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Supplementary Information

Synthesis of water-soluble fluorescent carbon dots from Setcreasea purpurea boom and its application for Br₂ detection

Haoying Zhaia, Baozhan Zhengb, Feng Yangb, Meina Wangb, Dan Xiaob,c,*

^a College of Chemistry and Chemical Engineering, Neijiang Normal University, Neijiang, Sichuan 641112, China.

E-mail: xiaodan@scu.edu.cn (Dan Xiao)

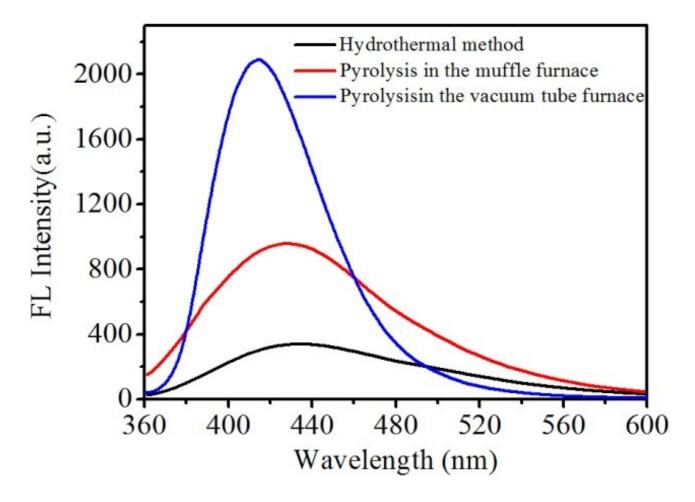


Fig. S1 The fluorescent spectra of CDs prepared with different mentods, including hydrothermal method (balck line), pyrolysis in the muffle furnace (red line) and pyrolysis in the vacuum tube furnace (blue line).

^b College of Chemistry, Sichuan University, 29 Wangjiang Road, Chengdu 610064, China.

^c College of Chemical Engineering, Sichuan University, 29 Wangjiang Road, Chengdu 610064, China.

^{*}Corresponding author:

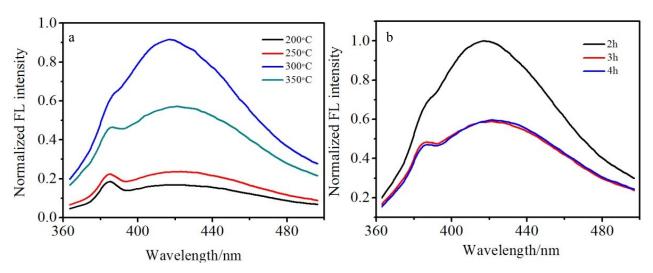


Fig. S2 Fluorescence spectra of CDs prepared under different temperature (a) and reaction time (b).

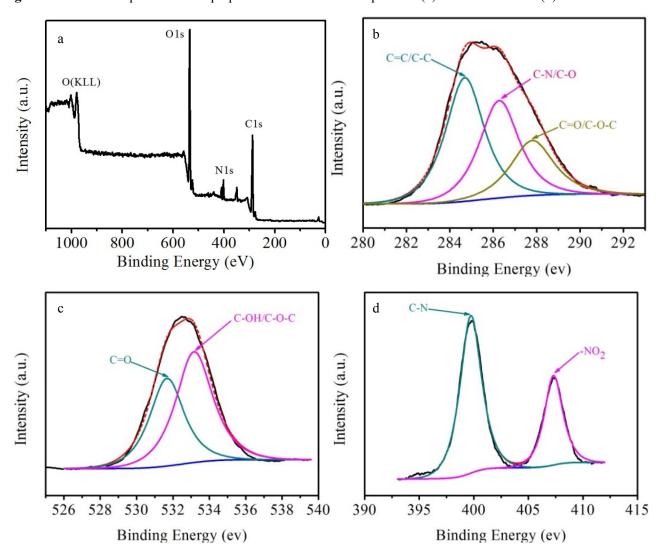


Fig. S3 XPS spectra of CDs (a), C1s (b), O1s (c) and N1s spectra (d).

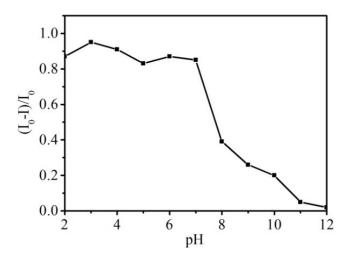


Fig. S4 Effects of pH values on the fluorescence quenching efficiency of CDs.

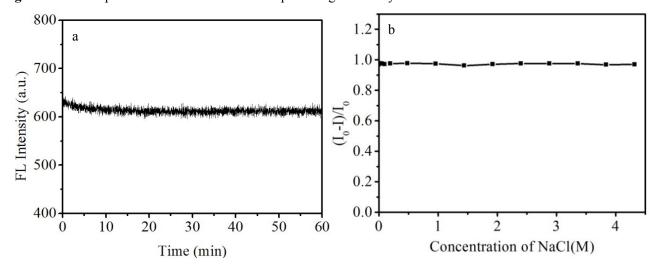


Fig. S5 (a) Photostability of CDs under the irradiation of 340 nm; (b) stability in different ionic strengths of NaCl.

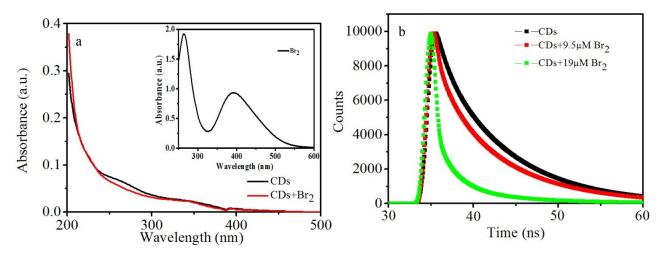


Fig. S6 UV-vis absorption spectra (a) (The inset shows the ultraviolet absorption of Br_2) and fluorescence lifetime (b) of CDs in the absence and presence of Br_2 .