

Supplementary information

Table S1. Structural parameters for $\text{Li}_6\text{CuB}_4\text{O}_{10}$ deduced from the Rietveld refinement of the synchrotron X-Ray diffraction pattern at 300 K.

Space Group <i>P</i> -1					
$a=4.821340(3) \text{ \AA}$, $b=9.234549(7) \text{ \AA}$, $c=13.964087(11) \text{ \AA}$, $\alpha=104.31439(5)^\circ$, $\beta=96.21839(5)^\circ$, $\gamma=94.57957(5)^\circ$					
$V=595.195(1) \text{ \AA}^3$ $Z=3$					
Atom	Wyckoff site	<i>x</i>	<i>y</i>	<i>z</i>	<i>B</i> (\AA^2)
B1	<i>2i</i>	0.0852(19)	0.4463(9)	0.3336(7)	0.39(14)
B2	<i>2i</i>	0.1779(20)	0.6796(10)	0.4822(7)	0.65(16)
B3	<i>2i</i>	0.190(2)	0.6607(10)	0.1849(7)	1.31(19)
B4	<i>2i</i>	0.276(2)	0.8972(10)	0.3284(7)	1.15(19)
B5	<i>2i</i>	0.4190(20)	0.7715(9)	0.9908(7)	0.66(17)
B6	<i>2i</i>	0.4720(19)	0.0004(9)	0.1452(6)	0.20(15)
Cu1	<i>1a</i>	0	0	0	0.55(3)
Cu2	<i>2i</i>	0.3000(2)	0.31614(10)	0.67137(8)	0.43(2)
Li1	<i>2i</i>	0.038(2)	0.8018(12)	0.8149(9)	-0.08(20)
Li2	<i>2i</i>	0.068(4)	0.3786(17)	0.0345(12)	3.2(4)
Li3	<i>2i</i>	0.078(3)	0.0713(14)	0.8176(10)	2.5(3)
Li4	<i>2i</i>	0.221(3)	0.9290(15)	0.6293(10)	1.7(3)
Li5	<i>2i</i>	0.333(3)	0.1367(12)	0.4883(9)	0.3(2)
Li6	<i>2i</i>	0.368(3)	0.4314(12)	0.9148(9)	1.0(2)
Li7	<i>2i</i>	0.370(2)	0.1719(12)	0.3115(9)	0.9(2)
Li8	<i>2i</i>	0.414(3)	0.5709(13)	0.6101(9)	1.8(3)
Li9	<i>2i</i>	0.475(3)	0.3997(14)	0.1973(9)	1.9(3)
O1	<i>2i</i>	0.0246(10)	0.5885(5)	0.0998(4)	0.55(10)
O2	<i>2i</i>	0.0668(10)	0.2575(5)	0.5419(3)	0.60(10)
O3	<i>2i</i>	0.1071(10)	0.7874(5)	0.2506(3)	0.26(9)
O4	<i>2i</i>	0.1604(11)	0.8056(5)	0.9530(4)	0.74(10)
O5	<i>2i</i>	0.1579(10)	0.0242(5)	0.3609(3)	0.26(9)
O6	<i>2i</i>	0.1917(10)	0.5242(5)	0.6976(4)	0.47(9)
O7	<i>2i</i>	0.1846(9)	0.0295(5)	0.1315(3)	0.13(8)
O8	<i>2i</i>	0.2232(10)	0.3413(5)	0.2815(3)	0.40(9)
O9	<i>2i</i>	0.2392(11)	0.5498(5)	0.4203(4)	0.55(10)
O10	<i>2i</i>	0.3505(10)	0.7487(5)	0.5606(4)	0.17(9)

O11	$2i$	0.3435(11)	0.9223(5)	0.7707(4)	0.22(10)
O12	$2i$	0.4361(10)	0.1228(5)	0.9209(3)	0.03(9)
O13	$2i$	0.4692(11)	0.1276(5)	0.6251(4)	0.48(9)
O14	$2i$	0.4751(11)	0.3536(5)	0.0435(4)	0.97(12)
O15	$2i$	0.4624(10)	0.6166(5)	0.2042(3)	0.24(8)

