

Supplementary information for

**Synergistic Sodiation of Cobalt Oxide Nanoparticles and Conductive Carbon  
Nanotubes (CNTs) for Sodium-ion Battery**

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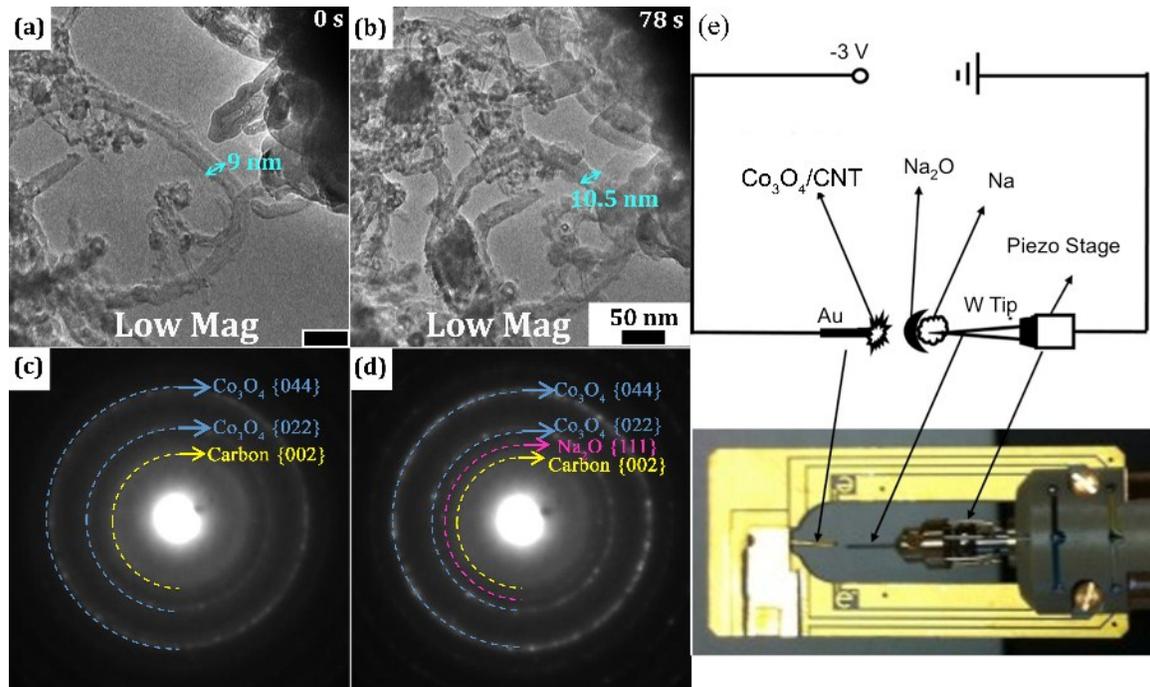
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## 1. METHODS

