

# Water-soluble transition metal complexes of Ruthenium(II), Osmium(II), Rhodium (III) and Iridium (III) with chelating *N*-heterocyclic carbene ligands in hydrogenation and transfer hydrogenation catalysis

## Supporting Information

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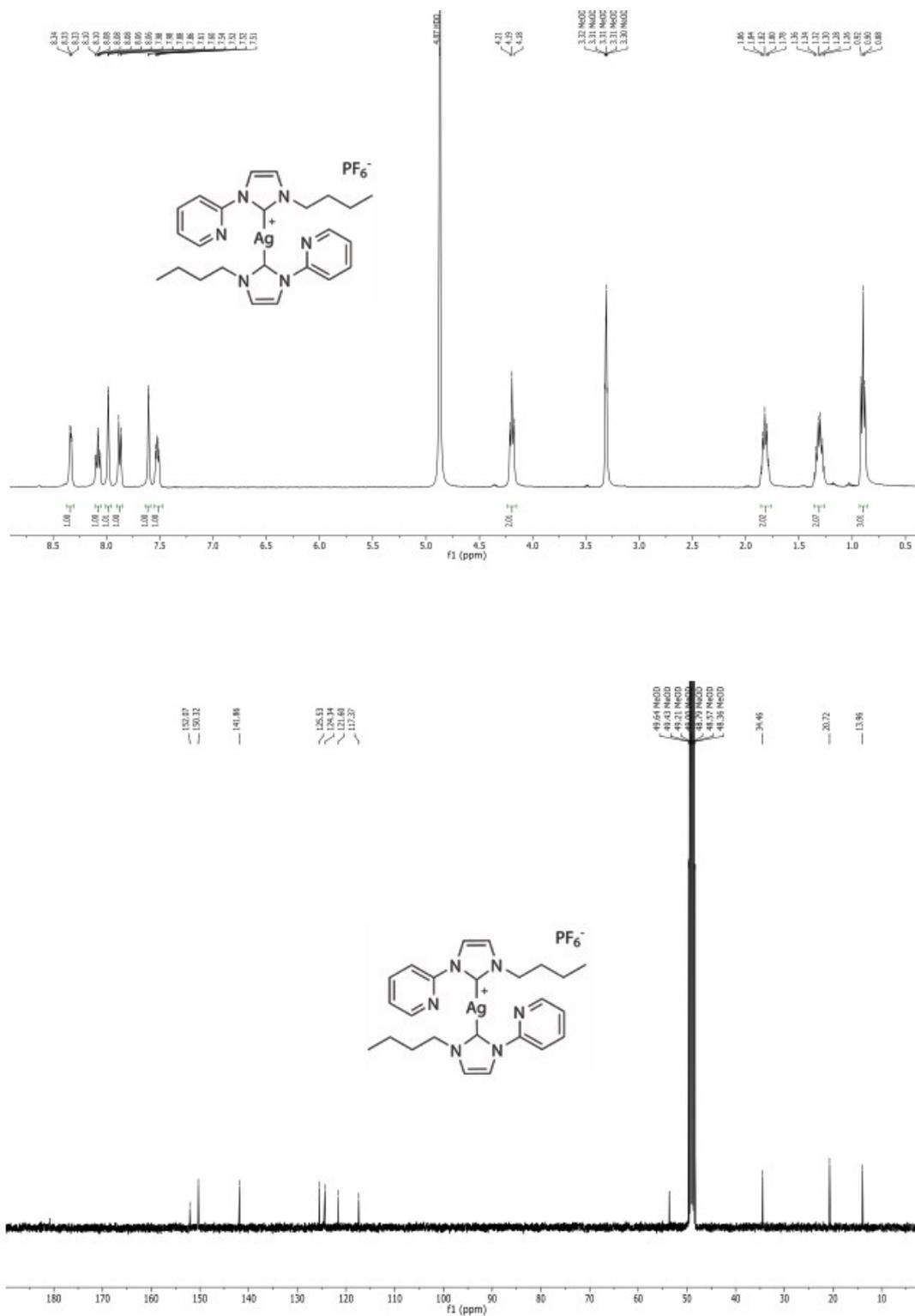
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# 1. $^1\text{H}$ and $^{13}\text{C}$ NMR spectra.