Table S2. Structural parameters for $\text{Li}_6\text{CuB}_4\text{O}_{10}$ deduced from the DFT calculations.

	a (Å)	b (Å)	c (Å)	α (°)	β (°)	γ (°)	V (Å ³)
Pristine DFT + U	4.8849 (1.2%)	9.3465 (1.3%)	14.1005 (0.7%)	104.789 (0.4%)	97.1720 (0.9%)	93.6110 (0.9%)	614.56 (2.9%)

Tab S3: Lattice parameters and the corresponding voltage of $\text{Li}_6\text{CuB}_4\text{O}_{10}$ during galvanostatic charge and following discharge.

	a (Å)	b (Å)	c (Å)	α (°)	β (°)	γ (°)	V (Å ³)
Charged 4.5V	4.837(5)	9.119(5)	14.088(5)	104.082(5)	96.936(5)	94.115(5)	595.052(5)
Discharged 1.0V	4.829(5)	9.251(5)	14.021(5)	104.36(5)	96.538(5)	94.472(5)	599.244(5)

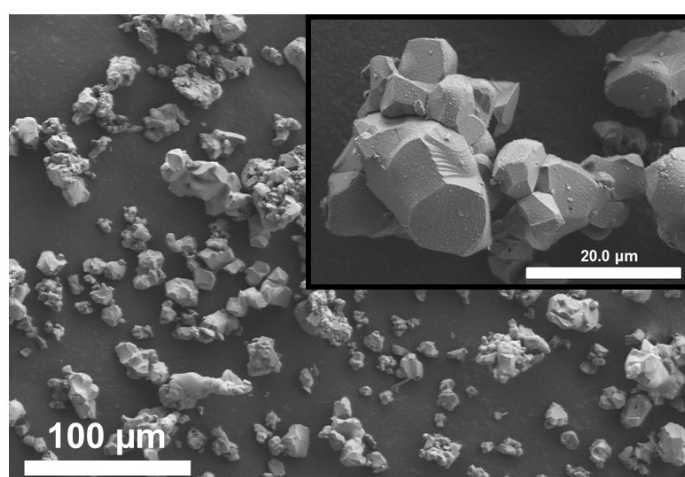


Figure S1: Representative SEM image of $\text{Li}_6\text{CuB}_4\text{O}_{10}$ particles after solid state synthesis.

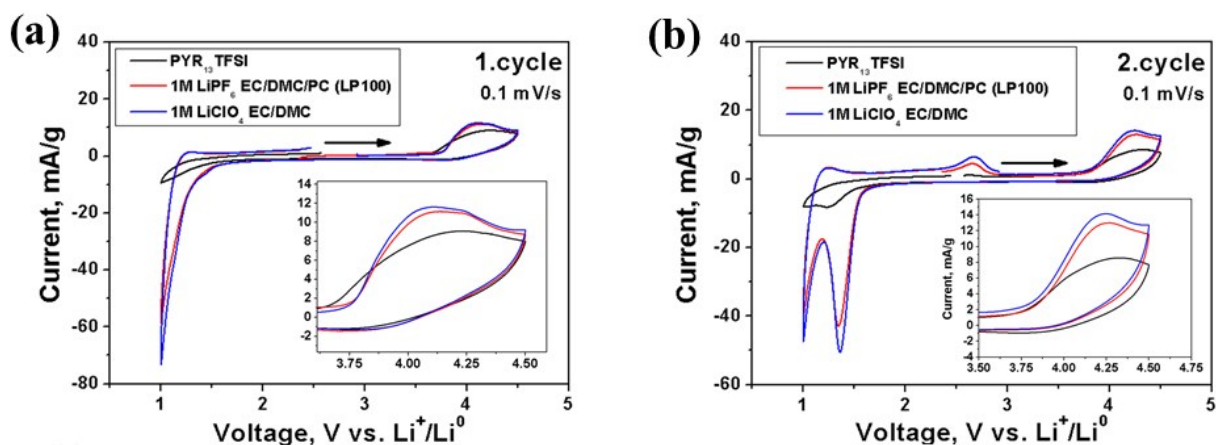


Figure S2: Cyclic voltammetry of Li/Li₆CuB₄O₁₀ half cells in different electrolytes in a potential window from 4.5-1.0 V vs. Li⁺/Li⁰ with a scan speed of 0.1 mV/s for the first (a) and second (b) cycle. Inset show the differences in anodic current due to electrolyte oxidation.

