## **Electronic Supplementary Information (ESI)**



Fig. S1 (a) XRD pattern and (b) SEM image of the ZnS.



Fig. S2 The energy-dispersive X-ray (EDX) spectrum of the mesoporous hierarchical  $ZnS@\beta-Ni(OH)_2$  microspheres.



Fig. S3 N<sub>2</sub> adsorption-desorption isotherms and pore size distribution curves of the samples: (a, b)  $ZnS@\beta-Ni(OH)_2$ ; (c, d) ZnS and (e, f)  $\beta-Ni(OH)_2$ .



Fig. S4 (a) XRD pattern and (b) SEM image of the  $\beta\text{-Ni(OH)}_2.$ 



Fig. S5 (a) Charge-discharge curves and (b) specific capacitances of ZnS at different current densities; (c) Charge-discharge curves and (d) specific capacitances of  $\beta$ -Ni(OH)<sub>2</sub> at different current densities.



Fig. S6 SEM image of the hierarchical  $ZnS@\beta$ -Ni(OH)<sub>2</sub> microspheres after 6000 charge-discharge cycles at 6.0 A g<sup>-1</sup>.



Fig. S7 The schematic crystal structures of  $\beta$ -Ni(OH)<sub>2</sub> super cell (2×2×2 slabs) projected based on data of ICSD-24015.



Fig. S8 Electrochemical impedance spectra of different electrodes materials at room temperature in 3.0 M KOH solutions



Fig. S9 Flexibility and stability of the flexible solid state  $ZnS@\beta$ -Ni(OH)<sub>2</sub>//activated carbons hybrid supercapacitors device at 30 mV s<sup>-1</sup> with different bending angles.



**Fig. S10** The CV curves after bending for certain times: (a) 30° for 10 times; (b) 30° for 30 times; (c) 60° for 10 times; (d) 60° for 30 times; (e) 90° for 10 times; (f) 90° for 30 times; (g) 180° for 10 times; (h) 180° for 30 times.