

## Bimetallic Gold(I) Complexes of Photoswitchable Phosphines: Synthesis and Uses in Cooperative Catalysis

Tanzeel Arif,<sup>a,b</sup> Clément Cazorla,<sup>a,b</sup> Nicolas Bogliotti,<sup>b</sup> Nidal Saleh,<sup>a</sup> Florent Blanchard,<sup>a</sup> Vincent Gandon,<sup>a,c</sup> Rémi Métivier,<sup>b</sup> Juan Xie,<sup>b</sup> Arnaud Voituriez,<sup>\*a</sup> Angela Marinetti<sup>\*a</sup>

<sup>a</sup> Institut de Chimie des Substances Naturelles, CNRS UPR 2301, Univ. Paris-Sud, Université Paris-Saclay, 1, av. de la Terrasse, 91198 Gif-sur-Yvette, France.

<sup>b</sup> PPSM, ENS Paris-Saclay, CNRS, Université Paris-Saclay, Cachan, 94235 France.

<sup>c</sup> Institut de Chimie Moléculaire et des Matériaux d'Orsay, CNRS UMR 8182, Univ. Paris-Sud, Université Paris-Saclay, bâtiment 420, 91405 Orsay cedex, France.

### Table of Contents:

<b>Part I:</b>	<b>General information</b>	<b>SI-2</b>
<b>Part II:</b>	<b>Experimental Procedures</b>	<b>SI-3</b>
<b>Part III:</b>	<b>NMR Spectra (<sup>1</sup>H, <sup>13</sup>C, <sup>19</sup>F and <sup>31</sup>P NMR)</b>	<b>SI-8</b>
<b>Part IV:</b>	<b>Photochromic Behaviour of complexes 3 and 6</b>	<b>SI-27</b>
<b>Part V:</b>	<b>Hydroamination of <i>N</i>-alkenyl ureas</b>	<b>SI-31</b>
<b>Part VI:</b>	<b>[4+2] cycloadditions of 1,6-ene substrate</b>	<b>SI-39</b>
<b>Part VII:</b>	<b>X-Ray crystal structure determinations</b>	<b>SI-40</b>

## **I. General information**

All reactions were run under an inert atmosphere (argon), by using standard techniques for manipulating air-sensitive compounds. Anhydrous solvents were obtained by filtration through drying columns (THF, Et<sub>2</sub>O, CH<sub>2</sub>Cl<sub>2</sub>). All reagents and solvents were of commercial quality and were used without further purification. Analytical thin-layer chromatography (TLC) was performed on plates precoated with silica gel. The developed chromatograms were visualized by UV absorbance and by immersion in phosphomolybdic acid or KMnO<sub>4</sub> solutions, followed by heating. Flash column chromatography was performed using 40-63 mesh silica gel. Purifications have been performed on a automated flash chromatography system (SiO<sub>2</sub>), unless otherwise stated. NMR spectra (<sup>1</sup>H, <sup>13</sup>C, <sup>31</sup>P) were recorded on either AV 500 or AV 300 spectrometers. NMR spectra were recorded at room temperature. Chemical shifts are reported in parts per million relative to an internal standard of residual chloroform ( $\delta$  = 7.27 ppm for <sup>1</sup>H NMR and 77.00 ppm for <sup>13</sup>C NMR). *J* couplings are reported in Hz. The <sup>13</sup>C NMR spectra were assigned by standard methods using HSQC or DEPT experiments.

Substrates **4**<sup>1</sup> and **7**<sup>2</sup> have been synthesized according to the reported procedures.

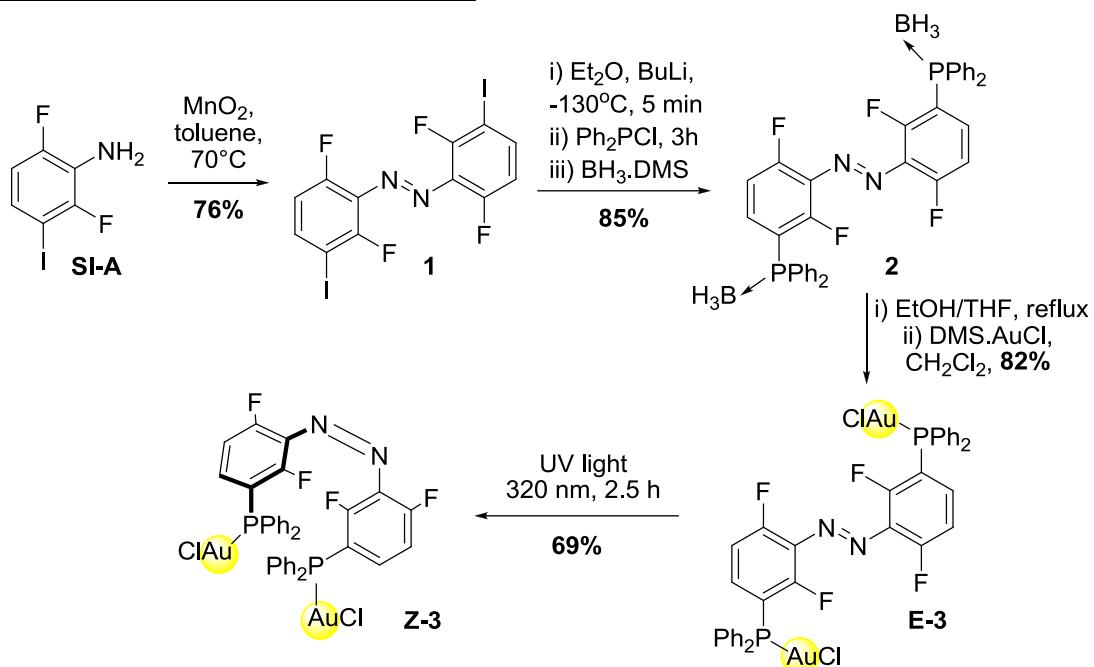
---

<sup>1</sup> Kojima, M.; Mikami, K. *Synlett* **2012**, 57.

<sup>2</sup> Nieto-Oberhuber, C.; López, S.; Echavarren, A. M. *J. Am. Chem. Soc.* **2005**, 127, 6178.

## II. Experimental Procedures

### A. Synthesis of complexes E-3 and Z-3:



#### II.1. 2,6-difluoro-3-iodoaniline (SI-A).

To a solution of 2,6-difluoroacetanilide<sup>3</sup> (210 mg, 1.23 mmol) in *conc.*  $\text{H}_2\text{SO}_4$  (12.3 mL) was added *N*-iodosuccinimide (277 mg, 1.23 mmol) portion-wise at room temperature. The resulting mixture was stirred over night at room temperature then poured into crushed ice (200 g). The aqueous layer was extracted with EtOAc (3x30 mL), washed with  $\text{H}_2\text{O}$  (100 mL), washed with *sat. aq.*  $\text{NaHCO}_3$  (100 mL), dried ( $\text{MgSO}_4$ ), filtered, evaporated to give the title compound (351 mg, 96%) as white solid. Mp 62–64°C;  $R_f$  0.42 (40% EtOAc/hexane); <sup>1</sup>H NMR (300 MHz,  $\text{CDCl}_3$ ) 7.60 (1H, dt,  $J_1$  9,  $J_2$  6,  $\text{CH}_{\text{Ar}}$ ), 7.28 (1H, br. s.  $\text{NH}$ ), 6.78 (1H, td,  $J_1$  9,  $J_2$  1,  $\text{CH}_{\text{Ar}}$ ), 2.20 (3H, s,  $\text{CH}_3$ ); <sup>13</sup>C NMR (75 MHz,  $\text{CDCl}_3$ ) 160.0 ( $\text{C}=\text{O}$ ), 158.6 (Cq), 158.4 (Cq), 143.8 (Cq), 136.4 ( $\text{CH}_{\text{Ar}}$ ), 113.4 (d,  $J$  21,  $\text{CH}_{\text{Ar}}$ ), 77.2 (Cl), 23.0 ( $\text{CH}_3$ ); <sup>19</sup>F NMR (282 MHz) -96.1 (s), -116.6 (s); HRMS (ESI) calcd. for  $\text{C}_8\text{H}_7\text{F}_2\text{INO}$  [ $\text{M}+\text{H}]^+$ : 297.9540, found: 297.9515.

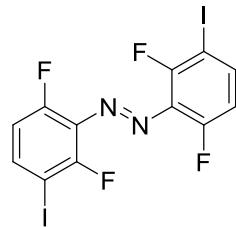
To a solution of *N*-(2,6-difluoro-3-iodophenyl)acetamide (351 mg, 1.18 mmol) in MeOH (1.5 mL) was added *conc.*  $\text{HCl}$  (1 mL). The reaction mixture was stirred at 60 °C for 12 h then allowed to reach room temperature, diluted with  $\text{H}_2\text{O}$  (50 mL), neutralized with solid  $\text{NaHCO}_3$ , extracted with EtOAc (3x50 mL), dried, evaporated and purified by chromatography to give the compound **SI-A** (290 mg, 96%) as white solid.

$R_f$  0.50 (20% EtOAc/hexane); <sup>1</sup>H NMR (300 MHz,  $\text{CDCl}_3$ ) 7.03 (1H, ddd,  $J_1$  9,  $J_2$  7,  $J_3$  6,  $\text{CH}_{\text{Ar}}$ ), 6.65 (1H, ddd,  $J_1$  10,  $J_2$  9,  $J_3$  2,  $\text{CH}_{\text{Ar}}$ ), 3.83 (2H, br.s.  $\text{NH}_2$ ); <sup>13</sup>C NMR (75 MHz,  $\text{CDCl}_3$ ) 153.1 (dd,  $J_1$  7,  $J_2$  93, CqF), 150.0 (dd,  $J_1$  6,  $J_2$  90, CqF), 132.3 (Cq), 125.1 (d,  $J$  8,  $\text{CH}_{\text{Ar}}$ ), 112.5 (d,  $J$  19,  $\text{CH}_{\text{Ar}}$ ), 74.6 (d,  $J$  20, CqI); <sup>19</sup>F NMR (282 MHz) -111.6 (s), -132.1 (s); HRMS (ESI) calcd. for  $\text{C}_6\text{H}_5\text{F}_2\text{IN}$  [ $\text{M}+\text{H}]^+$ : 255.9435, found: 255.9435.

<sup>3</sup> Finger, G. S.; Reed, F. H.; Finnerty, J. L. *J Am. Chem. Soc.* **1951**, 73, 153-155.

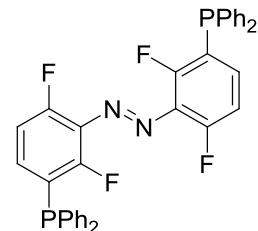
## II.2. (*E*)-1,2-bis(2,6-difluoro-3-iodophenyl)diazene (**1**).

To a solution of aniline **SI-A** (1.61 g, 6.3 mmol) in toluene (120 mL) was added MnO<sub>2</sub> (13.75 g, 158 mmol). The reaction mixture was stirred at 70°C for 48 h then allowed to reach room temperature, filtered through a pad of silica, evaporated and purified by chromatography to give the title compound (1.21 g, 76%, >95/5 : *E/Z*) as orange red solid; *R*<sub>f</sub> 0.46 (20% EtOAc/hexane). Mp 124–126°C; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>) 7.85–7.78 (2H, m, CH<sub>Ar</sub>), 7.00–6.90 (2H, m, CH<sub>Ar</sub>); *Z* isomer: 7.62 (2H, m, CH<sub>Ar</sub>), 6.73 (2H, m, CH<sub>Ar</sub>); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>) 156.0 (d, *J* 264, CqF), 154.0 (dd, *J*<sub>1</sub> 4, *J*<sub>2</sub> 262, CqF), 140.1 (d, *J* 7, CH<sub>Ar</sub>), 131.7 (t, *J* 12, CqN), 114.4 (d, *J* 20, CH<sub>Ar</sub>), 76.6 (dd, *J*<sub>1</sub> 5, *J*<sub>2</sub> 25, CqI); <sup>19</sup>F NMR (282 MHz) 140.3 (dd, *J*<sub>1</sub> 10, *J*<sub>2</sub> 3), 114.5 (dd, *J*<sub>1</sub> 4, *J*<sub>2</sub> 21); HRMS (ESI) calcd. for C<sub>12</sub>H<sub>5</sub>F<sub>4</sub>I<sub>2</sub>N<sub>2</sub> [M+H]<sup>+</sup>: 506.8478, found: 506.8491.



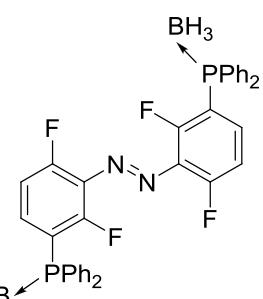
## II.3. (*E*)-1,2-bis(3-(diphenylphosphino)-2,6-difluorophenyl)diazene.

A stirred solution of diazene **1** (30 mg, 0.06 mmol) in Et<sub>2</sub>O (3 mL) was cooled to -130°C (pentane/liquid N<sub>2</sub>) and *n*-BuLi (74 μL, 0.119 mmol, 1.6 M in Hexanes) was added rapidly. To the resulting dark red reaction mixture was added after 5 min, a solution of distilled Ph<sub>2</sub>PCl (24 μL, 0.132 mmol) in Et<sub>2</sub>O (1 mL). After 10 min the cooling bath was removed and the reaction mixture was stirred at room temperature for 3 h; quenched with H<sub>2</sub>O (3 drops), dried (MgSO<sub>4</sub>), evaporated and purified by chromatography to give the desired product (31 mg, 80%, 95/5 : *E/Z*) as orange solid. Mp 168–170°C; *R*<sub>f</sub> 0.20 (5% EtOAc/pentane); <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>) 7.45–7.31 (20H, m, CH<sub>Ar</sub>), 6.99 (2H, t, *J* 9, CH<sub>Ar</sub>), 6.92–6.83 (2H, m, CH<sub>Ar</sub>); [Discernable data for *Z* isomer: 7.19–7.14 (2H, m, CH<sub>Ar</sub>), 7.10–7.04 (2H, m, CH<sub>Ar</sub>)]; <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>): 157.0 (ddd, *J*<sub>1</sub> 5, *J*<sub>2</sub> 16, *J*<sub>3</sub> 261, CqF), 155.7 (dd, *J*<sub>1</sub> 4, *J*<sub>2</sub> 264, CqF), 136.0–135.7 (m, CH<sub>Ar</sub>), 134.7 (d, *J* 11, CqP), 133.8 (d, *J* 21, CH<sub>Ar</sub>), 131.8–131.4 (m, CqP), 129.3 (s, CH<sub>Ar</sub>), 128.8 (d, *J* 7, CH<sub>Ar</sub>), 122.8–122.2 (m, N-C<sub>q</sub>), 113.0–112.5 (m, CH<sub>Ar</sub>); <sup>31</sup>P NMR (121.5 MHz): -18.9 (d, *J* 59); <sup>19</sup>F NMR (282 MHz): -110.5 (d, *J* 59), -120.0 (s); [Discernable data for *Z* isomer: -110.8 (d, *J* 58), -120.3 (s.)]; HRMS (ESI) calcd. for C<sub>36</sub>H<sub>25</sub>F<sub>4</sub>N<sub>2</sub>P<sub>2</sub> [M+H]<sup>+</sup>: 623.1429, found: 623.1444.



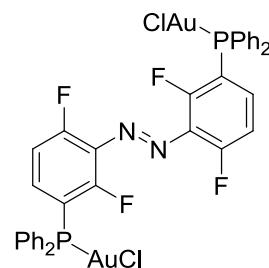
## II.4. (*E*)-1,2-bis(3-(diphenylphosphino)-2,6-difluorophenyl)diazene (2).

A stirred solution of diazene **1** (30 mg, 0.06 mmol) in Et<sub>2</sub>O (3 mL) was cooled to -130°C (pentane/liquid N<sub>2</sub>) and *n*-BuLi (74 μL, 0.119 mmol, 1.6 M in Hexanes) was added rapidly. To the resulting dark red reaction mixture was added after 5 min, a solution of distilled Ph<sub>2</sub>PCl (24 μL, 0.132 mmol) in Et<sub>2</sub>O (1 mL). After 10 min the cooling bath was removed and the reaction mixture was stirred at room temperature for 3 h; then re-cooled to 0 °C and BH<sub>3</sub>.Me<sub>2</sub>S (75 μL, 0.15 mmol, 2M in THF) was added and stirred at room temperature for 3 h. The excess borane was quenched with H<sub>2</sub>O (3 drops), dried (MgSO<sub>4</sub>), evaporated and purified by chromatography to give the desired complex (33 mg, 85%) as orange solid. *R*<sub>f</sub> 0.26 (20% EtOAc/hexane); <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>) 7.88–7.78 (2H, m, CH<sub>Ar</sub>), 7.70–7.64 (6H, m, CH<sub>Ar</sub>), 7.57–7.30 (14H, m, CH<sub>Ar</sub>), 7.17 (2H, t, *J* 9, CH<sub>Ar</sub>), 2.10–0.30 (6H, br. s. BH<sub>3</sub>); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>) 157.7 (dd, *J*<sub>1</sub> 5, *J*<sub>2</sub> 267, CqF), 157.1 (dd, *J* 267, CqF), 138.1–137.7 (m, CqP), 136.9–136.6 (m, CqP), 132.8 (d, *J* 10, CH<sub>Ar</sub>), 132.5–132.1 (m, CH<sub>Ar</sub>), 131.7 (d, *J* 2, CH<sub>Ar</sub>), 129.0 (d, *J* 11, CH<sub>Ar</sub>), 127.7–126.4 (m, CqN), 113.6 (m, CH<sub>Ar</sub>); <sup>31</sup>P NMR (121.5 MHz) 18.9 (m); <sup>19</sup>F NMR (282 MHz) -103.9 (dd, *J*<sub>1</sub> 7, *J*<sub>2</sub> 4), -114.2 (d, *J* 7); HRMS (ESI) calcd. for C<sub>36</sub>H<sub>30</sub>B<sub>2</sub>F<sub>4</sub>N<sub>2</sub>NaP<sub>2</sub> [M+Na]<sup>+</sup>: 673.1904, found: 673.1915.



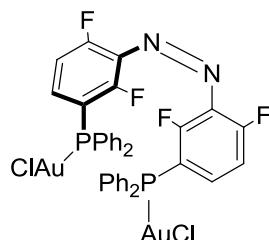
## II.5. (*E*)-1,2-bis(3-(diphenylphosphino)-2,6-difluorophenyl)diazene (*E*-3).

To a stirred solution of phosphine-borane **2** (30 mg, 0.046 mmol) in THF (5 mL) was added EtOH (10 mL). The resulting solution was degassed and refluxed for 3 h under argon in dark. The reaction was monitored by  $^{31}\text{P}$  NMR. After complete deprotection, the solvents and  $(\text{EtO})_3\text{B}$  were removed under high vacuum at 40°C. The crude diphosphine was dissolved in  $\text{CH}_2\text{Cl}_2$  (5 mL) and  $\text{AuCl} \cdot \text{Me}_2\text{S}$  (28 mg, 0.094 mmol, 2.05 equiv.) was added under argon, stirred at rt for 1 h, then solvent was removed and purified by chromatography to give desired gold complex (41 mg, 82%) as orange solid. Mp >250°C;  $R_f$  0.37 (70%  $\text{CH}_2\text{Cl}_2$ /hexane);  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ ) 7.66-7.58 (10H, m,  $\text{CH}_{\text{Ar}}$ ), 7.56-7.46 (10H, m,  $\text{CH}_{\text{Ar}}$ ), 7.32-7.25 (2H, m,  $\text{CH}_{\text{Ar}}$ ), 7.14 (2H, t,  $J$  9,  $\text{CH}_{\text{Ar}}$ );  $^{13}\text{C}$  NMR (75 MHz,  $\text{CDCl}_3$ ) 157.7 (d,  $J$  269, CqF), 156.9 (d,  $J$  265, CqF), 137.5-137.2 (m,  $\text{CH}_{\text{Ar}}$ ), 136.6-136.4 (m, CqP), 134.0 (d,  $J$  15,  $\text{CH}_{\text{Ar}}$ ), 132.8-132.5 (m,  $\text{CH}_{\text{Ar}}$ ), 132.6 (s,  $\text{CH}_{\text{Ar}}$ ), 129.6 (d,  $J$  12,  $\text{CH}_{\text{Ar}}$ ), 126.6 (d,  $J$  66, CqN), 113.9-113.5 (m,  $\text{CH}_{\text{Ar}}$ ). One CqP is missing;  $^{31}\text{P}$  NMR (121.5 MHz): 25.6 (d,  $J$  20);  $^{19}\text{F}$  NMR (282 MHz) -103.9 (dd,  $J_1$  7,  $J_2$  21), -112.5 (d,  $J$  7); HRMS (ESI) calcd. for  $\text{C}_{36}\text{H}_{24}\text{Au}_2\text{ClF}_4\text{N}_2\text{P}_2$  [M-Cl] $^+$ : 1051.0371, found: 1051.0452.

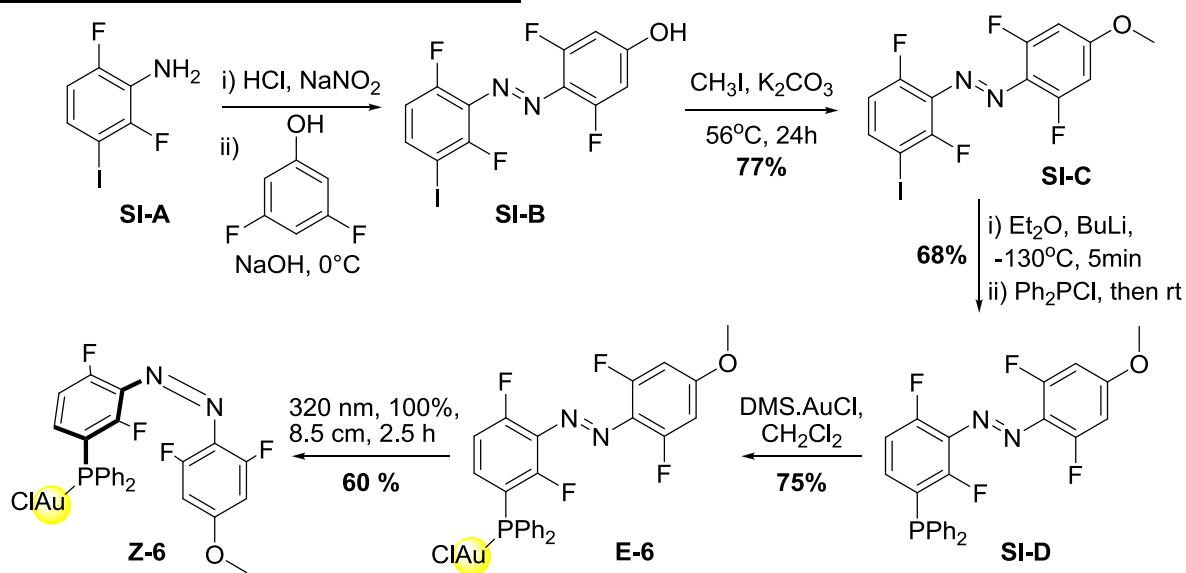


## II.6. (*Z*)-1,2-bis(3-(diphenylphosphino)-2,6-difluorophenyl)diazene (*Z*-3).

An orange solution of *E*-gold complex **E-3** (50 mg, 0.046 mmol) in  $\text{CH}_2\text{Cl}_2$  (50 mL) was stirred at room temperature and irradiated with UV light (Semrock® optical filter: 320/40, intensity: 100%, distance: 8.5 cm, time: 2.5 h). The resulting yellow solution was evaporated in the dark at 30 °C and purified by chromatography in dark to give desired **Z-3** (34 mg, 69%) as yellow solid;  $R_f$  0.49 (50% EtOAc/hexane) and recovered **E-3** (11 mg, 22%) as light orange solid.  $R_f$  0.40 (50% EtOAc/hexane);  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ ) 7.69-7.44 (20H, m,  $\text{CH}_{\text{Ar}}$ ), 7.15-7.05 (4H, m,  $\text{CH}_{\text{Ar}}$ );  $^{13}\text{C}$  NMR (75 MHz,  $\text{CDCl}_3$ ) 156.9 (d,  $J$  262, CqF), 149.6 (d,  $J$  254, CqF), 137.5-137.3 (m, CqP), 136.4 (m,  $\text{CH}_{\text{Ar}}$ ), 134.0 (d,  $J$  14,  $\text{CH}_{\text{Ar}}$ ), 132.8 (d,  $J$  2,  $\text{CH}_{\text{Ar}}$ ), 129.6 (d,  $J$  13,  $\text{CH}_{\text{Ar}}$ ), 126.1 (d,  $J$  64, CqN), 114.0-113.6 (m,  $\text{CH}_{\text{Ar}}$ ). One CqP is missing;  $^{31}\text{P}$  NMR (121.5 MHz): 24.8 (s);  $^{19}\text{F}$  NMR (282 MHz) -103.5 (s), -111.5 (s).

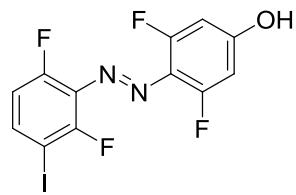


## B. Synthesis of complexes **E-6** and **Z-6**:



**II.7. (*E*)-4-((2,6-difluoro-3-iodophenyl)diazenyl)-3,5-difluorophenol (SI-B).**

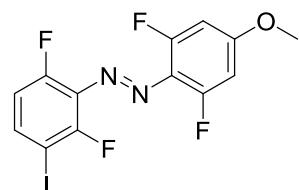
Finely powdered 2,6-difluoro-3-iodoaniline **SI-A** (255 mg, 1 mmol) was added to 3 mL of HCl (27%) at 0°C. A solution of NaNO<sub>2</sub> (83 mg, 1.2 mmol) in H<sub>2</sub>O (3 mL) was added dropwise and the temperature was kept below 5°C. The precipitate was dissolved by dropwise addition of HCl



(37%, 2 mL) and the resulting solution was stirred for 30 min at 0 °C. Excess NaNO<sub>2</sub> was quenched by addition of NH<sub>2</sub>SO<sub>3</sub>H (25 mg, 0.25 mmol). Meanwhile, in another flask, 3,5-difluorophenol (195 mg, 1.5 mmol) was dissolved in aqueous NaOH (prepared from 400 mg/5 mL of H<sub>2</sub>O). This solution was cooled to 0°C and a freshly prepared solution of the precooled (0°C) diazonium salt was added dropwise. During addition the pH was controlled to remain strongly alkaline. The reaction mixture was stirred for 1 h at room temperature. The pH was adjusted to 4 by addition of 2N CH<sub>3</sub>COOH. The precipitate was filtered, washed with water and dried under high vacuum to yield the desired azo compound **SI-B** as an orange red solid (324 mg; MS-EI: *m/z* 397 [M+H]<sup>+</sup>). The crude azo compound was used in next step without further purification. *Caution!* Solid diazonium salts are heat and shock sensitive towards explosion.

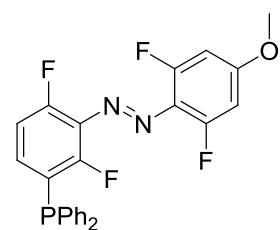
**II.8. (*E*-1-(2,6-difluoro-3-iodophenyl)-2-(2,6-difluoro-4-methoxyphenyl)diazene (SI-C).**

A solution of crude (*E*)-4-((2,6-difluoro-3-iodophenyl)diazenyl)-3,5-difluorophenol **SI-B** (324 mg, 0.82 mmol), methyl iodide (190 mg, 1.34 mmol), K<sub>2</sub>CO<sub>3</sub> (592 mg, 4.28 mmol), 18-crown-6 (55 mg, 0.21 mmol) in dry acetone (25 mL) was refluxed under argon for 24 h. After removal of solvent by evaporation, the residue was purified by chromatography to give the desired product **SI-C** (258 mg, 77%, >95/5 : *E/Z*) as orange solid. *R*<sub>f</sub> 0.52 (40% EtOAc/hexane); <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>) 7.75-7.66 (1H, m, *p*-CH<sub>Ar</sub>), 6.90 (1H, ddd, *J*<sub>1</sub> 10, *J*<sub>2</sub> 9, *J*<sub>3</sub> 2, *m*-CH<sub>Ar</sub>), 6.63-6.57 (2H, dd, *J*<sub>1</sub> 14, *J*<sub>2</sub> 3, CH<sub>Ar</sub>), 3.88 (3H, s, OCH<sub>3</sub>); [Discernable data for *Z* isomer: 7.85 (1H, dd, *J*<sub>1</sub> 9, *J*<sub>2</sub> 6, *p*-CH<sub>Ar</sub>), 7.02 (1H, dd, *J*<sub>1</sub> 10, *J*<sub>2</sub> 9, *m*-CH<sub>Ar</sub>), 6.50 (2H, dd, *J*<sub>1</sub> 14, *J*<sub>2</sub> 3, CH<sub>Ar</sub>), 3.84 (3H, s, OCH<sub>3</sub>)]; <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>) 163.1 (t, *J* 14, CqOCH<sub>3</sub>), 157.6 (dd, *J*<sub>1</sub> 9, *J*<sub>2</sub> 262, CqF), 156.6 (dd, *J*<sub>1</sub> 3, *J*<sub>2</sub> 135, CqF), 153.2 (dd, *J*<sub>1</sub> 4, *J*<sub>2</sub> 132, CqF), 138.6 (dd, *J*<sub>1</sub> 9, *J*<sub>2</sub> 3, CH<sub>Ar</sub>), 132.3 (t, *J* 12, CqN), 125.8 (t, *J* 9, CqN), 114.2 (dd, *J*<sub>1</sub> 4, *J*<sub>2</sub> 21, CHAr), 98.8 (dd, *J*<sub>1</sub> 3, *J*<sub>2</sub> 24, CH<sub>Ar</sub>), 76.2 (dd, *J*<sub>1</sub> 4, *J*<sub>2</sub> 26, CqI), 56.2 (s, OCH<sub>3</sub>); <sup>19</sup>F NMR (282 MHz) -100.2 (s), -116.4 (d, *J* 11), -121.6 (s); HRMS (ESI) calcd. for C<sub>13</sub>H<sub>8</sub>F<sub>4</sub>IN<sub>2</sub>O [M+H]<sup>+</sup>: 410.9617, found: 410.9609.



**II.9. (*E*-1-(2,6-difluoro-4-methoxyphenyl)-2-(3-(diphenylphosphino)-2,6-difluorophenyl)diazene (SI-D).**

A stirred solution of **SI-C** (220 mg, 0.54 mmol) in anhydrous Et<sub>2</sub>O (27 mL) was cooled to -130°C (pentane/liquid N<sub>2</sub>) and *n*-BuLi (370 μL, 0.59 mmol, 1.6 M in Hexanes) was added rapidly. To the resulting dark red reaction mixture was added after 5 min, a solution of freshly distilled Ph<sub>2</sub>PCl (116 μL, 0.65 mmol) in Et<sub>2</sub>O (5 mL). After 10 min the cooling bath was removed and the reaction mixture was stirred at room temperature for 3 h, evaporated and purified by chromatography to give the desired phosphine (171 mg, 68%, 95/5 : *E/Z*) as orange solid. *R*<sub>f</sub> 0.30 (10% EtOAc/hexane); <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>) 7.41-7.32 (10H, m, CH<sub>Ar</sub>), 7.00-6.93 (1H, m, *p*-CH<sub>Ar</sub>), 6.86-6.77 (1H, m, *m*-CH<sub>Ar</sub>), 6.69 (2H, dd, *J*<sub>1</sub> 3, *J*<sub>2</sub> 14, CH<sub>Ar</sub>), 3.88 (3H, s, OCH<sub>3</sub>); [Discernable data for *Z* isomer: 7.19 (1H, td, *J*<sub>1</sub> 8, *J*<sub>2</sub> 2, *p*-CH<sub>Ar</sub>), 6.37 (1H, dd, *J*<sub>1</sub> 2, *J*<sub>2</sub> 11, CH<sub>Ar</sub>), 3.80 (3H, s, OCH<sub>3</sub>)]; <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>) 162.8 (t, *J* 17, CqOCH<sub>3</sub>), 158.5 (dd, *J*<sub>1</sub> 8, *J*<sub>2</sub> 262, CqF), 158.6 (m, CqP), 155.7 (dd, *J*<sub>1</sub> 5, *J*<sub>2</sub> 262, CqF), 134.8 (d, *J* 11, CH<sub>Ar</sub>), 134.7-134.5 (m, CH<sub>Ar</sub>), 133.8 (d, *J* 21, CH<sub>Ar</sub>), 132.5-132.4 (m, CqP), 131.8 (t, *J* 8, CqN), 129.3 (s, CH<sub>Ar</sub>), 128.8 (d, *J* 7, CH<sub>Ar</sub>), 122.1 (t, *J* 20, CqN), 112.8 (dd, *J*<sub>1</sub> 3, *J*<sub>2</sub> 20, CH<sub>Ar</sub>), 98.9 (dd, *J*<sub>1</sub> 3, *J*<sub>2</sub> 24, CH<sub>Ar</sub>), 56.2 (OCH<sub>3</sub>);

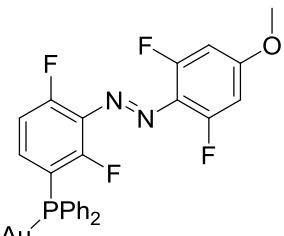


<sup>31</sup>P NMR (121.5 MHz) : -19.0 (d, *J* 62); <sup>19</sup>F NMR (282 MHz): -111.6 (d, *J* 59), -116.8 (d, *J* 12), -121.0 (s); HRMS (ESI) calcd. for C<sub>25</sub>H<sub>18</sub>F<sub>4</sub>N<sub>2</sub>OP [M+H]<sup>+</sup>: 469.1093, found: 469.1086.

## II.10.

### (*E*)-1-(2,6-difluoro-4-methoxyphenyl)-2-(3-

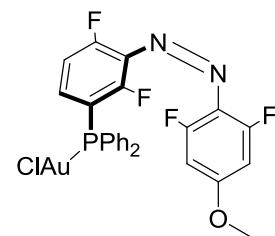
**(diphenylphosphino)- 2,6-difluorophenyl diazene (*E*-6).** To a stirred solution of phosphine **SI-D** in CH<sub>2</sub>Cl<sub>2</sub> (5 mL) and AuCl.Me<sub>2</sub>S (28 mg, 0.094 mmol, 2.05 equiv.) was added at 0 °C under argon. The resulting solution stirred at rt for 1 h, then solvent was removed and purified by chromatography to give desired gold complex (47 mg, 75%, >95/5 : *E/Z*) as orange-red solid. Mp 90-92°C; *R*<sub>f</sub> 0.54 (50% EtOAc/hexane); <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>): 7.69-7.45 (10H, m, CH<sub>Ar</sub>), 7.32-7.19 (1H, m, CH<sub>Ar</sub>), 7.12 (1H, t, *J* 8, CH<sub>Ar</sub>), 6.59 (2H, dd, *J*<sub>1</sub> 3, *J*<sub>2</sub> 14, CH<sub>Ar</sub>), 3.89 (3H, s, OCH<sub>3</sub>); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>): 163.5 (t, *J* 15, CqOCH<sub>3</sub>), 160.0 (dd, *J*<sub>1</sub> 14, *J*<sub>2</sub> 230, CqF), 157.7 (dd, *J*<sub>1</sub> 7, *J*<sub>2</sub> 264, CqF), 158.7-158.5 (m, CqP), 152.7 (dd, *J*<sub>1</sub> 8, *J*<sub>2</sub> 253, CqF), 135.7 (td, *J*<sub>1</sub> 11, *J*<sub>2</sub> 5, CH<sub>Ar</sub>), 134.0 (d, *J* 15, CH<sub>Ar</sub>), 132.4 (s, CH<sub>Ar</sub>), 129.5 (d, *J* 13, CH<sub>Ar</sub>), 127.4 (s, CqN), 126.5 (s, CqN), 113.8-113.3 (m, CH<sub>Ar</sub>), 99.2-98.5 (m, CH<sub>Ar</sub>), 56.3 (s, OCH<sub>3</sub>); <sup>31</sup>P NMR (121.5 MHz): 26.4 (s); <sup>19</sup>F NMR (282 MHz): -105.7 (m), -114.4 (d, *J* 7), -115.7 (d, *J* 11); HRMS (ESI) calcd. for C<sub>27</sub>H<sub>20</sub>AuF<sub>4</sub>N<sub>3</sub>OP [M-Cl+CH<sub>3</sub>CN]<sup>+</sup>: 706.0946, found: 706.0938.



