

Electronic supplementary information (ESI)

Oxygen vacancy rich Cu₂O based composite material with nitrogen doped carbon as matrix for photocatalytic H₂ production and organic pollutant removal

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X-Ray crystallography

Suitable single crystal of **CuCP** was carefully selected under an optical microscope and glued on a glass fiber. Structural measurement was performed on a Bruker AXS SMART APEX II CCD diffractometer at 293 K. The structure was solved with the direct method and refined by the full-matrix least-squares method on F^2 using the SHELXTL crystallographic software package. Anisotropic thermal parameters were used to refine all non-hydrogen atoms. Carbon-bound hydrogen atoms were placed in geometrically calculated positions; Oxygen-bound hydrogen atoms were located in the difference Fourier maps, kept in that position and refined with isotropic temperature factors. The details have been deposited to the Cambridge Crystallographic Data Centre (CCDC 1812937). The Cambridge Crystallographic Data Centre via www.ccdc.cam.ac.uk/data_request/cif

Table S1. Crystal data and structure refinement results for **CuCP**

Empirical formula	C ₂₀ H ₁₈ CuN ₂ O ₅
Formula weight	429.90
Space group	Monoclinic
Crystal system	C2/c
Z	4
a (Å)	5.6629(3)
b (Å)	19.4888(10)
c (Å)	17.2426(8)
β(°)	99.1860(10)
Volume (Å ³)	1878.54(16)
F(000)	884
Reflections collected	6850
Reflections unique	2326
R(int)	0.0197
Goodness-of-fit	0.884
R ₁	0.0262
wR ₂	0.0971
R ₁ (all data)	0.0299
wR ₂ (all data)	0.1029

Note. $R_1 = \Sigma ||F_o| - |F_c|| / \Sigma |F_o|$; $wR_2 = \Sigma [w(F_o^2 - F_c^2)^2] / \Sigma [w(F_o^2)^2]^{1/2}$

Table S2. Selected bond lengths and angles of **CuCP**

Cu(1)-O(1)#1	1.9538(11)	Cu(1)-O(1)	1.9538(11)
Cu(1)-N(1)	1.9932(13)	Cu(1)-N(1)#1	1.9932(13)
O(1)#1-Cu(1)-O(1)	93.00(7)	O(1)#1-Cu(1)-N(1)#1	93.83(5)
O(1)-Cu(1)-N(1)	93.83(5)	O(1)-Cu(1)-N(1)#1	167.88(5)
O(1)#1-Cu(1)-N(1)	167.88(5)	N(1)-Cu(1)-N(1)#1	81.31(7)

Symmetry transformations used to generate equivalent atoms: #1 -x, y, -z+1/2.

Table S3 Comparison of photocatalytic H₂ production rate reported in the literature with Cu₂O based composite photocatalysts and our works.

Photocatalyst	Catalyst dose (g)	H ₂ Production (μmol·h ⁻¹ ·g ⁻¹)	Ref.
NPs/Cu ₂ O	0.003	47.0	S1
C ₃ N ₄ /Pd/Cu ₂ O	0.015	32.5	S2
g-C ₃ N ₄ /Cu ₂ O	0.300	265.0	S3
NCs/Cu ₂ O	0.150	132.7	S4
HD-Au-NR/Cu ₂ O	0.005	80.2	S5
HD-Au-NR/TiO ₂ /Cu ₂ O	0.005	105.1	S5
Cu ₂ O	0.100	29.78	S6
NCs/Cu ₂ O	0.025	15.0	S7
Cu/Cu₂O@NC(A)	0.020	379.6	Our work

Table 4 The contents of Cu, C, N and H in **Cu/Cu₂O@NC**.

