

Supporting information

Novel and Easy Access to Highly Luminescent Eu and Tb Doped Ultra-small CaF_2 , SrF_2 and BaF_2 Nanoparticles – Structure and Luminescence

*Benjamin Ritter¹, Philipp Haida¹, Friedrich Fink¹, Thoralf Krah^{1,2}, Kornelia Gawlitza³, Knut Rurack³, Gudrun Scholz¹ and Erhard Kemnitz^{1,2} **

¹ Humboldt-Universität zu Berlin, Institut für Chemie, Brook-Taylor- Str. 2, D-12489 Berlin, Germany

² Nanofluor GmbH, Rudower Chaussee 29, D-12489 Berlin, Germany

³ Bundesanstalt für Materialforschung und -prüfung (BAM), Fachbereich Chemische und optische Sensorik, Richard-Willstätter-Str. 11, D-12489 Berlin, Germany

* Corresponding author: Prof. Dr. E. Kemnitz, Mail: erhard.kemnitz@chemie.hu-berlin.de, Phone: +49 30 2093 7555, Fax: +49 30 2093 7277

Contents

Figure S1. Powder diffractograms of SrF_2 and $\text{SrF}_2\text{:Eu10}$ xerogels with and without $\text{Al}(\text{O}^i\text{Pr})_3$ annealed at 400°C and 700°C.

Figure S2. ^{19}F MAS NMR spectrum ($\nu_{\text{rot}} = 30$ kHz) of the SrF_2 xerogel with 15 mol% $\text{Al}(\text{O}^i\text{Pr})_3$.

Figure S3. Luminescence excitation spectra. Left: $\text{SrF}_2\text{:Eu10}$ ($\lambda_{\text{em}} = 590$ nm), right: $\text{SrF}_2\text{:Tb10}$ ($\lambda_{\text{em}} = 542$ nm). Both pure and with 15 mol% $\text{Al}(\text{O}^i\text{Pr})_3$ added.

Figure S4. Comparison of measured and simulated ^{19}F NMR spectra of 0.2 M *nano*- $\text{Ca}_{1-x}\text{Sr}_x\text{F}_2$ sols in ethylene glycol with $0.10 \leq \chi_c \leq 0.90$.

Figure S5. Absorbance of $\text{Ca}_x\text{Sr}_{1-x}\text{F}_2\text{:Eu10}$ at 393 nm (0.2 M sol, 1 cm sample thickness).

Table S1. Labelling of the samples $\text{Ca}_x\text{Sr}_{1-x}\text{F}_2\text{:RE10}$.

Table S2. Particle diameter of 0.2 M sols of $\text{SrF}_2\text{:EuX}$ and $\text{SrF}_2\text{:TbX}$ ($X = 1\text{--}15$ mol%) by DLS.

Table S3. Lifetime changes of the $\text{SrF}_2\text{:Eu10}$ sols in ethylene glycol with 15 mol% of different additives.

Table S4. DLS particle diameters and luminescence lifetimes for $\text{SrF}_2\text{:EuX}$ and $\text{SrF}_2\text{:TbX}$ in ethylene glycol with $X = 0\text{--}40$.

Table S5. Properties of 0.2 M $\text{AEF}_2\text{:Eu10}$ sols in ethylene glycol ($\text{AE} = \text{Ca}, \text{Sr}, \text{Ba}$).

Table S6. Properties of 0.2 $\text{Ca}_{1-x}\text{Sr}_x\text{F}_2\text{:Eu10}$ and $\text{Ca}_{1-x}\text{Sr}_x\text{F}_2\text{:Tb10}$ sols in ethylene glycol ($0 \leq x \leq 1$).

