

SUPPORTING INFORMATION

A Core Switching Strategy to Pyrrolo[2,3-*b*]quinolines and Diazocino[1,2-*a*]indolinones

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Scheme S1 Chiral memory experiment with enantio-enriched *M*-atropisomer of **1a**^{S1} with NaOMe lead to **±2a**

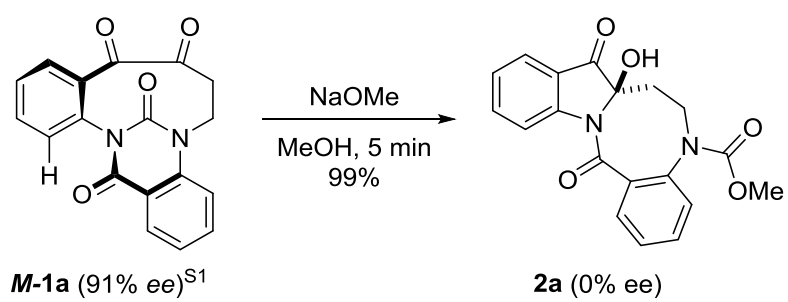


Table S1 Nucleophile screen of the scope of the transformation of **1a** to examples of **2**

Nucleophile	Expected m/z			Observed m/z
	$[M+H]^+$	$[M+Na]^+$	$[M-H]^-$	
sodium methoxide	353.3	375.3	351.3	351.3 ^a
<i>n</i> -butylamine	394.4	416.4	392.4	392.4 ^a
1,4-diaminobutane	409.4	431.4	407.4	407.4 ^a
acetylenediamine	423.4	445.4	421.4	421.4 ^a
Benzylamine	428.4	450.4	426.4	426.4 ^a
Aniline ^d	414.4	436.4	412.4	394.0 ^{*,d}
morpholine	408.4	430.4	406.4	406.4 ^a
(1 <i>S</i> ,2 <i>S</i>)-2-benzyloxycyclopentylamine ^d	512.6	534.6	510.6	492.0 ^{*,d}
<i>N</i> -methylhydrazine	367.4	389.4	365.4	365.4 ^a
benzyl carbazate	487.5	509.5	485.5	485.5 ^a
sodium azide	364.3	386.3	362.3	362.3 ^a
glycine <i>t</i> -butyl ester. HCl ^d	452.5	474.5	450.5	466.0 ^{*,d}
propan-2-ol ^c	381.4	403.4	379.4	319.0 ^c
propargyl alcohol ^c	377.4	399.4	375.4	319.0 ^c
methylithium.LiBr complex	337.3	359.3	335.3	335.3 ^a
allyl magnesium bromide	363.4	385.4	361.4	361.4 ^a
ethynyl magnesium chloride	347.3	369.3	345.3	345.3 ^a
methyl magnesium bromide	337.3	359.3	335.3	335.3 ^a
trimethylsilyl cyanide ^c	348.3	370.3	346.3	319.0 ^c
triphenylmethanethiol	597.7	619.7	595.7	595.7 ^a
Ethylthioglycolate ^d	427.4	449.4	425.4	439.0 ^{*,d}
Benzylmercaptan	445.5	467.5	443.5	445.5 ^b

Details: Using a Radleys® Greenhouse parallel synthesiser, to 18 stirred solutions of **1a** (20 mg) in dry THF (5mL) were added 1.1 equivalents of the relevant nucleophile at room temperature for 16 hours. In a separate study, the four reactions using Grignard reactions were carried out at -78 °C and then allowed to warm to 0 °C after 3 hours. LCMS analysis (ES^+/ES^-) of the crude reaction mixtures revealed that 14 / 22 reaction mixtures contained a new product with a m/z from either the ES^- (^a in table) or ES^+ (^b in table) spectra coupled with the UV_{254nm} trace;

c no reaction was observed and observed m/z consistent with the presence of starting material **1a**; ^d structure of product not determined.

Table S2 Optimisation of the sodium methoxide-induced rearrangement of **1a** to **2a**

Entry	NaOMe (eq.)	Solvent	Temperature (°C)	Time (min)	Yield (%)
1	1	MeOH	25	10	95
2	2	MeOH	25	10	99
3	10	MeOH	25	10	95
4	2	THF	25	960	45

Details: Variation of reaction conditions and the resulting isolated yields for the methoxide induced azepinoindole rearrangement of **1a** to give **2a**.

Table S3 Optimisation of the butylamine-induced rearrangement of **1a** to **2i**

entry	eq. <i>n</i> BuNH ₂	time (min)	percentage conversion
1	1	10	40
2	1	60	80
3	1	180	95 (74 ^a)
4	1	960	^b
5	2	180	70 (51 ^a)
6	4	180	^b
7	10	180	^b

Details: Percentage conversion as estimated by ¹H NMR spectra of the transformation of **1a** to **2i**. ^aIsolated yield (after flash column chromatography); ^bSubstantial degradation was observed.

Table S4 ^1H and ^{13}C NMR based assignment of **4** (Derived from 2D ^1H - ^1H COSY, ^1H - ^{13}C HSQC and ^1H - ^{13}C HMBC NMR experiments).

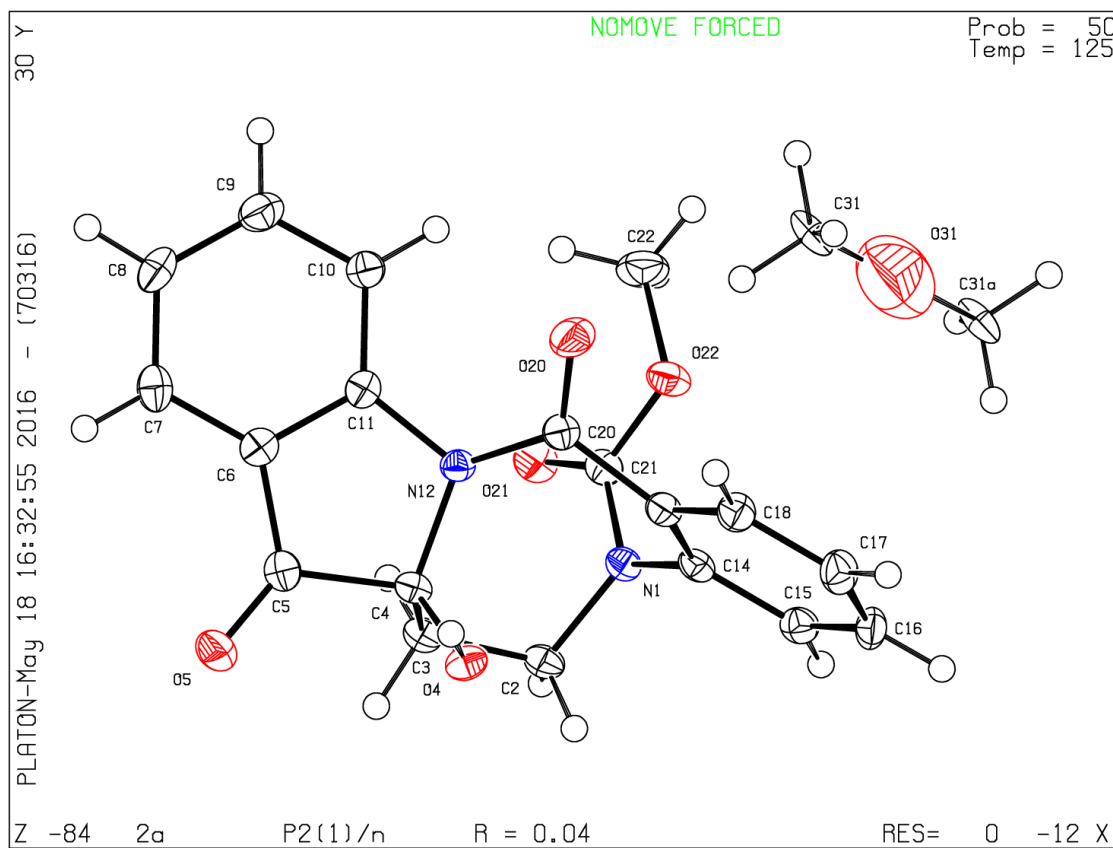
Proton Assignment	δ ^1H (ppm)	Multiplicity	Integration	J value (Hz)	Carbon Assignment	δ ^{13}C (ppm)
C2- <u>H</u> ₂	4.19	t	2H	7.5	C2	49.8
C3- <u>H</u> ₂	3.36	t	2H	7.4	C3	25.3
					C3a	119.6
C4-O <u>H</u>	8.55	br s	1H	-	C4	153.1
					C5	178.3
					C5a	126.5
C6- <u>H</u>	8.42	d	1H	2.6	C6	129.1
					C7	129.6
C8- <u>H</u>	7.41	dd	1H	8.9, 2.6	C8	134.8
C9- <u>H</u>	7.28	d	1H	8.8	C9	134.7
					C9a	148.6
					C10a	157.8
					C1'	139.1
					C2'	129.6
C3'- <u>H</u>	7.90	dd	1H	7.7, 1.5	C3'	130.5
C4'- <u>H</u>	7.37	dd	1H	7.6, 1.2	C4'	126.5
C5'- <u>H</u>	7.58	ddd	1H	7.7, 7.7, 1.7	C5'	132.5
C6'- <u>H</u>	7.32	dd	1H	8.0, 1.0	C6'	125.4
					ArC(=O)OEt	167.0
OCH ₂ CH ₃	3.96	q	2H	7.2	OCH ₂ CH ₃	61.1
OCH ₂ CH ₃	1.06	q	3H	7.2	OCH ₂ CH ₃	14.1

Table S5 Optimisation of the sodium alkoxide-induced rearrangement of **1e** to **3a**

Entry	NaOMe (eq.)	Solvent	Temperature (°C)	Time (h)	Yield (%)
1	2	MeOH	25	16	77
2	2	MeOH	64	16	59
3	10	MeOH	25	16	71

Details: Variation of the reaction conditions for the alkoxide induced pyrrolo[2,3-*b*]quinoline rearrangement **3a** from **1e** and the resulting isolated yields.

Table S6 Crystal data and structure refinement details for compound **2a**.



Empirical formula

C_{19.50} H₁₈ N₂ O_{5.50}

Formula weight

368.36

Temperature

125(2) K

Wavelength

0.71073 Å

Crystal system

Monoclinic

Space group

P2(1)/n

Unit cell dimensions

a = 7.7594(12) Å

α = 90°.

b = 19.459(3) Å

β = 106.142(2)°.

c = 11.8044(19) Å

γ = 90°.

Volume

1712.1(5) Å³

Z

4

Density (calculated)

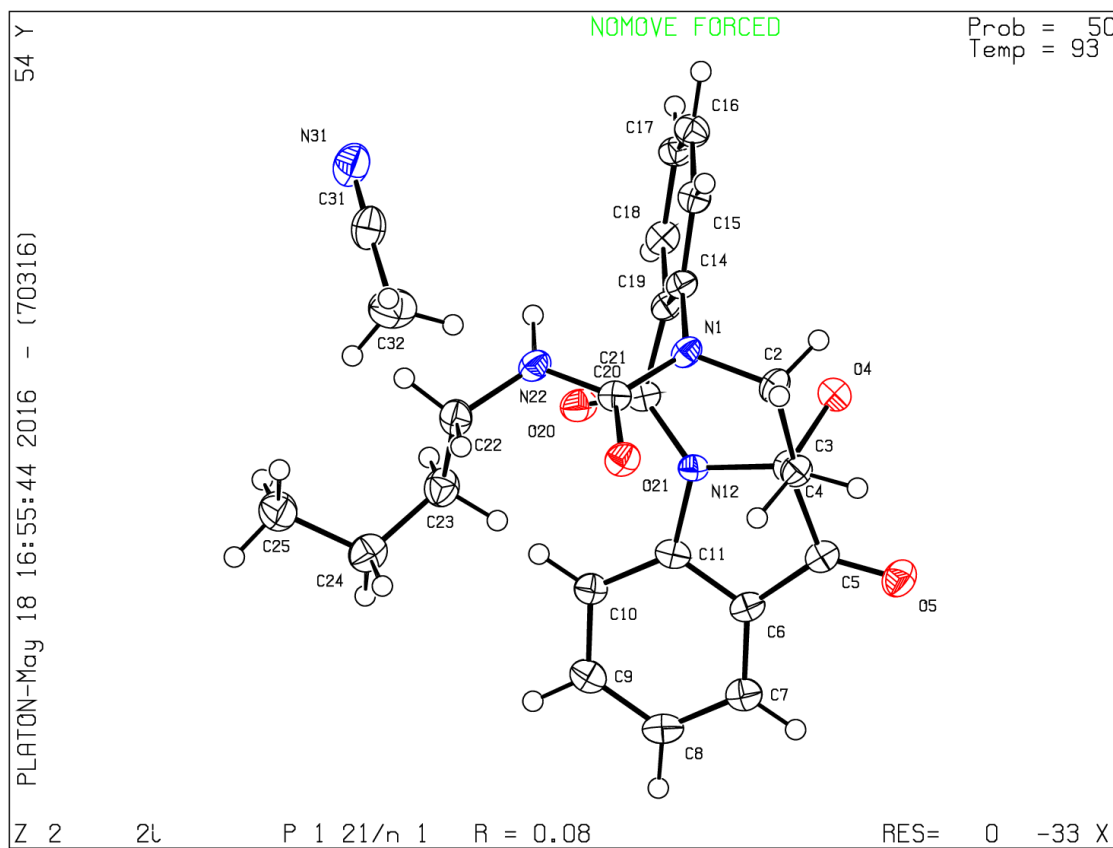
1.429 Mg/m³

Absorption coefficient

0.106 mm⁻¹

F(000)	772
Crystal size	.1 x .1 x .01 mm ³
Theta range for data collection	2.08 to 25.49°.
Index ranges	-9<=h<=9, -23<=k<=16, -12<=l<=14
Reflections collected	9839
Independent reflections	3068 [R(int) = 0.0244]
Completeness to theta = 25.49°	96.2 %
Absorption correction	MULTISCAN
Max. and min. transmission	1.00000 and 0.891243
Refinement method	Full-matrix least-squares on F ²
Data / restraints / parameters	3068 / 1 / 257
Goodness-of-fit on F ²	1.024
Final R indices [I>2sigma(I)]	R1 = 0.0416, wR2 = 0.0971
R indices (all data)	R1 = 0.0552, wR2 = 0.1051
Extinction coefficient	0.0039(9)
Largest diff. peak and hole	0.455 and -0.445 e.Å ⁻³

Table S7 Crystal data and structure refinement details for compound **2i**.



Empirical formula

C₂₄ H₂₆ N₄ O₄

Formula weight

434.49

Temperature

93(2) K

Wavelength

0.71073 Å

Crystal system

Monoclinic

Space group

P2(1)/n

Unit cell dimensions

a = 10.8627(13) Å $\angle = 90^\circ$.

b = 19.295(2) Å $\angle = 111.323(6)^\circ$.

c = 11.3504(14) Å $\angle = 90^\circ$.

Volume

2216.1(5) Å³

Z

4

Density (calculated)

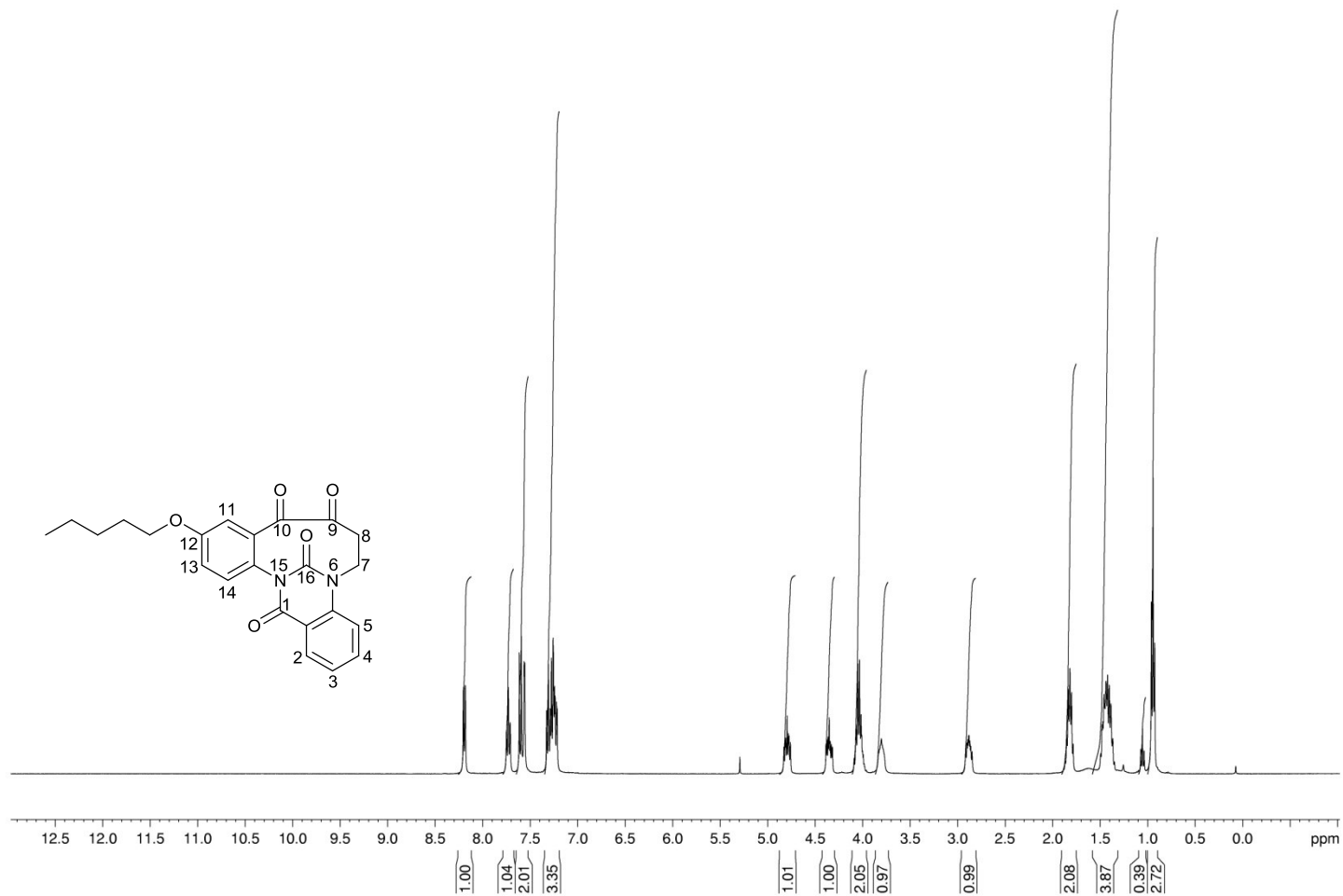
1.302 Mg/m³

Absorption coefficient

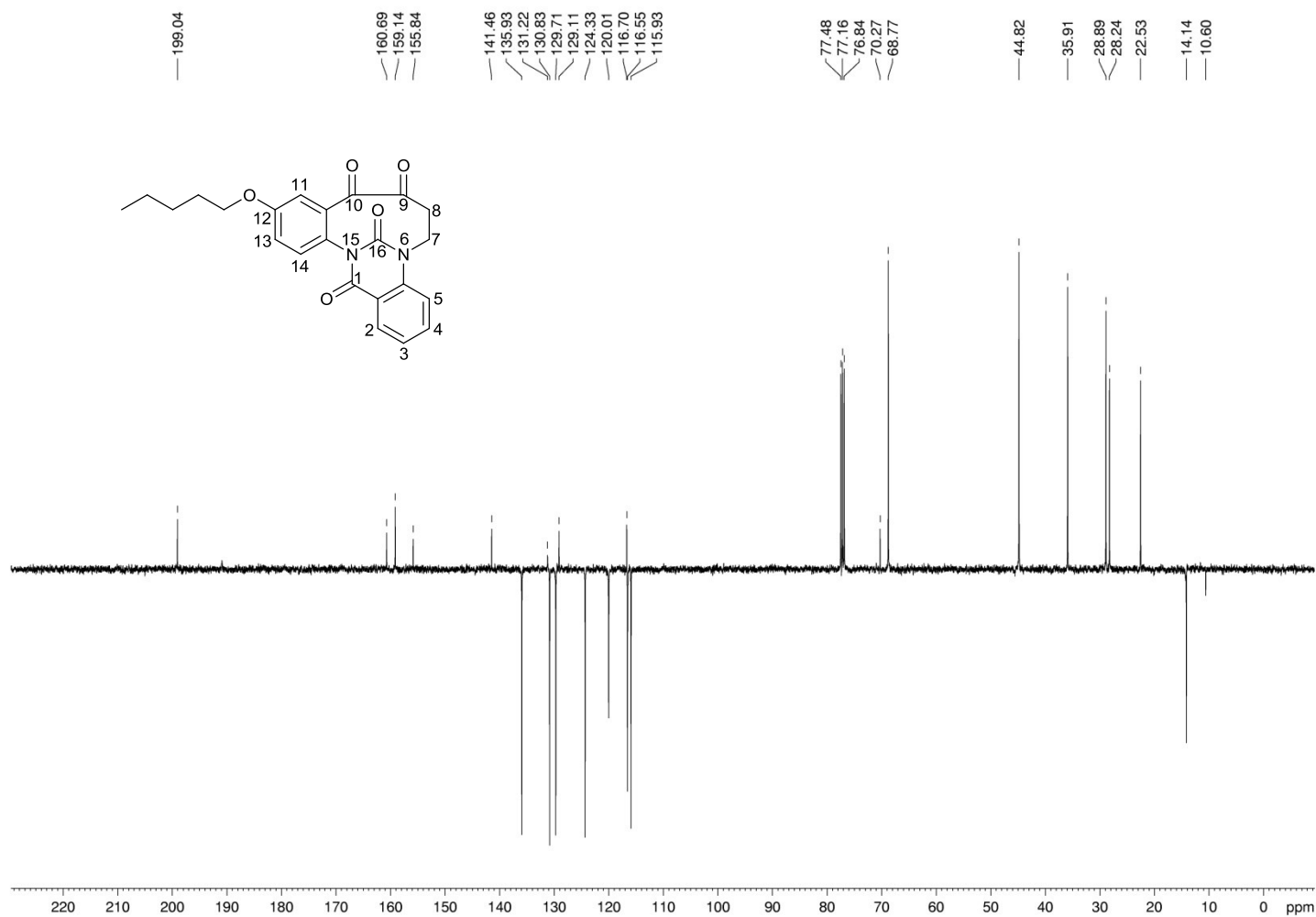
0.090 mm⁻¹

F(000)	920
Crystal size	0.200 x 0.010 x 0.010 mm ³
Theta range for data collection	2.11 to 25.34°.
Index ranges	-13<=h<=12, -16<=k<=23, -12<=l<=13
Reflections collected	15426
Independent reflections	3727 [R(int) = 0.0552]
Completeness to theta = 25.34°	92.0 %
Absorption correction	Multiscan
Max. and min. transmission	1.0000 and 0.8690
Refinement method	Full-matrix least-squares on F ²
Data / restraints / parameters	3727 / 2 / 300
Goodness-of-fit on F ²	1.192
Final R indices [I>2sigma(I)]	R1 = 0.0654, wR2 = 0.1112
R indices (all data)	R1 = 0.0893, wR2 = 0.1211
Extinction coefficient	0.0060(8)
Largest diff. peak and hole	0.234 and -0.224 e.Å ⁻³

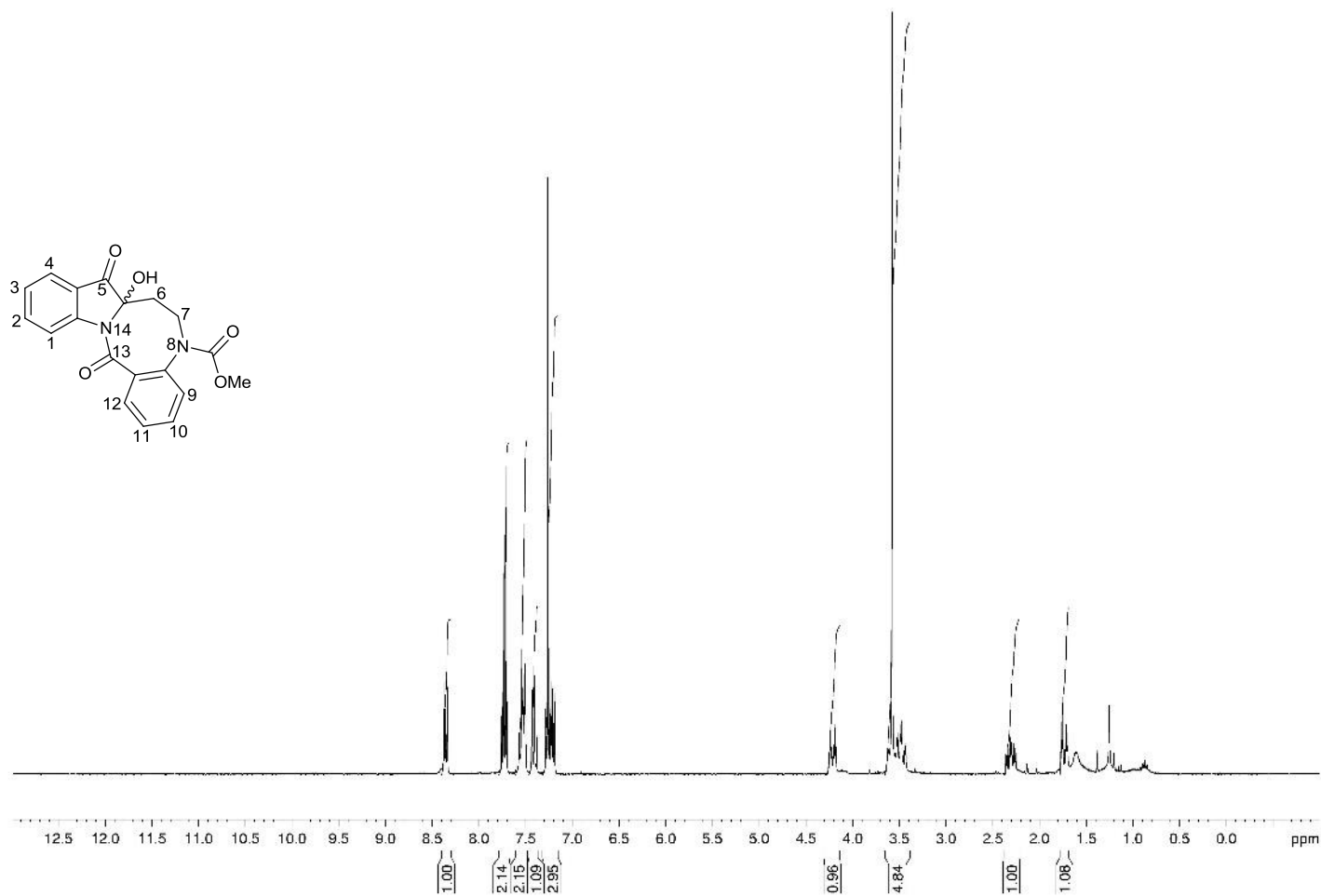
Copies of ^1H and ^{13}C NMR spectra: **Compound 1b** ^1H NMR (CDCl_3)



