

Electronic Supplementary Information (ESI)

**Dioxotungsten(VI) complexes with isoniazid-related hydrazones
as (pre)catalysts for olefin epoxidation: solvent and ligand
substituent effects**

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X-Ray Crystallography. Single crystal diffraction.

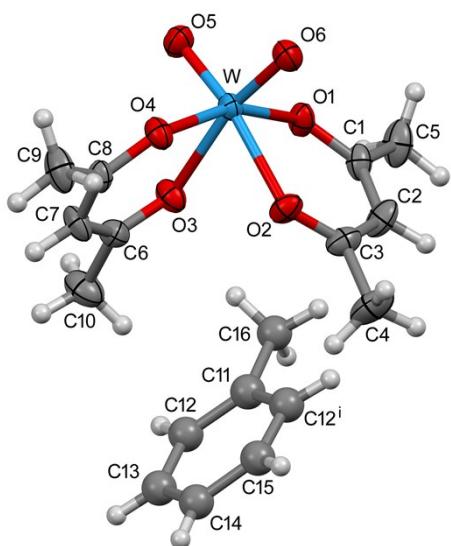


Fig. S1 ORTEP drawings of the $[\text{WO}_2(\text{acac})_2] \cdot 0.5\text{C}_6\text{H}_5\text{Me}$ with the atom-labeling scheme (displacement ellipsoids of non-hydrogen atoms are drawn at the 50 % probability level).

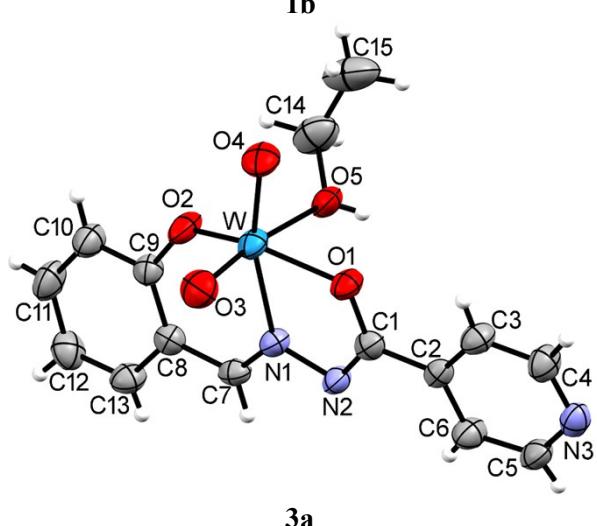
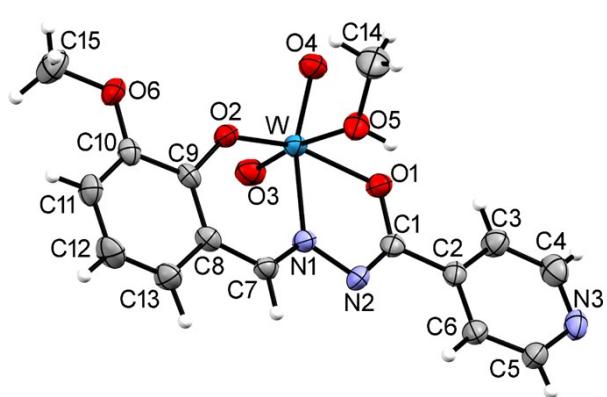
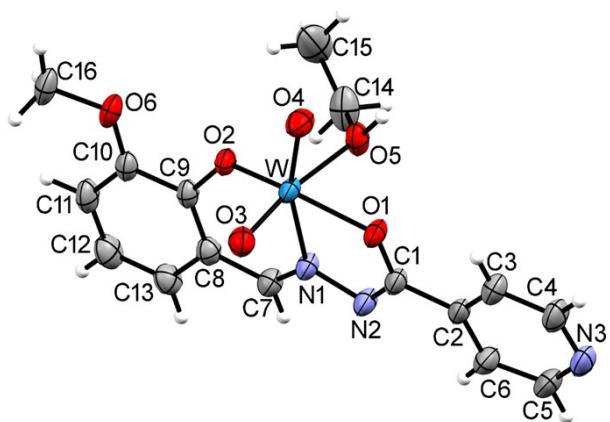


Fig. S2 ORTEP drawings of **1a**, **1b** and **3a** with the atom-labeling scheme (displacement ellipsoids of non-hydrogen atoms are drawn at the 50% probability level).