	Cu/Cu₂O@NC(A)	Cu/Cu₂O@NC(B)	Cu/Cu₂O@NC(C)
Cu (%)	4.5	4.4	4.2
C (%)	89.2	89.5	89.9
N (%)	3.0	2.9	2.8
H (%)	1.1	1.3	1.1

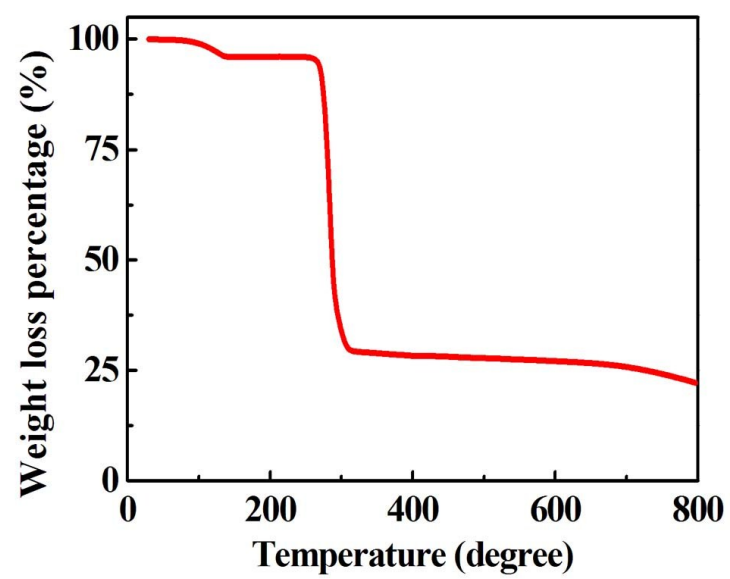


Fig. S1 TGA of CuCP

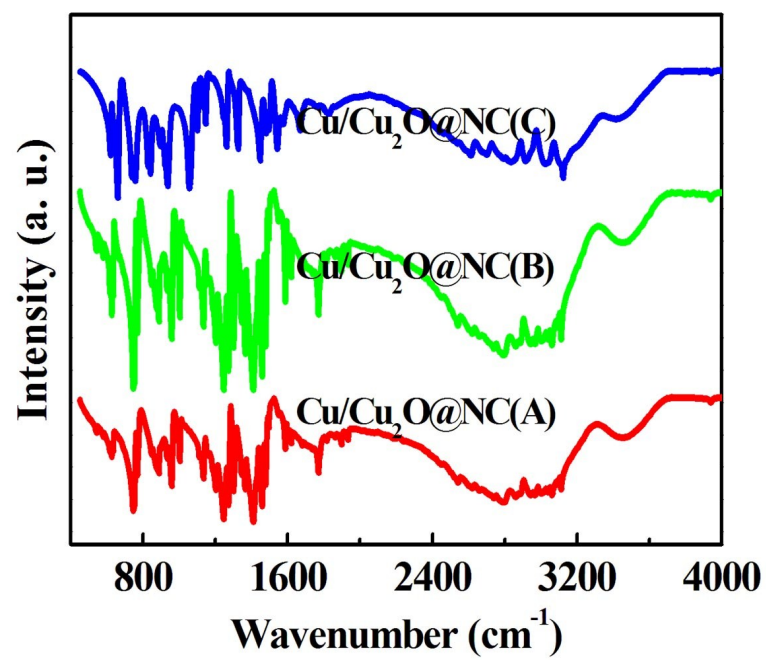


Fig. S2 FTIR of Cu/Cu₂O@NC

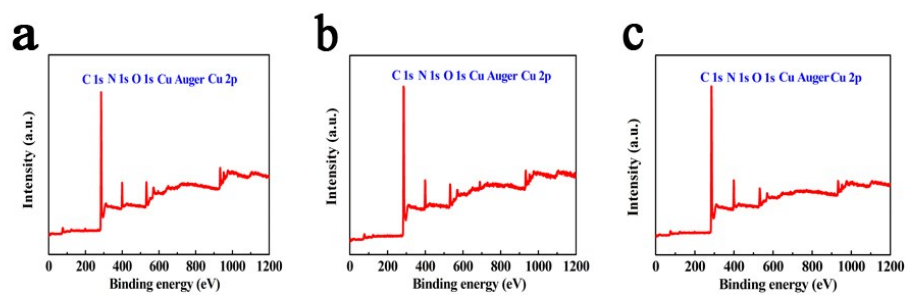


Fig. S3 XPS survey of (a) **Cu/Cu₂O@NC(A)**; (b) **Cu/Cu₂O@NC(B)**; (c) **Cu/Cu₂O@NC(C)**.