## II.11.

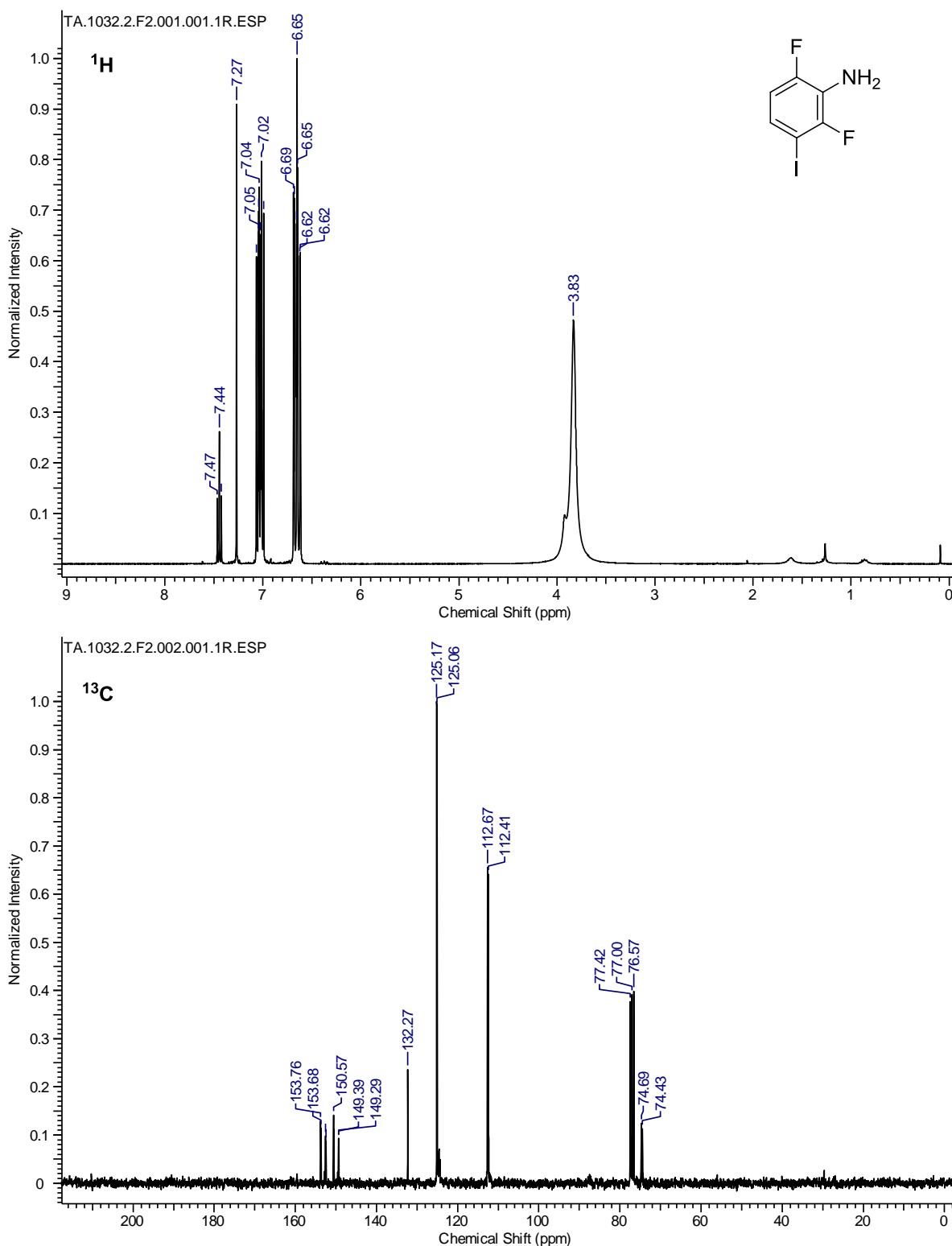
### (*Z*)-1-(2,6-difluoro-4-methoxyphenyl)-2-(3-(diphenylphosphino)-

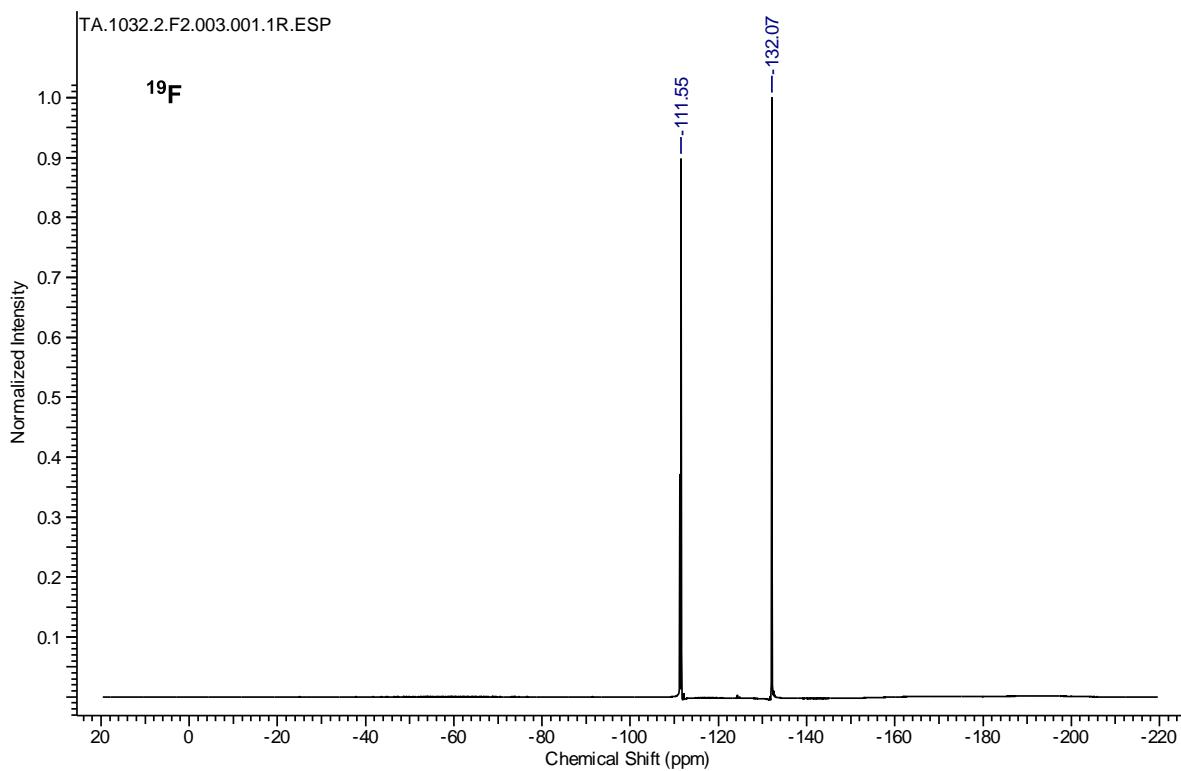
**2,6-difluorophenyl diazene (*Z*-6).** An orange solution of *E*-6 (25 mg, 0.035 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (50 mL) was stirred at room temperature and irradiated with UV light (*Semrock*<sup>®</sup> optical filter: 320/40, intensity: 100%, distance: 8.5 cm, time: 2 h). The resulting solution yellow solution was evaporated in the dark at 30°C and purified by HPLC to give *Z*-6 (15 mg, 60%, >95/5 : *Z/E*) as yellow solid and recovered *E*-6 (9 mg, 35%) as light orange solid. <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>) 7.58-7.53 (2H, m, CH<sub>Ar</sub>), 7.51-7.44 (8H, m, CH<sub>Ar</sub>), 7.23 (1H, dt, *J*<sub>1</sub> 14, *J*<sub>2</sub> 7, CH<sub>Ar</sub>), 7.04 (1H, t, *J* 9, CH<sub>Ar</sub>), 6.40 (2H, d, *J* 10, CH<sub>Ar</sub>), 3.85 (3H, s, CH<sub>3</sub>O); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>) 161.8-161.5 (m, CqP), 156.9 (dd, *J*<sub>1</sub> 262, *J*<sub>2</sub> 7, CqF), 152.8 (dd, *J*<sub>1</sub> 253, *J*<sub>2</sub> 8, CqF), 150.8 (d, *J* 263, CqF), 135.6-135.3 (m, CH<sub>Ar</sub>), 133.8 (d, *J* 14, CH<sub>Ar</sub>), 132.4 (d, *J* 2, CH<sub>Ar</sub>), 129.4 (d, *J* 13, CH<sub>Ar</sub>), 127.0 (CqN), 126.2 (CqN), 113.3 (dd, *J*<sub>1</sub> 17, *J*<sub>2</sub> 12, CH<sub>Ar</sub>), 98.7 (dd, *J*<sub>1</sub> 24, *J*<sub>2</sub> 3, CH<sub>Ar</sub>), 56.2 (s, CH<sub>3</sub>O). One CqP is missing; <sup>31</sup>P NMR (121.5 MHz) 27.0 (s); <sup>19</sup>F NMR (282 MHz): -103.3 (s), -111.5 (s), -117.0 (s).



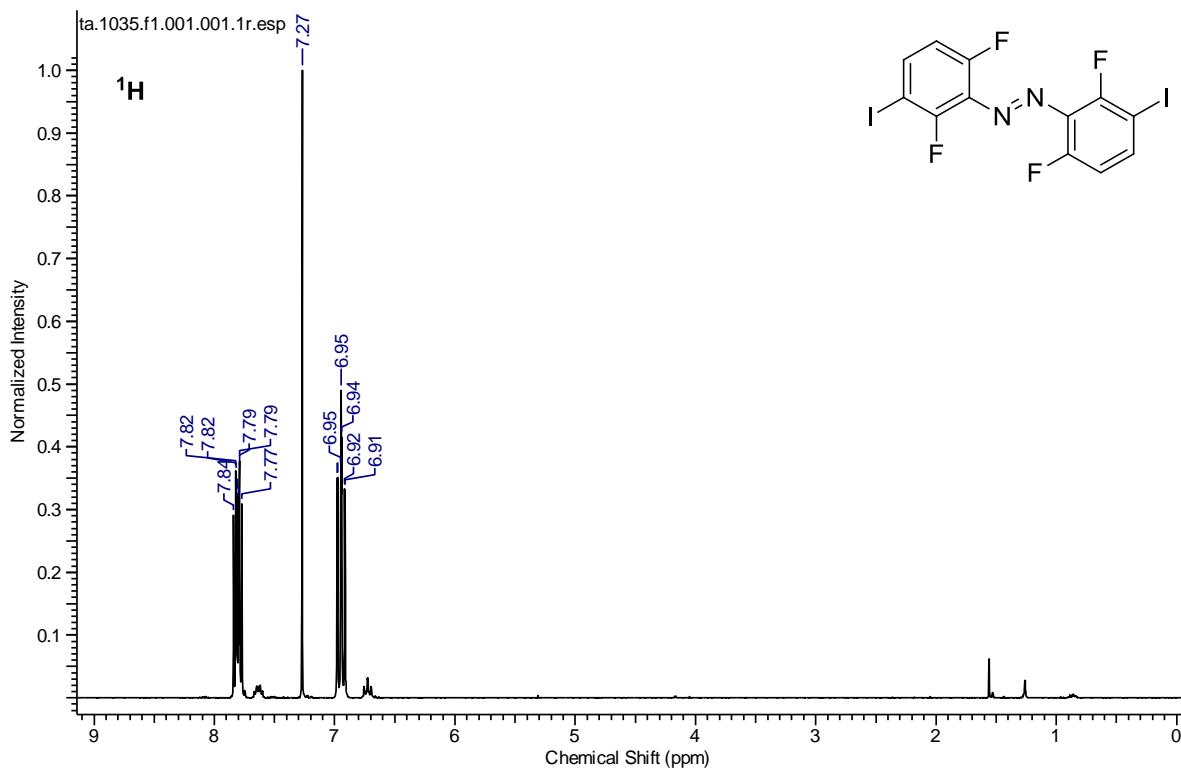
### III. NMR Spectra (<sup>1</sup>H, <sup>13</sup>C, <sup>19</sup>F, and <sup>31</sup>P NMR)

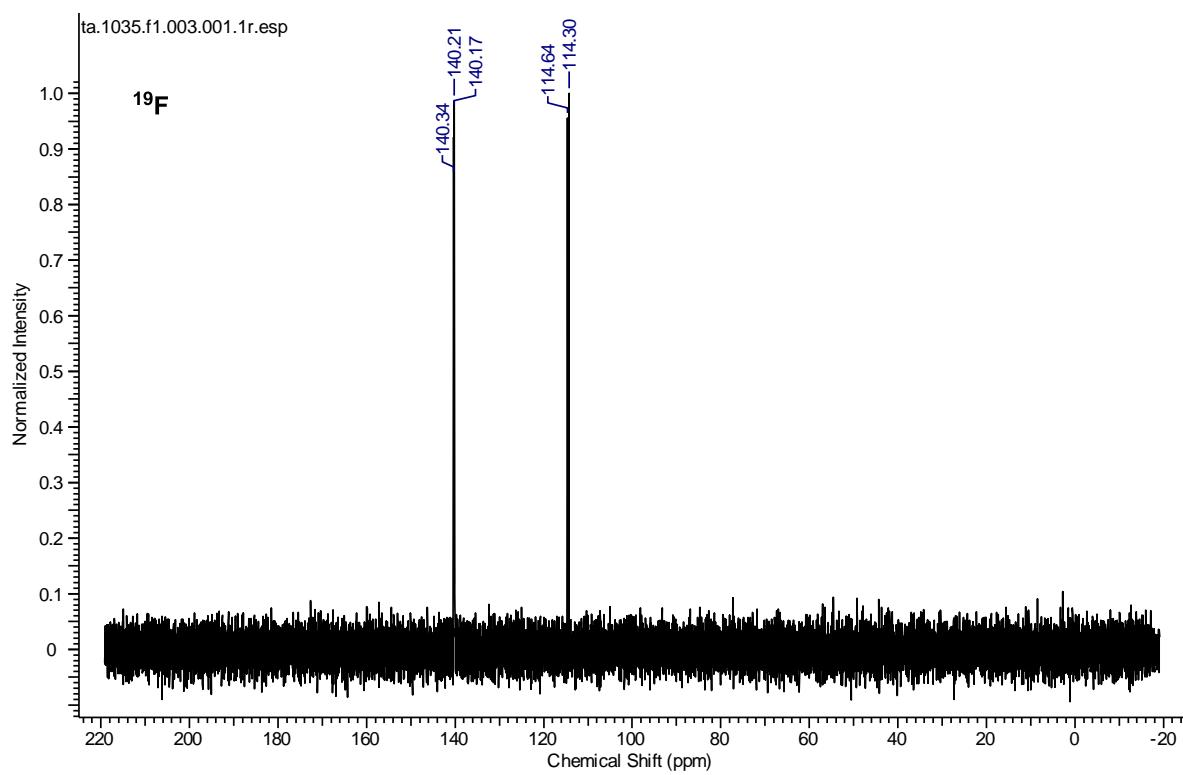
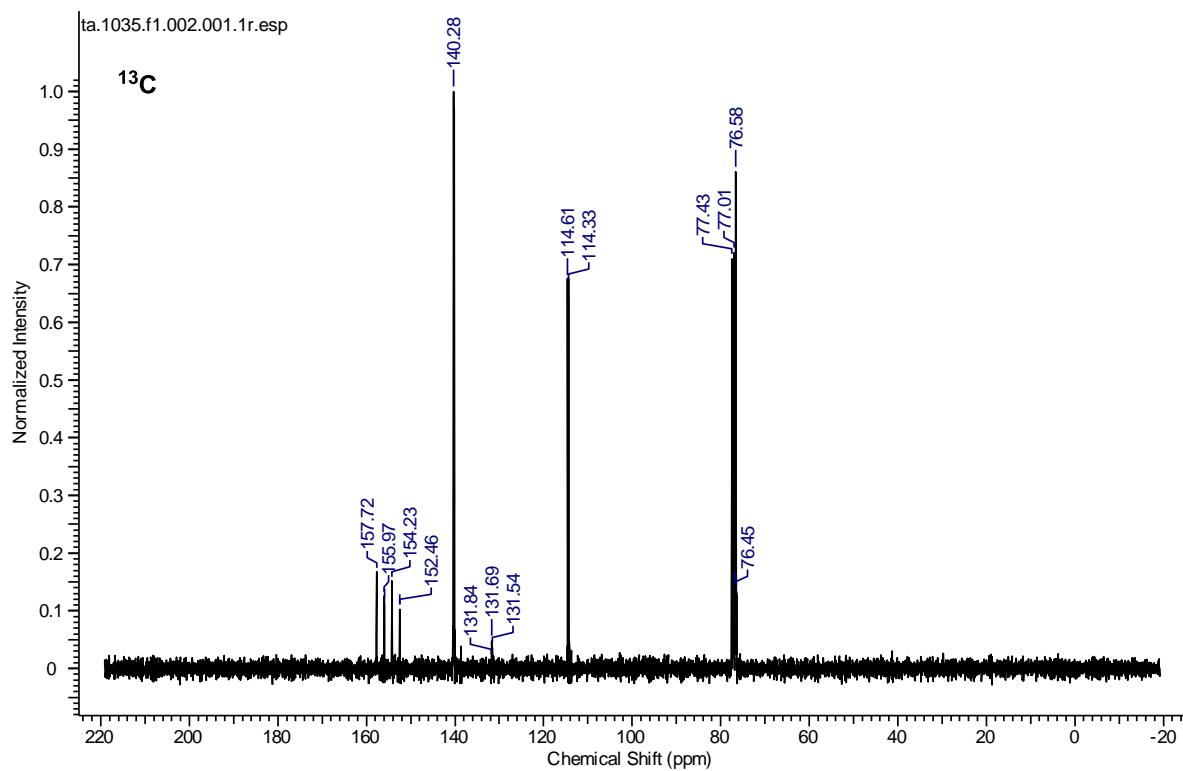
#### III.1. (SI-A).



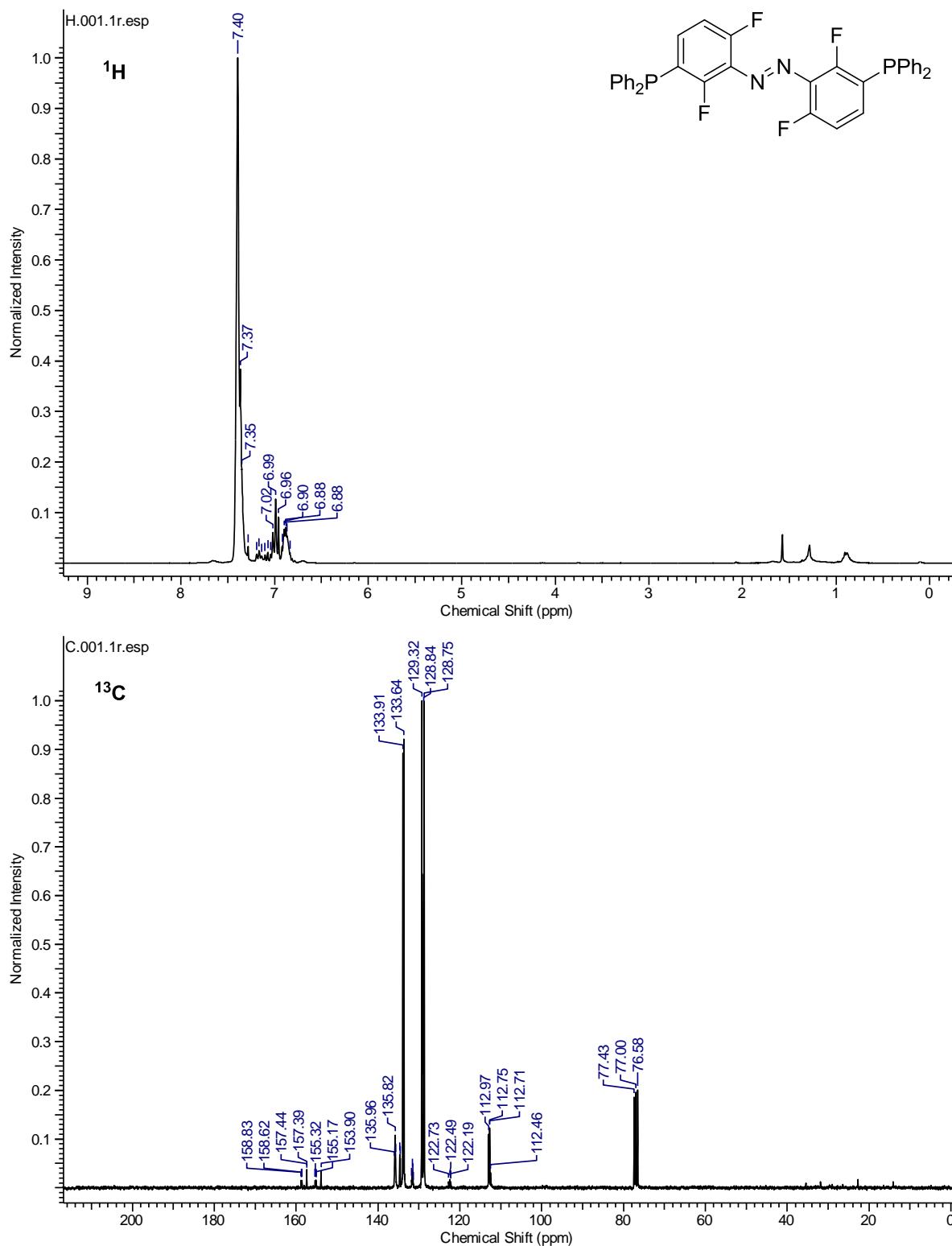


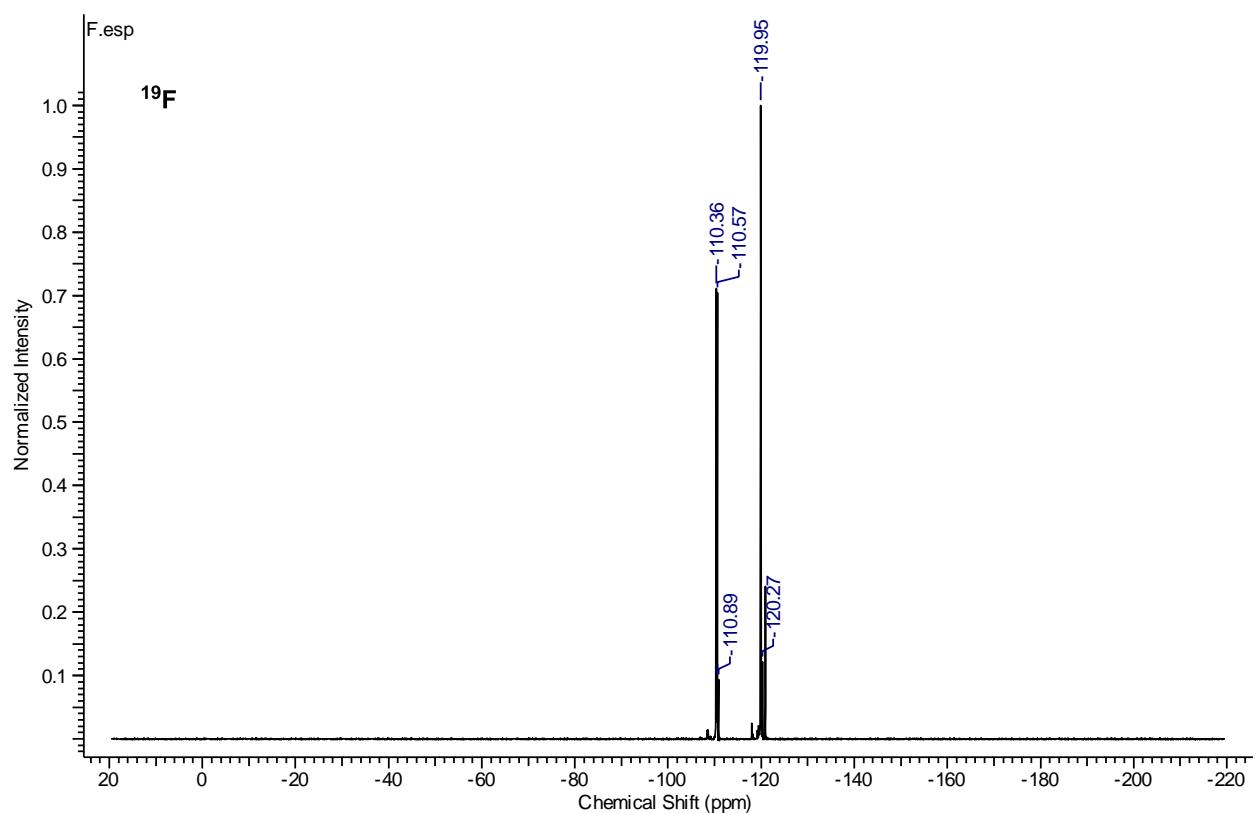
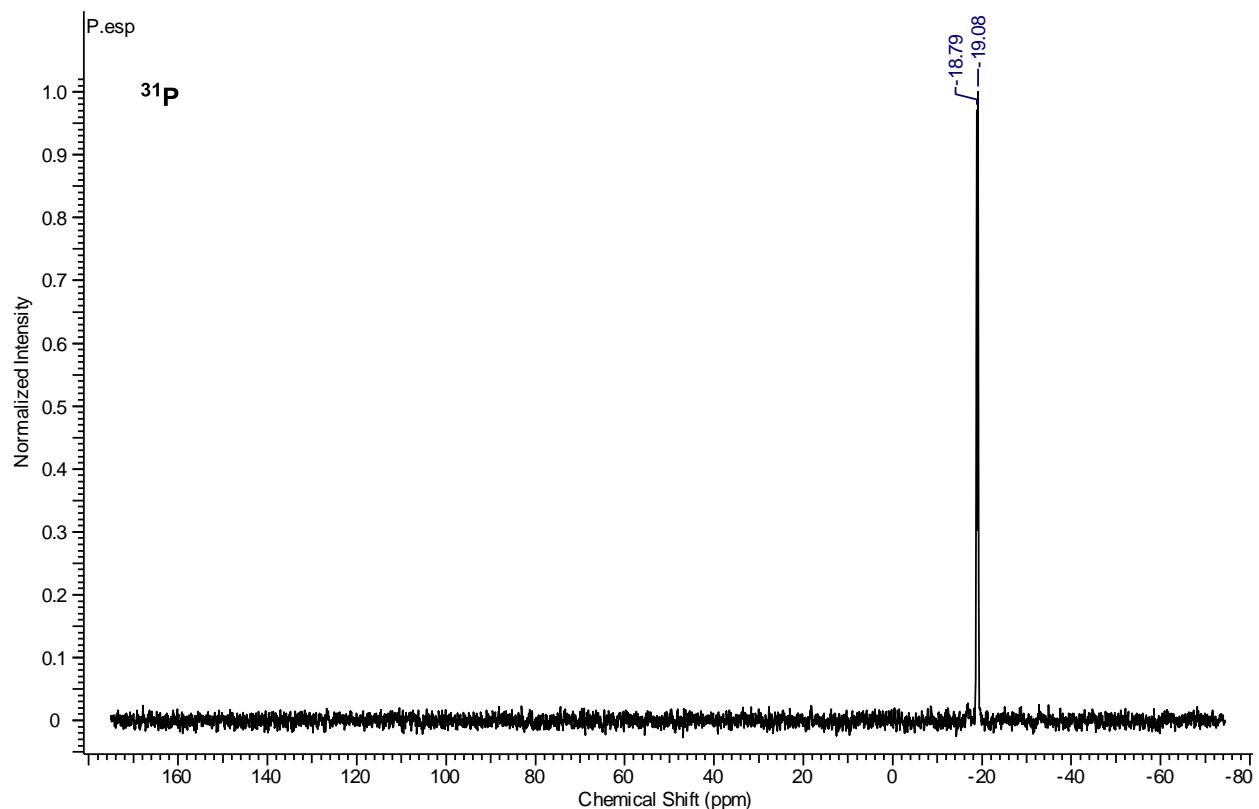
**III.2. (*E*)-1,2-bis(2,6-difluoro-3-iodophenyl)diazene (1).**



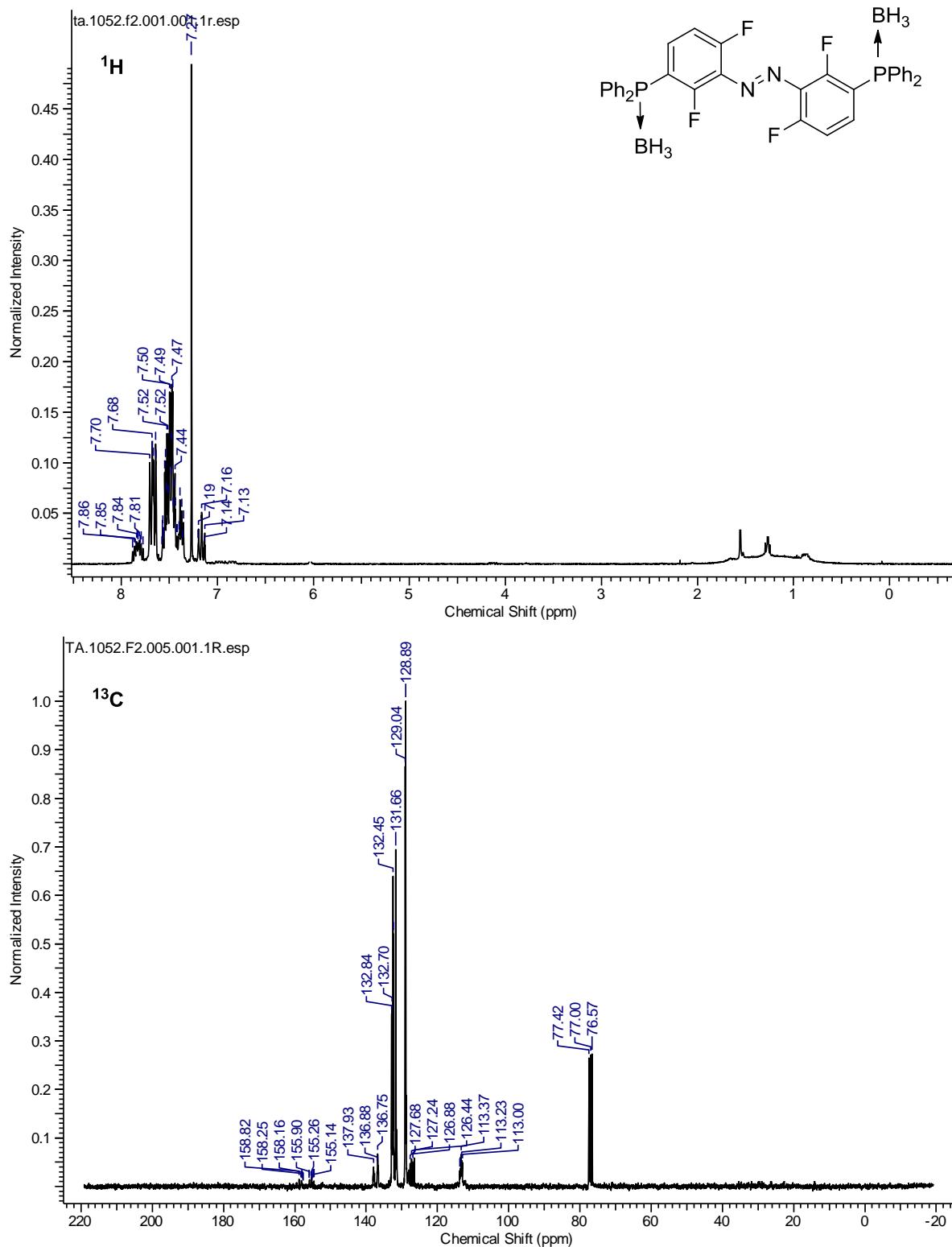


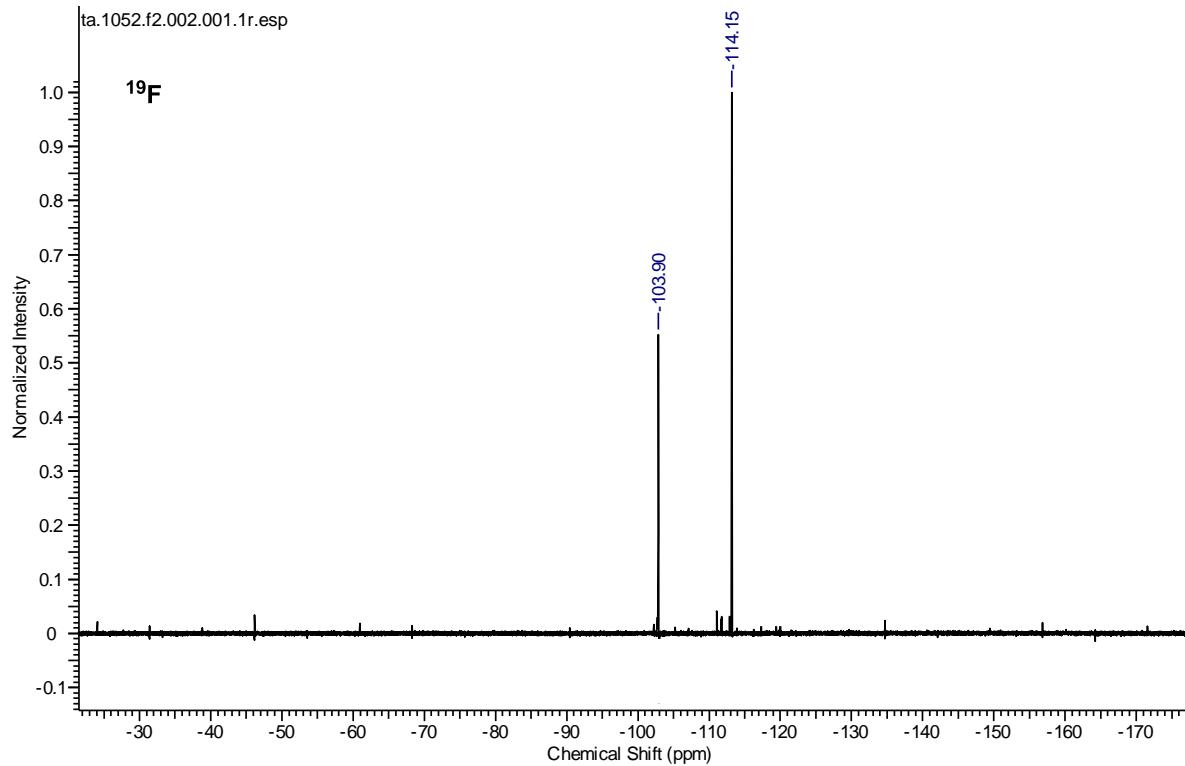
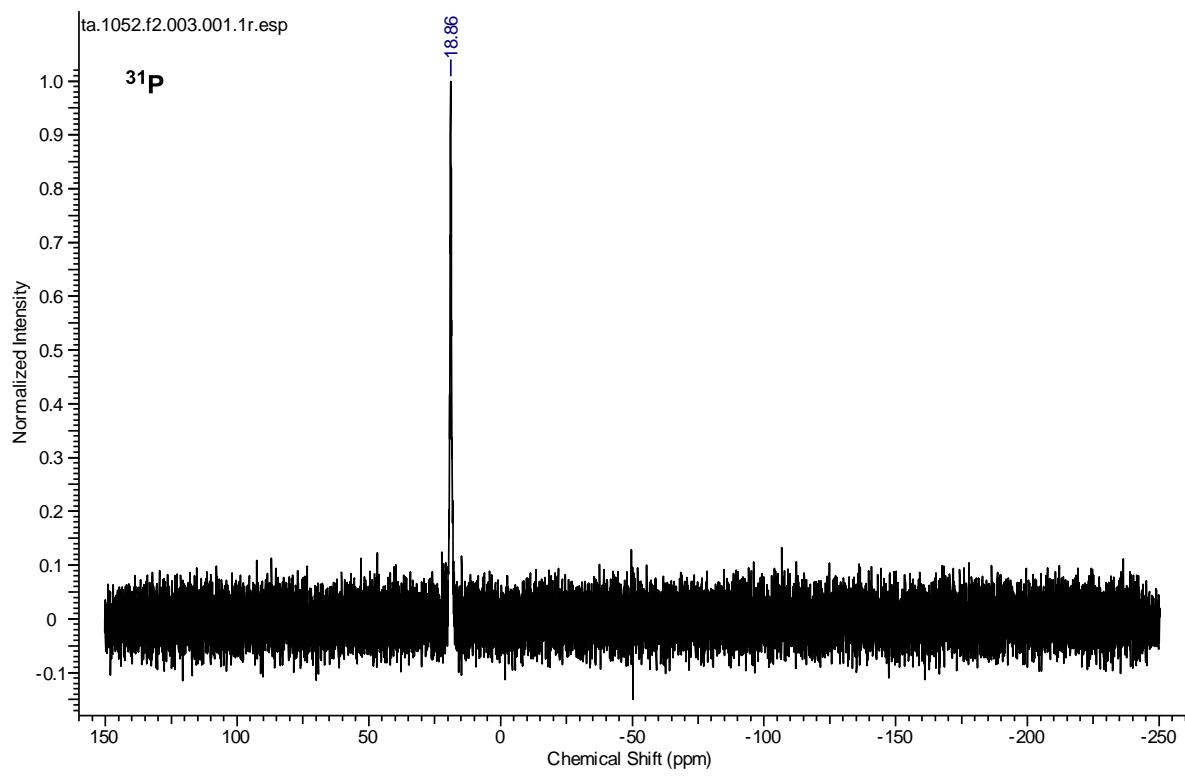
**III.3. (*E*)-1,2-bis(3-(diphenylphosphino)-2,6-difluorophenyl)diazene.**