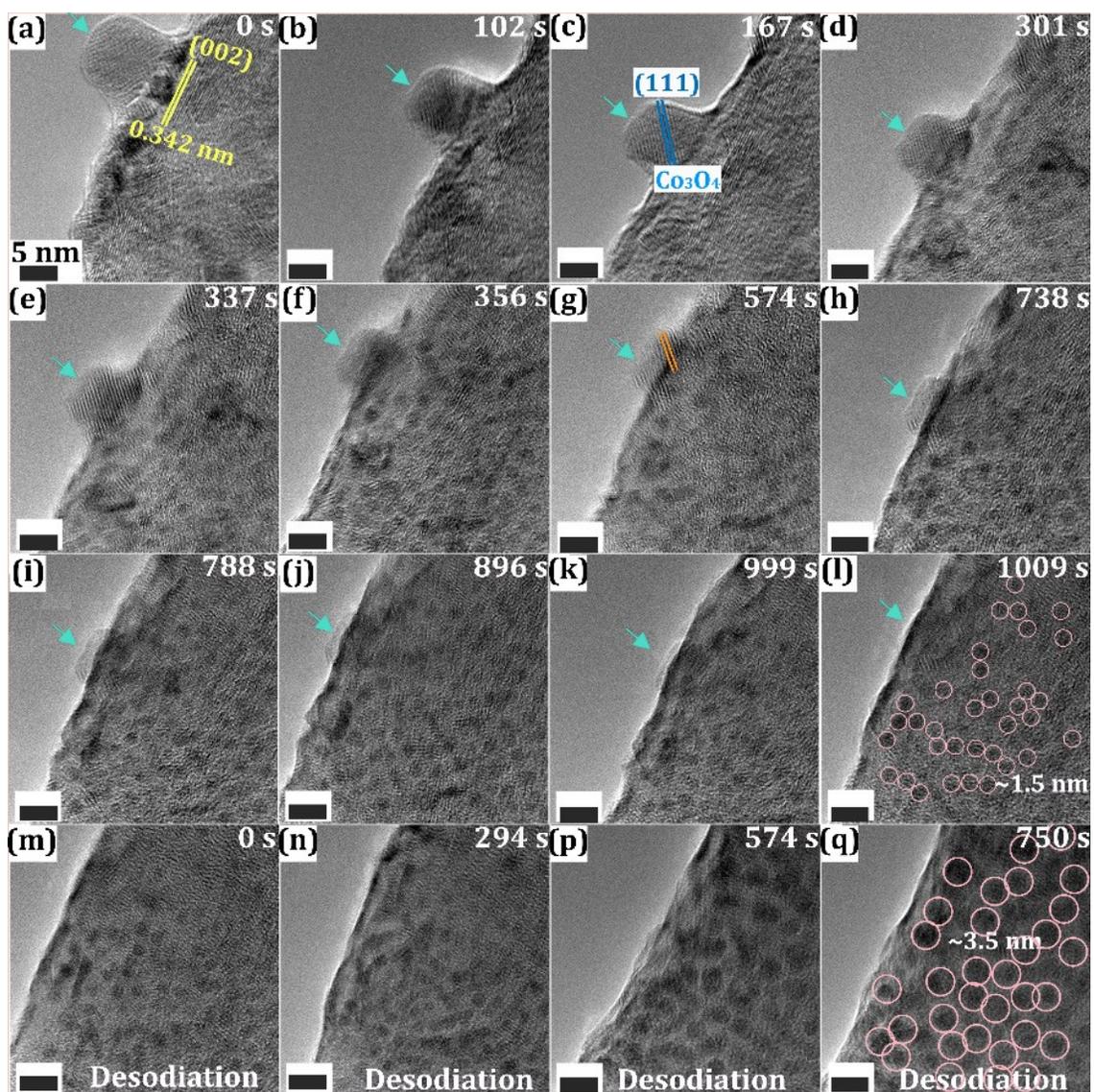
**Nanocomposite preparation.**  $\text{Co}_3\text{O}_4/\text{CNT}$  nanocomposites were synthesized by a simple and convenient hydrothermal method according to our group prior publication.<sup>1, 2</sup> In a typical experimental procedure, 200 mg cobalt (III) acetate tetrahydrated and 15 mg carbon nanotube were dissolved in 20 mL mixed solution of dimethylformamide (DMF) and water with volume ratio 8:2 by magnetic stirring for 30 min. Then the solution was transferred to a Teflon-lined autoclave of 25 mL capacity. The stainless steel autoclave was sealed and maintained at 120 °C for 2 h. After the autoclave was subsequently cooled to ambient temperature, newly produced precipitates were collected by filtration after being washed with ethanol and deionized water repeatedly, and dried in an oven at 60 °C.

***In situ* TEM experiments.** A nanobattery system was built inside TEM by using the nanofactory STM-TEM holder, which could provide a positive/negative bias between two electrodes in the circuit.  $\text{Co}_3\text{O}_4/\text{CNT}$  were adhered to one end of Au rod by using the conductive epoxy, acted as the working electrode. Sodium scratched directly by W rod was used as the counter electrode. Two electrodes of the nanobattery could contact each other by controlling precisely the pizo-motor with movement resolution by ~1nm inside TEM. The bias range of this holder is from -10 V to +10 V, which is enough for driven sodium ions moving from sodium source to react with active material in the nanobattery open cell. During the sodiation process, a negative bias -2.5 V was applied and a positive +3-4 V was applied to the circuit on desodiation. The electron dosage is carefully monitored in the experiments to minimize its effects on the current observations. In a controlled experiment, a similar reaction mechanism has been observed when the electron dose rate is  $5 \times 10^5 \text{ e}^-/\text{nm}^2$ .