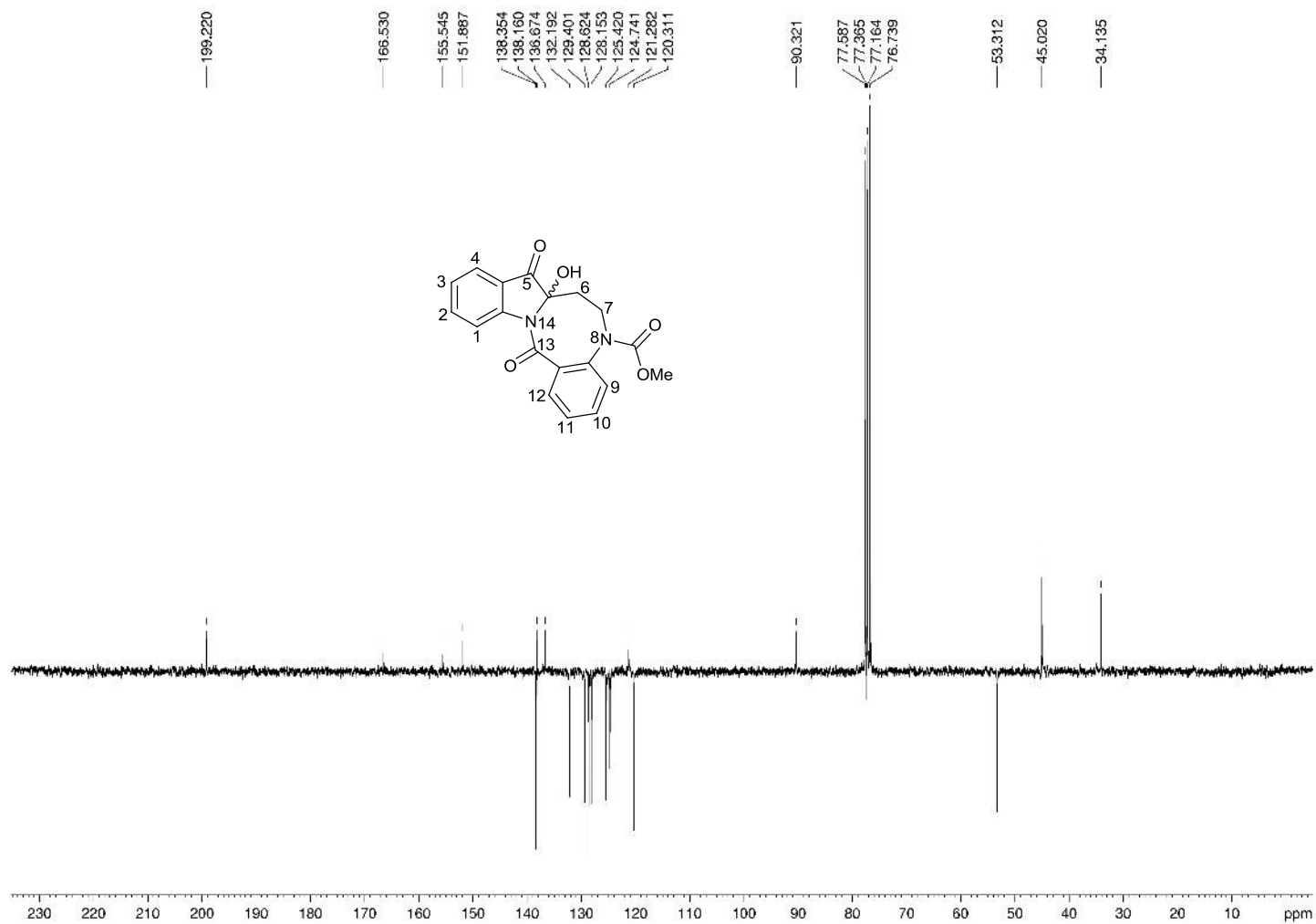
Compound 1b ^{13}C NMR (CDCl_3)



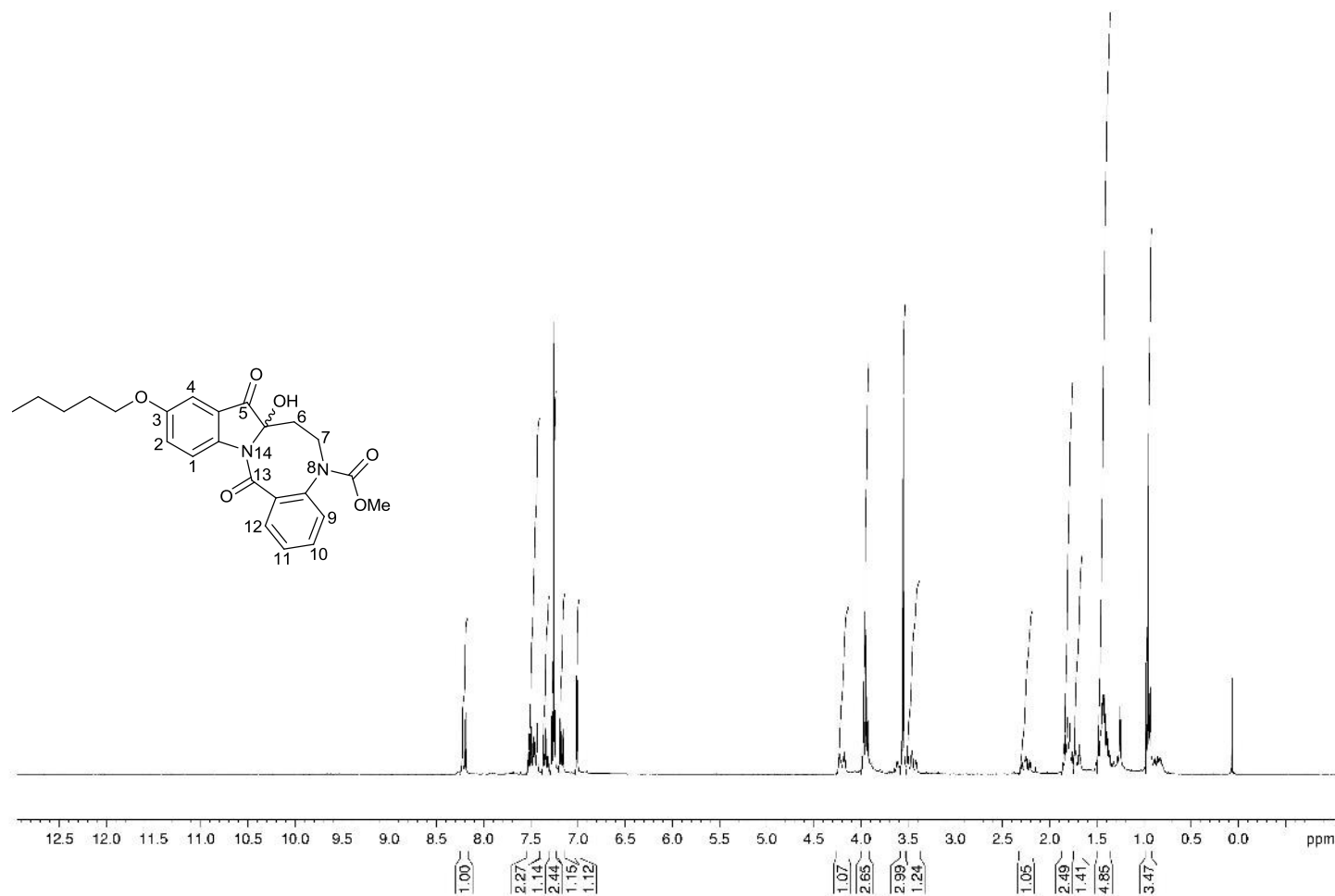
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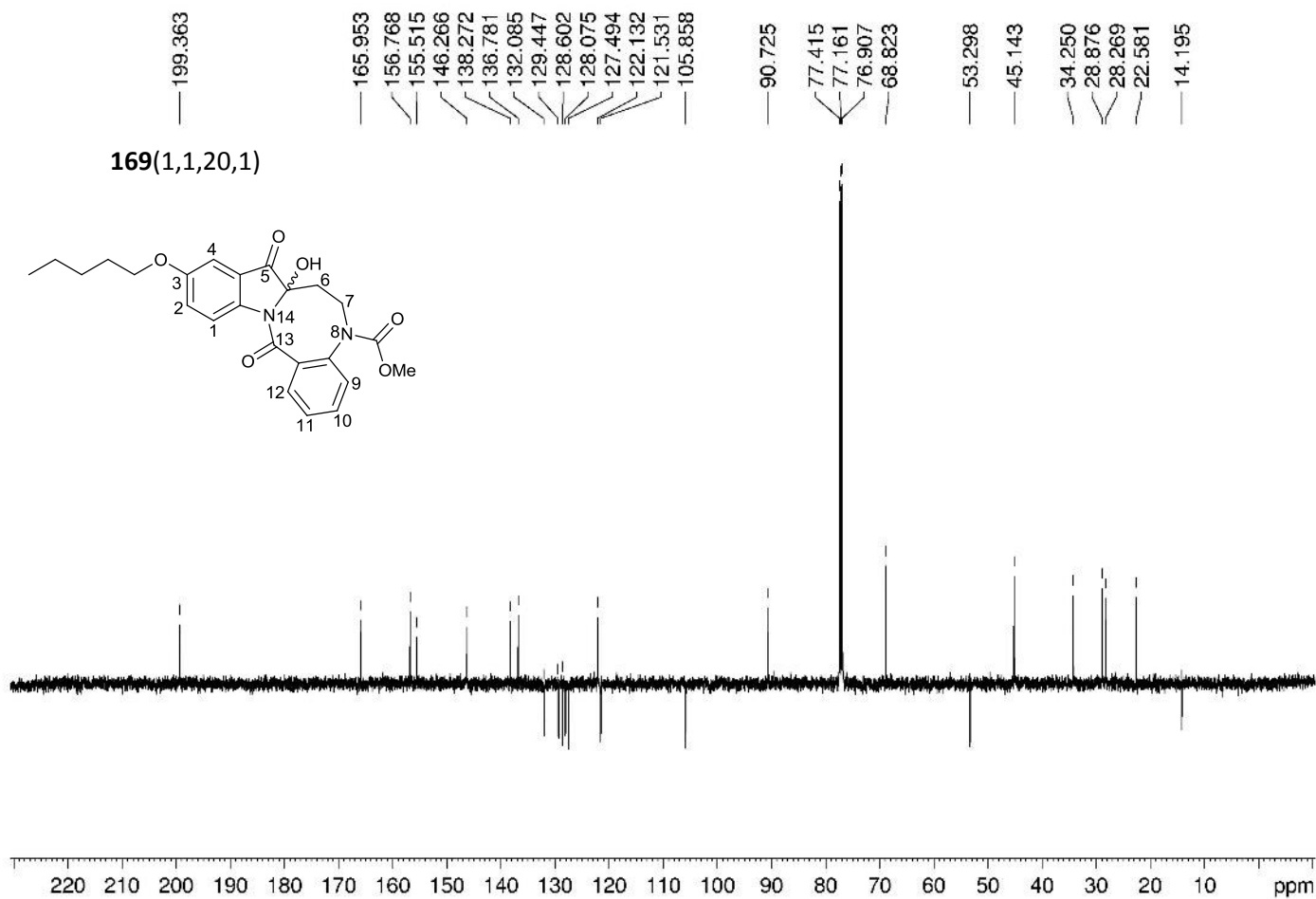
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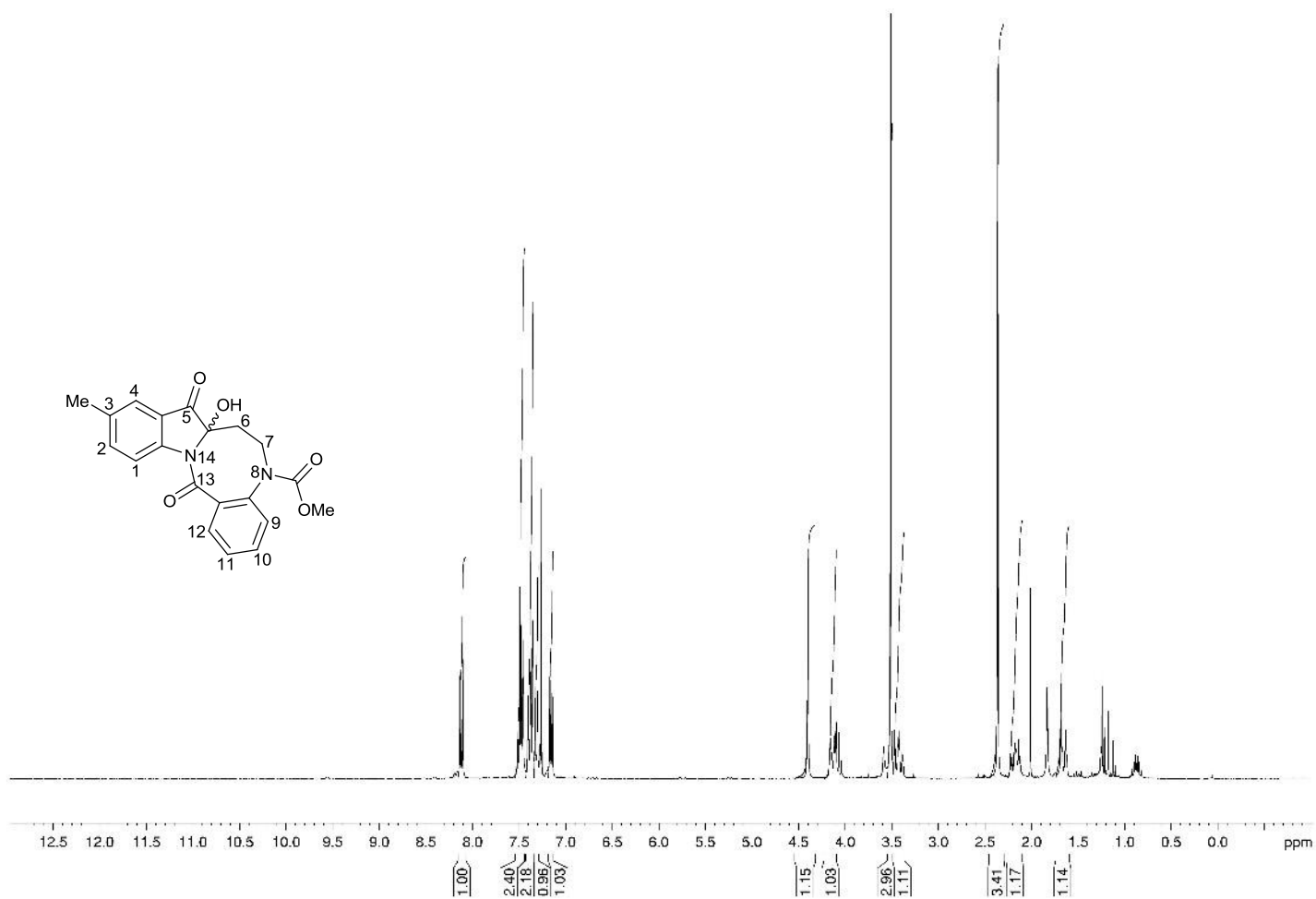
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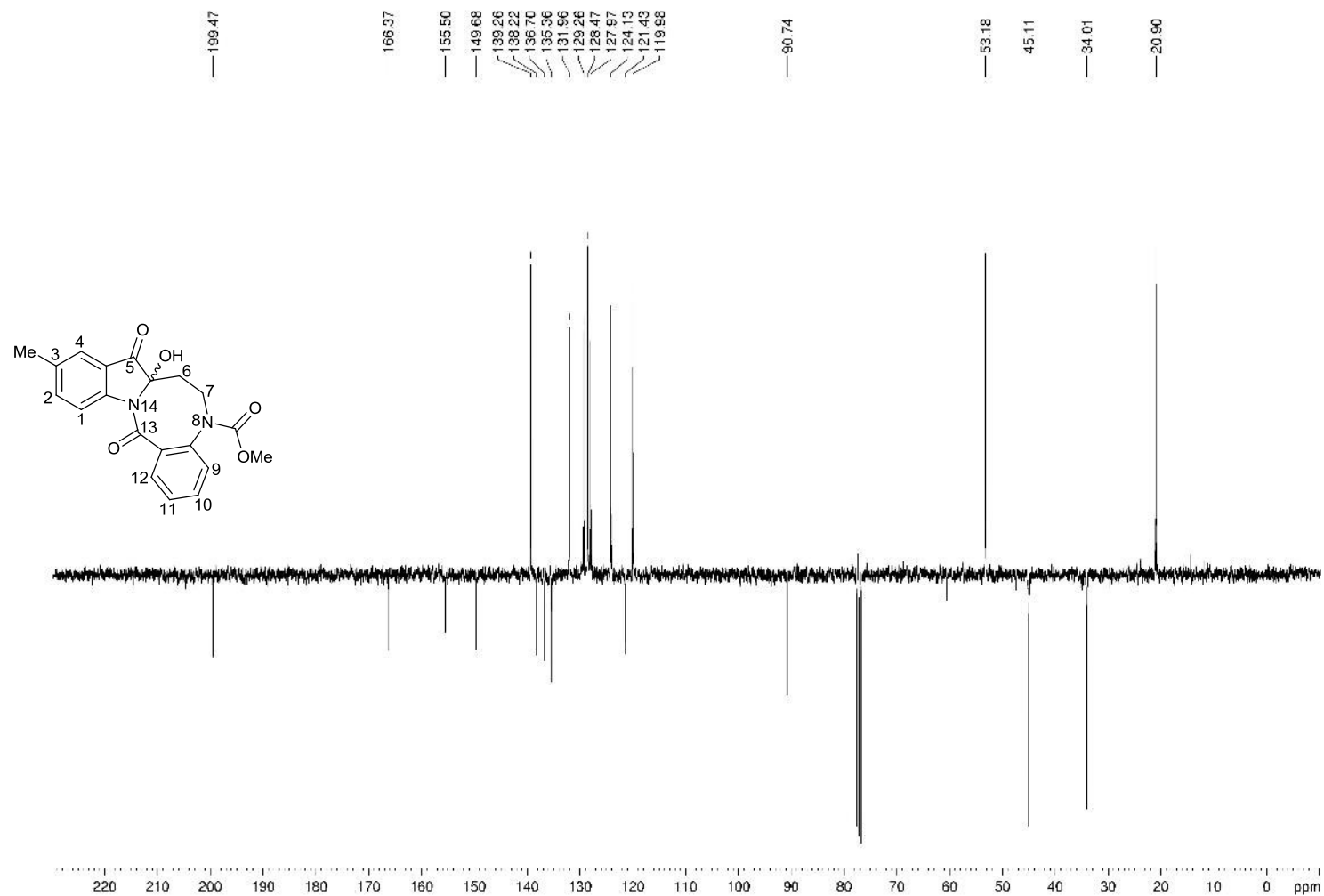
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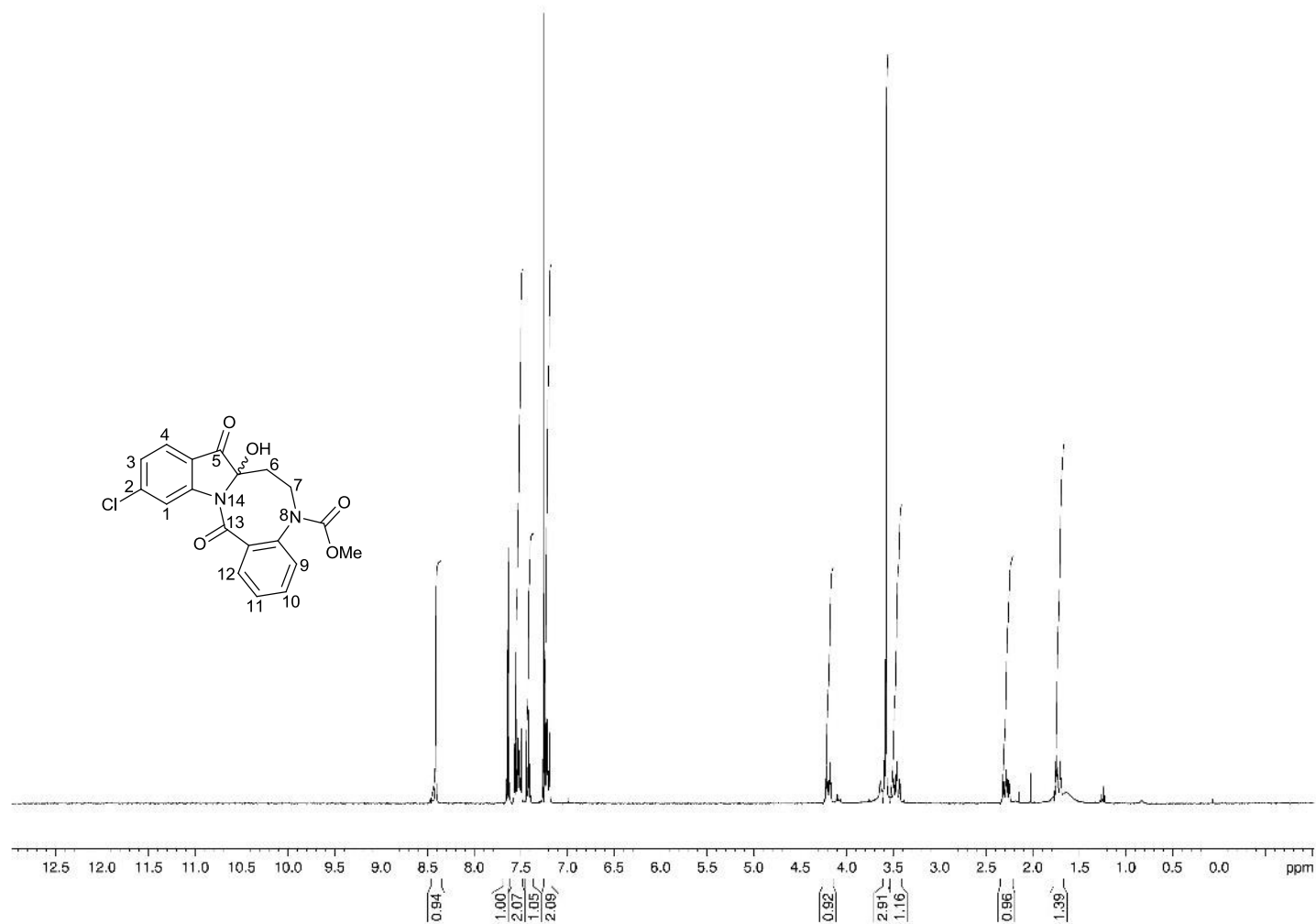
Compound 2c ^1H NMR (CDCl_3)