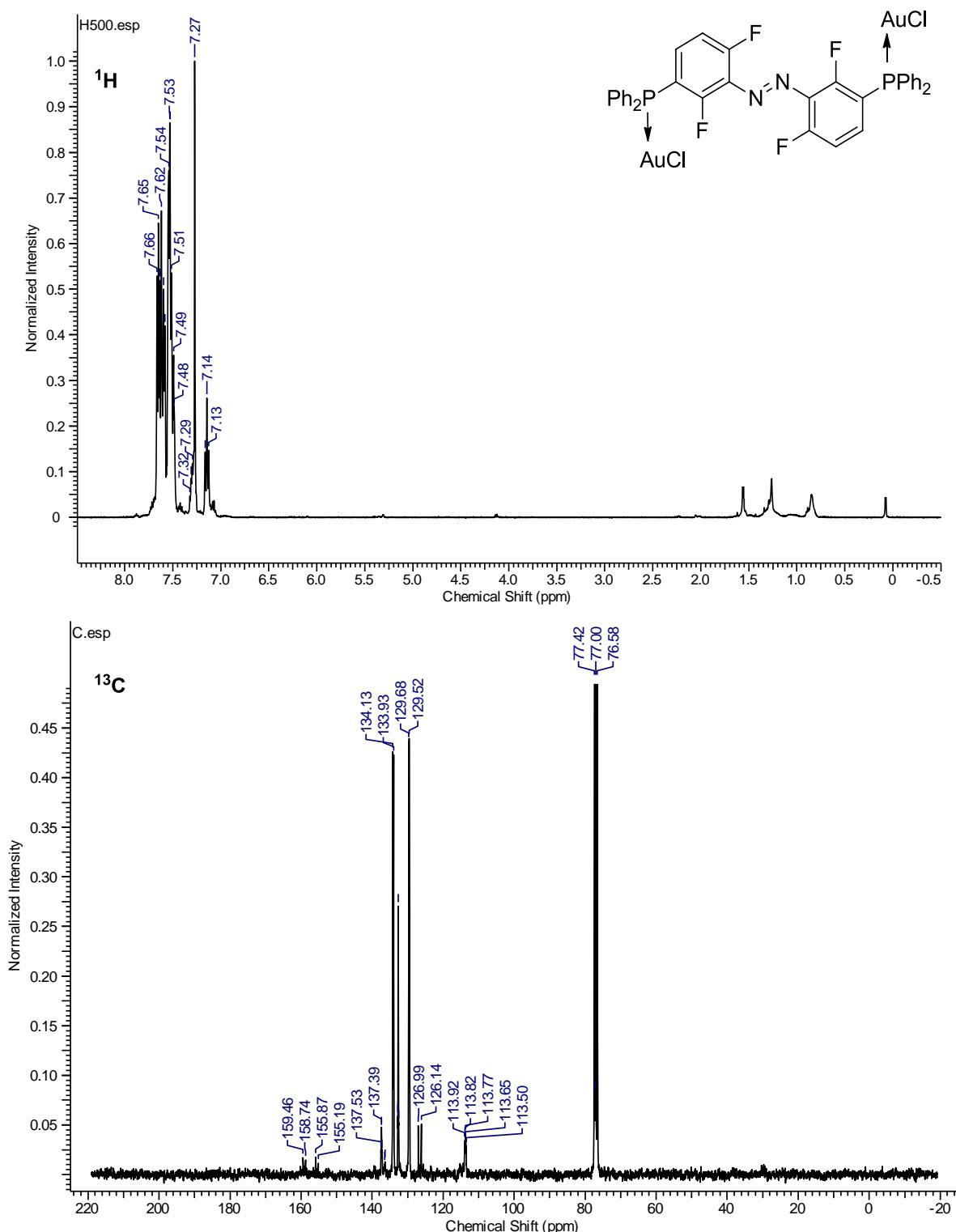


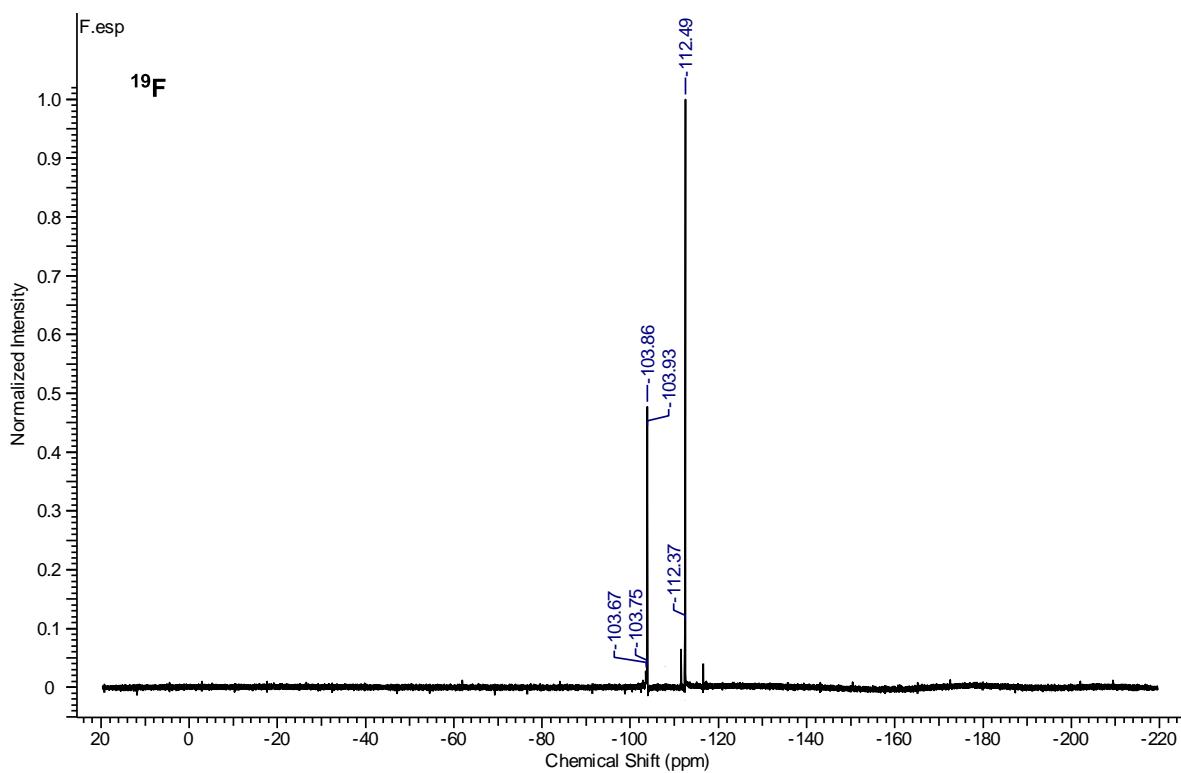
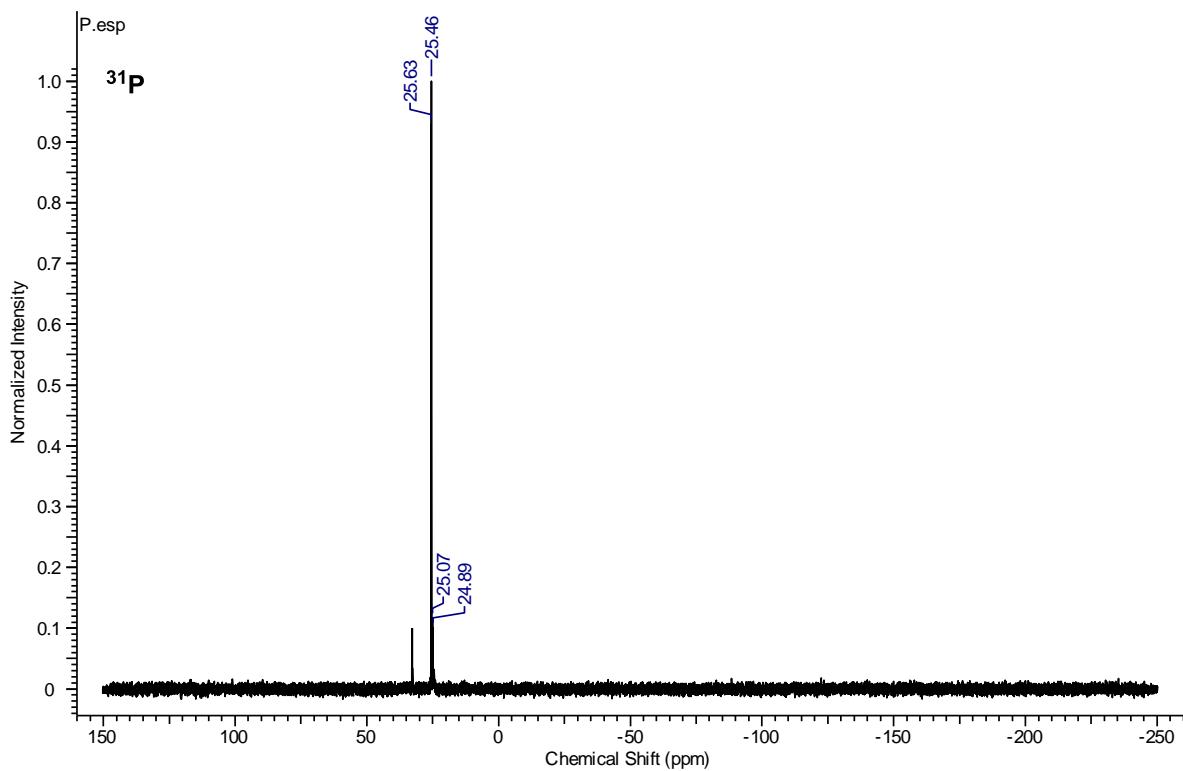
**III.4. (*E*)-1,2-bis(3-(diphenylphosphino)-2,6-difluorophenyl)diazene (2).**



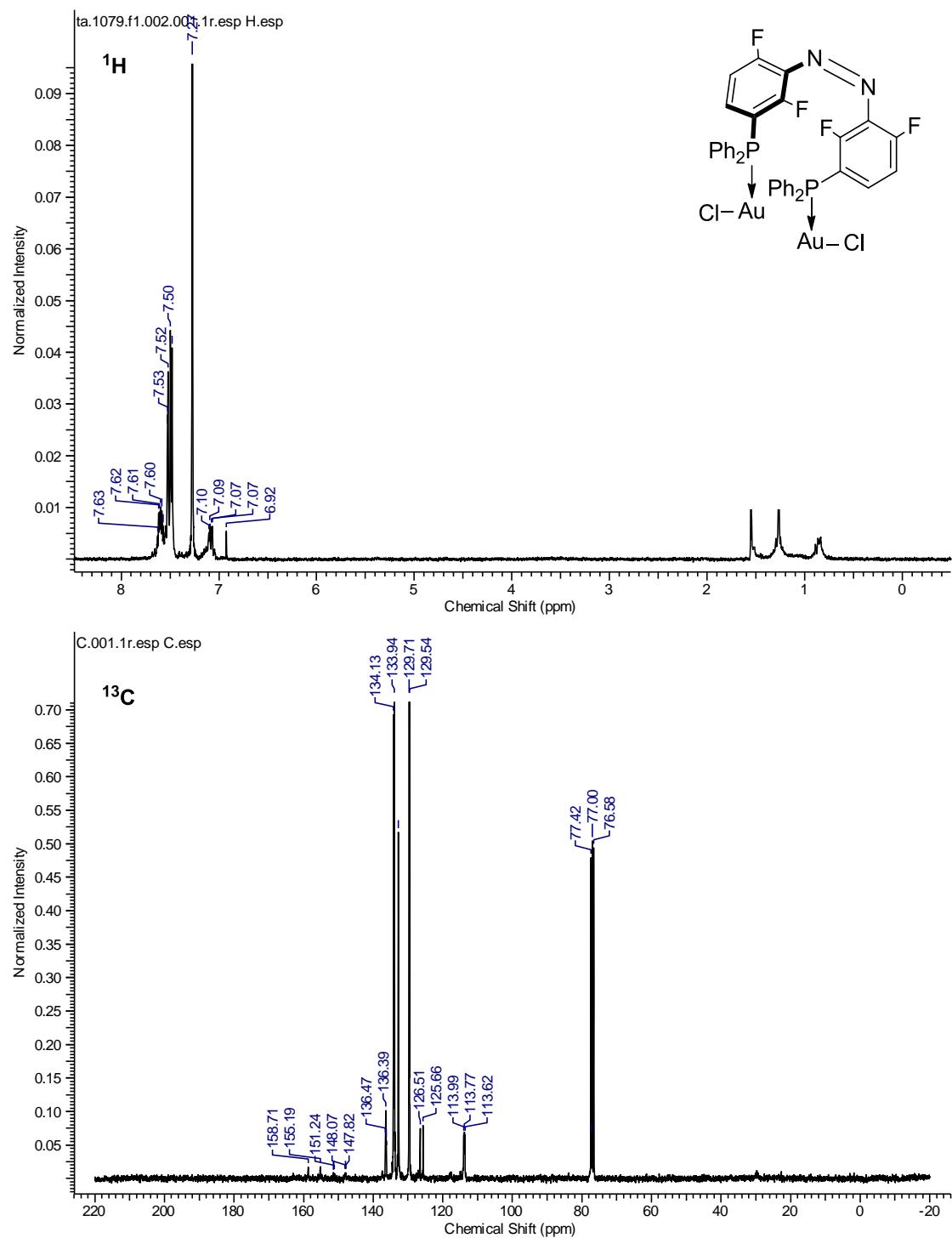


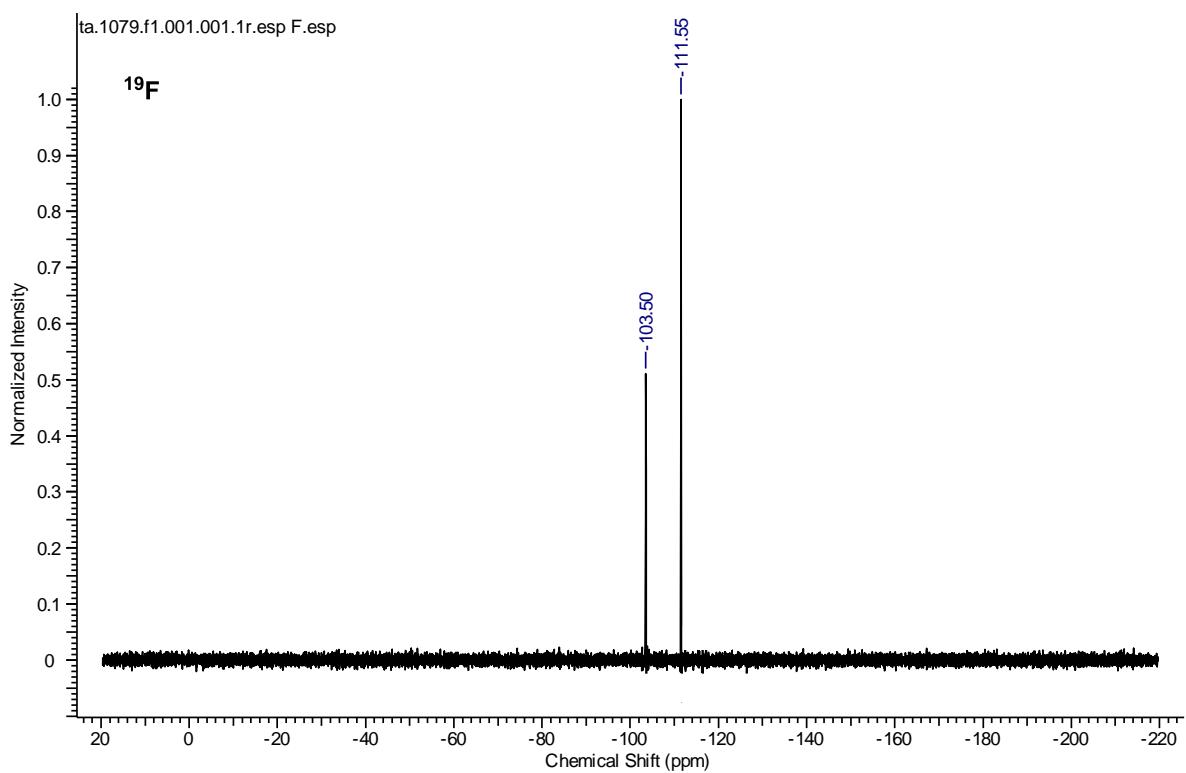
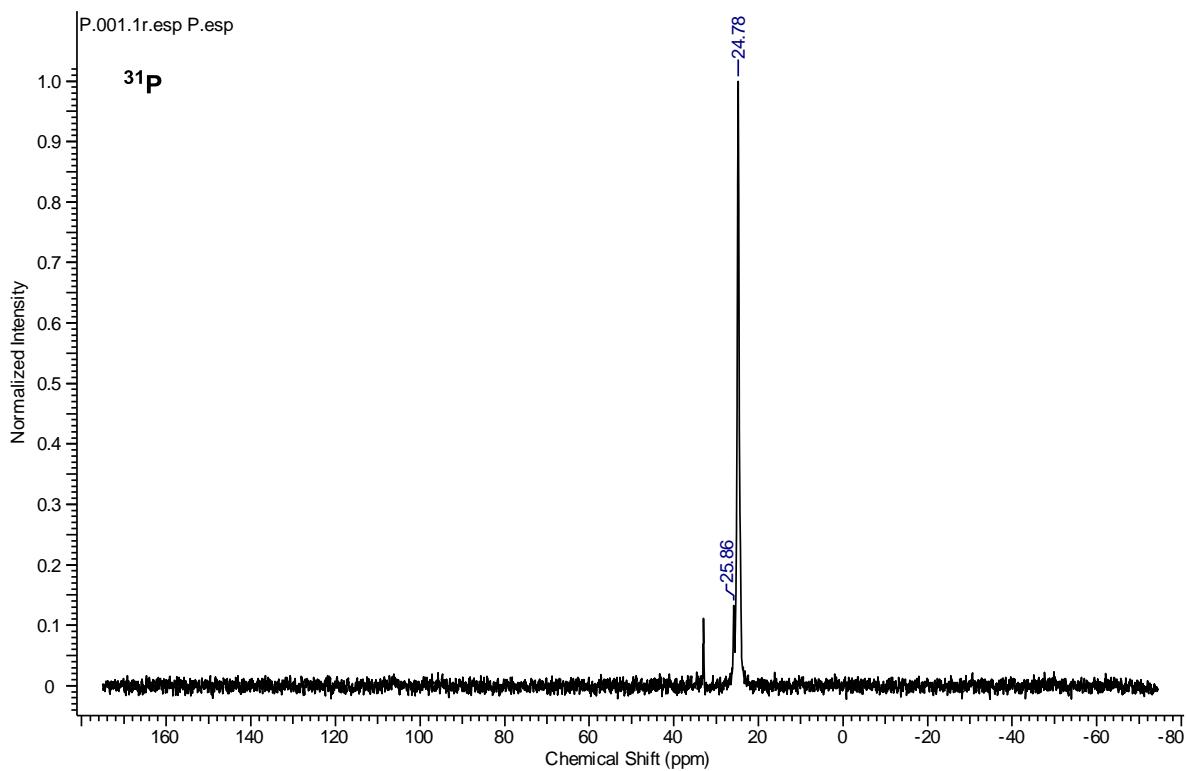
**III.5. (*E*)-1,2-bis(3-(diphenylphosphino)-2,6-difluorophenyl)diazene (*E*-3).**



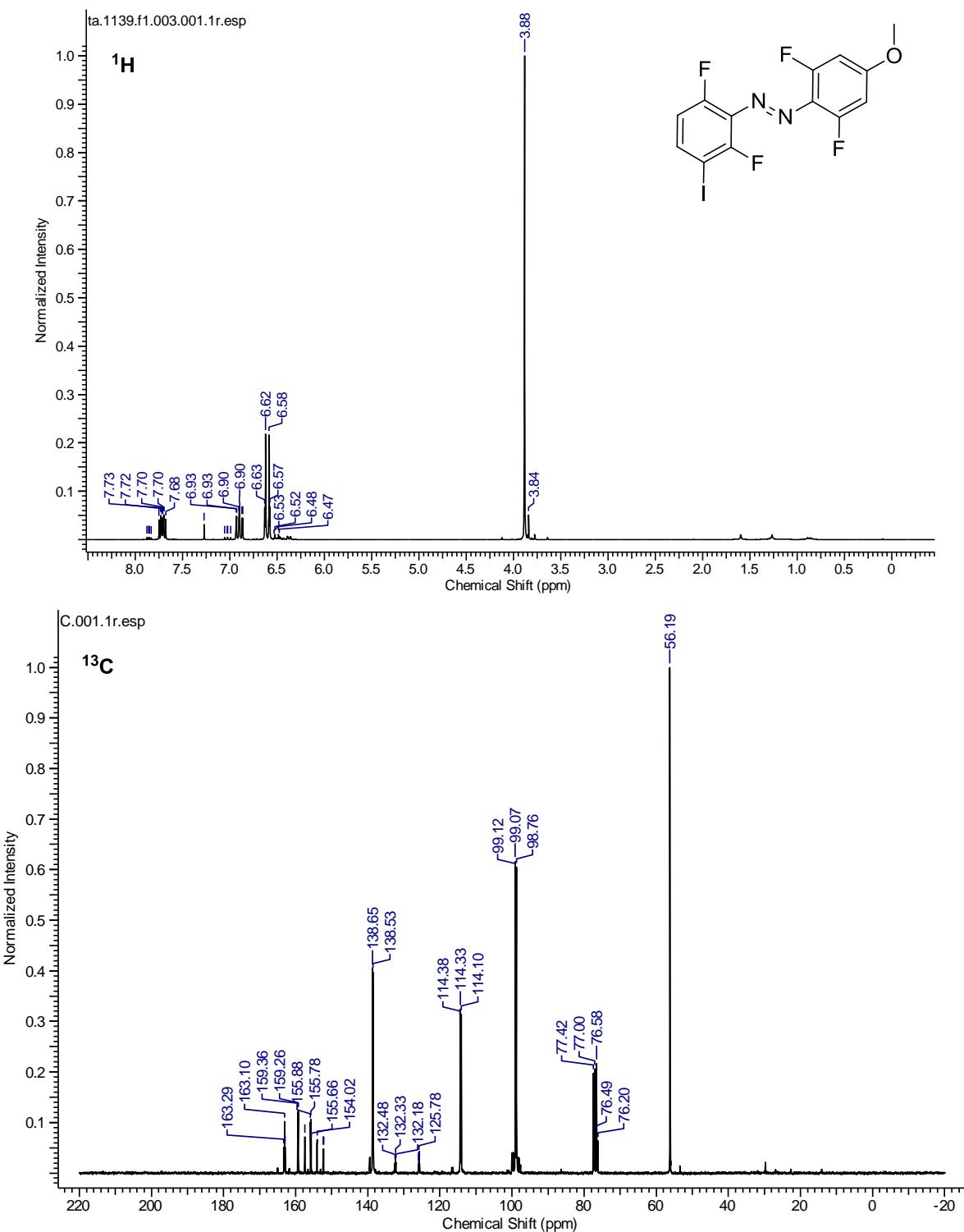


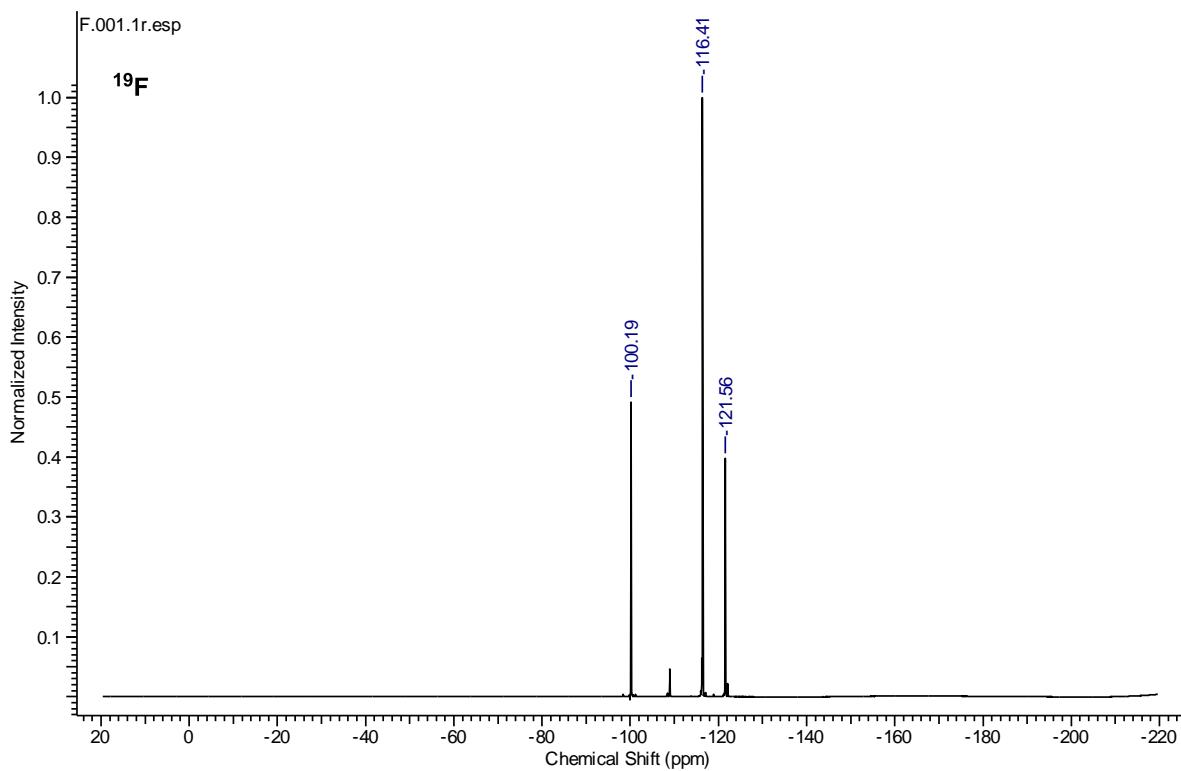
**III.6. (*Z*)-1,2-bis(3-(diphenylphosphino)-2,6-difluorophenyl)diazene (*Z*-3).**



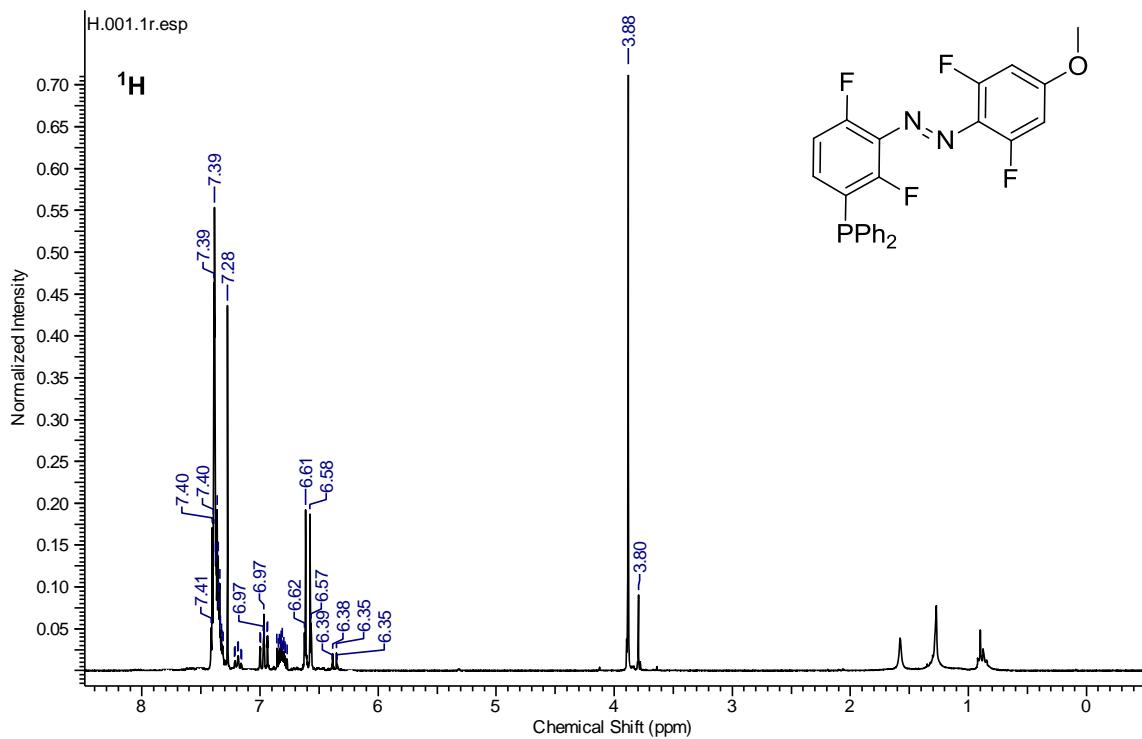


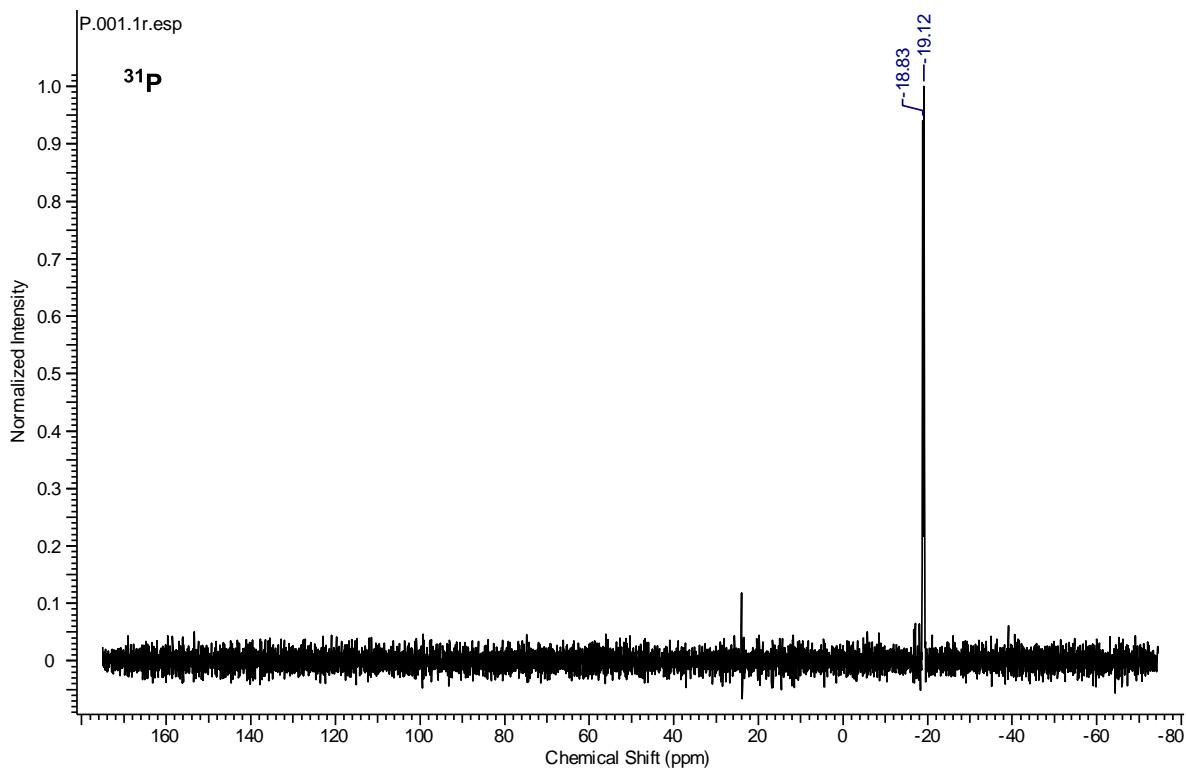
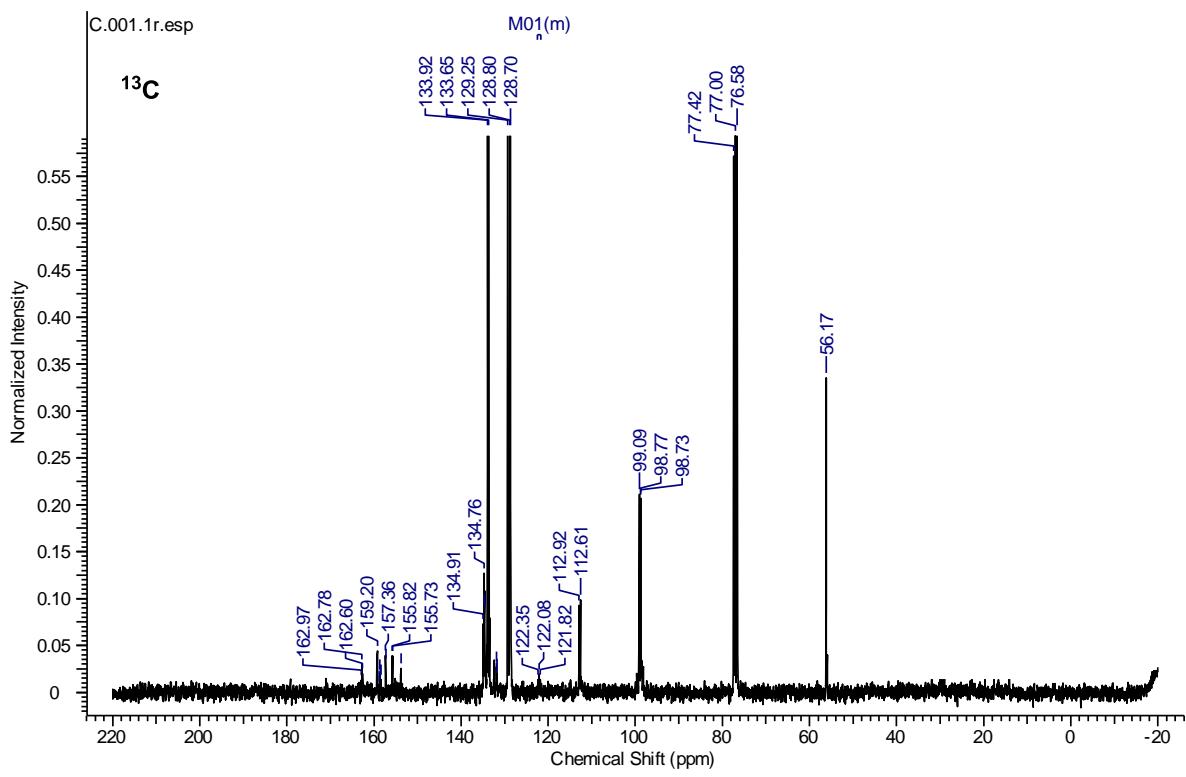
**III.7. (*E*)-1-(2,6-difluoro-3-iodophenyl)-2-(2,6-difluoro-4-methoxyphenyl)diazene (SI-C).**