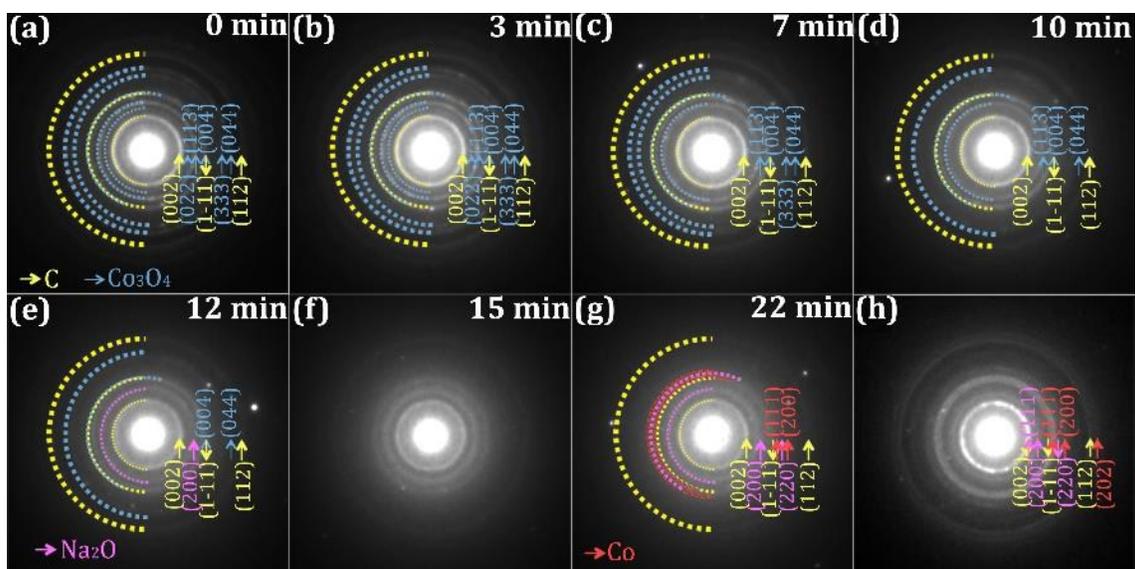
## 2. Supplementary Figures



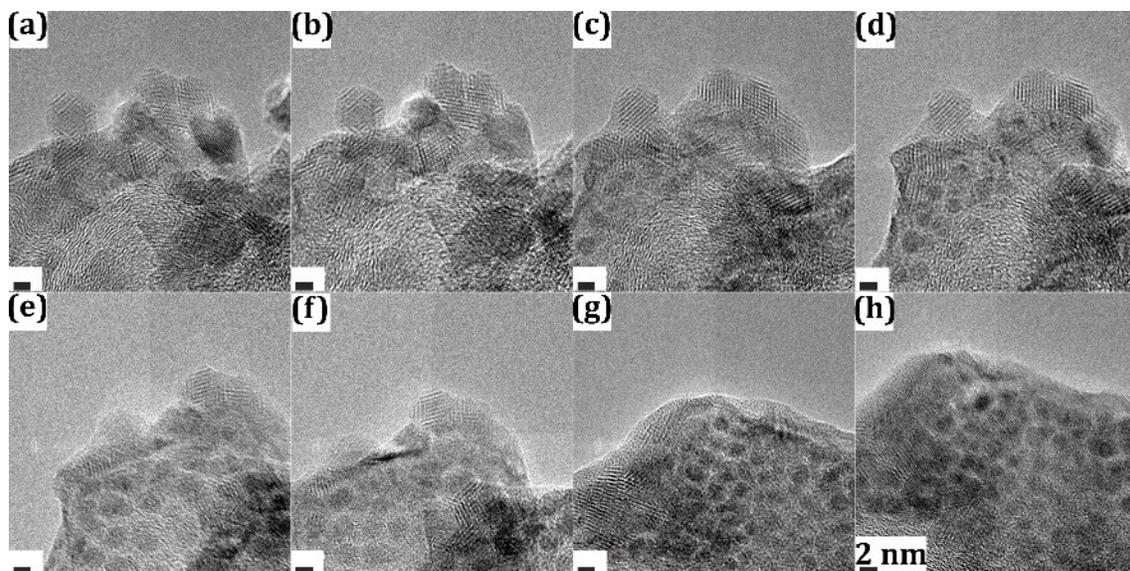
**Figure S1.** The first stage of sodiation process with Na<sub>2</sub>O layer coating on both surfaces of Co<sub>3</sub>O<sub>4</sub> nanoparticles and CNT. (a-b) The morphology before and after Na<sub>2</sub>O layer coverage, accompanying the diameter of CNT increased from 9 nm to 10.5 nm. (c-d) The electron diffraction patterns of the Co<sub>3</sub>O<sub>4</sub>/CNT composite before and after Na<sub>2</sub>O coating. (e) Schematic diagram of the nanobattery built inside the TEM along with the tip of the Nanofactory in-situ TEM holder.