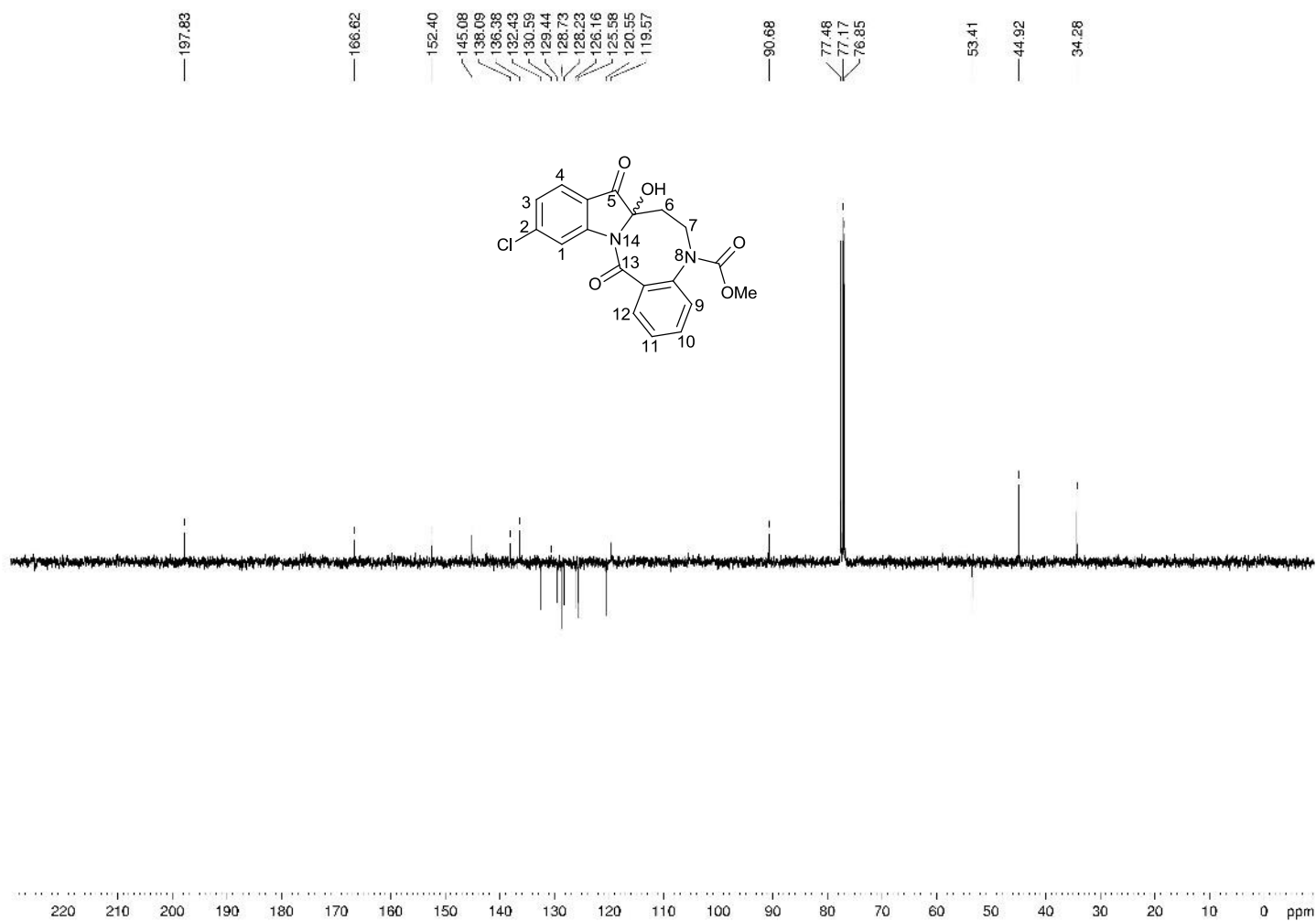
Compound 2c ^{13}C NMR (CDCl_3)



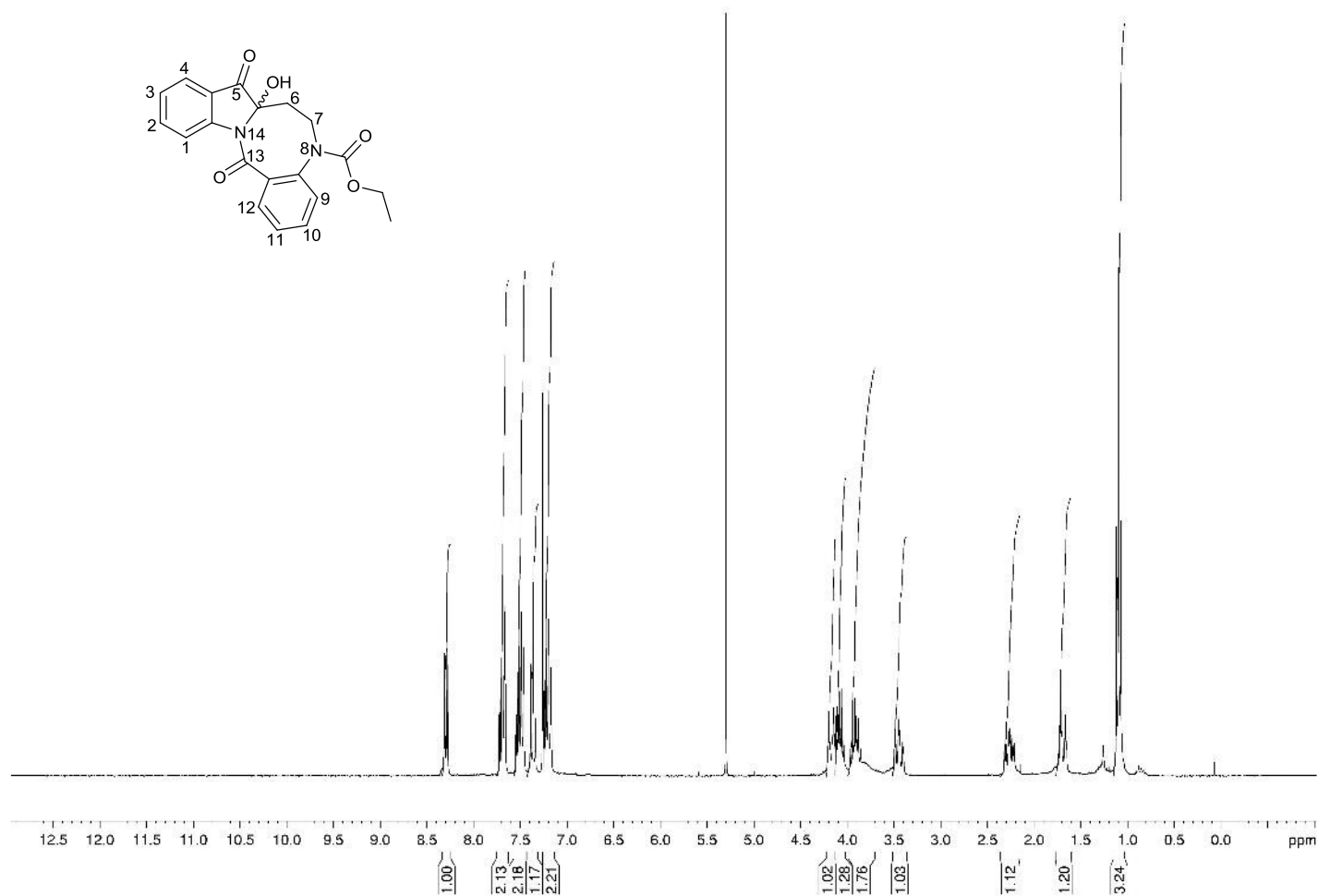
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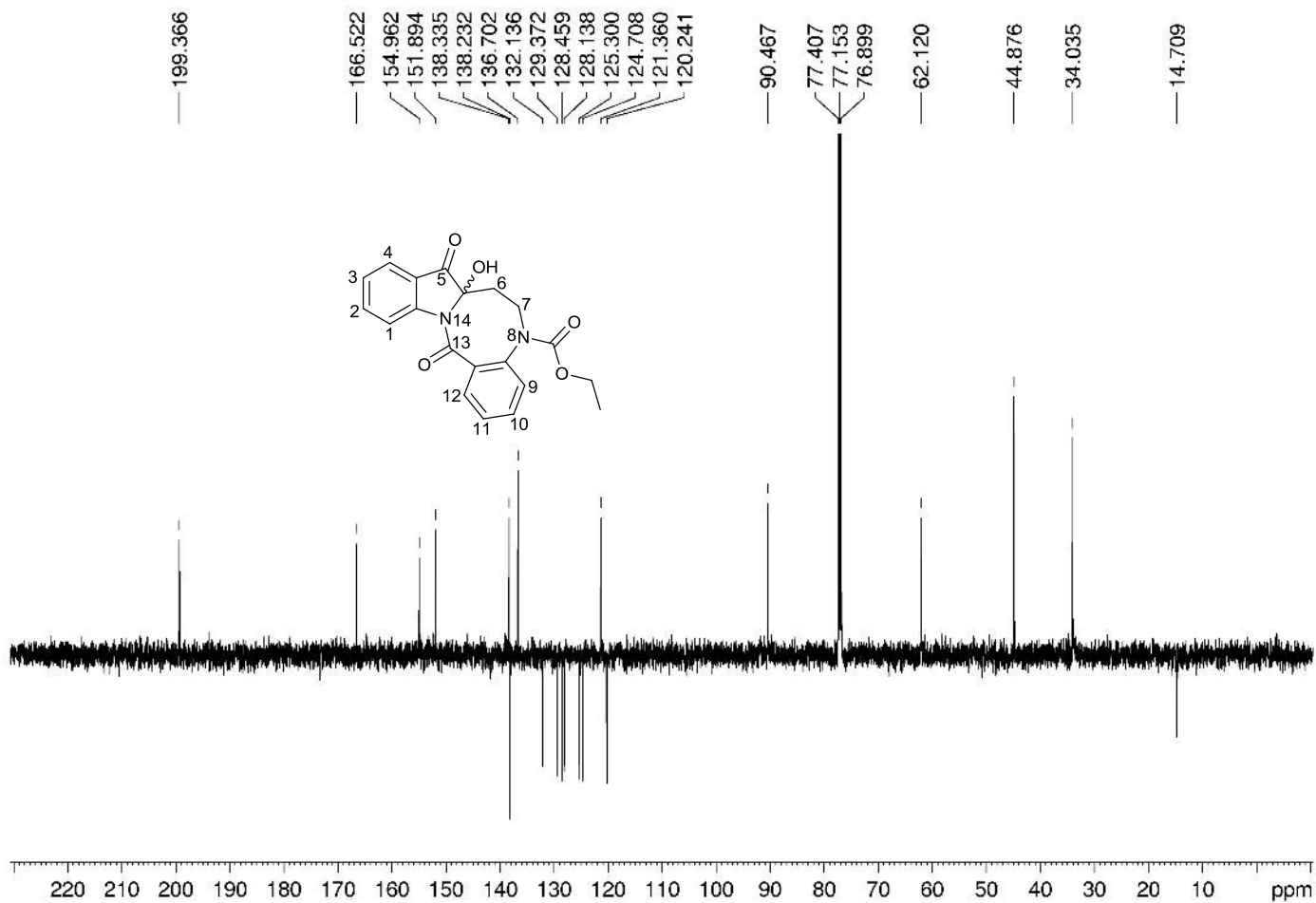
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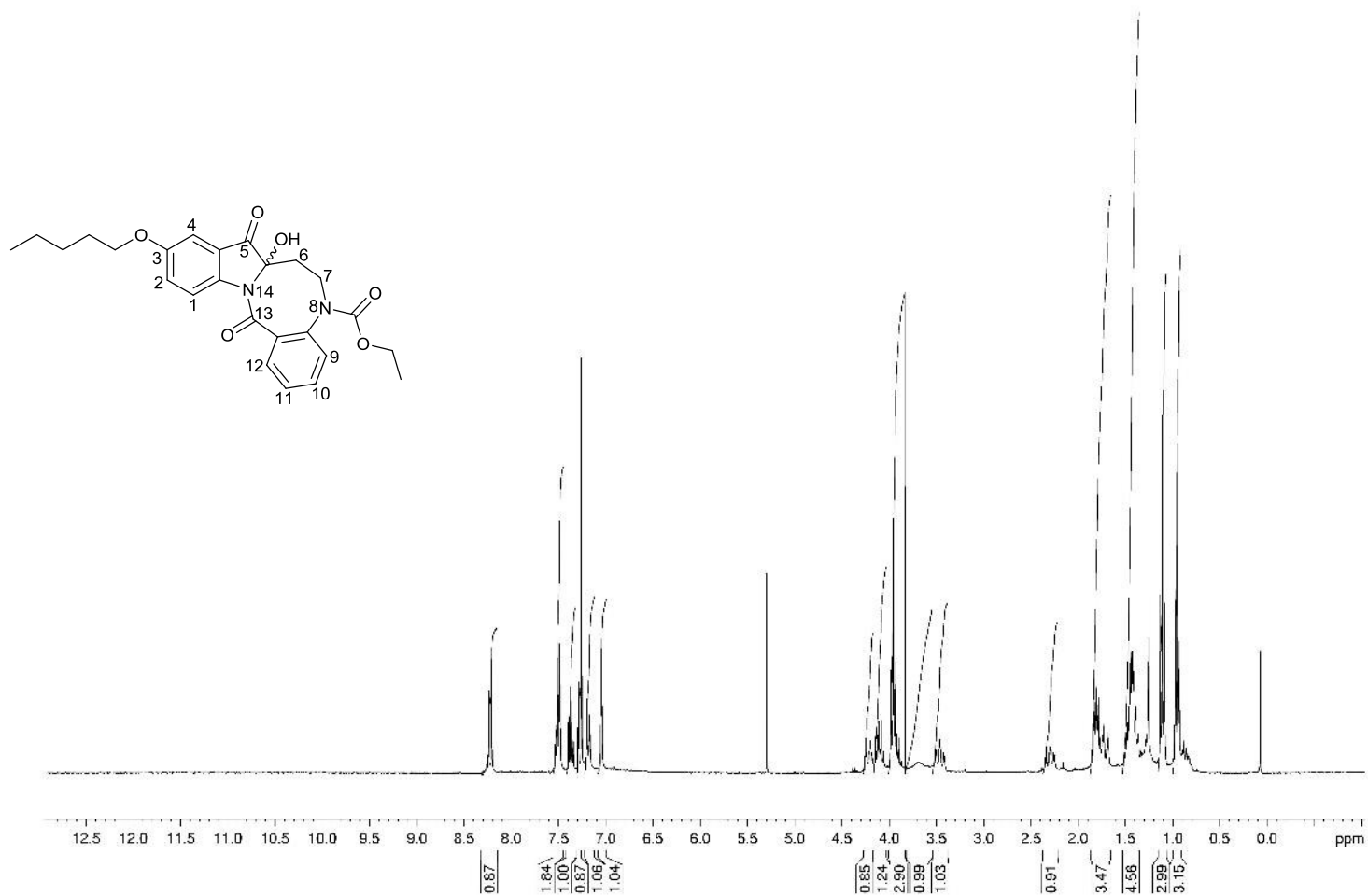
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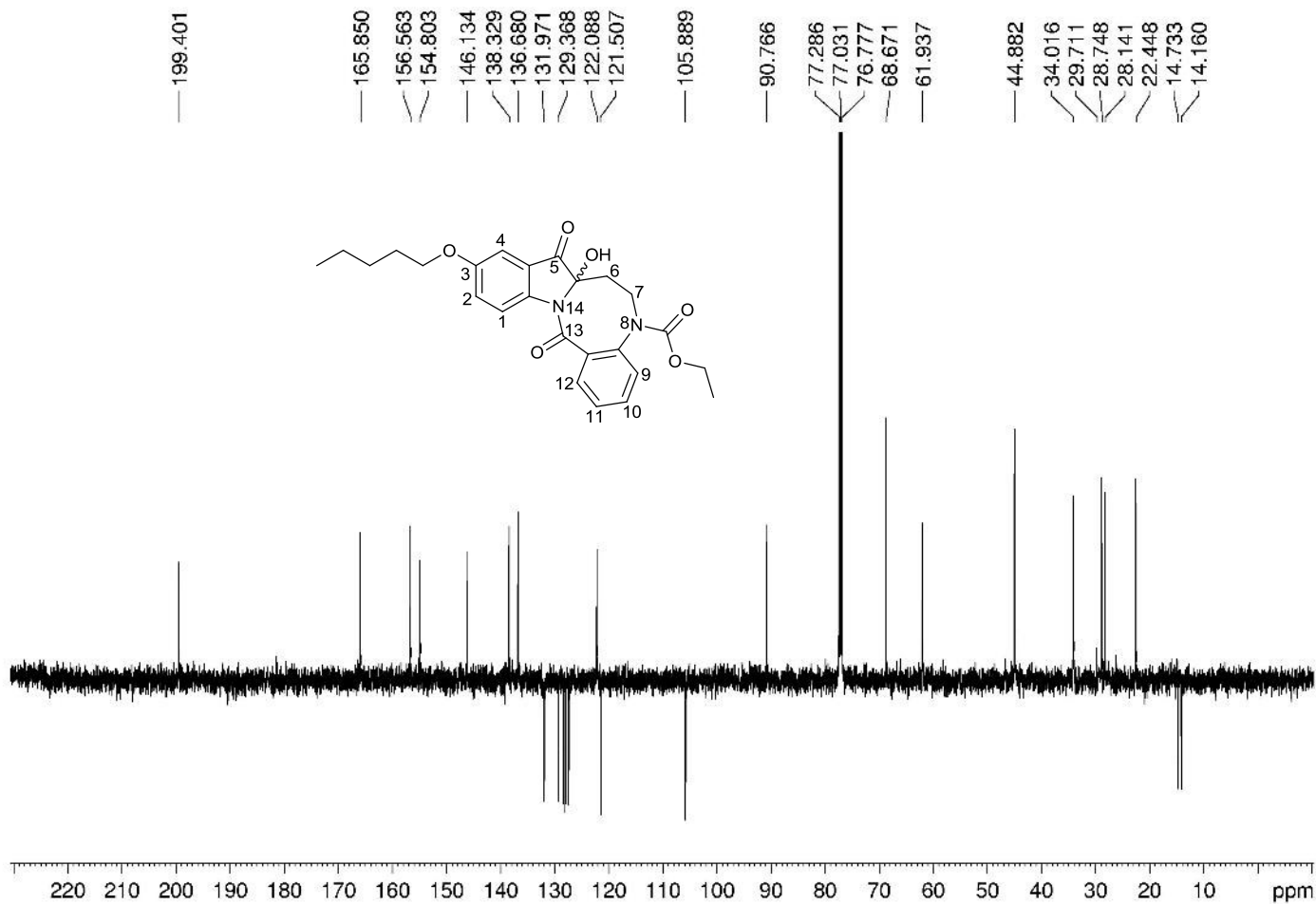
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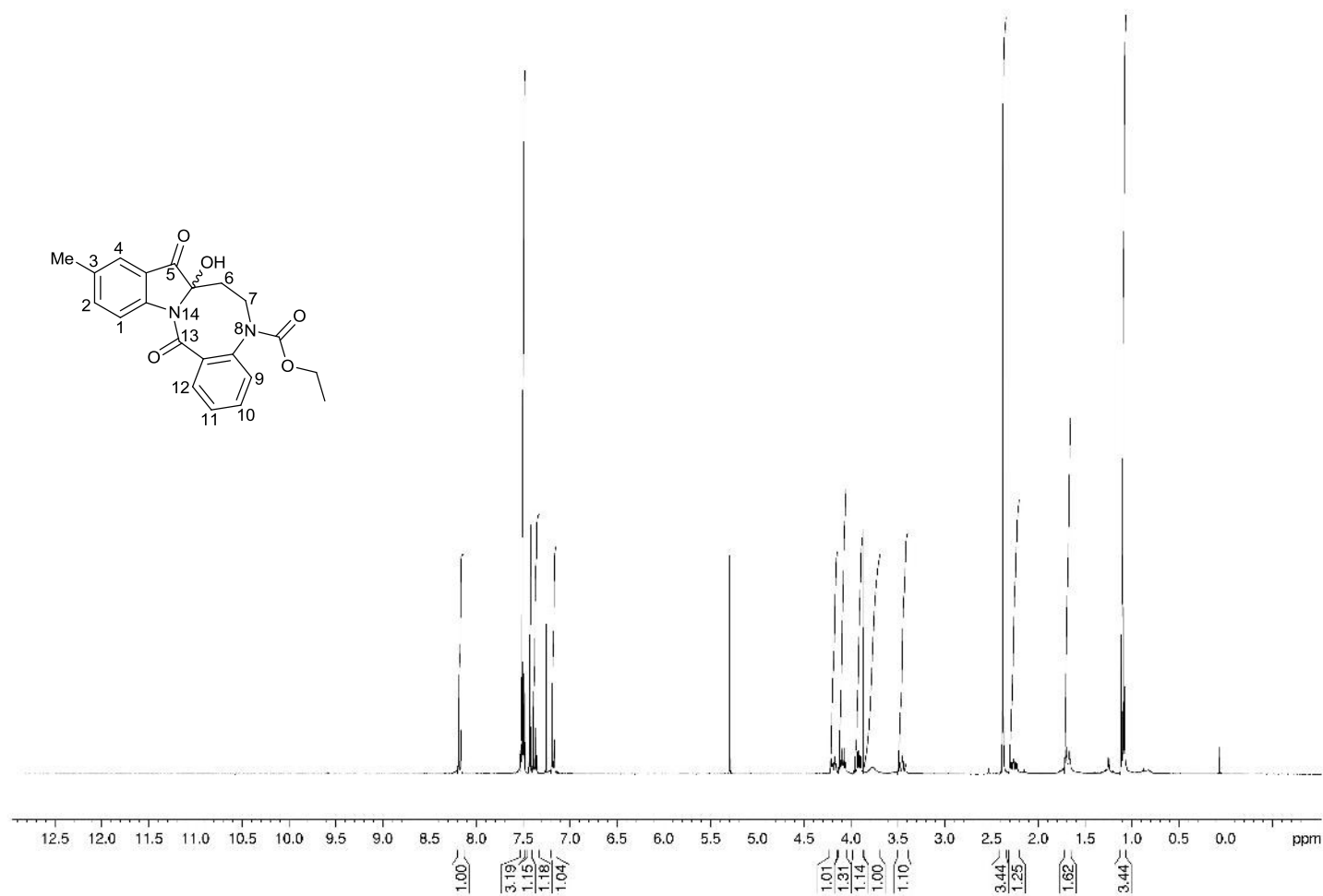
Compound 2f ^1H NMR (CDCl_3)



