Supplementary Information:

Mechanisms of Distinct Activated Carbon and Biochar

Amendment Effects on Petroleum Vapor Biofiltration in Soil

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- 10

11 Additional method descriptions:

12 **Batches experiments.** Amber vials (65 mL, Jencons, a VWR Division, Leicestershire, UK) 13 closed with Teflon Mininert valves (Supelco, Bellefonte, USA) were prepared in triplicates, and contained wet construction sand without sorbent amendment (soil), and with 2% d.w. biochar 14 amendment (soil & BC) or 2% d.w. activated carbon amendment (soil & AC). Similarly, 15 16 sterilized controls were prepared by autoclaving the same soil types at 121 °C for 30 min. Water contents of the vials were determined by the weight loss of control vials upon drying in the oven 17 at 105 °C. Three kinds of batch experiments were performed. Firstly, VPH headspace 18 concentrations were monitored after injecting 0.002 mL of VPH mixture as liquid into the 19 batches containing 40, 25 and 13 g of wet construction sand for soil, soil & BC, and soil & AC, 20respectively, to determine VPH vapor biodegradation rates (live soils) and sorption coefficients 21 22 (sterile soils). Secondly, to study the effects of nutrients on the biodegradation of VPHs, VPH 23 residuals were extracted 30 days after the addition of 0.03 mL of VPH mixture to 15 g wet 24 construction sand, equivalent to 0.018 g VPH-carbon, from the various soil types (soil, soil & BC, soil & AC) with and without the addition of 0.0018 g of nitrogen in the form of NH₄Cl 25 and/or 0.00018 g of phosphorus in the form of KH₂PO₄ (i.e., nitrogen and phosphorus were 26 27 added together or separately). Autoclaved controls with and without VPH mixture addition were also set up in parallel. On day 4 and day 6, 10 mL of the headspace air in the nutrient amended 28 29 batches containing soil with or without biochar were replaced with pure air using gas-tight 30 syringes to keep the batch aerobic, and the amount of VPHs removed with the syringe was 31 measured and considered as non-degraded VPHs in the mass balance. Thirdly, batch experiments 32 were performed to investigate soil microbiology. CO₂ was monitored in the headspace of batches 33 with and without 0.03 mL of VPH mixture addition, with and without nutrient addition, or with

the addition of 0.03 mL of equal weight mixtures of either the straight chain alkanes, the 34 branched and cyclic alkanes, or the monoaromatic hydrocarbons. After between 8 and 15 days, 35 soil DNA was extracted for 16s rRNA gene sequencing. Batch study data interpretation followed 36 protocols described by Bushnaf et al.¹. The apparent first-order biodegradation rates in the gas 37 38 phase ^{2, 3} was estimated by linear regression of the natural logarithm of the VPH concentration in 39 soil air versus time in the part of the experimental data that showed a clear decrease, e.g. after an eventual lag phase, and the rate had to be greater than the one observed in the sterile control. 40 Exemplary data is shown in Figure S1 for hexane (no lag phase) and isooctane (lag phase). First-41 42 order biodegradation rates in the water were derived from these measurements as explained in Bushnaf et al.¹. 43

Ammonium quantification. Readily available nutrients were extracted using 1:1 w:w ratios of 44 45 soil:CaCl₂ 0.01 M (CaCl₂ from VWR, Leicestershire, UK) in centrifuge tubes which were shaken for 20 h. The slurries were centrifuged at 4,000 rpm for 10 min and the supernatant was 46 47 immediately filtered through syringe filters (pore size 0.2µm, VWR, Leicestershire, UK). Ammonium determination was carried out using an ammonium reagent kit (Merck no. 48 1.4752.002, Merck, Darmstadt, Germany) and a Spectroquant Nova 60 photometer (Merck, 49 50 Germany). The sample of supernatant solution (5 ml) was put into test tube, and 0.6 ml of sodium hydroxide buffer solution and 0.5 g of EDTA was added. After 5 min of shaking 3 drops 51 52 of sodium nitroprusside indicator were added. 53 Nitrate, sulphate and phosphate quantification. Nitrate, sulphate and phosphate analysis was performed on a Dionex ICS 1000 Ion Chromatography system provided with a conductivity 54 55 detector (Dionex, Sunnyvale, CA, USA). An aliquot of the sample (5 mL) was injected by a

56 Dionex AS40 automated sampler. The separation was performed on an Ionic pac AS14A

57 analytical column (250 mm × 4 mm i.d., Dionex, Sunnyvale, CA, USA). The Ion Chromatograph 58 was held isothermally at 20°C with 8.0 mM Na₂CO₃/1.0 mM NaHCO₃ solution as the mobile 59 phase (flow rate of 1 ml min⁻¹, initial pressure 1800 psi).