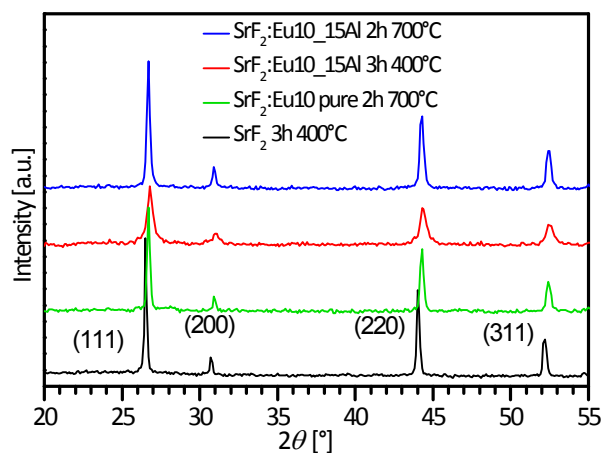


Figure S1. Powder diffractograms of SrF_2 and $\text{SrF}_2:\text{Eu10}$ xerogels with and without $\text{Al}(\text{O}^i\text{Pr})_3$ annealed at 400°C and 700°C.

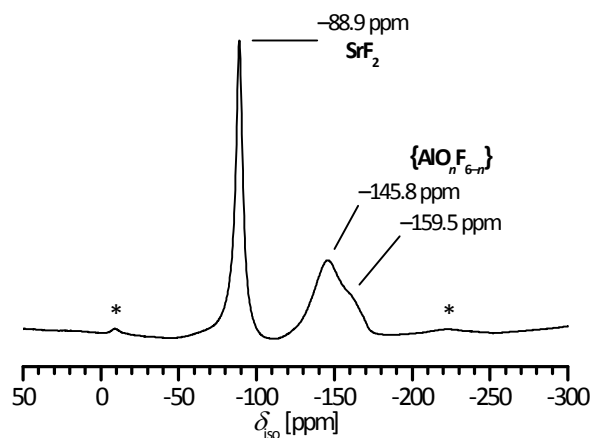


Figure S2. ^{19}F MAS NMR spectrum ($\nu_{\text{rot}} = 30$ kHz) of the SrF_2 xerogel with 15 mol% $\text{Al}(\text{O}^i\text{Pr})_3$. * ~spinning side bands.

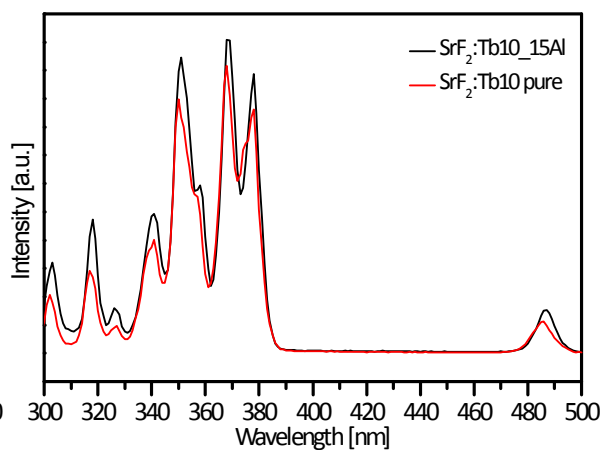
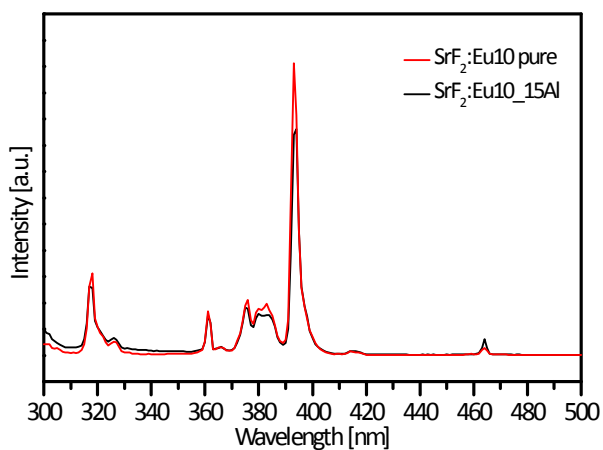


Figure S3. Luminescence excitation spectra. left: $\text{SrF}_2:\text{Eu10}$ ($\lambda_{\text{em}} = 590$ nm), right: $\text{SrF}_2:\text{Tb10}$ ($\lambda_{\text{em}} = 542$ nm). Both pure and with 15 mol% $\text{Al}(\text{O}^i\text{Pr})_3$ added.