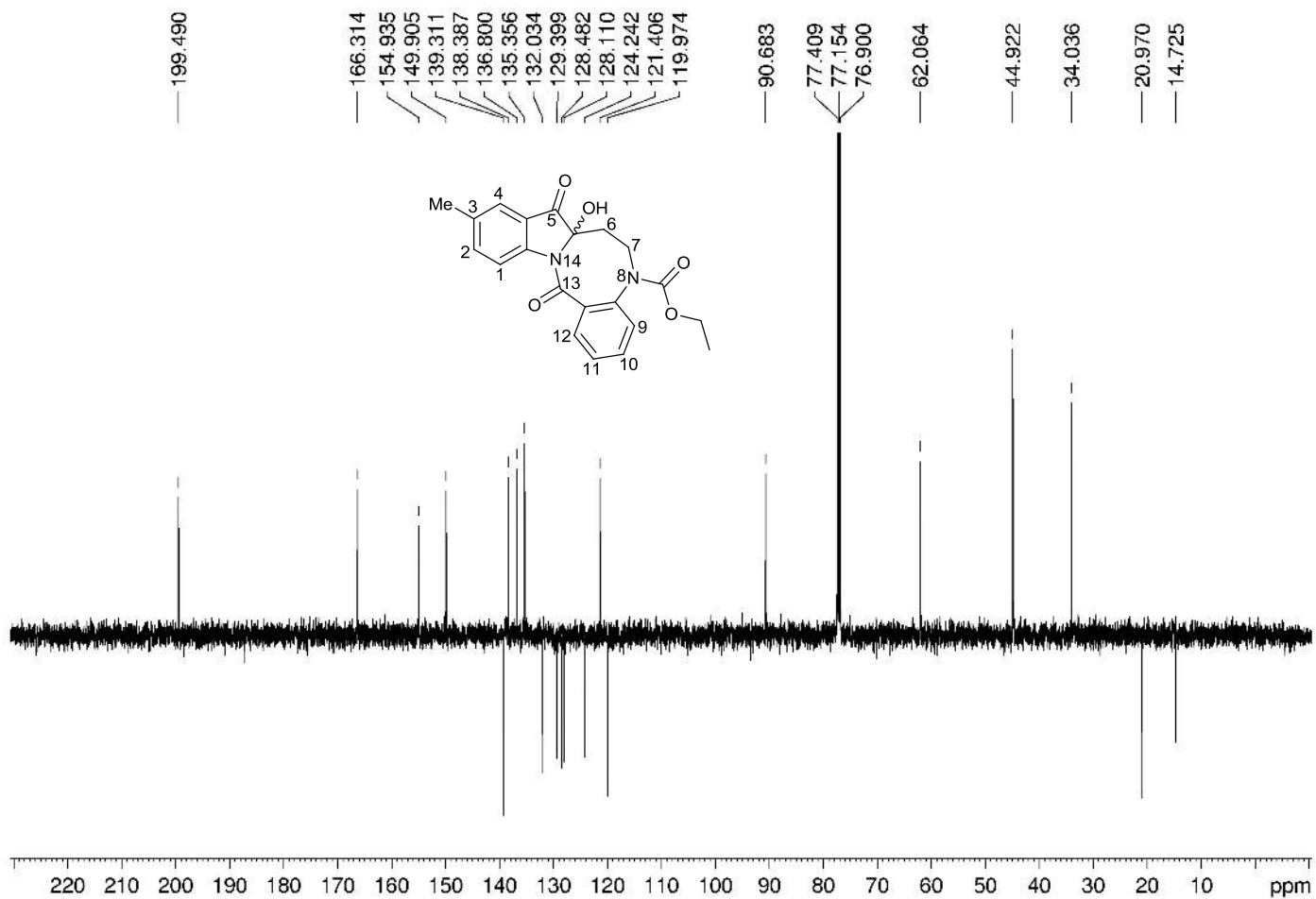
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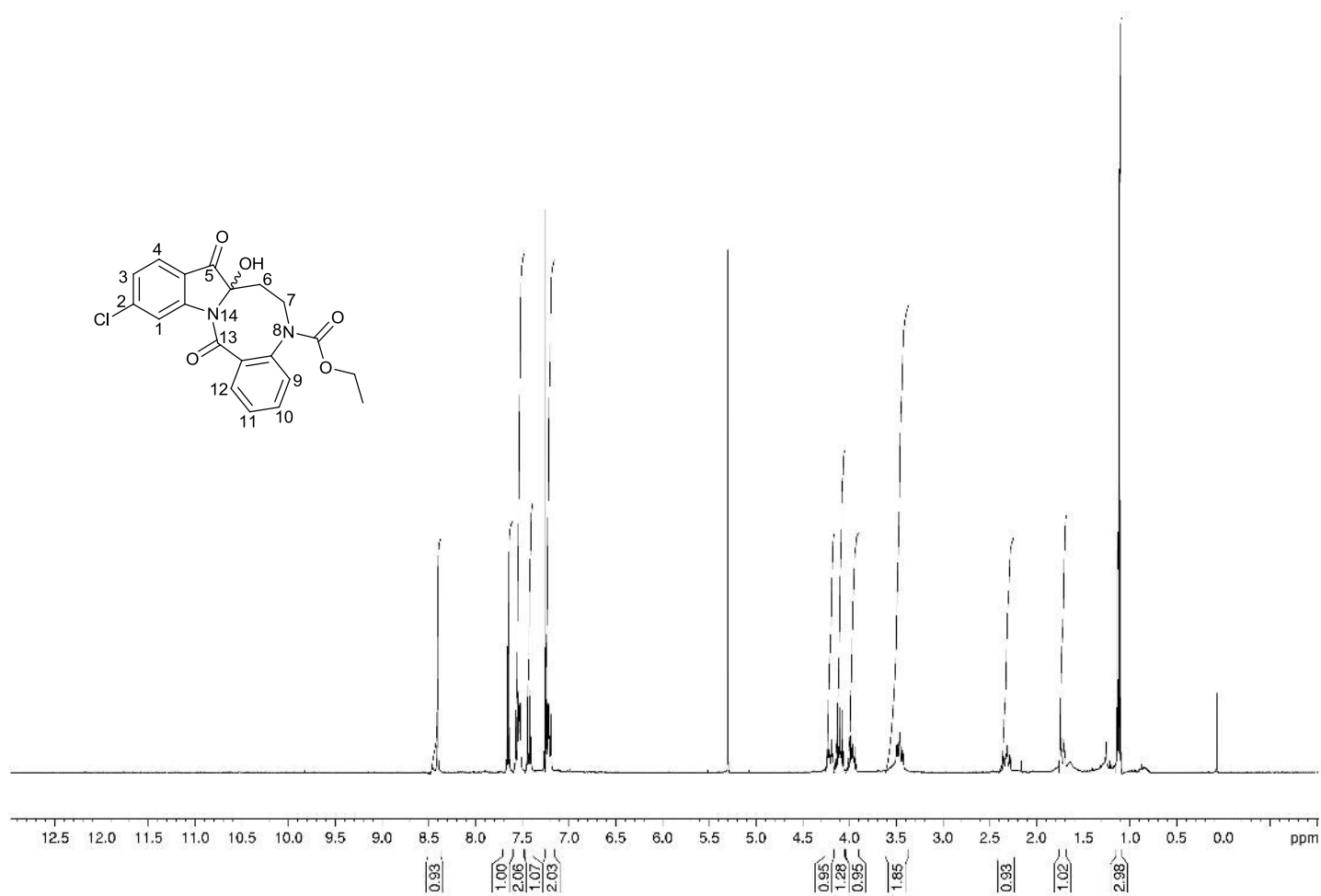
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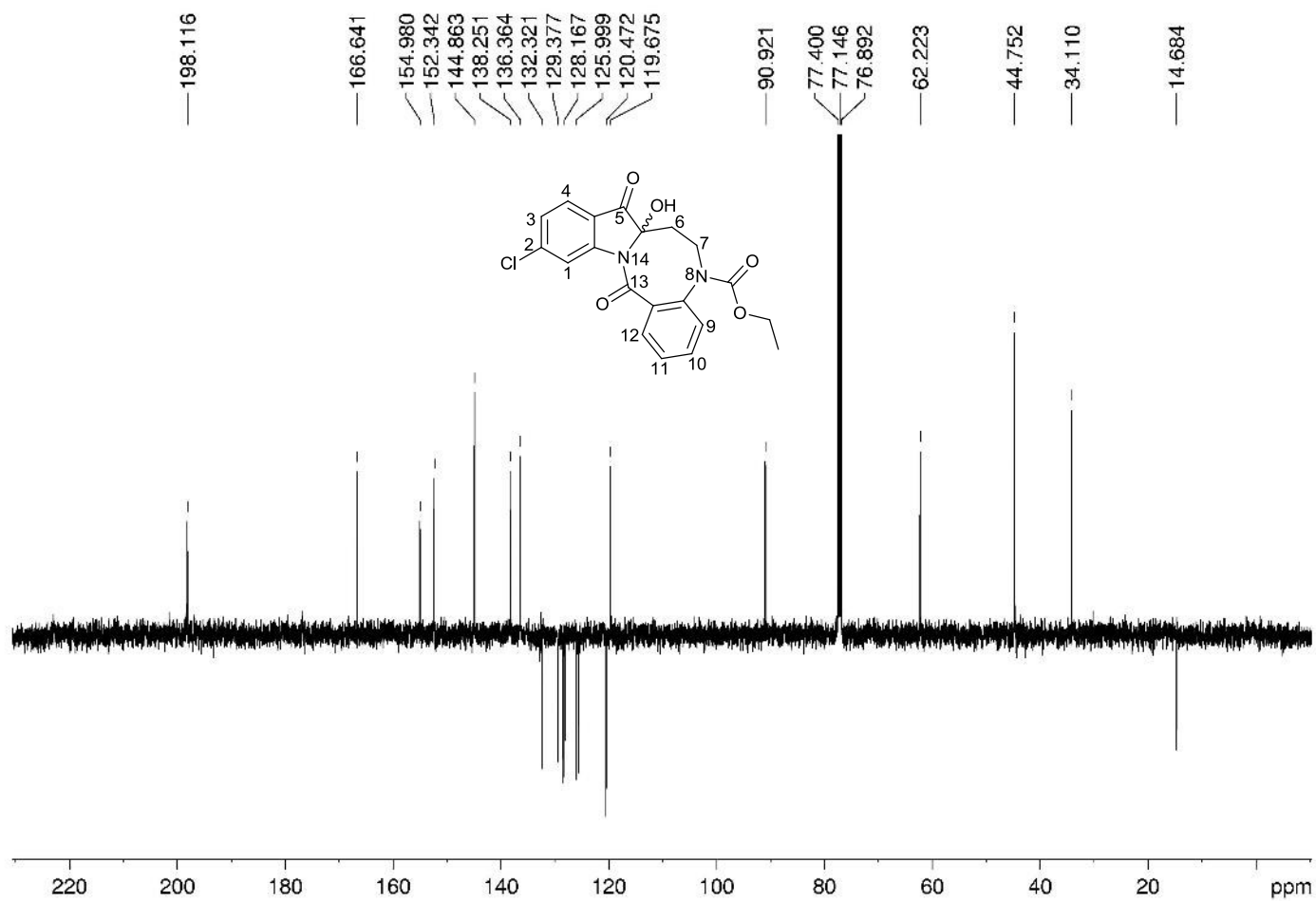
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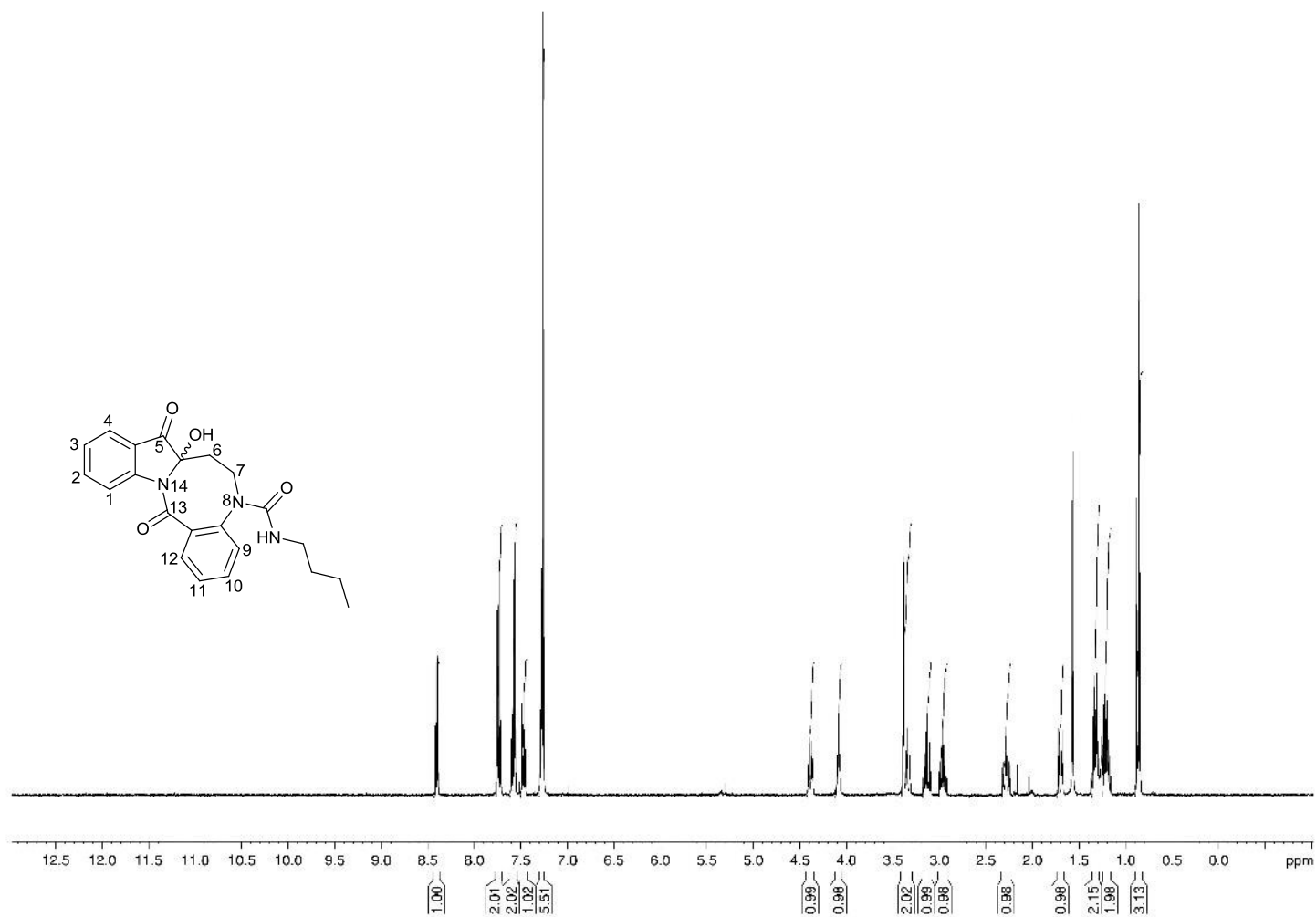
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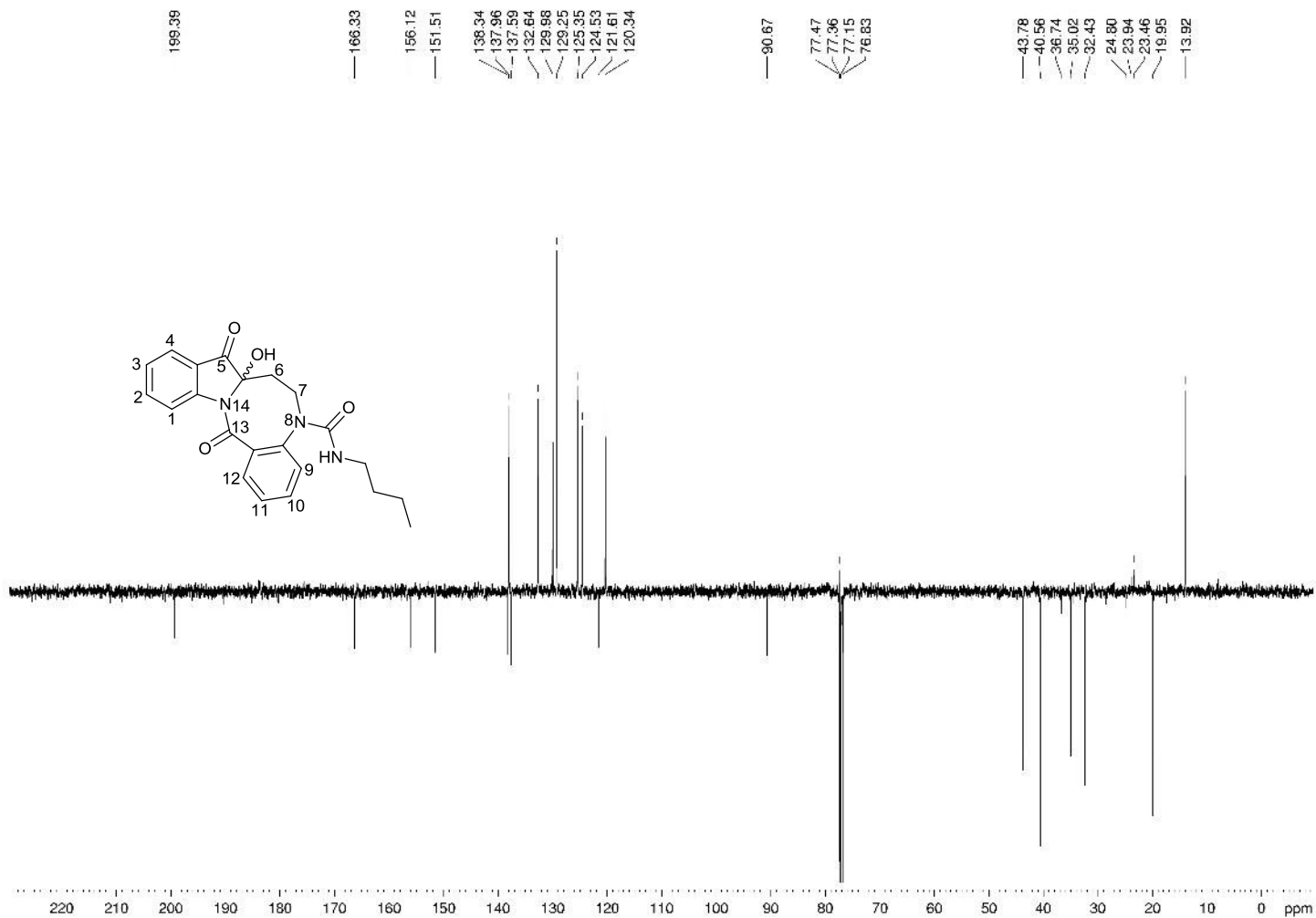
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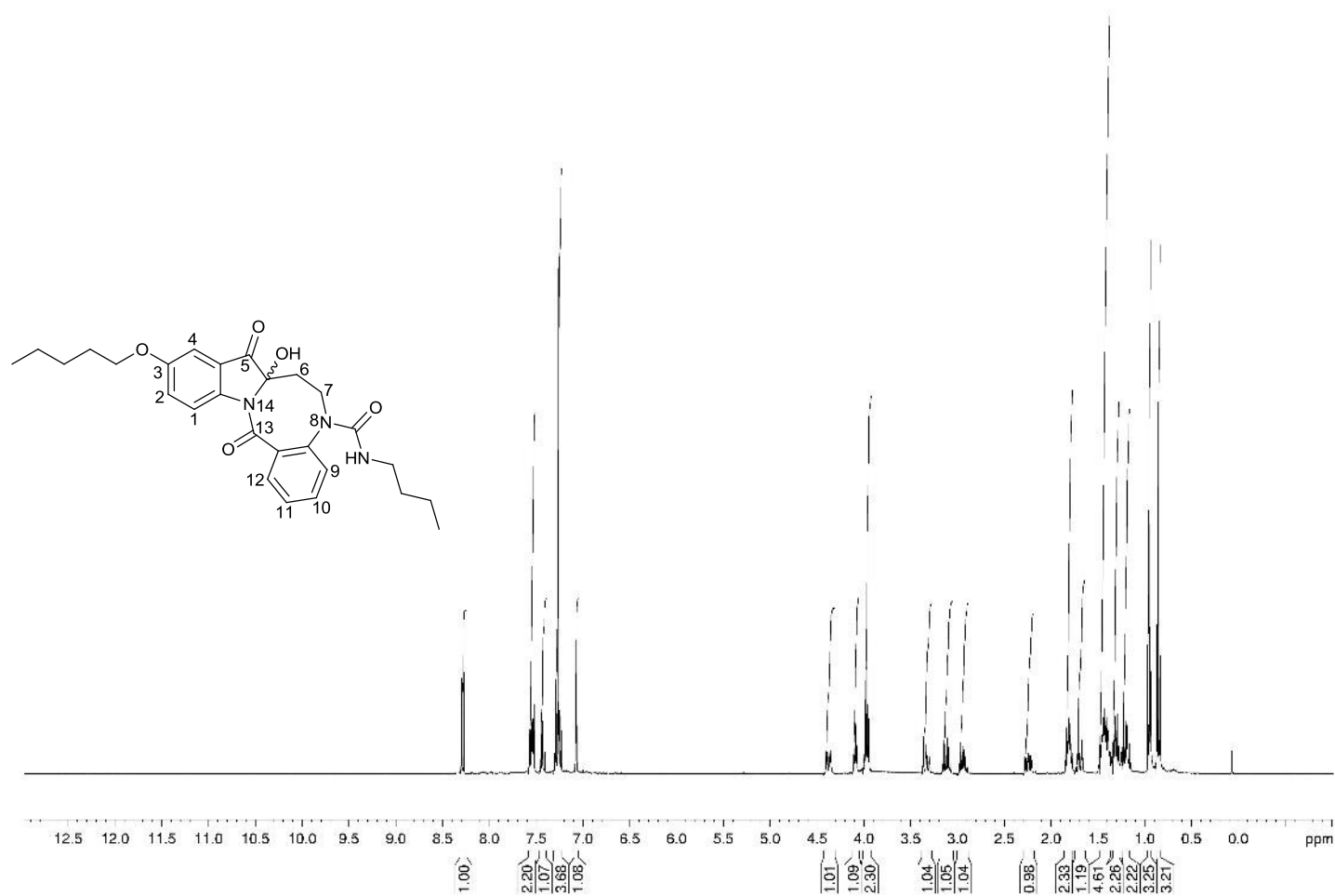
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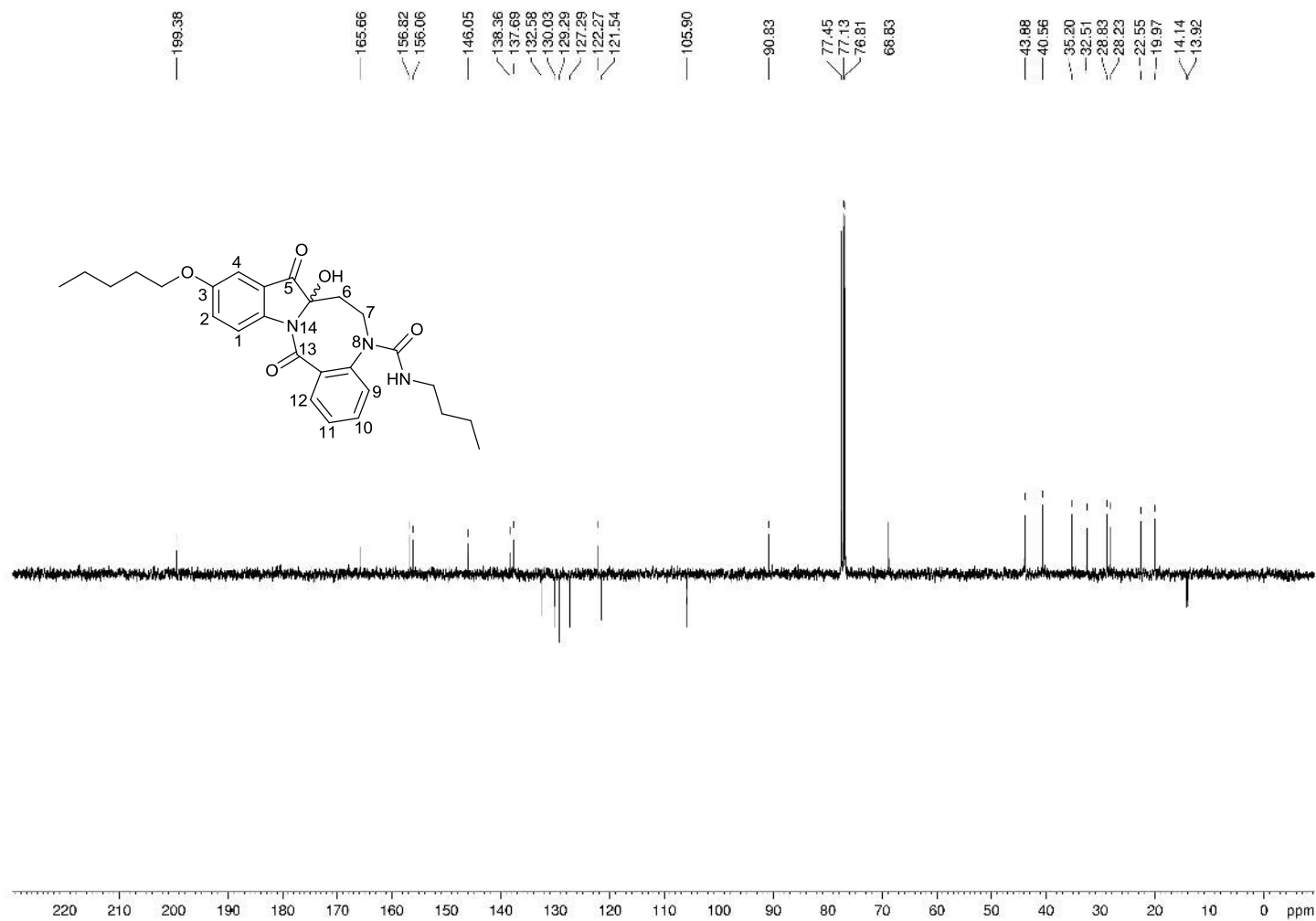
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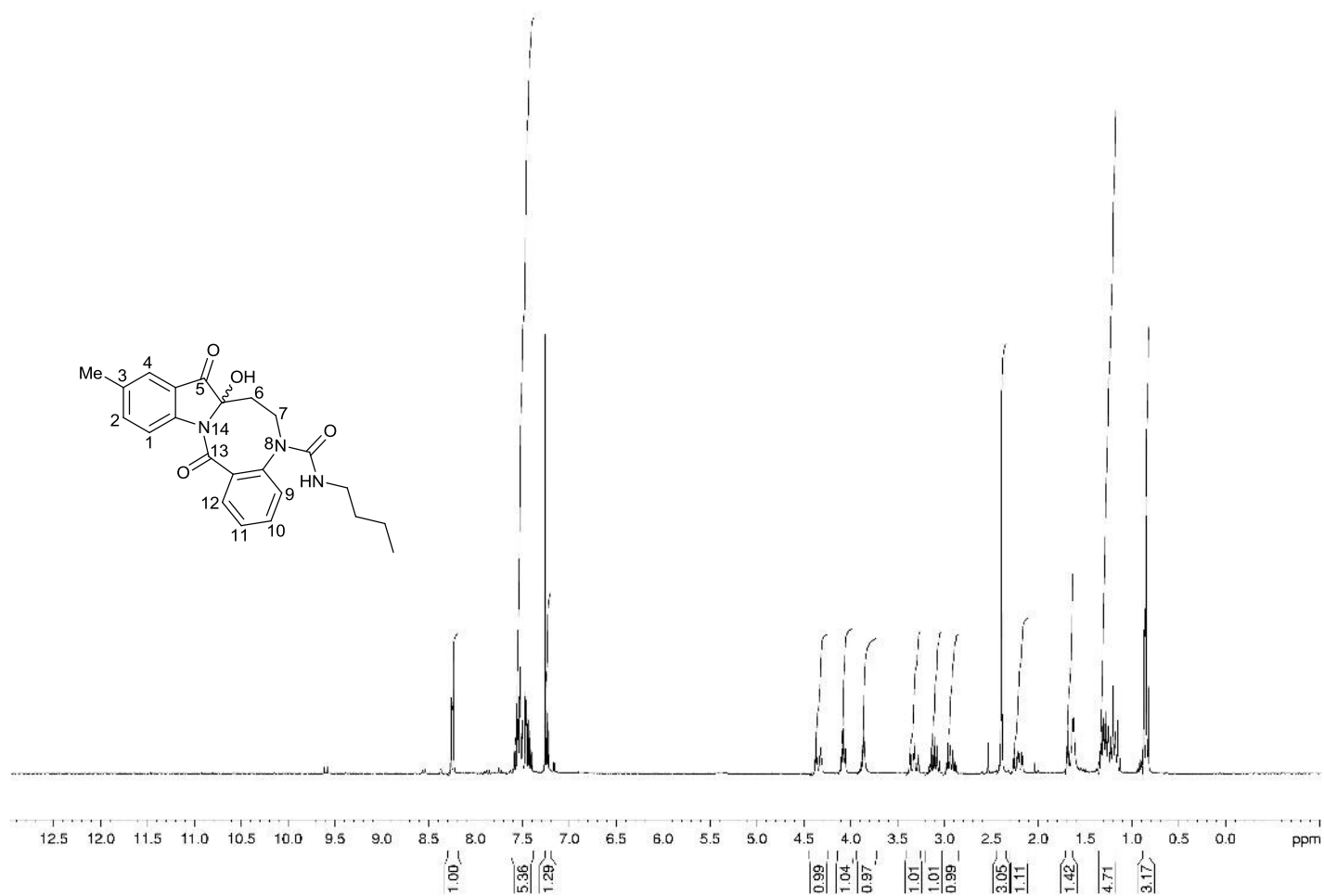
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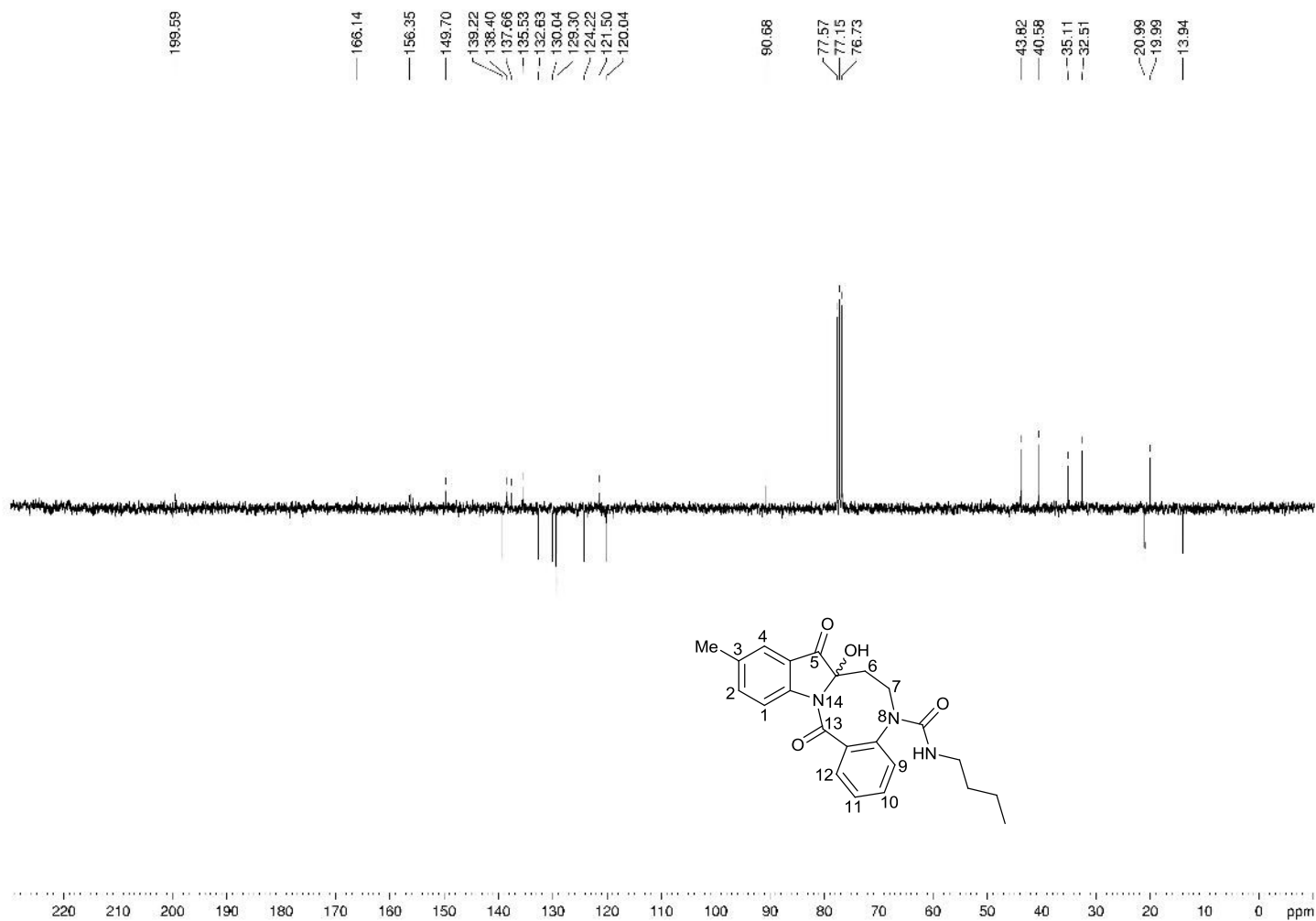
Compound 2j ^{13}C NMR (CDCl_3)



