

Supporting Information:

Structurally versatile phosphine and amine donors constructed from N-heterocyclic olefin units

Nathan R. Paisley, Melanie W. Lui, Robert McDonald, Michael J. Ferguson, Eric Rivard*

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Figure S1. ^1H NMR spectrum of (IPr=CH)P*i*Pr₂ (**2**) in C₆D₆.

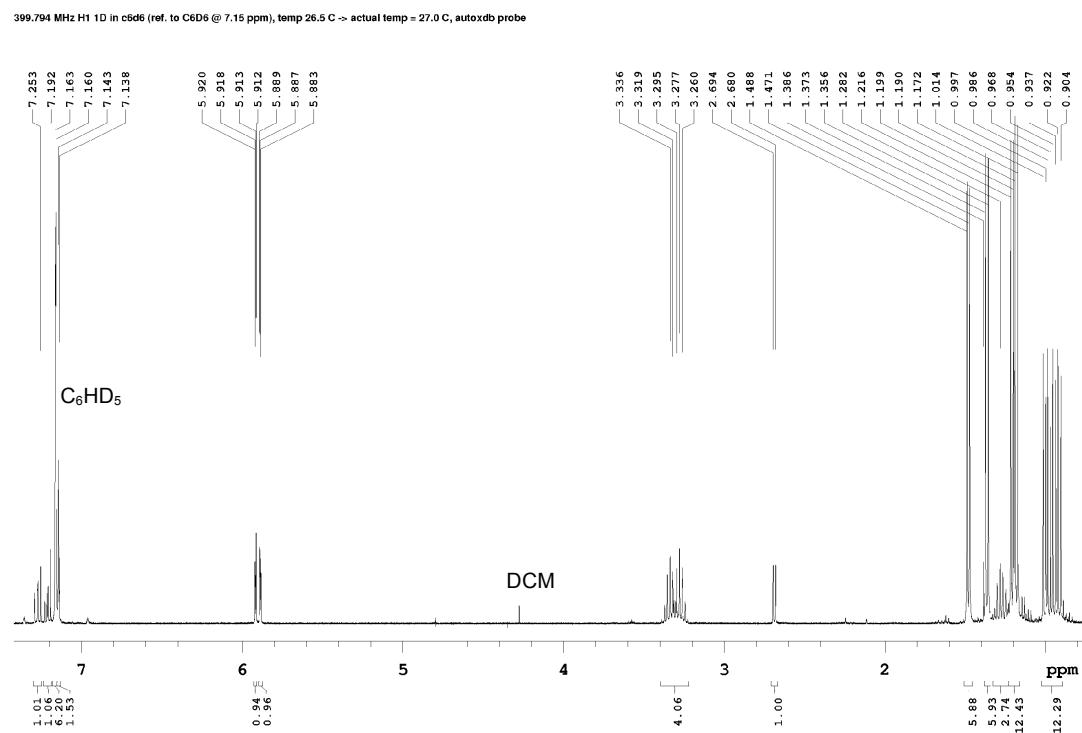


Figure S2. $^{13}\text{C}\{\text{H}\}$ DEPTQ NMR spectrum of (IPr=CH)P*i*Pr₂ (**2**) in C₆D₆.

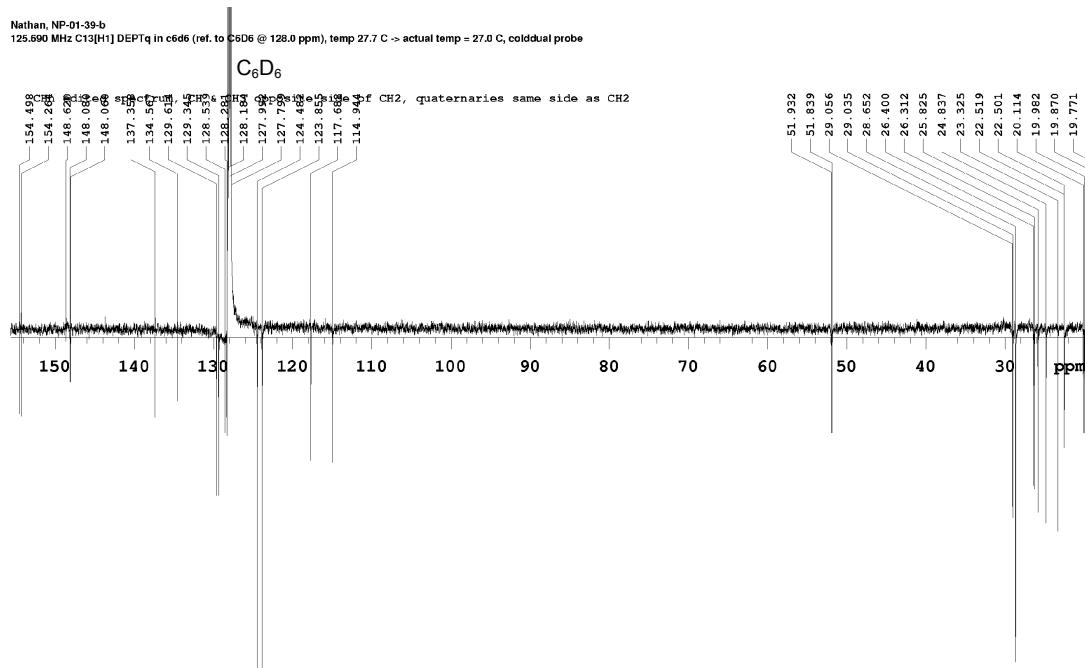


Figure S3. $^{31}\text{P}\{\text{H}\}$ NMR spectrum of (IPr=CH)P i Pr₂ (**2**) in C₆D₆.

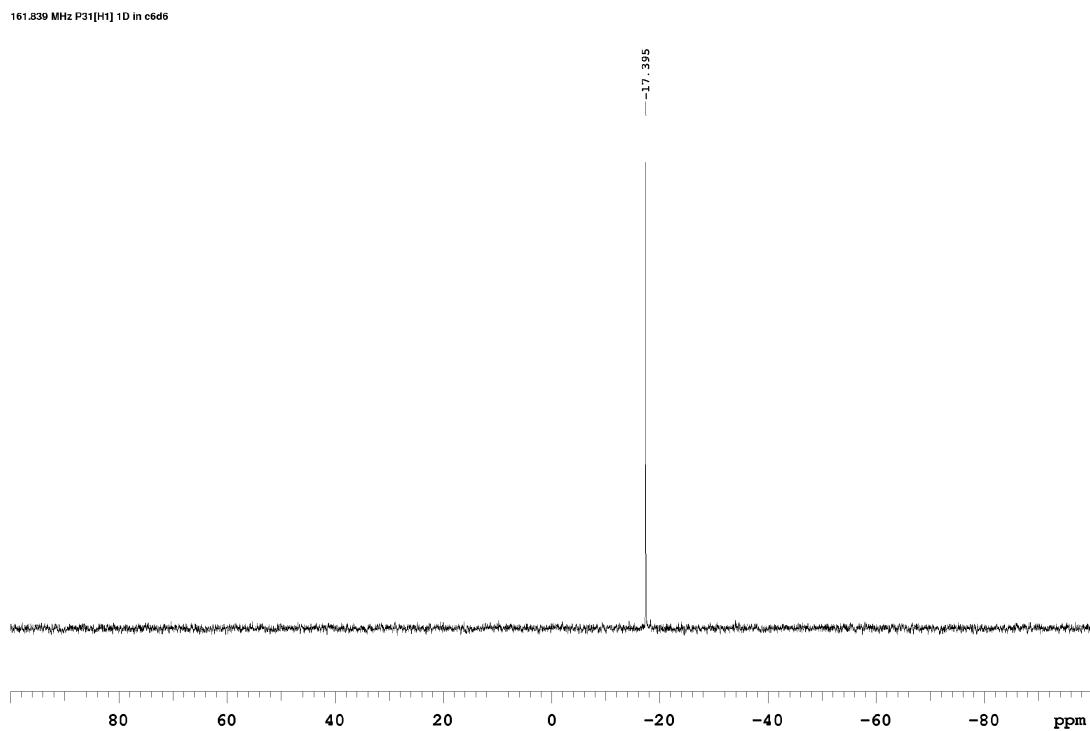


Figure S4. ^1H NMR spectrum of (IPr=CH)PPh₂ (**3**) in C₆D₆.

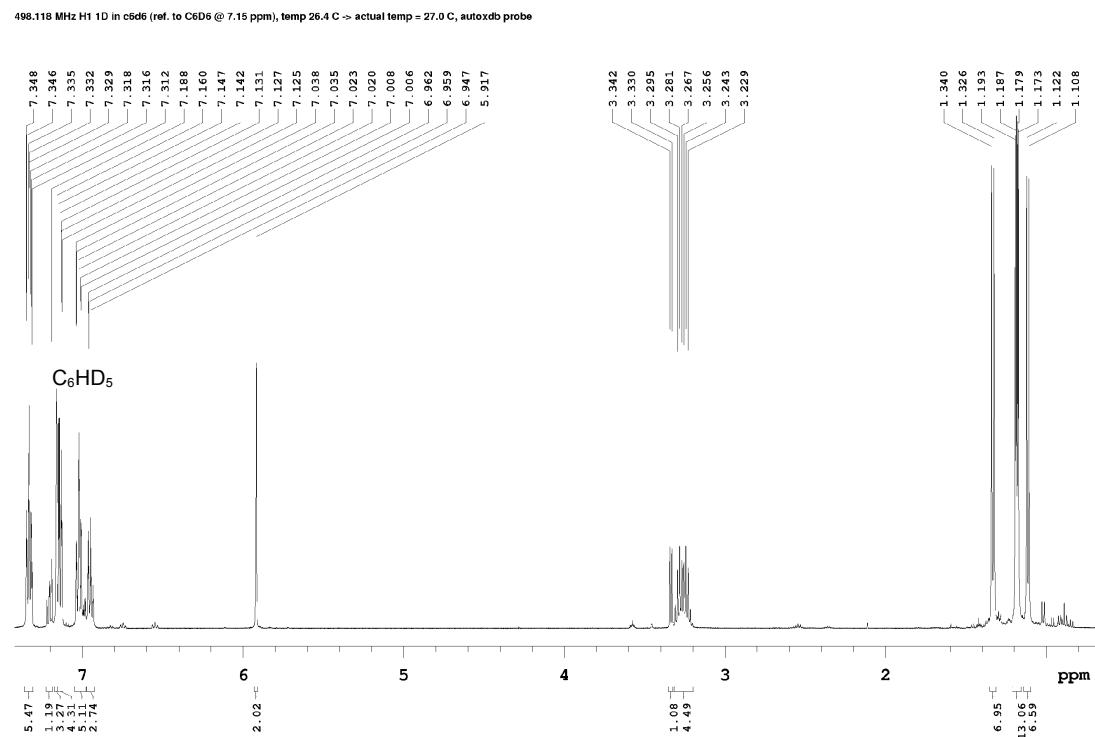


Figure S5. $^{13}\text{C}\{\text{H}\}$ DEPTQ NMR spectrum of (IPr=CH)PPh₂ (**3**) in C₆D₆.