Figure S1.  $^1\text{H}$  (MeOD- $d_4$ , 400MHz) and  $^{13}\text{C}$  (MeOD- $d_4$ , 100 MHz) NMR spectra of **2a**.



**Figure S2.**  $^1\text{H}$  (DMSO- $d_6$ , 400MHz) and  $^{13}\text{C}$  (DMSO- $d_6$ , 100MHz) NMR spectra of **2b**.

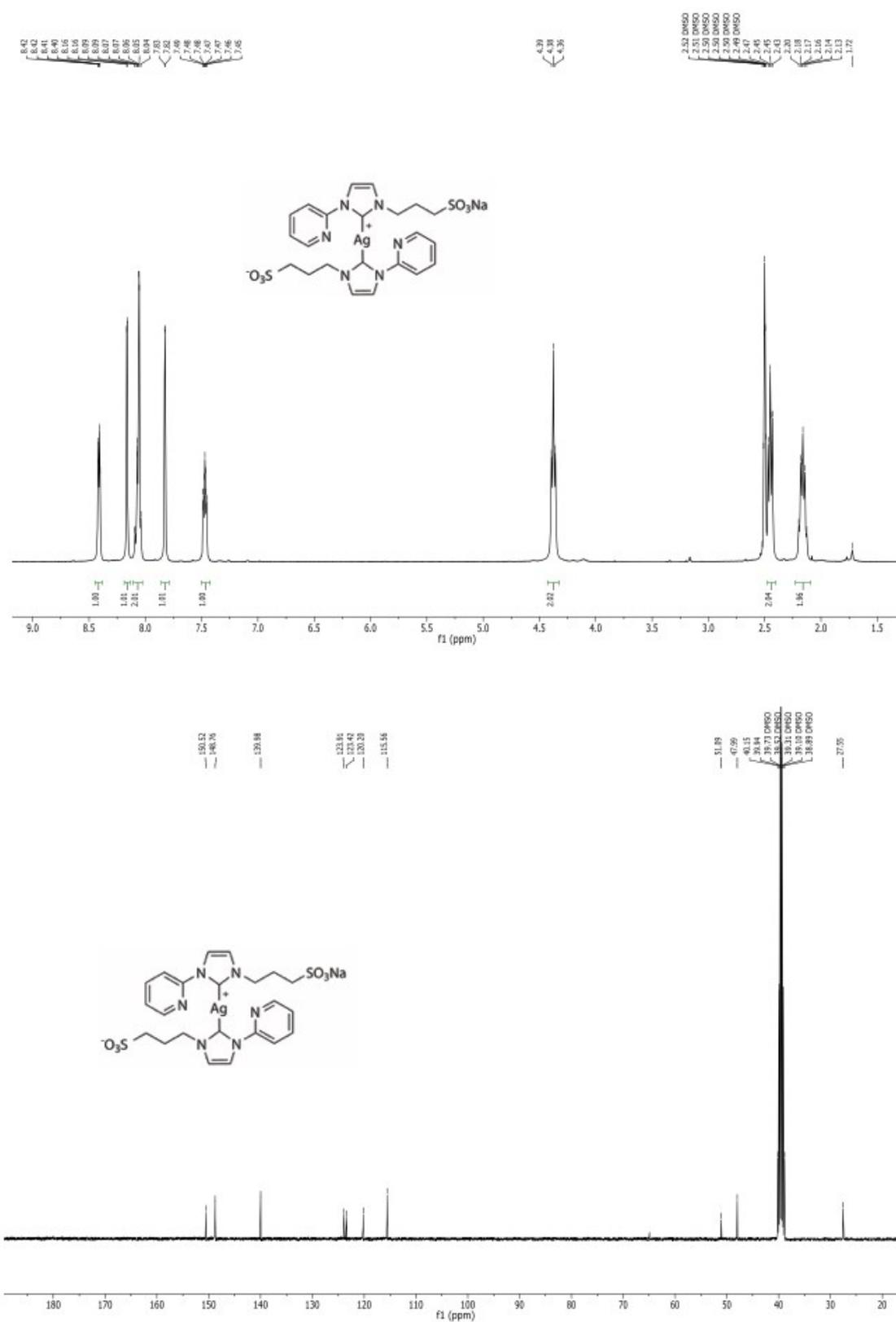
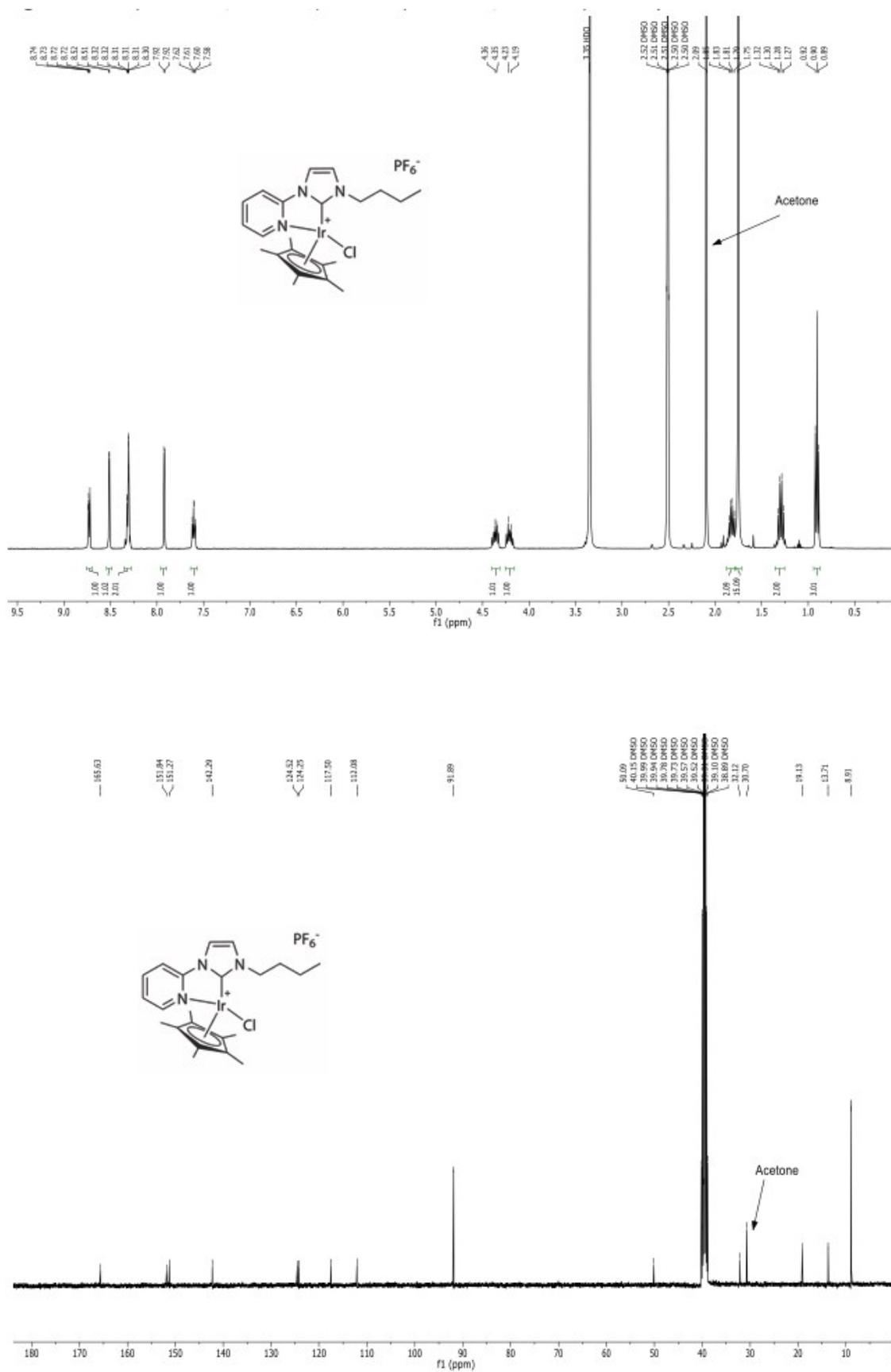








Figure S6.  $^1\text{H}$  (DMSO- $d_6$ , 400MHz) and  $^{13}\text{C}$  (DMSO- $d_6$ , 100MHz) NMR spectra of **6a**.