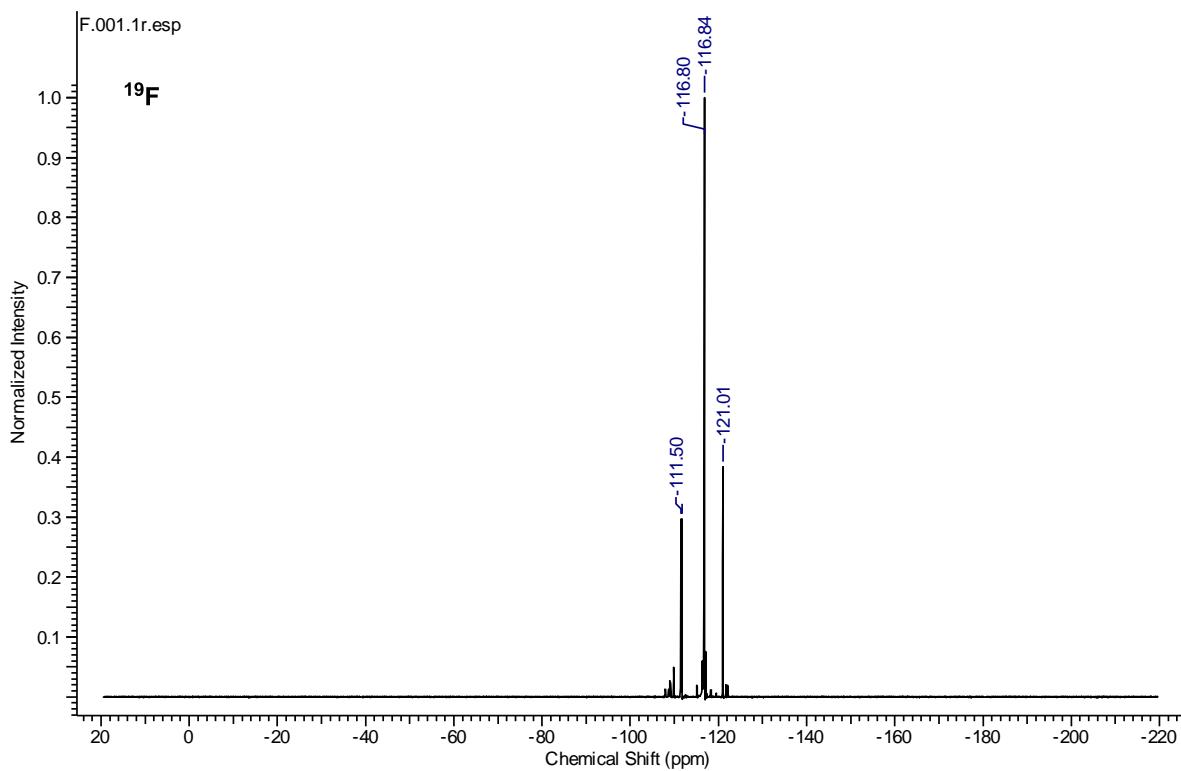




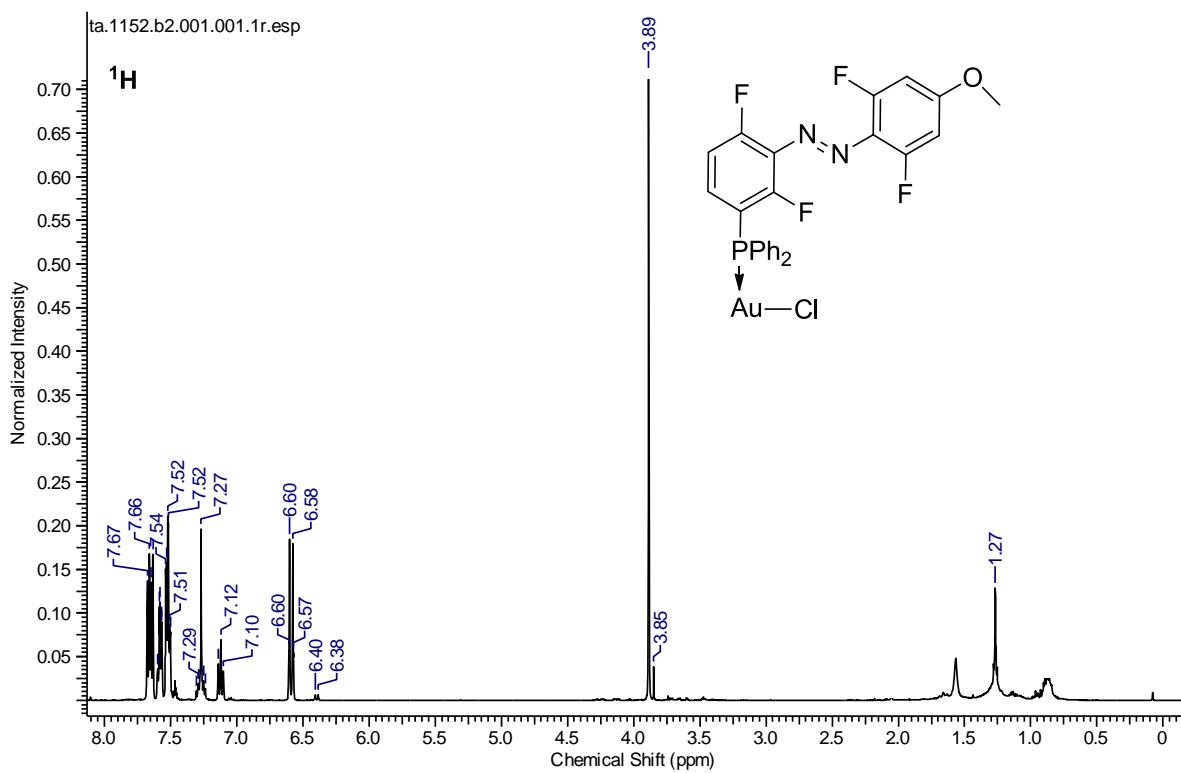
**III.8.(E)-1-(2,6-difluoro-4-methoxyphenyl)-2-(diphenylphosphino)-2,6-difluorophenyl)diazene (SI-D).**

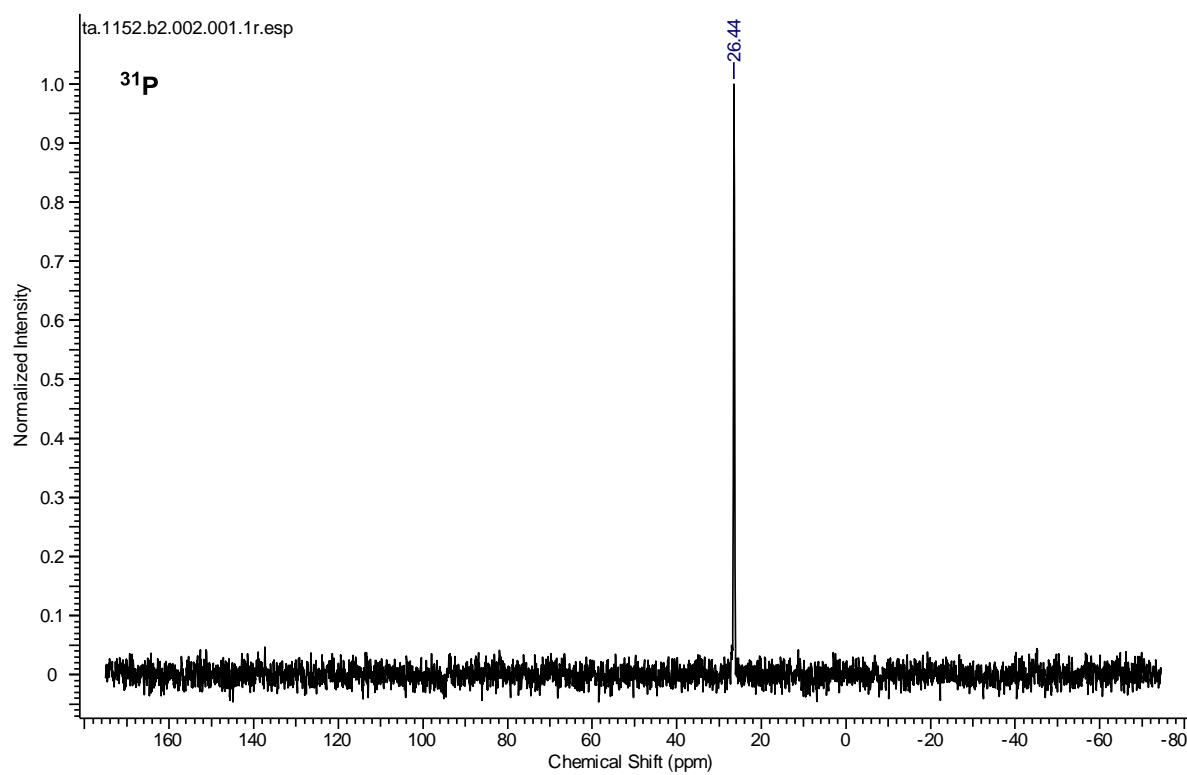
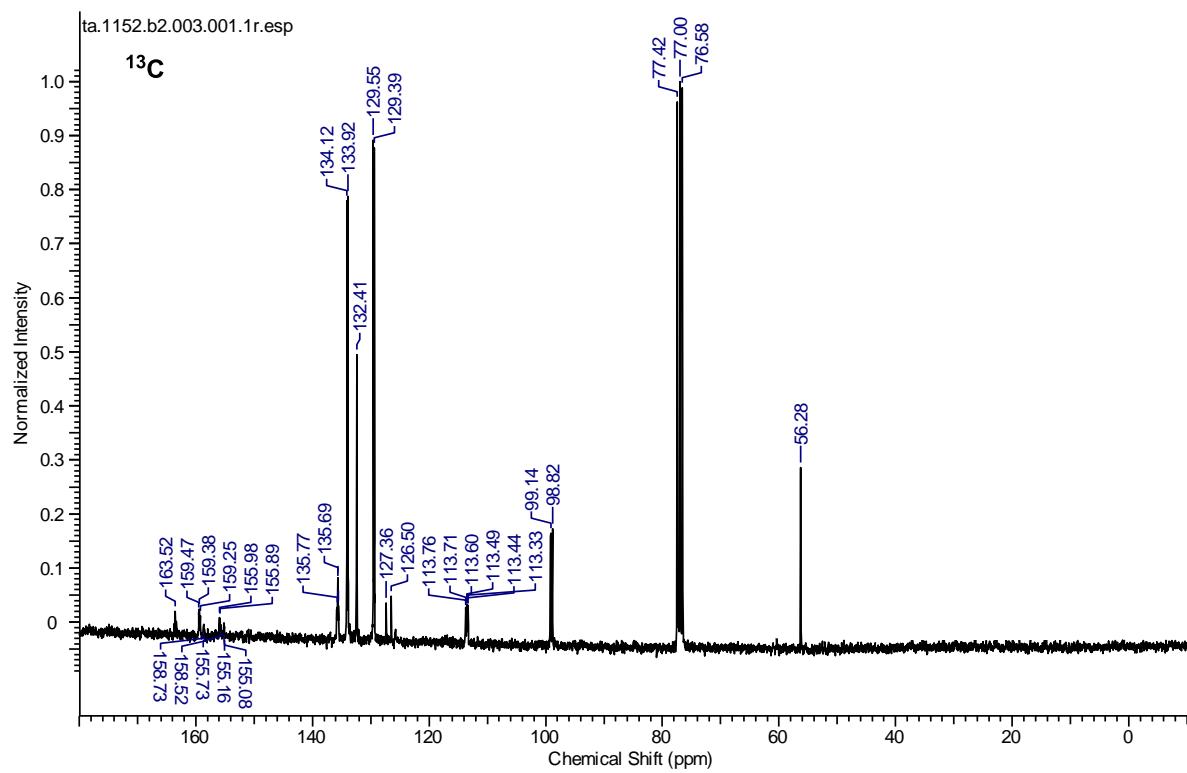


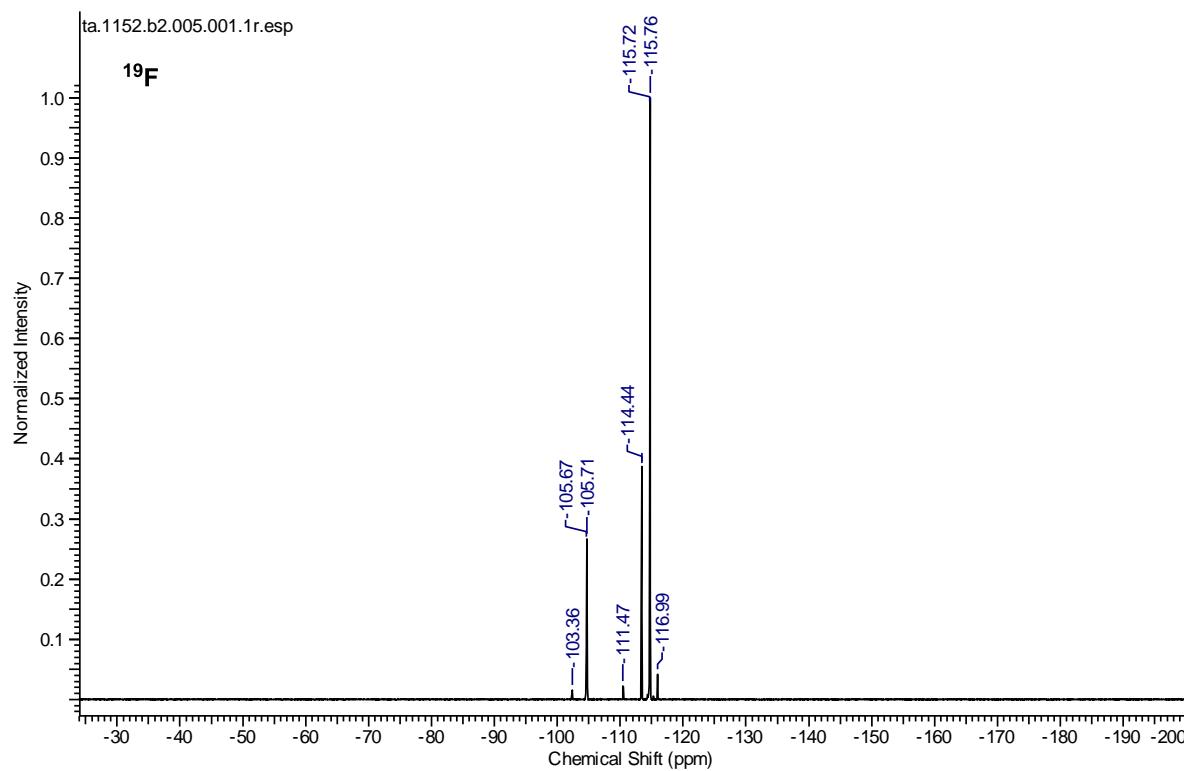




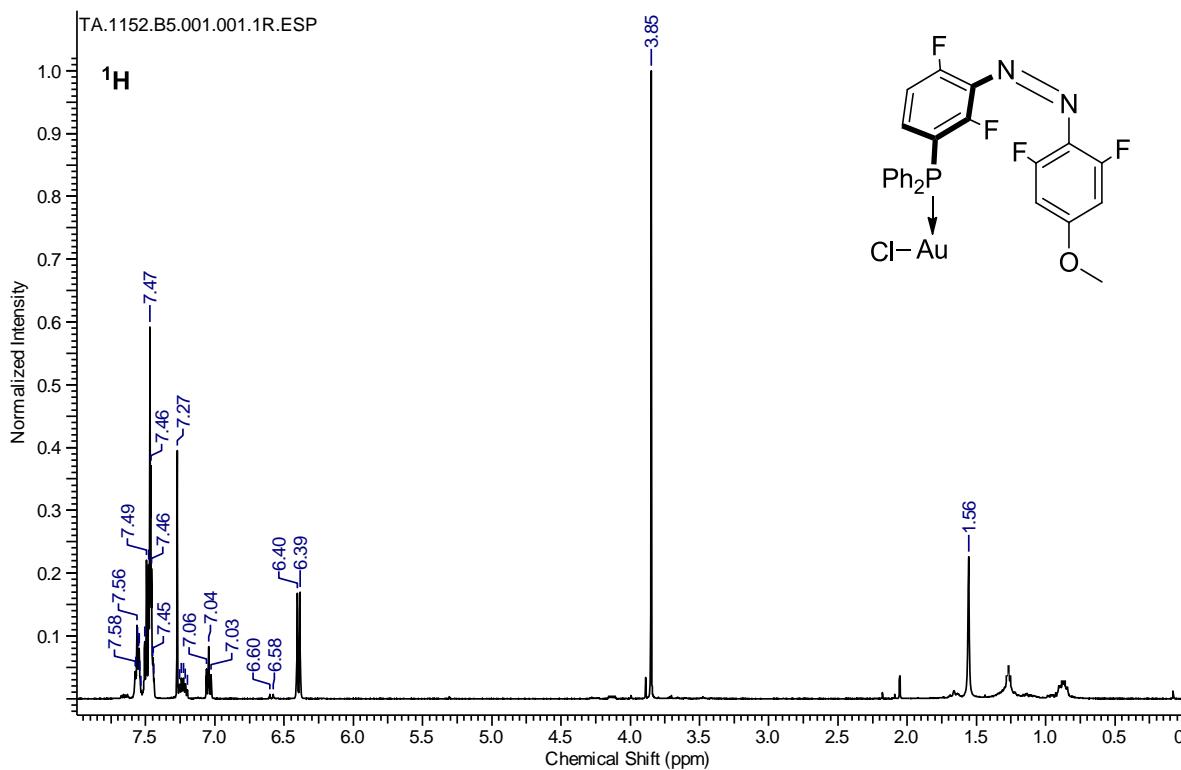
**III.9.** (*E*)-1-(2,6-difluoro-4-methoxyphenyl)-2-(3-(diphenylphosphino)-2,6-difluorophenyl)diazene (*E*-6).

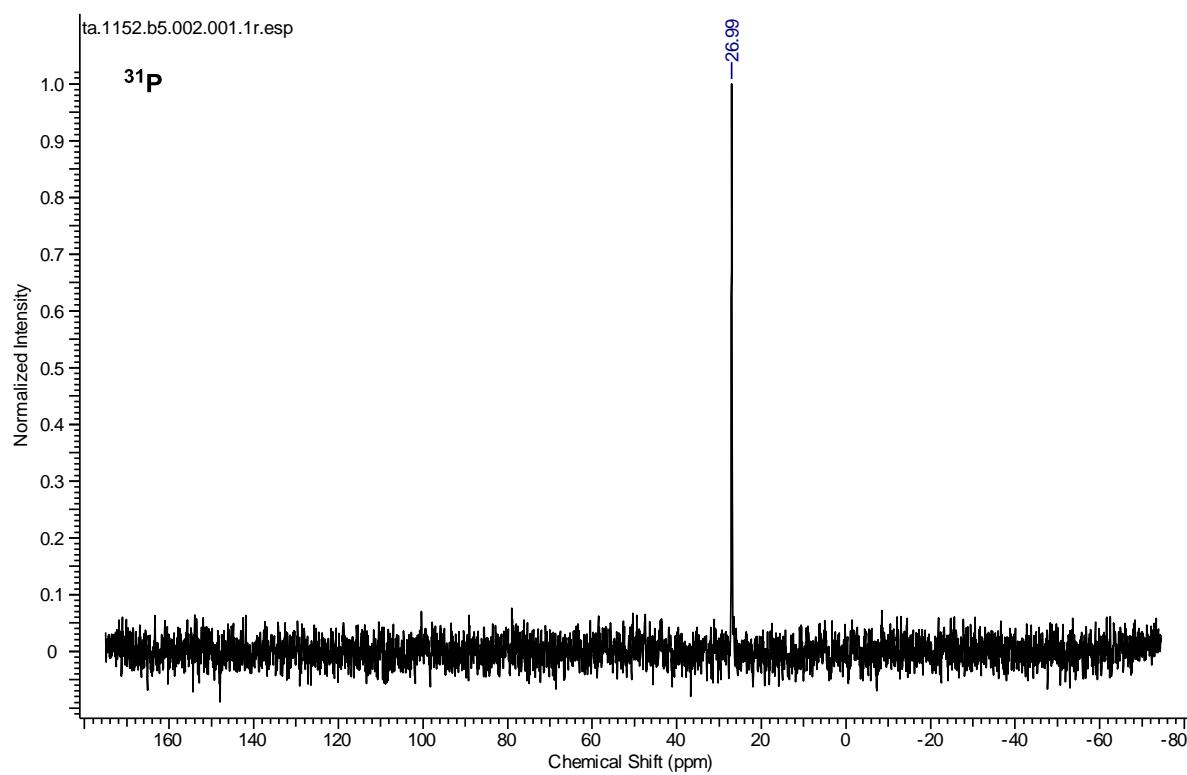
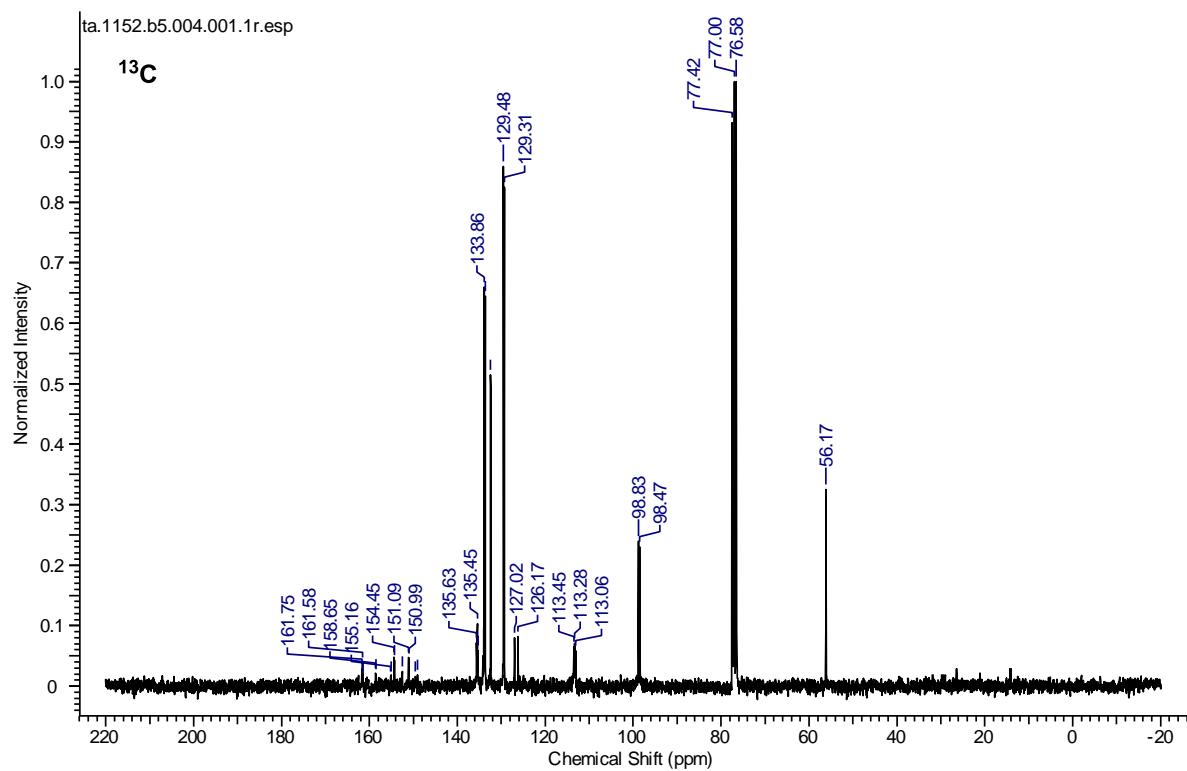


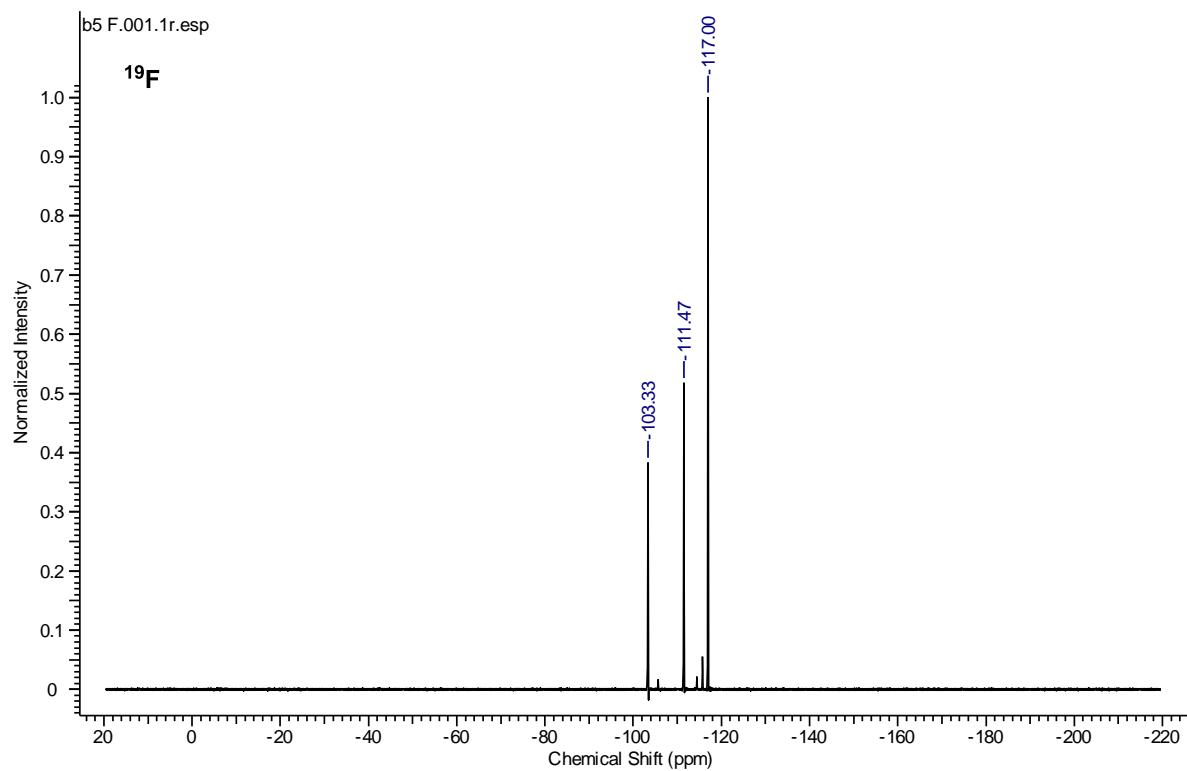




**III.10. (*Z*)-1-(2,6-difluoro-4-methoxyphenyl)-2-(3-(diphenylphosphino)-2,6-difluorophenyl) diazene (*Z*-6).**



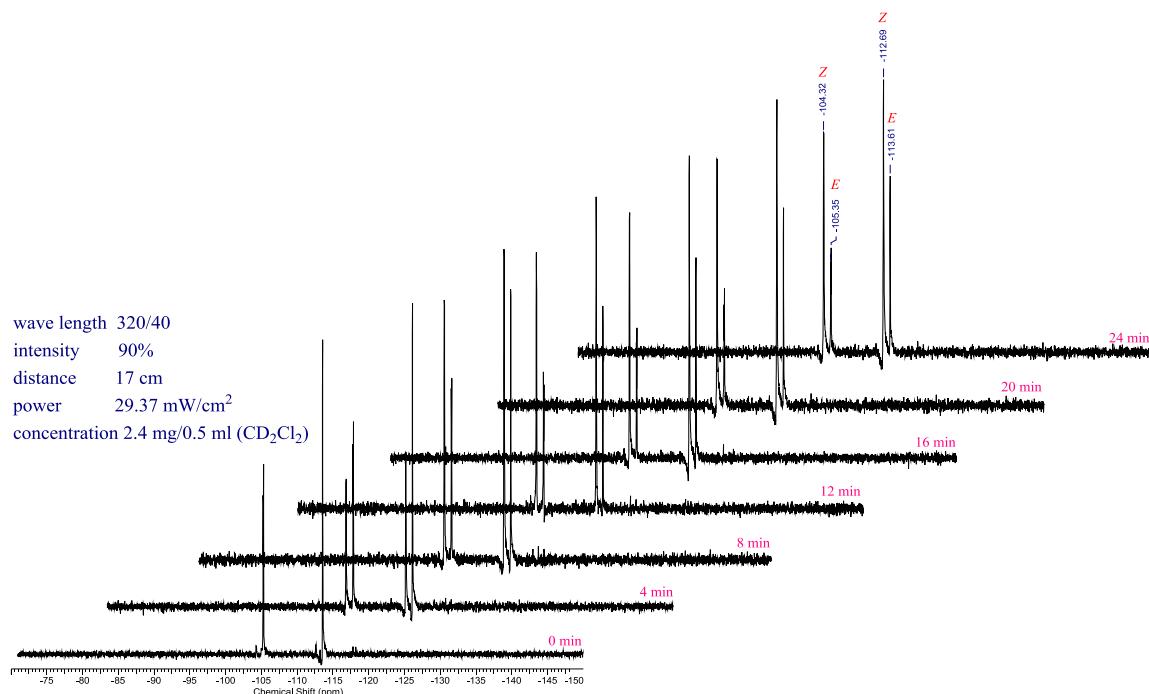
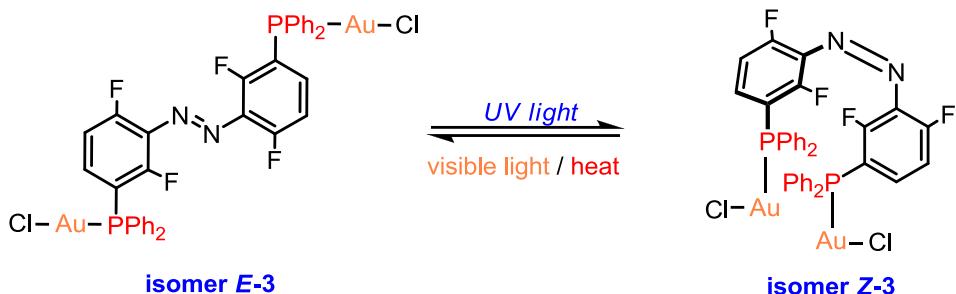




#### **IV. Photochromic Behaviour of complexes 3 and 6.**

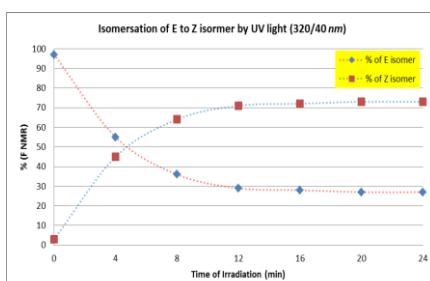
##### **Complexes E-3 and Z-3: study of isomeric transition by $^{19}\text{F}$ NMR.**

A solution of **E-3** (2.4 mg) in  $\text{CD}_2\text{Cl}_2$  (0.5 mL) was placed in NMR tube and irradiated with UV light (320/40 nm). The rate of isomerization was monitored by  $^{19}\text{F}$  NMR. The photo-stationary state (PPS) was reached after 16 minutes of irradiation. At PPS the isomeric ratio obtained was  $E:Z = 27:73$ .



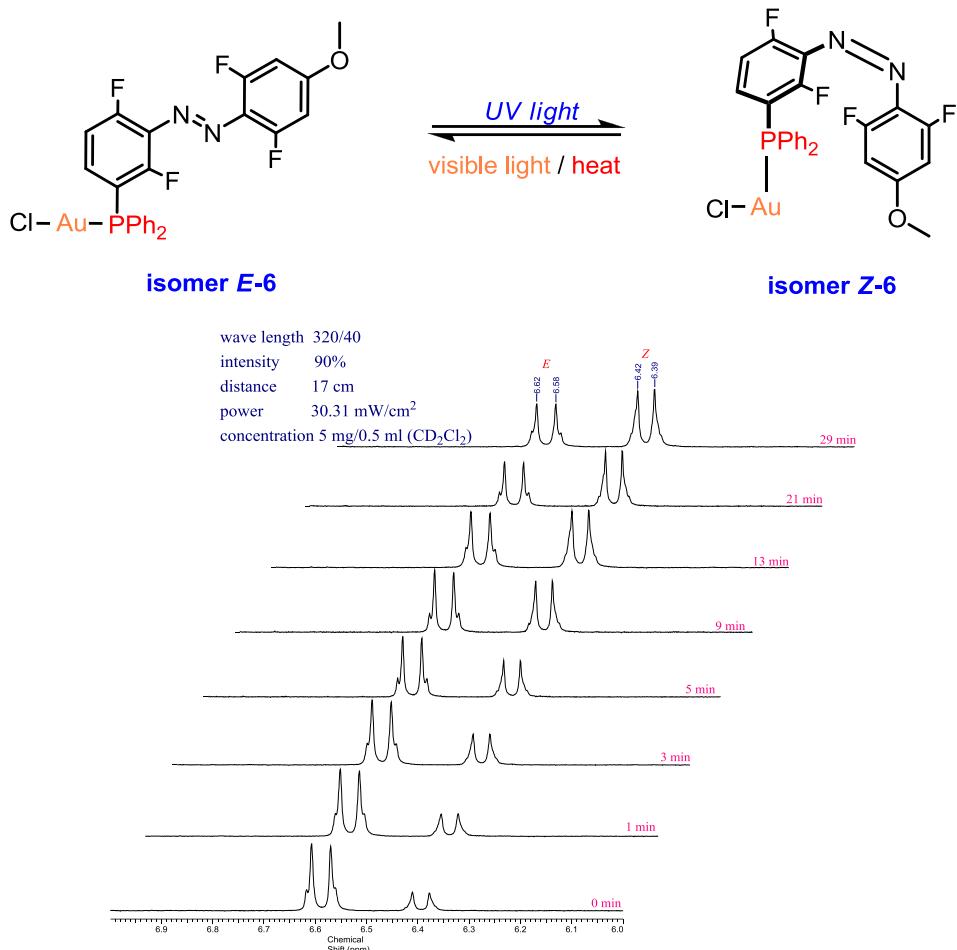
Changes in  $^{19}\text{F}$  NMR spectrum upon photo-isomerization in  $[\text{d}_2]$ -dichloromethane.

Time of irradiation	% E isomer	% Z isomer
0 min	97	3
4 min	55	45
8 min	36	64
12 min	29	71
16 min	28	72
20 min	27	73
24 min	27	73



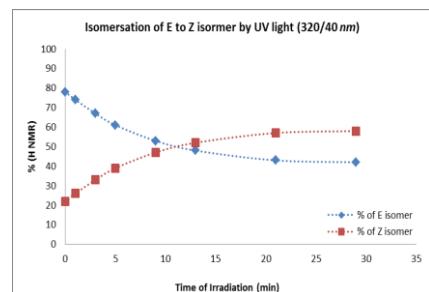
### Complexes E-6 and Z-6: study of isomeric Transition by $^1\text{H}$ NMR:

A solution of **E-6** (5 mg) in  $\text{CD}_2\text{Cl}_2$  (0.5 mL) was placed in NMR tube and irradiated with UV light (320/40 nm). The rate of isomerization was monitored by  $^1\text{H}$  NMR. The photo-stationary state (PPS) was reached after 29 minutes of irradiation. At PPS the isomeric ration obtained was  $E:Z = 42:58$ .



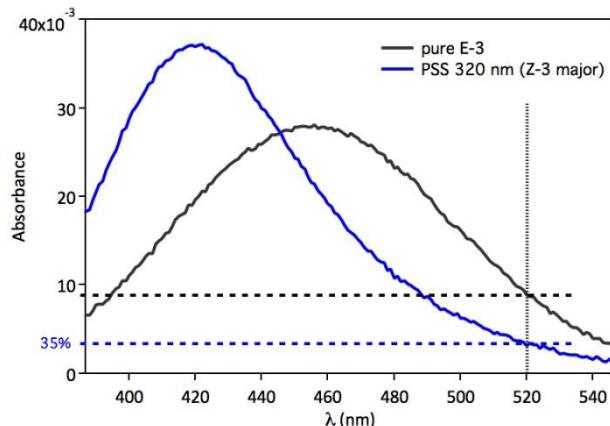
Changes in  $^1\text{H}$  NMR spectrum upon photo-isomerization in  $[\text{d}_2]$ -dichloromethane.

Time of irradiation	% of E isomer	% of Z isomer
0	78	22
1	74	26
3	67	33
5	61	39
9	53	47
13	48	52
21	43	57
29	42	58

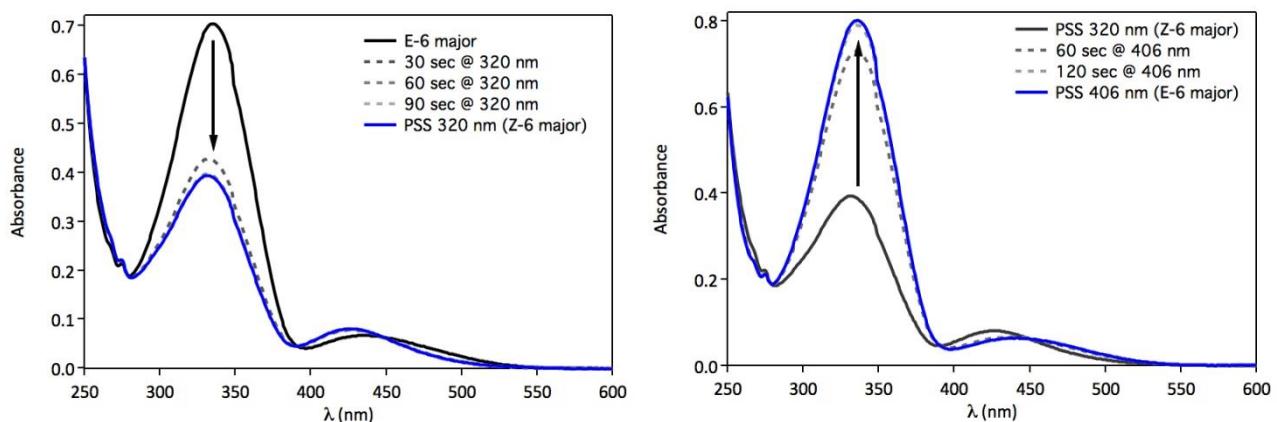


### Spectroscopic Measurements:

Acetonitrile was purchased from Aldrich (spectrometric grade), UV/Vis absorption spectra were recorded on a Cary5000 spectrophotometer from Agilent Technologies using a Peltier cuvette holder to monitor the temperature. Photo-isomerizations were induced by a continuous irradiation Hg/Xe lamp (Hamamatsu, LC6 Lightningcure, 200 W) equipped with narrow band interference filters of appropriate wavelengths (Semrock FF01-320/40 for  $\lambda_{\text{irr}} = 320 \text{ nm}$ ; FF01-406/15-25 for  $\lambda_{\text{irr}} = 406 \text{ nm}$ ). The irradiation power was measured using a photodiode from Ophir (PD300-UV).