Total carbon and total organic carbon. Total carbon (TC) and total organic carbon (TOC)
contents of soil, biochar, activated carbon and soil with 2% biochar or activated carbon were
measured using a Leco CS230 carbon-sulphur analyser (LECO Corporation, Michigan, USA).
Samples for TOC were treated with 50% hydrochloric acid to remove carbonate before
measuring the carbon content.
CO₂, O₂ and SF₆ quantification. GC-MS analysis of CO₂, O₂ and SF₆ was performed on a
Fisons 8060 Gas Chromatograph linked to a Fisons MD800 MS (electron voltage 70 eV,

67 filament current 4A, source current 800uA, source temperature 200°C, multiplier voltage 500V,

68 interface temperature 150°C). The sample (60 μ L) was injected in split mode with a 100 μ L

69 Hamilton gastight syringe. The separation was performed on a HP-PLOT-Q capillary column (30

70 m x 0.32 mm i.d) packed with 20 µm Q phase (Agilent Technologies, Palo Alto, USA). The GC

71 was held isothermally at 35°C with helium as the carrier gas (flow rate of 30 mL min⁻¹, initial

72 pressure 65 kPa, split at 100 mL min⁻¹). The instrument was calibrated using standard CO₂, O₂,

73 and N₂O (Scientific and Technical Gases, Staffordshire, UK) and SF₆ (Sigma –Aldrich, Dorset,

74 UK) gases for a five point calibration.

75 VPH quantification: GC-FID analysis was performed on an Agilent HP-7890 Gas

76 Chromatograph (Agilent Technologies, Palo Alto, USA). The gas sample (60 µL) of headspace

77 gas was injected with a 100 μ L Hamilton gastight syringe. The separation was performed on a

78 HP-5 capillary column (30 m \times 0.25 mm i.d.) with 0.25 μ m film thickness (Agilent

79 Technologies, Palo Alto, USA). The injection port used a split ratio of 10 and was heated to

200°C. The gas chromatography column temperature was held at 30°C for 5 minutes, increased
to 120°C at a rate of 10°C min⁻¹, and then held constant for 6 min with hydrogen as carrier gas.
The instrument was calibrated using dilutions of headspace gas concentrations sampled from
Mininert-valve (Supelco, Bellefonte, USA) closed batches containing the VPH mixture liquid for
a five point calibration.

85 **VPH residual quantification.** To examine VPHs residuals, 15 g soil samples were extracted twice for 24 hours in 10 mL dichloromethane/pentane mixture (40:60 by volume). The combined 86 extracts were passed through a glass chromatographic column (45 cm×10 mm i.d.) pre-packed 87 88 with 3 g activated silica gel topped by 1.5 g sodium sulphate (Sigma–Aldrich, Dorset, UK). Columns were rinsed with an additional 20 mL of the dichloromethane/pentane solvent and all 89 solvent was collected in 40 mL amber vials from which 1 mL was transferred into a 1.0 mL GC-90 91 vial for analysis. Pentane could not be quantified by this method because of incomplete separation from the solvent peak. Compounds in cleaned-up solvent extracts were identified and 92 93 quantified by GC-FID analysis on a Hewlett Packard 5890 series II in split less mode, injector 94 temperature at 280°C. The separation was performed on an Agilent fused silica capillary column $(30 \text{ m} \times 0.25 \text{ mm i.d.})$ coated with 0.25 µm diethyl poly-siloxane (HP-5) phase (Agilent 95 96 Technologies, Palo Alto, USA). The GC temperature was programmed from 50-310°C at 97 5°C/min and held at the final temperature for 20 min with hydrogen as the carrier gas. 98 **Amplicons library preparation for next-generation sequencing assays.** For both 99 pyrosequencing and Ion Torrent assays, individual sequencing libraries were prepared from each DNA sample by amplifying the V4 and most of the V5 regions of the 16S rRNA gene, using 100 101 primer set 515f (5'-GTGNCAGCMGCCGCGGTAA-3') and 926r (5'-

