

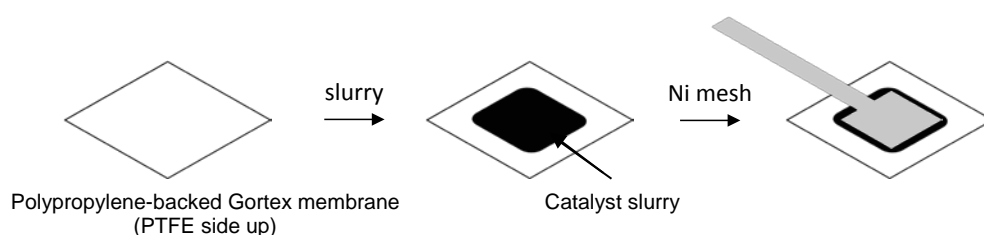
- SUPPLEMENTARY MATERIAL -

# An Electrochemical Cell with Gortex-based Electrodes Capable of Extracting Pure Hydrogen from Highly Dilute Hydrogen-Methane Mixtures

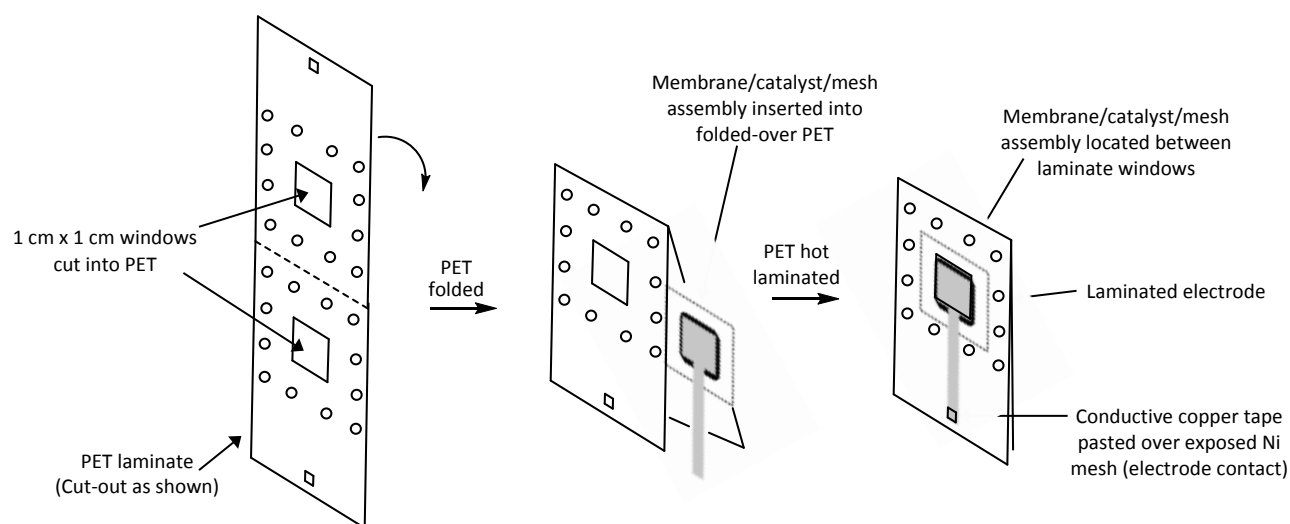
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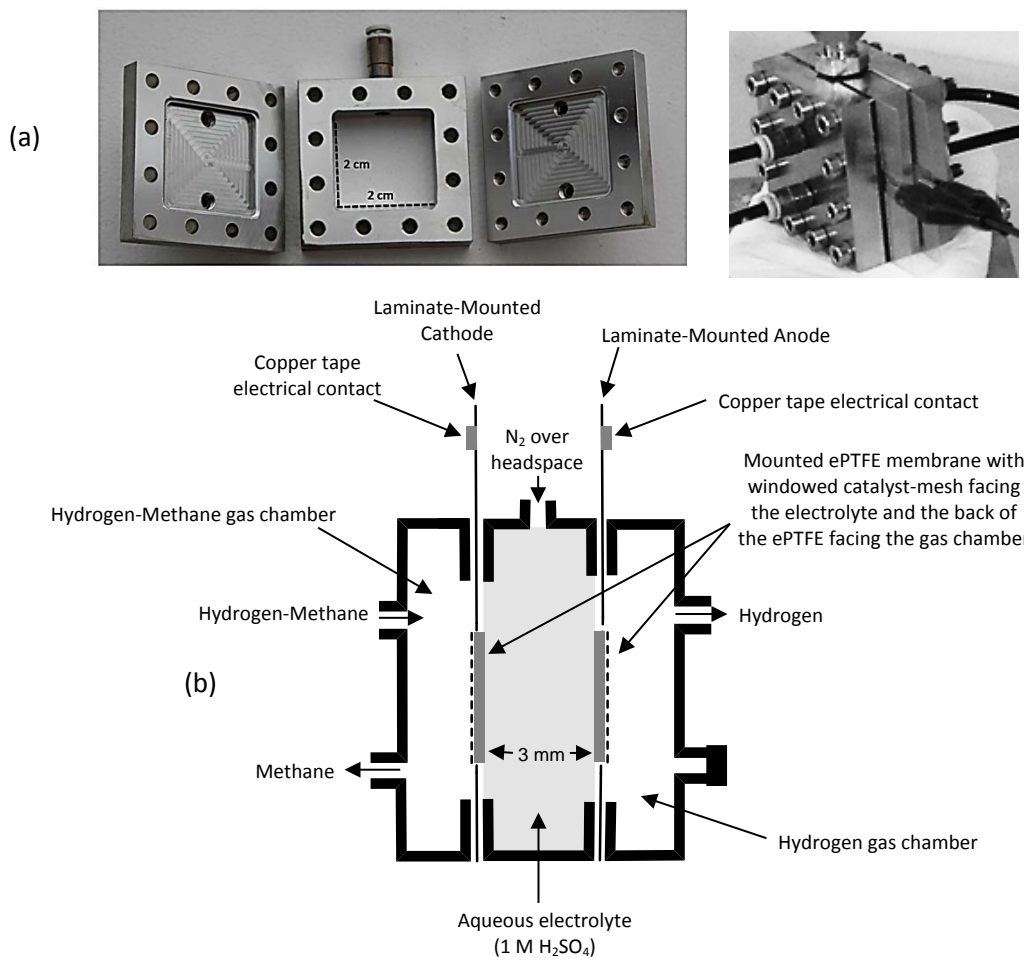
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**Figure S1.** Preparation of slurry/mesh/Gortex membrane assemblies.



