

*Supporting Information*

# Solid-state characterization of 17 $\beta$ -estradiol co-crystals presenting improved dissolution and bioavailability

Jian-Rong Wang,<sup>a,†</sup> Xiaojuan Wang,<sup>a,‡</sup> Yong Yang,<sup>b</sup> Xiaoyan Chen,<sup>\*,b</sup> and Xuefeng

Mei <sup>\*,a</sup>

<sup>†</sup> Pharmaceutical Analytical & Solid-State Chemistry Research Center, Shanghai Institute of Materia Medica, Chinese Academy of Sciences, Shanghai 201203, China

<sup>‡</sup> Center for Drug Metabolism and Pharmacokinetics, Shanghai Institute of Materia Medica, Chinese Academy of Sciences, Shanghai 201203, China

E-mail address: xychen@simm.ac.cn (X. Chen), xuefengmei@simm.ac.cn (X. Mei); fax: +86-21-50800934; Tel: +86-21-50800934

**Table S1.** CSD statistics of possible supramolecular synthons in E2 with phenolic hydroxyl group <sup>a</sup>

Synthon	Both groups containing hits	Synthon hits with %
<b>I</b> (OH + N <sub>arom</sub> )	1771	906 (51.2%)
<b>II</b> (OH + amide)	3266	1280 (39.2%)
<b>III</b> (OH + acid)	1136	210 (18.5%)
<b>IV</b> (OH + 1°/2°/3° amine)	941/2095/2516	164/94/188 (8.0%)

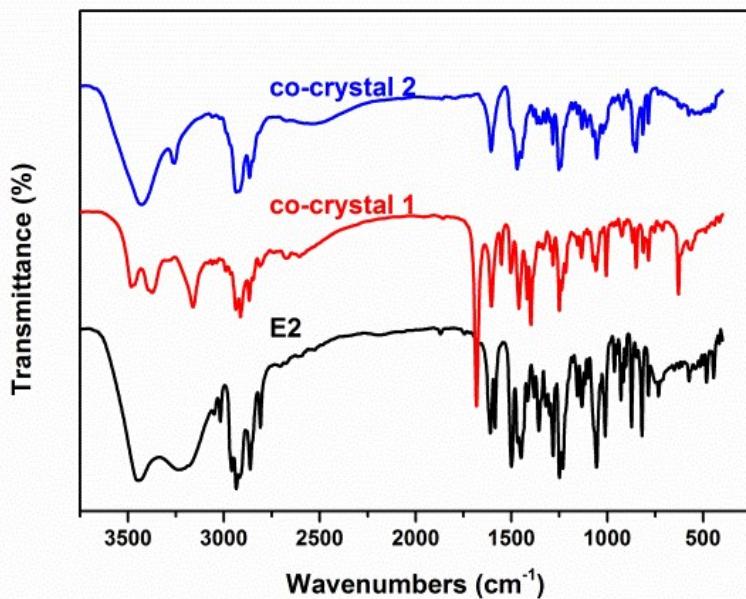
<sup>a</sup> Survey constrains: intermolecular shorter than sum of VdW radii + 0.0, no disordered, no errors, no ions, and only organics.

**Table S2.** Results of the screening experiments for the formation of co-crystals with E2 in MeOH, ethyl acetate (EA), and tetrahydrofuran (THF).

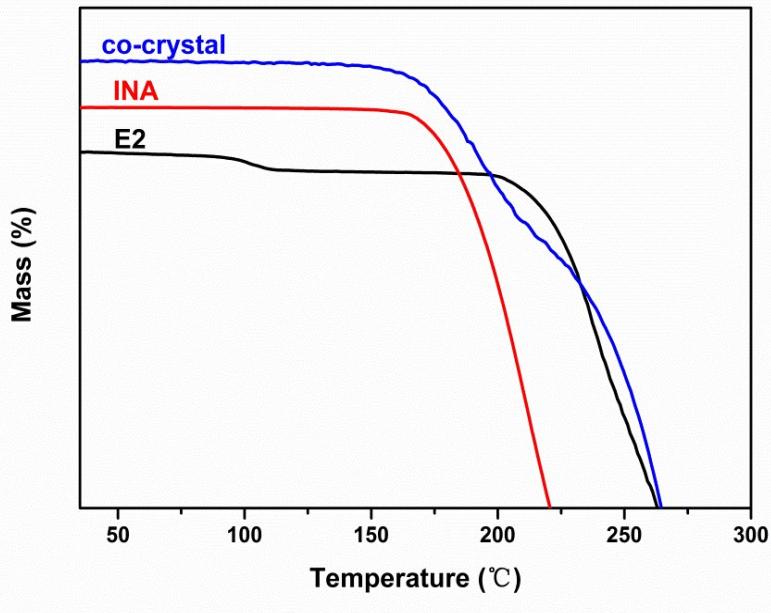
CCF	MeOH	EA	THF	CCF	MeOH	EA	THF
	✓	✓	X		X	X	X
isonicotinamide				maleic acid			
	X	X	X		X	X	X
pyrazinamide				fumaric acid			
	X	X	X		X	X	X
nicotinamide				succinic acid			
	X	X	X		X	X	X
isoniazide				adipic acid			
	X	X	X		✓	✓	✓
picolinamide				piperazine			
	X	X	X		X	X	X
nictinic acid				allophanamide			