Table S1 Selected bond lengths (\AA) and angles ($^\circ$) for compound $[\text{WO}_2(\text{acac})_2] \cdot 0.5\text{C}_6\text{H}_5\text{Me}$

$[\text{WO}_2(\text{acac})_2] \cdot 0.5\text{C}_6\text{H}_5\text{Me}$	
W–O1	1.989(4)
W–O2	2.166(3)
W–O3	2.170(4)
W–O4	1.971(4)
W–O5	1.718(4)
W–O6	1.712(4)
O1–C1	1.296(7)
C1–C2	1.355(9)
C2–C3	1.386(9)
C3–O2	1.267(7)
O3–C6	1.247(8)
C6–C7	1.415(8)
C7–C8	1.350(9)
C8–O4	1.296(8)
O1–W–O2	80.97(14)
O1–W–O3	82.27(17)
O1–W–O4	159.45(17)
O1–W–O5	92.67(18)
O1–W–O6	98.84(19)
O2–W–O3	75.54(17)
O2–W–O4	83.15(17)
O2–W–O5	164.71(17)
O2–W–O6	90.27(19)
O3–W–O4	81.19(17)
O3–W–O5	89.89(19)
O3–W–O6	165.5(2)
O4–W–O5	99.4(2)
O4–W–O6	94.18(19)
O5–W–O6	104.5(2)

Table S2 Angle between the phenyl and the pyridine moieties, φ ($^{\circ}$) and angle between the five- and six-membered chelate rings, ψ ($^{\circ}$) for compounds **1a**, **1b** and **3a**

	$\varphi /{}^{\circ}$	$\psi /{}^{\circ}$
1a	7.39(11)	10.94(17)
1b	5.20(14)	9.34(19)
3a	8.94(12)	16.87(16)

Table S3 Geometry of intra- and intermolecular hydrogen bonds (\AA , $^{\circ}$) for compounds **1a**, **1b** and **3a**

	D–H \cdots A	D–H (\AA)	H \cdots A (\AA)	D \cdots A (\AA)	D–H \cdots A($^{\circ}$)
1a	O5–H50 \cdots N3 ^a	0.83	1.9	2.723(4)	172
	C5–H5 \cdots O4 ^b	0.93	2.47	3.097(4)	125
	C7–H7 \cdots O3 ^c	0.93	2.49	3.097(4)	123
	C15–H18b \cdots O4 ^d	0.96	2.52	3.374(5)	148
1b	O5–H50 \cdots O1	0.83	2.38	2.811(4)	113
	O5–H50 \cdots N3 ^e	0.83	2.01	2.692(4)	139
	C4–H4 \cdots O4 ^f	0.93	2.58	3.421(5)	151
	C7–H7 \cdots O3 ^g	0.93	2.35	3.253(4)	165
	C15–H15b \cdots O4 ^h	0.96	2.57	3.408(5)	146
3a	O5–H50 \cdots N3 ⁱ	0.72(3)	1.98(3)	2.700(4)	175(4)
	C3–H3 \cdots O2 ^j	0.93	2.51	3.380(4)	156
	C7–H7 \cdots O4 ^h	0.93	2.57	3.461(4)	161

^a2-x,2-y,2-z; ^bx,1+y,z; ^c2-x,1-y,1-z; ^d2-x,1-y,2-z; ^e-x,-y,1-z; ^f1-x,-y,1-z; ^g-1+x,y,z; ^h-1/2+x,1/2-y,-1/2+z; ⁱ1-x,-y,-z; ^j1/2-x,-1/2+y,1/2-z

Table S4 Geometry of $\pi\cdots\pi$ interactions involved in the formation of dimers and between neighbouring dimer molecules for compounds **1a**, **1b** and **3a**

d($\pi\cdots\pi$) /Å	
1a	d(Cg3…Cg3) ⁱ =3.608(2)
	d(Cg1…Cg4) ⁱⁱ =4.2013(18)
1b	d(Cg3…Cg3) ⁱⁱⁱ =3.549(2)
	d(Cg1…Cg3) ^{iv} =4.284(2)
3a	d(Cg3…Cg3) ^v =3.5516(19)
	(Cg4…Cg4) ^{vi} =3.9292(19)

ⁱ 2-x,2-y,2-z; ⁱⁱ 2-x,1-y,1-z; ⁱⁱⁱ-x,-y,1-z; ^{iv} 1-x,-y,-z; ^v-x,1-y,2-z; ^{vi}-x,1-y,-z;

Cg1 is the centroid of the ring W, O1, C1, N1 and N2, Cg3 is the centroid of the ring N3, C2–C5 and Cg4 is the centroid of the ring C8–C13 in **1a**, **1b** and **3a**.