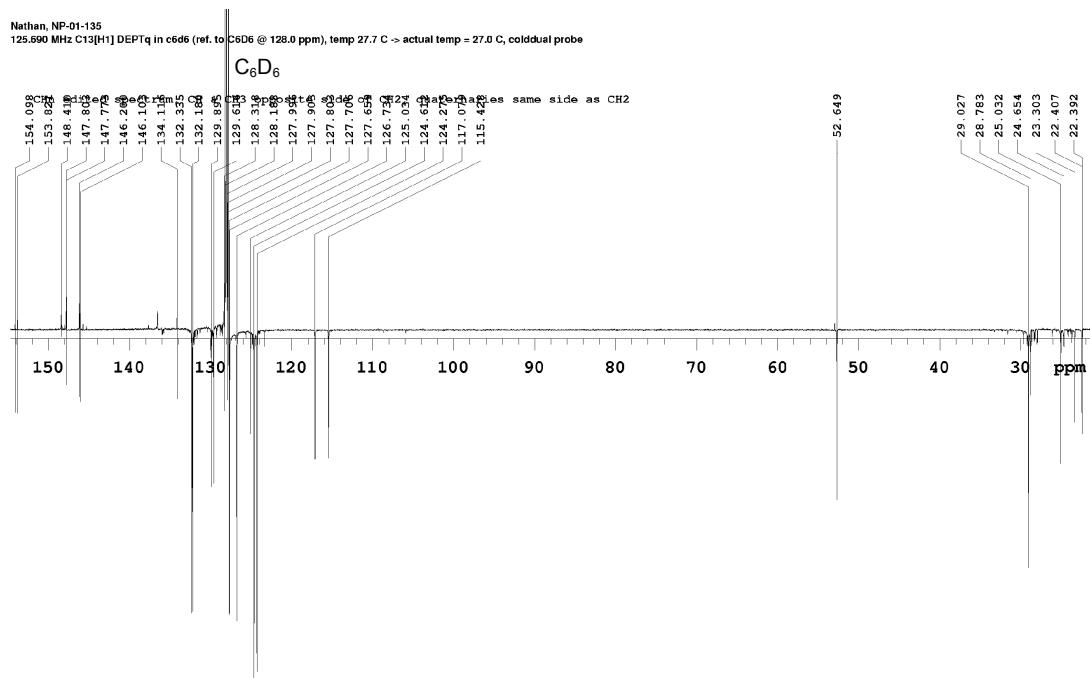


Figure S6. $^{31}\text{P}\{\text{H}\}$ NMR spectrum of (IPr=CH)PPh₂ (**3**) in C₆D₆.

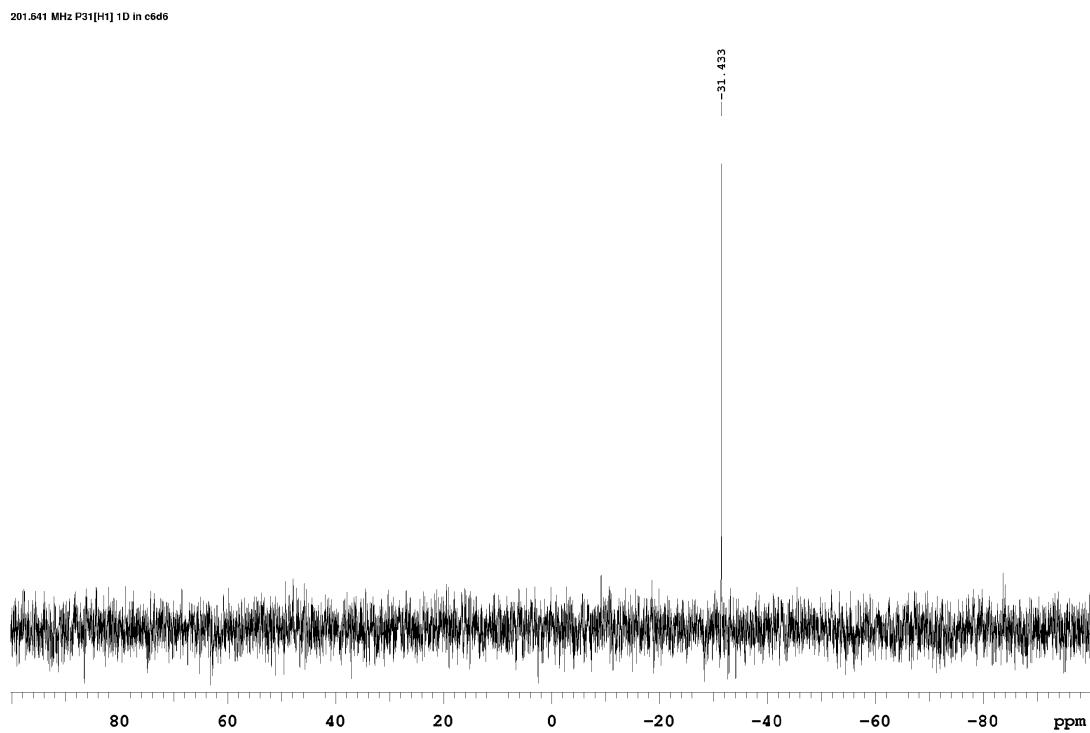


Figure S7. ^1H NMR spectrum of (IPr=CH)NMe₂ (**4**) in C₆D₆.

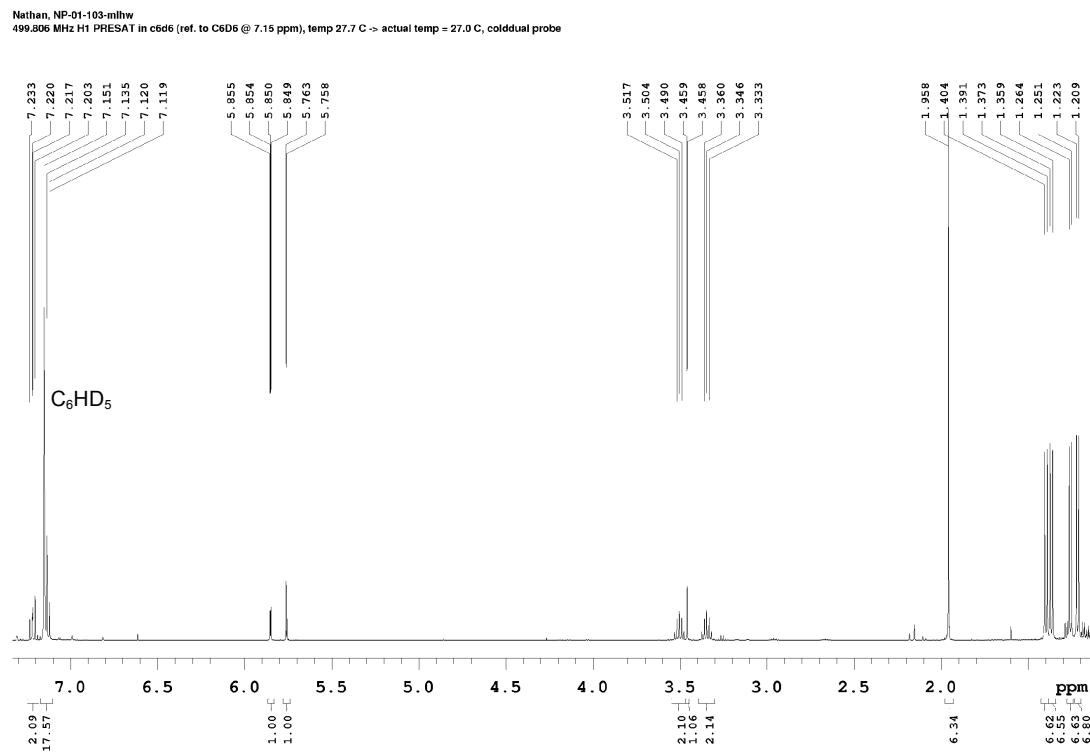


Figure S8. $^{13}\text{C}\{\text{H}\}$ DEPTQ NMR spectrum of (IPr=CH)NMe₂ (**4**) in C₆D₆.

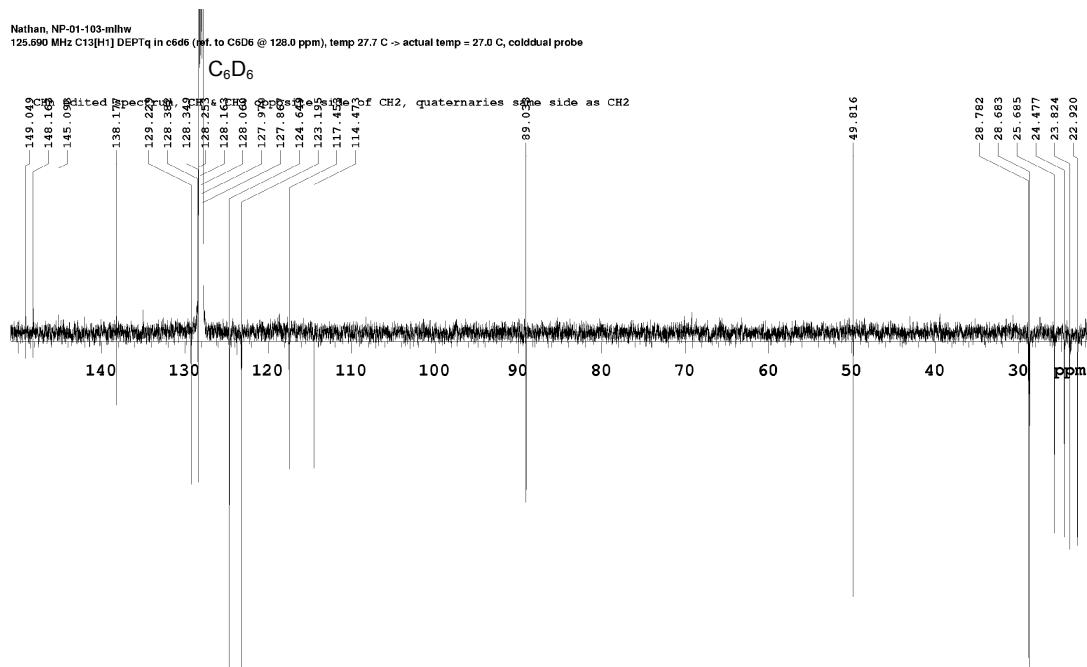


Figure S9. ^1H NMR spectrum of $(\text{IPr}=\text{CH})^i\text{Pr}_2\text{P}\bullet\text{BH}_3$ (**5**) in C_6D_6 .

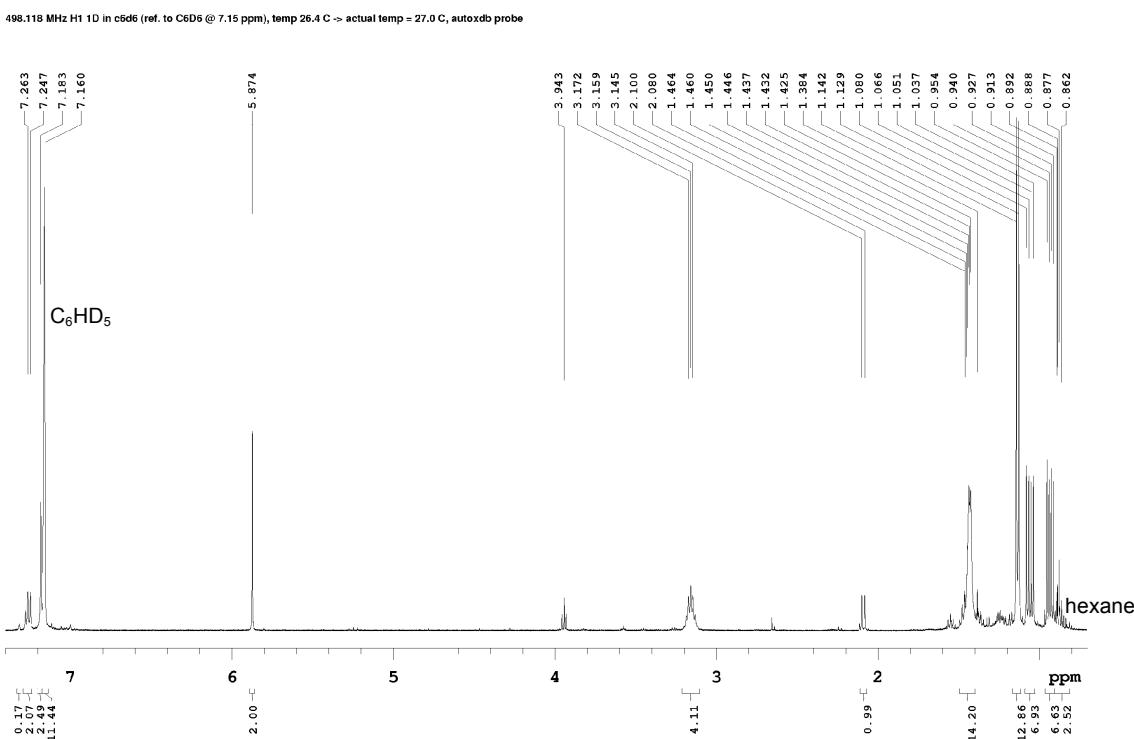


Figure S10. $^{11}\text{B}\{\text{H}\}$ NMR spectrum of $(\text{IPr}=\text{CH})^i\text{Pr}_2\text{P}\bullet\text{BH}_3$ (**5**) in C_6D_6 .

