with Torsion in Tri-n-

ButylammoniumTrichloroacetate (TBAT)

Muhammad Adnan Asghar,^{ab} Chengmin Ji,^a Yuelan Zhou,^a Zhihua Sun,^a Tariq Khan,^{ab}

Shuquan Zhang,^a Sangen Zhao^a and Junhua Luo^{a*}

^a Key Laboratory of Optoelectronic Materials Chemistry and Physics, Fujian Institute of Research on the Structure of Matter, Chinese Academy of Sciences, Fuzhou, Fujian, 350002, P. R. China. E-mail: <u>jhluo@fjirsm.ac.cn.</u> Fax: (+86) 0591-83730955. Tel: (+86) 0591-83730955.

^b University of Chinese Academy of Sciences, Beijing 100049, China.

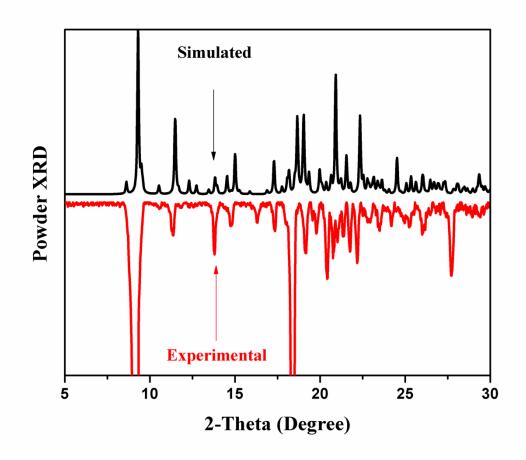


Figure S1. PXRD patterns of TBAT.

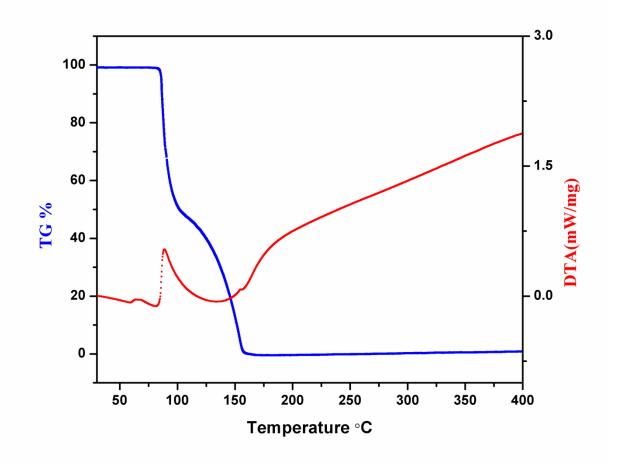


Figure S2.TG-DTA curves of TBAT.

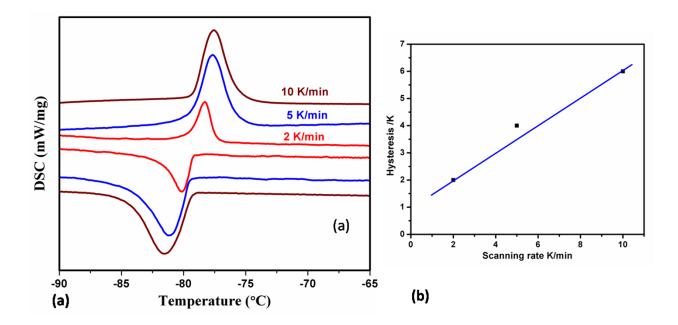


Figure S3. DSC curves of **TBAT** at (a) different scanning rates and (b) thermal hysteresis from the scans.

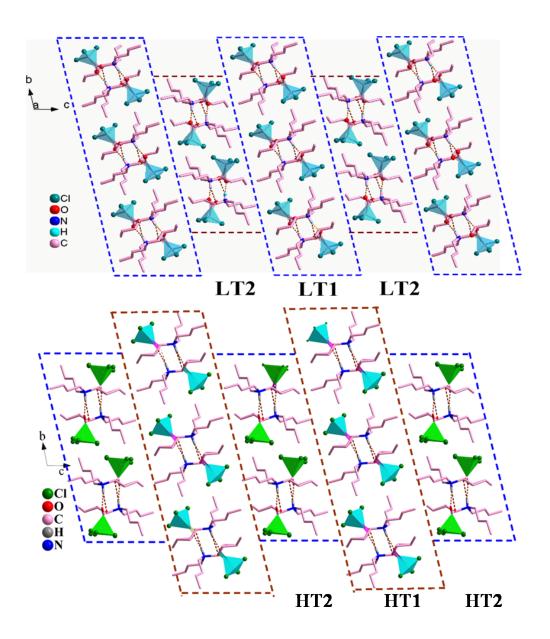


Figure S4.Unit cell packing diagrams viewed along the a-axis at (a) 100 and (b) 220 K of **TBAT**. The dotted lines stand for the hydrogen bonds.