Powder X-ray diffraction patterns

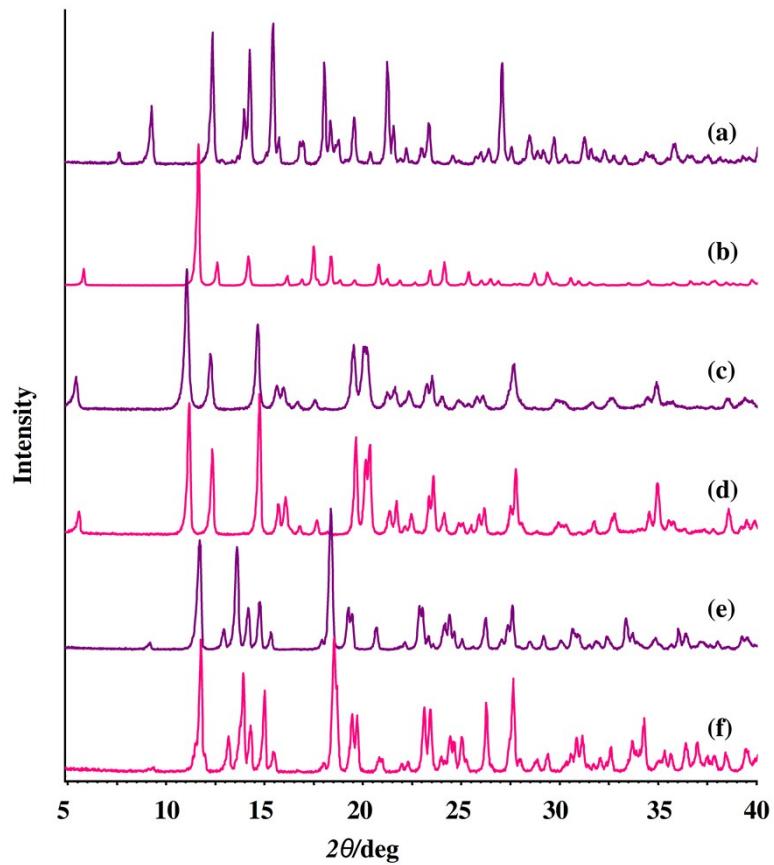
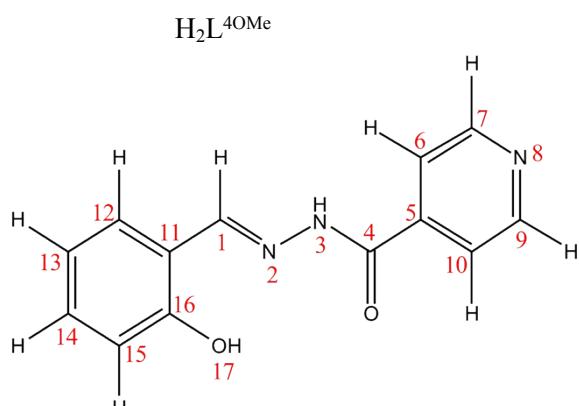
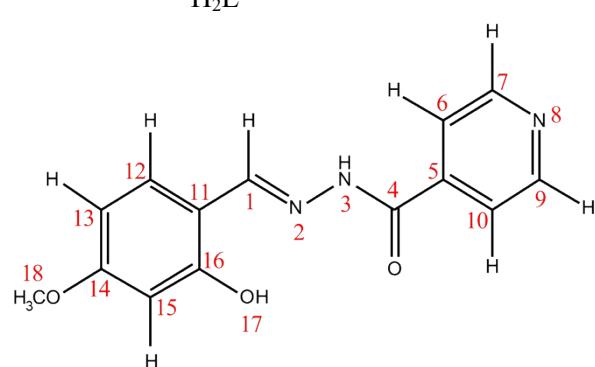
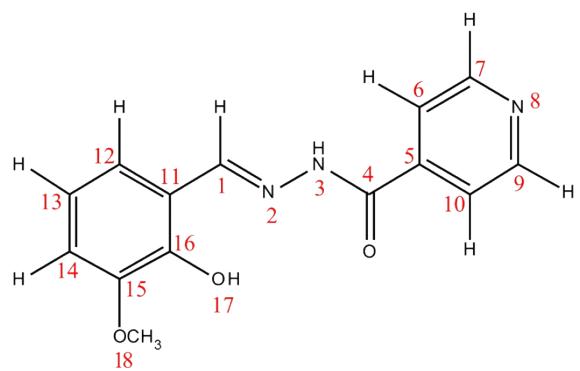


Fig. S3 Powder X-ray diffraction patterns of **1a** (a), **1b** (b), **2a** (c), **2b** (d), **3a** (e) and **3b** (f).

NMR numbering scheme



Scheme S1 The structural formula of $\text{H}_2\text{L}^{\text{R}}$ with the NMR numbering scheme.

NMR spectroscopy

Table S5 ^1H and ^{13}C chemical shifts (ppm) of $\text{H}_2\text{L}^{3\text{OMe}}$, $\text{H}_2\text{L}^{4\text{OMe}}$ and $\text{H}_2\text{L}^\text{H}$