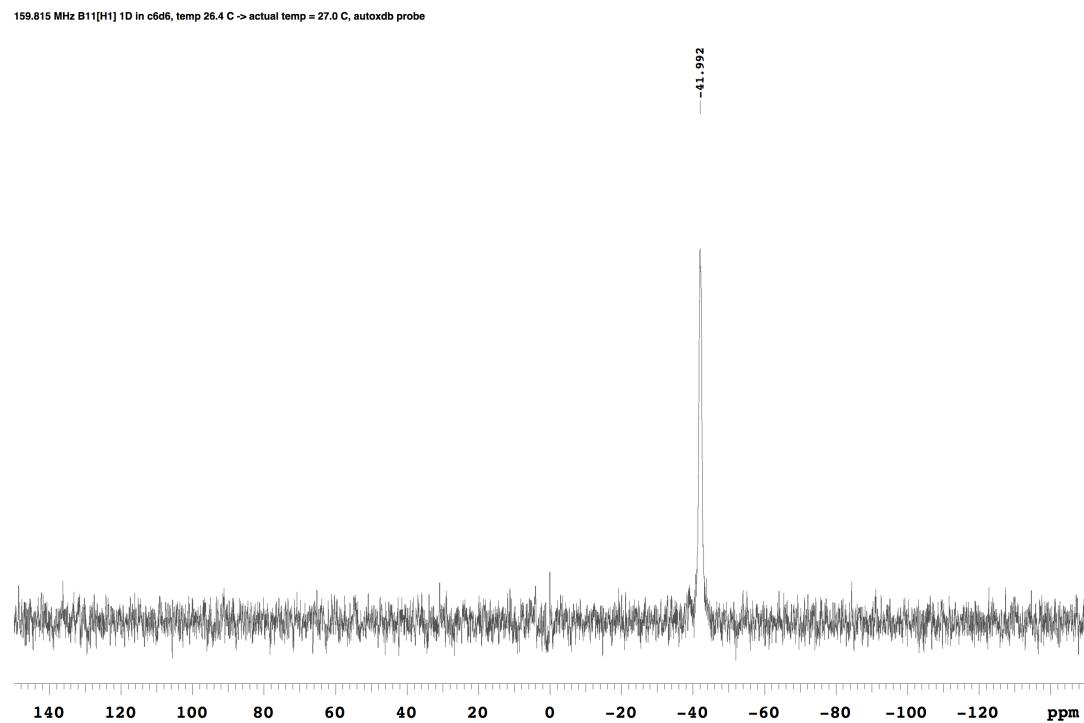


Figure S11. $^{13}\text{C}\{\text{H}\}$ DEPTQ NMR spectrum of $(\text{IPr}=\text{CH})^i\text{Pr}_2\text{P}\bullet\text{BH}_3$ (**5**) in C_6D_6 .

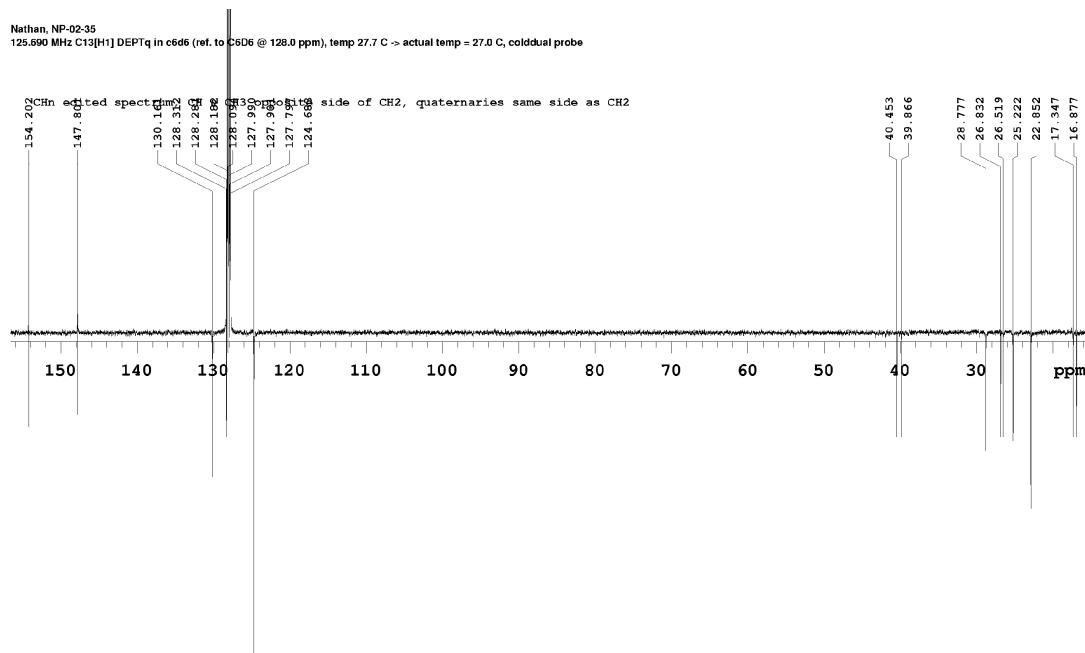


Figure S12. $^{31}\text{P}\{\text{H}\}$ NMR spectrum of $(\text{IPr}=\text{CH})^i\text{Pr}_2\text{P}\bullet\text{BH}_3$ (**5**) in C_6D_6 .

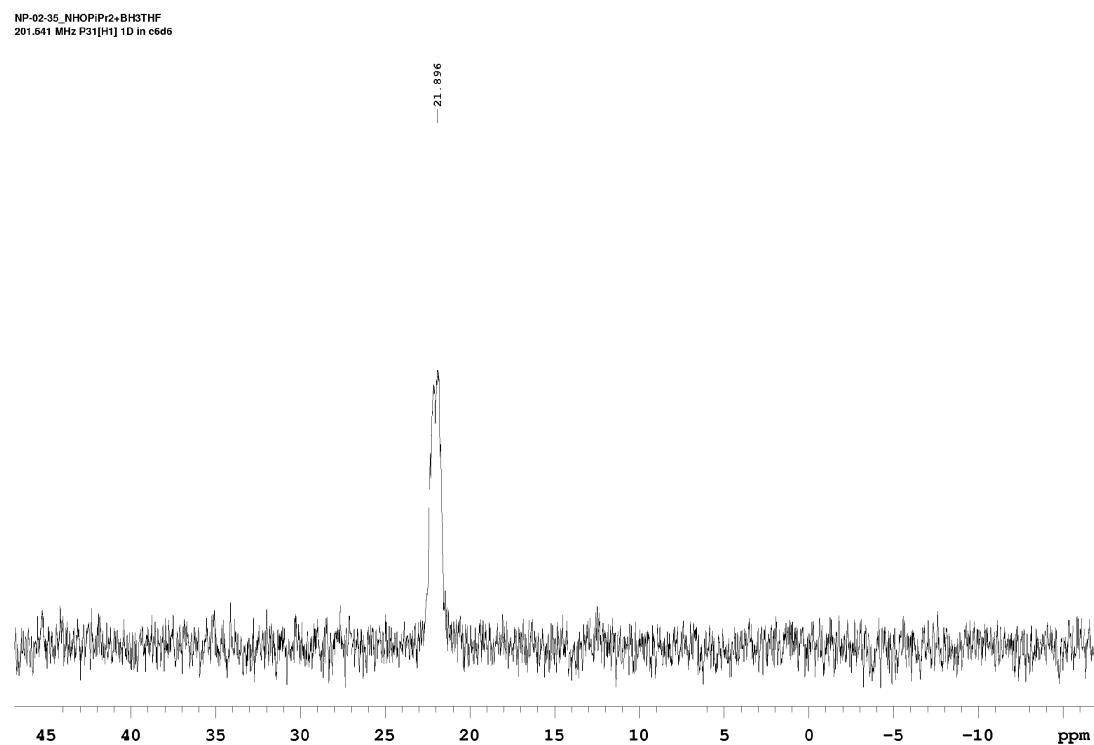


Figure S13. ^1H NMR spectrum of (IPr=CH)Ph₂P•BH₃ (**6**) in C₆D₆.

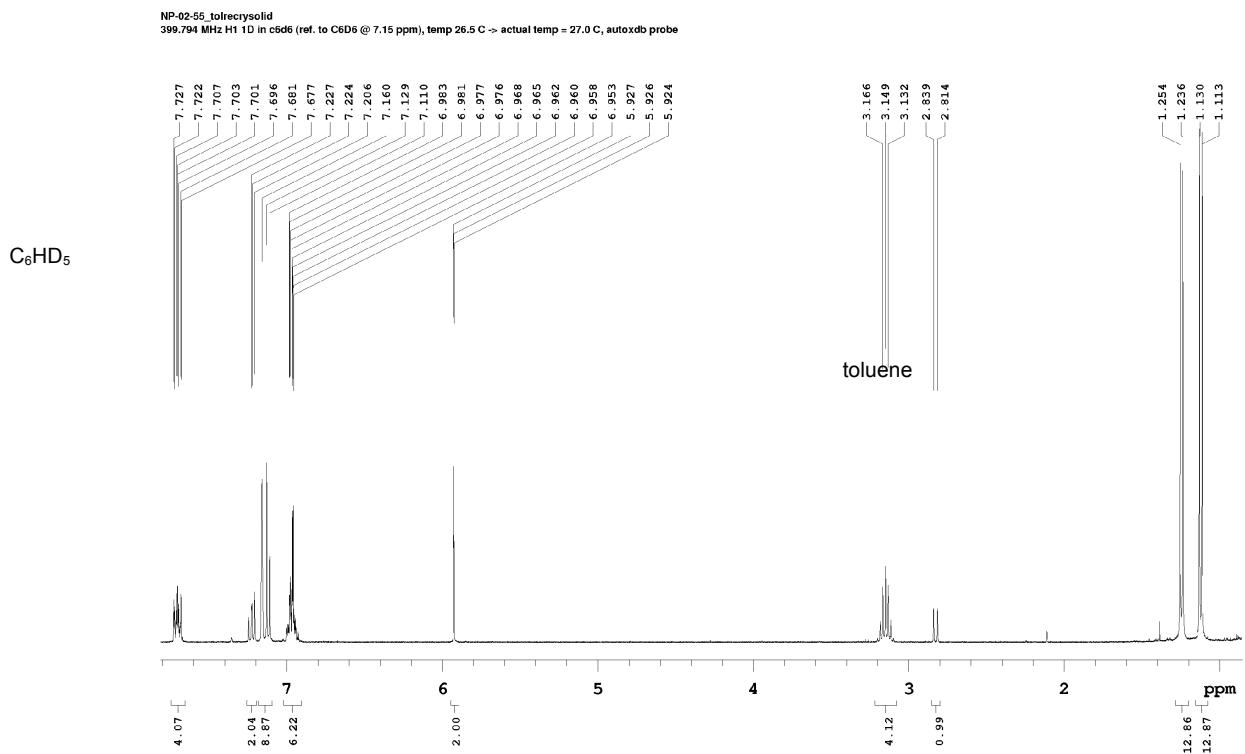


Figure S14. $^{11}\text{B}\{\text{H}\}$ NMR spectrum of (IPr=CH)Ph₂P•BH₃ (**6**) in C₆D₆.