## 2. X-Ray Crystallographic Data

Data were collected on an X-ray single crystal diffractometer equipped with a CCD detector (APEX II,  $\kappa$ -CCD), a rotating anode FR591 equipped with a Montel mirror optic (**5a**, **5b**, **6a**, **6b**) or a fine focused sealed tube equipped with a graphite monochromator (**2a**, **2b**, **4b**) by using the APEXII software package.<sup>S1</sup> The measurements were performed on single crystals coated with perfluorinated ether. The crystals were fixed on the top of a glass fiber and transferred to the diffractometer. Crystals were frozen under a stream of cold nitrogen. A matrix scan was used to determine the initial lattice parameters. Reflections were corrected for Lorentz and polarization effects, scan speed, and background using SAINT.<sup>S2</sup> Absorption corrections, including odd and even ordered spherical harmonics were performed using SADABS.<sup>S2</sup> Space group assignments were based upon systematic absences,  $E$  statistics, and successful refinement of the structures. Structures were solved by direct methods with the aid of successive difference Fourier maps,<sup>S3</sup> and were refined against all data using the APEX 2 software<sup>S1</sup> in conjunction with SHELXL-97<sup>S4</sup> and SHELXLE.<sup>S5</sup> Methyl hydrogen atoms were refined as part of rigid rotating groups, with a C–H distance of 0.98 Å and  $U_{\text{iso(H)}} = 1.5 \cdot U_{\text{eq(C)}}$ . Other H atoms were placed in calculated positions and refined using a riding model, with methylene and aromatic C–H distances of 0.99 and 0.95 Å, respectively, and  $U_{\text{iso(H)}} = 1.2 \cdot U_{\text{eq(C)}}$ . Non-hydrogen atoms were refined with anisotropic displacement parameters. Full-matrix least-squares refinements were carried out by minimizing  $\sum w(F_o^2 - F_c^2)^2$  with SHELXL-97 weighting scheme.<sup>S4</sup> Neutral atom scattering factors for all atoms and anomalous dispersion corrections for the non-hydrogen atoms were taken from *International Tables for Crystallography*.<sup>S6</sup> Images of the crystal structures were generated by Mercury.<sup>S7</sup> Crystallographic data (including structure factors) for the structures reported in this paper have been deposited with the Cambridge Crystallographic Data Centre as supplementary publication No.'s 1404189-1404195. Copies of the data can be obtained free of charge on application to CCDC, 12 Union Road, Cambridge CB2 1EZ, UK (fax: (+44)1223-336-033; e-mail: [deposit@ccdc.cam.ac.uk](mailto:deposit@ccdc.cam.ac.uk).)

