

Supporting Information

3D Charged Grid Induces a High Performance Catalyst: Ruthenium Clusters Enclosed in X-zeolite for Hydrogenation of Phenol to Cyclohexanone

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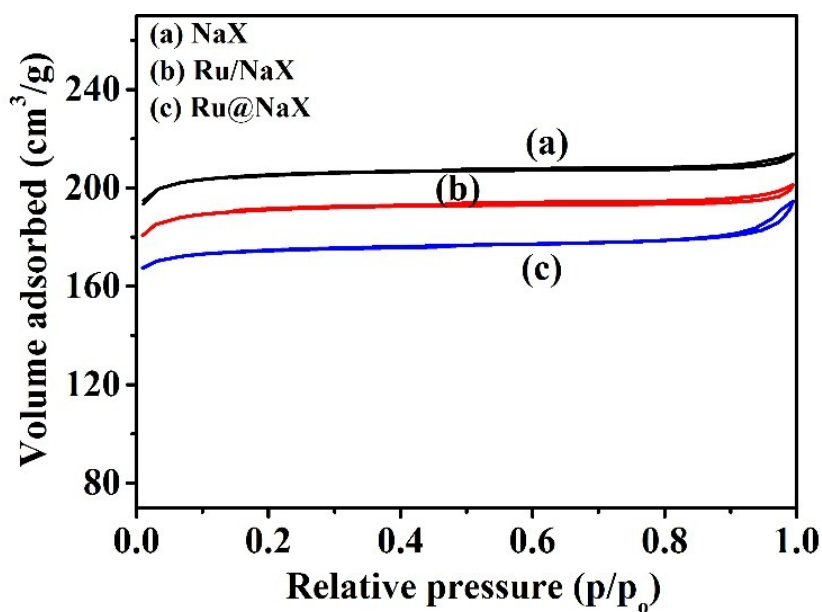


Figure S1. The N₂ adsorption/desorption isotherms of different samples.

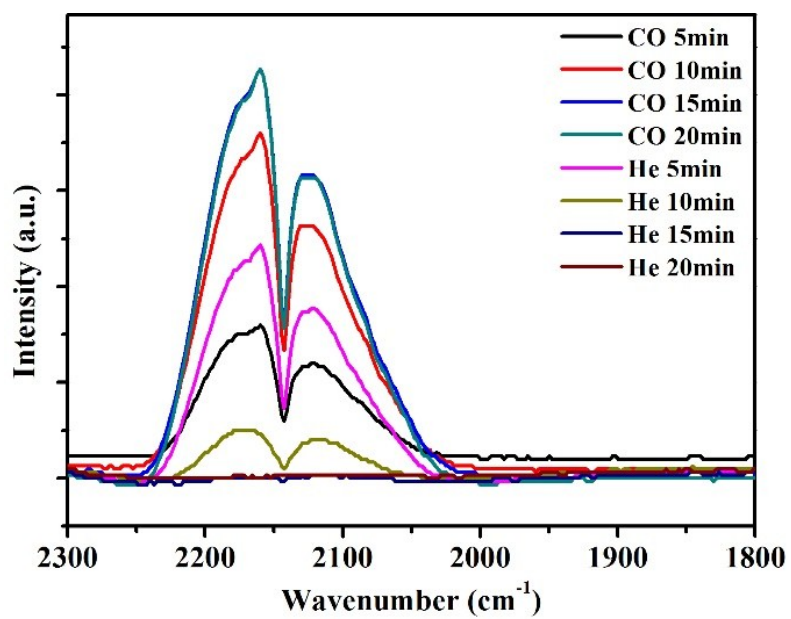


Figure S2. The in-situ DRIFTS of CO adsorption on NaX zeolite. (There are two CO bands located at 2173 cm^{-1} and 2118 cm^{-1} , associated respectively with the formation of OH-CO adducts and gas phase CO.)