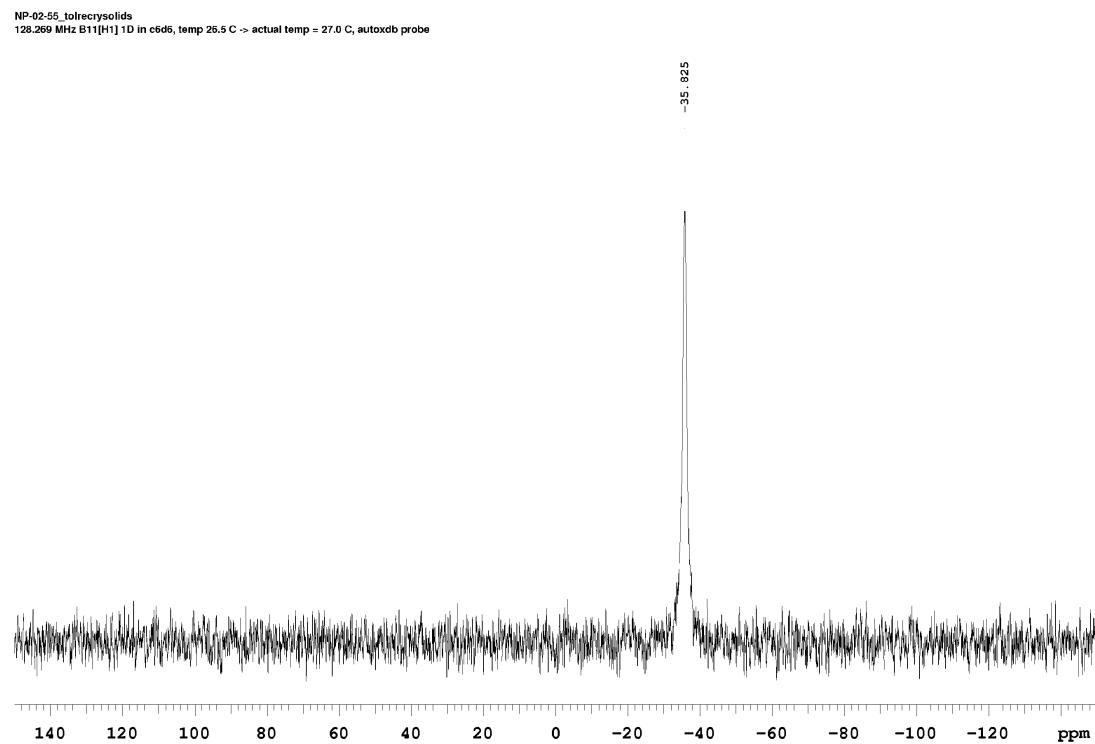


Figure S15. ¹³C{¹H} DEPTQ NMR spectrum of (IPr=CH)Ph₂P•BH₃ (**6**) in C₆D₆.

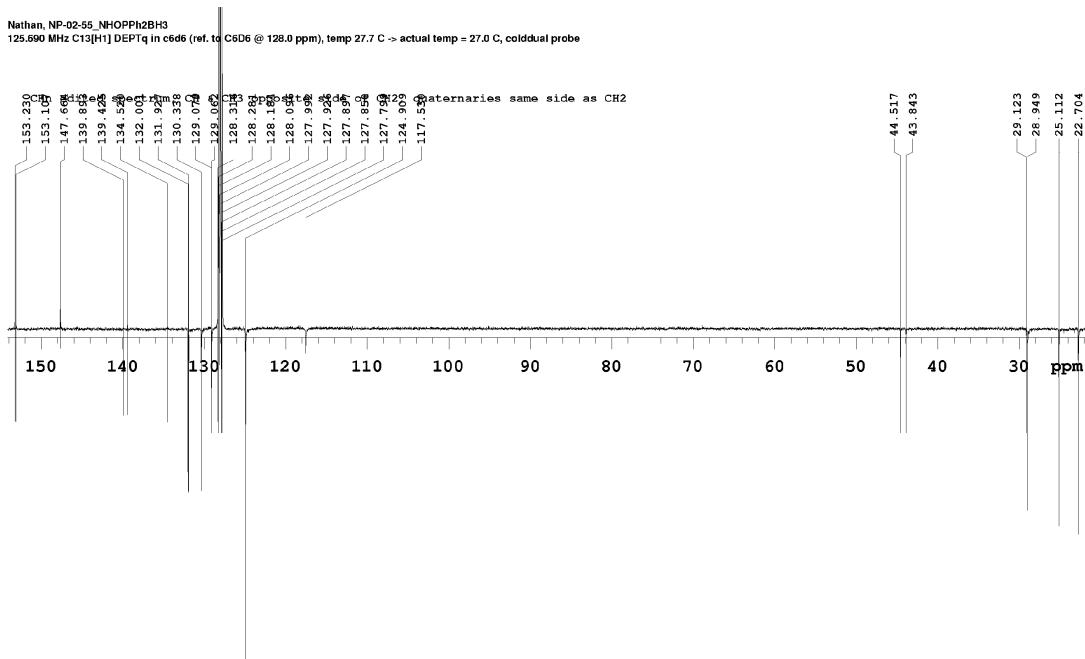


Figure S16. $^{31}\text{P}\{\text{H}\}$ NMR spectrum of (IPr=CH)Ph₂P•BH₃ (**6**) in C₆D₆.

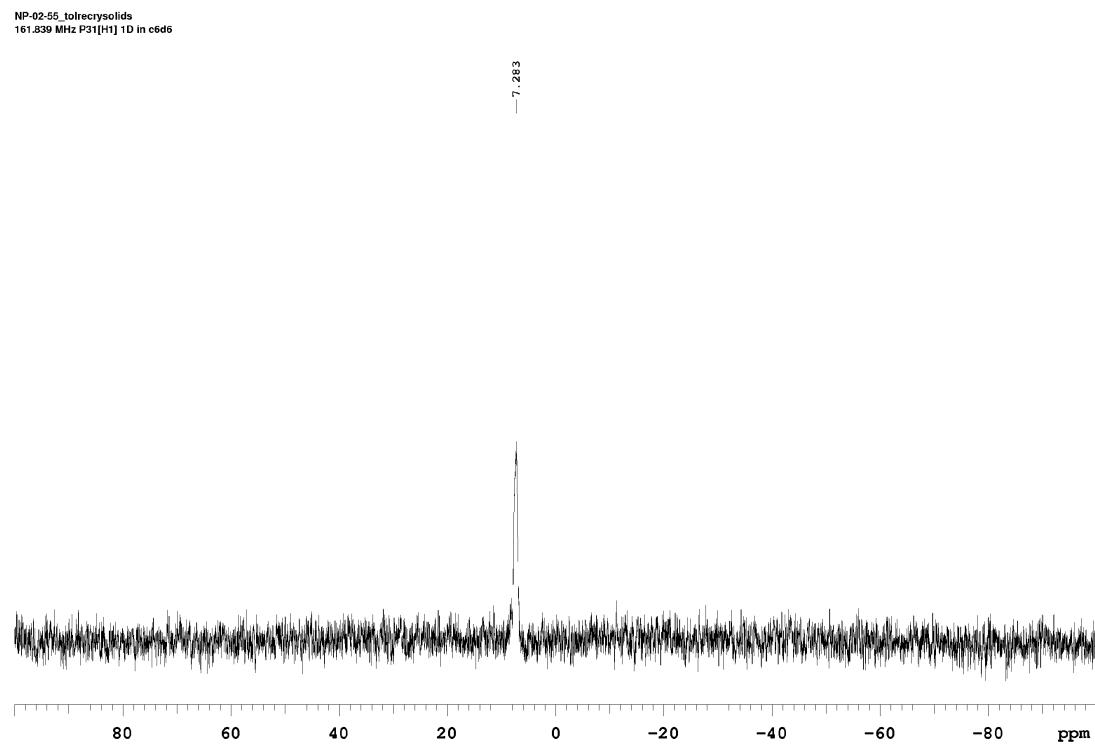


Figure S17. ^1H NMR spectrum of $(\text{IPr}=\text{CH})^i\text{Pr}_2\text{P}\bullet\text{PdCl}(\text{cinnamyl})$ (**7**) in C_6D_6 .

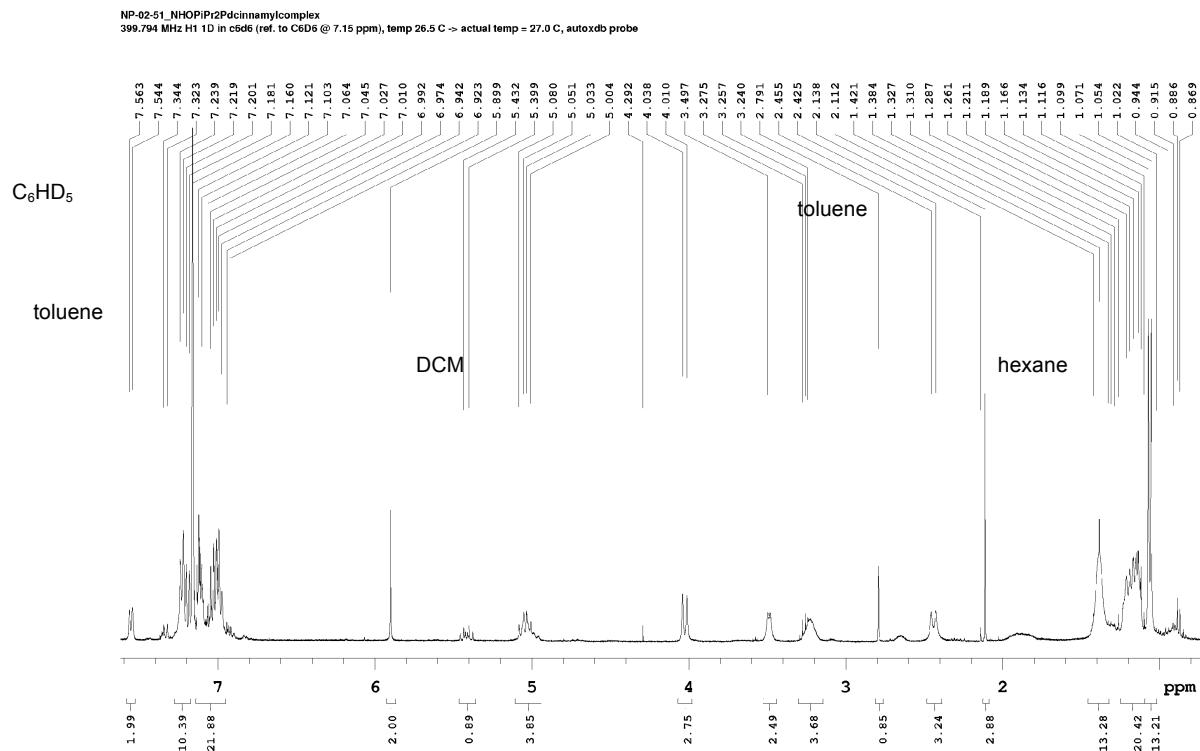
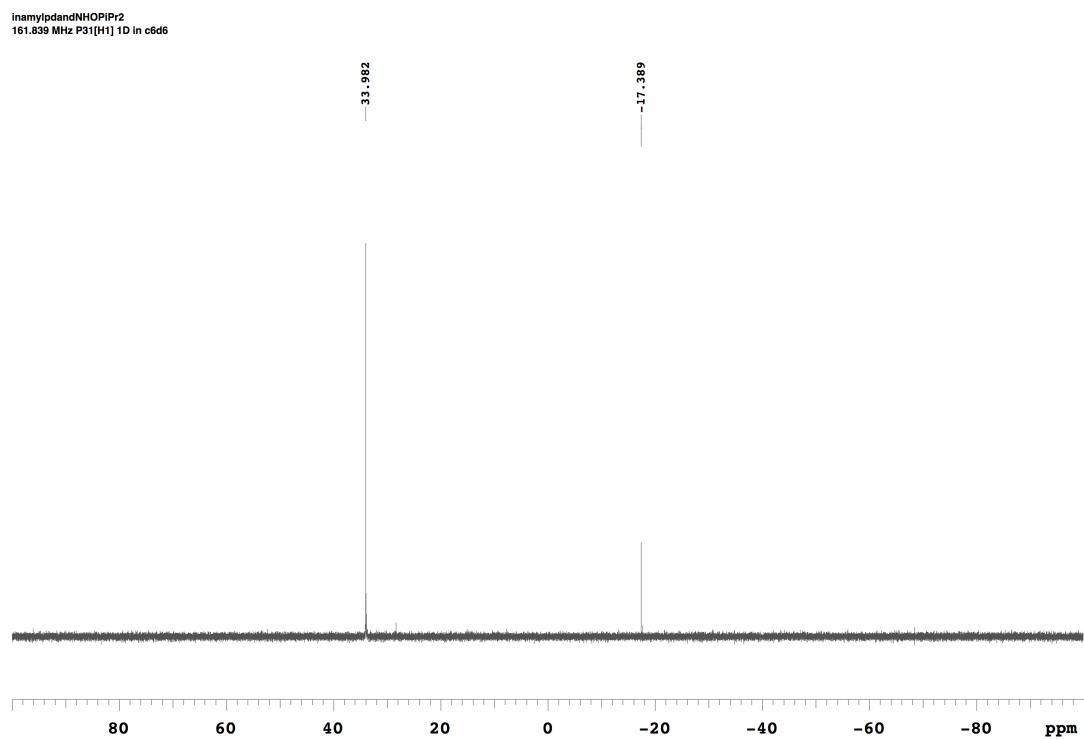