Atom	$\text{H}_2\text{L}^{3\text{OMe}}$		$\text{H}_2\text{L}^{4\text{OMe}}$		$\text{H}_2\text{L}^\text{H}$	
	δ / ppm (^1H)	δ / ppm (^{13}C)	δ / ppm (^1H)	δ / ppm (^{13}C)	δ / ppm (^1H)	δ / ppm (^{13}C)
1	8.71	149.30	8.59	150.06	8.70	149.46
2	-	-	-	-	-	-
3	12.27	-	12.20	-	12.30	-
4	-	161.78	-	161.55	-	161.81
5	-	140.60	-	140.57	-	140.45
6	7.85	121.57	7.84	121.93	7.85	121.97
7	8.80	150.85	8.80	150.83	8.81	150.86
8	-	-	-	-	-	-
9	8.80		8.80	150.83	8.81	150.86
10	7.85	121.97	7.84	121.93	7.85	121.97
11	-	119.45	-	112.17	-	119.16
12	7.21	120.90	7.48	131.50	7.61	129.70
13	6.88	119.45	6.53	107.09	6.94	119.90
14	7.05	114.45	-	162.83	7.32	132.21
15	-	148.46	6.51	101.64	6.95	116.92
16	-	147.65	-	159.93	-	157.95
17	10.71	-	11.43	-	11.09	-
18	3.83	56.31	3.79	55.81	-	-

Table S6 ^1H and ^{13}C chemical shifts (ppm) of complexes **1a–3a** and **1b–3b**

Atom	1a and 1b		2a and 2b		3a and 3b	
	δ / ppm (^1H)	δ / ppm (^{13}C)	δ / ppm (^1H)	δ / ppm (^{13}C)	δ / ppm (^1H)	δ / ppm (^{13}C)
1	9.01	159.64	8.91	159.03	9.04	159.62
2	-	-	-	-	-	-
3	-	-	-	-	-	-
4	-	167.30	-	165.98	-	167.28
5	-	137.51	-	137.62	-	137.50
6	7.91	122.05	7.88	121.92	7.97	122.06
7	8.80	151.19	8.77	151.12	8.79	151.18
8	-	-	-	-	-	-
9	8.80	151.19	8.77	151.12	8.79	151.18
10	7.91	122.05	7.88	121.92	7.91	122.06
11	-	121.18	-	114.32	-	120.94
12	7.36	126.18	7.70	136.51	7.80	135.35
13	7.08	122.39	6.74	110.42	7.14	122.55
14	7.36	118.58	-	166.62	7.64	136.53
15	-	149.75	6.66	104.54	7.06	120.06
16	-	148.40	-	160.91	-	158.76
17	-	-	-	-	-	-
18	3.85	56.48	3.86	56.48	-	-

* Signals belonging to EtOH and MeOH were also detected in ^1H NMR spectra in dmso solutions.

