Appendix A. Supplementary data for “3D-QSAR predictions for bovine serum albumin-water partitioning coefficients of organic anions using quantum mechanically based descriptors”

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The number of local sigma profiles depends on the default $\sigma$-interval \textit{(delsig parameter in COSMOsar3D)} and the range of sigma values of all training set chemicals. We chose a $\text{delsig}$ value of 0.006 $\sigma$ [e/Å$^2$] as it was proposed as sufficient by Klamt \textit{et al.} (Klamt, A. \textit{et al.}; \textit{Journal of Chemical Information and Modeling}, 2012, \textbf{52} (8), 2157-2164). For the derivation of the LSPs a grid size of 2 Å was chosen because this grid size was also proposed by Klamt \textit{et al.} and it is a common setting for 3D-QSAR models (Verma, J. \textit{et al.}; \textit{Current Topics in Medicinal Chemistry}, 2010, \textbf{10} (1), 95-115).

\begin{figure}[h]
\centering
\includegraphics[width=\textwidth]{figure.png}
\caption{Generation of the sigma profile (top) and the local sigma profiles (LSPs) (bottom). The LSPs are shown for two exemplary grid points indicated with a black plus sign. The lattice is moved through the whole molecule in steps of 2 Å and the MIFs are derived at each grid point. The figure is designed following Linden \textit{et al.} [CD paper model] and omitting LSP 1, LSP 2, and LSP 10 for the sake of clarity.}
\end{figure}
The alignment of the five template chemicals is illustrated in Figures SI2.

Figure SI 2 Result of the COSMOsim3D alignment for the template chemicals. Note that the order of the chemicals is arbitrary.
Comparison of the alignment of 2,4,6-trimethylbenzene sulfonate and 2,4,6-trimethylbenzoate:

Figure SI 3 Alignment of 2,4,6-trimethylbenzene sulfonate and 2,4,6-trimethylbenzoate. The green lines in the pictures show the template chemicals while the blue sticks show 2,4,6-trimethylbenzene sulfonate and the red sticks show 2,4,6-trimethylbenzoate. The teal area indicates the area where LSP7 is positively correlated to log $K_{BSA/water}$ and the violet area indicates the area where LSP8 is positively correlated to log $K_{BSA/water}$. The alignment figures were generated using Pymol38.

S-3 Results of the random test set predictions

Table SI 1 Statistical results for the random test set predictions

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<tr>
<td>phenoxyacetic acid (anion)</td>
<td>2.57</td>
<td>3.14</td>
<td></td>
</tr>
<tr>
<td>pyrene</td>
<td>4.76</td>
<td>4.49</td>
<td></td>
</tr>
<tr>
<td>sulcotrione (anion)</td>
<td>1.72</td>
<td>3.70</td>
<td></td>
</tr>
<tr>
<td>1,2,4-trimethylbenzene</td>
<td>3.35</td>
<td>2.39</td>
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</tr>
<tr>
<td>1,4-dibromobenzene</td>
<td>3.97</td>
<td>3.31</td>
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<tr>
<td>1-naphthoic acid (anion)</td>
<td>2.81</td>
<td>3.01</td>
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<tr>
<td>1-nonene</td>
<td>4.22</td>
<td>3.20</td>
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<td>2,4,5-T (anion)</td>
<td>3.83</td>
<td>3.20</td>
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<tr>
<td>2,4-DB (anion)</td>
<td>4.12</td>
<td>3.05</td>
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<tr>
<td>2-chloroaniline</td>
<td>1.95</td>
<td>2.76</td>
<td></td>
</tr>
<tr>
<td>Compound</td>
<td>log $K_{BSA/water}$ experimental</td>
<td>log $K_{BSA/water}$ predicted</td>
<td></td>
</tr>
<tr>
<td>---------------------------------------</td>
<td>----------------------------------</td>
<td>-------------------------------</td>
<td></td>
</tr>
<tr>
<td>2-chlorobenzoic acid (anion)</td>
<td>1.84</td>
<td>1.67</td>
<td></td>
</tr>
<tr>
<td>2-octanone</td>
<td>2.09</td>
<td>2.48</td>
<td></td>
</tr>
<tr>
<td>4-aminobiphenyl</td>
<td>2.55</td>
<td>3.58</td>
<td></td>
</tr>
<tr>
<td>4-bromobenzoic acid (anion)</td>
<td>3.48</td>
<td>3.87</td>
<td></td>
</tr>
<tr>
<td>4-chlorobenzoic acid (anion)</td>
<td>3.21</td>
<td>3.75</td>
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<tr>
<td>4-fluorophenol</td>
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<td>2.34</td>
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<tr>
<td>4-n-octylbenzenesulfonate (anion)</td>
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<td>4.33</td>
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<tr>
<td>anisole</td>
<td>2.16</td>
<td>2.11</td>
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<tr>
<td>bisphenol A</td>
<td>2.88</td>
<td>2.98</td>
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<tr>
<td>coumachtymor (anion)</td>
<td>3.37</td>
<td>2.80</td>
<td></td>
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<tr>
<td>coumactyfurl (anion)</td>
<td>2.79</td>
<td>2.67</td>
<td></td>
</tr>
<tr>
<td>diazepam</td>
<td>2.68</td>
<td>1.94</td>
<td></td>
</tr>
<tr>
<td>dibenzocthophene</td>
<td>4.16</td>
<td>4.20</td>
<td></td>
</tr>
<tr>
<td>enflurane</td>
<td>1.59</td>
<td>2.28</td>
<td></td>
</tr>
<tr>
<td>indole</td>
<td>2.25</td>
<td>2.90</td>
<td></td>
</tr>
<tr>
<td>mesitylenesulfonate (anion)</td>
<td>4.23</td>
<td>3.82</td>
<td></td>
</tr>
<tr>
<td>metolachlor</td>
<td>1.74</td>
<td>2.44</td>
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<tr>
<td>n-heptane</td>
<td>3.59</td>
<td>3.26</td>
<td></td>
</tr>
<tr>
<td>n-hexane</td>
<td>3.09</td>
<td>2.87</td>
<td></td>
</tr>
<tr>
<td>n-propylbenzene</td>
<td>2.95</td>
<td>3.35</td>
<td></td>
</tr>
<tr>
<td>phenoxyacetic acid (anion)</td>
<td>2.57</td>
<td>3.29</td>
<td></td>
</tr>
<tr>
<td>pyrene</td>
<td>4.76</td>
<td>4.44</td>
<td></td>
</tr>
<tr>
<td>styrene</td>
<td>2.76</td>
<td>2.82</td>
<td></td>
</tr>
<tr>
<td>tetrachloroethene</td>
<td>2.4</td>
<td>2.65</td>
<td></td>
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<tr>
<td>tri-n-butyl phosphate</td>
<td>2.47</td>
<td>2.55</td>
<td></td>
</tr>
</tbody>
</table>