Figure S18. $^{31}\text{P}\{\text{H}\}$ NMR spectrum of $(\text{IPr}=\text{CH})^i\text{Pr}_2\text{P}\bullet\text{PdCl}(\text{cinnamyl})$ (**7**) in C_6D_6 .



* impurity at -17.4 ppm is free ligand $(\text{IPr}=\text{CH})\text{P}^i\text{Pr}_2$ (**2**)

Figure S19. ^1H NMR spectrum of ($\text{IPr}=\text{CH}$) $^i\text{Pr}_2\text{P}\bullet\text{AuCl}$ (**8**) in C_6D_6 .

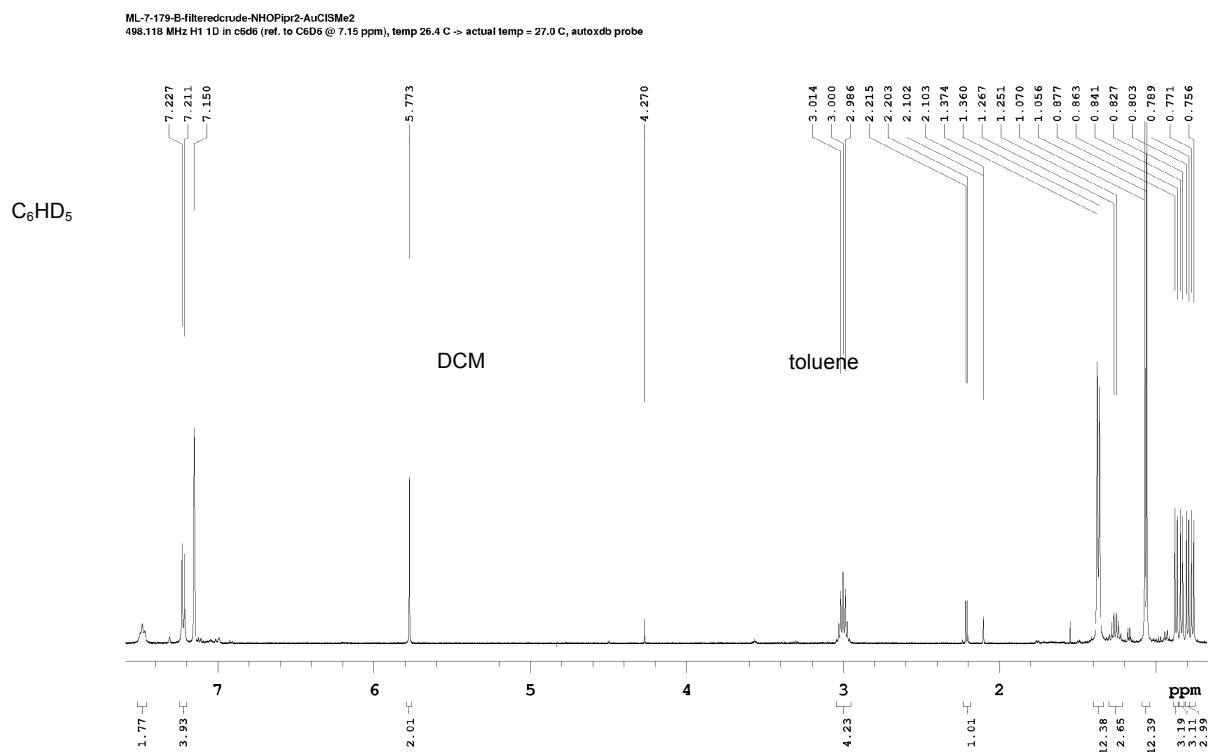


Figure S20. $^{13}\text{C}\{\text{H}\}$ NMR spectrum of $(\text{IPr}=\text{CH})^i\text{Pr}_2\text{P}\bullet\text{AuCl}$ (**8**) in C_6D_6 .

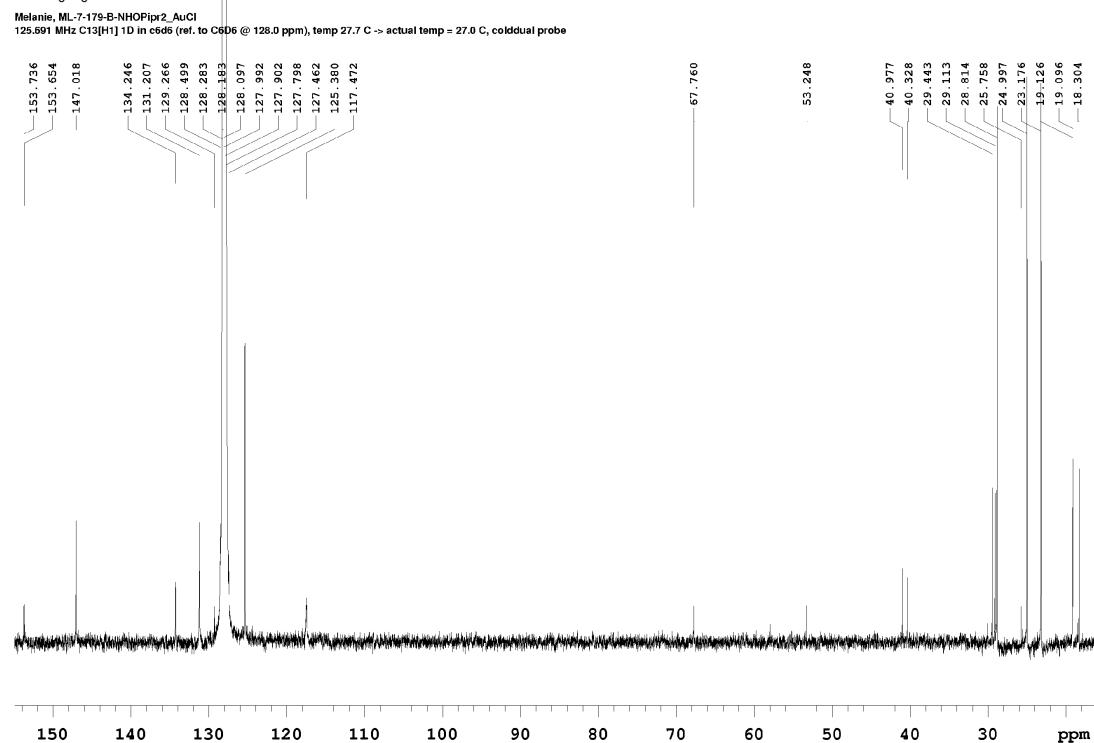


Figure S21. $^{31}\text{P}\{\text{H}\}$ NMR spectrum of $(\text{IPr}=\text{CH})^i\text{Pr}_2\text{P}\bullet\text{AuCl}$ (**8**) in C_6D_6 .

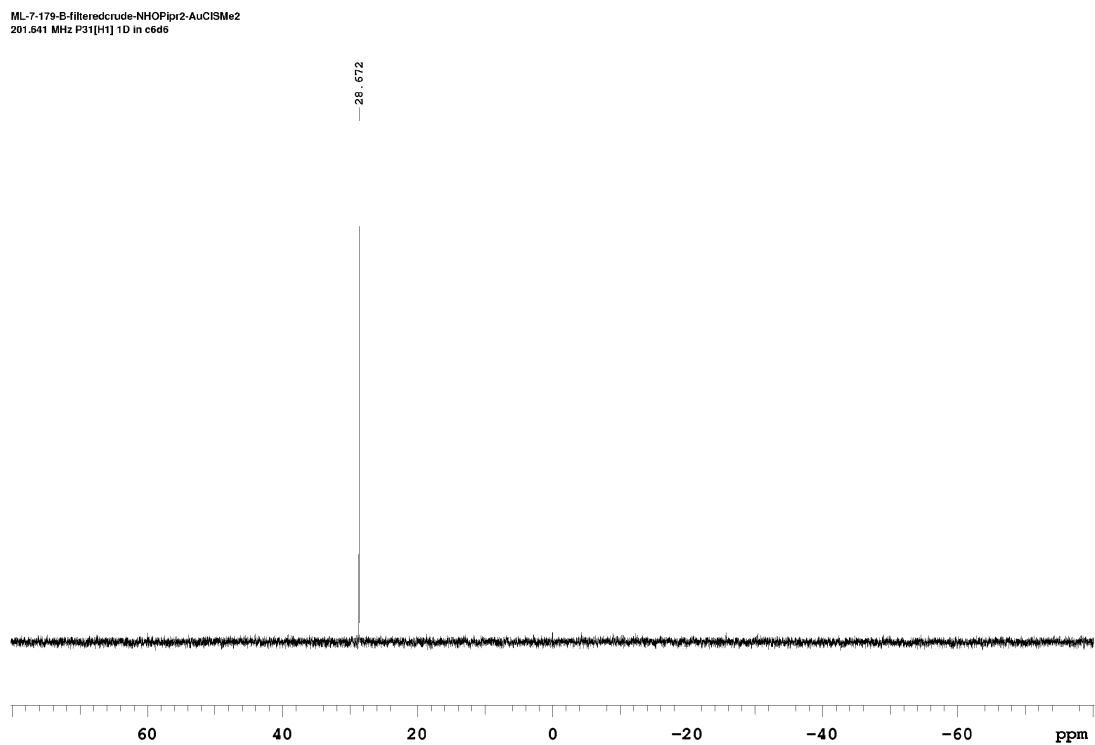


Figure S22. ^1H NMR spectrum of ($\text{IPr}=\text{CH}$) $\text{Ph}_2\text{P}\bullet\text{AuCl}$ (**9**) in C_6D_6 .