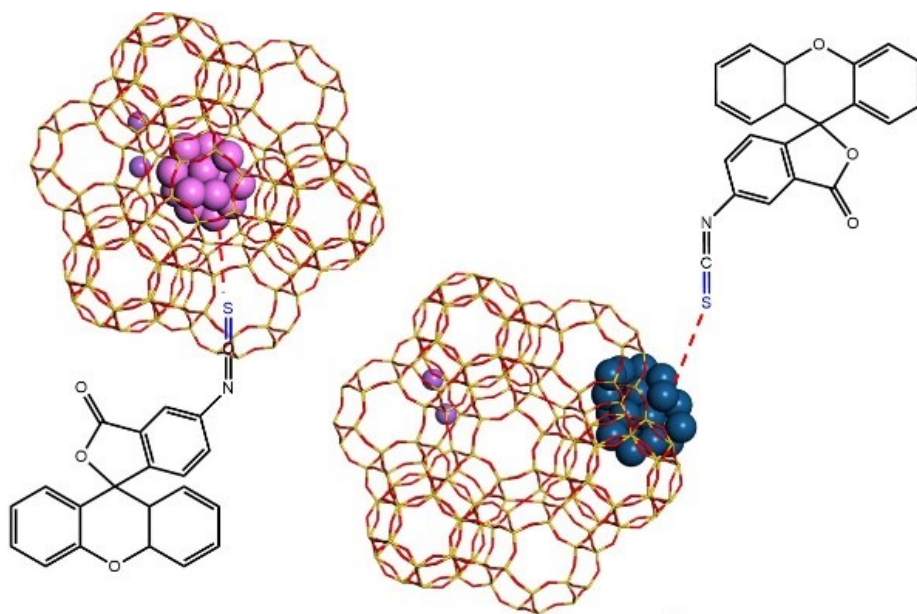


Figure S3. The adsorption model of FITC over the Ru@NaX and Ru/NaX.

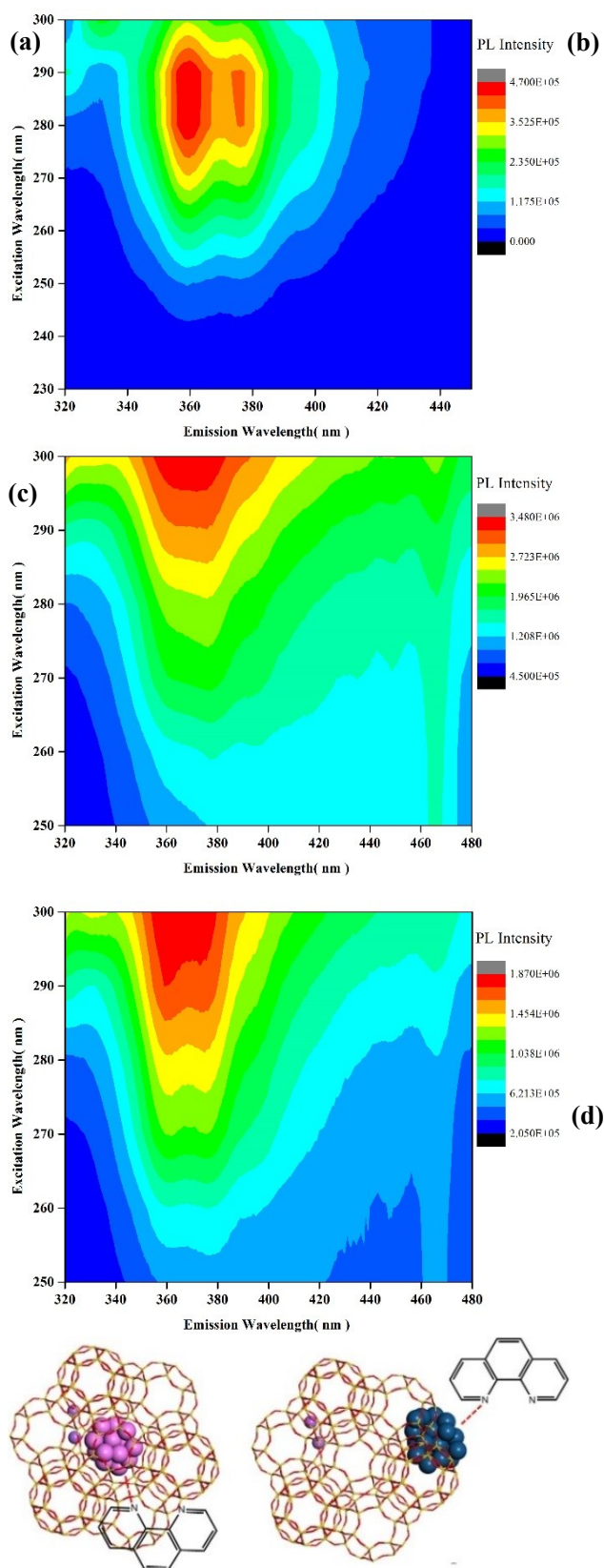


Figure S4. The three-dimensional fluorescence spectrum of (a) 1,10-phenanthroline monohydrate (PM); (b) PM/Ru@NaX; (c) PM/Ru/NaX; (d) The adsorption model of PM over the Ru@NaX and Ru/NaX.