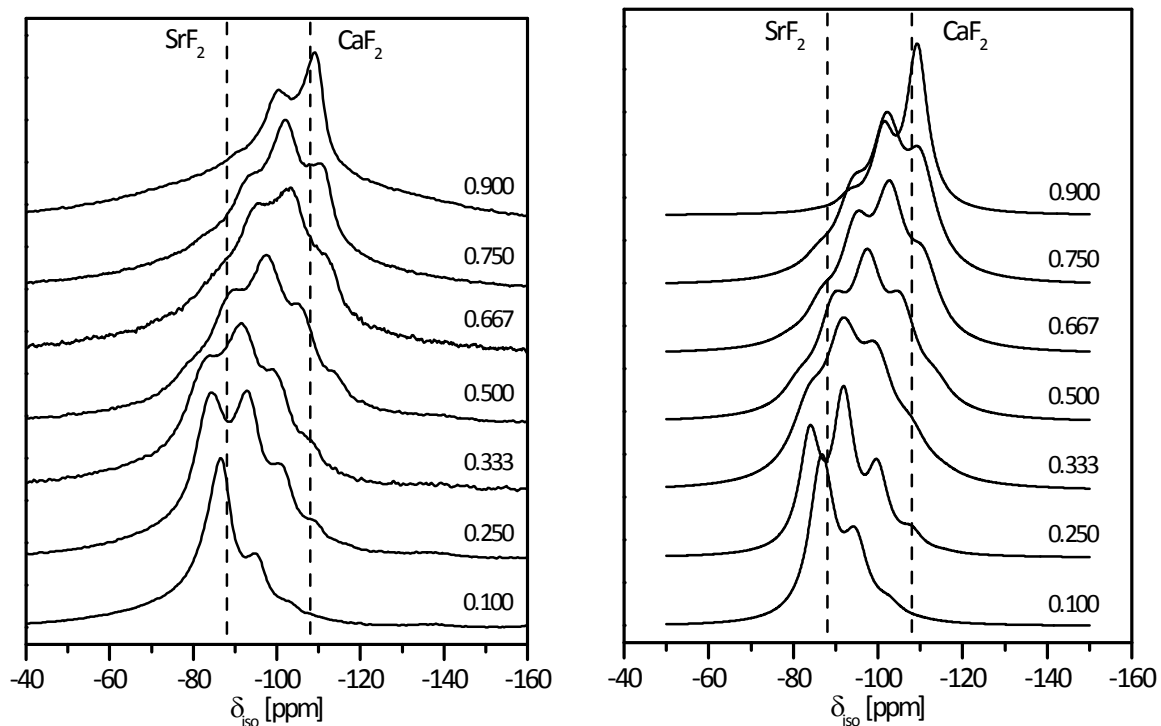


Figure S4. Comparison of measured (left) and simulated (right) ^{19}F NMR spectra of *nano*- $\text{Ca}_{1-x}\text{Sr}_x\text{F}_2$ sols in ethylene glycol with $0.10 \leq \chi_c \leq 0.90$ (dashed lines show the chemical shifts for pure CaF_2 and SrF_2).

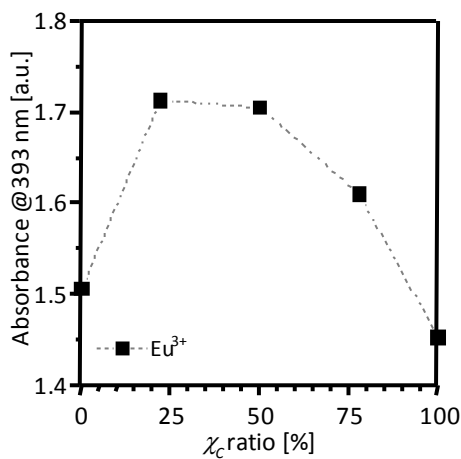


Figure S5. Absorbance of $\text{Ca}_x\text{Sr}_{1-x}\text{F}_2:\text{Eu}10$ at 393 nm (0.2 M sol, 1 cm sample thickness).