**Figure S2.** Preparation of laminate-mounted electrodes.

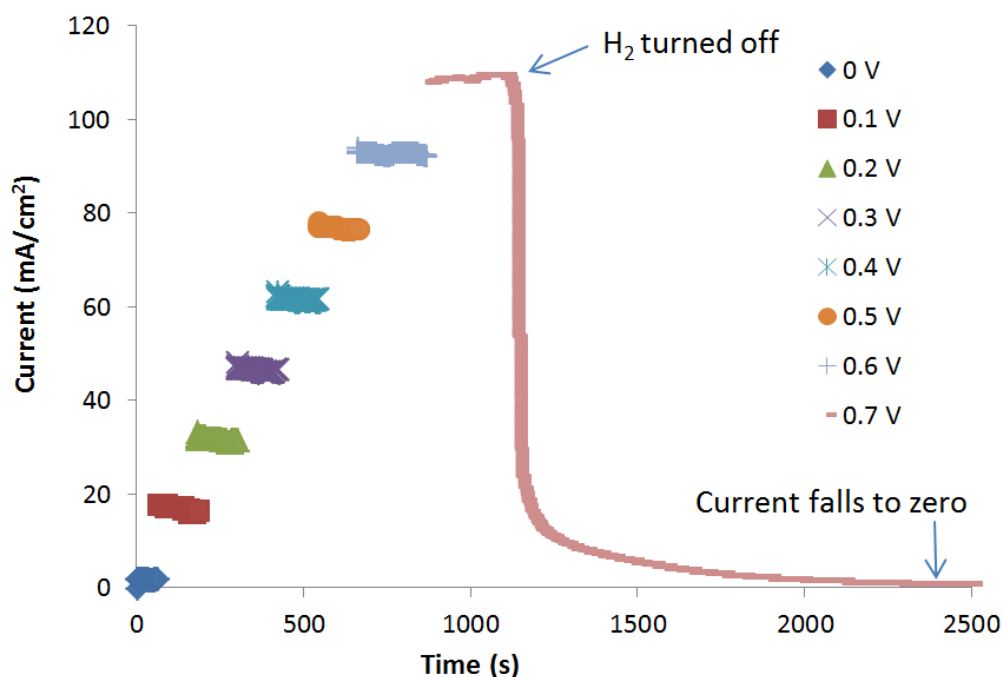


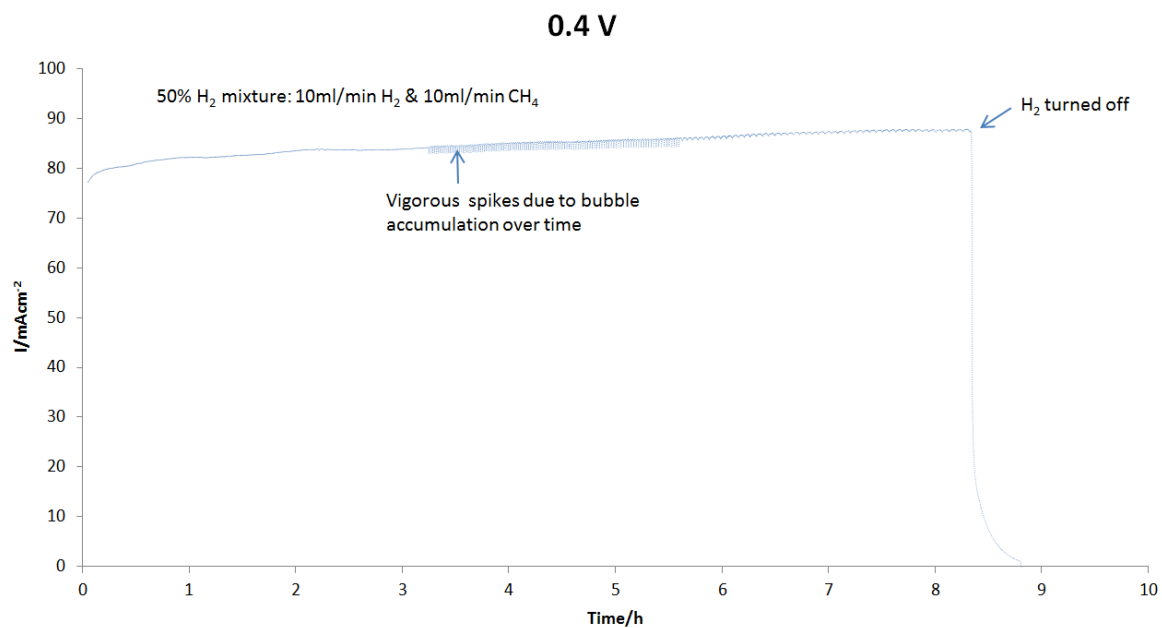
**Figure S3.** Photographs (a) and cross-sectional schematic (b) of the test cell, showing the electrical and gas connections.

Experiment number	Flow conditions at anode	Potential applied	Observation
1	10 mL/min, pure (100%) CH <sub>4</sub>	None (3-electrode; vs Ag/AgCl)	No bubbles due to CH <sub>4</sub> crossover observed at cathode when monitored for 20 min
2	50 mL/min, pure (100%) CH <sub>4</sub>	None (3-electrode; vs Ag/AgCl)	No bubbles due to CH <sub>4</sub> crossover observed at cathode when monitored for 20 min
3	10 mL/min, pure (100%) CH <sub>4</sub>	-0.2 to +0.4 V (3-electrode; vs Ag/AgCl)	No current observed and no bubbles observed at cathode
4	50 mL/min, pure (100%) CH <sub>4</sub>	-0.2 to +0.4 V (3-electrode; vs Ag/AgCl)	No current observed and no bubbles observed at cathode
