Electronic Supplementary Material (ESI) for Chemical Science. This journal is © The Royal Society of Chemistry 2016

Electronic Supplementary Information

Competing Multiple Pathways of Chemical Reaction: Drastic Shortcut of Reaction for Self-Catalytic Double-Helix Formation of Helicene Oligomers

Yo Kushida, Nozomi Saito, Masanori Shigeno, and Masahiko Yamaguchi*

Department of Organic Chemistry, Graduate School of Pharmaceutical Sciences,

Tohoku University, Aoba, Sendai, 980-8578, Japan

*E-mail: yama@m.tohoku.ac.jp, Fax: (+81) 22-795-6811.

Table of contents

General method	S2
1. Materials	S2
2. Freezing-defrosting experiments	S2
3. Simple-cooling experiments	
4. The molecular structures at the initial stage at 25 $^{\circ}C$	
5. A-to-C and A-to-B-to-C reaction after cooling at -10 °C	S7
6. Experiments at the concentration of 2.5 \times 10^{-4} M	
7. Temperature effect in the A-to-C reaction and A-to-B-to-C reaction	
8. Self-catalysis in the A-to-C reaction	
9. Self-catalysis in the A-to-B reaction	
10. References	S17

General method. CD and UV-Vis spectra were measured on a JASCO J-720 spectropolarimeter. The path length 0.0107 cm of quartz round cell and 0.0217 cm of quartz square cell were used. Dynamic light scattering (DLS) determined at 173° scattering angle was observed by a Zetasizer Nano S.

1. Materials

(P)-1 and (M)-2 were synthesized as described in our previous work.^{S1}

2. Freezing-defrosting experiments

< Sample preparation >

A 1:1 (*P*)-1/(*M*)-2 (total concentration 5.0×10^{-4} M) mixture: (*P*)-1 (0.70 mg, 2.5×10^{-4} mmol) and (*M*)-2 (0.84 mg, 2.5×10^{-4} mmol) were dissolved in fluorobenzene (1.0 mL) with heating at 70 °C for 20 min.

< Freezing-defrosting/rapid experiment >

A solution of 1:1 (*P*)-1/(*M*)-2 mixture (total 5.0×10^{-4} M, fluorobenzene) was heated at 70 °C for 30 min, snap cooled at -35 °C by immersing in a cooled bath at -35 °C, transferred to a CD cell adjusted to -25 °C, and allowed to settle for 15 min. The solution was rapidly heated at 25 °C, and allowed to settle for CD analysis (Fig. 3).

< Freezing-defrosting/constant-rate experiment (heating from –25 to 25 °C) at the rate of 2 Kmin⁻¹>

A solution of 1:1 (*P*)-1/(*M*)-2 mixture (total 5.0×10^{-4} M, fluorobenzene) was heated at 70 °C for 30 min, snap cooled at -35 °C by immersing in a cooled bath at -35 °C, transferred to a CD cell adjusted to -25 °C, and allowed to settle for 15 min. The solution was heated from -25 °C to 25 °C at the rate of 2 Kmin⁻¹, and allowed to settle for 120 min for CD analysis (Fig. S1).



Fig. S1 The $\Delta \varepsilon$ (315 nm)/time profile of a 1:1 (*P*)-1/(*M*)-2 mixture in fluorobenzene (total 5.0 × 10⁻⁴ M). The solution was heated from -25 °C to 25 °C at the constant-rate of 2 Kmin⁻¹ (red line), and settled at 25 °C for 120 min (green circles). The green lines are drawn between points.

3. Simple-cooling experiments

< Simple-cooling/rapid experiment >

A solution of 1:1 (*P*)-1/(*M*)-2 mixture (total 5.0×10^{-4} M, fluorobenzene) was first heated at 70 °C for 30 min, rapidly cooled at 25 °C, then allowed to settle for CD analysis.