Table S1. Labelling of the samples $\text{Ca}_x\text{Sr}_{1-x}\text{F}_2\text{:RE10}$.

Shortcut <i>RE</i> = Eu, Tb	Sum formula <i>RE</i> = Eu, Tb	χ_c
$\text{SrF}_2\text{:RE10}$	$\text{Sr}_{0.9}\text{RE}_{0.1}\text{F}_{2.1}$	0.000
$\text{Ca}_{0.1}\text{Sr}_{0.9}\text{F}_2\text{:RE10}$	$\text{Ca}_{0.05}\text{Sr}_{0.85}\text{RE}_{0.1}\text{F}_{2.1}$	0.056
$\text{Ca}_{0.25}\text{Sr}_{0.75}\text{F}_2\text{:RE10}$	$\text{Ca}_{0.20}\text{Sr}_{0.70}\text{RE}_{0.1}\text{F}_{2.1}$	0.222
$\text{Ca}_{0.33}\text{Sr}_{0.67}\text{F}_2\text{:RE10}$	$\text{Ca}_{0.28}\text{Sr}_{0.62}\text{RE}_{0.1}\text{F}_{2.1}$	0.311
$\text{Ca}_{0.5}\text{Sr}_{0.5}\text{F}_2\text{:RE10}$	$\text{Ca}_{0.45}\text{Sr}_{0.45}\text{RE}_{0.1}\text{F}_{2.1}$	0.500
$\text{Ca}_{0.67}\text{Sr}_{0.33}\text{F}_2\text{:RE10}$	$\text{Ca}_{0.62}\text{Sr}_{0.28}\text{RE}_{0.1}\text{F}_{2.1}$	0.689
$\text{Ca}_{0.75}\text{Sr}_{0.25}\text{F}_2\text{:RE10}$	$\text{Ca}_{0.70}\text{Sr}_{0.20}\text{RE}_{0.1}\text{F}_{2.1}$	0.778
$\text{Ca}_{0.9}\text{Sr}_{0.1}\text{F}_2\text{:RE10}$	$\text{Ca}_{0.85}\text{Sr}_{0.05}\text{RE}_{0.1}\text{F}_{2.1}$	0.944
$\text{CaF}_2\text{:RE10}$	$\text{Ca}_{0.9}\text{RE}_{0.1}\text{F}_{2.1}$	1.000

Table S2. Particle diameter of 0.2 M sols of $\text{SrF}_2\text{:EuX}$ and $\text{SrF}_2\text{:TbX}$ ($X = 1\text{-}15$ mol%) by DLS (number).

X [mol%]	DLS diam. (nm) $\text{SrF}_2\text{:EuX}$	DLS diam. (nm) $\text{SrF}_2\text{:TbX}$	Solvent and additive
1	3	3	synthesis 2 in methanol, +15 mol% $\text{Al}(\text{O}^i\text{Pr})_3$
2	3	3	
4	11	14	
10	24	5	
15	24	24	
10	5	5	synthesis 1 in ethylene glycol, no additive

Table S3. Lifetime changes of the $\text{SrF}_2\text{:Eu10}$ sols in ethylene glycol with 15 mol% of different additives. Note the large asymmetry parameter in the case of $\text{Al}(\text{O}^i\text{Pr})_3$. **T1**: $^5D_0 \rightarrow ^7F_1$, **T2**: $^5D_0 \rightarrow ^7F_2$.

Additive	Luminescence lifetime	T1 : T2 intensity
<i>none</i>	t_1 3.1 ms q_1 1.121	63 : 37
Citric acid	t_1 3.1 ms q_1 1.137	62 : 38
Benzoic acid	t_1 3.1 ms q_1 1.148	63 : 37
Phenylphosphonic acid	t_1 3.3 ms q_1 1.141	51 : 49
Tetramethyl orthosilicate	t_1 3.5 ms q_1 1.144	56 : 44
Aluminium isopropoxide	t_1 2.2 ms q_1 1.259	44 : 56
Titanium isopropoxide	t_1 3.0 ms q_1 1.183	50 : 50

Table S4. DLS particle diameters and luminescence lifetimes for SrF₂:EuX and SrF₂:TbX in ethylene glycol with X = 0...40. ($\lambda_{ex}(\text{Eu}) = 393 \text{ nm}$, $\lambda_{ex}(\text{Tb}) = 350 \text{ nm}$, t_1 = lifetime and q_1 = asymmetry parameter).