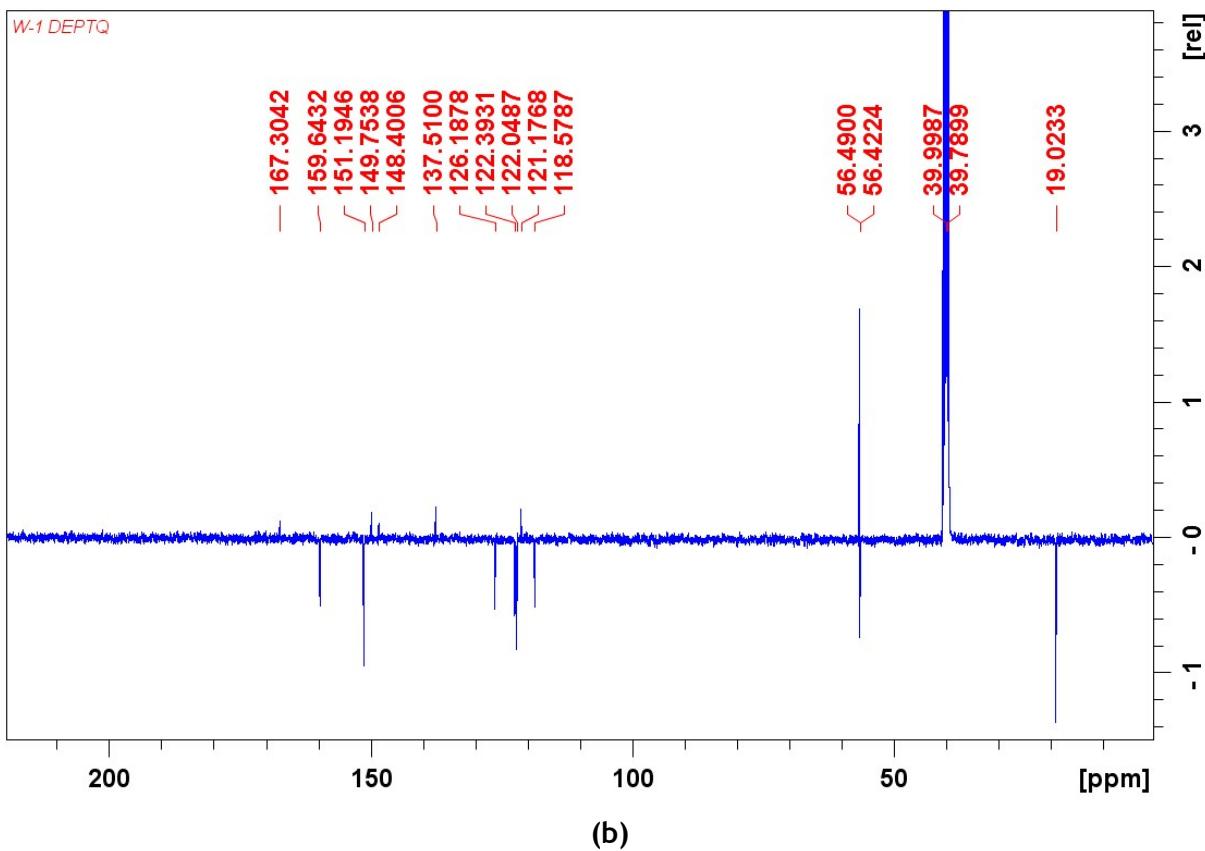
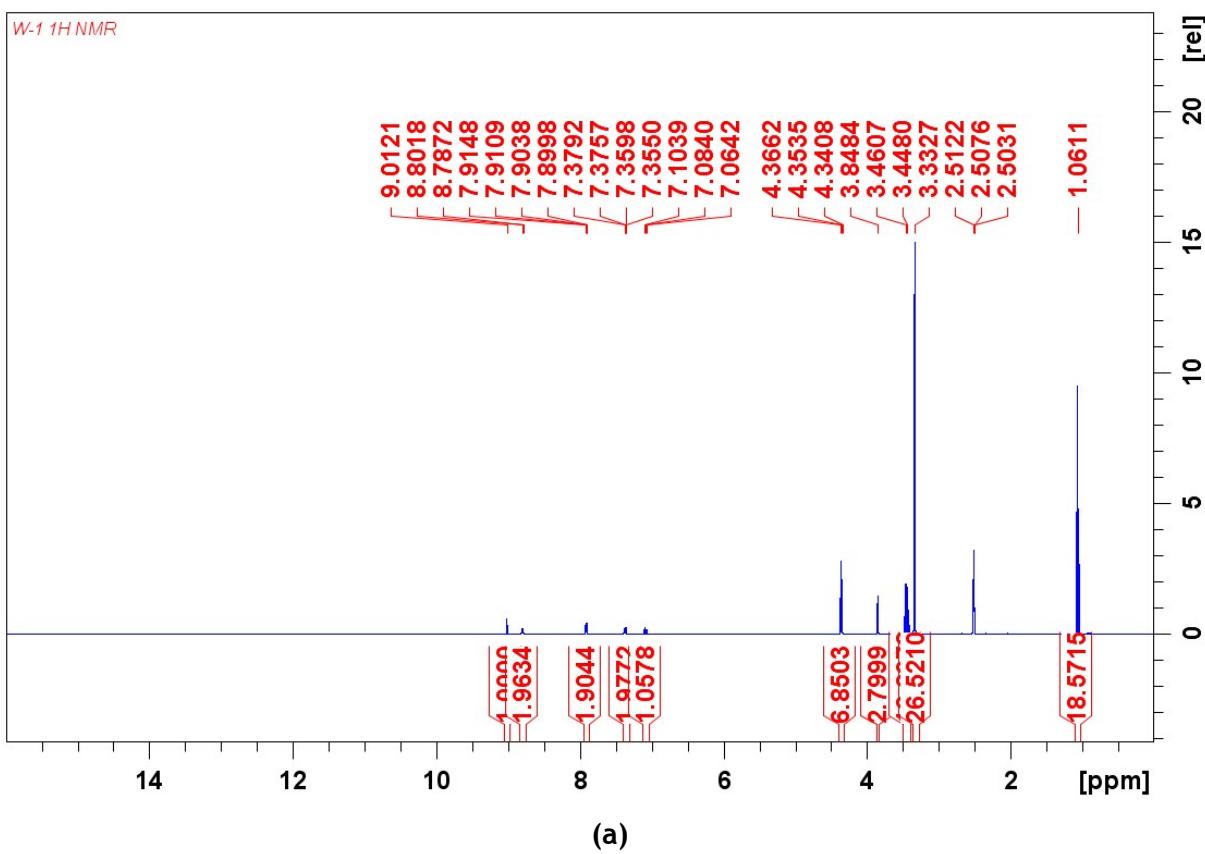