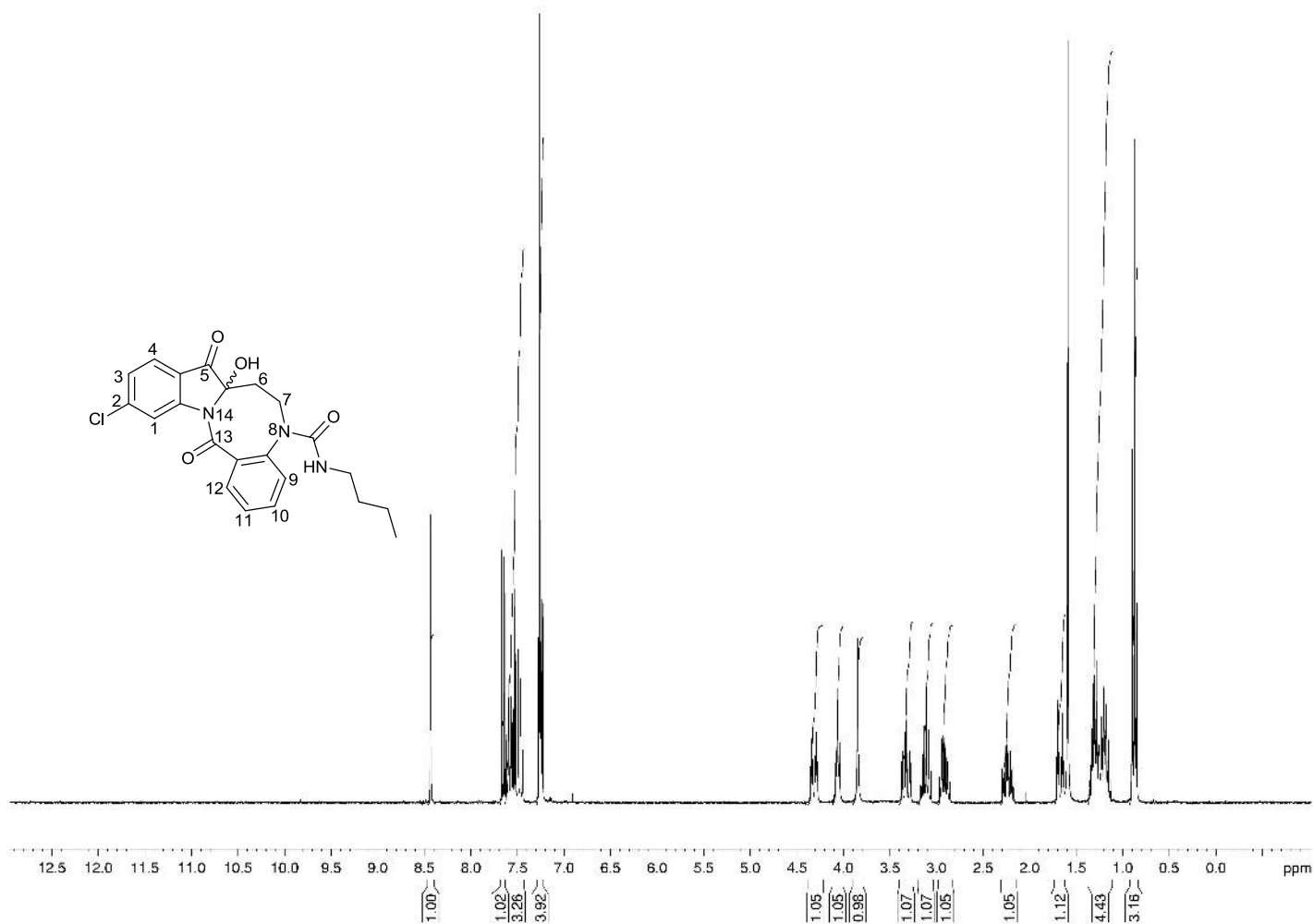
Compound 2k ^1H NMR (CDCl_3)



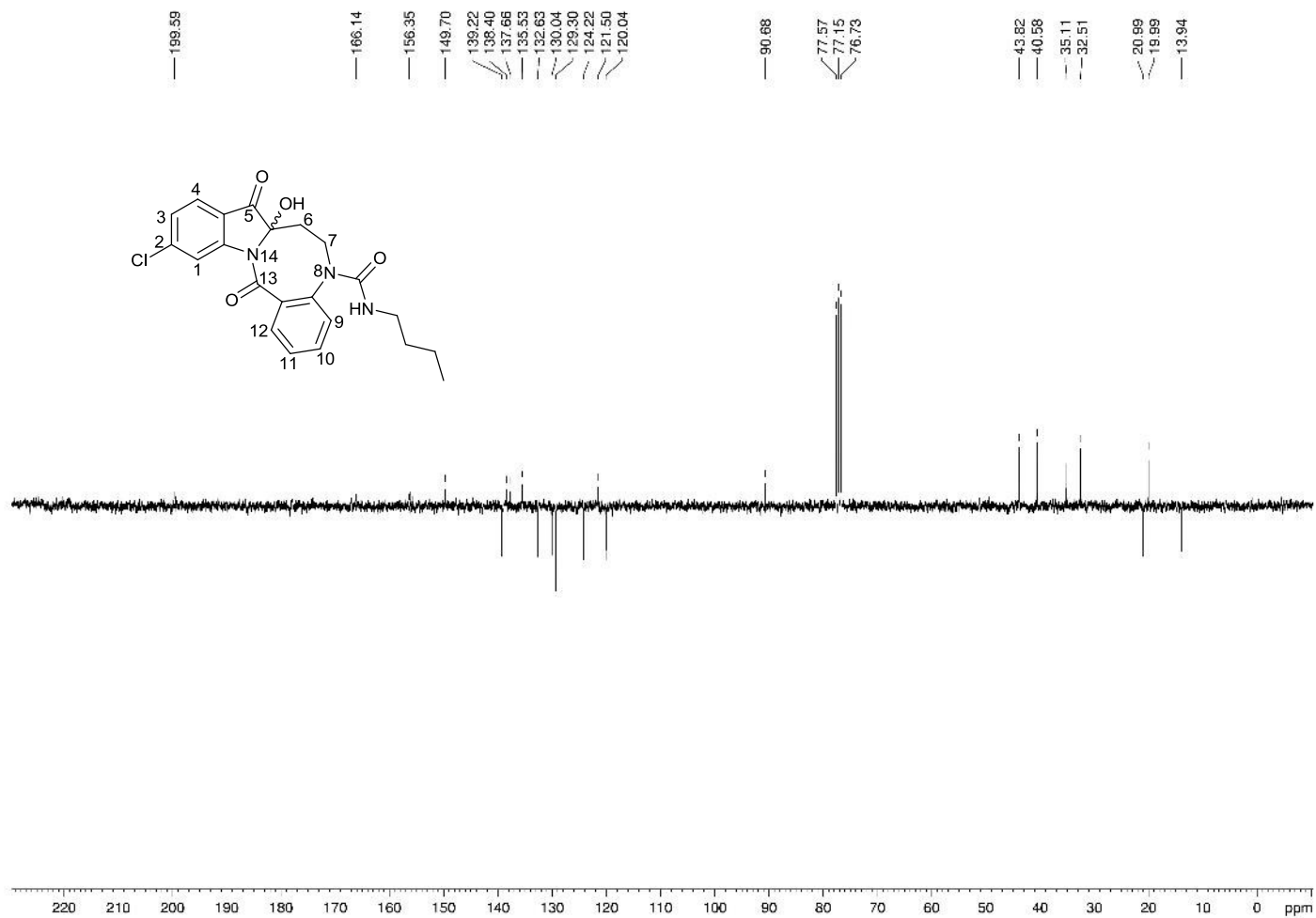
Compound 2k ^{13}C NMR (CDCl_3)



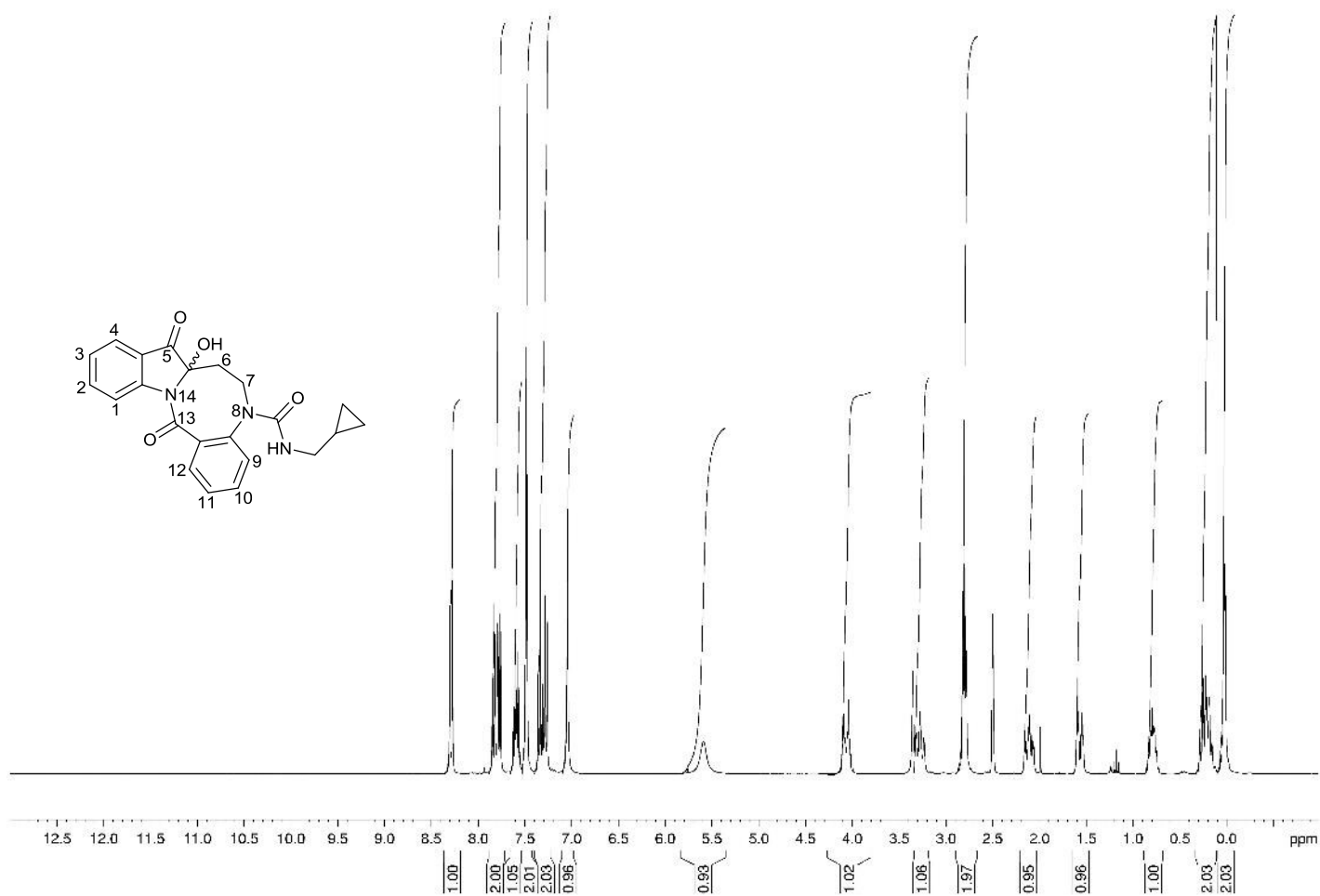
Compound 21 ^1H NMR (CDCl_3)



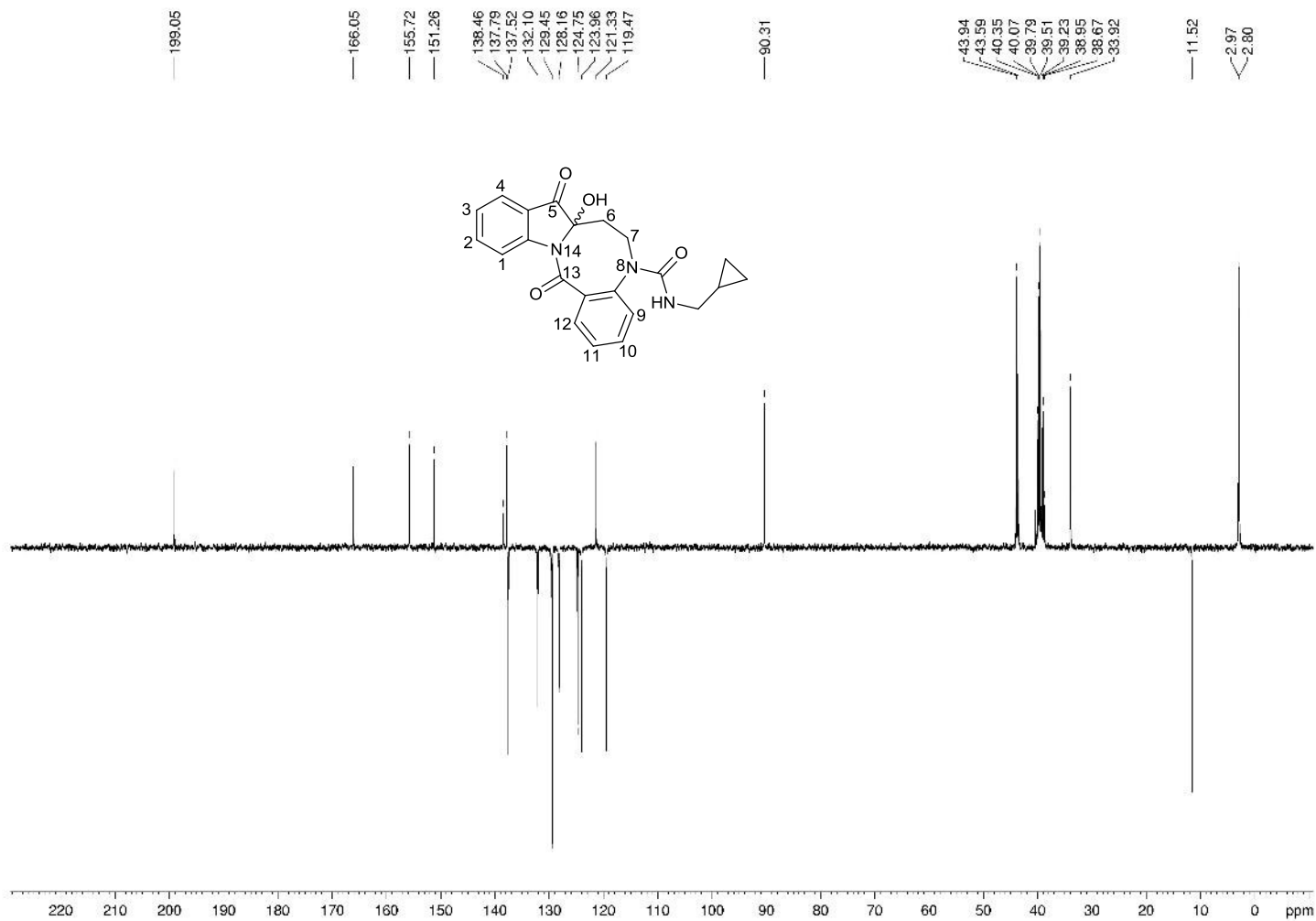
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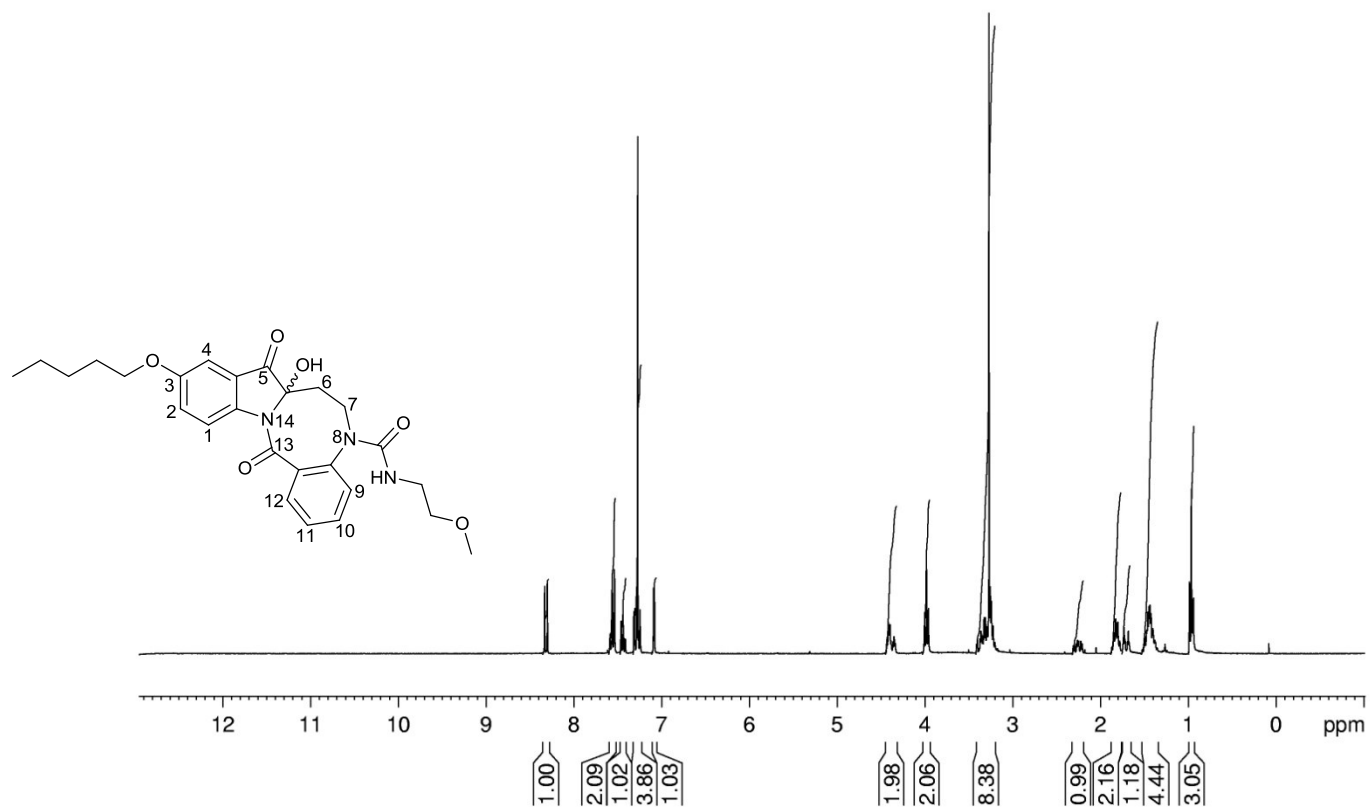
Compound 2m ^1H NMR ($\text{d}_6\text{-DMSO}$)



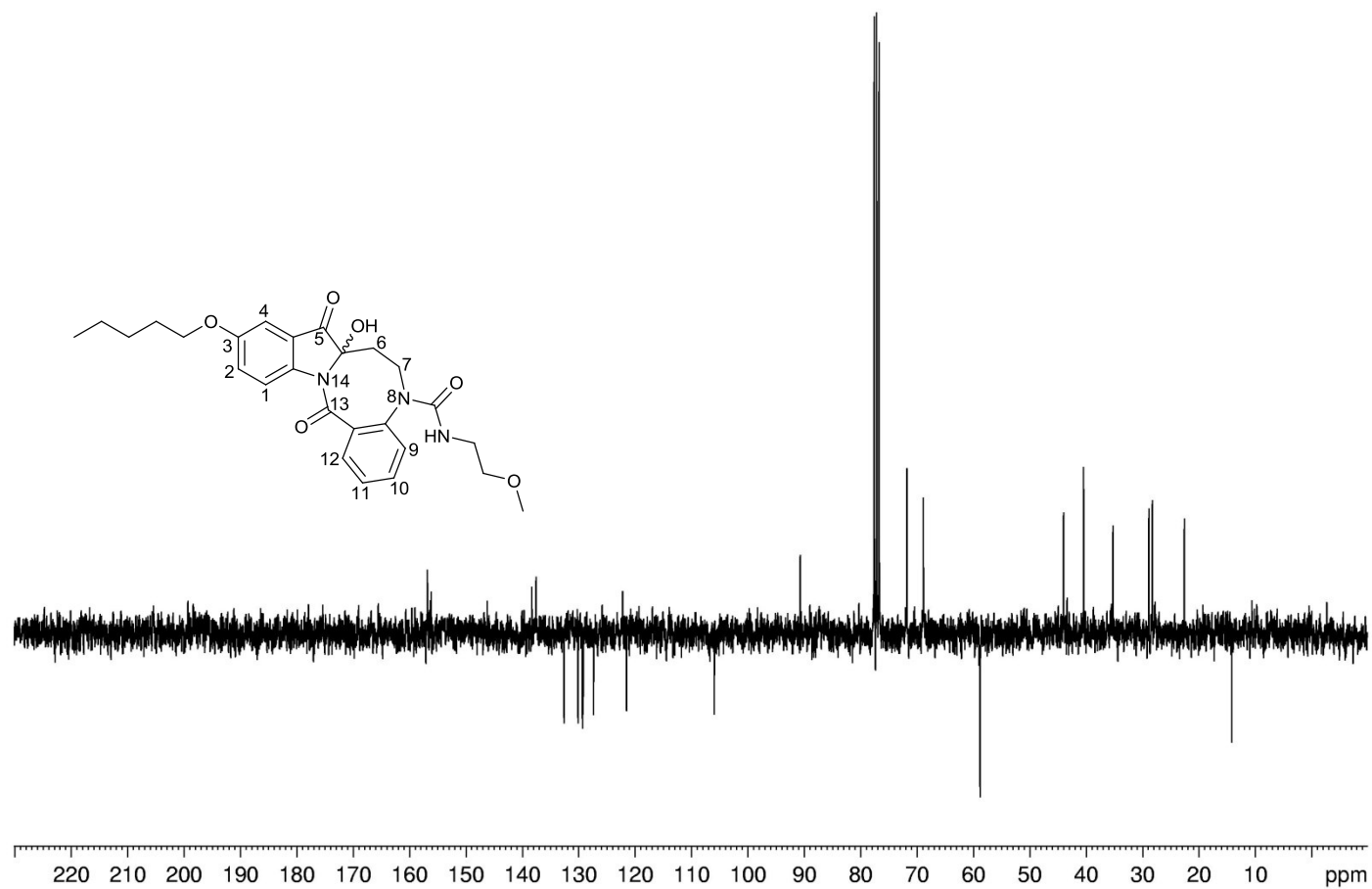
Compound 2m ^{13}C NMR ($\text{d}_6\text{-DMSO}$)