**Figure SI1.** Magnification of the absorption spectrum of *E*-3 and PSS 320 nm (*Z*-3 major) in MeCN at 25 °C allowing determination of a ratio  $Z\text{-}3:E\text{-}3 \approx 65:35$  in PSS 320 nm.



**Figure SI2.** Left: Absorption spectrum of *E*-6 (major) in MeCN at 25 °C and its stepwise evolution upon irradiation pulses at 320 nm ( $4 \text{ mW.cm}^{-2}$ ). Right: Evolution of absorption spectrum of the previous solution at PSS 320 nm (*Z*-6 major) in MeCN at 25 °C and its stepwise evolution upon irradiation pulses at 406 nm ( $10 \text{ mW.cm}^{-2}$ ).

### General method for the determination of the half-lives of the complexes **3** and **6**.

A solution of complex **E-3** or **E-6** ( $C \sim 20 \mu\text{M}$ ) in MeCN at  $25^\circ\text{C}$  was irradiated at  $320 \text{ nm}$  ( $4 \text{ mW.cm}^{-2}$ ) for 6 min. Then, the sample was placed in a thermostated cuvette holder at  $60^\circ\text{C}$ , and  $Z \rightarrow E$  thermal back-isomerization was monitored by recording absorption at an appropriate wavelength ( $310 \text{ nm}$  and  $335 \text{ nm}$  for **3** and **6**, respectively).

Absorption was plotted versus time and fitted with an exponential curve corresponding to a first order kinetics reaction.

$$A(t) = A_\infty + (A_0 - A_\infty)e^{-kt}$$

where  $A(t)$  is the absorption at time  $t$

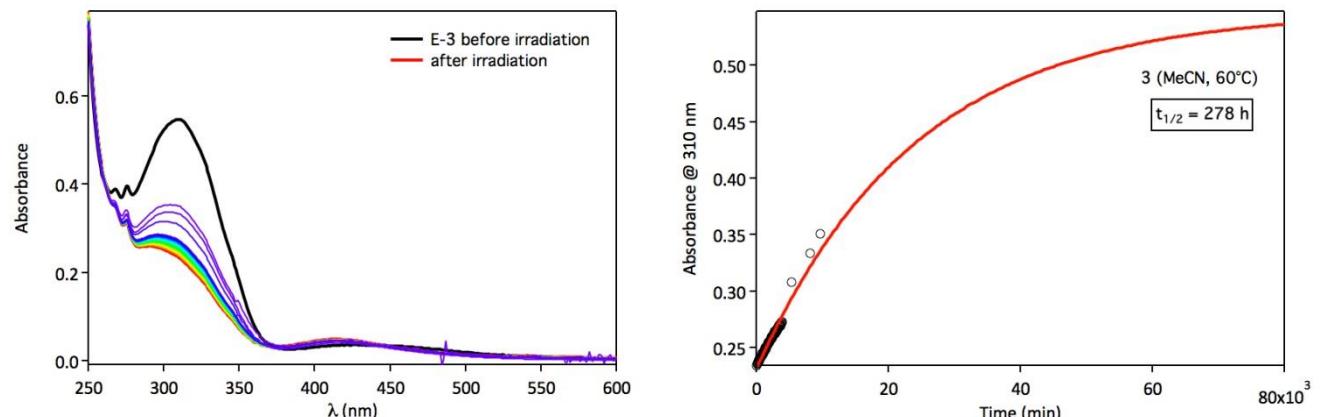
$A_0$  the absorption after irradiation

$A_\infty$  the absorbance after full thermal relaxation in the dark, corresponding to the absorption before irradiation (*E*-isomer)

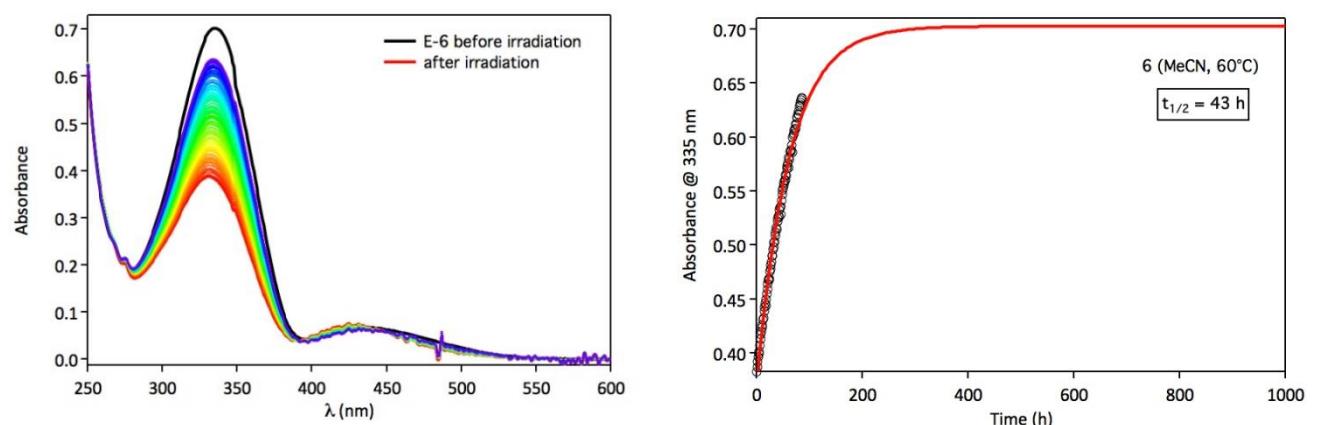
$k$  the kinetic constant in  $\text{min}^{-1}$

The half-life  $t_{1/2}$  in minutes of the (*Z*) isomer was calculated with the following equation:

$$t_{1/2} = \frac{\ln 2}{k}$$



**Figure SI3.** Left: Absorption spectra of **E-3** in MeCN at  $25^\circ\text{C}$  before (black line) and after irradiation (red line) at  $320 \text{ nm}$  (6 min at  $4 \text{ mW.cm}^{-2}$ ), and its evolution in the dark at  $60^\circ\text{C}$  (red to violet lines). Right: Evolution of absorption at  $310 \text{ nm}$  as a function of time in the dark at  $60^\circ\text{C}$ .

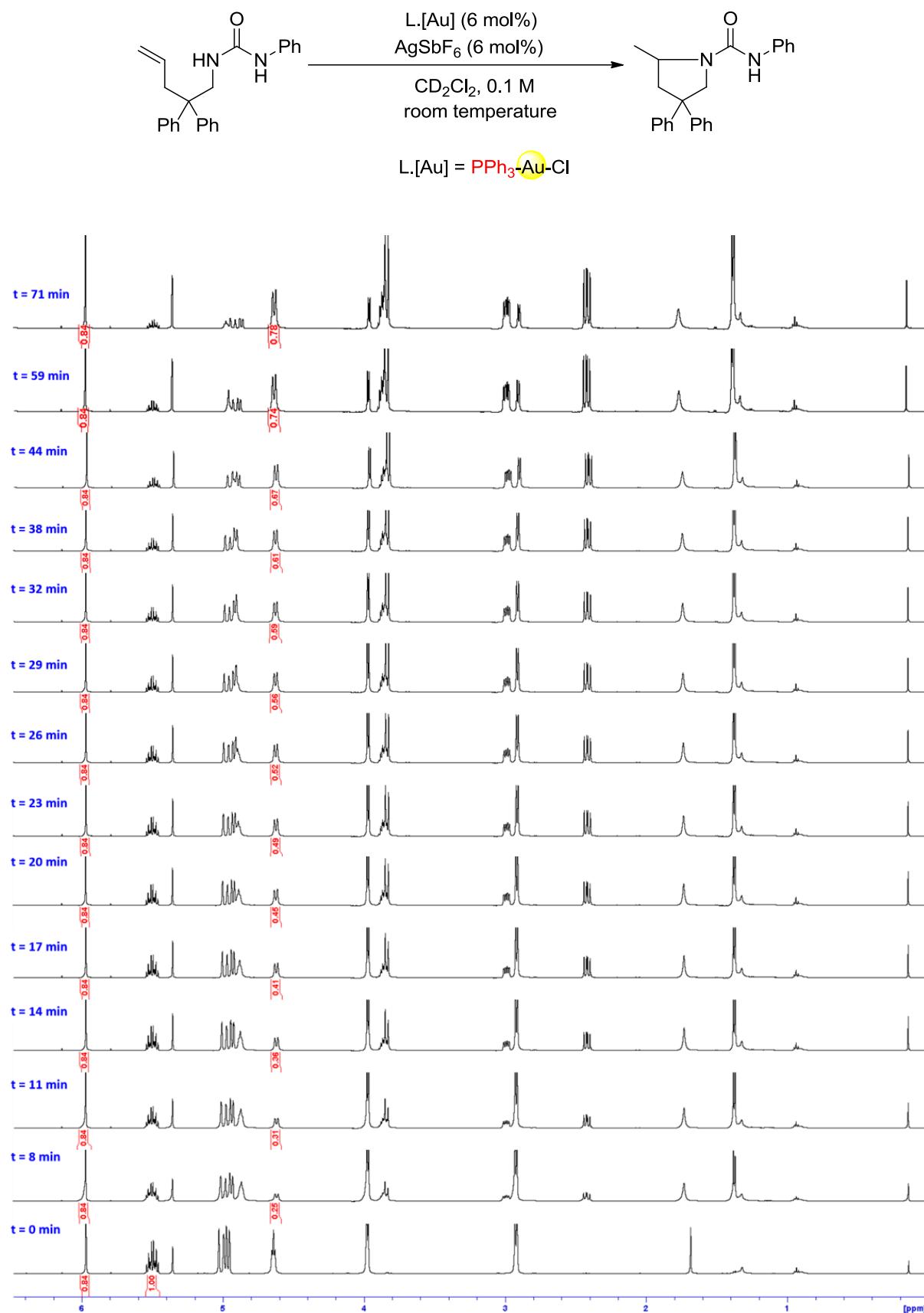


**Figure SI4.** Left: Absorption spectra of **E-6** in MeCN at  $25^\circ\text{C}$  before (black line) and after irradiation (red line) at  $320 \text{ nm}$  (6 min at  $4 \text{ mW.cm}^{-2}$ ), and its evolution upon resting in the dark at  $60^\circ\text{C}$  (red to violet lines). Right: Evolution of absorption at  $335 \text{ nm}$  as a function of time in the dark at  $60^\circ\text{C}$ .

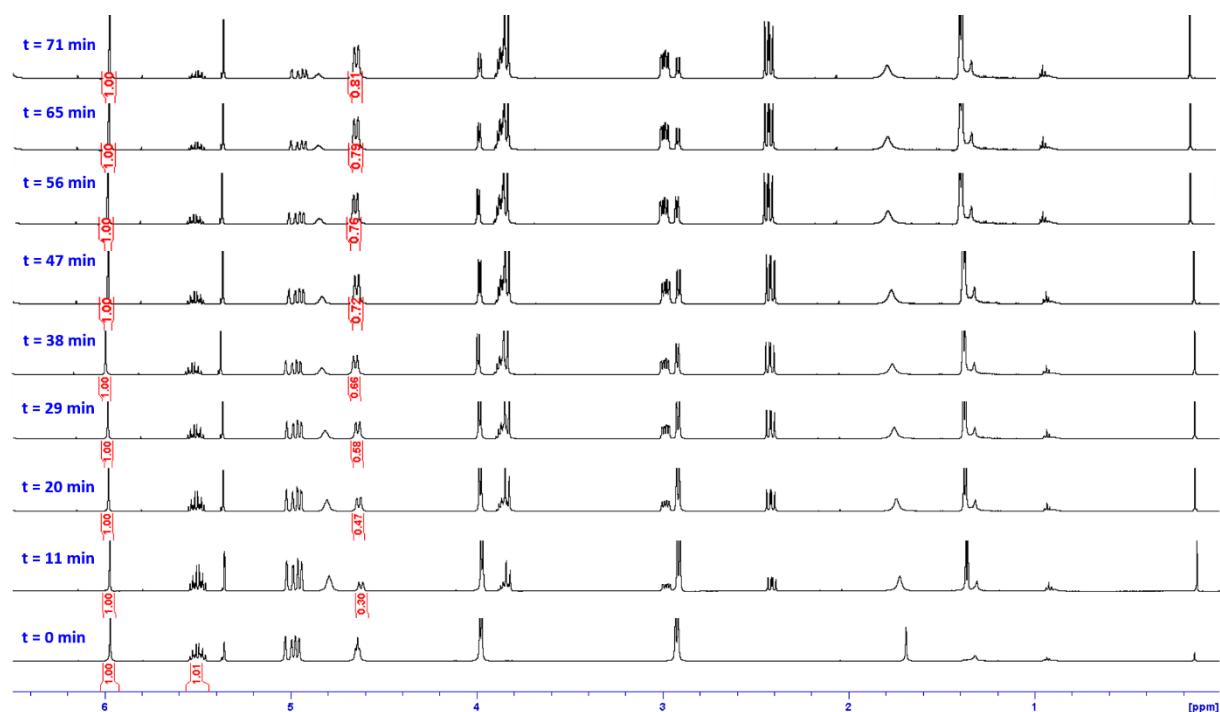
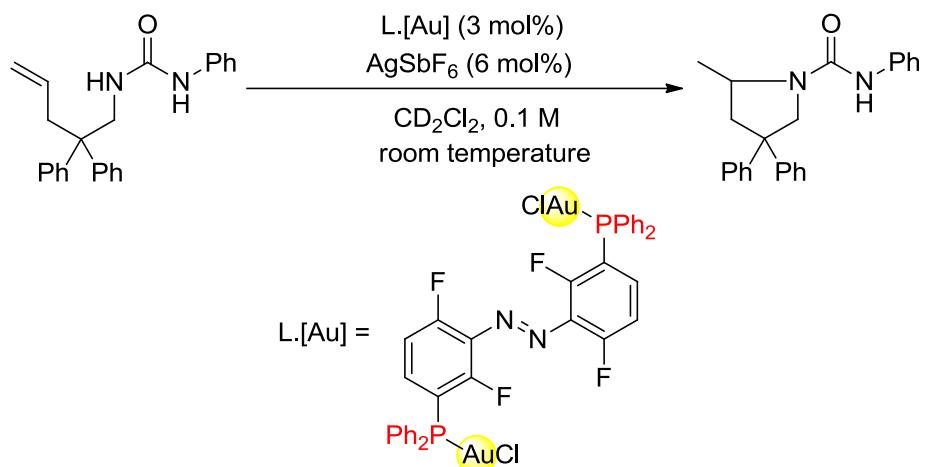
## **V. Intramolecular hydroamination of *N*-alkenyl ureas **4**.**

We have investigated the comparative rate of reaction for the hydroamination of *N*-alkenyl urea **4** with complexes **E-3**, **Z-3**, **E-6**, **Z-6** and Ph<sub>3</sub>PAuCl. A careful weighed sample of *N*-alkenyl urea (0.15 mmol), gold complex and internal standard were dissolved in CD<sub>2</sub>Cl<sub>2</sub> (1.5 ml). 0.5 ml of the solution was taken out and <sup>1</sup>H NMR was recorded before initiating the reaction. In order to initiate the reaction, the silver salt (6 mol% of AgSbF<sub>6</sub> with respect to the remaining 0.1 mmol of substrate) was weighed in 1 mL screw capped vial and then the 1 ml solution remained was transferred to this 1 mL screw capped vial and time was noted (t = 0). The reaction mixture was quickly mixed (2-3 min) and 0.5 ml of the reaction mixture transferred back to an NMR tube with a syringe and the reaction was monitored by automatic recording of <sup>1</sup>H NMR spectra at regular time intervals.

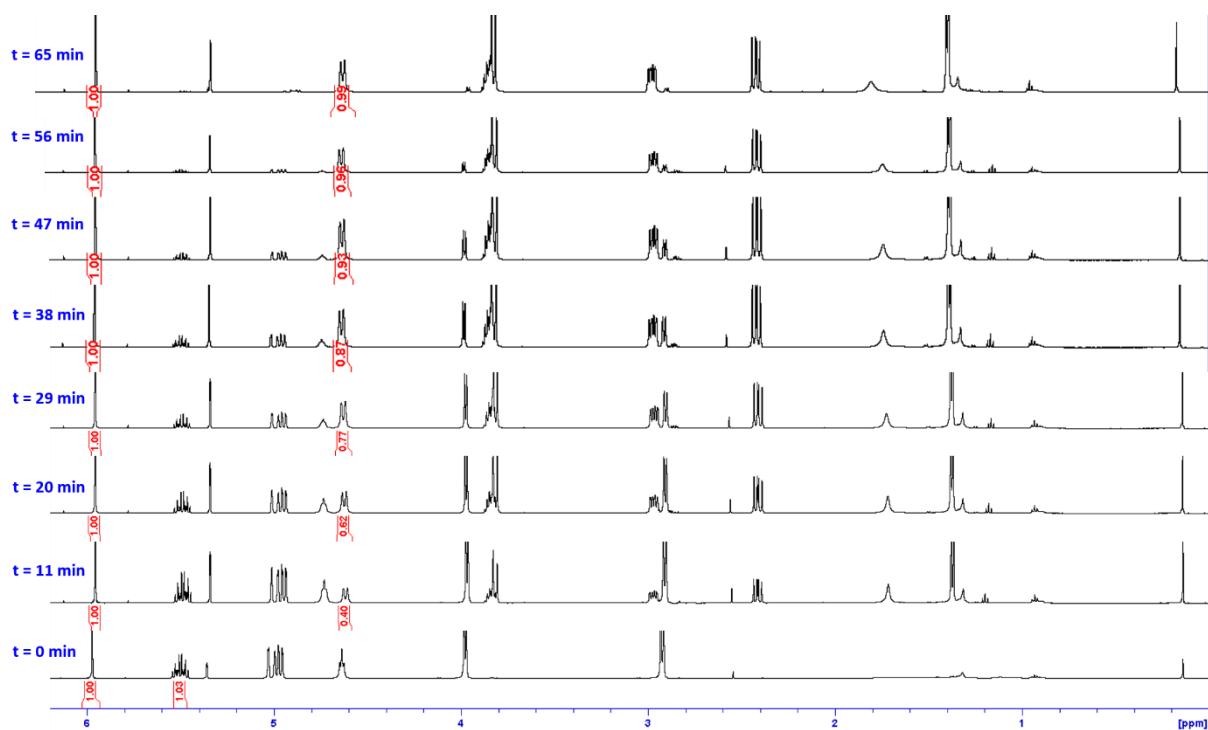
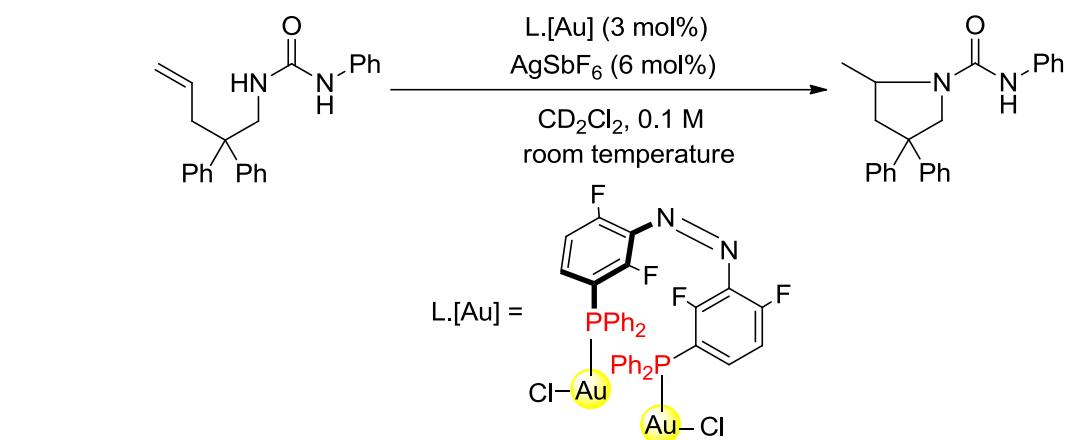
**<sup>1</sup>H NMR monitoring of the reaction with PPh<sub>3</sub>-Au-Cl**



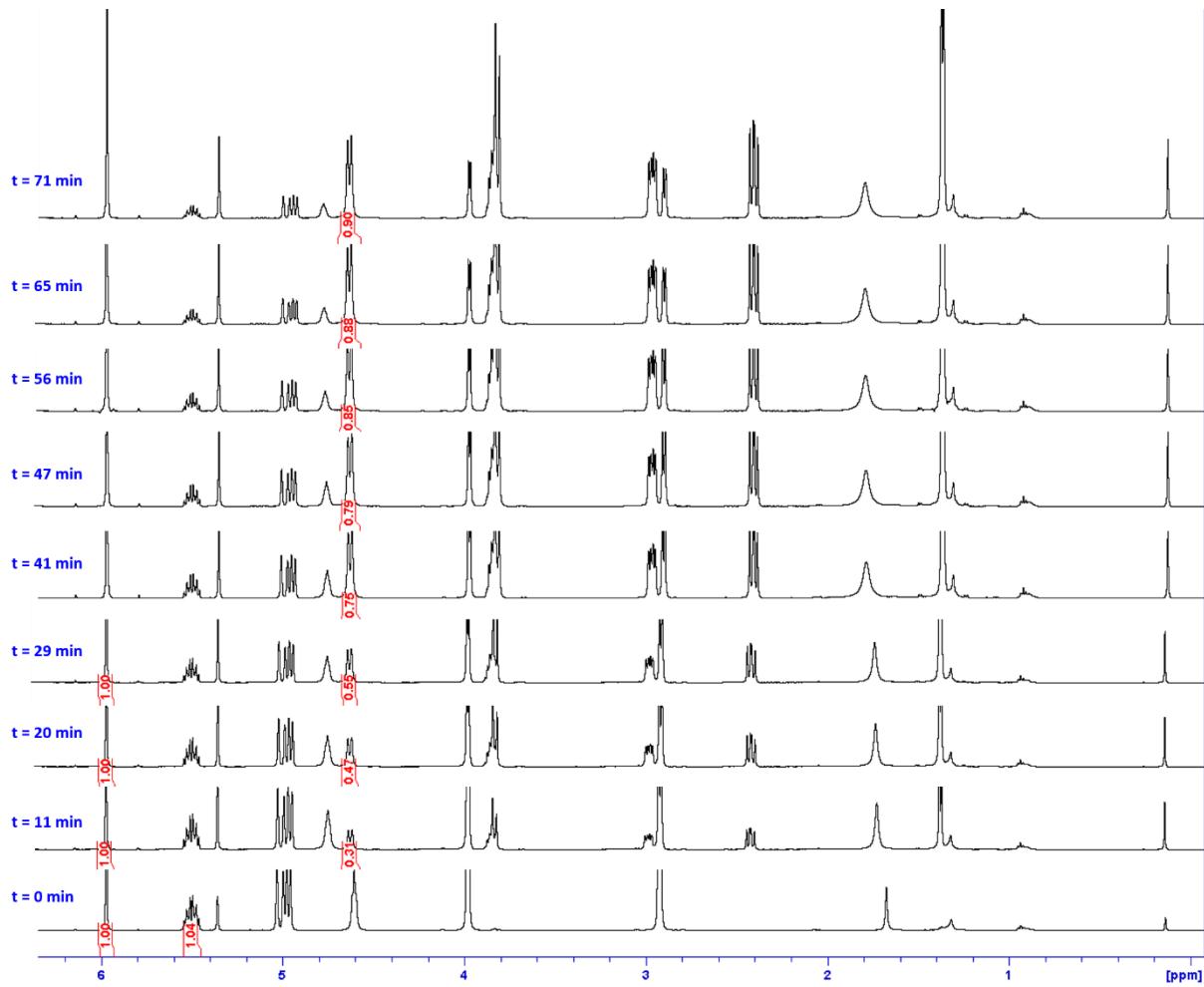
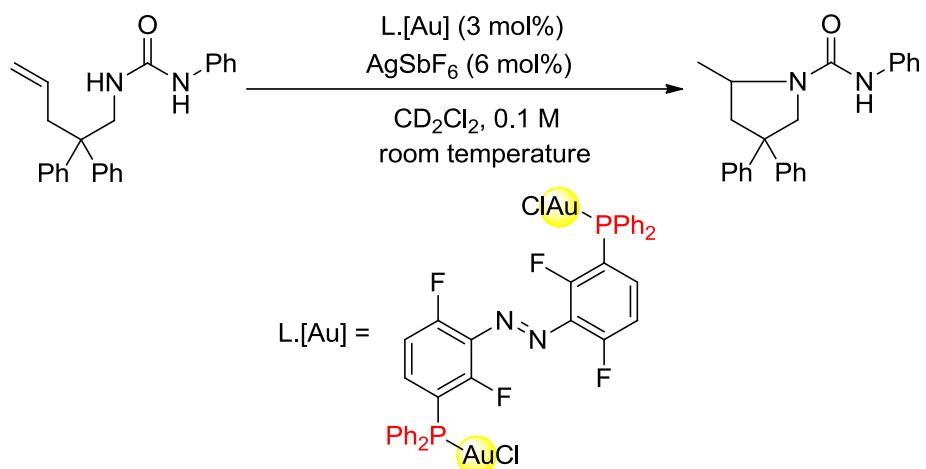
<sup>1</sup>H NMR monitoring of the reaction with *E*-3



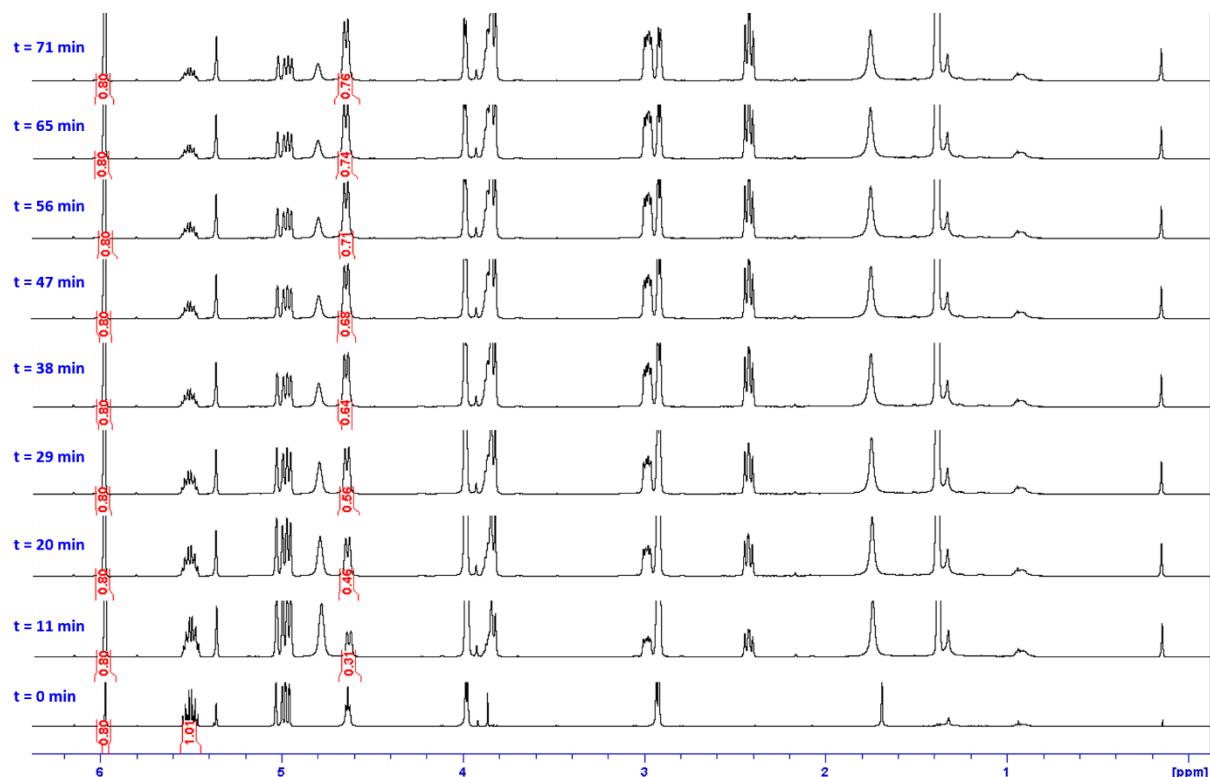
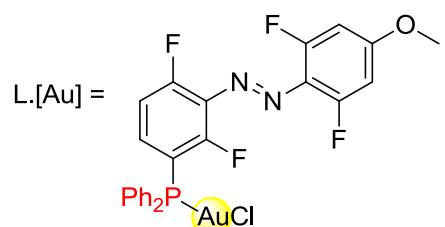
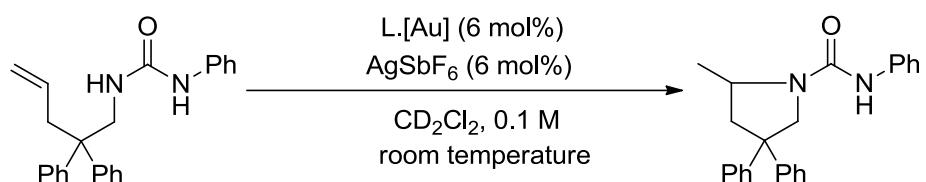
<sup>1</sup>H NMR monitoring of the reaction with Z-3



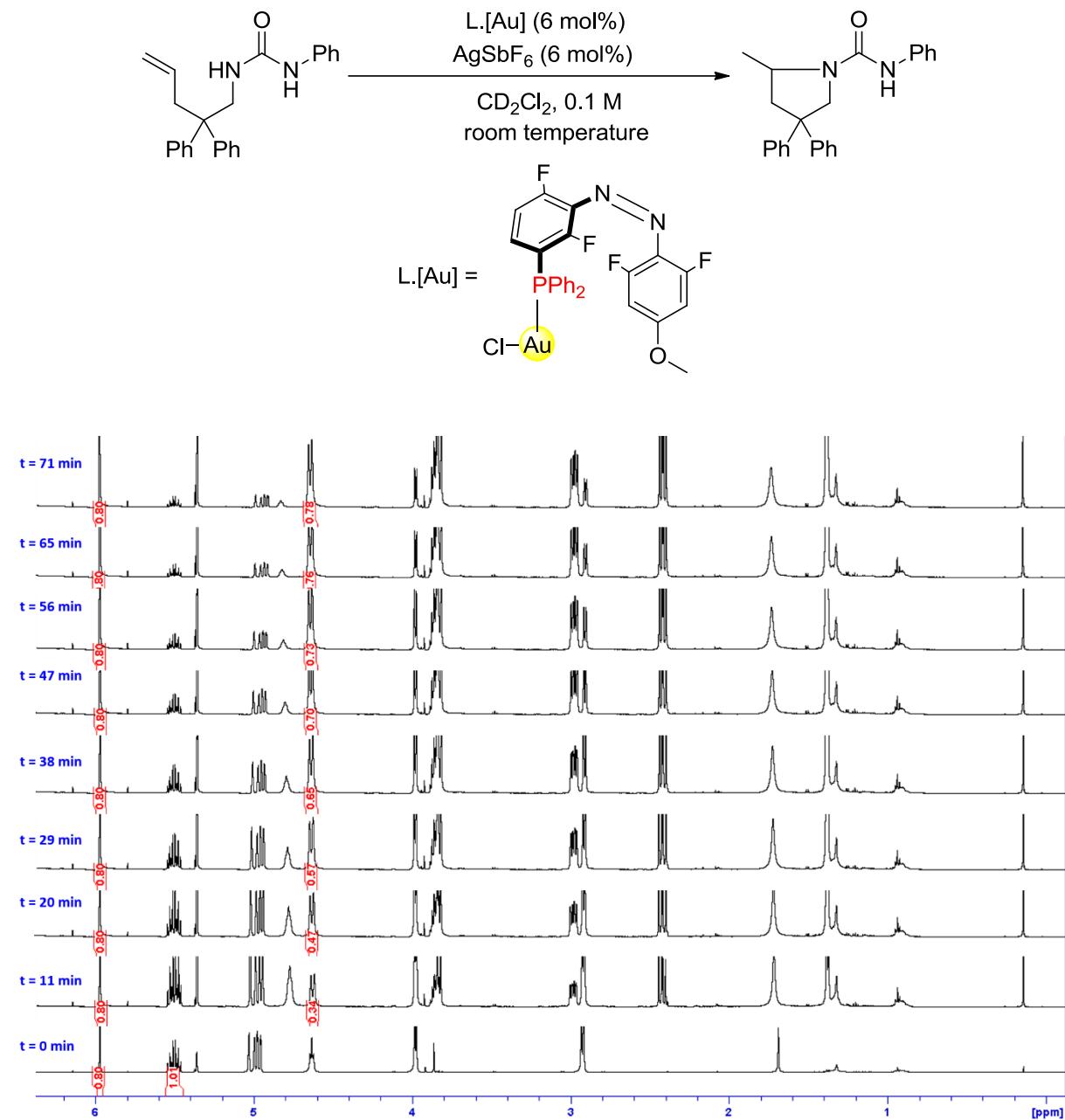
**<sup>1</sup>H NMR monitoring of the reaction with E-3: irradiation at 320 nm after 30 min reaction.**



**<sup>1</sup>H NMR monitoring of the reaction with E-6**

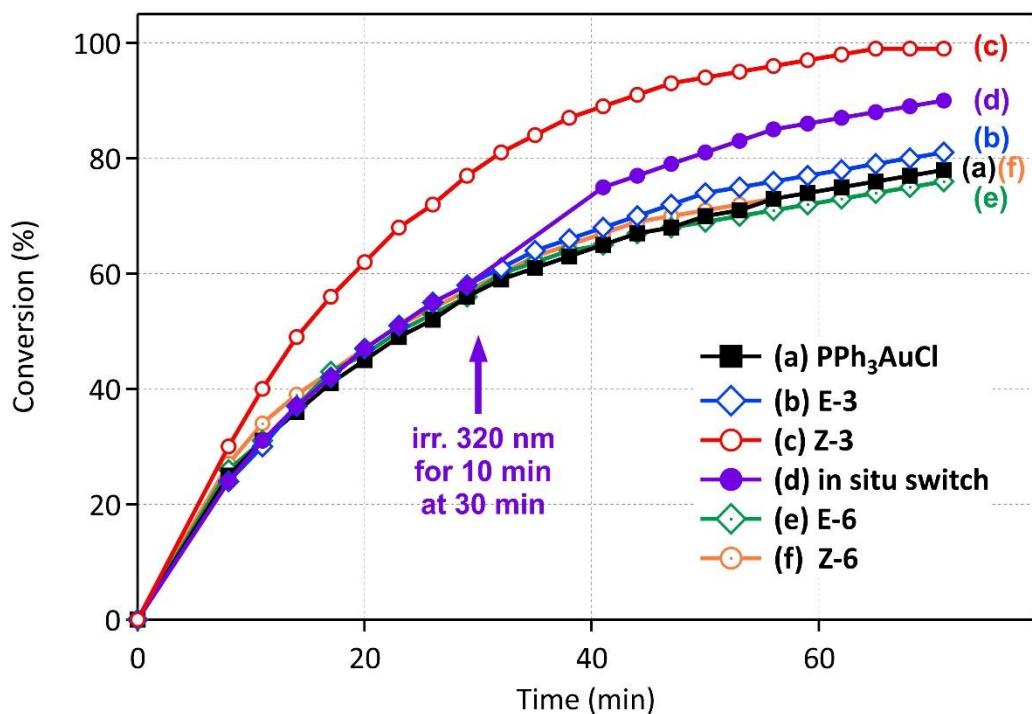


**<sup>1</sup>H NMR monitoring of the reaction with Z-6**



**Reaction profiles for the intramolecular Hydroamination of *N*-Alkenyl Ureas:**

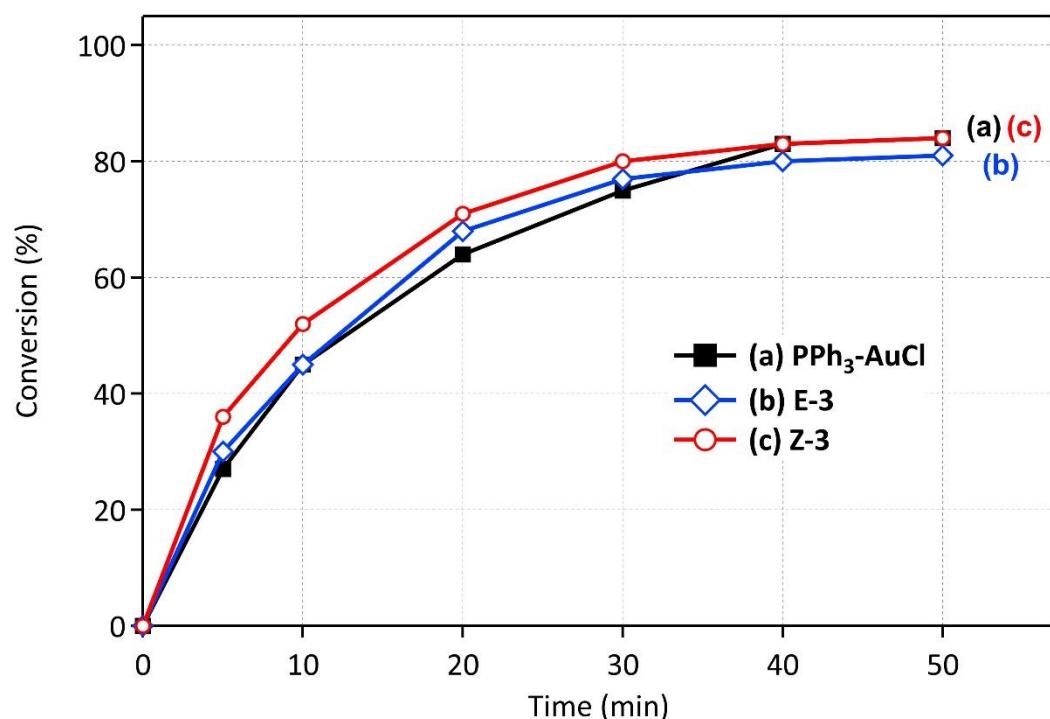
The conversion rate for the hydroamination reactions with complexes **E-3**, **Z-3**, **E-6**, **Z-6** and  $\text{Ph}_3\text{PAuCl}$  are plotted against time.