Fig S4 ¹H (a) and ¹³C NMR (b) spectra of complex **1a** in DMSO-*d*₆

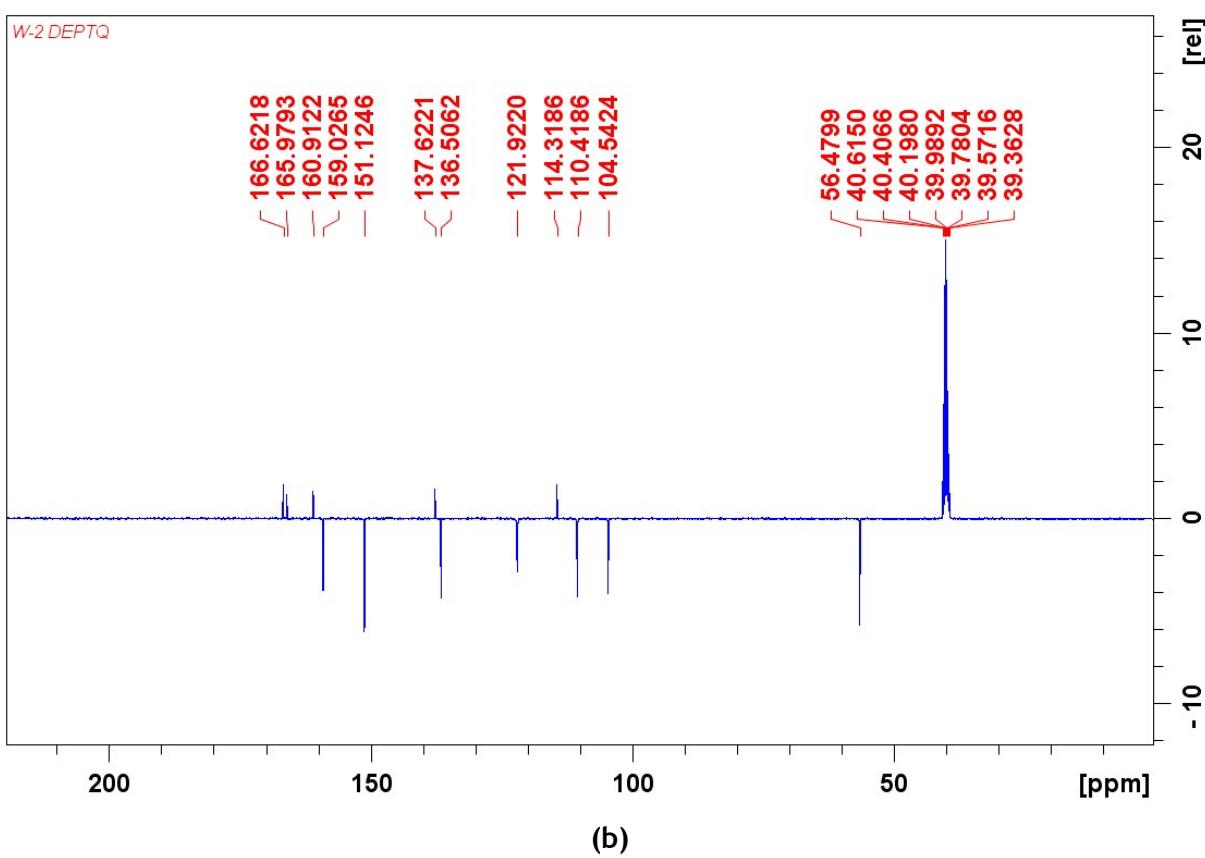
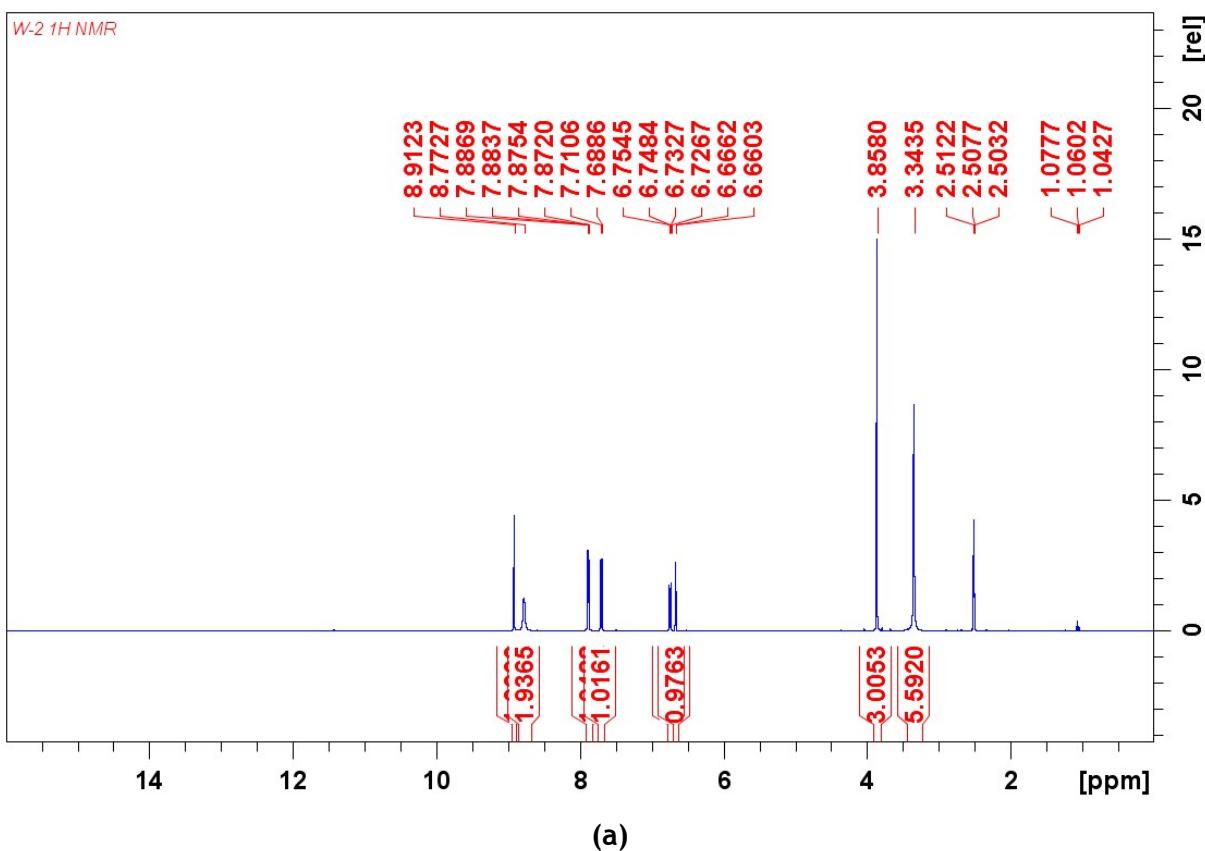