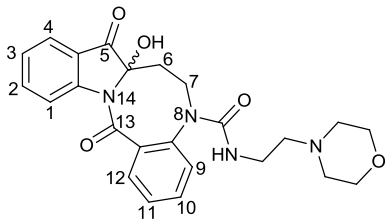
Compound 2n ^1H NMR (CDCl_3)



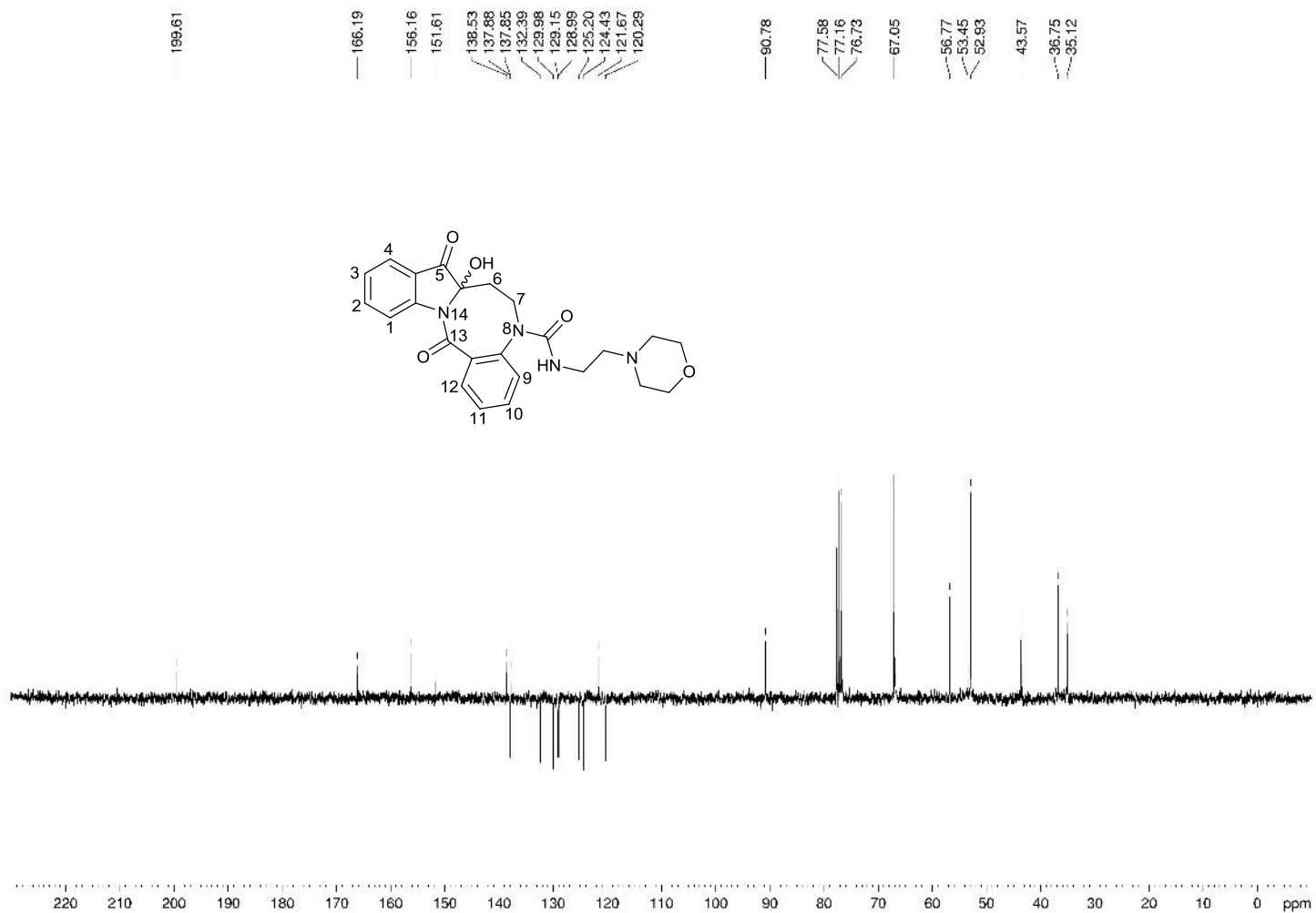
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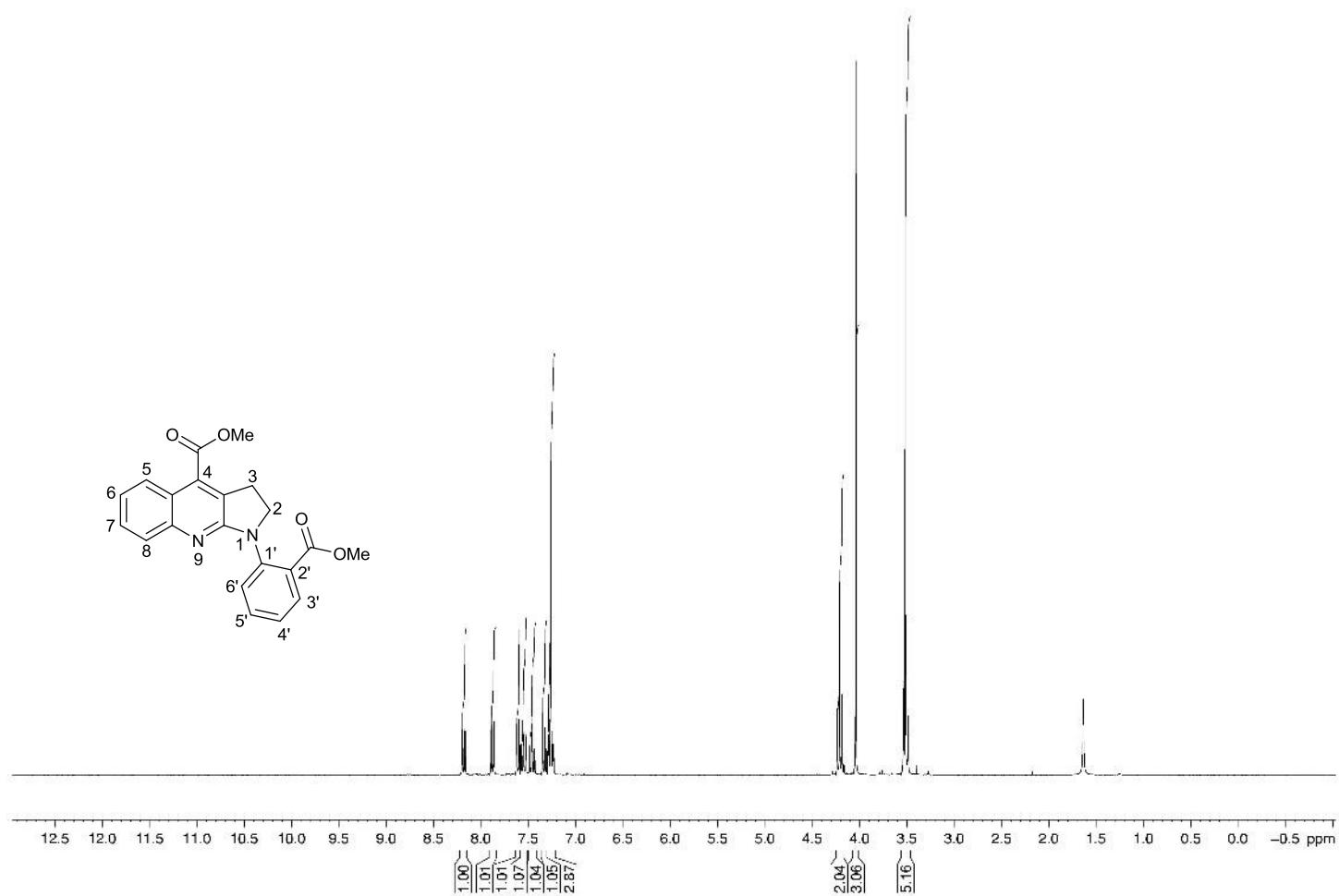
Compound 2o ¹H NMR (CDCl₃)



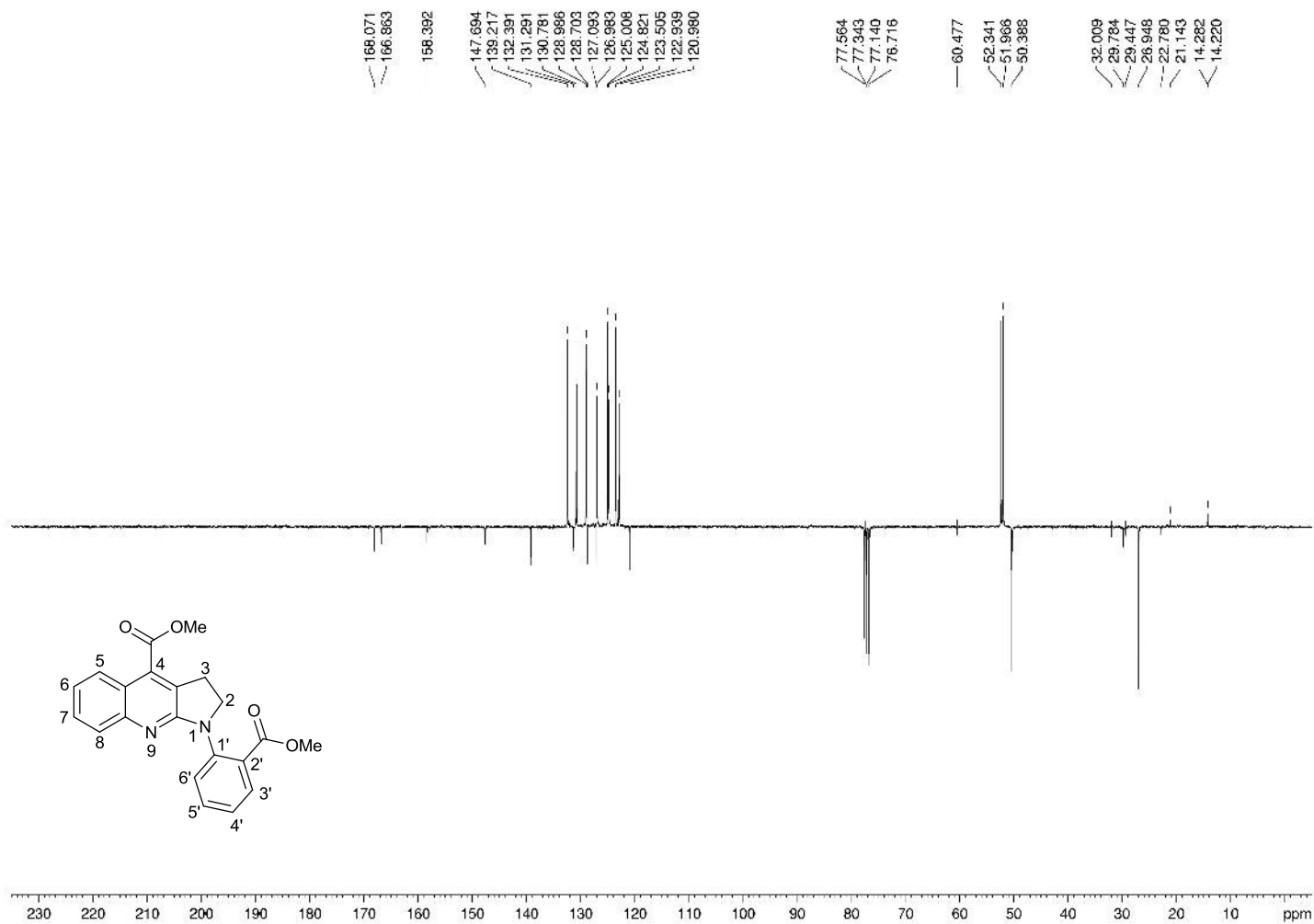
Compound 2o ¹³C NMR (CDCl₃)



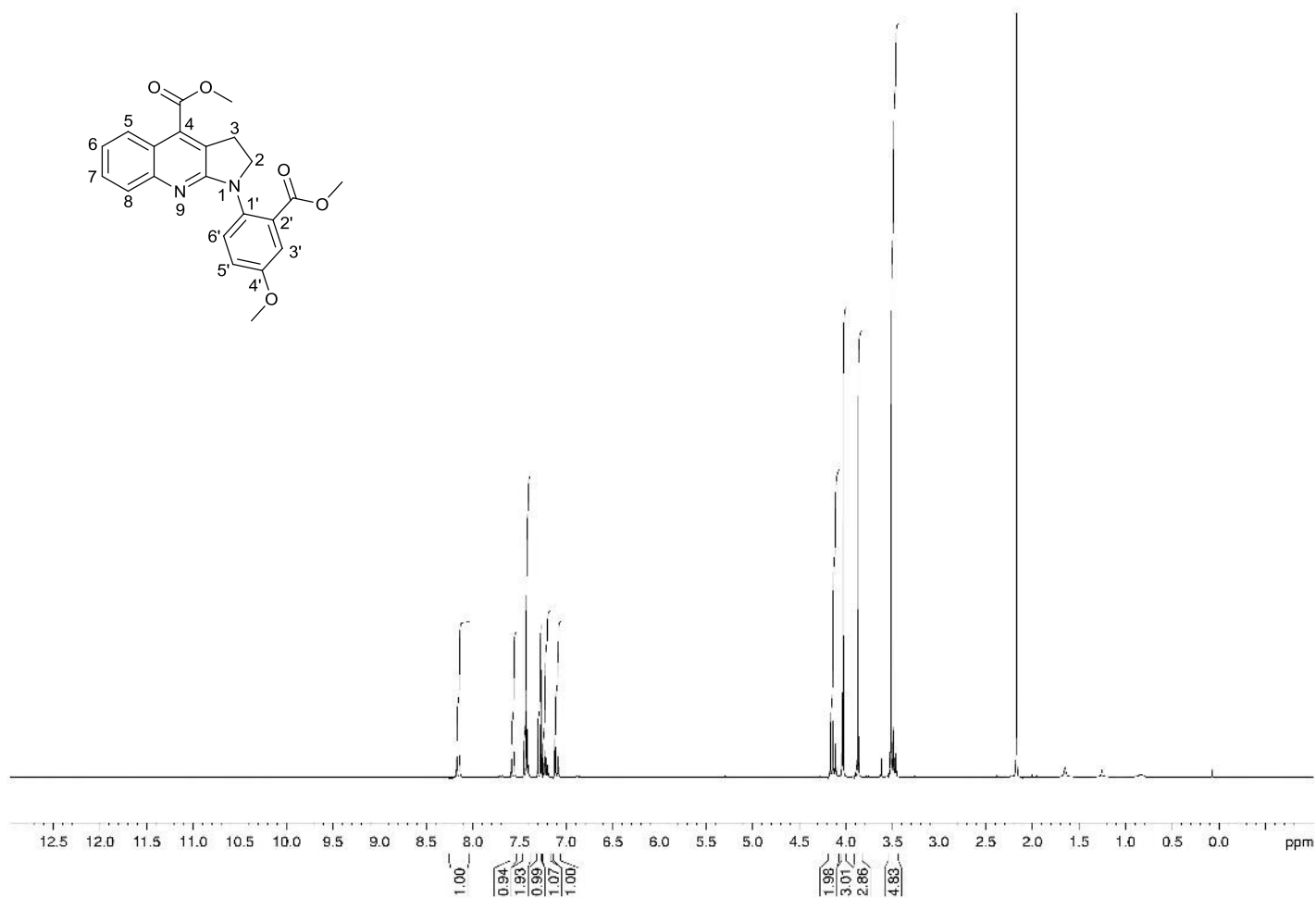
Compound 3a ^1H NMR (CDCl_3)



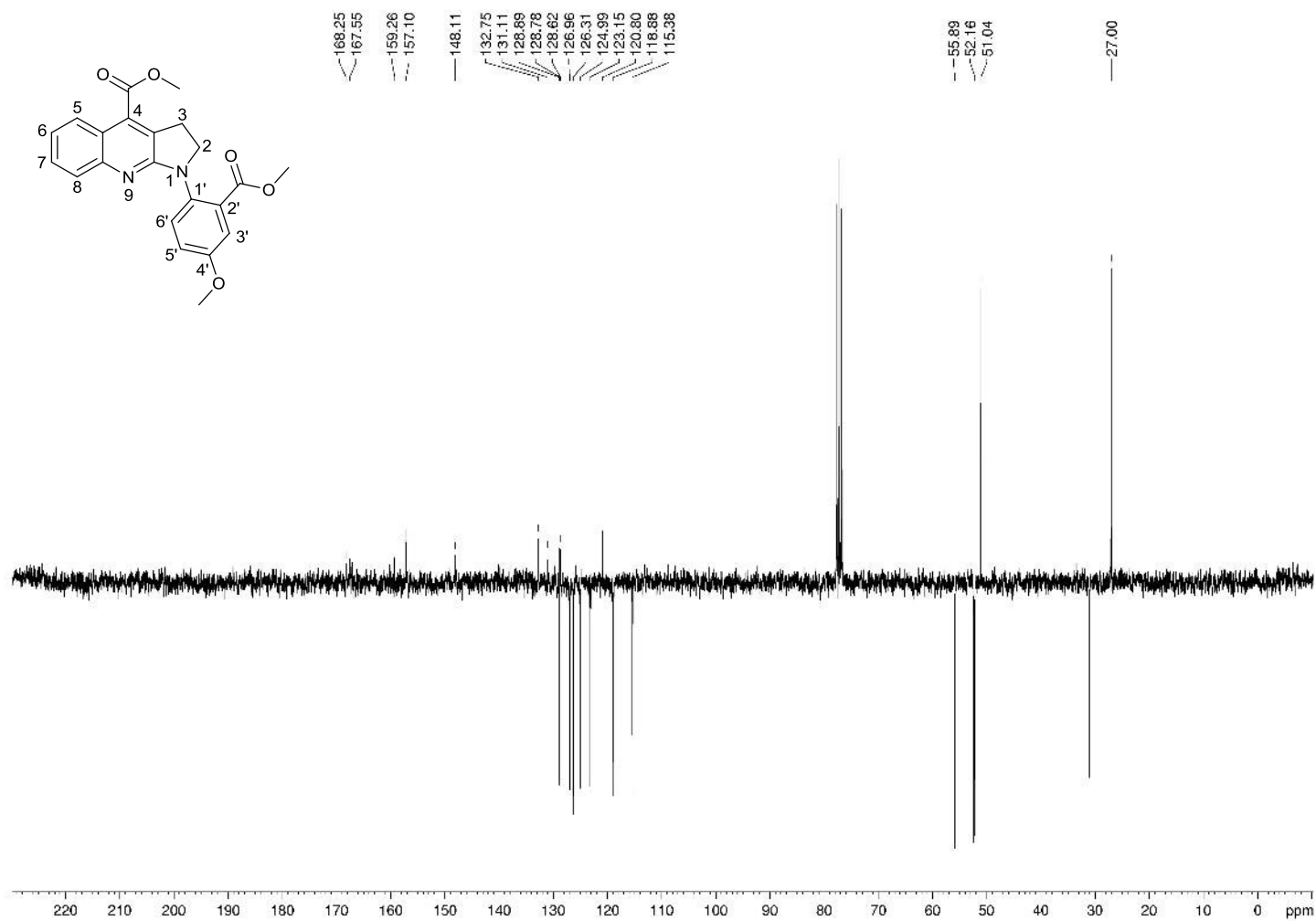
Compound 3a ^{13}C NMR (CDCl_3)



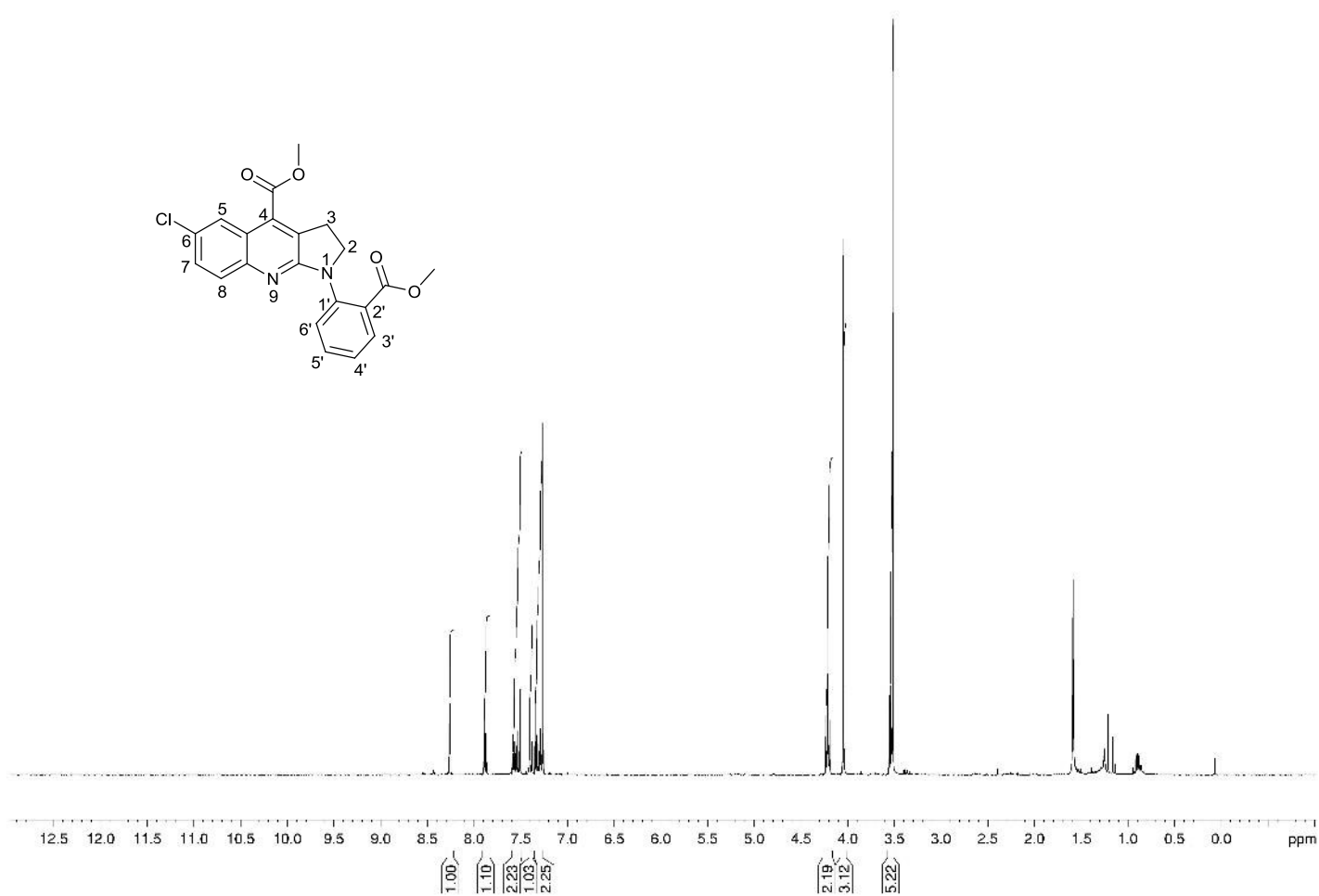
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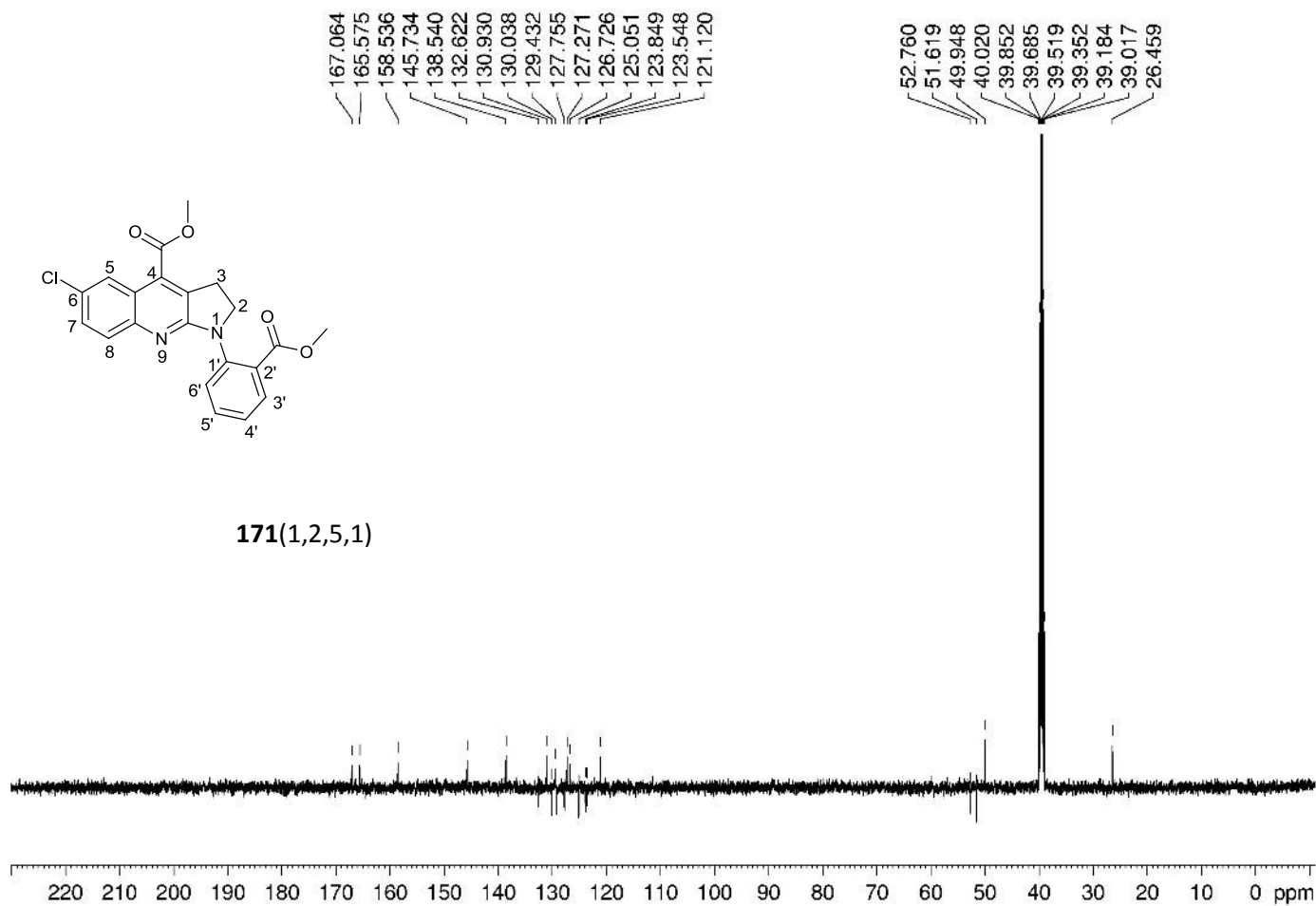
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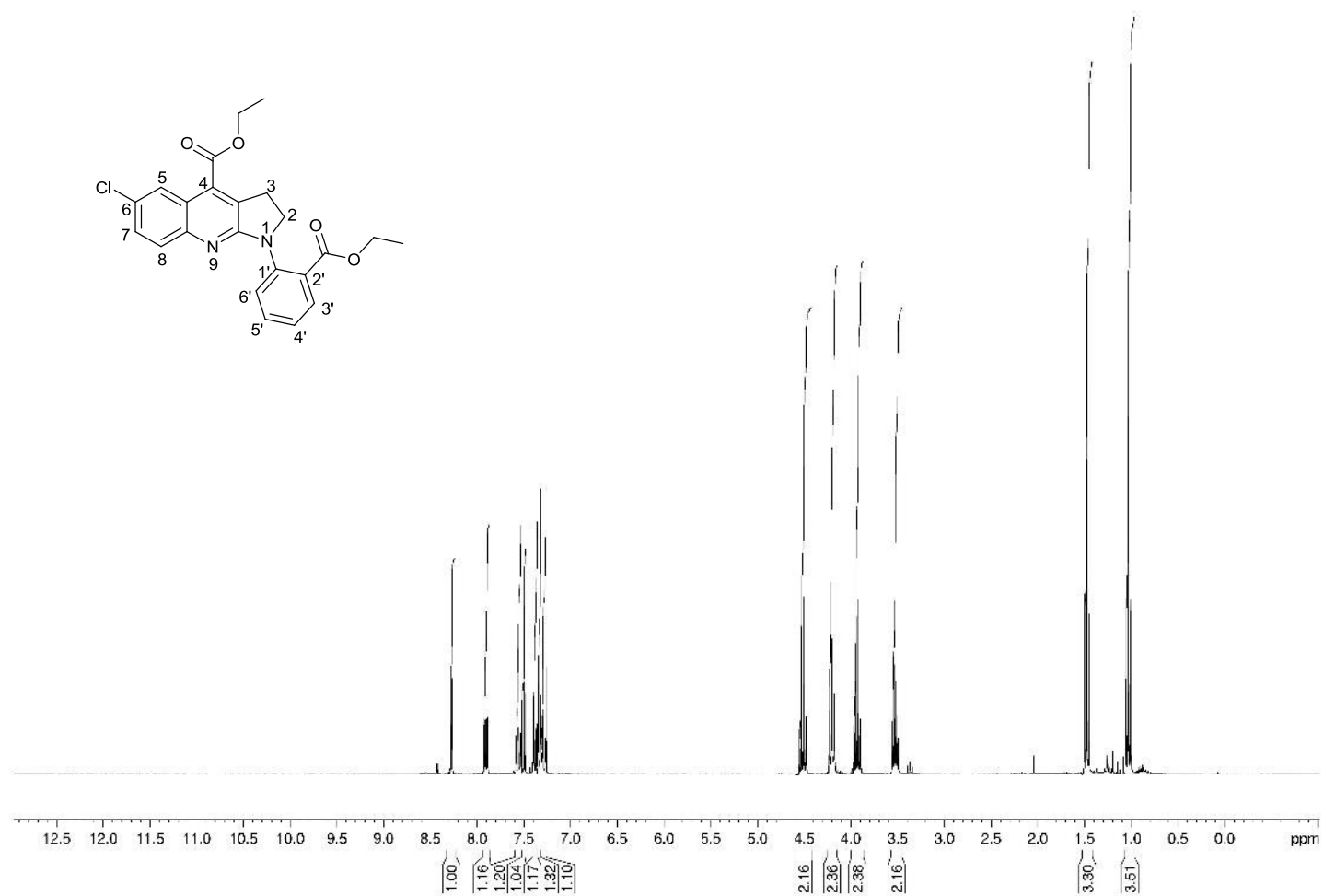
Compound 3c ^1H NMR (CDCl_3)