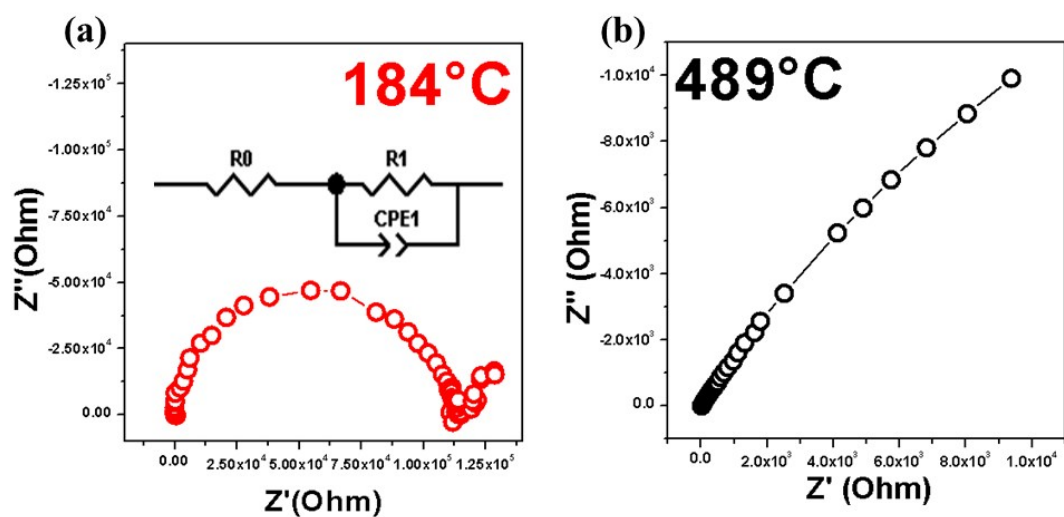


Figure S3: Complex a.c. impedance spectra for Li₆CuB₄O₁₀ measured at 184°C (a) and at 489°C (b).

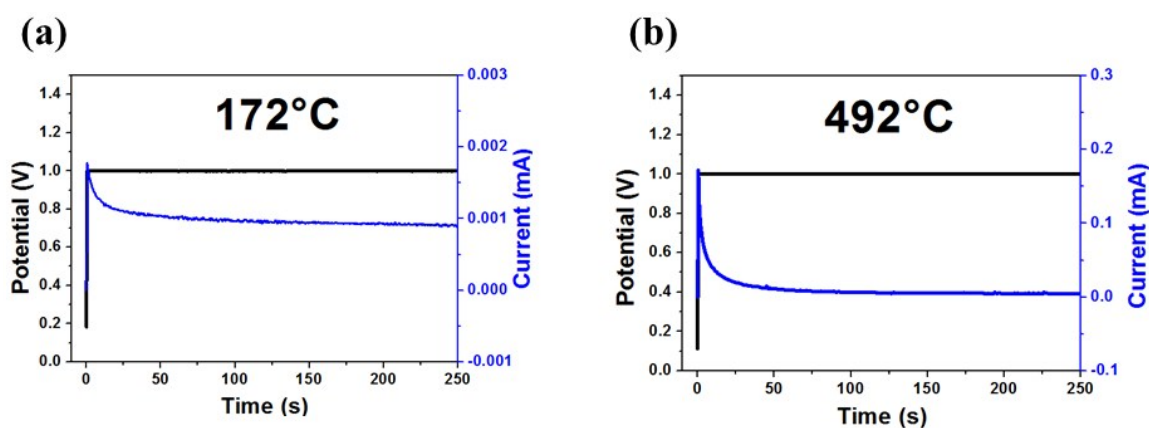


Figure S4: D.c. polarization measurements for Li₆CuB₄O₁₀ measured at 172°C (a) and at 492°C (b).