

**Supplementary information (ESI)**

**A facile room temperature chemical transformation approach for binder-free thin film formation of Ag<sub>2</sub>Te and lithiation/delithiation chemistry of the film**

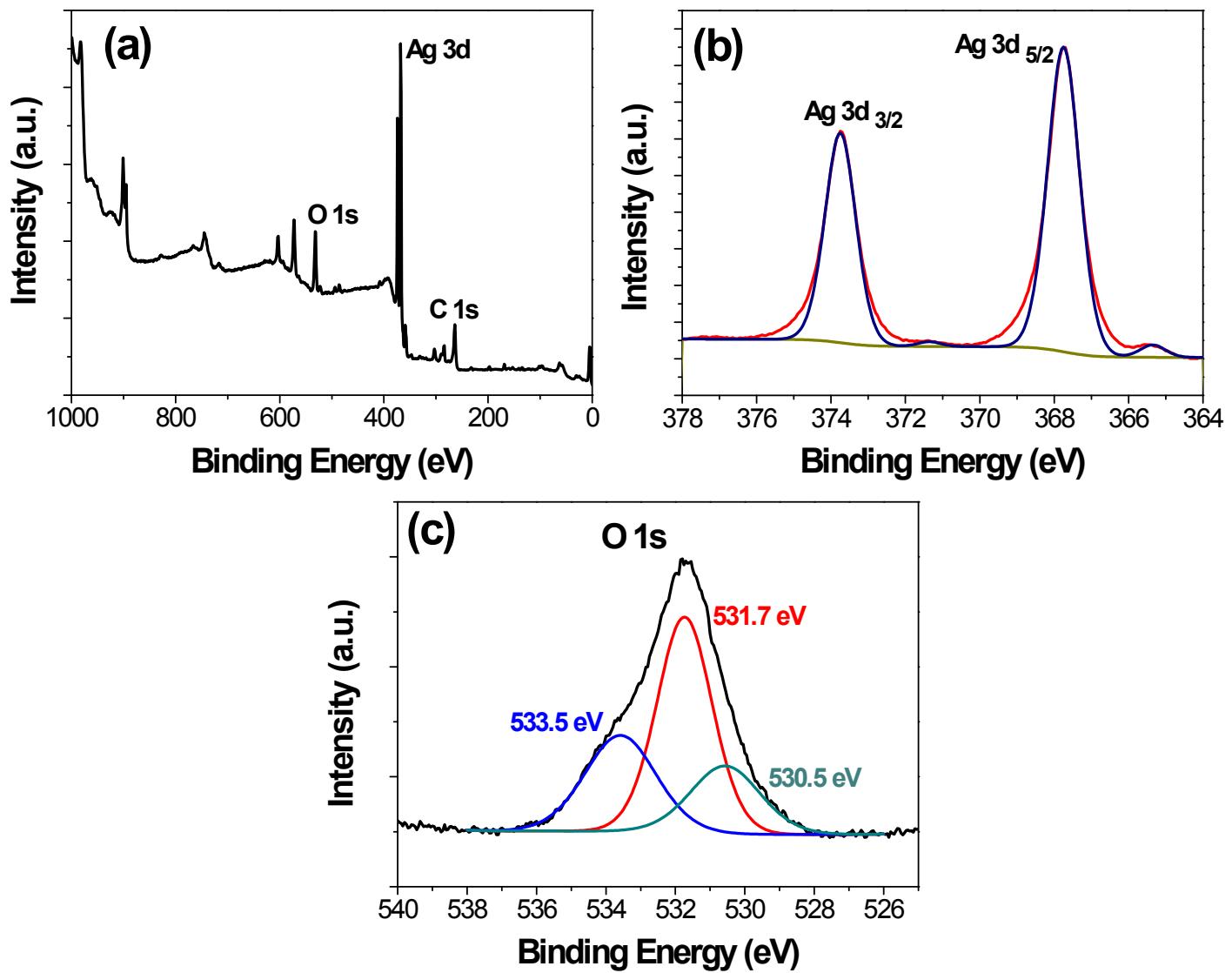
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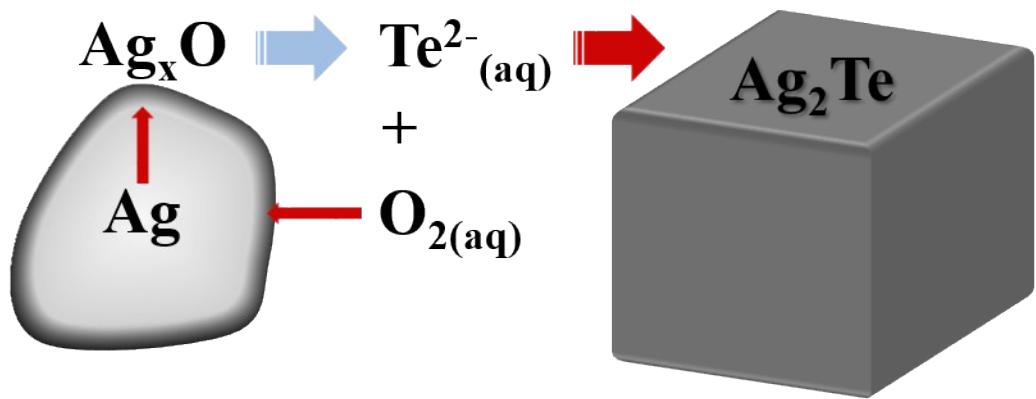
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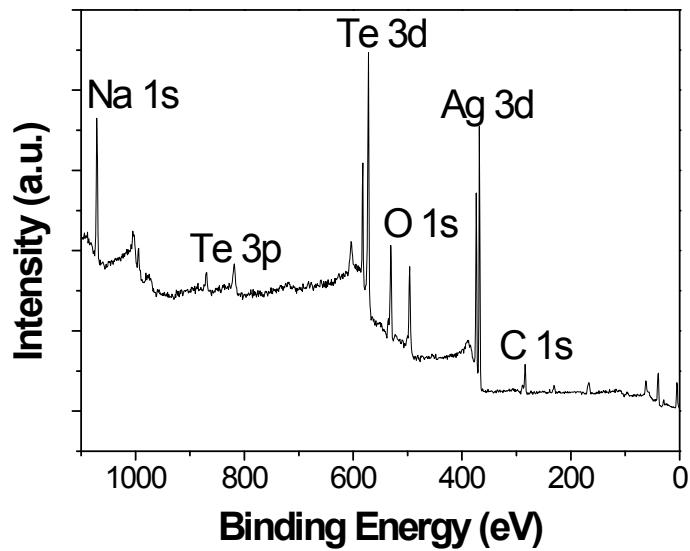
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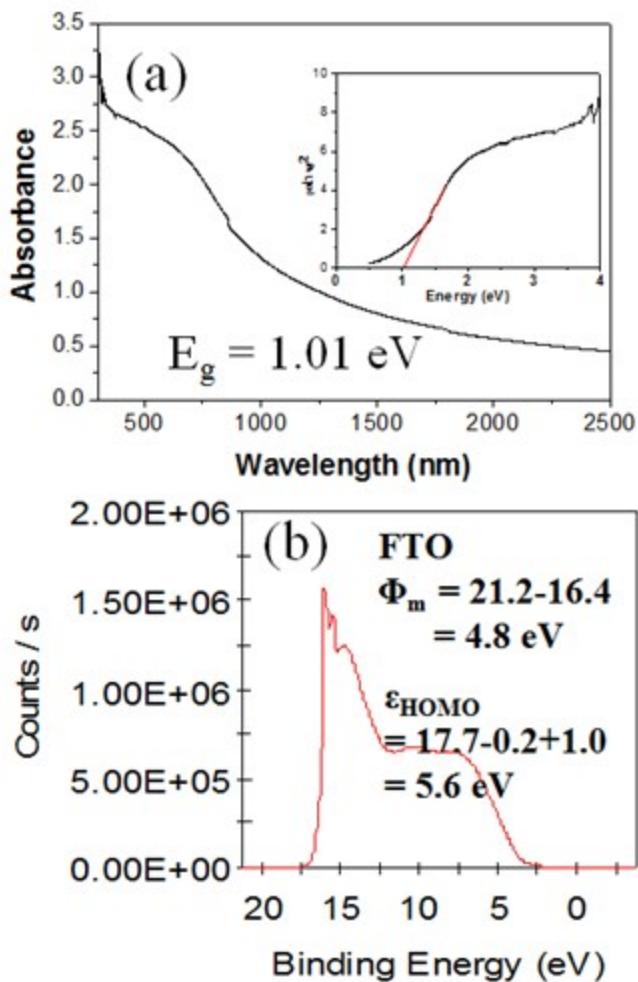
**Fig. S1.** XPS spectra of Ag/Ag<sub>x</sub>O film. (a) Survey spectrum, (b) Ag 3d core level spectrum, and (c) O 1s core level spectrum.