**Table S1.** Crystallographic data for compounds **2a**, **2b**, **5a**, **6a**, **4b**, **5b** and **6b**.

	<b>7</b>	<b>2b</b>	<b>5a</b>	<b>6a</b>	<b>4b</b>	<b>5b</b>	<b>6b</b>
<b>Formula</b>	C <sub>36</sub> H <sub>45</sub> Ag <sub>2</sub> F <sub>12</sub> N <sub>9</sub> P <sub>2</sub>	C <sub>23</sub> H <sub>28</sub> AgN <sub>6</sub> NaO <sub>7</sub> S <sub>2</sub>	C <sub>22</sub> H <sub>30</sub> ClF <sub>6</sub> N <sub>3</sub> PRh	C <sub>22</sub> H <sub>30</sub> ClF <sub>6</sub> IrN <sub>3</sub> P	C <sub>22</sub> H <sub>30</sub> ClN <sub>3</sub> O <sub>4</sub> OsS	C <sub>21</sub> H <sub>31</sub> ClN <sub>3</sub> O <sub>5.1</sub> RhS	C <sub>21</sub> H <sub>31</sub> ClIrN <sub>3</sub> O <sub>5.09</sub> S
<b>fw</b>	1109.49	695.51	619.82	709.13	658.24	577.55	666.72
<b>Colour/habit</b>	brown fragment	colourless fragment	orange fragment	yellow plate	yellow block	orange block	yellow plate
<b>Cryst. dimensions (mm<sup>3</sup>)</b>	0.090 x 0.120 x 0.260	0.080 x 0.250 x 0.310	0.134 x 0.232 x 0.277	0.090 x 0.179 x 0.290	0.186 x 0.230 x 0.370	0.227 x 0.231 x 0.279	0.164 x 0.378 x 0.445
<b>Crystal system</b>	triclinic	triclinic	orthorhombic	orthorhombic	monoclinic	monoclinic	monoclinic
<b>Space group</b>	<i>P</i> -1	<i>P</i> -1	<i>P bca</i>	<i>P bca</i>	<i>P</i> 2 <sub>1</sub> / <i>c</i>	<i>P</i> 2 <sub>1</sub> / <i>n</i>	<i>P</i> 2 <sub>1</sub> / <i>n</i>
<b>a, Å</b>	12.6948(6)	9.0008(3)	14.9552(4)	15.0548(6)	10.5278(2)	11.3895(3)	11.3664(6)
<b>b, Å</b>	13.5073(6)	12.2235(4)	15.0093(4)	14.9891(6)	17.4635(4)	12.1721(3)	12.2127(6)
<b>c, Å</b>	14.4562(7)	13.4457(5)	21.7963(6)	21.8205(9)	13.6800(3)	17.5563(5)	17.5641(9)
<b>α, deg</b>	94.699(2)	71.899(2)	90	90	90	90	90
<b>β, deg</b>	105.537(2)	82.767(2)	90	90	93.009(1)	93.134(1)	93.028(2)
<b>γ, deg</b>	112.654(2)	76.054(2)	90	90	90	90	90
<b>V, Å<sup>3</sup></b>	2155.71(18)	1362.57(8)	4892.6(2)	4924.0(3)	2511.63(9)	2430.26(11)	2434.8(2)
<b>Z</b>	2	2	8	8	4	4	4
<b>T, K</b>	123	123	123	123	123	123	123
<b>D<sub>calcd</sub>, g cm<sup>-3</sup></b>	1.709	1.695	1.683	1.913	1.741	1.579	1.819
<b>μ, mm<sup>-1</sup></b>	1.074	0.964	0.935	5.658	5.299	0.936	5.716
<b>F(000)</b>	1112	708	2512	2768	1296	1187.3	1315

<b><math>\theta</math> range, deg</b>	1.84 – 26.87	2.02 – 25.35	1.87 – 25.35	1.87 – 26.02	1.89 – 26.37	2.04 – 33.14	2.32 – 25.35
<b>Index ranges (<i>h</i>, <i>k</i>, <i>l</i>)</b>	±16, ±17, ±18	±10, ±14, ±16	±18, ±18, ±26	±18, -16 – 18, ±26	±13, ±21, ±17	-17 – 16, ±18, -24 – 27	±13, ±14, ±21
<b>No. of rflns collected</b>	65786	50061	82866	92223	92387	51535	24095
<b>No. of independent rflns/<math>R_{int}</math></b>	9239/0.0717	4965/0.0748	4466/0.0649	4859/0.0476	5125/0.0589	9255/0.0494	4452/0.0461
<b>No. of observed rflns (<math>I &gt; 2\sigma(I)</math>)</b>	6537	3948	3634	4133	4445	8308	4266
<b>No. of data/restraints/parameters</b>	9239/0/ 553	4965/0/363	4466/0/313	4859/0/313	5125/0/294	9255/6/311	4452/6/320
<b>R1/wR2 (<math>I &gt; 2\sigma(I)</math>)<sup>a</sup></b>	0.0390/ 0.0819	0.0303/0.0546	0.0336/0.0669	0.0179/0.0360	0.0227/0.0491	0.0237/0.0595	0.0201/0.0501
<b>R1/wR2 (all data)<sup>a</sup></b>	0.0729/ 0.0927	0.0492/0.0593	0.0497/0.0751	0.0251/0.0384	0.0306/0.0515	0.0281/0.0617	0.0210/0.0506
<b>GOF (on <math>F^2</math>)<sup>a</sup></b>	1.007	1.022	1.113	1.026	1.040	1.071	1.208
<b>Largest diff peak and hole (e <math>\text{\AA}^{-3}</math>)</b>	1.054/ -0.959	0.435/-0.403	0.094/-0.646	0.797/-0.515	1.368/-0.745	0.708/-0.878	1.803/-0.745