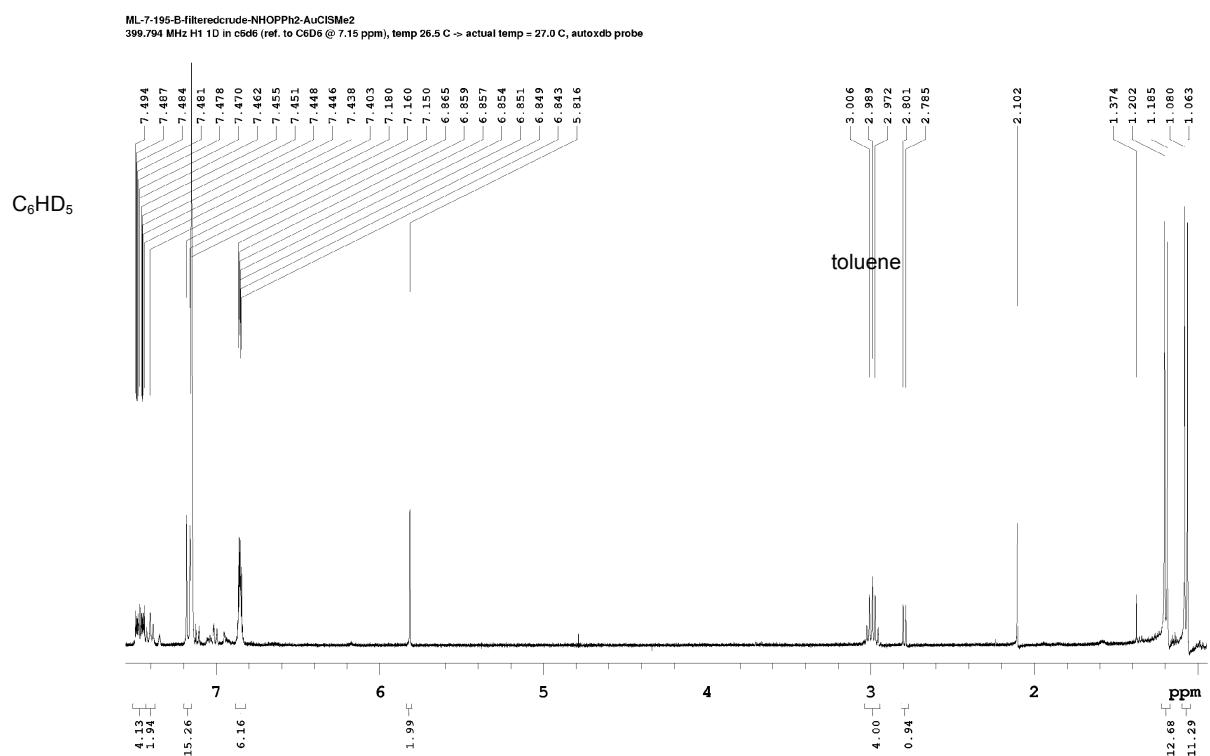


Figure S23. $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum of (IPr=CH)Ph₂P•AuCl (**9**) in C₆D₆.

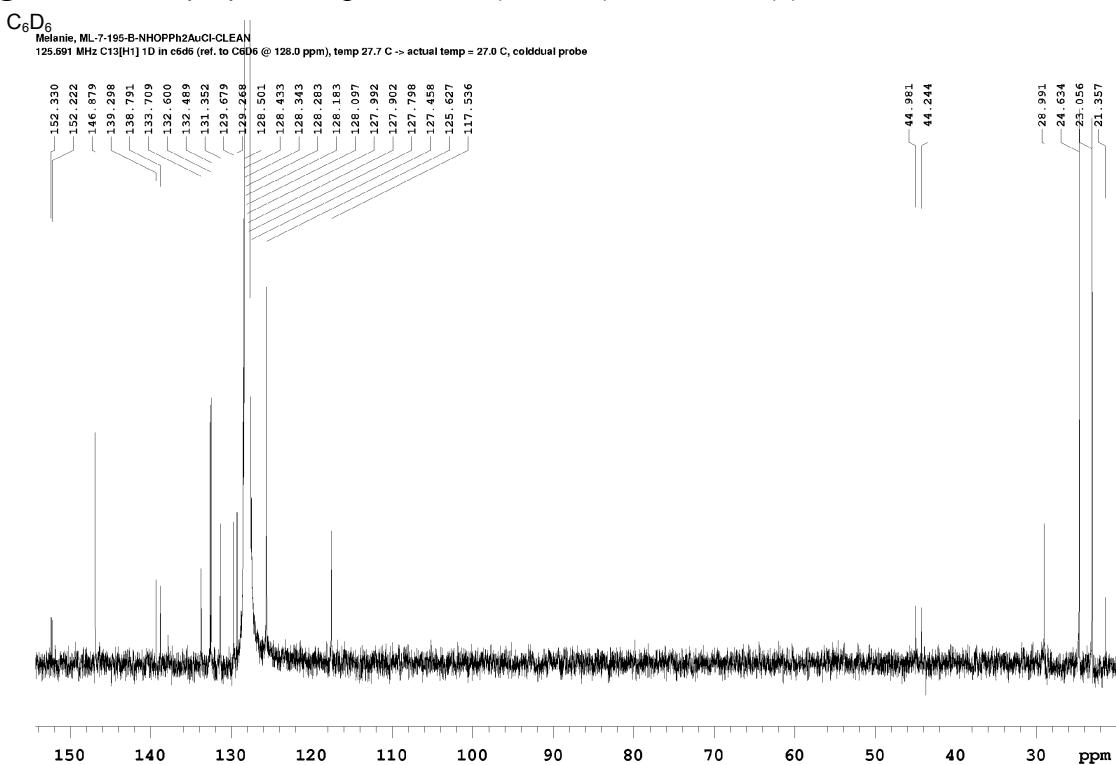


Figure S24. $^{31}\text{P}\{\text{H}\}$ NMR spectrum of (IPr=CH)Ph₂P•AuCl (**9**) in C₆D₆.

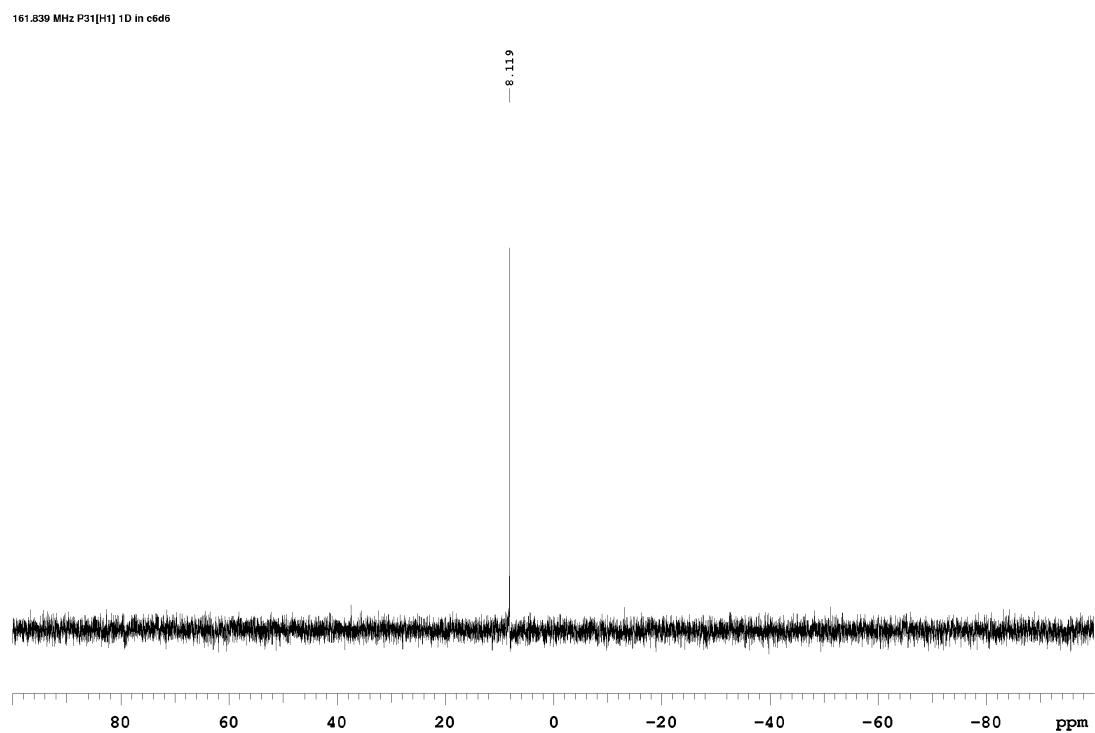


Figure S25. ^1H NMR spectrum of [IPr-CH₂-PPh₂•AuAr^F][BAr^F₄] (**10**) in CDCl₃.

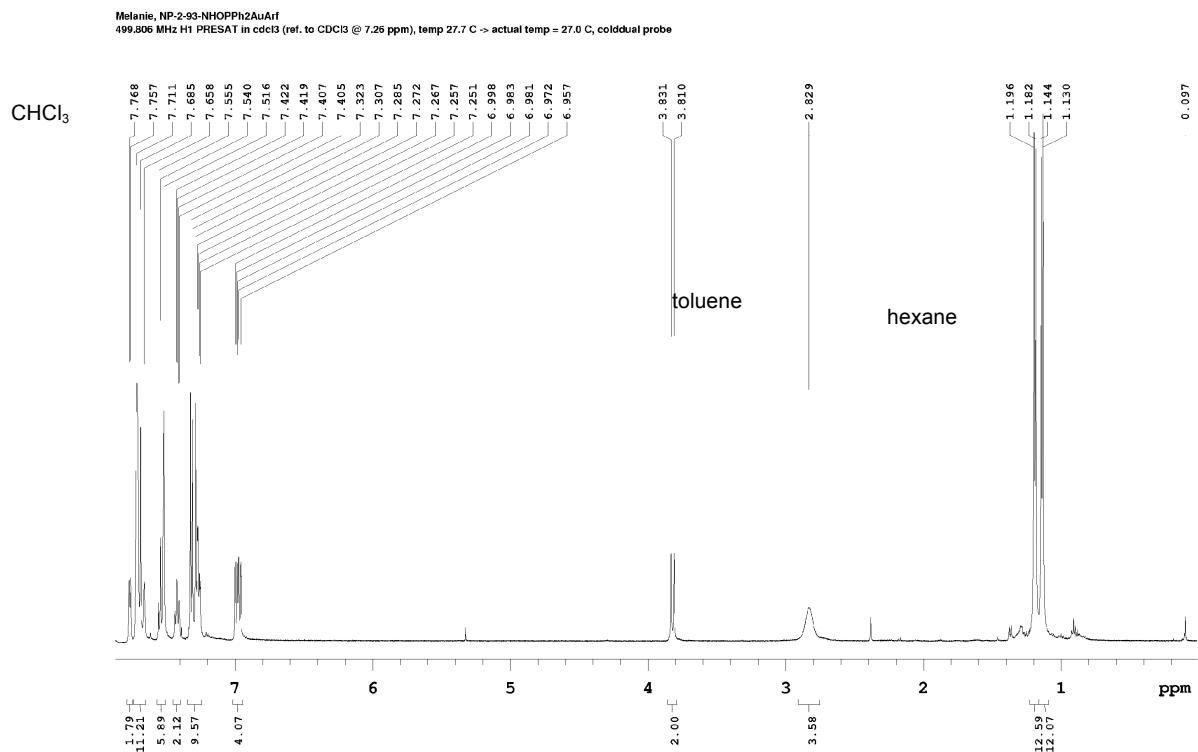


Figure S26. $^{11}\text{B}\{\text{H}\}$ NMR spectrum of [IPr-CH₂-PPh₂•AuAr^F][BAr^F₄] (**10**) in CDCl₃.