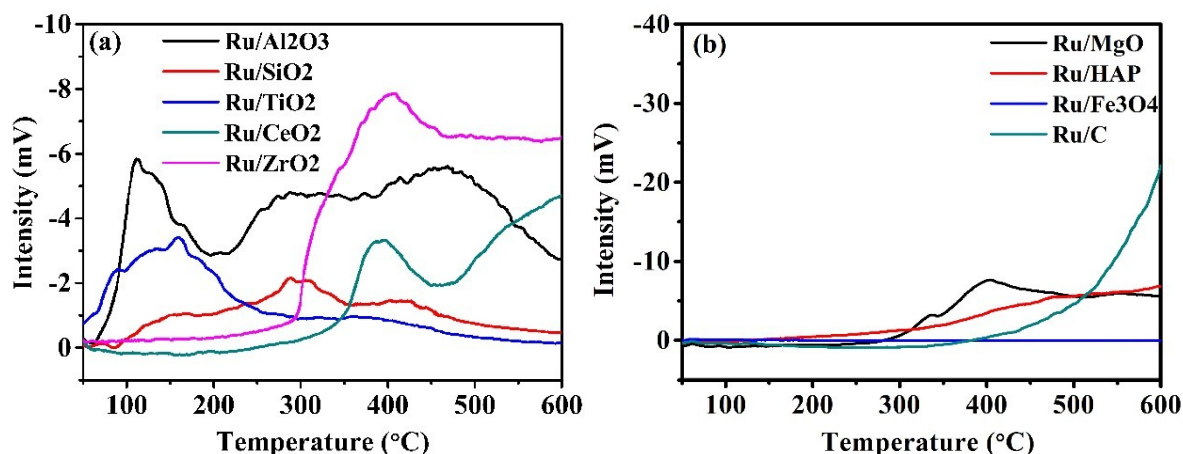


Figure S5. The H₂-TPD experimental curves for different samples.

Table S1. The catalytic performance of phenol selective hydrogenation for ruthenium catalysts in literatures

Catalyst	Loading (%)	Reaction conditions			Catalytic performance			Ref.
		Temperature	Pressure	Time	Solvent	Conversion	Selectivity	
		(K)	(MPa)	(h)		(%)	(%)	
Ru/ γ -Al ₂ O ₃	5.0	433	5.0	2	water	81.5	87.0	1
Ru/MIL-101	5.0	323	0.5	4	water	90.0	90.0	2
Ru/MSN	2.1	323	1.0	1	CH ₂ Cl ₂	1.1	64.9	3
Ru/K-10	~	298	0.4	3	water	23.0	38.5	4
Ru/ γ -Al ₂ O ₃	~	433	5.0	2	cyclohexane	88.0	83.0	5
Ru/C	5.0	328	10.0	2	CO ₂	30.0	5.0	6
Ru@NDCs	3.0	313	0.5	2	water	52.0	2.0	7
Ru@C	5.0	328	10	2	water	30.0	5.0	8

Table S2. The catalytic performance of phenol selective hydrogenation for different catalysts in literatures

Catalyst	Loading (%)	Reaction conditions			Catalytic performance			Ref.
		Temperature	Pressure	Time	Solvent	Conversion	Selectivity	
		(K)	(MPa)	(h)		(%)	(%)	
Pd@sMMT-1	1.5	308	0.1	15	water	99.0	98.0	9

Pd/CeO ₂	1.0	298	0.1	4	water	86.2	96.3	10
Pd@CN	~	318	0.1	4	water	100.0	99.0	11
Pd/C-AlCl ₃	5.0	303	1.0	12	water	99.9	99.9	12
Pd/FDU-N	5.0	373	0.1	1	water	80.0	99.0	13
Pd/UHT	2.0	453	0.1	4	vapour-phase	95.0	85.0	14
Pd/TiO ₂ -C	1.1	373	0.5	~	water	99.0	98.0	15
Pd-NPs	2.0	363	0.1	16	water	99.7	99.5	16
Pd/MIL-101	4.9	323	0.5	2	water	90.0	98.6	17
Pd/Al ₂ O ₃	~	490	no	~	CH ₃ OH/H ₂ O	82.5	77.1	18
Pd-La-B		393	1.0	4	water	79.0	38.0	19
Pd@mpg-C ₃ N ₄	5.0	373	0.1	1	water	99.0	99.0	20

Table S3. Textural properties and Ru content (%) of different samples.

Entry	Sample	S _{BET} ^a (m ² g ⁻¹)	PV ^b (cm ³ g ⁻¹)	PS ^c (nm)	Ru Content (% ICP-AES)
1	NaX	682.2	0.14	2.07	0
2	Ru/NaX	632.3	0.18	2.14	0.57
3	Ru@NaX	584.5	0.19	2.01	0.58
4	Ru/Al ₂ O ₃	134.3	0.45	11.43	0.59
5	Ru/TiO ₂	60.8	0.14	3.45	0.57
6	Ru/SiO ₂	387.3	0.87	2.56	0.59
7	Ru/CeO ₂	92.2	0.17	3.8	0.64
8	Ru/HAP	105.0	0.44	12.8	0.56
9	Ru/ZrO ₂	72.9	0.09	3.6	0.63
10	Ru/C	1510.9	0.98	0.75	0.57
11	Ru/Fe ₃ O ₄	49.5	0.67	10.5	0.55

12	Ru/MgO	55.6	0.54	3.2	0.61
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^a BET surface area, ^b Pore volume, ^c Pore size.

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