< Simple-cooling/constant-rate experiment (cooling from 70 to 25 °C) at the rate of 2 Kmin⁻¹>

A solution of 1:1 (*P*)-1/(M)-2 mixture (total 5.0 × 10⁻⁴ M, fluorobenzene) was heated at 70 °C for 30 min. The solution was cooled from 70 °C to 25 °C at the rate of 2 Kmin⁻¹, and allowed to settle for 25 min for CD analysis (Fig. S2).



Fig. S2 The $\Delta \varepsilon$ (315 nm)/time profile of a 1:1 (*P*)-1/(*M*)-2 mixture in fluorobenzene (total 5.0 × 10⁻⁴ M). The solution was cooled from 70 °C to 25 °C at the constant-rate of 2 Kmin⁻¹ (blue line), and settled at 25 °C for 25 min (green circles). The green lines are drawn between points.

4. The molecular structures at the initial stage at 25 °C

<< CD Studies >>>

< Freezing-defrosting/rapid experiment >

A solution of 1:1 (*P*)-1/(*M*)-2 mixture (total 5.0×10^{-4} M, fluorobenzene) was heated at 70 °C for 30 min, snap cooled at -35 °C by immersing in a cooled bath at -35 °C, transferred to a CD cell adjusted to -25 °C, and allowed to settle for 15 min. The solution was rapidly heated at 25 °C and allowed to settle for 20 min, during which CD spectra were obtained at 20 sec intervals until 5 min and at 5 min intervals between 5 and 20 min (Fig. S3).



Fig. S3 (a) CD spectra of 1:1 (*P*)-1/(*M*)-2 mixture (total 5.0×10^{-4} M, fluorobenzene) showing the initial state in the structural changes from random coil A at -25 °C to hetero-double helix C at 25 °C under the freezing-defrosting/rapid conditions. CD spectrum of S-random-coil A was obtained in our previous work.^{S2} (b) Extraction and magnification of (a).

< Simple-cooling/rapid experiment >

A solution of 1:1 (*P*)-1/(M)-2 mixture (total 5.0 × 10⁻⁴ M, fluorobenzene) was heated at 70 °C for 30 min providing an S-random-coil solution. The solution was rapidly cooled at 25 °C then allowed to settle for 5 min, during which CD spectra were obtained at 20 sec intervals (Fig. S4).



Fig. S4 (a) CD spectra of 1:1 (*P*)-1/(*M*)-2 mixture (total 5.0×10^{-4} M, fluorobenzene) showing the initial state in the structural changes from random coil **A** at 70 °C to hetero-double helix **B** at 25 °C under the simple-cooling/rapid conditions. CD spectrum of S-random-coil **A** was obtained in our previous work.^{S2} (b) Extraction and magnification of (a).

The spectra in Fig. S3 and S4 were very similar during the initial 300 seconds.

<< NMR study >>

< Freezing-defrosting/rapid experiments >

A solution of 1:1 (*P*)-1/(M)-2 mixture (total 5.0 × 10⁻⁴ M, fluorobenzene) was heated at 70 °C for 30 min, snap cooled at -35 °C by immersing in a cooled bath at -35 °C, transferred to a CD cell adjusted to -25 °C, and allowed to settle for 15 min. The solution was rapidly heated at 25 °C, allowed to settle for 2 min, and analyzed by NMR.

<Simple-cooling/rapid experiment >

A solution of 1:1 (*P*)-1/(*M*)-2 mixture (total 5.0×10^{-4} M, fluorobenzene) was first heated at 70 °C for 30 min, rapidly cooled at 25 °C, allowed to settle for 2 min, and analyzed by NMR (Fig. S5).



Fig. S5 ¹H-NMR spectra of 1:1 (*P*)-1/(*M*)-2 mixture in d_5 -fluorobenzene (total 5.0×10^{-4} M) in freezing-defrosting and simple-cooling experiments at 25 °C. In the spectra (a) and (b), the solvent residual peaks were adjusted to the same height.