**Table S3.** Hydrogen bonding distances and angles for E2 co-crystals

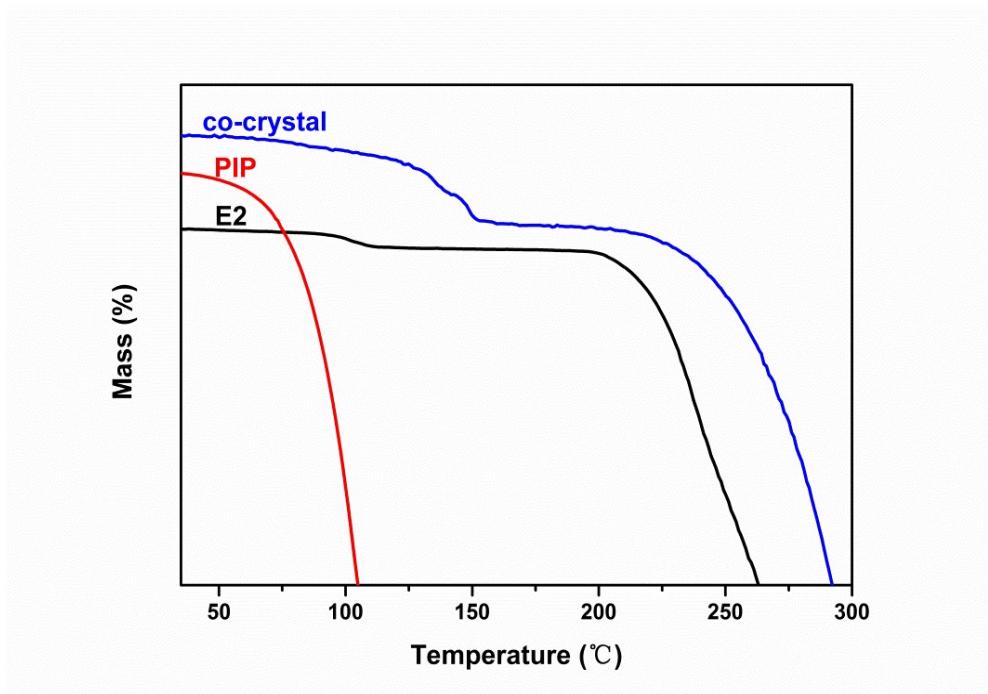
H bond	d(D-H)/Å	d(H···A)/Å	d(D···A) /Å	<DHA/°
<b>co-crystal 1</b>				
O1-H1···N1	0.877	1.894	2.764	171.50
O1'-H1'A···N3	0.820	1.948	2.767	176.18
O2'-H2'A···O5	0.851	2.021	2.852	164.84
O2-H2···O6	0.820	2.047	2.811	154.98
N4-H4B···O2'	0.863	2.058	2.908	167.75
N4-H4A···O1	0.900	2.121	3.004	166.53
N2-H2A···O2	0.923	1.990	2.884	162.40
N2-H2B···O1'	0.922	2.376	3.177	145.11
<b>co-crystal 2</b>				
O1-H1···N1	0.820	1.910	2.721	170.06
O3-H3···O1S	0.820	2.037	2.844	167.59
O2-H2C···N2	0.820	1.901	2.712	169.54
O4-H4A···O2S	0.820	2.140	2.860	146.51
O1S-H2S···O4	0.813	2.042	2.796	154.07
N1-H6S···O1S	0.880	2.245	3.096	163.00
N2-H5S···O2S	0.851	2.326	3.163	168.02
O1S-H1S···O2	0.794	2.110	2.897	171.27
O2S-H4S···O1	0.797	2.290	3.044	157.90
O2S-H3S···O3	0.806	2.031	2.836	177.46