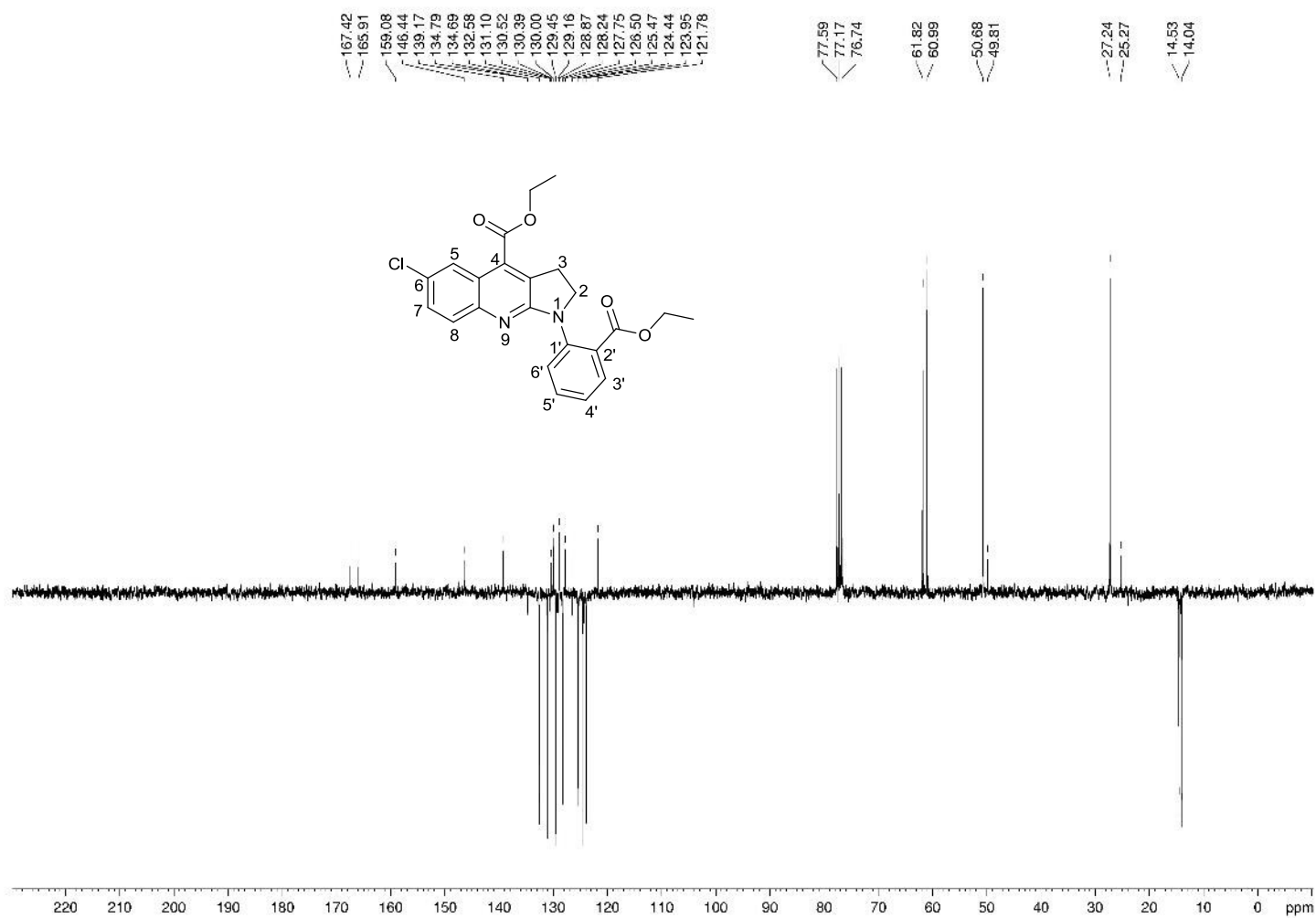
Compound 3c ^{13}C NMR ($\text{d}_6\text{-DMSO}$)



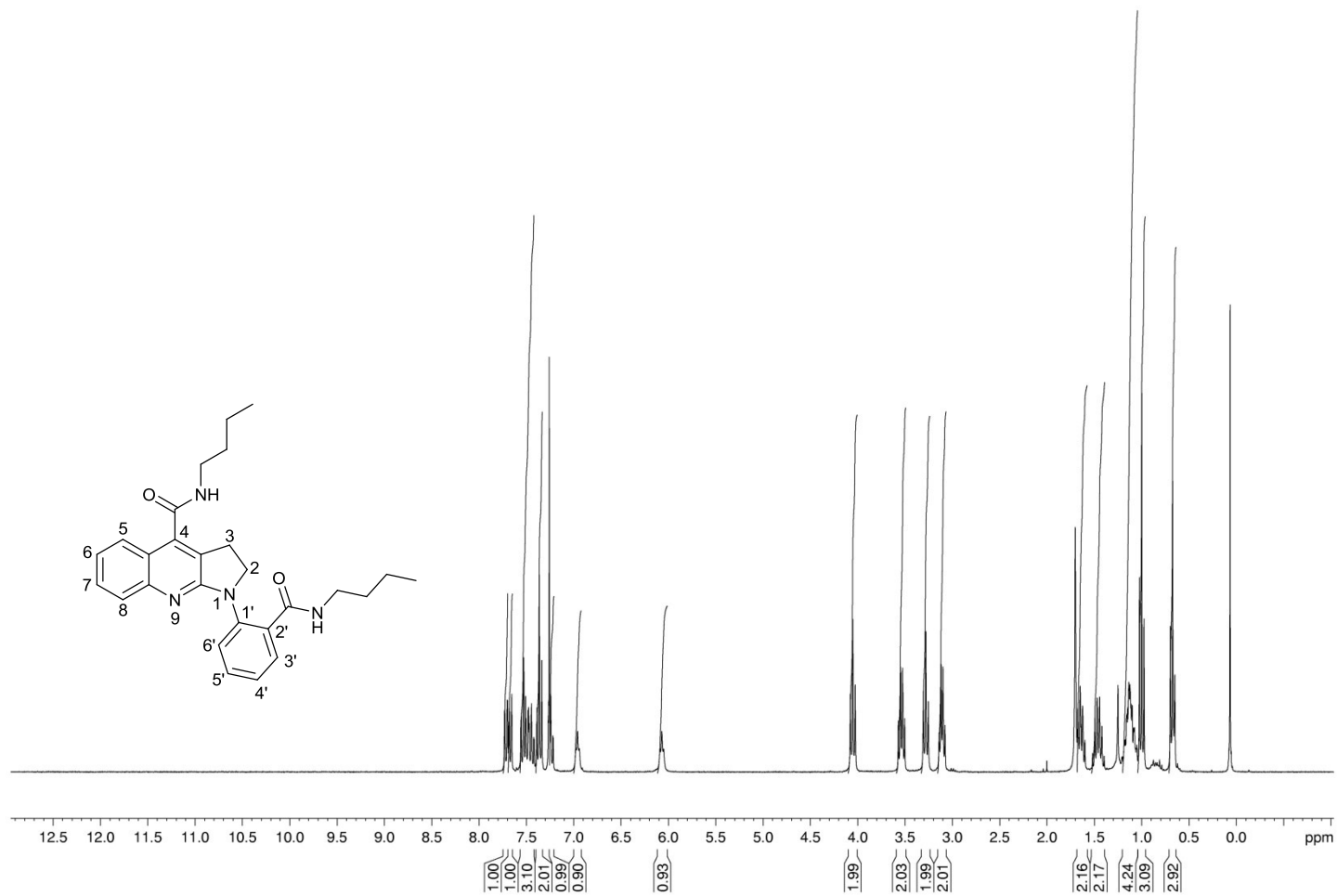
Compound 3d ^1H NMR (CDCl_3)



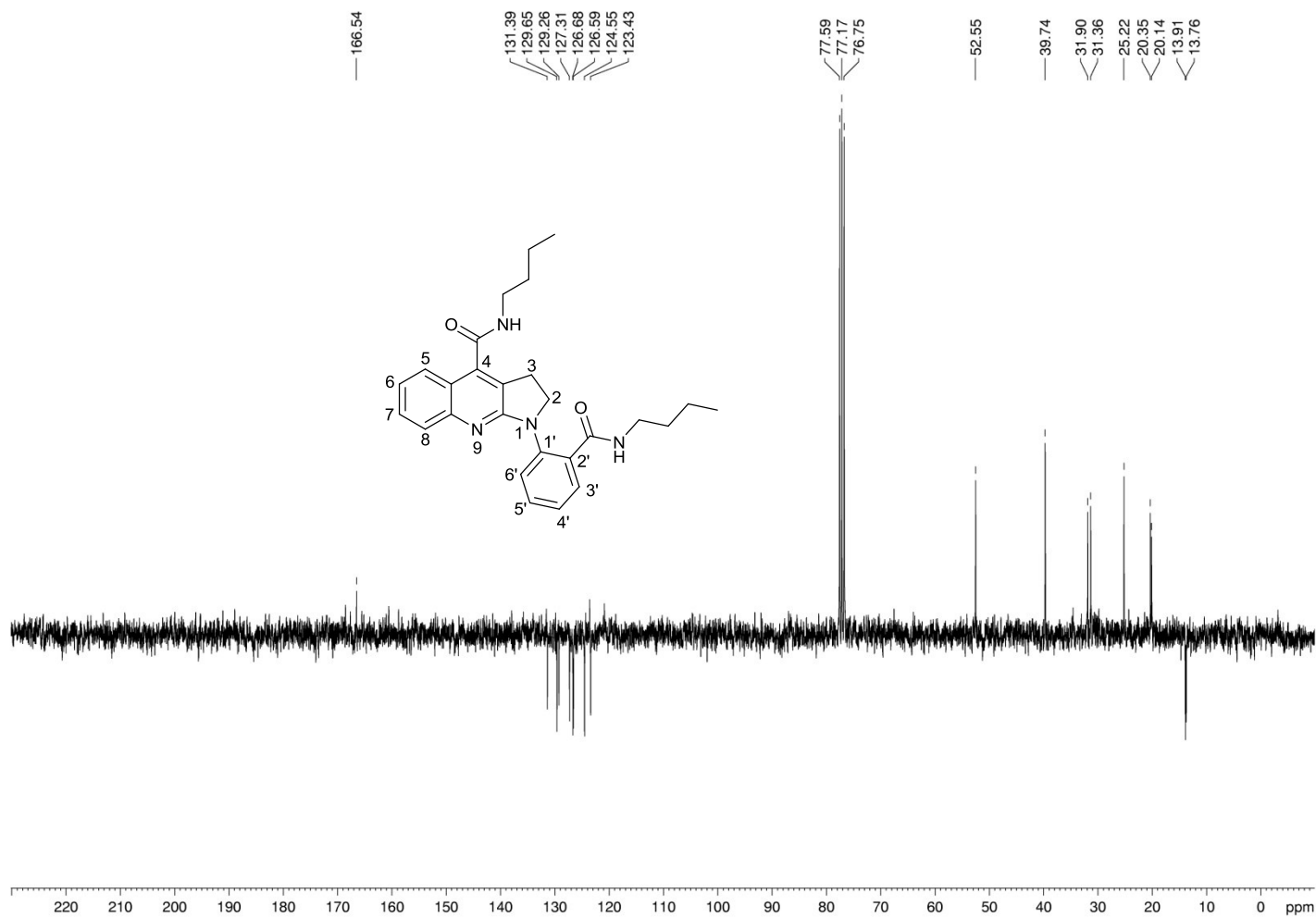
Compound 3d ^{13}C NMR (CDCl_3)



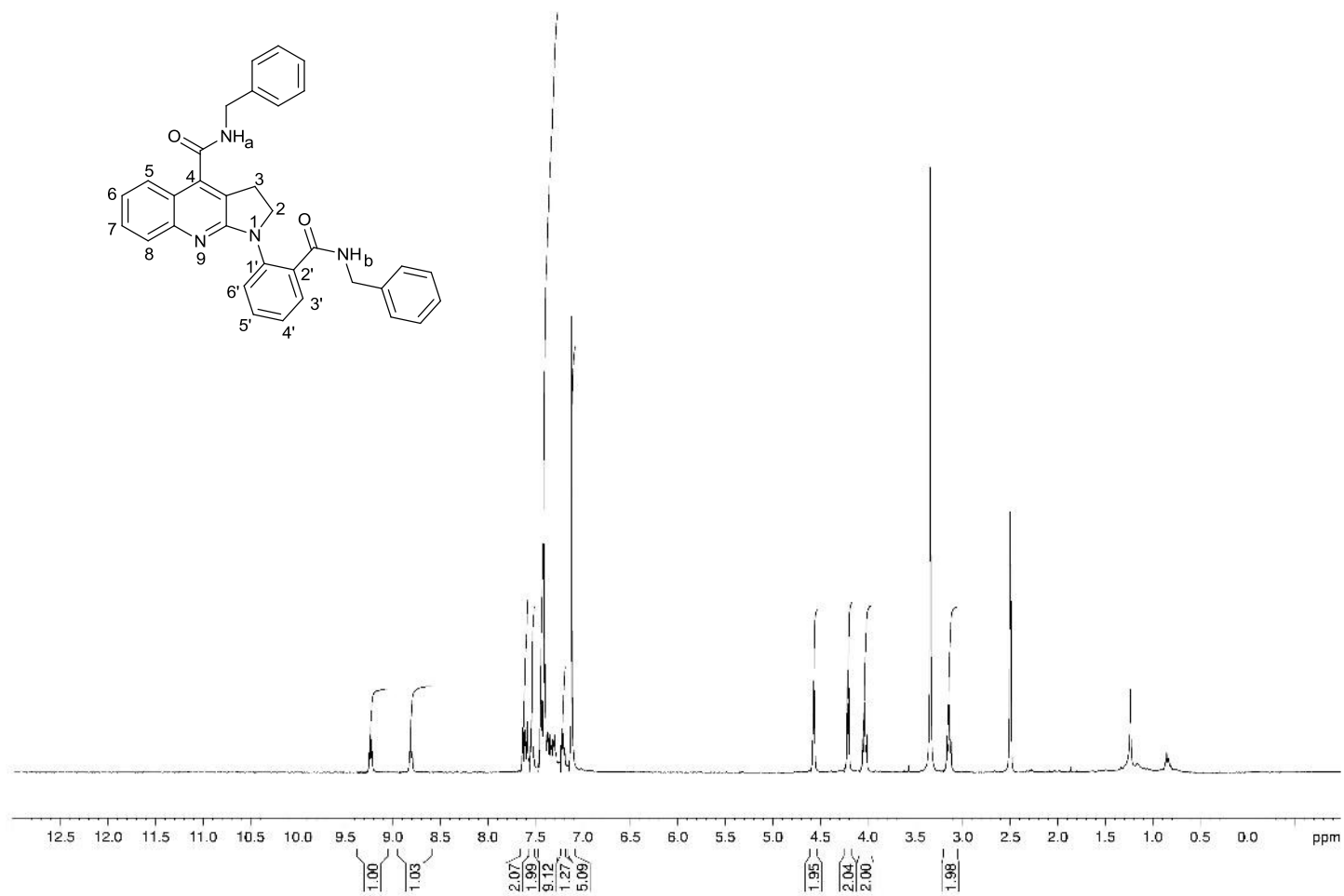
Compound 3e ^1H NMR (CDCl_3)



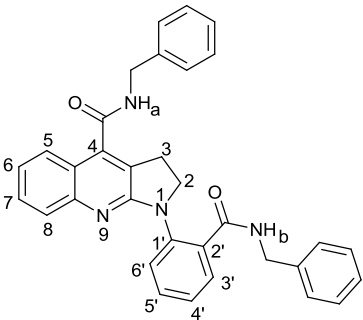
Compound 3e ^{13}C NMR (CDCl_3)



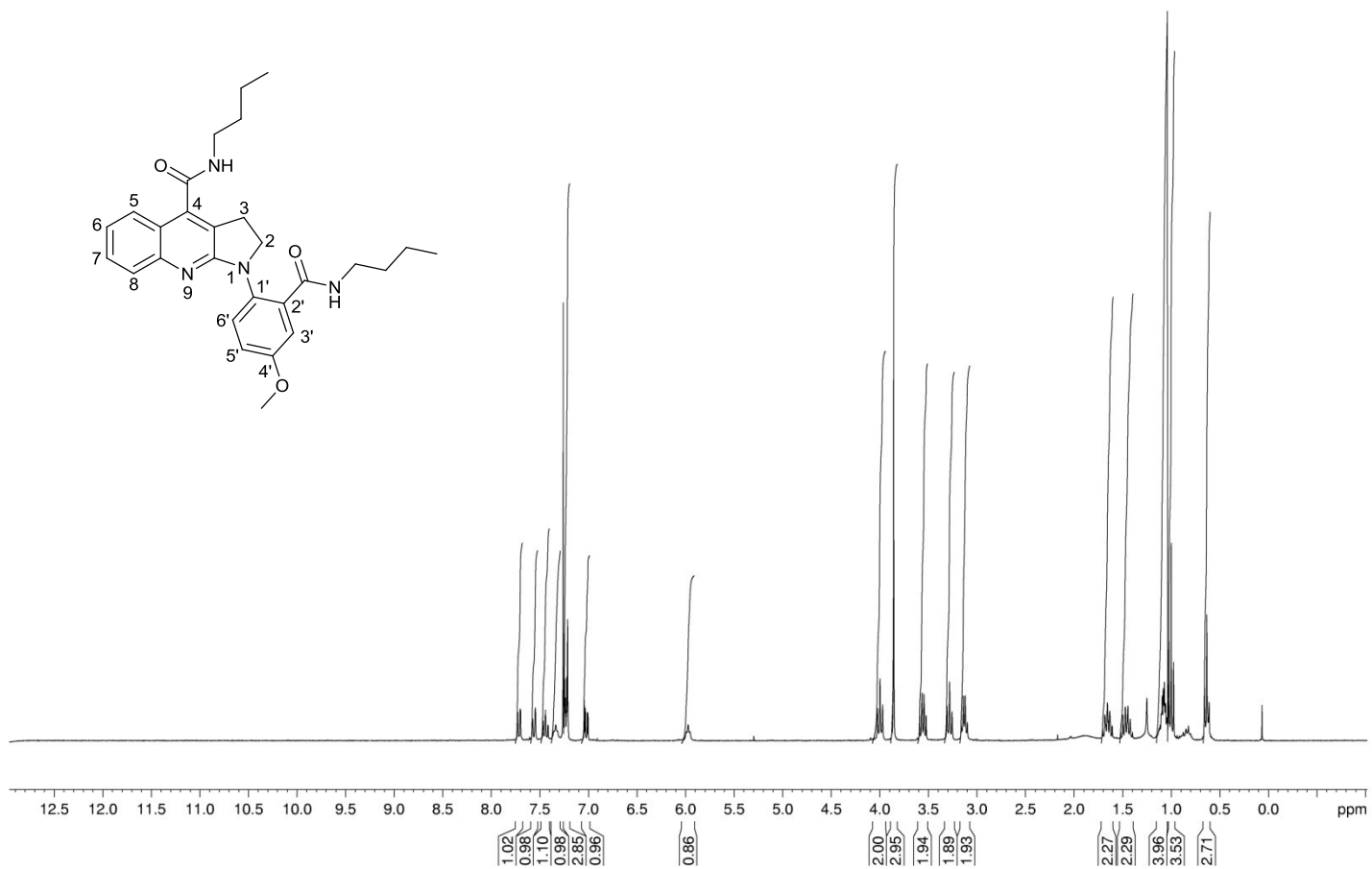
Compound 3f ^1H NMR ($\text{d}_6\text{-DMSO}$)