< Freezing-defrosting/rapid experiment >

A solution of 1:1 (*P*)-1/(*M*)-2 mixture (total 5.0×10^{-4} M, fluorobenzene) was heated at 70 °C for 30 min, snap cooled at -35 °C by immersing in a cooled bath at -35 °C, transferred to a CD cell adjusted to -25 °C, and allowed to settle for 15 min. The solution was rapidly heated at 25 °C then allowed to settle for 2 min for DLS analysis.

< Simple-cooling/rapid experiment >

A solution of 1:1 (*P*)-1/(*M*)-2 mixture (total 5.0×10^{-4} M, fluorobenzene) was first heated at 70 °C for 30 min, rapidly cooled at 25 °C, and allowed to settle at that temperature for DLS analysis.



Fig. S6 DLS analysis of 1:1 (*P*)-1/(*M*)-2 mixture in fluorobenzene (total 5.0×10^{-4} M) in freezing-defrosting/rapid and simple-cooling/rapid experiments. The average diameters of 5 nm and 6 nm were obtained at 2 min after heating the solution from -25 °C to 25 °C and at 5 min after cooling the solution from 70 °C to 25 °C, respectively.

5. A-to-C and A-to-B-to-C reaction after cooling at -10 °C

< A-to-C reaction: Freezing-defrosting/rapid experiment with snap-cooling at -10 °C >

A solution of 1:1 (*P*)-1/(*M*)-2 mixture (total 5.0×10^{-4} M, fluorobenzene) was first heated at 70 °C for 30 min, snap cooled to -35 °C for 15 min by immersing in a cooled bath at -35 °C, transferred to a CD cell adjusted to -10 °C, and allowed to settle for 30 min. Then, the solution was rapidly heated at 25 °C, and allowed to settle for 300 min (Fig. 5, green circles; Fig. S7a).

<A-to-B-to-C reaction: Trapping B by constant-rate cooling to -10 °C at the rate of 2 Kmin⁻¹ then rapid-heating experiment >

A solution of 1:1 (*P*)-1/(*M*)-2 mixture (total 5.0×10^{-4} M, fluorobenzene) was first heated at 70 °C for 30 min, cooled from 70 °C to -10 °C at the rate of 2 K/min, and allowed to settle for 30 min at -10 °C. Then, the solution was rapidly heated at 25 °C, and allowed to settle for 5040 min (Fig. 5, blue circles; Fig. S7b).



Fig. S7 CD spectra of 1:1 (*P*)-1/(*M*)-2 mixture (total 5.0×10^{-4} M, fluorobenzene) showing the structural changes to hetero-double helix C at 25 °C. The solution was prepared by (a) freezing-defrosting/rapid method or (b) constant-rate cooling method at the rate of 2 Kmin⁻¹.

6. Experiments at the concentration of 2.5×10^{-4} M

< Sample preparation >

A 1:1 (*P*)-1/(*M*)-2 (total concentration 2.5×10^{-4} M) mixture: (*P*)-1 (0.35 mg, 1.25×10^{-4} mmol) and (*M*)-2 (0.420 mg, 1.25×10^{-4} mmol) were dissolved in fluorobenzene (1.0 mL) with heating at 70 °C for 20 min.

< Equilibrium states >

A solution of 1:1 (*P*)-1/(*M*)-2 mixture (total 2.5×10^{-4} M, fluorobenzene) was heated at 70 °C for 30 min, cooled at 25 °C, and allowed to settle for 86 h providing a solution of hetero-double helix C. The solution was cooled at 5 °C for 90 min, 25 °C for 60 min, and 40 °C for 115 min (Fig. S8). At each temperature, the solution was allowed to settle and analyzed by CD for the times described in parentheses (Fig. S8), when $\Delta\varepsilon$ change was less than 2 cm⁻¹M⁻¹ within a 10 min period. Then, the equilibrium at 5 °C with $\Delta\varepsilon = +410$ cm⁻¹M⁻¹, 25 °C with $\Delta\varepsilon = +360$ cm⁻¹M⁻¹, and 40 °C with +140 cm⁻¹M⁻¹ were determined.