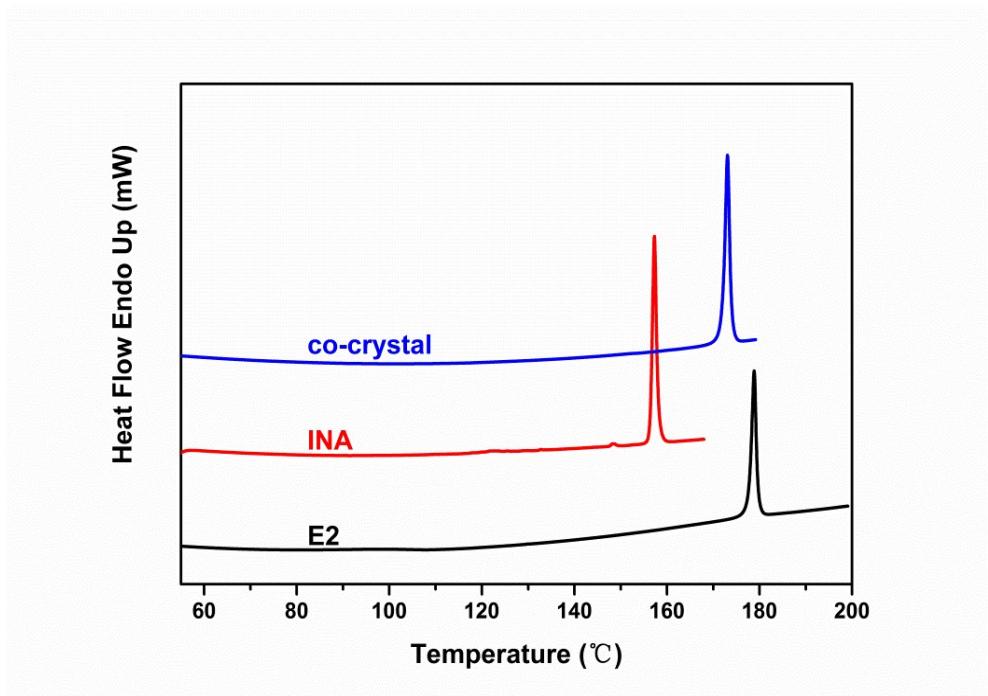
**Figure S1.** FT-IR spectroscopy of E2 co-crystals



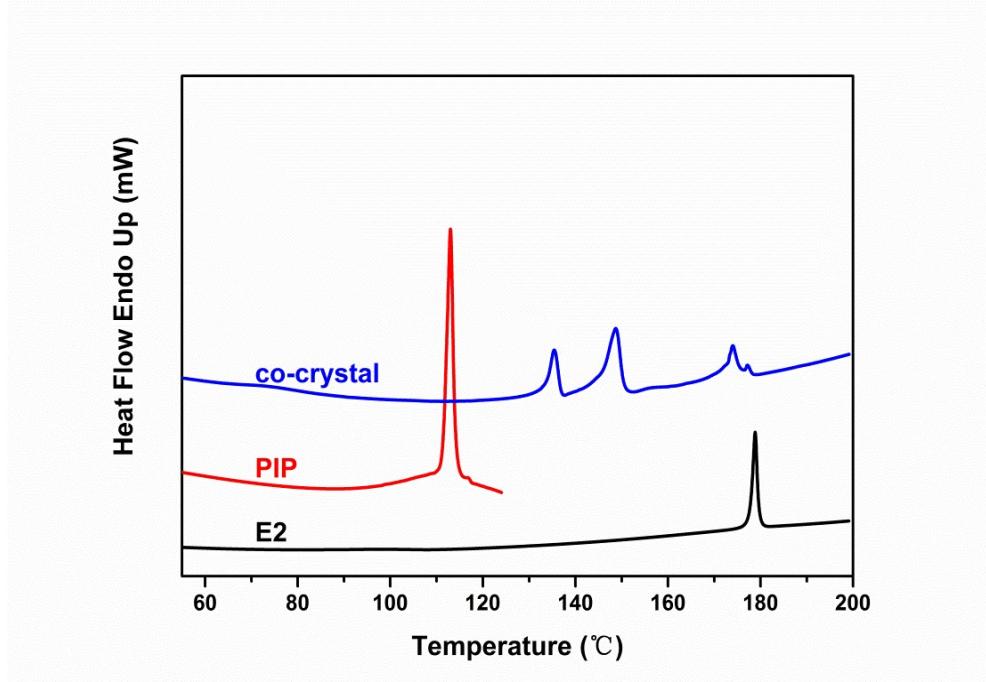
(a)



(b)