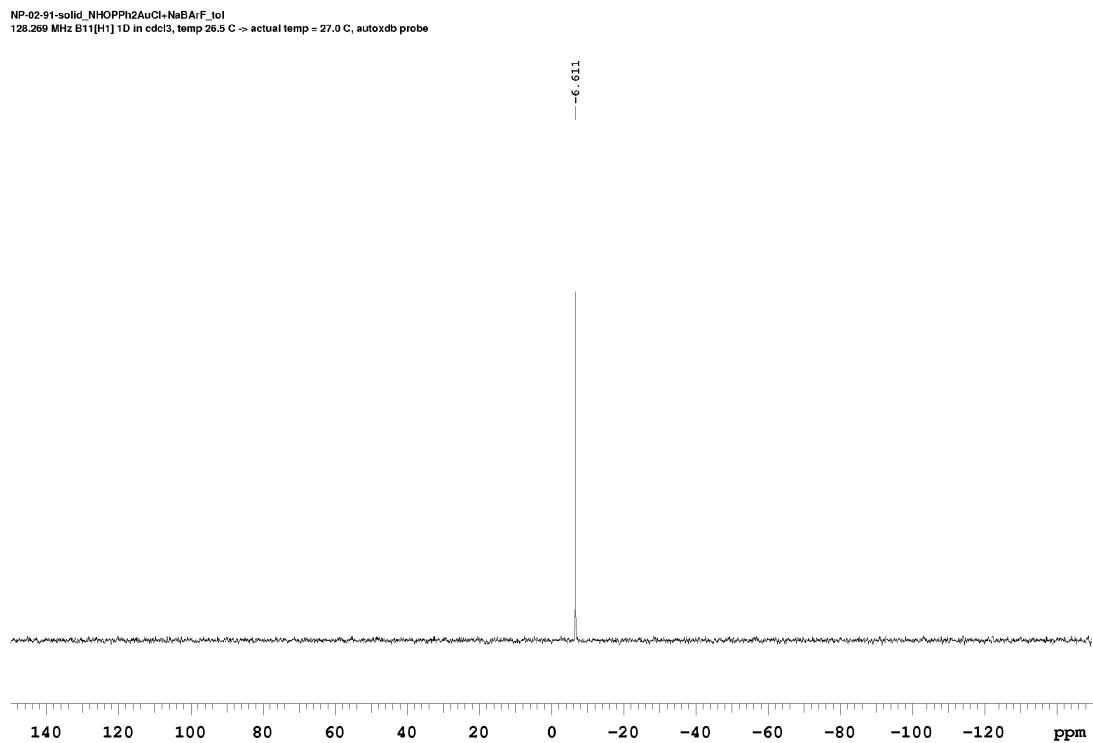


Figure S27. $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum of [IPr- CH_2 -PPh₂•AuAr^F][BAr^F₄] (**10**) in CDCl₃.

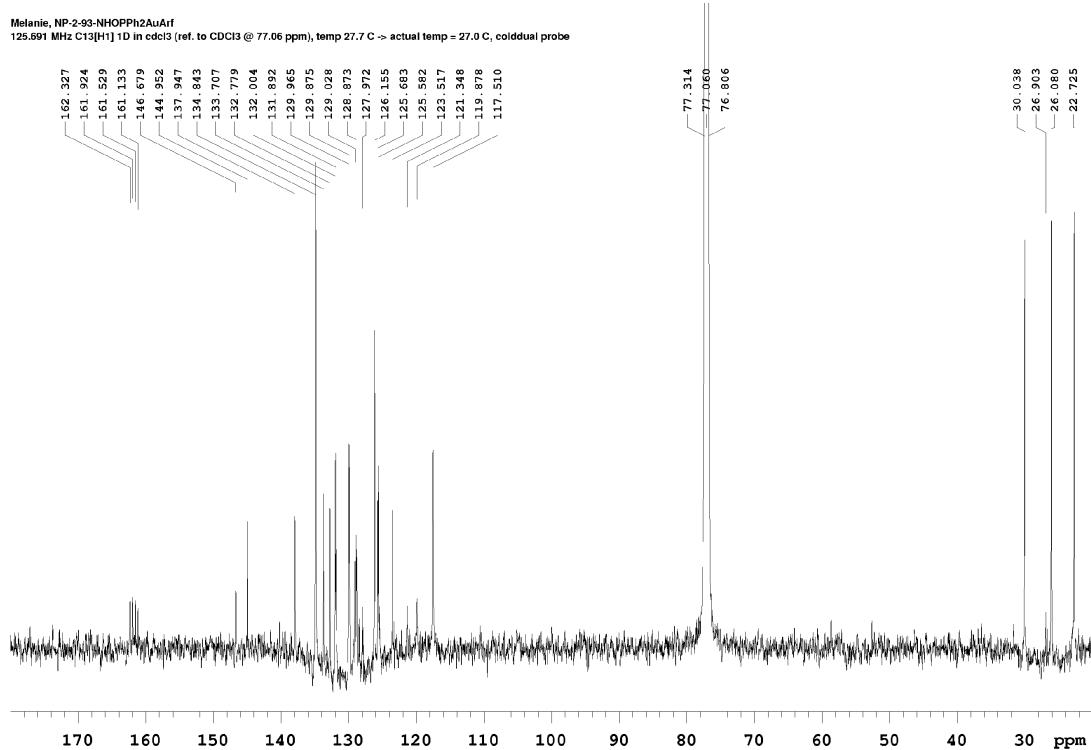


Figure S28. ^{19}F NMR spectrum of [IPr-CH₂-PPh₂•AuAr^F][BAr^F₄] (**10**) in CDCl₃.

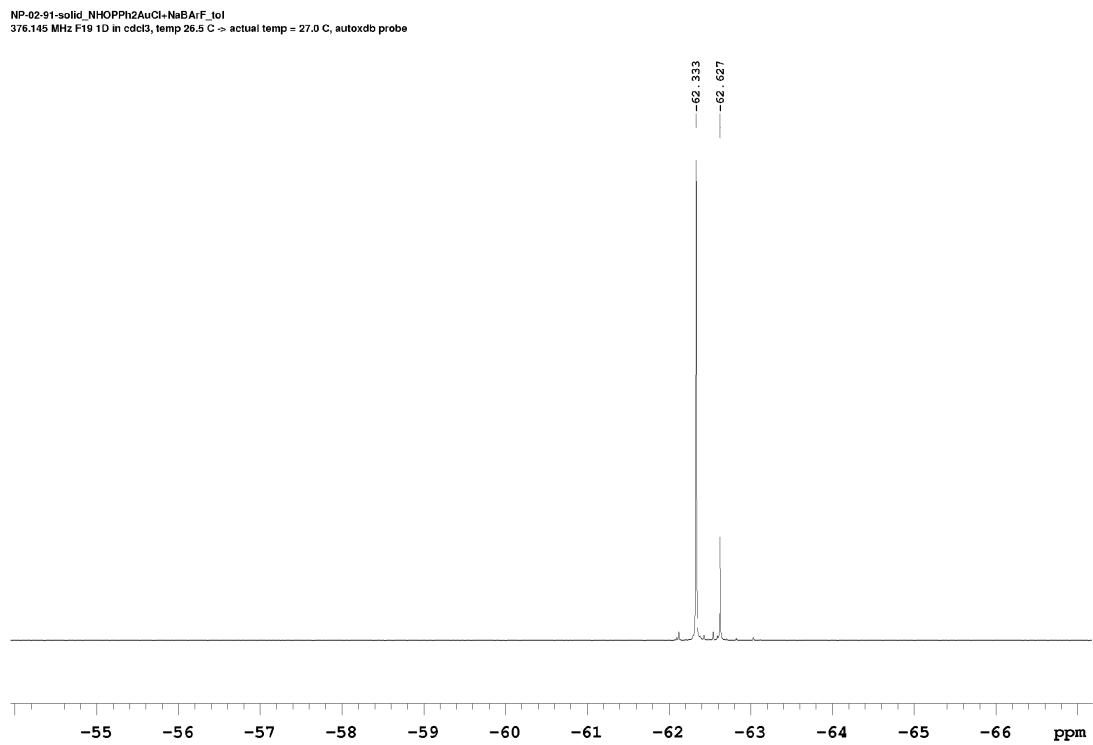


Figure S29. $^{31}\text{P}\{\text{H}\}$ NMR spectrum of [IPr-CH₂-PPh₂•AuAr^F][BAr^F₄] (**10**) in CDCl₃.

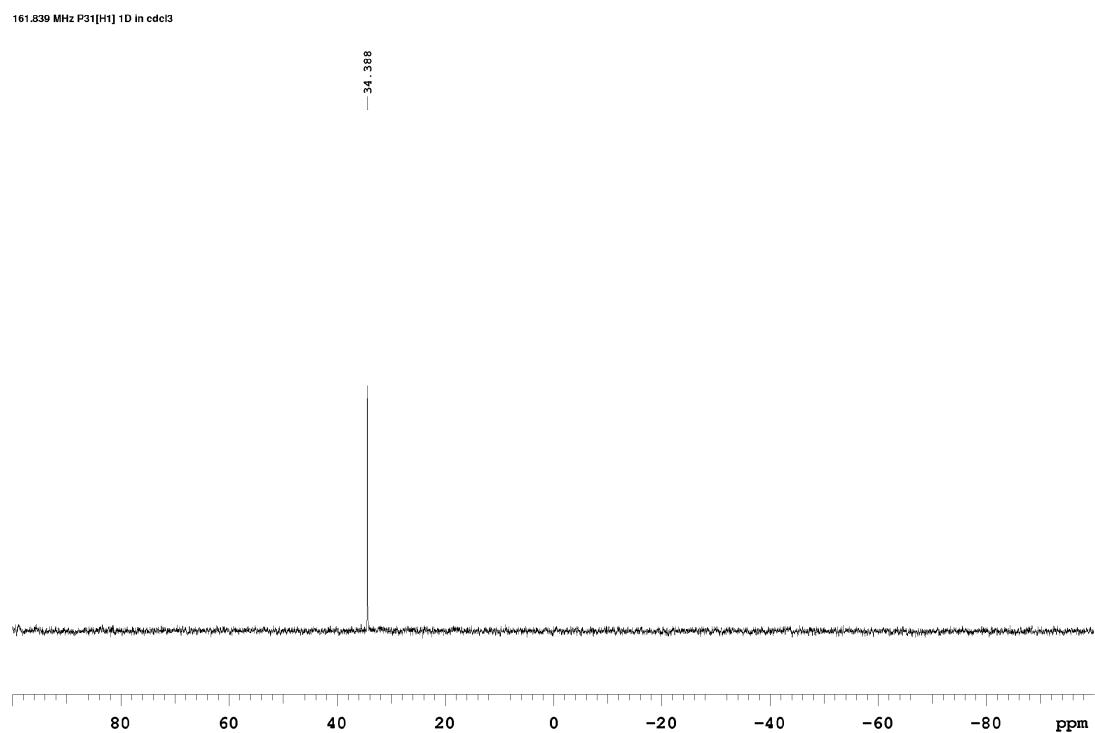


Figure S30. ^1H NMR spectrum of $(\text{IPr}=\text{CH})\text{NMe}_2 \bullet \text{AuCl}$ (**11**) in C_6D_6 .