Fig. S8 (a) CD and (b) UV-Vis spectra of 1:1 (*P*)-1/(M)-2 mixture in fluorobenzene (total 2.5 × 10⁻⁴ M) at equilibrium. The spectra were obtained at each temperature after settled for the time shown in parenthesis.

< Freezing-defrosting/rapid experiments >

A solution of 1:1 (*P*)-1/(*M*)-2 mixture (total 2.5×10^{-4} M, fluorobenzene) was heated at 70 °C for 30 min, snap cooled at -35 °C by immersing in a cooled bath at -35 °C, transferred to a CD cell adjusted to -25 °C, and allowed to settle for 15 min. The solution was rapidly heated either at 5 °C, 25 °C, or 40 °C, and allowed to settle for 50-780 min for CD analysis (Fig. S9).



Fig. S9 The $\Delta \varepsilon$ (315 nm)/time profiles of 1:1 (*P*)-1/(*M*)-2 mixtures in fluorobenzene (total 2.5 × 10⁻⁴ M) under freezing-defrosting conditions at different temperatures. The profiles at 5, 25, and 40 °C are shown by blue, green, red circles, respectively. The lines are drawn between points. The inset shows the magnification of the profiles between 0 and 120 min. The $\Delta \varepsilon$ values at equilibrium at 5, 25, and 40 °C are shown in purple, pale green, and brown lines, respectively, which were obtained in Fig. S8.

< Simple-cooling/rapid experiments >

A solution of 1:1 (*P*)-1/(*M*)-2 mixture (total 2.5×10^{-4} M, fluorobenzene) was first heated at 70 °C for 30 min, rapidly cooled either at 40, 25, 15, 5, 0, -10, -15, -17.5, or -20 °C, and allowed to settle. The $\Delta \varepsilon$ (315 nm) value at each temperature was followed by CD for 190 min at 40 °C, 40 min between 35 and 5 °C, and 30 min between 0 and -20 °C (Fig. S10). The obtained $\Delta \varepsilon$ values at 20 min were plotted in red circles in Fig. S11.



Fig. S10 (a) The $\Delta \varepsilon$ (315 nm)/time profiles of 1:1 (*P*)-1/(*M*)-2 mixtures in fluorobenzene (total 2.5 × 10^{-4} M) under simple-cooling/rapid conditions. The lines are drawn between points. (b) Magnification

of (a).



Fig. S11 The $\Delta\varepsilon$ (315 nm)/temperature profiles of 1:1 (*P*)-1/(*M*)-2 mixtures in fluorobenzene (total 2.5 × 10⁻⁴ M), which were obtained by plotting the $\Delta\varepsilon$ values reached after 20 min periods under the freezing-defrosting/rapid conditions (red circles). The $\Delta\varepsilon$ values obtained under simple-cooling/rapid conditions (red squares) and the values at quilibrium states are also shown (blue circles). The $\Delta\varepsilon$ values of S-random-coil A, S-Hetero-double helices B and C, and at equilibrium between 5 and 50 were obtained in our previous work.^{S2}

< Effect of "freezing" temperature >

A solution of 1:1 (*P*)-1/(*M*)-2 mixture (total 2.5×10^{-4} M, fluorobenzene) was first heated at 70 °C for 30 min, rapidly cooled at -25, -20, -17.5, or -15 °C, and allowed to settle for 30 min. Then, the solution was rapidly heated at 25 °C and the $\Delta \varepsilon$ values at 315 nm was followed by CD (Fig. S12).



Fig. S12 Time course of the A-to-C reaction from different "freezing" temperatures obtained by the $\Delta \varepsilon$ (315 nm)/temperature profiles of 1:1 (*P*)-1/(*M*)-2 mixture in fluorobenzene (total 2.5 × 10⁻⁴ M) at 25 °C. The lines are drawn between points.