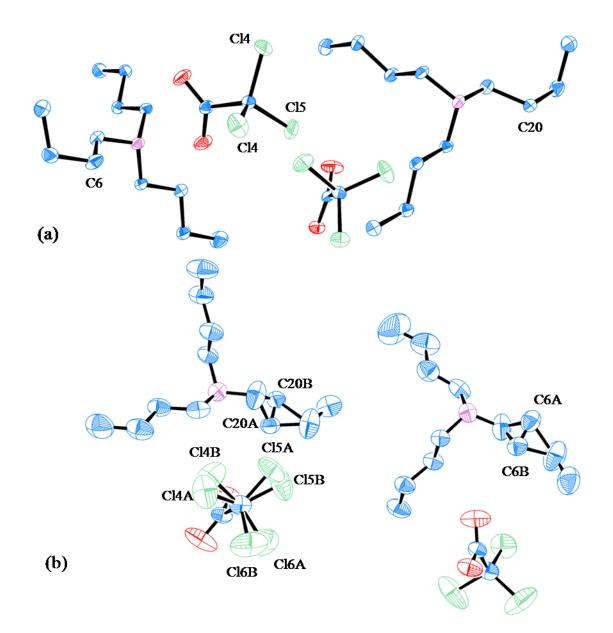


Figure S5. Thermal ellipsoidal view of **TBAT** at (a) 100 K and (b) at 220 K after splitting of appropriate atoms.

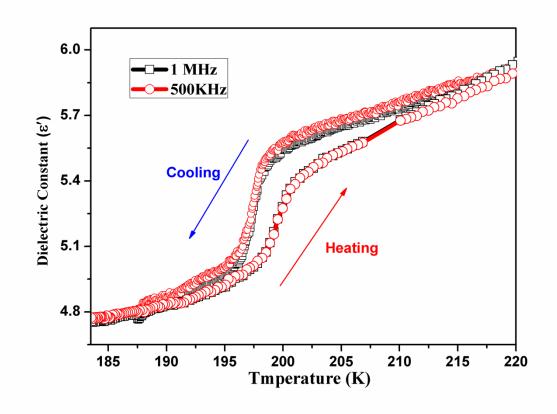


Figure S6. Temperature dependent cycle of dielectric constant of TBAT at 1 MHz and 500 KHz.

| Sum formula | C14 H28 Cl3 N O2 | C14 H28 Cl3 N O2 |
|--|------------------|------------------|
| Formula weight | 348.72 | 348.72 |
| Temperature (K) | 100 | 220 |
| Crystal system | Triclinic | Triclinic |
| Space group | PĪ | PĪ |
| a/Å | 9.6337 | 9.6704 |
| b/Å | 10.7137 | 10.5407 |
| c/Å | 18.8281 | 19.7334 |
| a/deg | 98.742 | 98.677 |
| β /deg | 100.488 | 101.66 |
| γ/deg | 100.213 | 98.350 |
| Volume (Å ³) | 1846.15 | 1915.04 |
| Ζ | 2 | 2 |
| D _{calcd} , g cm ⁻¹ | 1.255 | 1.031 |
| <i>F</i> (000) | 744 | 744 |
| Completeness (%) | 98.3 | 99.8 % |
| Goodness-of-fit on F^2 | 1.038 | 1.056 |
| $R_1 (\text{on } F_0^2, I \ge 2\sigma(I))$ | 0.0492 | 0.0652 |
| wR_2 (on F_o^2 , $I > 2\sigma(I)$ | 0.1265 | 0.2573 |

Table S1.Crystal Data and structure refinement details of TBAT at 100 and 220 K.

| 100K | D-H | Н…А | D····A | D-H····A |
|--------------------|--------------|--------------|-----------|----------------|
| N1-H101 | 0.93 | 1.79 | 2.716 (3) | 170.5 |
| N2-H2O4 | 0.93 | 1.75 | 2.677 (3) | 171.1 |
| 220K | D-H | Н…А | D····A | D-H····A |
| | | | | |
| N1-H1O1 | 0.93 | 1.77 | 2.699 (3) | 170.5 |
| N1—H1O1 N2—H2O4 | 0.93 0.93 | 1.77 1.81 | | 170.5 171.1 |

Table S2. Hydrogen-bond geometry (Å, deg) at 100 and 220 K in TBAT.

| Atoms | U(equiv) | Atoms | U(equiv) |
|-------|----------|-------|----------|
| C1 | 0.064 | C15 | 0.052 |
| C2 | 0.072 | C16 | 0.062 |
| C3 | 0.100 | C17 | 0.072 |
| C4 | 0.178 | C18 | 0.098 |
| C5 | 0.081 | C19 | 0.093 |
| C7 | 0.100 | C21 | 0.102 |
| C8 | 0.100 | C22 | 0.117 |
| C9 | 0.048 | C23 | 0.057 |
| C10 | 0.054 | C24 | 0.062 |
| C11 | 0.061 | C25 | 0.082 |
| C12 | 0.080 | C26 | 0.101 |
| C13 | 0.055 | C27 | 0.051 |
| C14 | 0.065 | C28 | 0.061 |
| N1 | 0.052 | N2 | 0.050 |
| 01 | 0.063 | 03 | 0.061 |
| 02 | 0.083 | 04 | 0.981 |
| Cl1 | 0.133 | Cl2 | 0.133 |
| Cl3 | 0.093 | | |
| | | | |

Table S3. Ueq values of TBAT at 220 K.