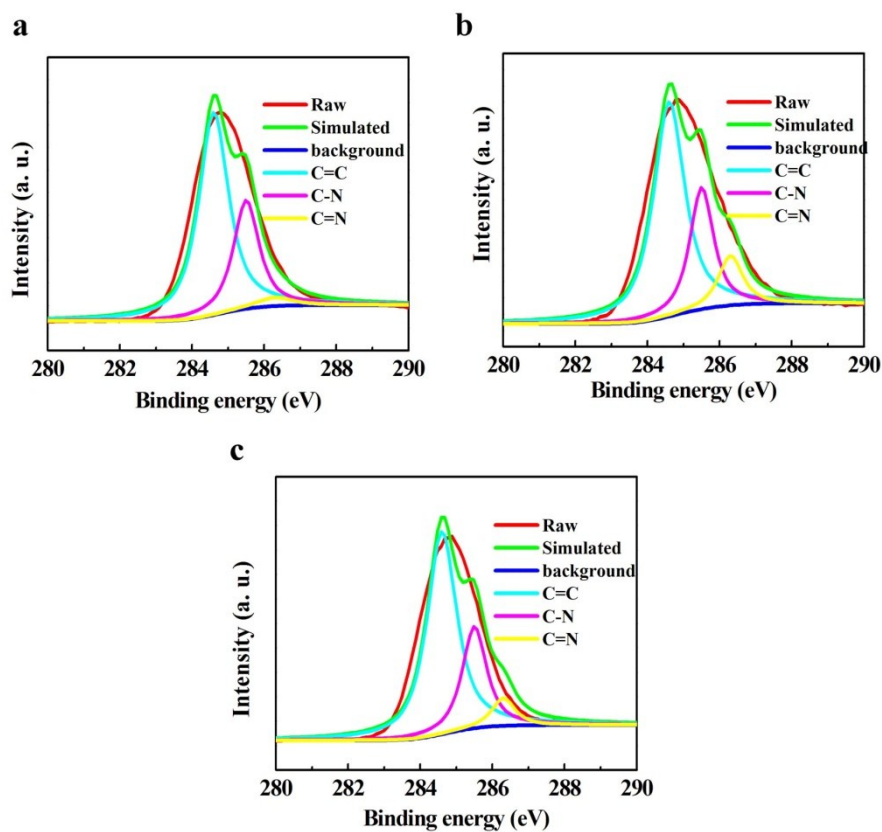


Fig. S4 C1s of (a) Cu/Cu₂O@NC(A); (b) Cu/Cu₂O@NC(B); (c) Cu/Cu₂O@NC(C).

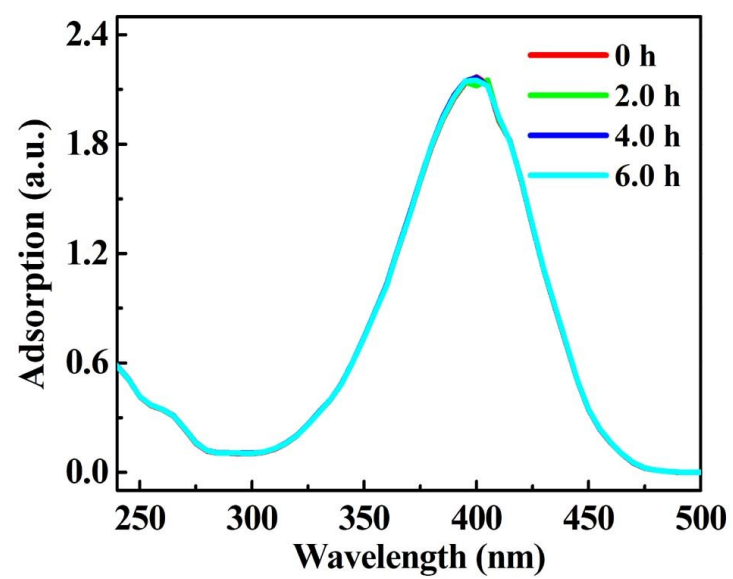


Fig. S5 UV-vis spectra of 4-nitrophenol without **Cu/Cu₂O@NC**.

References

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