7. Temperature effect in the A-to-C reaction and A-to-B-to-C reaction

< Freezing-defrosting/rapid experiment: A-to-C reaction >

A solution of 1:1 (*P*)-1/(*M*)-2 mixture (total 5.0×10^{-4} M, fluorobenzene) was heated at 70 °C for 30 min, snap cooled at -35 °C by immersing in a cooled bath at -35 °C, transferred to a CD cell adjusted to -25 °C, and allowed to settle for 15 min. The solution was rapidly heated either at 5 °C, 25 °C, 40 °C, or 50 °C and allowed to settle for 40-300 min for CD analysis (Fig. S13).



Fig. S13 The $\Delta \varepsilon$ (315 nm)/time profiles of 1:1 (*P*)-1/(*M*)-2 mixtures in fluorobenzene (total 5.0 × 10⁻⁴ M) under freezing-defrosting conditions at different temperatures. The profiles at 5, 25, 40, and 50 °C are shown by blue, green, red circles, respectively. The lines are drawn between points. The inset shows the magnification of the profiles between 0 and 60 min. The $\Delta \varepsilon$ values at equilibrium at 5, 25, 40, and 50 °C are shown in purple, pale green, and brown lines, respectively, which were obtained in our previous work.^{S2}

< Simple-cooling/rapid experiment: A-to-B-to-C reaction >

A solution of 1:1 (*P*)-1/(*M*)-2 mixture (total 5.0×10^{-4} M, fluorobenzene) was first heated at 70 °C for 30 min, rapidly cooled either at 50, 40, 25, 15, 5, 0, -5, -10, -15, -16, -18, or -20 °C, and allowed to settle. The $\Delta \varepsilon$ (315 nm) value at each temperature was followed by CD for 600 min at 50 °C and 30 min between 40 and -20 °C (Fig. S14). The obtained $\Delta \varepsilon$ values at 315 nm after 20 min were plotted as red circles in Fig. 6.



Fig. S14 (a) The $\Delta \varepsilon$ (315 nm)/time profiles of 1:1 (*P*)-**1**/(*M*)-**2** mixtures in fluorobenzene (total 5.0 × 10⁻⁴ M) under simple-cooling conditions. The lines are drawn between points. (b) Magnification of (a).

< Freezing-defrosting/constant-rate experiment >

A solution of 1:1 (*P*)-1/(*M*)-2 mixture (total 5.0×10^{-4} M, fluorobenzene) was first heated at 70 °C for 30 min, snap cooled to -35 °C for 15 min by immersing in a cooled bath at -35 °C, transferred to a CD cell adjusted to -25 °C, and allowed to settle for 15 min. The solution was (1) heated from -25 to 70 °C at the constant-rate of 2 Kmin⁻¹, and then (2) cooled to from 70 to -25 °C at the constant-rate of 2 Kmin⁻¹, during which $\Delta \varepsilon$ values at 315 nm were followed (Fig. 7).

8. Self-catalysis in the A-to-C reaction

< Kinetic analysis >

A solution of 1:1 (*P*)-1/(*M*)-2 mixture (total 5.0×10^{-4} M, fluorobenzene) was heated at 70 °C for 30 min, snap cooled at -35 °C by immersing in a cooled bath at -35 °C, transferred to a CD cell adjusted to -25 °C, and allowed to settle for 15 min. The solution was rapidly heated at 5 °C, and allowed to settle for 450 min. The structural change was monitored by the $\Delta\varepsilon$ values at 315 nm (Fig. S15).

Slightly sigmoidal $\Delta \varepsilon$ /time profiles were obtained.