Abbreviation (RE = Eu, Tb)	Sum formula (RE = Eu, Tb)	DLS particle diameter (nm)		Luminescence lifetime	
		RE = Eu	RE = Tb	RE = Eu	RE = Tb
SrF ₂	SrF ₂	13.5		–	
SrF ₂ :RE0.1	Sr _{0.999} RE _{0.001} F _{2.001}	24.4		t_1 1.2 ms q_1 1.300	
SrF ₂ :RE0.2	Sr _{0.998} RE _{0.002} F _{2.002}	18.2		t_1 1.5 ms q_1 1.300	
SrF ₂ :RE1	Sr _{0.99} RE _{0.01} F _{2.01}	6.5	4.2	t_1 2.3 ms q_1 1.300	t_1 2.59 ms q_1 1.300
SrF ₂ :RE2	Sr _{0.98} RE _{0.02} F _{2.02}	5.6		t_1 3.23 ms q_1 1.160	
SrF ₂ :RE4	Sr _{0.96} RE _{0.04} F _{2.04}	8.7		t_1 3.05 ms q_1 1.166	
SrF ₂ :RE5	Sr _{0.95} RE _{0.05} F _{2.05}	5.6		t_1 2.2 ms q_1 1.296	
SrF ₂ :RE10	Sr _{0.9} RE _{0.1} F _{2.1}	5.0	8.5	t_1 2.7 ms q_1 1.174	t_1 3.81 ms q_1 1.013
SrF ₂ :RE20	Sr _{0.8} RE _{0.2} F _{2.2}	4.2	10.0	t_1 2.2 ms q_1 1.101	t_1 3.03 ms q_1 1.069
SrF ₂ :RE30	Sr _{0.7} RE _{0.3} F _{2.3}	4.2	10.1	t_1 2.2 ms q_1 1.061	t_1 3.43 ms q_1 1.041
SrF ₂ :RE40	Sr _{0.6} RE _{0.4} F _{2.4}	3.6	11.7	t_1 2.0 ms q_1 1.026	t_1 2.41 ms q_1 1.092

Table S5. Properties 0.2 M AEF₂:Eu10 sols (AE = Ca, Sr, Ba) in ethylene glycol (t_1 = lifetime, q_1 = asymmetry parameter).

Shortcut	Sum formula	DLS particle diameter (nm)	Luminescence lifetime	Relative quantum yield
BaF ₂ :Eu10_15Al	Ba _{0.9} Eu _{0.1} F _{2.1} +15 mol% Al(O'Pr) ₃	28.2 nm	t_1 2.87 ms q_1 1.230	
SrF ₂ :Eu10_15Al	Sr _{0.9} Eu _{0.1} F _{2.1} +15 mol% Al(O'Pr) ₃	6.5 nm	t_1 2.38 ms q_1 1.289	
CaF ₂ :Eu10_15Al	Ca _{0.9} Eu _{0.1} F _{2.1} +15 mol% Al(O'Pr) ₃	4.6 nm	t_1 1.95 ms q_1 1.300	
BaF ₂ :Eu10_5BA	Ba _{0.9} Eu _{0.1} F _{2.1} +5 mol% BA	15.7 nm	t_1 3.28 ms q_1 1.021	34.6%
SrF ₂ :Eu10_5BA	Sr _{0.9} Eu _{0.1} F _{2.1} +5 mol% BA	4.8 nm	t_1 2.80 ms q_1 1.150	29.8%
CaF ₂ :Eu10_5BA	Ca _{0.9} Eu _{0.1} F _{2.1} +5 mol% BA	3.6 nm	t_1 2.12 ms q_1 1.300	29.4%

Table S6. Properties of 0.2 M $\text{Ca}_{1-x}\text{Sr}_x\text{F}_2\text{:Eu10}$ and $\text{Ca}_{1-x}\text{Sr}_x\text{F}_2\text{:Tb10}$ sols in ethylene glycol ($0 \leq x \leq 1$). ($\lambda_{ex}(\text{Eu}) = 393 \text{ nm}$, $\lambda_{ex}(\text{Tb}) = 350 \text{ nm}$, t_1 = lifetime and q_1 = asymmetry parameter).