## **VI. Intramolecular [4+2] cycloadditions of 7:**

A careful weighed sample of dimethyl 2-(3-methylbut-2-enyl)-2-(3-phenylprop-2-ynyl)malonate, gold complex and internal standard was dissolved in  $\text{CD}_2\text{Cl}_2$  and proton NMR was taken. In order to initiate the reaction, the silver salt ( $\text{AgSbF}_6$ ) was weighed in a 1 mL screw capped vial and then the solution from NMR tube was transferred to this 1 mL screw capped vial and time was noted. The reaction mixture was quickly mixed and transferred back to the NMR tube with a syringe. NMR spectra were recorded at the given time intervals.

The NMR conversion rates are plotted against the time.



## **VII. X-Ray crystal structure determination**

Single crystals suitable for X-ray diffraction were obtained by recrystallization from CHCl<sub>3</sub>-CH<sub>2</sub>Cl<sub>2</sub>-heptane mixture. X-ray crystallographic data were collected on a Rigaku XtaLabPro diffractometer equipped with a microfocus source (MicroMax003\_Mo) and multilayer confocal mirrors (Mo K $\alpha$  radiation,  $\lambda = 0.71075 \text{ \AA}$ ) using  $\omega$ -scans. Data were indexed, integrated and scaled using *d\*TREK* from the *CrystalClear*<sup>4</sup> software suite (*E-3*) or *CrysAlisPRO*<sup>5</sup> (*Z-3, E-6* and *Z-6*). They were also corrected for polarization, Lorentz and absorption effects (*REQAB* (*E-3*) or *CrysAlisPRO* (*Z-3, E-6* and *Z-6*)). For each complex, the structure was solved by direct methods with *SHELXT*<sup>3</sup>, and refined with *SHELXL-2014/7*<sup>6</sup>. The model was refined using full-matrix least-squares, all non-hydrogen atoms were refined with anisotropic displacement parameters. H atoms have been added geometrically and treated as riding on their parent atoms ( $U_H = 1.2 U_C$ ).

For *E-3*, data quality was not so good despite suitable redundancy (ca. 4). This may be due to poor crystals quality and/or twin crystals. Attempts to detwin data using *CrysAlisPRO* were unsuccessful and lead to high *R*<sub>int</sub>, so a TWIN refinement was done (scales: 0.9454(12), 0.0546(12)). Restraints were applied along the entire connectivity set, using RIGU command (with standard deviation values), leading to more reasonable anisotropic displacement parameters. CHCl<sub>3</sub> and DCM molecules were refined with free occupancy factors using FVAR variables (CHCl<sub>3</sub> occupancy factors: 0.77(2) and 0.64(2), DCM occupancy factor: 0.44(4)).

For *Z-3*, Restraints were applied on CHCl<sub>3</sub> molecule, using RIGU command (with standard deviation values), leading to more reasonable anisotropic displacement parameters. CHCl<sub>3</sub> molecule was refined over two positions using PART command and its occupancy factor fixed at 0.4 for each position.

For *E-6*, Phenyl rings (C1-C6, C7-C12, C26-C31 and C32-37) were refined as rigid bodies (AFIX 66, C-C distance of 1.39  $\text{\AA}$  and bond angles of 120°). Due to a disorder on N=N bond, we refined N atoms over two positions with restraints (DFIX for C-N and N=N bond length, distances determined from Mogul database (Cambridge Structural Database)) and using EADP command to equate their anisotropic displacement parameters.

Molecular graphics were computed with Ortep-3<sup>7</sup>. CCDC 1484587-1484590 contains the supplementary crystallographic data for this paper. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre via [www.ccdc.cam.ac.uk/data\\_request/cif](http://www.ccdc.cam.ac.uk/data_request/cif).

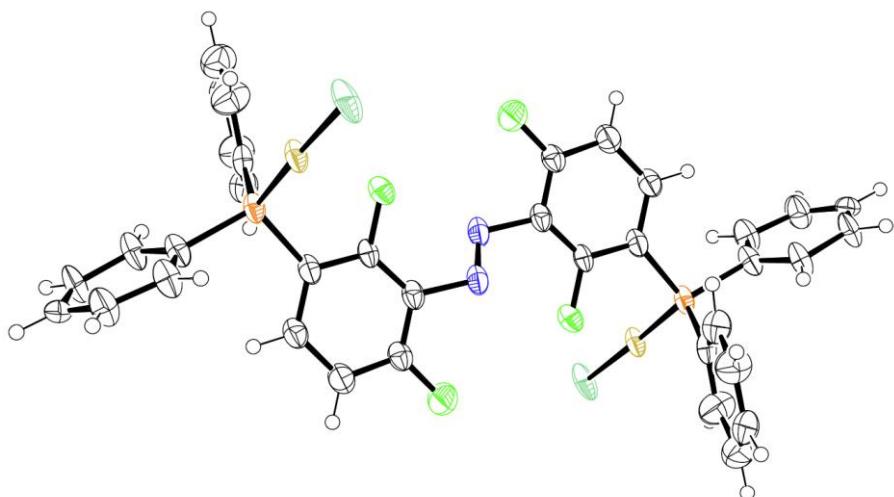
<sup>4</sup> Rigaku, CrystalClear-SM Expert 2.1 b43. (2015).

<sup>5</sup> Rigaku Oxford Diffraction, CrysAlisPro Software system, version 38.41, Rigaku Corporation, Oxford, UK. (2015).

<sup>6</sup> Sheldrick, G. M. *SHELXT* – Integrated space-group and crystal-structure determination. *Acta Crystallogr. Sect. Found. Adv.* **2015**, *71*, 3–8.

<sup>7</sup> Farrugia, L. J. *ORTEP* -3 for Windows - a version of *ORTEP* -III with a Graphical User Interface (GUI). *J. Appl. Crystallogr.* **1997**, *30*, 565–565.

Crystal Data for E-3 :



**Figure SI5 : Ortep view of the complex (Color scheme: C, gray; N, blue; F, green; Au, brown; Cl, pale green; P, orange)**

$C_{37.85}H_{26.29}Au_2Cl_{7.115}F_4N_2P_2$  ( $M=1293.03$  g/mol): triclinic, space group P-1 (no. 2),  $a = 11.051(3)$  Å,  $b = 15.816(4)$  Å,  $c = 16.857(5)$  Å,  $\alpha = 109.459(3)^\circ$ ,  $\beta = 103.933(5)^\circ$ ,  $\gamma = 104.747(5)^\circ$ ,  $V = 2509.4(12)$  Å $^3$ ,  $Z = 2$ ,  $T = 130.0$  K,  $\mu(\text{MoK}\alpha) = 6.325$  mm $^{-1}$ ,  $D_{\text{calc}} = 1.713$  g/cm $^3$ , 46126 reflections measured ( $7.232^\circ \leq 2\Theta \leq 50.702^\circ$ ), 9151 unique ( $R_{\text{int}} = 0.1024$ ,  $R_{\text{sigma}} = 0.0966$ ) which were used in all calculations. The final  $R_1$  was 0.1038 ( $I > 2\sigma(I)$ ) and  $wR_2$  was 0.2689 (all data).

**Table 1 Crystal data and structure refinement for E-3**

CCDC number	1484589
Empirical formula	$C_{37.85}H_{26.29}Au_2Cl_{7.12}F_4N_2P_2$
Formula weight	1293.03
Temperature/K	130.0
Crystal system	triclinic
Space group	P-1
$a/\text{\AA}$	11.051(3)
$b/\text{\AA}$	15.816(4)
$c/\text{\AA}$	16.857(5)
$\alpha/^\circ$	109.459(3)
$\beta/^\circ$	103.933(5)
$\gamma/^\circ$	104.747(5)
Volume/Å $^3$	2509.4(12)
Z	2
$\rho_{\text{calc}}/\text{cm}^3$	1.713
$\mu/\text{mm}^{-1}$	6.325
F(000)	1224.5
Crystal size/mm $^3$	0.2 × 0.09 × 0.07
Radiation	MoK $\alpha$ ( $\lambda = 0.71075$ )
2 $\Theta$ range for data collection/°	7.232 to 50.702
Index ranges	-14 ≤ h ≤ 14, -20 ≤ k ≤ 19, -21 ≤ l ≤ 21
Reflections collected	46126
Independent reflections	9151 [ $R_{\text{int}} = 0.1024$ , $R_{\text{sigma}} = 0.0966$ ]
Data/restraints/parameters	9151/423/533
Goodness-of-fit on $F^2$	1.119
Final R indexes [I>=2σ (I)]	$R_1 = 0.1038$ , $wR_2 = 0.2582$
Final R indexes [all data]	$R_1 = 0.1239$ , $wR_2 = 0.2689$
Largest diff. peak/hole / e Å $^{-3}$	5.21/-2.74

**Table 2 Fractional Atomic Coordinates and Equivalent Isotropic Displacement Parameters for E-3.**  
**U<sub>eq</sub> is defined as 1/3 of the trace of the orthogonalised U<sub>ij</sub> tensor.**

Atom	x	y	z	U(eq)
C1	0.473(2)	0.2100(18)	0.7931(16)	0.038(4)
C2	0.495(3)	0.297(2)	0.860(2)	0.061(7)
C3	0.562(3)	0.388(2)	0.860(2)	0.061(6)
C4	0.596(2)	0.380(2)	0.7851(19)	0.050(5)
C5	0.571(3)	0.294(2)	0.717(2)	0.059(6)
C6	0.517(2)	0.211(2)	0.7215(17)	0.046(5)
C7	0.655(2)	0.099(2)	0.8426(18)	0.047(6)
C8	0.531(2)	0.0584(19)	0.8432(16)	0.039(4)
C9	0.500(3)	-0.010(2)	0.8787(18)	0.049(6)
C10	0.611(3)	-0.031(2)	0.915(2)	0.053(6)
C11	0.733(2)	0.0010(16)	0.9162(15)	0.036(4)
C12	0.755(2)	0.066(2)	0.8767(19)	0.050(6)
C13	0.302(2)	0.0092(17)	0.6857(15)	0.035(4)
C14	0.1972(18)	0.0204(15)	0.6336(13)	0.026(3)
C15	0.114(2)	-0.0462(17)	0.5487(14)	0.032(4)
C16	0.133(2)	-0.1310(17)	0.5162(14)	0.033(4)
C17	0.239(2)	-0.1490(19)	0.5636(16)	0.043(5)
C18	0.325(2)	-0.0753(17)	0.6485(15)	0.037(4)
C19	0.748(2)	-0.0912(16)	0.6345(15)	0.035(4)
C20	0.639(2)	-0.0768(16)	0.5801(15)	0.033(4)
C21	0.645(2)	0.0127(17)	0.5838(16)	0.037(4)
C22	0.761(2)	0.0896(17)	0.6430(16)	0.036(4)
C23	0.873(2)	0.0826(18)	0.6970(17)	0.042(5)
C24	0.859(3)	-0.0100(19)	0.6938(19)	0.051(6)
C25	0.701(2)	-0.2913(17)	0.5207(16)	0.037(4)
C26	0.748(2)	-0.253(2)	0.4678(18)	0.046(5)
C27	0.738(3)	-0.318(2)	0.3826(19)	0.053(5)
C28	0.684(3)	-0.410(2)	0.360(2)	0.058(6)
C29	0.633(4)	-0.447(2)	0.416(2)	0.070(8)
C30	0.643(3)	-0.3866(19)	0.496(2)	0.052(5)
C31	0.871(2)	-0.1937(17)	0.7124(16)	0.036(4)
C32	0.880(2)	-0.184(2)	0.7997(18)	0.050(6)
C33	0.999(2)	-0.171(2)	0.8618(19)	0.051(6)
C34	1.112(3)	-0.165(2)	0.8351(19)	0.056(7)
C35	1.106(2)	-0.1817(19)	0.7509(17)	0.042(5)
C36	0.983(2)	-0.195(2)	0.6860(18)	0.046(6)
C37	0.146(6)	-0.457(4)	0.437(5)	0.087(10)
C38	0.411(4)	-0.340(3)	0.871(3)	0.066(7)
Au1	0.25512(8)	0.10393(7)	0.87630(6)	0.0366(3)
Au2	0.53296(9)	-0.25386(7)	0.66724(7)	0.0428(3)
Cl1	0.1039(6)	0.1063(6)	0.9471(5)	0.0548(17)
Cl2	0.3330(7)	-0.3002(6)	0.6871(6)	0.062(2)
Cl3	0.2851(18)	-0.4199(13)	0.4250(11)	0.101(6)
Cl4	0.0614(15)	-0.5810(11)	0.3885(15)	0.107(7)
Cl5	0.033(2)	-0.4039(15)	0.4169(18)	0.145(10)
Cl7	0.5784(13)	-0.3214(12)	0.8871(9)	0.099(5)
Cl8	0.3841(11)	-0.2454(9)	0.9453(7)	0.073(4)
Cl6	0.3521(16)	-0.4432(9)	0.8884(9)	0.092(4)
F1	0.1736(12)	0.1032(9)	0.6727(8)	0.036(3)
F2	0.0511(15)	-0.2017(11)	0.4358(9)	0.050(3)
F3	0.7651(15)	0.1782(10)	0.6512(10)	0.051(3)
F4	0.5301(13)	-0.1533(10)	0.5256(9)	0.044(3)
N1	-0.0094(18)	-0.0425(14)	0.4956(13)	0.036(4)
N2	0.5388(18)	0.0337(13)	0.5386(12)	0.034(4)
P1	0.3956(5)	0.0990(5)	0.8013(4)	0.0320(12)
P2	0.7154(6)	-0.2086(4)	0.6338(4)	0.0337(12)
Cl9	0.917(3)	0.321(3)	0.879(3)	0.154(18)

<b>C110</b>	1.166(3)	0.325(3)	0.846(4)	0.170(19)
<b>C39</b>	1.086(4)	0.336(6)	0.925(3)	0.13(2)

**Table 3 Anisotropic Displacement Parameters for E-3**

Atom	U <sub>11</sub>	U <sub>22</sub>	U <sub>33</sub>	U <sub>23</sub>	U <sub>13</sub>	U <sub>12</sub>
<b>C1</b>	0.022(8)	0.051(6)	0.039(8)	0.021(5)	0.004(6)	0.014(6)
<b>C2</b>	0.064(16)	0.052(7)	0.057(10)	0.016(6)	0.026(10)	0.011(7)
<b>C3</b>	0.055(15)	0.053(9)	0.065(11)	0.018(7)	0.021(10)	0.011(8)
<b>C4</b>	0.029(11)	0.048(9)	0.064(10)	0.020(7)	0.015(9)	0.005(8)
<b>C5</b>	0.061(15)	0.056(8)	0.062(11)	0.028(7)	0.026(10)	0.018(8)
<b>C6</b>	0.040(11)	0.053(9)	0.045(9)	0.023(7)	0.014(8)	0.013(8)
<b>C7</b>	0.033(7)	0.075(13)	0.057(14)	0.041(12)	0.023(7)	0.034(8)
<b>C8</b>	0.034(6)	0.059(10)	0.042(10)	0.028(9)	0.018(6)	0.029(6)
<b>C9</b>	0.041(8)	0.068(12)	0.055(14)	0.038(12)	0.019(8)	0.030(8)
<b>C10</b>	0.044(8)	0.064(14)	0.066(15)	0.039(13)	0.017(8)	0.029(8)
<b>C11</b>	0.039(8)	0.031(9)	0.031(10)	0.004(8)	0.008(7)	0.019(7)
<b>C12</b>	0.032(8)	0.069(13)	0.070(15)	0.042(13)	0.020(8)	0.033(8)
<b>C13</b>	0.027(7)	0.047(7)	0.035(5)	0.019(5)	0.010(5)	0.019(5)
<b>C14</b>	0.018(6)	0.041(6)	0.030(5)	0.021(5)	0.016(4)	0.013(5)
<b>C15</b>	0.024(7)	0.047(7)	0.031(6)	0.018(5)	0.013(5)	0.016(5)
<b>C16</b>	0.030(7)	0.047(7)	0.031(6)	0.019(5)	0.016(5)	0.019(6)
<b>C17</b>	0.039(8)	0.052(9)	0.040(7)	0.017(6)	0.012(6)	0.024(7)
<b>C18</b>	0.032(8)	0.051(7)	0.035(7)	0.019(6)	0.014(6)	0.022(6)
<b>C19</b>	0.035(6)	0.042(6)	0.040(8)	0.024(5)	0.021(5)	0.019(5)
<b>C20</b>	0.034(6)	0.041(6)	0.038(8)	0.022(5)	0.020(6)	0.020(5)
<b>C21</b>	0.034(6)	0.041(6)	0.048(9)	0.025(5)	0.020(6)	0.021(5)
<b>C22</b>	0.034(7)	0.044(6)	0.044(8)	0.025(6)	0.022(6)	0.018(5)
<b>C23</b>	0.034(8)	0.046(8)	0.054(10)	0.028(7)	0.017(7)	0.016(6)
<b>C24</b>	0.045(8)	0.048(7)	0.057(11)	0.025(6)	0.008(8)	0.017(6)
<b>C25</b>	0.018(8)	0.048(7)	0.049(6)	0.019(5)	0.013(6)	0.020(6)
<b>C26</b>	0.039(11)	0.056(9)	0.055(8)	0.025(7)	0.023(8)	0.029(8)
<b>C27</b>	0.044(13)	0.059(9)	0.060(9)	0.020(7)	0.027(9)	0.025(9)
<b>C28</b>	0.054(14)	0.055(9)	0.065(10)	0.018(8)	0.026(10)	0.025(9)
<b>C29</b>	0.08(2)	0.048(9)	0.073(10)	0.016(7)	0.040(11)	0.019(9)
<b>C30</b>	0.050(12)	0.047(7)	0.063(9)	0.021(6)	0.025(9)	0.024(7)
<b>C31</b>	0.029(6)	0.043(11)	0.046(7)	0.022(7)	0.017(5)	0.021(6)
<b>C32</b>	0.029(8)	0.082(19)	0.049(7)	0.031(9)	0.019(6)	0.028(9)
<b>C33</b>	0.032(8)	0.080(18)	0.051(8)	0.033(10)	0.019(6)	0.027(9)
<b>C34</b>	0.032(8)	0.10(2)	0.052(9)	0.039(10)	0.020(7)	0.034(10)
<b>C35</b>	0.032(8)	0.063(15)	0.053(8)	0.033(9)	0.021(6)	0.030(8)
<b>C36</b>	0.033(7)	0.074(16)	0.052(9)	0.035(9)	0.023(6)	0.035(8)
<b>C37</b>	0.093(13)	0.063(13)	0.11(2)	0.055(13)	0.019(12)	0.020(9)
<b>C38</b>	0.067(12)	0.070(11)	0.067(14)	0.029(9)	0.027(10)	0.032(9)
<b>Au1</b>	0.0247(4)	0.0584(6)	0.0340(5)	0.0210(4)	0.0127(4)	0.0224(4)
<b>Au2</b>	0.0339(5)	0.0513(6)	0.0704(7)	0.0403(5)	0.0312(5)	0.0262(4)
<b>Cl1</b>	0.037(3)	0.098(5)	0.063(4)	0.047(4)	0.036(3)	0.044(4)
<b>Cl2</b>	0.044(3)	0.073(5)	0.108(6)	0.057(5)	0.051(4)	0.034(3)
<b>Cl3</b>	0.094(10)	0.096(11)	0.081(10)	0.041(9)	0.010(8)	-0.001(7)
<b>Cl4</b>	0.066(9)	0.062(8)	0.175(18)	0.051(9)	0.015(9)	0.021(6)
<b>Cl5</b>	0.123(13)	0.090(12)	0.18(2)	0.043(12)	-0.007(12)	0.050(10)
<b>Cl7</b>	0.076(7)	0.150(12)	0.088(9)	0.055(9)	0.038(6)	0.055(7)
<b>Cl8</b>	0.071(7)	0.084(7)	0.062(6)	0.026(5)	0.016(5)	0.040(6)
<b>Cl6</b>	0.129(11)	0.072(7)	0.085(8)	0.036(6)	0.044(8)	0.040(7)
<b>F1</b>	0.034(6)	0.046(6)	0.039(6)	0.021(5)	0.019(5)	0.022(5)
<b>F2</b>	0.050(7)	0.058(7)	0.035(5)	0.013(5)	0.008(5)	0.022(6)
<b>F3</b>	0.051(8)	0.043(6)	0.063(9)	0.027(6)	0.020(7)	0.019(5)
<b>F4</b>	0.037(6)	0.043(6)	0.054(7)	0.021(5)	0.014(5)	0.020(5)
<b>N1</b>	0.025(7)	0.047(10)	0.036(8)	0.018(8)	0.007(6)	0.016(7)

<b>N2</b>	0.034(7)	0.046(10)	0.042(9)	0.028(8)	0.025(6)	0.024(7)
<b>P1</b>	0.022(2)	0.049(3)	0.031(3)	0.020(2)	0.012(2)	0.017(2)
<b>P2</b>	0.028(3)	0.040(3)	0.044(3)	0.022(3)	0.017(2)	0.019(2)
<b>Cl9</b>	0.14(2)	0.19(4)	0.15(3)	0.08(3)	0.05(2)	0.06(2)
<b>Cl10</b>	0.16(3)	0.12(3)	0.26(4)	0.10(3)	0.09(3)	0.04(2)
<b>C39</b>	0.12(3)	0.06(4)	0.22(4)	0.09(3)	0.05(2)	0.03(3)

**Table 4 Bond Lengths for E-3**

Atom	Atom	Length/Å	Atom	Atom	Length/Å
C1	C2	1.37(4)	C22	C23	1.40(3)
C1	C6	1.41(3)	C22	F3	1.35(3)
C1	P1	1.81(2)	C23	C24	1.42(3)
C2	C3	1.44(4)	C25	C26	1.36(3)
C3	C4	1.38(4)	C25	C30	1.35(4)
C4	C5	1.35(4)	C25	P2	1.86(2)
C5	C6	1.33(4)	C26	C27	1.42(4)
C7	C8	1.37(3)	C27	C28	1.30(4)
C7	C12	1.41(3)	C28	C29	1.41(4)
C8	C9	1.40(3)	C29	C30	1.32(4)
C8	P1	1.85(2)	C31	C32	1.41(3)
C9	C10	1.42(3)	C31	C36	1.41(3)
C10	C11	1.30(4)	C31	P2	1.81(2)
C11	C12	1.41(3)	C32	C33	1.38(3)
C13	C14	1.37(3)	C33	C34	1.41(3)
C13	C18	1.39(3)	C34	C35	1.34(4)
C13	P1	1.82(2)	C35	C36	1.44(3)
C14	C15	1.37(3)	C37	Cl3	1.58(7)
C14	F1	1.38(2)	C37	Cl4	1.73(6)
C15	C16	1.36(3)	C37	Cl5	1.71(6)
C15	N1	1.47(3)	C38	Cl7	1.73(4)
C16	C17	1.40(3)	C38	Cl8	1.75(4)
C16	F2	1.34(3)	C38	Cl6	1.74(4)
C17	C18	1.41(3)	Au1	Cl1	2.276(6)
C19	C20	1.44(3)	Au1	P1	2.226(5)
C19	C24	1.38(4)	Au2	Cl2	2.287(6)
C19	P2	1.79(2)	Au2	P2	2.231(6)
C20	C21	1.38(3)	N1	N1 <sup>1</sup>	1.26(4)
C20	F4	1.32(3)	N2	N2 <sup>2</sup>	1.28(4)
C21	C22	1.38(3)	Cl9	C39	1.7603
C21	N2	1.42(3)	Cl10	C39	1.7599

<sup>1</sup>-X,-Y,1-Z; <sup>2</sup>1-X,-Y,1-Z

**Table 5 Bond Angles for E-3**

Atom	Atom	Atom	Angle/°	Atom	Atom	Atom	Angle/°
C2	C1	C6	118(2)	C22	C23	C24	116(2)
C2	C1	P1	120(2)	C19	C24	C23	122(2)
C6	C1	P1	122.4(19)	C26	C25	P2	118.8(19)
C1	C2	C3	123(3)	C30	C25	C26	124(3)
C4	C3	C2	113(3)	C30	C25	P2	118(2)
C5	C4	C3	124(3)	C25	C26	C27	117(3)
C6	C5	C4	121(3)	C28	C27	C26	118(3)
C5	C6	C1	120(3)	C27	C28	C29	123(3)
C8	C7	C12	117(2)	C30	C29	C28	119(3)
C7	C8	C9	124(2)	C29	C30	C25	119(3)
C7	C8	P1	119.2(18)	C32	C31	C36	120(2)
C9	C8	P1	117.0(17)	C32	C31	P2	119.7(17)

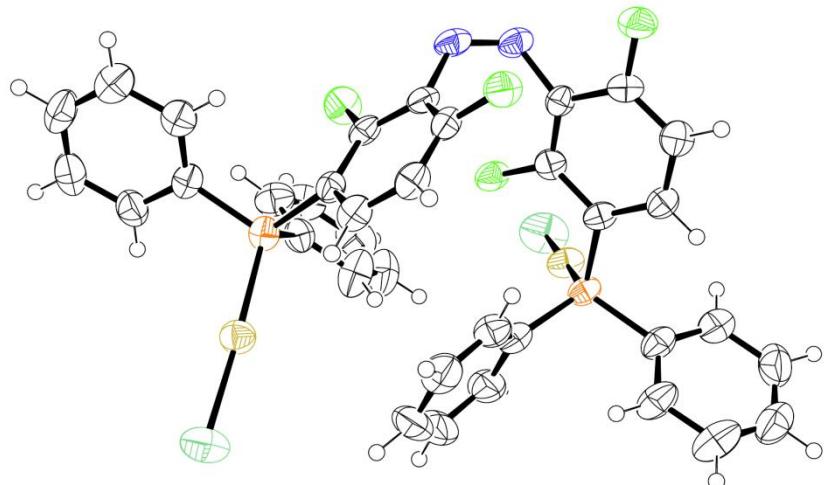
C8	C9	C10	113(2)	C36	C31	P2	120.7(18)
C11	C10	C9	128(3)	C33	C32	C31	121(2)
C10	C11	C12	115(2)	C32	C33	C34	118(2)
C11	C12	C7	122(2)	C35	C34	C33	124(2)
C14	C13	C18	117(2)	C34	C35	C36	119(2)
C14	C13	P1	119.3(17)	C31	C36	C35	119(2)
C18	C13	P1	123.6(16)	C13	C37	Cl4	117(4)
C13	C14	F1	116.1(18)	C13	C37	Cl5	119(3)
C15	C14	C13	124(2)	C15	C37	Cl4	108(3)
C15	C14	F1	119.7(17)	C17	C38	Cl8	114(2)
C14	C15	N1	125.5(19)	C17	C38	Cl6	105(2)
C16	C15	C14	117.8(19)	C16	C38	Cl8	109(2)
C16	C15	N1	115.7(19)	P1	Au1	Cl1	177.5(2)
C15	C16	C17	122(2)	P2	Au2	Cl2	173.7(3)
F2	C16	C15	121.3(19)	N1 <sup>1</sup>	N1	C15	110(2)
F2	C16	C17	116(2)	N2 <sup>2</sup>	N2	C21	114(2)
C16	C17	C18	117(2)	C1	P1	C8	107.5(11)
C13	C18	C17	122(2)	C1	P1	C13	105.7(11)
C20	C19	P2	117.1(17)	C1	P1	Au1	115.6(8)
C24	C19	C20	117(2)	C8	P1	Au1	114.3(8)
C24	C19	P2	124.9(18)	C13	P1	C8	104.4(11)
C21	C20	C19	123(2)	C13	P1	Au1	108.5(7)
F4	C20	C19	117.5(19)	C19	P2	C25	106.9(11)
F4	C20	C21	120.0(19)	C19	P2	C31	102.6(11)
C20	C21	N2	126(2)	C19	P2	Au2	110.6(7)
C22	C21	C20	117(2)	C25	P2	Au2	113.1(7)
C22	C21	N2	117(2)	C31	P2	C25	106.5(10)
C21	C22	C23	125(2)	C31	P2	Au2	116.2(8)
F3	C22	C21	117.6(19)	Cl10	C39	Cl9	112.9
F3	C22	C23	118(2)				

<sup>1</sup>-X,-Y,1-Z; <sup>2</sup>1-X,-Y,1-Z

Table 6 Atomic Occupancy for E-3

Atom	Occupancy	Atom	Occupancy	Atom	Occupancy
C37	0.64(2)	C38	0.77(2)	C39	0.44(4)
H37	0.64(2)	H38	0.77(2)	H39A	0.44(4)
Cl3	0.64(2)	Cl6	0.77(2)	H39B	0.44(4)
Cl4	0.64(2)	Cl7	0.77(2)	Cl9	0.44(4)
Cl5	0.64(2)	Cl8	0.77(2)	Cl10	0.44(4)

Crystal Data for Z-3 :



**Figure SI6 : Ortep view of the complex (Color scheme: C, gray; N, blue; F, green; Au, brown; Cl, pale green; P, orange)**

$C_{36.75}H_{24.75}Au_2Cl_{4.25}F_4N_2P_2$  ( $M = 1182.85$  g/mol): monoclinic, space group  $P2_1/c$  (no. 14),  $a = 18.4530(16)$  Å,  $b = 18.5065(19)$  Å,  $c = 11.1568(13)$  Å,  $\beta = 97.085(10)^\circ$ ,  $V = 3781.0(7)$  Å $^3$ ,  $Z = 4$ ,  $T = 250$  K,  $\mu(\text{MoK}\alpha) = 8.198$  mm $^{-1}$ ,  $D_{\text{calc}} = 2.078$  g/cm $^3$ , 55840 reflections measured ( $6.978^\circ \leq 2\Theta \leq 59.556^\circ$ ), 9695 unique ( $R_{\text{int}} = 0.0781$ ,  $R_{\text{sigma}} = 0.0596$ ) which were used in all calculations. The final  $R_1$  was 0.0403 ( $I > 2\sigma(I)$ ) and  $wR_2$  was 0.1115 (all data).