Fig. S15 $\Delta \varepsilon$ (315 nm)/time profiles of 1:1 (*P*)-**1**/(*M*)-**2** mixtures (total 5.0 × 10⁻⁴ M, fluorobenzene) at 5 °C under freezing-defrosting conditions. The $\Delta \varepsilon$ values of equilibrium was obtained in our previous work.^{S2}

< Seeding experiments >

A solution of 1:1 (*P*)-1/(*M*)-2 mixture (total 5.0×10^{-4} M, fluorobenzene, 0.60 mL) was transferred to a round quartz cell, heated at 70 °C for 30 min, cooled at 25 °C for 3-4 days, and heated at 40 °C for 100 min providing a solution of hetero-double helix C (Fig. S16). Another solution of (*P*)-1/(*M*)-2 mixture (total 5.0×10^{-4} M, fluorobenzene, 0.60 mL) was heated at 70 °C for 20 min providing a solution of Srandom-coil **A**. The solution of S-random-coil **A** was cooled at 40 °C for 5 sec, and then was added to the solution of hetero-double helix C in the quartz cell in a ratio of 1:1 (hetero-double helix C:random coil **A** = 0.60 mL:0.60 mL), and the solution was allowed to settle for 240 min for CD analysis (Fig. S17). Experiments in the mixing ratio of 1:2 (hetero-double helix C:random coil **A** = 0.40 mL:0.80 mL) and 1:3 (hetero-double helix C:random coil **A** = 0.30 mL:0.90 mL) were also conducted (Fig. S17).



Fig. S16 (a) CD and (b) UV-Vis spectra of (P)-1/(M)-2 mixture in fluorobenzene (total 5.0×10^{-4} M) at

40 °C, which were obtained at each time shown after heating the S-hetero-double-helix solution at 25 °C. S2



Fig. S17 Seeding experiments for the A-to-C reaction. (a) A schematic illustration of experimental procedure. (b) $\Delta \varepsilon$ (315 nm)/time profiles of 1:1 (*P*)-1/(*M*)-2 mixture in fluorobenzene (total 5.0 × 10⁻⁴ M) at 40 °C after an addition of random coil A. The inset shows the magnification of the profiles between 0 and 60 min. The circles show $\Delta \varepsilon$ change in the experiments. The pink, pale blue, and pale green squares are calculated value based on the initial ratio of hetero-double helix C:random coil A = 1:1, 1:2, and 1:3, respectively. The orange circles show $\Delta \varepsilon$ change obtained under simple-cooling conditions at 40 °C (Fig. S13, red circles), where the solution of hetero-double helix C is not added. The lines are drawn between points. The $\Delta \varepsilon$ value at equilibrium is shown in a purple line.^{S2}

9. Self-catalysis in the A-to-B reaction

< Seeding experiments >

At 45 °C:

A solution of 1:1 (*P*)-1/(*M*)-2 mixture (total 5.0×10^{-4} M, fluorobenzene, 0.60 mL) was transferred to a round quartz cell, heated at 70 °C for 20 min, and rapidly cooled at 45 °C for 20 min providing a solution of hetero-double helix **B**. Another solution of (*P*)-1/(*M*)-2 mixture (total 5.0×10^{-4} M, fluorobenzene, 0.60 mL) was heated at 70 °C for 20 min providing a solution of S-random-coil **A**. The S-random-coil **A** solution was cooled at 45 °C for 5 sec, and then was added to the solution of heterodouble helix **B** in the quartz cell in a ratio of 1:1 (hetero-double helix **B**:random coil **A** = 0.60 mL:0.60 mL), and the solution was allowed to settle for 20 min for CD analysis (Fig. S18b). At 25 °C:

A solution of 1:1 (*P*)-1/(*M*)-2 mixture (total 5.0×10^{-4} M, fluorobenzene, 0.60 mL) was transferred to a round quartz cell, heated at 70 °C for 20 min, and rapidly cooled at 25 °C for 20 min providing a solution of hetero-double helix **B**. Another solution of (*P*)-1/(*M*)-2 mixture (total 5.0×10^{-4} M, fluorobenzene, 0.60 mL) was heated at 70 °C for 20 min providing a solution of S-random-coil **A**. The S-random-coil **A** solution was cooled at 25 °C for 5 sec, and then was added to the solution of heterodouble helix **B** in the quartz cell in a ratio of 1:1 (hetero-double helix **B**:random coil **A** = 0.60 mL:0.60 mL), and the solution was allowed to settle for 20 min for CD analysis (Fig. S18c).