Shortcut $RE = \text{Eu, Tb}$	Sum formula $RE = \text{Eu, Tb}$	χ_C	DLS particle diameter (nm)		Luminescence lifetime t_1 (ms)		Relative quantum yield	
			$RE = \text{Eu}$	$RE = \text{Tb}$	$RE = \text{Eu}$	$RE = \text{Tb}$	$RE = \text{Eu}$	$RE = \text{Tb}$
$\text{SrF}_2\text{:RE10}$	$\text{Sr}_{0.9}\text{RE}_{0.1}\text{F}_{2.1}$	0.000	7.5	6.5	t_1 2.24 q_1 1.157	t_1 3.66 q_1 1.059	27.3%	79.2%
$\text{Ca}_{0.1}\text{Sr}_{0.9}\text{F}_2\text{:RE10}$	$\text{Ca}_{0.05}\text{Sr}_{0.85}\text{RE}_{0.1}\text{F}_{2.1}$	0.056	7.5	6.5	t_1 2.21 q_1 1.160	t_1 3.64 q_1 1.064		
$\text{Ca}_{0.25}\text{Sr}_{0.75}\text{F}_2\text{:RE10}$	$\text{Ca}_{0.20}\text{Sr}_{0.70}\text{RE}_{0.1}\text{F}_{2.1}$	0.222	7.5	4.8	t_1 2.18 q_1 1.178	t_1 3.59 q_1 1.065	26.9%	
$\text{Ca}_{0.33}\text{Sr}_{0.67}\text{F}_2\text{:RE10}$	$\text{Ca}_{0.28}\text{Sr}_{0.62}\text{RE}_{0.1}\text{F}_{2.1}$	0.311	5.6	4.1	t_1 2.12 q_1 1.189	t_1 3.57 q_1 1.061		
$\text{Ca}_{0.5}\text{Sr}_{0.5}\text{F}_2\text{:RE10}$	$\text{Ca}_{0.45}\text{Sr}_{0.45}\text{RE}_{0.1}\text{F}_{2.1}$	0.500	5.6	3.6	t_1 2.08 q_1 1.191	t_1 3.54 q_1 1.069		
$\text{Ca}_{0.67}\text{Sr}_{0.33}\text{F}_2\text{:RE10}$	$\text{Ca}_{0.62}\text{Sr}_{0.28}\text{RE}_{0.1}\text{F}_{2.1}$	0.689	4.2	3.1	t_1 2.01 q_1 1.206	t_1 3.48 q_1 1.073		
$\text{Ca}_{0.75}\text{Sr}_{0.25}\text{F}_2\text{:RE10}$	$\text{Ca}_{0.70}\text{Sr}_{0.20}\text{RE}_{0.1}\text{F}_{2.1}$	0.778	3.6	2.6	t_1 1.92 q_1 1.218	t_1 3.45 q_1 1.078		
$\text{Ca}_{0.9}\text{Sr}_{0.1}\text{F}_2\text{:RE10}$	$\text{Ca}_{0.85}\text{Sr}_{0.05}\text{RE}_{0.1}\text{F}_{2.1}$	0.944	2.4	2.6	t_1 1.70 q_1 1.189	t_1 3.29 q_1 1.083		
$\text{CaF}_2\text{:RE10}$	$\text{Ca}_{0.9}\text{RE}_{0.1}\text{F}_{2.1}$	1.000	2.4	2.6	t_1 1.64 q_1 1.261	t_1 3.20 q_1 1.086	24.7%	76.5%