**Table 1 Crystal data and structure refinement for Z-3**

CCDC number	1484590
Empirical formula	$C_{36.8}H_{24.8}Au_2Cl_{4.4}F_4N_2P_2$
Formula weight	1182.85
Temperature/K	250
Crystal system	monoclinic
Space group	$P2_1/c$
$a/\text{\AA}$	18.4530(16)
$b/\text{\AA}$	18.5065(19)
$c/\text{\AA}$	11.1568(13)
$\alpha/^\circ$	90
$\beta/^\circ$	97.085(10)
$\gamma/^\circ$	90
Volume/Å $^3$	3781.0(7)
$Z$	4
$\rho_{\text{calc}}/\text{cm}^3$	2.078
$\mu/\text{mm}^{-1}$	8.198
$F(000)$	2234.0
Crystal size/mm $^3$	$0.24 \times 0.16 \times 0.11$
Radiation	$\text{MoK}\alpha (\lambda = 0.71073)$
$2\Theta$ range for data collection/°	6.978 to 59.556
Index ranges	$-25 \leq h \leq 25, -24 \leq k \leq 25, -15 \leq l \leq 15$
Reflections collected	55840
Independent reflections	9695 [ $R_{\text{int}} = 0.0781$ , $R_{\text{sigma}} = 0.0596$ ]
Data/restraints/parameters	9695/36/493
Goodness-of-fit on $F^2$	1.029
Final R indexes [I>=2σ (I)]	$R_1 = 0.0403$ , $wR_2 = 0.0991$
Final R indexes [all data]	$R_1 = 0.0727$ , $wR_2 = 0.1115$
Largest diff. peak/hole / e Å $^{-3}$	1.25/-0.97

**Table 2 Fractional Atomic Coordinates and Equivalent Isotropic Displacement Parameters for Z=3.**  
 **$U_{eq}$  is defined as 1/3 of the trace of the orthogonalised  $U_{ij}$  tensor.**

Atom	x	y	z	$U(eq)$
<b>Au1</b>	0.59832(2)	0.60511(2)	0.51272(2)	0.05450(9)
<b>Au2</b>	0.97394(2)	0.38836(2)	0.66358(2)	0.05508(9)
<b>C1</b>	0.5510(3)	0.5387(3)	0.7651(4)	0.0483(12)
<b>C2</b>	0.5449(3)	0.6070(3)	0.8148(6)	0.0608(15)
<b>C3</b>	0.4915(4)	0.6222(4)	0.8859(6)	0.0709(17)
<b>C4</b>	0.4418(4)	0.5687(4)	0.9057(5)	0.0703(18)
<b>C5</b>	0.4464(3)	0.5011(4)	0.8576(5)	0.0643(16)
<b>C6</b>	0.5022(3)	0.4850(4)	0.7881(5)	0.0573(14)
<b>C7</b>	0.7072(3)	0.5320(3)	0.7557(5)	0.0537(13)
<b>C8</b>	0.7688(3)	0.5266(4)	0.6955(6)	0.0733(18)
<b>C9</b>	0.8374(3)	0.5273(4)	0.7597(7)	0.081(2)
<b>C10</b>	0.8458(4)	0.5326(4)	0.8834(7)	0.0723(18)
<b>C11</b>	0.7863(4)	0.5380(4)	0.9426(6)	0.0718(17)
<b>C12</b>	0.7164(3)	0.5371(3)	0.8770(5)	0.0626(15)
<b>C13</b>	0.6133(3)	0.4337(3)	0.6102(4)	0.0460(11)
<b>C14</b>	0.5891(3)	0.4185(3)	0.4889(5)	0.0502(12)
<b>C15</b>	0.5903(3)	0.3503(3)	0.4435(5)	0.0510(13)
<b>C16</b>	0.6178(3)	0.2951(3)	0.5186(5)	0.0458(11)
<b>C17</b>	0.6437(3)	0.3070(3)	0.6382(4)	0.0459(11)
<b>C18</b>	0.6381(3)	0.3760(3)	0.6819(4)	0.0463(12)
<b>C19</b>	0.8457(3)	0.4457(3)	0.4410(5)	0.0503(12)
<b>C20</b>	0.8751(3)	0.5149(3)	0.4554(6)	0.0686(17)
<b>C21</b>	0.8334(4)	0.5741(4)	0.4126(8)	0.087(2)
<b>C22</b>	0.7635(4)	0.5640(4)	0.3565(8)	0.088(2)
<b>C23</b>	0.7343(4)	0.4951(4)	0.3405(6)	0.0707(17)
<b>C24</b>	0.7757(3)	0.4367(3)	0.3855(5)	0.0545(13)
<b>C25</b>	0.9554(3)	0.3549(3)	0.3599(5)	0.0531(13)
<b>C26</b>	1.0126(3)	0.3083(4)	0.3723(6)	0.0671(16)
<b>C27</b>	1.0492(4)	0.2942(4)	0.2704(8)	0.085(2)
<b>C28</b>	1.0290(4)	0.3283(4)	0.1650(6)	0.0757(19)
<b>C29</b>	0.9733(4)	0.3757(5)	0.1532(6)	0.081(2)
<b>C30</b>	0.9346(4)	0.3902(4)	0.2505(5)	0.0653(17)
<b>C31</b>	0.8485(3)	0.2909(3)	0.4949(4)	0.0443(11)
<b>C32</b>	0.8025(3)	0.2858(3)	0.5836(4)	0.0455(11)
<b>C33</b>	0.7628(3)	0.2245(3)	0.6033(4)	0.0453(11)
<b>C34</b>	0.7726(3)	0.1649(3)	0.5293(5)	0.0487(12)
<b>C35</b>	0.8162(3)	0.1670(3)	0.4394(4)	0.0514(13)
<b>C36</b>	0.8539(3)	0.2300(3)	0.4225(5)	0.0504(12)
<b>Cl1</b>	0.57157(10)	0.68576(9)	0.35919(16)	0.0770(5)
<b>Cl2</b>	1.03851(13)	0.40628(12)	0.84780(15)	0.0911(6)
<b>F1</b>	0.6582(2)	0.38673(18)	0.7997(3)	0.0629(8)
<b>F2</b>	0.61820(18)	0.22678(17)	0.4764(3)	0.0567(7)
<b>F3</b>	0.79638(17)	0.34324(16)	0.6558(3)	0.0540(7)
<b>F4</b>	0.7364(2)	0.10434(16)	0.5506(3)	0.0622(8)
<b>N1</b>	0.6699(3)	0.2504(3)	0.7220(4)	0.0526(11)
<b>N2</b>	0.7228(3)	0.2125(2)	0.7040(4)	0.0500(10)
<b>P1</b>	0.61864(7)	0.52669(8)	0.66464(12)	0.0484(3)
<b>P2</b>	0.90543(7)	0.37095(8)	0.48767(11)	0.0461(3)
<b>Cl8</b>	0.1948(3)	0.2908(5)	0.9662(7)	0.129(3)
<b>C38</b>	0.2719(3)	0.2554(3)	0.9061(6)	0.114(6)

<b>C16</b>	0.2489(5)	0.2287(4)	0.7536(7)	0.121(3)
<b>C17</b>	0.3454(3)	0.3178(6)	0.9216(9)	0.103(4)
<b>C15</b>	0.3278(5)	0.3333(5)	0.9070(9)	0.095(3)
<b>C14</b>	0.2156(7)	0.2700(6)	1.0377(10)	0.220(7)
<b>C13</b>	0.2546(5)	0.1994(4)	0.8205(10)	0.167(5)
<b>C37</b>	0.2867(5)	0.2521(4)	0.9489(7)	0.157(8)

**Table 3 Anisotropic Displacement Parameters for Z-3**

<b>Atom</b>	<b>U<sub>11</sub></b>	<b>U<sub>22</sub></b>	<b>U<sub>33</sub></b>	<b>U<sub>23</sub></b>	<b>U<sub>13</sub></b>	<b>U<sub>12</sub></b>
<b>Au1</b>	0.05442(14)	0.05611(15)	0.05254(14)	0.00253(9)	0.00496(10)	-0.00033(9)
<b>Au2</b>	0.05944(14)	0.06476(16)	0.03929(12)	0.00068(9)	-0.00087(9)	-0.01185(10)
<b>C1</b>	0.046(3)	0.059(3)	0.040(3)	-0.001(2)	0.005(2)	0.004(2)
<b>C2</b>	0.058(3)	0.065(4)	0.059(4)	-0.005(3)	0.007(3)	0.010(3)
<b>C3</b>	0.073(4)	0.075(4)	0.064(4)	-0.008(3)	0.007(3)	0.020(4)
<b>C4</b>	0.061(4)	0.099(5)	0.052(3)	0.002(3)	0.012(3)	0.024(4)
<b>C5</b>	0.052(3)	0.087(5)	0.053(3)	0.015(3)	0.003(3)	0.006(3)
<b>C6</b>	0.052(3)	0.070(4)	0.049(3)	-0.004(3)	0.002(2)	0.002(3)
<b>C7</b>	0.052(3)	0.052(3)	0.055(3)	-0.005(3)	0.000(2)	-0.001(2)
<b>C8</b>	0.059(4)	0.098(5)	0.061(4)	-0.017(4)	-0.002(3)	0.003(4)
<b>C9</b>	0.051(3)	0.094(5)	0.097(5)	-0.032(4)	0.004(3)	0.001(3)
<b>C10</b>	0.060(4)	0.062(4)	0.087(5)	-0.010(3)	-0.021(3)	-0.008(3)
<b>C11</b>	0.075(4)	0.078(5)	0.057(4)	0.002(3)	-0.010(3)	-0.011(4)
<b>C12</b>	0.060(3)	0.068(4)	0.059(3)	0.001(3)	0.004(3)	-0.009(3)
<b>C13</b>	0.040(3)	0.054(3)	0.044(3)	-0.001(2)	0.007(2)	0.000(2)
<b>C14</b>	0.046(3)	0.063(3)	0.041(3)	0.001(2)	0.002(2)	0.004(2)
<b>C15</b>	0.042(3)	0.072(4)	0.037(3)	-0.006(3)	-0.001(2)	-0.001(2)
<b>C16</b>	0.044(3)	0.051(3)	0.043(3)	-0.005(2)	0.011(2)	-0.008(2)
<b>C17</b>	0.047(3)	0.055(3)	0.037(2)	0.003(2)	0.012(2)	0.000(2)
<b>C18</b>	0.046(3)	0.060(3)	0.032(2)	0.000(2)	0.006(2)	-0.004(2)
<b>C19</b>	0.050(3)	0.059(3)	0.043(3)	0.003(2)	0.011(2)	-0.008(3)
<b>C20</b>	0.054(3)	0.059(4)	0.095(5)	0.010(3)	0.015(3)	-0.007(3)
<b>C21</b>	0.075(5)	0.061(4)	0.129(7)	0.014(4)	0.031(5)	-0.007(4)
<b>C22</b>	0.081(5)	0.083(5)	0.103(6)	0.035(5)	0.027(4)	0.020(4)
<b>C23</b>	0.067(4)	0.081(5)	0.063(4)	0.020(3)	0.002(3)	0.012(3)
<b>C24</b>	0.060(3)	0.055(3)	0.049(3)	-0.003(3)	0.007(2)	-0.001(3)
<b>C25</b>	0.047(3)	0.067(4)	0.047(3)	-0.004(3)	0.010(2)	-0.013(3)
<b>C26</b>	0.063(4)	0.069(4)	0.072(4)	0.007(3)	0.020(3)	0.001(3)
<b>C27</b>	0.072(4)	0.070(4)	0.120(7)	-0.013(4)	0.041(4)	0.001(4)
<b>C28</b>	0.084(5)	0.085(5)	0.064(4)	-0.010(4)	0.031(4)	-0.022(4)
<b>C29</b>	0.083(5)	0.113(6)	0.050(4)	0.004(4)	0.019(3)	-0.014(4)
<b>C30</b>	0.065(4)	0.091(5)	0.040(3)	0.005(3)	0.009(3)	-0.003(3)
<b>C31</b>	0.045(3)	0.051(3)	0.036(2)	0.001(2)	0.000(2)	-0.006(2)
<b>C32</b>	0.053(3)	0.049(3)	0.034(2)	0.003(2)	0.003(2)	0.001(2)
<b>C33</b>	0.051(3)	0.050(3)	0.035(2)	0.002(2)	0.006(2)	0.000(2)
<b>C34</b>	0.052(3)	0.049(3)	0.044(3)	0.002(2)	0.003(2)	-0.004(2)
<b>C35</b>	0.056(3)	0.053(3)	0.044(3)	-0.007(2)	0.004(2)	0.002(2)
<b>C36</b>	0.050(3)	0.061(3)	0.040(3)	-0.001(2)	0.006(2)	0.000(3)
<b>C11</b>	0.0938(12)	0.0665(10)	0.069(1)	0.0152(8)	0.0039(9)	-0.0052(9)
<b>C12</b>	0.1186(15)	0.0989(14)	0.0474(9)	-0.0014(8)	-0.0229(9)	-0.0193(11)
<b>F1</b>	0.078(2)	0.066(2)	0.0427(17)	-0.0033(14)	0.0004(15)	0.0028(16)
<b>F2</b>	0.0700(19)	0.0546(18)	0.0461(16)	-0.0064(14)	0.0098(14)	-0.0036(15)
<b>F3</b>	0.0681(19)	0.0532(18)	0.0428(15)	-0.0060(13)	0.0151(14)	-0.0103(15)

<b>F4</b>	0.073(2)	0.0494(18)	0.067(2)	0.0001(15)	0.0171(17)	-0.0069(15)
<b>N1</b>	0.070(3)	0.055(3)	0.036(2)	0.0009(19)	0.015(2)	-0.010(2)
<b>N2</b>	0.060(3)	0.049(2)	0.043(2)	0.0035(19)	0.013(2)	-0.001(2)
<b>P1</b>	0.0460(7)	0.0533(8)	0.0450(7)	-0.0031(6)	0.0021(5)	0.0018(6)
<b>P2</b>	0.0467(7)	0.0549(7)	0.0371(6)	0.0014(5)	0.0060(5)	-0.0050(6)
<b>Cl8</b>	0.079(3)	0.160(7)	0.152(6)	-0.075(6)	0.026(4)	-0.021(4)
<b>C38</b>	0.088(9)	0.117(13)	0.141(11)	-0.019(9)	0.034(8)	-0.011(8)
<b>Cl6</b>	0.130(6)	0.095(5)	0.147(6)	-0.034(4)	0.051(5)	-0.031(4)
<b>Cl7</b>	0.082(5)	0.099(5)	0.123(6)	0.033(4)	-0.003(4)	-0.003(4)
<b>Cl5</b>	0.093(4)	0.105(6)	0.088(4)	-0.012(5)	0.005(4)	0.000(5)
<b>Cl4</b>	0.224(13)	0.179(10)	0.268(14)	0.067(9)	0.067(11)	-0.022(9)
<b>Cl3</b>	0.142(7)	0.126(7)	0.208(9)	-0.092(7)	-0.079(7)	0.046(5)
<b>C37</b>	0.183(16)	0.130(13)	0.151(13)	-0.022(9)	-0.015(10)	-0.067(11)

**Table 4 Bond Lengths for Z-3**

Atom	Atom	Length/Å	Atom	Atom	Length/Å
<b>Au1</b>	<b>Cl1</b>	2.2797(16)	<b>C19</b>	<b>P2</b>	1.805(6)
<b>Au1</b>	<b>P1</b>	2.2278(14)	<b>C20</b>	<b>C21</b>	1.389(10)
<b>Au2</b>	<b>Cl2</b>	2.2693(16)	<b>C21</b>	<b>C22</b>	1.376(11)
<b>Au2</b>	<b>P2</b>	2.2230(13)	<b>C22</b>	<b>C23</b>	1.386(11)
<b>C1</b>	<b>C2</b>	1.390(8)	<b>C23</b>	<b>C24</b>	1.382(9)
<b>C1</b>	<b>C6</b>	1.385(8)	<b>C25</b>	<b>C26</b>	1.358(9)
<b>C1</b>	<b>P1</b>	1.792(5)	<b>C25</b>	<b>C30</b>	1.395(8)
<b>C2</b>	<b>C3</b>	1.368(9)	<b>C25</b>	<b>P2</b>	1.815(5)
<b>C3</b>	<b>C4</b>	1.386(10)	<b>C26</b>	<b>C27</b>	1.416(10)
<b>C4</b>	<b>C5</b>	1.369(10)	<b>C27</b>	<b>C28</b>	1.346(11)
<b>C5</b>	<b>C6</b>	1.395(8)	<b>C28</b>	<b>C29</b>	1.345(11)
<b>C7</b>	<b>C8</b>	1.394(8)	<b>C29</b>	<b>C30</b>	1.397(9)
<b>C7</b>	<b>C12</b>	1.346(8)	<b>C31</b>	<b>C32</b>	1.384(7)
<b>C7</b>	<b>P1</b>	1.817(5)	<b>C31</b>	<b>C36</b>	1.397(7)
<b>C8</b>	<b>C9</b>	1.375(9)	<b>C31</b>	<b>P2</b>	1.824(5)
<b>C9</b>	<b>C10</b>	1.374(10)	<b>C32</b>	<b>C33</b>	1.383(7)
<b>C10</b>	<b>C11</b>	1.353(10)	<b>C32</b>	<b>F3</b>	1.346(6)
<b>C11</b>	<b>C12</b>	1.404(8)	<b>C33</b>	<b>C34</b>	1.402(7)
<b>C13</b>	<b>C14</b>	1.401(7)	<b>C33</b>	<b>N2</b>	1.435(6)
<b>C13</b>	<b>C18</b>	1.378(7)	<b>C34</b>	<b>C35</b>	1.361(7)
<b>C13</b>	<b>P1</b>	1.823(5)	<b>C34</b>	<b>F4</b>	1.341(6)
<b>C14</b>	<b>C15</b>	1.361(8)	<b>C35</b>	<b>C36</b>	1.383(8)
<b>C15</b>	<b>C16</b>	1.378(8)	<b>N1</b>	<b>N2</b>	1.238(6)
<b>C16</b>	<b>C17</b>	1.378(7)	<b>Cl8</b>	<b>C38</b>	1.7710
<b>C16</b>	<b>F2</b>	1.350(6)	<b>C38</b>	<b>Cl6</b>	1.7724
<b>C17</b>	<b>C18</b>	1.376(7)	<b>C38</b>	<b>Cl7</b>	1.7743
<b>C17</b>	<b>N1</b>	1.447(7)	<b>Cl5</b>	<b>C37</b>	1.7717
<b>C18</b>	<b>F1</b>	1.337(6)	<b>Cl4</b>	<b>C37</b>	1.7695
<b>C19</b>	<b>C20</b>	1.392(8)	<b>Cl3</b>	<b>C37</b>	1.7748
<b>C19</b>	<b>C24</b>	1.372(8)			

**Table 5 Bond Angles for Z-3**

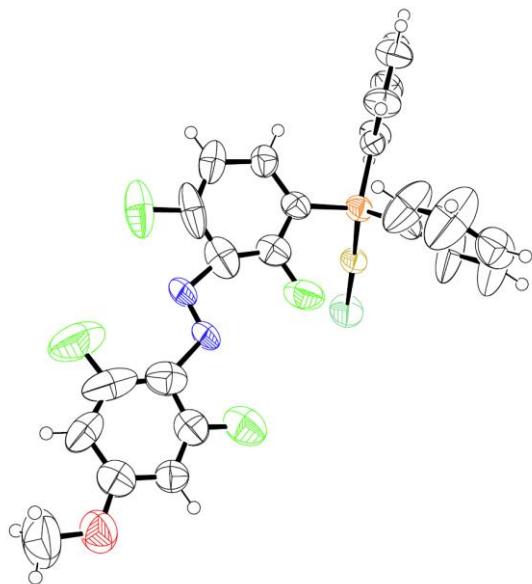
Atom	Atom	Atom	Angle/°	Atom	Atom	Atom	Angle/°
<b>P1</b>	<b>Au1</b>	<b>Cl1</b>	177.12(6)	<b>C26</b>	<b>C25</b>	<b>P2</b>	119.2(5)
<b>P2</b>	<b>Au2</b>	<b>Cl2</b>	177.04(7)	<b>C30</b>	<b>C25</b>	<b>P2</b>	119.9(5)
<b>C2</b>	<b>C1</b>	<b>P1</b>	117.3(4)	<b>C25</b>	<b>C26</b>	<b>C27</b>	118.5(7)

<b>C6</b>	<b>C1</b>	<b>C2</b>	119.5(5)	<b>C28</b>	<b>C27</b>	<b>C26</b>	120.4(7)
<b>C6</b>	<b>C1</b>	<b>P1</b>	123.1(4)	<b>C29</b>	<b>C28</b>	<b>C27</b>	121.2(7)
<b>C3</b>	<b>C2</b>	<b>C1</b>	121.2(6)	<b>C28</b>	<b>C29</b>	<b>C30</b>	120.4(7)
<b>C2</b>	<b>C3</b>	<b>C4</b>	118.9(6)	<b>C25</b>	<b>C30</b>	<b>C29</b>	118.6(7)
<b>C5</b>	<b>C4</b>	<b>C3</b>	121.0(6)	<b>C32</b>	<b>C31</b>	<b>C36</b>	117.0(5)
<b>C4</b>	<b>C5</b>	<b>C6</b>	119.9(6)	<b>C32</b>	<b>C31</b>	<b>P2</b>	119.6(4)
<b>C1</b>	<b>C6</b>	<b>C5</b>	119.4(6)	<b>C36</b>	<b>C31</b>	<b>P2</b>	123.1(4)
<b>C8</b>	<b>C7</b>	<b>P1</b>	117.3(4)	<b>C33</b>	<b>C32</b>	<b>C31</b>	123.5(5)
<b>C12</b>	<b>C7</b>	<b>C8</b>	118.7(5)	<b>F3</b>	<b>C32</b>	<b>C31</b>	118.6(4)
<b>C12</b>	<b>C7</b>	<b>P1</b>	124.0(5)	<b>F3</b>	<b>C32</b>	<b>C33</b>	117.9(4)
<b>C9</b>	<b>C8</b>	<b>C7</b>	120.2(6)	<b>C32</b>	<b>C33</b>	<b>C34</b>	116.3(4)
<b>C10</b>	<b>C9</b>	<b>C8</b>	120.4(7)	<b>C32</b>	<b>C33</b>	<b>N2</b>	125.7(4)
<b>C11</b>	<b>C10</b>	<b>C9</b>	119.9(6)	<b>C34</b>	<b>C33</b>	<b>N2</b>	117.1(5)
<b>C10</b>	<b>C11</b>	<b>C12</b>	119.6(6)	<b>C35</b>	<b>C34</b>	<b>C33</b>	122.8(5)
<b>C7</b>	<b>C12</b>	<b>C11</b>	121.3(6)	<b>F4</b>	<b>C34</b>	<b>C33</b>	116.7(5)
<b>C14</b>	<b>C13</b>	<b>P1</b>	120.7(4)	<b>F4</b>	<b>C34</b>	<b>C35</b>	120.5(5)
<b>C18</b>	<b>C13</b>	<b>C14</b>	116.6(5)	<b>C34</b>	<b>C35</b>	<b>C36</b>	118.6(5)
<b>C18</b>	<b>C13</b>	<b>P1</b>	122.5(4)	<b>C35</b>	<b>C36</b>	<b>C31</b>	121.8(5)
<b>C15</b>	<b>C14</b>	<b>C13</b>	121.9(5)	<b>N2</b>	<b>N1</b>	<b>C17</b>	120.8(4)
<b>C14</b>	<b>C15</b>	<b>C16</b>	118.9(5)	<b>N1</b>	<b>N2</b>	<b>C33</b>	122.1(4)
<b>C15</b>	<b>C16</b>	<b>C17</b>	121.9(5)	<b>C1</b>	<b>P1</b>	<b>Au1</b>	109.30(18)
<b>F2</b>	<b>C16</b>	<b>C15</b>	120.0(4)	<b>C1</b>	<b>P1</b>	<b>C7</b>	107.0(3)
<b>F2</b>	<b>C16</b>	<b>C17</b>	118.1(5)	<b>C1</b>	<b>P1</b>	<b>C13</b>	108.2(2)
<b>C16</b>	<b>C17</b>	<b>N1</b>	124.0(5)	<b>C7</b>	<b>P1</b>	<b>Au1</b>	116.5(2)
<b>C18</b>	<b>C17</b>	<b>C16</b>	117.3(5)	<b>C7</b>	<b>P1</b>	<b>C13</b>	104.2(2)
<b>C18</b>	<b>C17</b>	<b>N1</b>	118.5(4)	<b>C13</b>	<b>P1</b>	<b>Au1</b>	111.34(17)
<b>C17</b>	<b>C18</b>	<b>C13</b>	123.2(5)	<b>C19</b>	<b>P2</b>	<b>Au2</b>	113.85(18)
<b>F1</b>	<b>C18</b>	<b>C13</b>	119.2(5)	<b>C19</b>	<b>P2</b>	<b>C25</b>	104.5(3)
<b>F1</b>	<b>C18</b>	<b>C17</b>	117.5(5)	<b>C19</b>	<b>P2</b>	<b>C31</b>	107.5(2)
<b>C20</b>	<b>C19</b>	<b>P2</b>	117.1(4)	<b>C25</b>	<b>P2</b>	<b>Au2</b>	115.38(18)
<b>C24</b>	<b>C19</b>	<b>C20</b>	119.7(5)	<b>C25</b>	<b>P2</b>	<b>C31</b>	104.6(2)
<b>C24</b>	<b>C19</b>	<b>P2</b>	123.0(4)	<b>C31</b>	<b>P2</b>	<b>Au2</b>	110.28(16)
<b>C21</b>	<b>C20</b>	<b>C19</b>	119.7(6)	<b>Cl8</b>	<b>C38</b>	<b>Cl6</b>	111.0
<b>C22</b>	<b>C21</b>	<b>C20</b>	119.8(7)	<b>Cl8</b>	<b>C38</b>	<b>Cl7</b>	111.2
<b>C21</b>	<b>C22</b>	<b>C23</b>	120.8(7)	<b>Cl6</b>	<b>C38</b>	<b>Cl7</b>	111.4
<b>C24</b>	<b>C23</b>	<b>C22</b>	119.0(6)	<b>Cl5</b>	<b>C37</b>	<b>Cl3</b>	111.4
<b>C19</b>	<b>C24</b>	<b>C23</b>	121.1(6)	<b>Cl4</b>	<b>C37</b>	<b>Cl5</b>	111.1
<b>C26</b>	<b>C25</b>	<b>C30</b>	120.8(6)	<b>Cl4</b>	<b>C37</b>	<b>Cl3</b>	111.0

**Table 6 Atomic Occupancy for Z-3**

Atom	Occupancy	Atom	Occupancy
<b>C38</b>	0.4	<b>C37</b>	0.4
<b>H38</b>	0.4	<b>H1</b>	0.4
<b>Cl6</b>	0.4	<b>Cl3</b>	0.4
<b>Cl7</b>	0.4	<b>Cl4</b>	0.4
<b>Cl8</b>	0.4	<b>Cl5</b>	0.4

Crystal Data for *E*-6 :



**Figure SI7 : Ortep view of the complex (Color scheme: C, gray; N, blue; F, green; Au, brown; Cl, pale green; P, orange)**

$C_{25}H_{17}AuClF_4N_2OP$  ( $M = 700.79$  g/mol): monoclinic, space group  $P2_1/n$  (no. 14),  $a = 19.942(2)$  Å,  $b = 10.9697(7)$  Å,  $c = 24.135(2)$  Å,  $\beta = 112.700(11)^\circ$ ,  $V = 4870.6(8)$  Å $^3$ ,  $Z = 8$ ,  $T = 250$  K,  $\mu(\text{MoK}\alpha) = 6.268$  mm $^{-1}$ ,  $D_{\text{calc}} = 1.911$  g/cm $^3$ , 44413 reflections measured ( $6.666^\circ \leq 2\Theta \leq 52.738^\circ$ ), 9919 unique ( $R_{\text{int}} = 0.0720$ ,  $R_{\text{sigma}} = 0.0726$ ) which were used in all calculations. The final  $R_1$  was 0.0486 ( $I > 2\sigma(I)$ ) and  $wR_2$  was 0.1371 (all data).

**Table 1 Crystal data and structure refinement for *E*-6**

CCDC number	1484587
Empirical formula	$C_{25}H_{17}AuClF_4N_2OP$
Formula weight	700.79
Temperature/K	250
Crystal system	monoclinic
Space group	$P2_1/n$
$a/\text{\AA}$	19.942(2)
$b/\text{\AA}$	10.9697(7)
$c/\text{\AA}$	24.135(2)
$\alpha/^\circ$	90
$\beta/^\circ$	112.700(11)
$\gamma/^\circ$	90
Volume/Å $^3$	4870.6(8)
$Z$	8
$\rho_{\text{calcd}}/\text{cm}^3$	1.911
$\mu/\text{mm}^{-1}$	6.268
$F(000)$	2688.0
Crystal size/mm $^3$	$0.23 \times 0.12 \times 0.11$
Radiation	$\text{MoK}\alpha (\lambda = 0.71073)$
$2\Theta$ range for data collection/°	6.666 to 52.738
Index ranges	$-24 \leq h \leq 24$ , $-13 \leq k \leq 12$ , $-30 \leq l \leq 30$
Reflections collected	44413
Independent reflections	9919 [ $R_{\text{int}} = 0.0720$ , $R_{\text{sigma}} = 0.0726$ ]
Data/restraints/parameters	9919/10/585
Goodness-of-fit on $F^2$	1.040
Final R indexes [I $>= 2\sigma(I)$ ]	$R_1 = 0.0486$ , $wR_2 = 0.1096$
Final R indexes [all data]	$R_1 = 0.0895$ , $wR_2 = 0.1371$
Largest diff. peak/hole / e Å $^{-3}$	1.02/-1.23

**Table 2 Fractional Atomic Coordinates and Equivalent Isotropic Displacement Parameters for E-6**

Atom	<i>x</i>	<i>y</i>	<i>z</i>	U(eq)
<b>Au1</b>	0.06879(2)	0.24009(3)	0.79035(2)	0.05406(13)
<b>Cl1</b>	0.00203(18)	0.0658(2)	0.76163(13)	0.0892(9)
<b>P1</b>	0.13434(13)	0.4093(2)	0.82145(10)	0.0512(5)
<b>F1</b>	0.2504(3)	0.2352(5)	0.8858(3)	0.0913(19)
<b>F2</b>	0.2840(4)	0.3388(7)	1.0845(3)	0.122(3)
<b>F3</b>	0.4044(6)	0.0679(8)	1.1394(5)	0.162(4)
<b>F4</b>	0.3851(4)	-0.0226(6)	0.9457(3)	0.108(2)
<b>O1</b>	0.5454(5)	-0.2437(8)	1.1018(4)	0.103(3)
<b>N1</b>	0.3179(8)	0.1983(15)	1.0266(7)	0.062(2)
<b>N2</b>	0.3378(10)	0.1269(16)	0.9967(11)	0.062(2)
<b>N3</b>	0.3241(10)	0.1742(16)	0.9888(11)	0.062(2)
<b>N4</b>	0.3482(8)	0.1293(13)	1.0415(7)	0.062(2)
<b>C1</b>	0.1960(3)	0.4411(6)	0.7857(3)	0.056(2)
<b>C6</b>	0.1785(4)	0.3946(8)	0.7282(3)	0.129(6)
<b>C5</b>	0.2215(5)	0.4225(10)	0.6964(3)	0.142(6)
<b>C4</b>	0.2821(5)	0.4970(9)	0.7220(4)	0.099(4)
<b>C3</b>	0.2996(4)	0.5435(9)	0.7795(5)	0.170(9)
<b>C2</b>	0.2566(4)	0.5155(8)	0.8113(3)	0.132(6)
<b>C7</b>	0.0778(3)	0.5445(4)	0.8099(3)	0.050(2)
<b>C8</b>	0.0028(3)	0.5303(4)	0.7876(3)	0.064(2)
<b>C9</b>	-0.0420(2)	0.6320(6)	0.7769(3)	0.076(3)
<b>C10</b>	-0.0117(4)	0.7481(5)	0.7885(3)	0.073(3)
<b>C11</b>	0.0634(4)	0.7623(4)	0.8108(3)	0.082(3)
<b>C12</b>	0.1081(3)	0.6605(5)	0.8215(3)	0.072(3)
<b>C13</b>	0.1877(5)	0.4016(7)	0.9023(4)	0.053(2)
<b>C14</b>	0.2372(5)	0.3091(9)	0.9257(5)	0.066(3)
<b>C15</b>	0.2731(6)	0.2777(10)	0.9853(6)	0.085(3)
<b>C16</b>	0.2544(7)	0.3549(15)	1.0233(5)	0.103(5)
<b>C17</b>	0.2055(6)	0.4534(12)	1.0046(5)	0.083(3)
<b>C18</b>	0.1734(5)	0.4746(9)	0.9436(4)	0.062(2)
<b>C19</b>	0.3900(6)	0.0354(10)	1.0379(7)	0.088(3)
<b>C20</b>	0.4150(6)	-0.0420(10)	1.0061(5)	0.078(3)
<b>C21</b>	0.4632(5)	-0.1321(9)	1.0247(5)	0.068(3)
<b>C22</b>	0.4947(6)	-0.1512(10)	1.0860(6)	0.081(3)
<b>C23</b>	0.4733(7)	-0.0837(13)	1.1242(6)	0.096(4)
<b>C24</b>	0.4225(9)	0.0049(13)	1.0998(9)	0.115(5)
<b>C25</b>	0.5892(10)	-0.2614(18)	1.1649(7)	0.171(8)
<b>Au2</b>	0.44221(2)	0.37654(3)	0.39042(2)	0.04605(12)
<b>Cl2</b>	0.50957(12)	0.20410(19)	0.39845(10)	0.0604(6)
<b>P2</b>	0.37402(11)	0.53786(19)	0.39002(9)	0.0456(5)
<b>F5</b>	0.2716(3)	0.3488(5)	0.3941(3)	0.0811(16)
<b>F6</b>	0.2857(3)	0.4130(6)	0.5902(3)	0.100(2)
<b>F7</b>	0.1591(4)	0.1380(6)	0.5642(3)	0.122(3)
<b>F8</b>	0.1235(5)	0.1326(7)	0.3602(4)	0.132(3)
<b>O2</b>	0.0135(4)	-0.1623(6)	0.4409(3)	0.0797(19)
<b>N5</b>	0.2350(8)	0.2833(13)	0.5035(6)	0.059(2)
<b>N6</b>	0.1904(9)	0.2427(15)	0.4542(7)	0.059(2)
<b>N7</b>	0.2205(10)	0.2736(15)	0.4630(7)	0.059(2)
<b>N8</b>	0.1965(8)	0.2379(12)	0.4985(6)	0.059(2)
<b>C26</b>	0.4247(2)	0.6798(4)	0.4108(2)	0.0442(18)
<b>C31</b>	0.4994(3)	0.6735(4)	0.4427(2)	0.052(2)
<b>C30</b>	0.5397(2)	0.7799(5)	0.4613(2)	0.060(2)
<b>C29</b>	0.5054(3)	0.8926(4)	0.4481(3)	0.058(2)
<b>C28</b>	0.4306(3)	0.8989(3)	0.4163(3)	0.068(3)
<b>C27</b>	0.3903(2)	0.7925(5)	0.3977(3)	0.059(2)
<b>C32</b>	0.2985(3)	0.5684(5)	0.3200(2)	0.052(2)
<b>C33</b>	0.2351(3)	0.6240(6)	0.3184(2)	0.066(3)

<b>C34</b>	0.1812(3)	0.6566(6)	0.2637(3)	0.084(3)
<b>C35</b>	0.1907(3)	0.6338(7)	0.2105(2)	0.096(4)
<b>C36</b>	0.2540(4)	0.5782(7)	0.2121(2)	0.094(4)
<b>C37</b>	0.3079(3)	0.5456(6)	0.2668(3)	0.070(3)
<b>C38</b>	0.3386(4)	0.5138(7)	0.4496(4)	0.0472(19)
<b>C39</b>	0.3629(4)	0.5788(8)	0.5018(4)	0.051(2)
<b>C40</b>	0.3455(5)	0.5448(8)	0.5496(4)	0.059(2)
<b>C41</b>	0.3023(5)	0.4481(9)	0.5433(4)	0.064(3)
<b>C42</b>	0.2721(5)	0.3791(8)	0.4912(5)	0.066(3)
<b>C43</b>	0.2932(5)	0.4160(8)	0.4455(4)	0.057(2)
<b>C44</b>	0.1471(6)	0.1433(10)	0.4643(7)	0.097(4)
<b>C45</b>	0.1297(6)	0.0859(11)	0.5093(6)	0.084(3)
<b>C46</b>	0.0877(6)	-0.0129(11)	0.5021(5)	0.079(3)
<b>C47</b>	0.0565(5)	-0.0641(9)	0.4464(5)	0.071(3)
<b>C48</b>	0.0687(6)	-0.0125(9)	0.3977(5)	0.080(3)
<b>C49</b>	0.1134(6)	0.0878(10)	0.4087(7)	0.090(4)
<b>C50</b>	-0.0207(7)	-0.2157(10)	0.3834(6)	0.101(4)

**Table 3 Anisotropic Displacement Parameters for E-6**

<b>Atom</b>	<b>U<sub>11</sub></b>	<b>U<sub>22</sub></b>	<b>U<sub>33</sub></b>	<b>U<sub>23</sub></b>	<b>U<sub>13</sub></b>	<b>U<sub>12</sub></b>
<b>Au1</b>	0.0651(3)	0.0421(2)	0.0558(2)	0.00004(15)	0.02437(19)	-0.00493(16)
<b>Cl1</b>	0.121(2)	0.0546(15)	0.0912(19)	-0.0083(13)	0.0394(18)	-0.0293(16)
<b>P1</b>	0.0565(14)	0.0433(12)	0.0543(13)	0.0015(10)	0.0219(11)	-0.0039(11)
<b>F1</b>	0.087(4)	0.068(4)	0.100(5)	0.000(3)	0.016(4)	0.023(3)
<b>F2</b>	0.113(5)	0.167(7)	0.057(4)	0.028(4)	0.001(4)	-0.053(5)
<b>F3</b>	0.197(9)	0.140(7)	0.207(9)	-0.041(7)	0.142(8)	0.011(7)
<b>F4</b>	0.094(5)	0.096(5)	0.116(6)	0.014(4)	0.020(4)	0.013(4)
<b>O1</b>	0.089(6)	0.128(7)	0.086(6)	0.018(5)	0.027(5)	0.021(5)
<b>N1</b>	0.060(6)	0.063(8)	0.054(6)	0.019(7)	0.012(6)	0.027(5)
<b>N2</b>	0.060(6)	0.063(8)	0.054(6)	0.019(7)	0.012(6)	0.027(5)
<b>N3</b>	0.060(6)	0.063(8)	0.054(6)	0.019(7)	0.012(6)	0.027(5)
<b>N4</b>	0.060(6)	0.063(8)	0.054(6)	0.019(7)	0.012(6)	0.027(5)
<b>C1</b>	0.062(6)	0.045(5)	0.058(5)	0.010(4)	0.018(5)	-0.003(4)
<b>C6</b>	0.126(11)	0.219(16)	0.059(7)	-0.030(8)	0.053(8)	-0.086(11)
<b>C5</b>	0.115(11)	0.257(19)	0.066(8)	-0.011(10)	0.048(8)	-0.045(13)
<b>C4</b>	0.087(9)	0.098(9)	0.129(11)	0.024(8)	0.063(8)	-0.004(7)
<b>C3</b>	0.144(14)	0.24(2)	0.166(16)	-0.101(15)	0.102(13)	-0.119(15)
<b>C2</b>	0.098(10)	0.190(15)	0.136(12)	-0.080(11)	0.079(9)	-0.067(10)
<b>C7</b>	0.063(6)	0.042(5)	0.052(5)	0.002(4)	0.030(4)	-0.001(4)
<b>C8</b>	0.062(6)	0.052(6)	0.075(6)	0.004(5)	0.024(5)	0.002(5)
<b>C9</b>	0.066(7)	0.076(8)	0.084(7)	0.015(6)	0.026(6)	0.014(6)
<b>C10</b>	0.094(8)	0.058(7)	0.070(7)	0.017(5)	0.037(6)	0.018(6)
<b>C11</b>	0.092(9)	0.053(6)	0.098(8)	0.002(5)	0.032(7)	0.007(6)
<b>C12</b>	0.065(6)	0.052(6)	0.098(8)	-0.003(5)	0.032(6)	-0.006(5)
<b>C13</b>	0.053(5)	0.040(5)	0.060(5)	-0.001(4)	0.017(4)	-0.011(4)
<b>C14</b>	0.057(6)	0.055(6)	0.076(7)	-0.004(5)	0.015(5)	-0.012(5)
<b>C15</b>	0.070(7)	0.082(8)	0.077(8)	0.013(7)	0.000(7)	-0.025(6)
<b>C16</b>	0.083(9)	0.136(12)	0.060(8)	0.044(8)	-0.007(7)	-0.062(9)
<b>C17</b>	0.086(8)	0.112(9)	0.047(6)	-0.004(6)	0.023(6)	-0.031(7)
<b>C18</b>	0.065(6)	0.064(6)	0.055(6)	0.002(5)	0.023(5)	-0.008(5)
<b>C19</b>	0.082(8)	0.071(8)	0.120(11)	0.001(8)	0.050(8)	-0.001(6)
<b>C20</b>	0.077(7)	0.072(7)	0.076(8)	0.003(6)	0.019(6)	-0.005(6)
<b>C21</b>	0.064(6)	0.071(7)	0.071(7)	0.002(5)	0.027(6)	0.006(5)
<b>C22</b>	0.076(7)	0.072(7)	0.093(9)	0.001(6)	0.029(7)	-0.008(6)
<b>C23</b>	0.118(10)	0.105(9)	0.085(8)	-0.029(7)	0.063(8)	-0.020(9)
<b>C24</b>	0.117(11)	0.078(9)	0.195(17)	-0.051(11)	0.109(13)	-0.027(8)
<b>C25</b>	0.159(16)	0.24(2)	0.091(12)	0.038(12)	0.022(12)	0.038(14)
<b>Au2</b>	0.0482(2)	0.0369(2)	0.0509(2)	-0.00624(14)	0.01665(16)	0.00190(14)