Figure SI 4 Prediction of the test set random1. The solid line indicates the 1:1 line and the dashed lines indicate a deviation of 1 log unit from the 1:1 line.
Figure SI 5 Prediction of the test set random2. The solid line indicates the 1:1 line and the dashed lines indicate a deviation of 1 log unit from the 1:1 line.

Figure SI 6 Prediction of the test set random3. The solid line indicates the 1:1 line and the dashed lines indicate a deviation of 1 log unit from the 1:1 line.
Figure SI 7 Prediction of the test set random4. The solid line indicates the 1:1 line and the dashed lines indicate a deviation of 1 log unit from the 1:1 line.

Figure SI 8 Prediction of the test set random5. The solid line indicates the 1:1 line and the dashed lines indicate a deviation of 1 log unit from the 1:1 line.
S-4 Conformers of isomer pairs

Figure SI 9 Side and top view of the used conformer of 2-methylbenzoate and 4-methylbenzoate
Figure SI 10 Side and top view of the used conformer of 2-chlorobenzoate and 4-chlorobenzoate
Figure SI 11 Side and top view of the used conformer of 2,6-dichlorobenzoate and 3,4-dichlorobenzoate
S-5 Comparison of 2,4,6-trimethylbenzene sulfonate and 2,4,6-trimethylbenzoate

The COSMOfiles of 2,4,6-trimethylbenzene sulfonate and 2,4,6-trimethylbenzoate show the surface polarisation, i.e., the distribution of the partial charges. The benzoic ring of 2,4,6-trimethylbenzene sulfonate (Fig. SI 12) is less polarized than the benzoic ring of 2,4,6-trimethylbenzoate (Fig. SI 13), which is indicated by the greenish patches of 2,4,6-trimethylbenzene sulfonate compared to the yellow patches of 2,4,6-trimethylbenzoate.

Figure SI 12 COSMOfile of 2,4,6-trimethylbenzene sulfonate. The red colour indicates a negative partial charge, the green colour indicates a neutral surface segment, and yellow patches indicate a partial charge in between.

Figure SI 13 COSMOfile of 2,4,6-trimethylbenzoate. The red colour indicates a negative partial charge, the green colour indicates a neutral surface segment, yellow patches indicate a partial charge in between.
S-6 Results of Tanimoto Index

Table SI 3 Statistical results for the comparison of the different Tanimoto mean groups using the five random test sets