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<sup>[a]</sup>  $R1 = \sum(|F_o| - |F_c|) / \sum |F_o|$ ;  $wR2 = \{ \sum [w(F_o^2 - F_c^2)^2] / \sum [w(F_o^2)^2] \}^{1/2}$ ;  $GOF = \{ \sum [w(F_o^2 - F_c^2)^2] / (n-p) \}^{1/2}$

**Table S2.** Selected bond lengths (Å) and bond angles (°) for rhodium and iridium complexes.

Complex		5a <sup>a</sup>	5b <sup>b</sup>	6a <sup>a</sup>	6b <sup>b</sup>	
Bond lengths (Å)	M(1)–C(1)	2.023(3)	2.0263(11)	2.018(3)	2.021(2)	
	M(1)–N(3)	2.128(3)	2.118(10)	2.122(2)	2.107(2)	
	M(1)–Cl(1)	2.4065(8)	2.4053(3)	2.4090(7)	2.4088(6)	
	M(1)–C(12) <sup>b</sup> or M(1)–C(13) <sup>a</sup>	2.155(3)	2.1992(11)	2.165(3)	2.173(2)	
	M(1)–C(13) <sup>b</sup> or M(1)–C(14) <sup>a</sup>	2.213(3)	2.1566(12)	2.232(3)	2.227(3)	
	M(1)–C(14) <sup>b</sup> or M(1)–C(15) <sup>a</sup>	2.236(3)	2.2264(12)	2.250(3)	2.233(2)	
	M(1)–C(15) <sup>b</sup> or M(1)–C(16) <sup>a</sup>	2.154(3)	2.2159(11)	2.163(2)	2.166(3)	
	M(1)–C(16) <sup>b</sup> or M(1)–C(17) <sup>a</sup>	2.181(3)	2.1673(11)	2.180(3)	2.198(2)	
	Bond angles (°)	C(1)–M(1)–N(3)	76.78(12)	76.82(4)	76.33(10)	76.22(9)
		C(1)–M(1)–Cl(1)	85.53(9)	86.28(3)	85.20(7)	86.12(7)
		N(3)–M(1)–Cl(1)	89.03(8)	87.99(3)	86.78(6)	86.02(5)
		C(1)–M(1)–C(12) <sup>a</sup> or C(1)–M(1)–C(13) <sup>b</sup>	117.04(13)	102.54(4)	117.33(11)	124.35(10)
		C(1)–M(1)–C(13) <sup>a</sup> or C(1)–M(1)–C(14) <sup>b</sup>	155.12(13)	113.39(4)	155.12(11)	163.12(10)
C(1)–M(1)–C(14) <sup>a</sup> or C(1)–M(1)–C(15) <sup>b</sup>		154.84(13)	149.56(4)	156.47(10)	150.05(10)	
C(1)–M(1)–C(15) <sup>a</sup> or C(1)–M(1)–C(16) <sup>b</sup>		117.07(13)	162.11(5)	118.51(10)	114.26(9)	
C(1)–M(1)–C(16) <sup>a</sup> or C(1)–M(1)–C(17) <sup>b</sup>		100.50(13)	123.12(5)	101.47(10)	103.85(10)	
Torsion angles (°)		C(1)–N(2)–C(4)–N(3)	-2.2(4)	-0.04(15)	-2.0(3)	-0.05(3)

### 3. References

- S1 APEX suite of crystallographic software. APEX 2 Version 2008.4. Bruker AXS Inc., Madison, Wisconsin, USA, 2008.
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- S4 G. M. Sheldrick, "SHELXL-97", University of Göttingen, Göttingen, Germany, 1998, *or* Sheldrick, G. M. "SHELXL-2014", University of Göttingen, Göttingen, Germany, 2014.
- S5 C. B. Huebschle, G. M. Sheldrick and B. Dittrich, "SHELXLE", *J. Appl. Cryst.*, 2011, **44**, 1281.
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- S7 Mercury: visualization and analysis of crystal structures C.F. Macrae, P.R. Edgington, P McCabe, E. Pidcock, G.P. Shields, R. Taylor, M. Towler, J. van de Streek, *J. Appl. Cryst.* (2006), 39-3, 453-457.