<b>C12</b>	0.0695(14)	0.0414(12)	0.0739(15)	-0.0043(10)	0.0316(12)	0.0113(11)
<b>P2</b>	0.0461(12)	0.0403(12)	0.0476(12)	-0.0027(9)	0.0150(10)	0.0045(9)
<b>F5</b>	0.083(4)	0.064(3)	0.104(4)	-0.024(3)	0.045(3)	-0.029(3)
<b>F6</b>	0.097(5)	0.124(5)	0.102(5)	0.038(4)	0.065(4)	0.018(4)
<b>F7</b>	0.117(6)	0.105(5)	0.100(5)	-0.025(4)	-0.006(4)	-0.007(4)
<b>F8</b>	0.160(8)	0.106(6)	0.155(7)	-0.013(5)	0.088(6)	-0.056(5)
<b>O2</b>	0.085(5)	0.074(5)	0.077(5)	0.004(4)	0.028(4)	-0.024(4)
<b>N5</b>	0.074(8)	0.051(6)	0.059(5)	-0.001(5)	0.033(7)	-0.021(5)
<b>N6</b>	0.074(8)	0.051(6)	0.059(5)	-0.001(5)	0.033(7)	-0.021(5)
<b>N7</b>	0.074(8)	0.051(6)	0.059(5)	-0.001(5)	0.033(7)	-0.021(5)
<b>N8</b>	0.074(8)	0.051(6)	0.059(5)	-0.001(5)	0.033(7)	-0.021(5)
<b>C26</b>	0.050(5)	0.038(5)	0.040(4)	-0.001(4)	0.014(4)	0.002(4)
<b>C31</b>	0.053(5)	0.038(5)	0.057(5)	0.006(4)	0.012(4)	-0.001(4)
<b>C30</b>	0.055(5)	0.052(5)	0.065(6)	-0.003(5)	0.013(5)	0.000(5)
<b>C29</b>	0.066(6)	0.046(5)	0.065(6)	-0.008(4)	0.030(5)	-0.020(5)
<b>C28</b>	0.072(7)	0.036(5)	0.092(7)	-0.002(5)	0.027(6)	0.005(5)
<b>C27</b>	0.046(5)	0.042(5)	0.079(6)	0.007(5)	0.014(5)	0.006(4)
<b>C32</b>	0.043(5)	0.048(5)	0.055(5)	-0.004(4)	0.006(4)	0.004(4)
<b>C33</b>	0.053(6)	0.060(6)	0.072(7)	-0.008(5)	0.011(5)	-0.002(5)
<b>C34</b>	0.050(6)	0.079(7)	0.085(8)	0.004(6)	-0.016(6)	0.008(5)
<b>C35</b>	0.081(9)	0.115(10)	0.051(7)	0.012(6)	-0.019(6)	-0.001(7)
<b>C36</b>	0.078(8)	0.141(11)	0.051(6)	0.012(7)	0.012(6)	0.019(8)
<b>C37</b>	0.065(6)	0.091(7)	0.049(6)	0.001(5)	0.016(5)	0.009(5)
<b>C38</b>	0.047(5)	0.040(5)	0.055(5)	-0.007(4)	0.020(4)	0.007(4)
<b>C39</b>	0.046(5)	0.053(5)	0.054(5)	-0.002(4)	0.018(4)	0.004(4)
<b>C40</b>	0.061(6)	0.060(6)	0.060(6)	0.001(4)	0.026(5)	0.010(5)
<b>C41</b>	0.065(6)	0.074(7)	0.071(7)	0.029(6)	0.045(6)	0.023(5)
<b>C42</b>	0.058(6)	0.055(6)	0.092(8)	0.019(6)	0.036(6)	0.008(5)
<b>C43</b>	0.055(5)	0.040(5)	0.069(6)	-0.010(4)	0.017(5)	0.002(4)
<b>C44</b>	0.054(7)	0.067(8)	0.141(12)	-0.033(8)	0.006(7)	0.006(6)
<b>C45</b>	0.073(8)	0.068(8)	0.083(8)	-0.019(7)	-0.002(7)	0.007(6)
<b>C46</b>	0.063(6)	0.085(8)	0.078(8)	-0.011(6)	0.015(6)	-0.010(6)
<b>C47</b>	0.061(6)	0.064(7)	0.088(8)	0.001(6)	0.028(6)	-0.007(5)
<b>C48</b>	0.069(7)	0.064(7)	0.098(8)	-0.008(6)	0.022(6)	-0.007(6)
<b>C49</b>	0.088(8)	0.051(6)	0.145(12)	0.002(7)	0.062(9)	-0.008(6)
<b>C50</b>	0.115(10)	0.068(7)	0.106(10)	0.007(7)	0.026(8)	-0.023(7)

**Table 4 Bond Lengths for E-6**

Atom	Atom	Length/Å	Atom	Atom	Length/Å
<b>Au1</b>	<b>Cl1</b>	2.280(3)	<b>Au2</b>	<b>Cl2</b>	2.285(2)
<b>Au1</b>	<b>P1</b>	2.227(2)	<b>Au2</b>	<b>P2</b>	2.230(2)
<b>P1</b>	<b>C1</b>	1.788(5)	<b>P2</b>	<b>C26</b>	1.818(4)
<b>P1</b>	<b>C7</b>	1.818(5)	<b>P2</b>	<b>C32</b>	1.810(5)
<b>P1</b>	<b>C13</b>	1.829(9)	<b>P2</b>	<b>C38</b>	1.850(8)
<b>F1</b>	<b>C14</b>	1.360(11)	<b>F5</b>	<b>C43</b>	1.362(10)
<b>F2</b>	<b>C16</b>	1.374(12)	<b>F6</b>	<b>C41</b>	1.352(9)
<b>F3</b>	<b>C24</b>	1.339(14)	<b>F7</b>	<b>C45</b>	1.351(12)
<b>F4</b>	<b>C20</b>	1.361(12)	<b>F8</b>	<b>C49</b>	1.354(14)
<b>O1</b>	<b>C22</b>	1.379(13)	<b>O2</b>	<b>C47</b>	1.351(11)
<b>O1</b>	<b>C25</b>	1.447(16)	<b>O2</b>	<b>C50</b>	1.415(13)
<b>N1</b>	<b>N2</b>	1.23(3)	<b>N5</b>	<b>N6</b>	1.260(16)
<b>N1</b>	<b>C15</b>	1.365(14)	<b>N5</b>	<b>C42</b>	1.382(13)
<b>N2</b>	<b>C19</b>	1.509(14)	<b>N6</b>	<b>C44</b>	1.468(15)
<b>N3</b>	<b>N4</b>	1.27(3)	<b>N7</b>	<b>N8</b>	1.198(15)
<b>N3</b>	<b>C15</b>	1.505(15)	<b>N7</b>	<b>C42</b>	1.523(14)
<b>N4</b>	<b>C19</b>	1.348(14)	<b>N8</b>	<b>C44</b>	1.450(14)
<b>C1</b>	<b>C6</b>	1.3900	<b>C26</b>	<b>C31</b>	1.3900
<b>C1</b>	<b>C2</b>	1.3900	<b>C26</b>	<b>C27</b>	1.3900

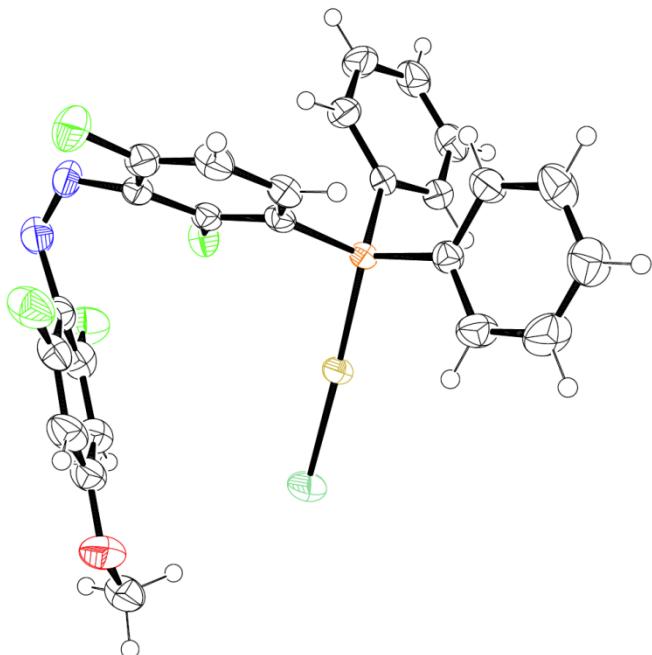
C6	C5	1.3900	C31	C30	1.3900
C5	C4	1.3900	C30	C29	1.3900
C4	C3	1.3900	C29	C28	1.3900
C3	C2	1.3900	C28	C27	1.3900
C7	C8	1.3900	C32	C33	1.3900
C7	C12	1.3900	C32	C37	1.3900
C8	C9	1.3900	C33	C34	1.3900
C9	C10	1.3900	C34	C35	1.3900
C10	C11	1.3900	C35	C36	1.3900
C11	C12	1.3900	C36	C37	1.3900
C13	C14	1.376(13)	C38	C39	1.364(11)
C13	C18	1.392(12)	C38	C43	1.382(12)
C14	C15	1.381(14)	C39	C40	1.377(11)
C15	C16	1.402(18)	C40	C41	1.339(13)
C16	C17	1.408(18)	C41	C42	1.391(13)
C17	C18	1.378(13)	C42	C43	1.384(13)
C19	C20	1.362(15)	C44	C45	1.410(17)
C19	C24	1.42(2)	C44	C49	1.388(17)
C20	C21	1.330(14)	C45	C46	1.340(16)
C21	C22	1.383(15)	C46	C47	1.366(14)
C22	C23	1.372(15)	C47	C48	1.408(14)
C23	C24	1.363(19)	C48	C49	1.377(14)

Table 5 Bond Angles for E-6

Atom	Atom	Atom	Angle/ <sup>o</sup>	Atom	Atom	Atom	Angle/ <sup>o</sup>
P1	Au1	Cl1	178.15(9)	P2	Au2	Cl2	174.41(8)
C1	P1	Au1	115.1(2)	C26	P2	Au2	113.94(19)
C1	P1	C7	105.5(3)	C26	P2	C38	104.2(3)
C1	P1	C13	107.4(4)	C32	P2	Au2	115.9(2)
C7	P1	Au1	112.3(2)	C32	P2	C26	105.8(3)
C7	P1	C13	105.8(4)	C32	P2	C38	108.8(3)
C13	P1	Au1	110.2(3)	C38	P2	Au2	107.4(2)
C22	O1	C25	118.0(11)	C47	O2	C50	118.3(8)
N2	N1	C15	104.3(15)	N6	N5	C42	107.9(13)
N1	N2	C19	109.5(18)	N5	N6	C44	110.6(13)
N4	N3	C15	109.5(17)	N8	N7	C42	108.4(13)
N3	N4	C19	103.7(15)	N7	N8	C44	101.4(13)
C6	C1	P1	117.1(4)	C31	C26	P2	118.2(3)
C6	C1	C2	120.0	C31	C26	C27	120.0
C2	C1	P1	122.8(4)	C27	C26	P2	121.7(3)
C1	C6	C5	120.0	C30	C31	C26	120.0
C6	C5	C4	120.0	C31	C30	C29	120.0
C3	C4	C5	120.0	C28	C29	C30	120.0
C4	C3	C2	120.0	C29	C28	C27	120.0
C3	C2	C1	120.0	C28	C27	C26	120.0
C8	C7	P1	118.6(3)	C33	C32	P2	121.8(3)
C8	C7	C12	120.0	C33	C32	C37	120.0
C12	C7	P1	121.4(3)	C37	C32	P2	117.9(3)
C9	C8	C7	120.0	C32	C33	C34	120.0
C8	C9	C10	120.0	C35	C34	C33	120.0
C11	C10	C9	120.0	C34	C35	C36	120.0
C12	C11	C10	120.0	C35	C36	C37	120.0
C11	C12	C7	120.0	C36	C37	C32	120.0
C14	C13	P1	120.1(7)	C39	C38	P2	122.5(6)
C14	C13	C18	116.3(9)	C39	C38	C43	117.5(8)
C18	C13	P1	122.8(7)	C43	C38	P2	119.4(6)
F1	C14	C13	116.9(9)	C38	C39	C40	121.1(8)
F1	C14	C15	115.2(10)	C41	C40	C39	118.7(9)

<b>C13</b>	<b>C14</b>	<b>C15</b>	127.8(10)	<b>F6</b>	<b>C41</b>	<b>C42</b>	116.3(9)
<b>N1</b>	<b>C15</b>	<b>C14</b>	148.2(14)	<b>C40</b>	<b>C41</b>	<b>F6</b>	119.1(10)
<b>N1</b>	<b>C15</b>	<b>C16</b>	99.9(13)	<b>C40</b>	<b>C41</b>	<b>C42</b>	124.6(8)
<b>C14</b>	<b>C15</b>	<b>N3</b>	108.3(13)	<b>N5</b>	<b>C42</b>	<b>C41</b>	107.4(10)
<b>C14</b>	<b>C15</b>	<b>C16</b>	111.7(10)	<b>N5</b>	<b>C42</b>	<b>C43</b>	138.3(11)
<b>C16</b>	<b>C15</b>	<b>N3</b>	139.9(14)	<b>C41</b>	<b>C42</b>	<b>N7</b>	143.1(10)
<b>F2</b>	<b>C16</b>	<b>C15</b>	120.9(14)	<b>C43</b>	<b>C42</b>	<b>N7</b>	103.1(10)
<b>F2</b>	<b>C16</b>	<b>C17</b>	113.7(15)	<b>C43</b>	<b>C42</b>	<b>C41</b>	113.8(8)
<b>C15</b>	<b>C16</b>	<b>C17</b>	125.4(10)	<b>F5</b>	<b>C43</b>	<b>C38</b>	117.0(8)
<b>C18</b>	<b>C17</b>	<b>C16</b>	116.8(11)	<b>F5</b>	<b>C43</b>	<b>C42</b>	118.7(8)
<b>C17</b>	<b>C18</b>	<b>C13</b>	121.9(10)	<b>C38</b>	<b>C43</b>	<b>C42</b>	124.2(8)
<b>N4</b>	<b>C19</b>	<b>C20</b>	151.9(14)	<b>C45</b>	<b>C44</b>	<b>N6</b>	142.6(12)
<b>N4</b>	<b>C19</b>	<b>C24</b>	98.9(13)	<b>C45</b>	<b>C44</b>	<b>N8</b>	101.3(12)
<b>C20</b>	<b>C19</b>	<b>N2</b>	110.7(14)	<b>C49</b>	<b>C44</b>	<b>N6</b>	105.2(13)
<b>C20</b>	<b>C19</b>	<b>C24</b>	109.2(10)	<b>C49</b>	<b>C44</b>	<b>N8</b>	146.5(14)
<b>C24</b>	<b>C19</b>	<b>N2</b>	140.2(15)	<b>C49</b>	<b>C44</b>	<b>C45</b>	112.2(10)
<b>F4</b>	<b>C20</b>	<b>C19</b>	113.7(11)	<b>F7</b>	<b>C45</b>	<b>C44</b>	115.1(12)
<b>C21</b>	<b>C20</b>	<b>F4</b>	116.0(11)	<b>C46</b>	<b>C45</b>	<b>F7</b>	118.7(13)
<b>C21</b>	<b>C20</b>	<b>C19</b>	130.3(11)	<b>C46</b>	<b>C45</b>	<b>C44</b>	126.2(11)
<b>C20</b>	<b>C21</b>	<b>C22</b>	116.4(10)	<b>C45</b>	<b>C46</b>	<b>C47</b>	118.9(11)
<b>O1</b>	<b>C22</b>	<b>C21</b>	112.9(10)	<b>O2</b>	<b>C47</b>	<b>C46</b>	117.1(10)
<b>C23</b>	<b>C22</b>	<b>O1</b>	126.6(12)	<b>O2</b>	<b>C47</b>	<b>C48</b>	123.1(10)
<b>C23</b>	<b>C22</b>	<b>C21</b>	120.5(12)	<b>C46</b>	<b>C47</b>	<b>C48</b>	119.8(10)
<b>C24</b>	<b>C23</b>	<b>C22</b>	117.7(13)	<b>C49</b>	<b>C48</b>	<b>C47</b>	117.9(11)
<b>F3</b>	<b>C24</b>	<b>C19</b>	119.4(15)	<b>F8</b>	<b>C49</b>	<b>C44</b>	119.9(11)
<b>F3</b>	<b>C24</b>	<b>C23</b>	114.8(17)	<b>F8</b>	<b>C49</b>	<b>C48</b>	115.1(12)
<b>C23</b>	<b>C24</b>	<b>C19</b>	125.8(11)	<b>C48</b>	<b>C49</b>	<b>C44</b>	124.9(12)

Crystal Data for Z-6:



**Figure SI8: Ortep view of the complex (Color scheme: C, gray; N, blue; F, green; Au, brown; Cl, pale green; P, orange)**

$C_{25}H_{17}AuClF_4N_2OP$  ( $M = 700.79$  g/mol): monoclinic, space group  $P2_1/n$  (no. 14),  $a = 16.0034(5)$  Å,  $b = 8.3754(2)$  Å,  $c = 19.2521(5)$  Å,  $\beta = 111.188(3)^\circ$ ,  $V = 2406.02(13)$  Å $^3$ ,  $Z = 4$ ,  $T = 200$  K,  $\mu(\text{MoK}\alpha) = 6.345$  mm $^{-1}$ ,  $D_{\text{calc}} = 1.935$  g/cm $^3$ , 25630 reflections measured ( $7.496^\circ \leq 2\Theta \leq 60.118^\circ$ ), 6263 unique ( $R_{\text{int}} = 0.0393$ ,  $R_{\text{sigma}} = 0.0405$ ) which were used in all calculations. The final  $R_1$  was 0.0258 ( $I > 2\sigma(I)$ ) and  $wR_2$  was 0.0598 (all data).

**Table 1 Crystal data and structure refinement for Z-6**

<b>CCDC number</b>	1484588
<b>Empirical formula</b>	C <sub>25</sub> H <sub>17</sub> AuClF <sub>4</sub> N <sub>2</sub> OP
<b>Formula weight</b>	700.79
<b>Temperature/K</b>	200
<b>Crystal system</b>	monoclinic
<b>Space group</b>	P2 <sub>1</sub> /n
<b>a/<math>\text{\AA}</math></b>	16.0034(5)
<b>b/<math>\text{\AA}</math></b>	8.3754(2)
<b>c/<math>\text{\AA}</math></b>	19.2521(5)
<b><math>\alpha/^\circ</math></b>	90
<b><math>\beta/^\circ</math></b>	111.188(3)
<b><math>\gamma/^\circ</math></b>	90
<b>Volume/<math>\text{\AA}^3</math></b>	2406.02(13)
<b>Z</b>	4
<b><math>\rho_{\text{calc}}/\text{cm}^3</math></b>	1.935
<b><math>\mu/\text{mm}^{-1}</math></b>	6.345
<b>F(000)</b>	1344.0
<b>Crystal size/mm<sup>3</sup></b>	0.28 × 0.14 × 0.12
<b>Radiation</b>	MoK $\alpha$ ( $\lambda = 0.71075$ )
<b>2<math>\Theta</math> range for data collection/<math>^\circ</math></b>	7.496 to 60.118
<b>Index ranges</b>	-21 ≤ h ≤ 20, -10 ≤ k ≤ 11, -27 ≤ l ≤ 25
<b>Reflections collected</b>	25630
<b>Independent reflections</b>	6263 [R <sub>int</sub> = 0.0393, R <sub>sigma</sub> = 0.0405]
<b>Data/restraints/parameters</b>	6263/0/317
<b>Goodness-of-fit on F<sup>2</sup></b>	1.058
<b>Final R indexes [I&gt;=2σ (I)]</b>	R <sub>1</sub> = 0.0258, wR <sub>2</sub> = 0.0566
<b>Final R indexes [all data]</b>	R <sub>1</sub> = 0.0389, wR <sub>2</sub> = 0.0598
<b>Largest diff. peak/hole / e <math>\text{\AA}^{-3}</math></b>	1.41/-0.84

**Table 2 Fractional Atomic Coordinates and Equivalent Isotropic Displacement Parameters Z-6**

<b>Atom</b>	<b>x</b>	<b>y</b>	<b>z</b>	<b>U(eq)</b>
<b>Au1</b>	0.14051(2)	0.42917(2)	0.53278(2)	0.02940(5)
<b>C1</b>	0.1642(2)	0.0942(3)	0.44570(15)	0.0295(6)
<b>C2</b>	0.1205(3)	0.0246(4)	0.48879(19)	0.0431(8)
<b>C3</b>	0.0904(3)	-0.1313(5)	0.4760(2)	0.0571(10)
<b>C4</b>	0.1025(3)	-0.2179(4)	0.4198(2)	0.0547(9)
<b>C5</b>	0.1468(3)	-0.1519(4)	0.3773(2)	0.0506(9)
<b>C6</b>	0.1777(2)	0.0030(4)	0.38979(17)	0.0394(7)
<b>C7</b>	0.18573(18)	0.3828(3)	0.37181(14)	0.0255(5)
<b>C8</b>	0.0985(2)	0.4145(3)	0.32438(17)	0.0335(6)
<b>C9</b>	0.0830(2)	0.4822(4)	0.25529(17)	0.0416(7)
<b>C10</b>	0.1535(2)	0.5189(4)	0.23358(17)	0.0434(8)
<b>C11</b>	0.2397(2)	0.4880(5)	0.27956(17)	0.0492(9)
<b>C12</b>	0.2565(2)	0.4169(4)	0.34913(17)	0.0380(7)
<b>C13</b>	0.32360(18)	0.2845(3)	0.50970(13)	0.0269(6)
<b>C14</b>	0.3709(2)	0.1412(4)	0.52255(15)	0.0321(6)
<b>C15</b>	0.4633(2)	0.1375(4)	0.55902(16)	0.0370(7)
<b>C16</b>	0.5075(2)	0.2780(4)	0.58126(15)	0.0372(7)
<b>C17</b>	0.4641(2)	0.4246(3)	0.57288(16)	0.0315(6)
<b>C18</b>	0.3723(2)	0.4236(3)	0.53534(15)	0.0283(6)
<b>C19</b>	0.4448(2)	0.6366(4)	0.67290(16)	0.0356(7)
<b>C20</b>	0.4632(2)	0.5138(4)	0.72552(16)	0.0397(7)
<b>C21</b>	0.4172(3)	0.4937(4)	0.77219(17)	0.0449(8)
<b>C22</b>	0.3475(2)	0.5982(4)	0.76686(17)	0.0404(8)
<b>C23</b>	0.3264(2)	0.7217(4)	0.71539(17)	0.0405(7)
<b>C24</b>	0.3767(2)	0.7385(4)	0.67184(17)	0.0405(7)
<b>C25</b>	0.2298(3)	0.6686(5)	0.8091(2)	0.0514(9)
<b>Cl1</b>	0.08525(6)	0.55569(10)	0.61219(5)	0.0494(2)
<b>F1</b>	0.32914(12)	0.56428(18)	0.52115(10)	0.0358(4)
<b>F2</b>	0.59769(12)	0.2788(3)	0.61422(11)	0.0520(5)
<b>F3</b>	0.53214(15)	0.4156(2)	0.73122(11)	0.0546(6)
<b>F4</b>	0.35809(16)	0.8603(3)	0.62255(12)	0.0613(6)
<b>N1</b>	0.51491(18)	0.5723(3)	0.58927(15)	0.0397(6)
<b>N2</b>	0.50042(19)	0.6731(3)	0.63133(15)	0.0433(7)
<b>O1</b>	0.30316(19)	0.5680(3)	0.81350(14)	0.0515(6)
<b>P1</b>	0.20218(5)	0.29764(9)	0.46283(3)	0.02505(14)

**Table 3 Anisotropic Displacement Parameters for Z-6.**

<b>Atom</b>	<b>U<sub>11</sub></b>	<b>U<sub>22</sub></b>	<b>U<sub>33</sub></b>	<b>U<sub>23</sub></b>	<b>U<sub>13</sub></b>	<b>U<sub>12</sub></b>
<b>Au1</b>	0.03056(7)	0.03458(7)	0.02486(6)	-0.00283(4)	0.01218(5)	0.00008(5)
<b>C1</b>	0.0299(15)	0.0301(15)	0.0272(13)	0.0003(11)	0.0090(12)	-0.0002(12)
<b>C2</b>	0.056(2)	0.0377(17)	0.0447(18)	0.0012(15)	0.0291(17)	-0.0026(16)
<b>C3</b>	0.075(3)	0.0372(18)	0.072(3)	0.0076(19)	0.043(2)	-0.008(2)
<b>C4</b>	0.070(3)	0.0311(16)	0.064(2)	-0.0028(17)	0.026(2)	-0.0054(18)
<b>C5</b>	0.067(2)	0.0330(17)	0.0491(19)	-0.0087(16)	0.0176(18)	-0.0022(18)
<b>C6</b>	0.0475(19)	0.0395(17)	0.0330(15)	-0.0048(14)	0.0166(14)	-0.0030(16)
<b>C7</b>	0.0271(14)	0.0251(13)	0.0222(12)	-0.0022(11)	0.0062(11)	-0.0009(11)
<b>C8</b>	0.0313(15)	0.0378(17)	0.0319(14)	-0.0024(12)	0.0122(12)	-0.0020(13)
<b>C9</b>	0.0409(18)	0.0457(18)	0.0279(15)	0.0009(14)	0.0001(13)	0.0060(16)
<b>C10</b>	0.054(2)	0.0476(19)	0.0240(14)	0.0047(14)	0.0081(14)	-0.0049(17)
<b>C11</b>	0.0421(19)	0.078(3)	0.0300(15)	0.0047(17)	0.0162(14)	-0.0143(19)
<b>C12</b>	0.0288(15)	0.060(2)	0.0238(13)	0.0046(13)	0.0076(12)	-0.0021(14)
<b>C13</b>	0.0294(14)	0.0337(14)	0.0170(11)	0.0014(11)	0.0078(10)	0.0026(12)
<b>C14</b>	0.0377(16)	0.0348(15)	0.0233(13)	-0.0003(12)	0.0102(12)	0.0014(14)
<b>C15</b>	0.0387(17)	0.0412(17)	0.0279(14)	0.0026(14)	0.0080(12)	0.0115(15)

<b>C16</b>	0.0287(14)	0.0561(19)	0.0246(13)	0.0023(14)	0.0070(12)	0.0025(15)
<b>C17</b>	0.0295(15)	0.0429(17)	0.0229(13)	-0.0007(12)	0.0105(11)	-0.0035(13)
<b>C18</b>	0.0289(14)	0.0356(16)	0.0204(12)	0.0007(11)	0.0090(11)	0.0011(12)
<b>C19</b>	0.0411(17)	0.0338(16)	0.0287(14)	-0.0061(13)	0.0090(13)	-0.0108(14)
<b>C20</b>	0.0425(18)	0.0439(18)	0.0275(15)	-0.0044(14)	0.0065(13)	0.0003(16)
<b>C21</b>	0.061(2)	0.0439(18)	0.0282(15)	0.0047(15)	0.0143(15)	-0.0014(18)
<b>C22</b>	0.051(2)	0.0411(18)	0.0306(15)	-0.0013(13)	0.0158(14)	-0.0069(15)
<b>C23</b>	0.0499(19)	0.0343(16)	0.0395(16)	-0.0009(14)	0.0188(15)	-0.0012(15)
<b>C24</b>	0.052(2)	0.0275(16)	0.0397(16)	0.0025(13)	0.0133(15)	-0.0081(14)
<b>C25</b>	0.061(2)	0.056(2)	0.0422(18)	-0.0103(16)	0.0237(17)	-0.0037(19)
<b>C11</b>	0.0498(5)	0.0634(6)	0.0442(4)	-0.0156(4)	0.0280(4)	-0.0007(4)
<b>F1</b>	0.0353(9)	0.0303(9)	0.0358(9)	-0.0013(7)	0.0059(7)	-0.0009(7)
<b>F2</b>	0.0256(9)	0.0691(13)	0.0540(11)	0.0015(10)	0.0055(8)	0.0054(9)
<b>F3</b>	0.0551(13)	0.0692(15)	0.0344(10)	0.0044(9)	0.0101(9)	0.0192(10)
<b>F4</b>	0.0866(17)	0.0414(11)	0.0702(14)	0.0199(11)	0.0454(13)	0.0084(12)
<b>N1</b>	0.0295(13)	0.0497(17)	0.0373(14)	-0.0018(12)	0.0088(11)	-0.0118(12)
<b>N2</b>	0.0426(15)	0.0495(17)	0.0359(13)	-0.0040(13)	0.0118(12)	-0.0116(13)
<b>O1</b>	0.0655(17)	0.0552(16)	0.0425(13)	0.0058(11)	0.0299(13)	0.0017(12)
<b>P1</b>	0.0261(3)	0.0288(3)	0.0203(3)	-0.0008(3)	0.0084(3)	-0.0013(3)

**Table 4 Bond Lengths for Z-6**

Atom	Atom	Length/Å	Atom	Atom	Length/Å
<b>Au1</b>	<b>Cl1</b>	2.2845(8)	<b>C14</b>	<b>C15</b>	1.389(4)
<b>Au1</b>	<b>P1</b>	2.2276(7)	<b>C15</b>	<b>C16</b>	1.361(5)
<b>C1</b>	<b>C2</b>	1.392(4)	<b>C16</b>	<b>C17</b>	1.391(4)
<b>C1</b>	<b>C6</b>	1.399(4)	<b>C16</b>	<b>F2</b>	1.350(3)
<b>C1</b>	<b>P1</b>	1.799(3)	<b>C17</b>	<b>C18</b>	1.383(4)
<b>C2</b>	<b>C3</b>	1.382(5)	<b>C17</b>	<b>N1</b>	1.451(4)
<b>C3</b>	<b>C4</b>	1.371(6)	<b>C18</b>	<b>F1</b>	1.343(3)
<b>C4</b>	<b>C5</b>	1.379(5)	<b>C19</b>	<b>C20</b>	1.398(5)
<b>C5</b>	<b>C6</b>	1.378(5)	<b>C19</b>	<b>C24</b>	1.380(5)
<b>C7</b>	<b>C8</b>	1.390(4)	<b>C19</b>	<b>N2</b>	1.428(4)
<b>C7</b>	<b>C12</b>	1.382(4)	<b>C20</b>	<b>C21</b>	1.363(5)
<b>C7</b>	<b>P1</b>	1.821(3)	<b>C20</b>	<b>F3</b>	1.348(4)
<b>C8</b>	<b>C9</b>	1.383(4)	<b>C21</b>	<b>C22</b>	1.391(5)
<b>C9</b>	<b>C10</b>	1.373(5)	<b>C22</b>	<b>C23</b>	1.387(4)
<b>C10</b>	<b>C11</b>	1.367(5)	<b>C22</b>	<b>O1</b>	1.355(4)
<b>C11</b>	<b>C12</b>	1.400(4)	<b>C23</b>	<b>C24</b>	1.363(5)
<b>C13</b>	<b>C14</b>	1.393(4)	<b>C24</b>	<b>F4</b>	1.351(4)
<b>C13</b>	<b>C18</b>	1.389(4)	<b>C25</b>	<b>O1</b>	1.423(4)
<b>C13</b>	<b>P1</b>	1.826(3)	<b>N1</b>	<b>N2</b>	1.248(4)

**Table 5 Bond Angles for Z-6**

Atom	Atom	Atom	Angle/°	Atom	Atom	Atom	Angle/°
<b>P1</b>	<b>Au1</b>	<b>Cl1</b>	175.70(3)	<b>C18</b>	<b>C17</b>	<b>N1</b>	121.6(3)
<b>C2</b>	<b>C1</b>	<b>C6</b>	118.7(3)	<b>C17</b>	<b>C18</b>	<b>C13</b>	123.0(3)
<b>C2</b>	<b>C1</b>	<b>P1</b>	119.8(2)	<b>F1</b>	<b>C18</b>	<b>C13</b>	118.8(3)
<b>C6</b>	<b>C1</b>	<b>P1</b>	121.5(2)	<b>F1</b>	<b>C18</b>	<b>C17</b>	118.2(2)
<b>C3</b>	<b>C2</b>	<b>C1</b>	120.4(3)	<b>C20</b>	<b>C19</b>	<b>N2</b>	123.6(3)
<b>C4</b>	<b>C3</b>	<b>C2</b>	120.1(3)	<b>C24</b>	<b>C19</b>	<b>C20</b>	115.1(3)
<b>C3</b>	<b>C4</b>	<b>C5</b>	120.4(3)	<b>C24</b>	<b>C19</b>	<b>N2</b>	120.5(3)
<b>C6</b>	<b>C5</b>	<b>C4</b>	120.1(3)	<b>C21</b>	<b>C20</b>	<b>C19</b>	123.1(3)
<b>C5</b>	<b>C6</b>	<b>C1</b>	120.2(3)	<b>F3</b>	<b>C20</b>	<b>C19</b>	117.4(3)
<b>C8</b>	<b>C7</b>	<b>P1</b>	118.0(2)	<b>F3</b>	<b>C20</b>	<b>C21</b>	119.5(3)
<b>C12</b>	<b>C7</b>	<b>C8</b>	119.7(3)	<b>C20</b>	<b>C21</b>	<b>C22</b>	118.9(3)

<b>C12</b>	<b>C7</b>	<b>P1</b>	122.3(2)	<b>C23</b>	<b>C22</b>	<b>C21</b>	120.2(3)
<b>C9</b>	<b>C8</b>	<b>C7</b>	119.9(3)	<b>O1</b>	<b>C22</b>	<b>C21</b>	115.6(3)
<b>C10</b>	<b>C9</b>	<b>C8</b>	120.2(3)	<b>O1</b>	<b>C22</b>	<b>C23</b>	124.2(3)
<b>C11</b>	<b>C10</b>	<b>C9</b>	120.5(3)	<b>C24</b>	<b>C23</b>	<b>C22</b>	118.1(3)
<b>C10</b>	<b>C11</b>	<b>C12</b>	120.0(3)	<b>C23</b>	<b>C24</b>	<b>C19</b>	124.5(3)
<b>C7</b>	<b>C12</b>	<b>C11</b>	119.6(3)	<b>F4</b>	<b>C24</b>	<b>C19</b>	116.7(3)
<b>C14</b>	<b>C13</b>	<b>P1</b>	123.6(2)	<b>F4</b>	<b>C24</b>	<b>C23</b>	118.8(3)
<b>C18</b>	<b>C13</b>	<b>C14</b>	117.4(3)	<b>N2</b>	<b>N1</b>	<b>C17</b>	119.8(3)
<b>C18</b>	<b>C13</b>	<b>P1</b>	119.1(2)	<b>N1</b>	<b>N2</b>	<b>C19</b>	121.2(3)
<b>C15</b>	<b>C14</b>	<b>C13</b>	121.3(3)	<b>C22</b>	<b>O1</b>	<b>C25</b>	118.0(3)
<b>C16</b>	<b>C15</b>	<b>C14</b>	118.6(3)	<b>C1</b>	<b>P1</b>	<b>Au1</b>	112.66(10)
<b>C15</b>	<b>C16</b>	<b>C17</b>	123.0(3)	<b>C1</b>	<b>P1</b>	<b>C7</b>	105.84(13)
<b>F2</b>	<b>C16</b>	<b>C15</b>	119.9(3)	<b>C1</b>	<b>P1</b>	<b>C13</b>	105.19(13)
<b>F2</b>	<b>C16</b>	<b>C17</b>	117.0(3)	<b>C7</b>	<b>P1</b>	<b>Au1</b>	116.52(9)
<b>C16</b>	<b>C17</b>	<b>N1</b>	120.7(3)	<b>C7</b>	<b>P1</b>	<b>C13</b>	104.90(12)
<b>C18</b>	<b>C17</b>	<b>C16</b>	116.5(3)	<b>C13</b>	<b>P1</b>	<b>Au1</b>	110.84(8)