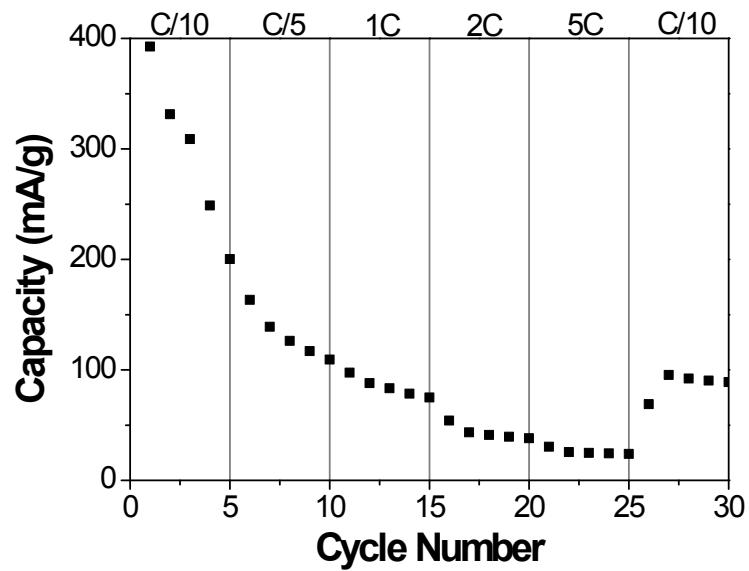
**Fig. S2.** Schematic diagram showing the surface oxidation of Ag particles followed by chemical transformation into  $\text{Ag}_2\text{Te}$  in presence of aqueous Te-precursor solution at room temperature.



**Fig. S3.** XPS survey spectrum of chemically transformed  $\text{Ag}_2\text{Te}$  film. The Na signal at 1071.12 eV is attributed to residual  $\text{NaOH}$ .



**Fig. S4.** (a) UV-vis absorption spectrum of  $\text{Ag}_2\text{Te}$  film and inset shows the Tauc plot for estimation of optical gap. (b) UPS spectrum for estimation of valance band edge position of  $\text{Ag}_2\text{Te}$  film. (c) Various parameters obtained from Hall measurement of the  $\text{Ag}_2\text{Te}$  film.



**Fig. S5** Charge–discharge curves of the Li/LiPF<sub>6</sub>/Ag<sub>2</sub>Te battery at various current densities.