Fig S5 ^1H (a) and ^{13}C NMR (b) spectra of complex **2a** in $\text{DMSO}-d_6$

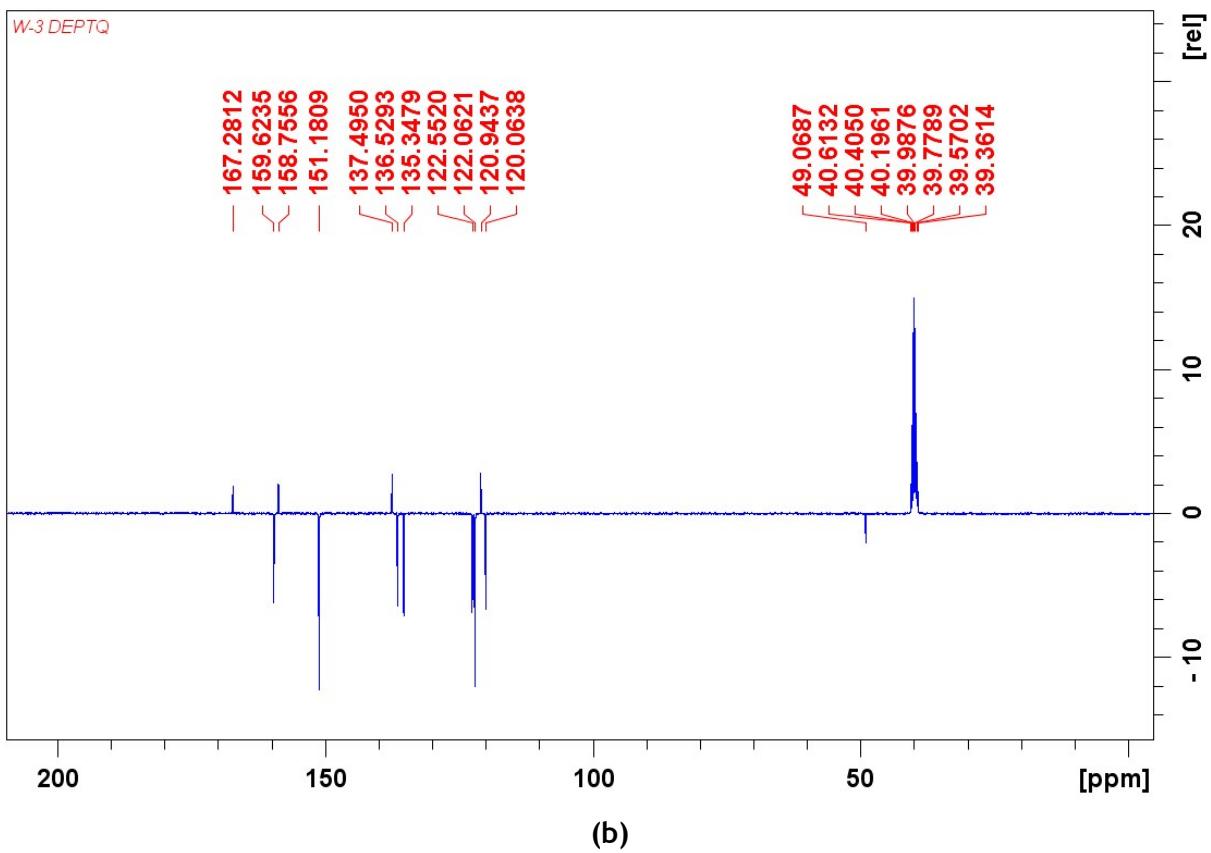
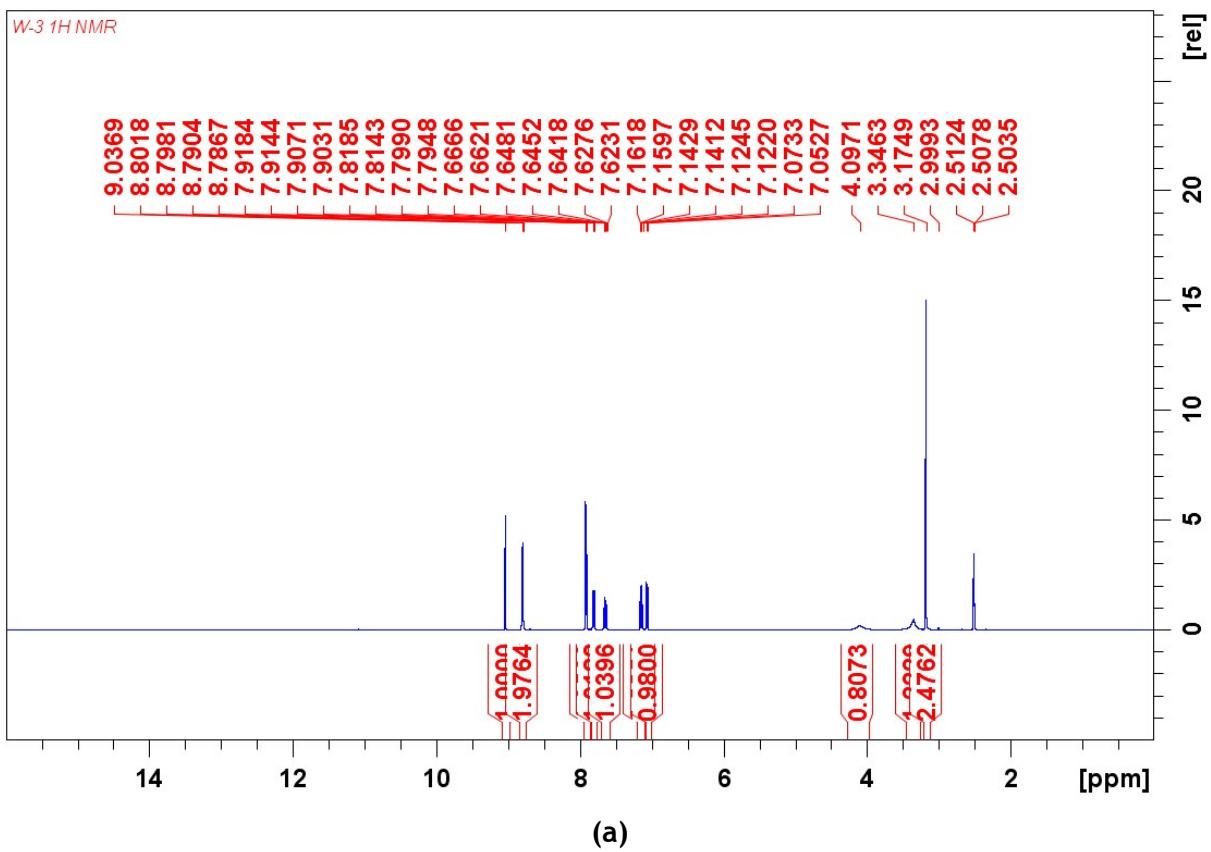