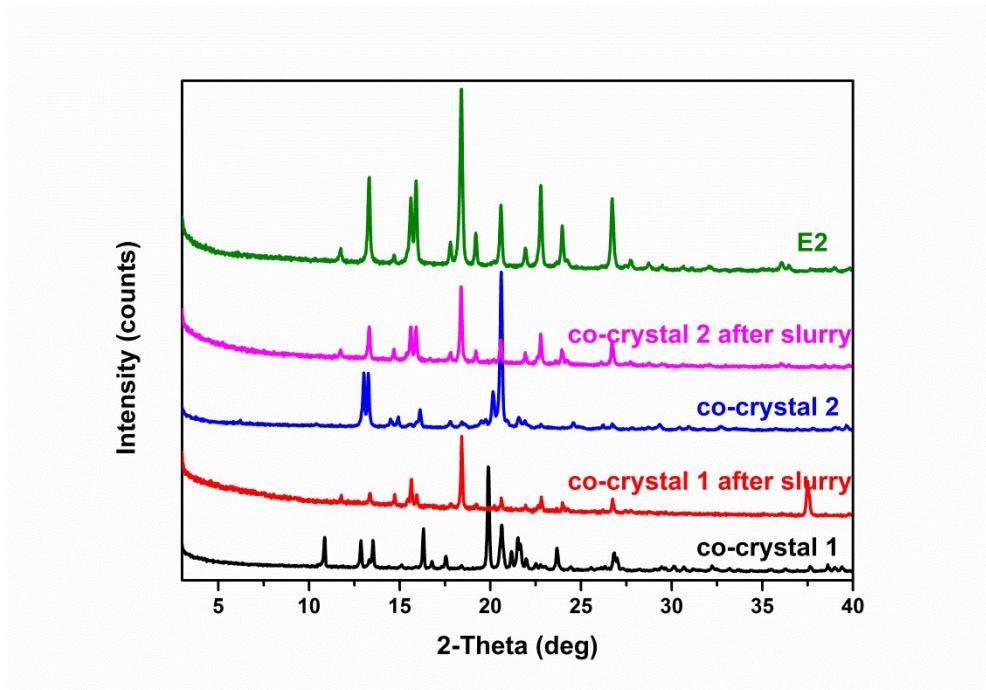


(c)

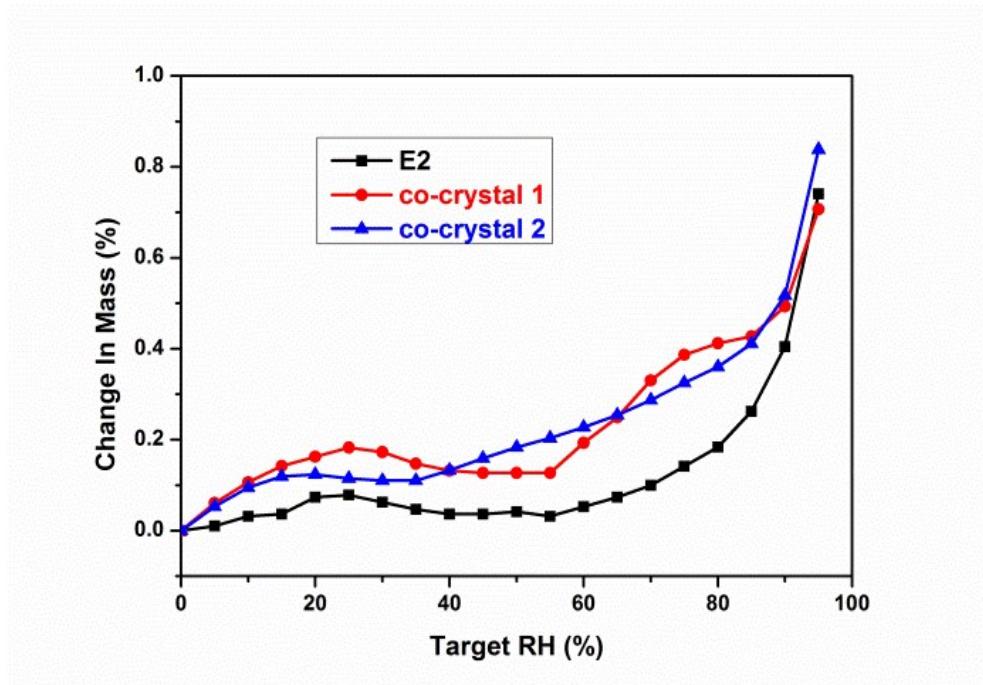


(d)

**Figure S2.** TGA and DSC thermograms of E2, co-former and its co-crystals (a) TGA of co-crystal **1**, (b) TGA of co-crystal **2**, (c) DSC of co-crystal **1** and (d) DSC of co-crystal **2**



**Figure S3.** PXRD patterns of co-crystals **1** and **2** after slurry



**Figure S4.** DVS profiles of E2 co-crystals