<table>
<thead>
<tr>
<th>Tanimoto mean group</th>
<th>Brown–Forsythe variance analysis</th>
<th>Mann–Whitney U median analysis</th>
</tr>
</thead>
<tbody>
<tr>
<td>reference group</td>
<td>Test group</td>
<td>p-value</td>
</tr>
<tr>
<td>0.60–0.70 (n=30)</td>
<td>0.10–0.20 (n= 10)</td>
<td>0.09</td>
</tr>
<tr>
<td>0.60–0.70 (n=30)</td>
<td>0.20–0.30 (n = 24)</td>
<td>0.84</td>
</tr>
<tr>
<td>0.60–0.70 (n=30)</td>
<td>0.30–0.40 (n= 42)</td>
<td>0.28</td>
</tr>
<tr>
<td>0.60–0.70 (n=30)</td>
<td>0.40–0.50 (n= 26)</td>
<td>0.24</td>
</tr>
<tr>
<td>0.60–0.70 (n=30)</td>
<td>0.50–0.60 (n= 24)</td>
<td>0.67</td>
</tr>
<tr>
<td>0.60–0.70 (n=30)</td>
<td>0.70–0.80 (n= 4)</td>
<td>0.20</td>
</tr>
</tbody>
</table>

For the analysis of the domain of applicability, the standard test set selection procedure was also modified to represent all chemicals in the test set once: the numbers one to four were randomly assigned to the four chemicals of each bin and then all chemicals with an identical number formed a test set while the rest of the chemicals formed the respective training set. This is necessary to get the highest possible number of individual test cases and thus an extended statistical evaluation of the domain of applicability.

Table SI 4 Statistical results for the comparison of the different Tanimoto mean groups using the four constructed test sets

<table>
<thead>
<tr>
<th>Tanimoto mean group</th>
<th>Brown–Forsythe variance analysis</th>
<th>Mann–Whitney U median analysis</th>
</tr>
</thead>
<tbody>
<tr>
<td>reference group</td>
<td>p-value</td>
<td>p-value</td>
</tr>
<tr>
<td>Range</td>
<td>Count</td>
<td>Energy</td>
</tr>
<tr>
<td>---------------</td>
<td>-------</td>
<td>--------</td>
</tr>
<tr>
<td>0.70-0.80</td>
<td>8</td>
<td>0.36</td>
</tr>
<tr>
<td>0.20-0.30</td>
<td>21</td>
<td>0.03</td>
</tr>
<tr>
<td>0.50-0.60</td>
<td>29</td>
<td>0.01</td>
</tr>
<tr>
<td>0.50-0.60</td>
<td>27</td>
<td>0.21</td>
</tr>
<tr>
<td>0.40-0.50</td>
<td>19</td>
<td>0.17</td>
</tr>
<tr>
<td>0.60-0.70</td>
<td>13</td>
<td>0.64</td>
</tr>
</tbody>
</table>

**S-7 COSMOconf template**

COSMOconf was used to create more conformers than in common applications by increasing the total number of possible conformers, reducing the energetic distance between conformers, and loosening the clustering steps.

```xml
<?xml version="1.0" encoding="UTF-8" standalone="no"?>
<job>
  <clean_up>1</clean_up>
  <info>BP-TZVPD-FINE-COSMO+GAS_new</info>
  <molecule_set_in>input_set.xml</molecule_set_in>
  <molecule_set_out extractable="join">output_set.xml</molecule_set_out>
  <job_schedule>
    <step>
      <number>1</number>
      <info>conf. creation using balloon</info>
      <method>BALLOON_CONF_GEN</method>
      <status>ready</status>
    </step>
    <step>
      <number>2</number>
      <info>consistency check (using UNIQUECODE)</info>
      <method>REDUCE_TO_UNIQUECODE</method>
      <status>ready</status>
    </step>
  </job_schedule>
  <reference>input_set.xml</reference>
</job>
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<step>
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  <method>SORT_BY_E</method>
  <status>ready</status>
</step>

<step>
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</step>

<step>
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  <energy_window>2000</energy_window>
  <n_max>150</n_max>
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<step>
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  <info>BP/SV(P) cosmo single points</info>
</step>
<step>
  <number>7</number>
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  <energy_window>25</energy_window>
  <n_max>50</n_max>
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<step>
  <number>8</number>
  <info>BP/SV(P) opt. with loose settings</info>
  <method>BP-SV_P-COSMO-LOOSE</method>
  <status>ready</status>
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<step>
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  <method>SORT_BY_E</method>
  <status>ready</status>
</step>

<step>
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<geodis_threshold2>1</geodis_threshold2>
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<step>
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  <geodis_threshold2>1</geodis_threshold2>
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<step>
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  <energy_window>25</energy_window>
  <n_max>50</n_max>
</step>

<step>
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  <method>PRINT_CONF_INFO</method>
  <status>ready</status>
</step>

<step>
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  <info/>
  <method>COPY_COSMO_FILE</method>
</step>
<step>
  <number>18</number>
  <info>BP-TZVPD-FINE COSMO single point</info>
  <method>BP-TZVPD-FINE-COSMO-SP</method>
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<step>
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<step>

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</step>

<step>

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<method>BP-TZVPD-GAS-SP</method>

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<step>

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<step>

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</step>

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<info/>

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</step>

</job_schedule>

</job>