Fig S6 ^1H (a) and ^{13}C NMR (b) spectra of complex **3a** in $\text{DMSO}-d_6$

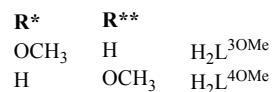
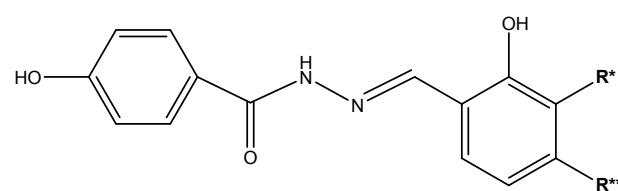
Table S7 Relevant catalytic results for the cyclooctene epoxidation by aqueous TBHP^a taken from Ref.12.

*H₂L^{3OMe} is 3-methoxysalicylaldehyde 4-hydroxybenzhydrazone

**H₂L^{4OMe} is 4-methoxysalicylaldehyde 4-hydroxybenzhydrazone

Compound	Conversion ^b	TOF _{20min} ^c /h ⁻¹	TON ^d
[WO ₂ (L ^{3OMe*})(EtOH)]	17	190	70
[WO ₂ (L ^{4OMe*})(EtOH)]	31	260	128

^a Reaction conditions: time, 5 h; temperature, 80 °C; [W]/cyclooctene/TBHP molar ratio: 0.25/100/200 for all compounds. ^b For cyclooctene, calculated after 5 h. ^cn(cyclooctene transformed)/n(catalyst)/time at 20 minutes. ^dn(cyclooctene transformed)/n(catalyst at 5 h).



Scheme S2 The structural formula of ligands H₂L taken from Ref.12.