Fig. S18 Seeding experiments for the A-to-B reaction shown (a) A schematic illustration of experimental procedure. $\Delta \varepsilon$ (315 nm)/time profiles of 1:1 (*P*)-1/(*M*)-2 mixtures in fluorobenzene (total 5.0×10^{-4} M) at (b) 45 °C and (c) 25 °C. The brown curves show $\Delta \varepsilon$ change in the experiments. The purple squares are calculated values based on the ratio of hetero-double helix B:random coil A = 1:1. Increase and decrease in $\Delta \varepsilon$ values on addition of random coil A are due to removal and setting of quartz cell from CD instrument. The initial increases of hetero-double helix B are shown by the downward red arrows from the calculated $\Delta \varepsilon$ values.

< Control experiments >

At 45 °C and 25 °C:

A solution of 1:1 (*P*)-1/(*M*)-2 mixture (total 5.0×10^{-4} M, fluorobenzene, 1.2 mL) was transferred to a round quartz cell, heated at 70 °C for 20 min, and rapidly cooled at 45 °C (Fig. S19a) or 25 °C (Fig. S19b) for 40 min. Compare Fig. S17b with S18a, and also S17c with S18b: No decrease of $\Delta \varepsilon$ as shown by the red arrows appeared in this control experiments.

Addition of solvent (fluorobenzene) at 25 °C:

A solution of 1:1 (*P*)-1/(*M*)-2 mixture (total 5.0×10^{-4} M, fluorobenzene, 0.60 mL) was transferred to a round quartz cell, heated at 70 °C for 20 min, and cooled at 25 °C for 20 min providing a solution of hetero-double helix **B**. Fluorobenzene was added to the hetero-double helix **B** solution in the quartz cell in a ratio of 1:1 (hetero-double helix **B**:fluorobenzene = 0.60 mL:0.60 mL), and the solution was allowed to settle for 20 min for CD analysis (Fig. S19d). Compare with Fig. S18c.

Addition of hetero-double helix B to hetero-double helix B at 25 °C:

A solution of 1:1 (*P*)-1/(*M*)-2 mixture (total 5.0×10^{-4} M, fluorobenzene, 0.60 mL) was transferred to a round quartz cell, heated at 70 °C for 20 min, and cooled at 25 °C for 20 min providing a solution of hetero-double-helix **B**. Another solution of (*P*)-1/(*M*)-2 mixture (total 5.0×10^{-4} M, fluorobenzene, 0.60 mL) was heated at 70 °C for 20 min, and cooled at 25 °C for 20 min providing a solution of heterodouble helix **B**, which was added to the hetero-double helix **B** solution in the quartz cell in a ratio of 1:1 (hetero-double helix **B**:hetero-double helix **B** = 0.60 mL:0.60 mL), and the solution was allowed to settle for 20 min for CD analysis (Fig. S19e). No decrease of $\Delta\varepsilon$ was observed in this control experiment. Compare with Fig. S18c.



Fig. S19 Control experiments for the A-to-B reaction shown by $\Delta \varepsilon$ (315 nm)/time profiles of (*P*)-1/(*M*)-2 mixtures in fluorobenzene (total 5.0 × 10⁻⁴ M) with simple-cooling from 70 °C to (a) 45 °C and (b) 25 °C and with an addition of (d) fluorobenzene or (e) hetero-double helix B at 25 °C. (c) A schematic illustration of the experimental procedure in Fig. S19e.

10. References

(S1) Shigeno, M.; Sato, M.; Kushida, Y.; Yamaguchi, M. Asian J. Org. Chem., 2014, 3, 797.

(S2) Shigeno, M.; Kushida, Y.; Yamaguchi, M. J. Am. Chem. Soc., 2014, 136, 7972.