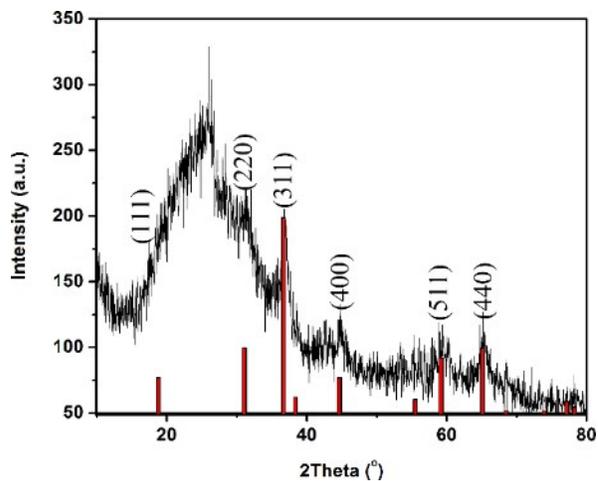
**Figure S2. The propagation of the electrochemical conversion reaction of one single  $\text{Co}_3\text{O}_4$  NP.** (a) A HRTEM image of one  $\text{Co}_3\text{O}_4$  NP and carbon nanotube, the d-spacing of 0.342 nm corresponded to the {002} lattice plane of carbon nanotube. (b-l) The evolutions of the  $\text{Co}_3\text{O}_4$  nanoparticle in size and morphology along with the reaction with Na-ions in the first sodiation. (m-q) The morphology and structure evolution in the first desodiation.



**Figure S3. Microstructure evolution of  $\text{Co}_3\text{O}_4/\text{CNT}$  in the first sodiation process.** (a) The diffraction pattern of pristine  $\text{Co}_3\text{O}_4/\text{CNT}$  nanocomposite. (b-e) Several diffraction rings of spinel  $\text{Co}_3\text{O}_4$  (blue semicircle) gradually disappeared along with sodiation. (f-g)  $\text{Na}_2\text{O}$  diffraction rings and a diffused ring corresponding to Co gradually appeared in the late stage of sodiation. (h) The diffraction pattern was taken from the region with more materials and giving a stronger diffraction signal.



**Figure S4.** (a-h) The corresponding morphology evolution during the first sodiation process.



**Figure S5.** X-ray powder diffraction pattern of the Co<sub>3</sub>O<sub>4</sub>/CNT composites. All the diffraction peaks labelled can be indexed by the Co<sub>3</sub>O<sub>4</sub> spinel phase, implying the high purity of the composite.

### 3. Supplementary Movies

**Movie 1** Atomic resolution observation of electrochemical sodiation of single  $\text{Co}_3\text{O}_4$  nanocube grown on carbon nanotube under bias of  $-2.5\text{V}$ .

**Movie 2** Atomic resolution observation of electrochemical desodiation of  $\text{Co}_3\text{O}_4$  nanocubes on carbon nanotube under bias of  $+4\text{V}$ .

**Movie 3** Atomic resolution observation of electrochemical sodiation of  $\text{Co}_3\text{O}_4$  nanocube clusters grown on carbon nanotube under bias of  $-2.5\text{V}$ .

## Reference

1. J. Xu, J. Wu, L. Luo, X. Chen, H. Qin, V. Dravid, S. Mi and C. Jia, *J. Power Sources*, 2015, **274**, 816-822.
2. J. Cheng, L. Liu, J. Zhang, F. Liu and X. Zhang, *Journal of Electroanalytical Chemistry*, 2014, **722**, 23-31.