5	During potential sweep: 10 ml/min CH <sub>4</sub> + 10 ml/min H <sub>2</sub> (i.e. 50% H <sub>2</sub> & 50% CH <sub>4</sub> ).  After potential sweep (at +0.4 V): H <sub>2</sub> flow turned off (i.e. becomes 100% CH <sub>4</sub> )	-0.2 V to +0.4 V (3-electrode; vs Ag/AgCl)	Increasing currents observed with increasing potential. At +0.4V vs Ag/AgCl, when H <sub>2</sub> flow was stopped, the current fell to zero (with CH <sub>4</sub> flow still on).
6	During potential sweep: 10 ml/min CH <sub>4</sub> + 10 ml/min H <sub>2</sub> (i.e. 50% H <sub>2</sub> & 50% CH <sub>4</sub> ).  After potential sweep (at +0.7 V): H <sub>2</sub> flow turned off (i.e. becomes 100% CH <sub>4</sub> )	-0.2 V to +0.7 V (2-electrode)	Increasing currents observed with increasing potential. At +0.7 V, when H <sub>2</sub> flow is stopped, current falls to zero. (Graph shown below)

**Table S1.** Experiments to assess methane reactivity at the anode and methane crossover to the cathode during cell operation.

Graphical data from experiment 6 above:





**Figure S4.** Long-term performance of the cell at 0.4 V (vs. Ag/AgCl) with a mixture of 10 mL/min hydrogen and 10 mL/min methane flowing through the anode.