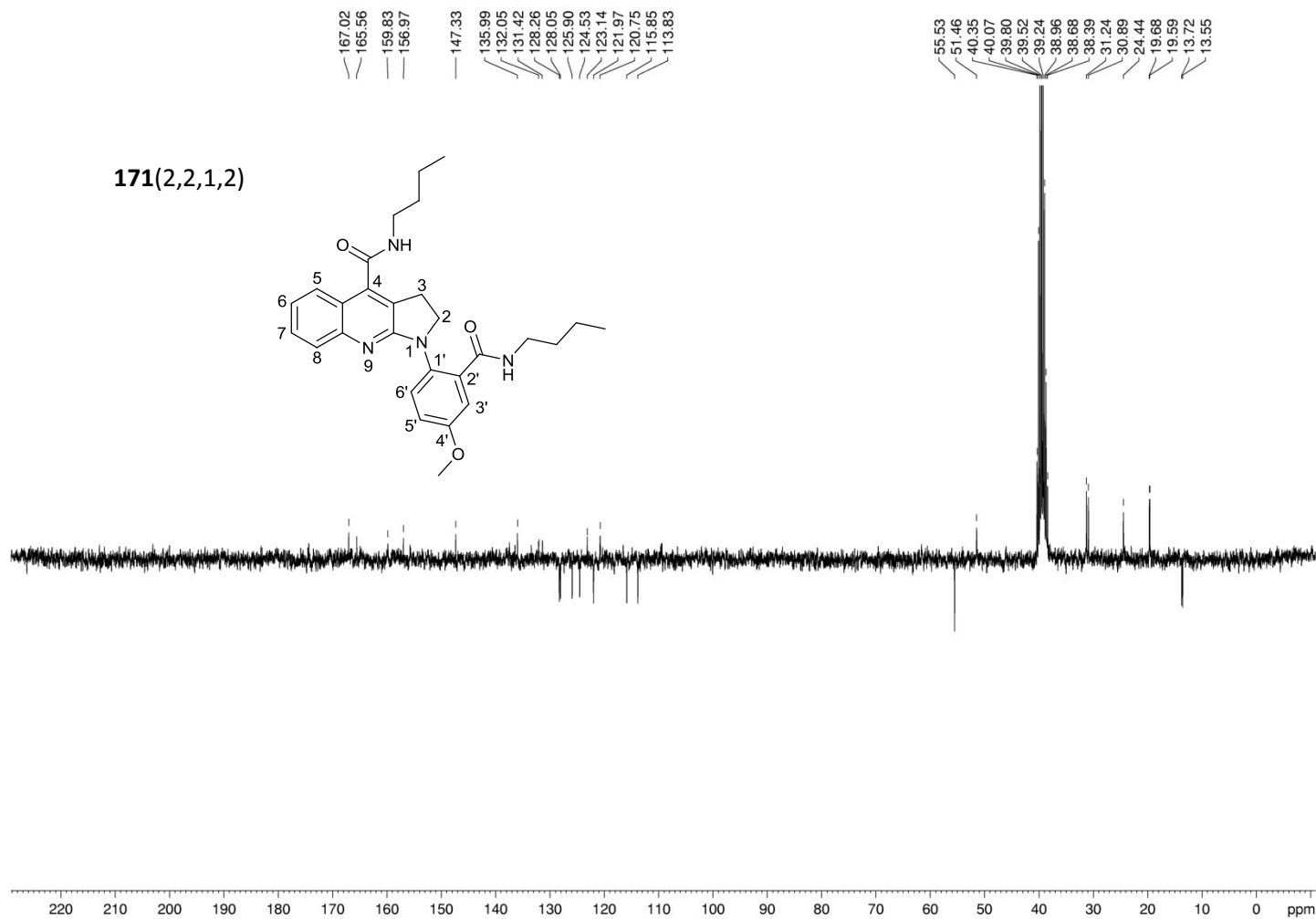
Compound 3f ¹³C NMR (d₆-DMSO)



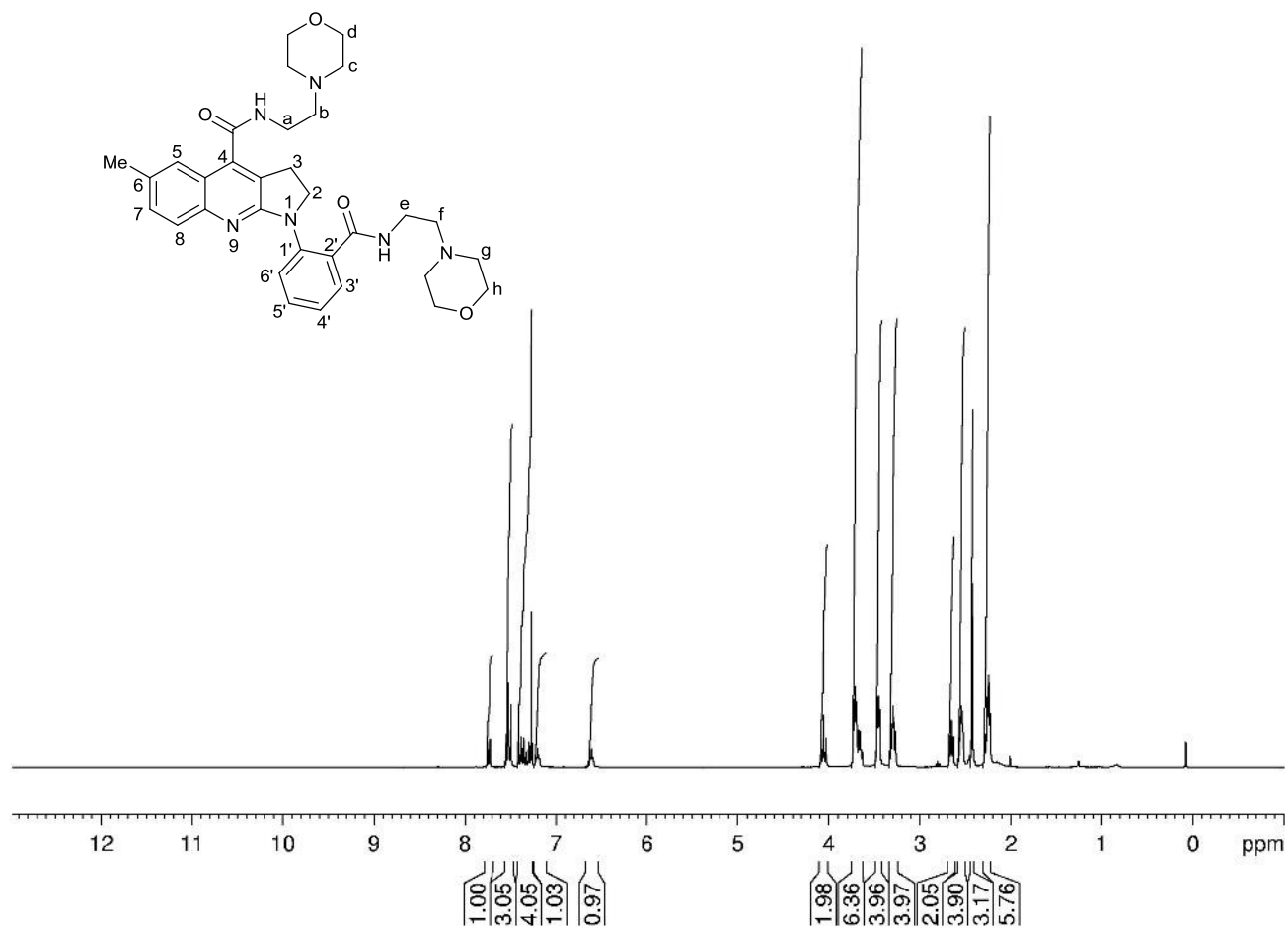
Compound 3g ^1H NMR (CDCl_3)



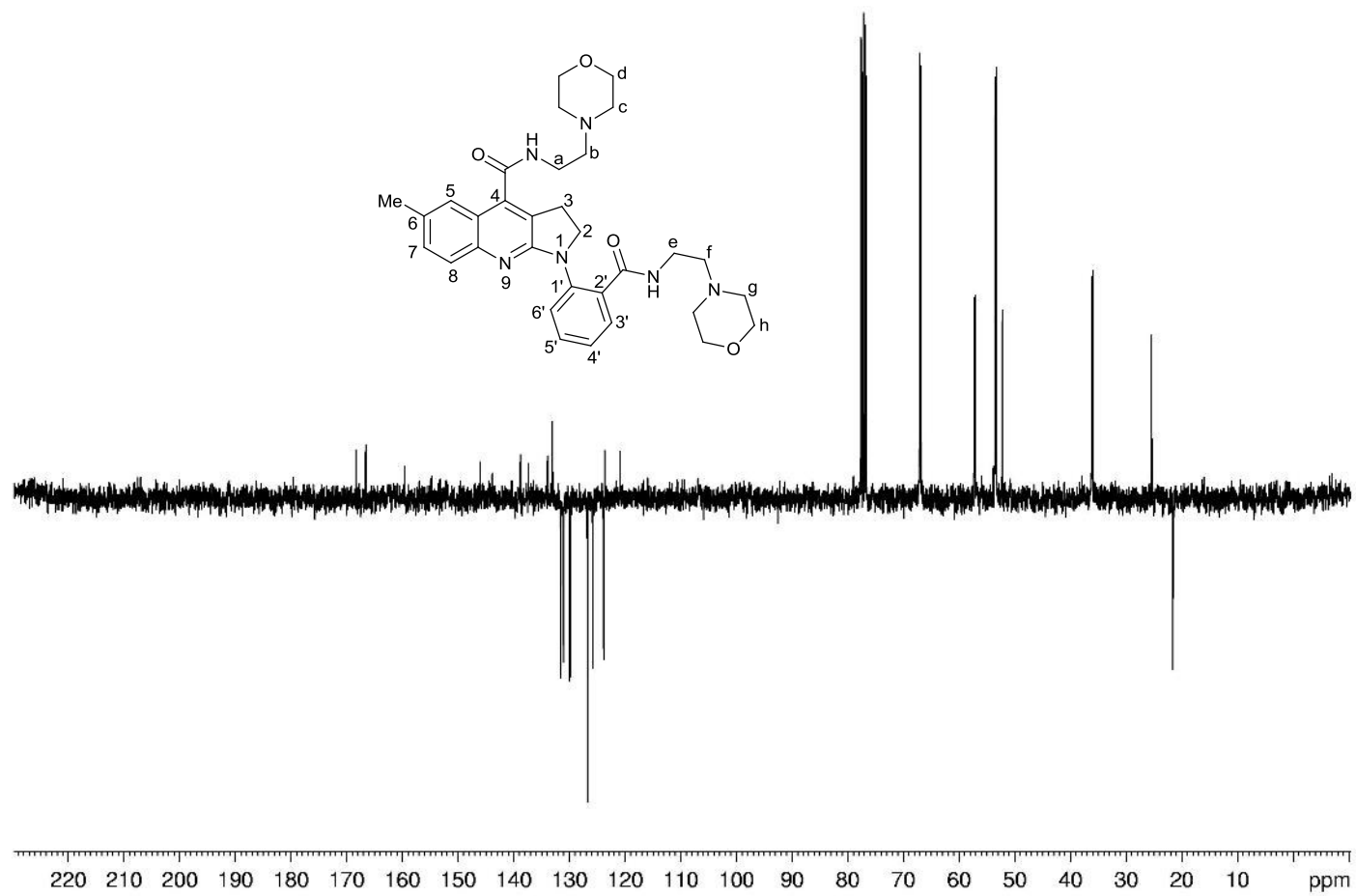
Compound 3g ^{13}C NMR ($\text{d}_6\text{-DMSO}$)



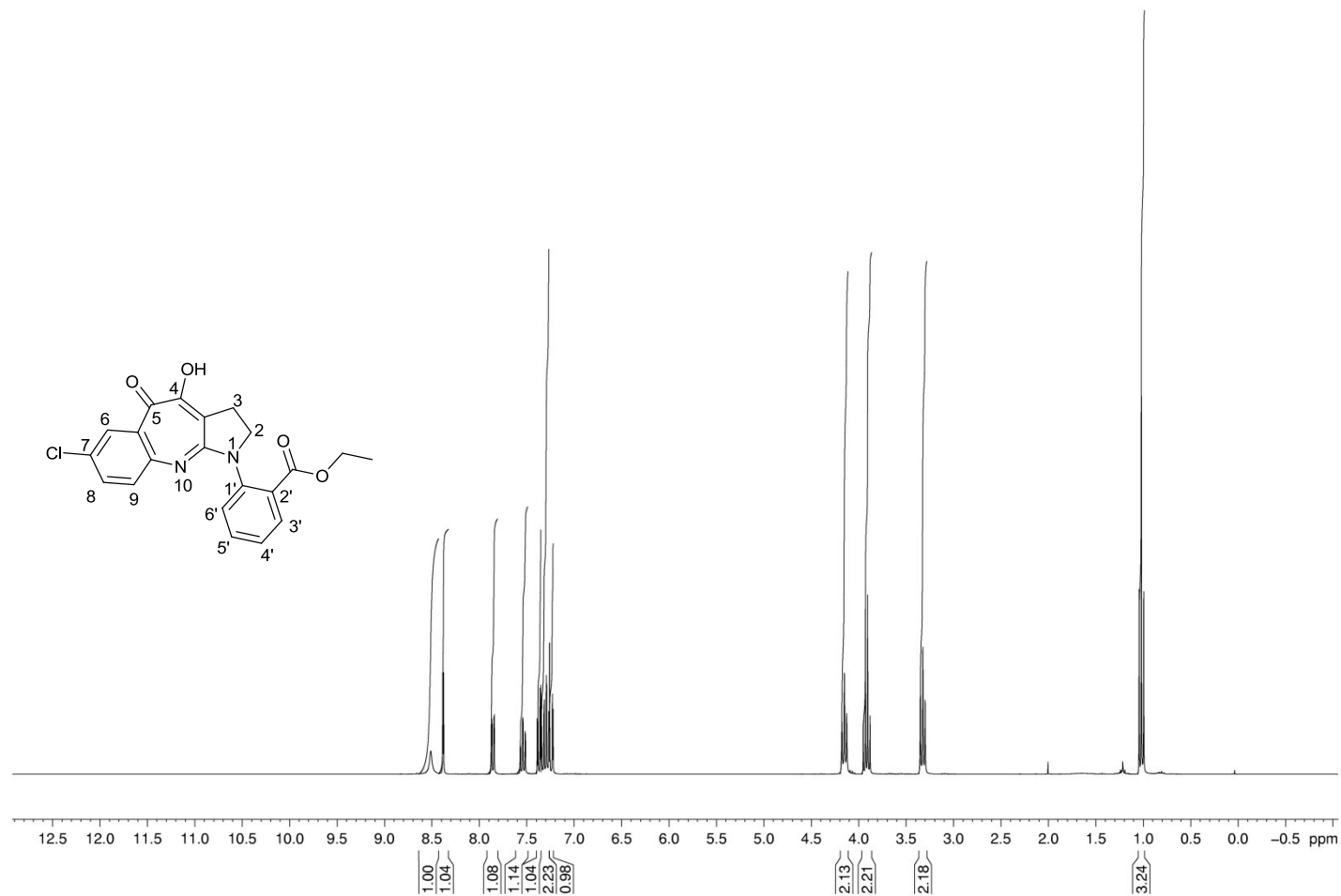
Compound 3h ^1H NMR (CDCl_3)



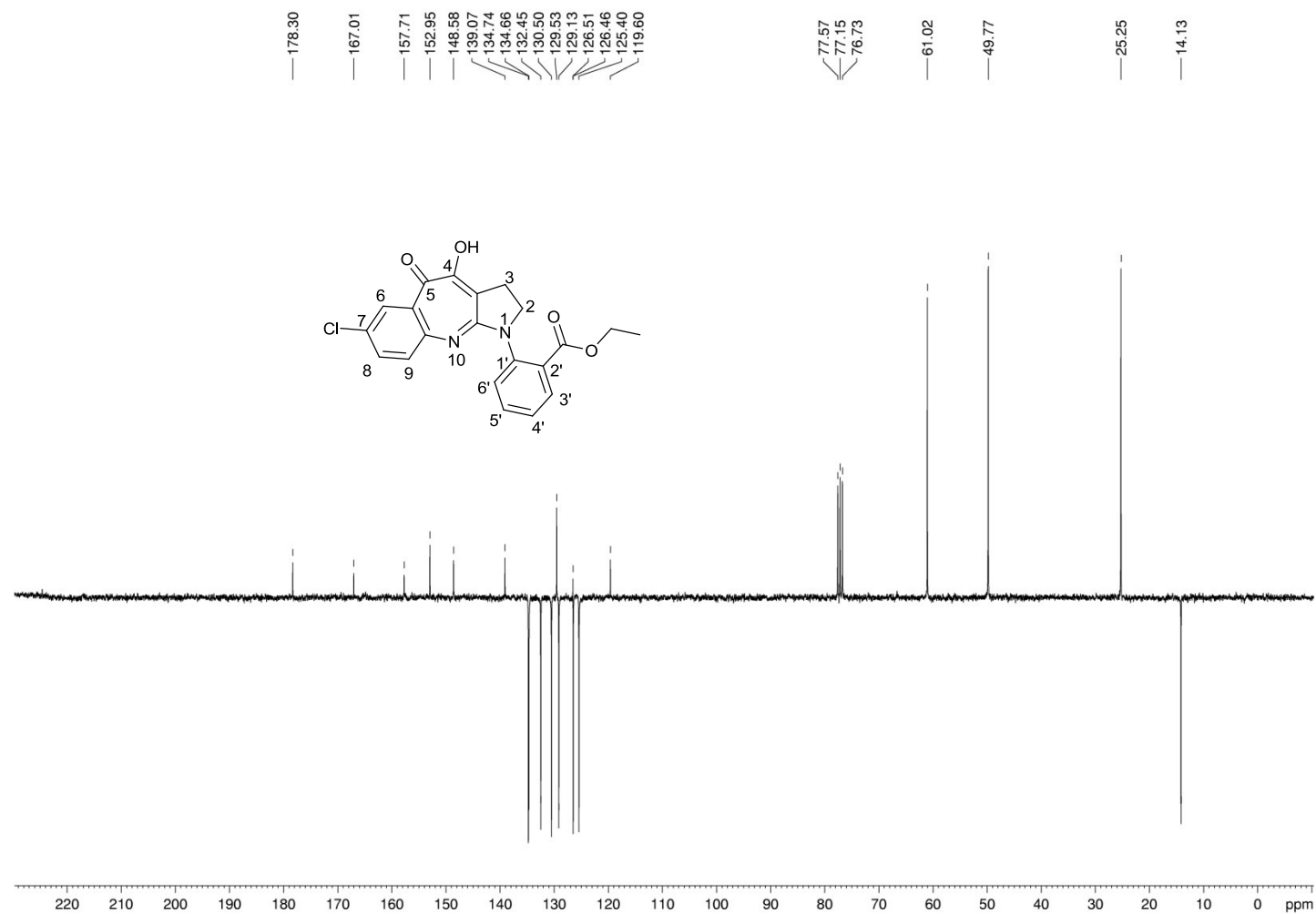
Compound 3h ^{13}C NMR (CDCl_3)



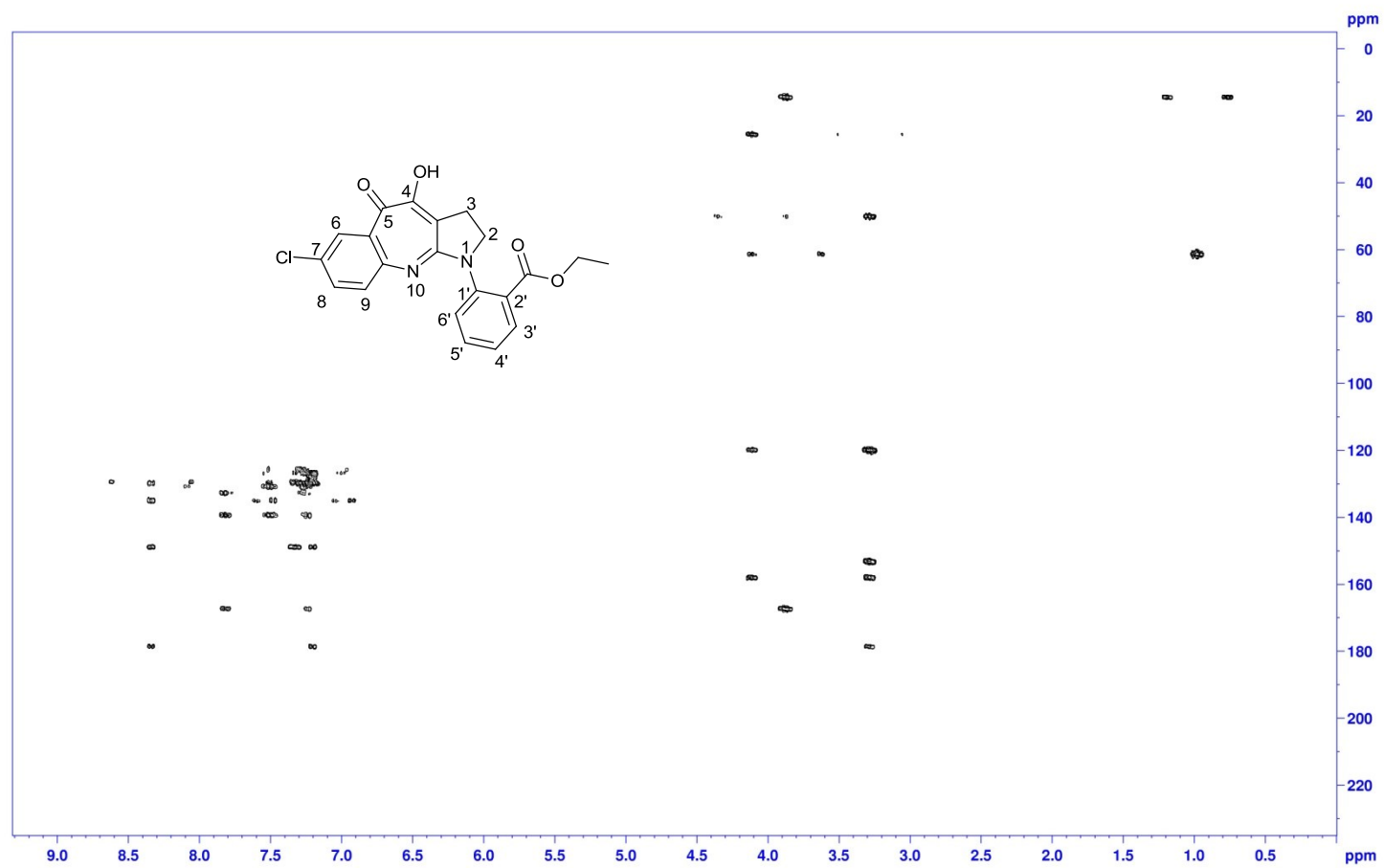
Compound 4 ^1H NMR (CDCl_3)



Compound 4 ^{13}C NMR (CDCl_3)



Compound 4 2D NMR (CDCl₃)



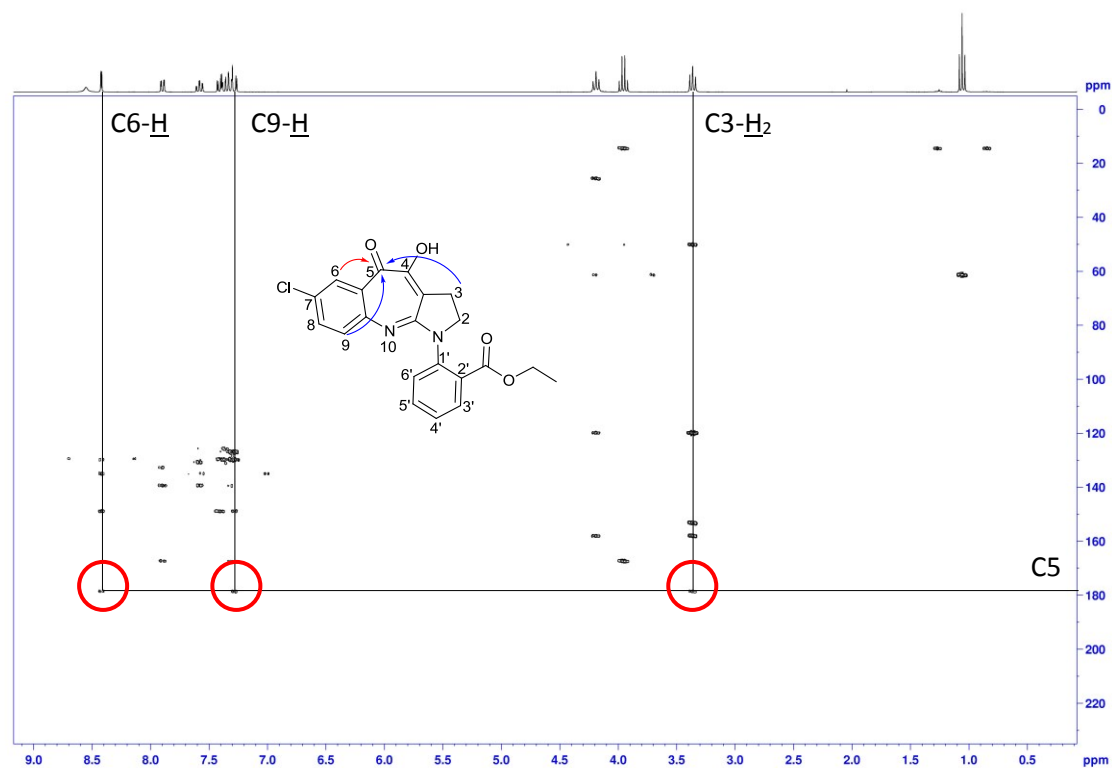


Figure S1: ^1H - ^{13}C HMBC NMR spectrum of **4** showing selected correlations that were used to determine the structure of ring B.

References

- S1** A. M. Jones, G. Liu, M. M. Lorion, S. Patterson, A. M. Z. Slawin, N. J. Westwood, *Chem. Eur. J.*, 2011, **17**, 5714.
- S2** J. L. C. Marais, W. Pickl, B. Staskun, *J. Org. Chem.*, 1990, **55**, 1969.