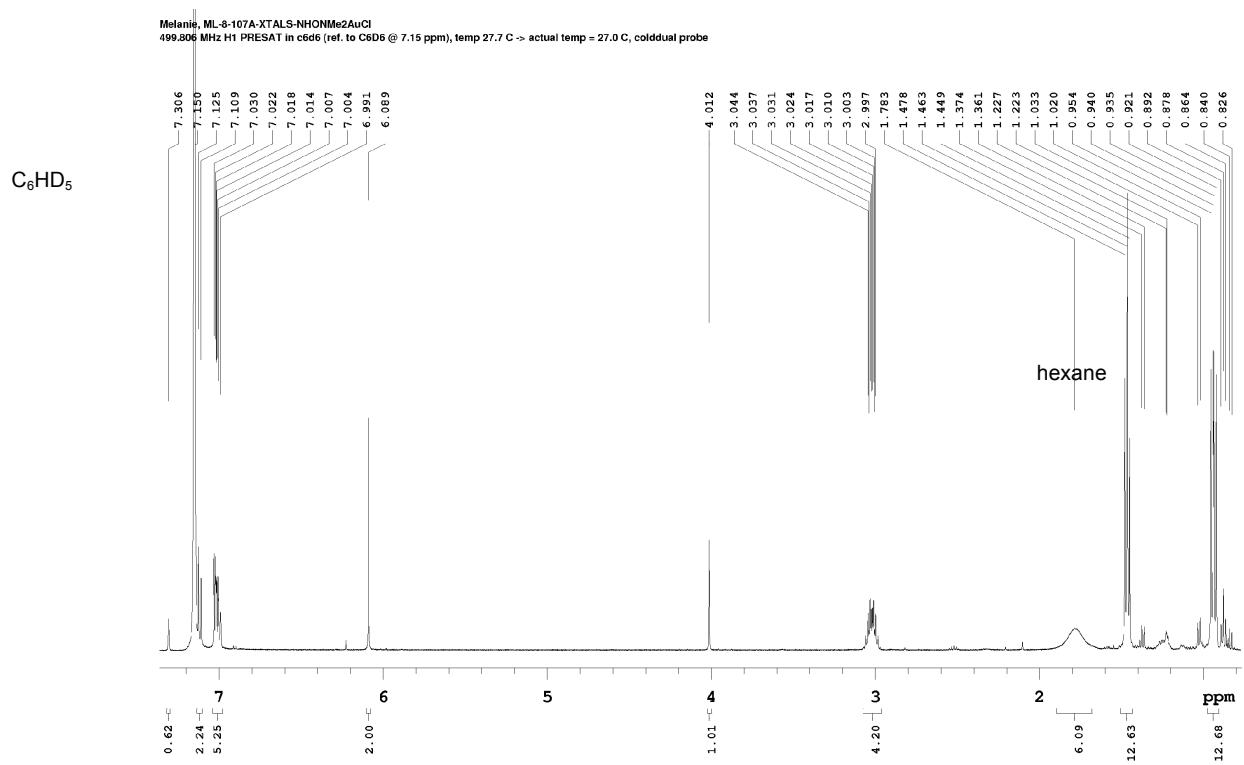
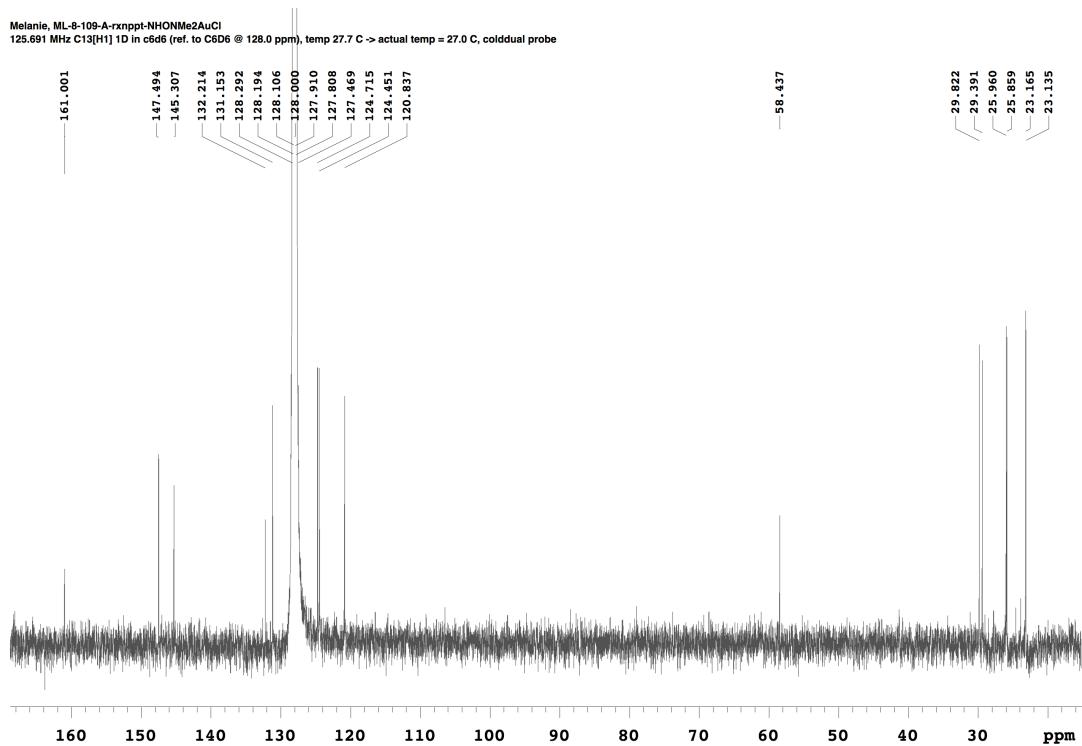


Figure S31. $^{13}\text{C}_{\text{C}_6\text{D}_6}\{1\text{H}\}$ NMR spectrum of (IPr=CH)NMe₂•AuCl (**11**) in C₆D₆.



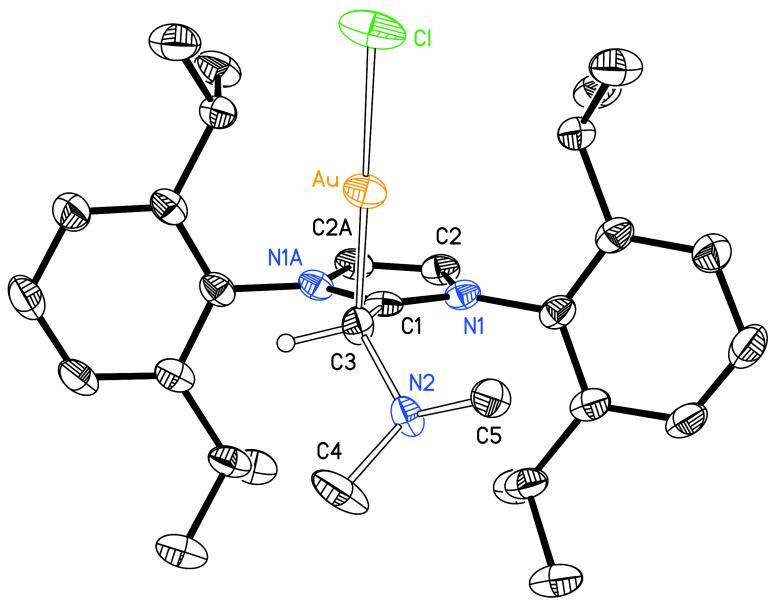


Figure S32. Molecular structure of $(\text{IPr}=\text{CH})\text{NMe}_2 \bullet \text{AuCl}$ (**11**) with thermal ellipsoids presented at a 30 % probability level. The hydrogen atom at C(3) is shown with an arbitrarily small thermal parameter; all remaining hydrogen atoms have been omitted for clarity. Selected bond lengths (Å) and angles (°): Au-C(3) 2.044(15), Au-Cl 2.300(4), C(3)-N(2) 1.44(2); Cl-Au-C(3) 177.6(4), C(1)-C(3)-Au 108.2(8), C(1)-C(3)-N(2) 110.9(16).

Table S1. Crystallographic Experimental Details for **11**.*A. Crystal Data*

formula	C ₃₀ H ₄₃ AuClN ₃
formula weight	678.09
crystal dimensions (mm)	0.11 × 0.10 × 0.05
crystal system	monoclinic
space group	P2 ₁ /m (No. 11)
unit cell parameters ^a	
a (Å)	8.9505 (3)
b (Å)	18.1602 (7)
c (Å)	9.8838 (3)
β (deg)	110.4098 (17)
V (Å ³)	1505.69 (9)
Z	2
ρ _{calcd} (g cm ⁻³)	1.496
μ (mm ⁻¹)	10.14

B. Data Collection and Refinement Conditions

diffractometer	Bruker D8/APEX II CCD ^b
radiation (λ [Å])	Cu K α (1.54178) (microfocus source)
temperature (°C)	-100
scan type	ω and ϕ scans (1.0°) (5 s exposures)
data collection 2θ limit (deg)	146.18
total data collected	10578 (-11 ≤ h ≤ 10, -22 ≤ k ≤ 22, -12 ≤ l ≤ 12)
independent reflections	3026 ($R_{\text{int}} = 0.0306$)
number of observed reflections (NO)	2927 [$F_o^2 \geq 2\sigma(F_o^2)$]
structure solution method	intrinsic phasing (<i>SHELXT-2014^c</i>)
refinement method	full-matrix least-squares on F^2 (<i>SHELXL-2014^d</i>)
absorption correction method	Gaussian integration (face-indexed)
range of transmission factors	0.7213–0.4106
data/restraints/parameters	3026 / 0 / 189
goodness-of-fit (S) ^e [all data]	1.359
final R indices ^f	
R_1 [$F_o^2 \geq 2\sigma(F_o^2)$]	0.0606
wR_2 [all data]	0.1230
largest difference peak and hole	2.060 and -4.228 e Å ⁻³

^aObtained from least-squares refinement of 9990 reflections with 9.54° < 2θ < 144.30°.

(continued)

Table S1. Crystallographic Experimental Details for **11** (continued)

^bPrograms for diffractometer operation, data collection, data reduction and absorption correction were those supplied by Bruker. The crystal used for data collection was the 'best' of a batch of low quality crystals. The unit cell was indexed using the program CELL_NOW, and the major component fit ~60% of the thresholded reflections. There were at least an additional six components, and attempts to integrate a multicomponent dataset were not particularly successful. The noisy difference map can most likely be attributed to the fact that there are a number of additional partially overlapping components contributing to the measured intensities (this is also apparent in the list of most disagreeable reflections in the SHELXL-2014 output with I_{obs} larger than I_{calc} for the top 50 reflections). Attempts to refine the structure in $P2_1$ instead to $P2_1/m$ gave massive correlations of the ADPs and a more poorly-behaved structure. New crystallization experiments are currently underway in the hope of yielding better quality single crystals.

^cSheldrick, G. M. *Acta Crystallogr.* **2015**, *A71*, 3–8. (SHELXT-2014)

^dSheldrick, G. M. *Acta Crystallogr.* **2015**, *C71*, 3–8. (SHELXL-2014)

^e $S = [\sum w(F_o^2 - F_c^2)^2 / (n - p)]^{1/2}$ (n = number of data; p = number of parameters varied; $w = [\sigma^2(F_o^2) + 11.7041P]^{-1}$ where $P = [\text{Max}(F_o^2, 0) + 2F_c^2]/3$).

^f $R_1 = \sum |F_o| - |F_c| / \sum |F_o|$; $wR_2 = [\sum w(F_o^2 - F_c^2)^2 / \sum w(F_o^4)]^{1/2}$.