102 CCGYCAATTYMTTTRAGTTT-3'). The primers contained the specific adapters for

pyrosequencing (GS FLX Titanium adapter A, 5' - CGTATCGCCTCCCTCGCGCCATCAG –
3', and adapter B, 5' – CTATGCGCCTTGCCAGCCCGCTCAG – 3') or for the Ion Torrent (Ion
adapter A, 5' - CCATCTCATCCCTGCGTGTCTCCGACTCAG-3', and Ion adapter trP1, 5'CCTCTCTATGGGCAGTCGGTGAT-3'). Samples were differentiated by a unique 8 bp

107 (pyrosequencing) or 12bp (Ion Torrent) barcode, added to the 5' –end of both the forward and

108 reverse primers through a GA linker (pyrosequencing) or to the 5'-end of the forward primers

109 through a GAT spacer (Ion Torrent). Each PCR reaction was composed of 0.5 µl DNA template,

110 0.4 µmol l⁻¹ primers, 0.2 mmol l⁻¹ dNTPs (PCR grade Nucleotide Mix, Roche), FastStart High

111 Fidelity Enzyme Blend (2.5U/reaction), 1.8 mM MgCl₂ in the FastStart High Fidelity Reaction

112 Buffer (Roche Diagnostics GmbH, Mannheim, Germany). The following PCR thermal cycling

113 programme was used: an initial denaturation cycle at 95 °C for 4 minutes; 25 cycles of

114 denaturation at 95 °C for 1 minute, annealing of primers at 55 °C for 45 seconds, and elongation

115 at 72 °C for 1 minute; one final elongation step at 72 °C for 7 minutes. For samples in which the

116 product was not sufficiently concentrated, multiplex PCR were prepared, pooled together and

117 cleaned using the QIAquick PCR purification kit (QUIAGEN, Crawley, UK). For the Ion

118 Torrent, single libraries were further purified and size selected using double-sided solid-phase

119 reversible immobilisation (SPRI) beads (Agencourt AMPure XP system, Beckman Coulter),

120 following the manufacturer's instructions. The single libraries were then individually quantified

121 using a Qubit dsDNA HS Assay on a Qubit 2.0 Fluorometer (Invitrogen, UK), following the

122 manufacturer's instruction, pooled together in equimolar amounts and sequenced on a Roche 454

123 GS FLX+ System by NewGene Ltd (International Centre for Life, Newcastle upon Tyne, UK) or

124 (for the batch study in which the biodegradation of the straight chain alkanes, the branched and

125 cyclic alkanes, and the monoaromatic hydrocarbons in soil was investigated separately) on a

126 Personal Genome Machine (PGM) at the School of Civil Engineering and Geosciences, 127 Newcastle University, using a 316 ion chip. The raw sequences were processed using the QIIME (v 1.7.0) bioinformatics pipeline. Briefly, the reads were filtered for quality (filtering criteria: 128 perfect match to sequence barcode/primer, 200bp minimum sequencing length). For 129 pyrosequencing data, QIIME Denoiser ⁴ was used to detect and correct sequencing errors, and 130 131 the data were reintegrated into the QIIME pipeline by inflation. Sequences were then clustered into Operational Taxonomic Unit (OTU) at 97% sequence similarity level by the uclust 132 133 algorithm. A representative sequence from each OTU was selected and taxonomically identified using the Greengenes database ^{5, 6}. Representative sequences and correspondent taxonomic 134 assignment were used to build a table of OTU abundances at different levels of taxonomy. 135 Microbial community diversity within and across the samples was determined by using the 136 137 QIIME (v.1.7.0) pipeline ⁷ and Primer v6 software ⁸. The Bray Curtis similarity metric was generated and represented by a 2-dimensional non-metric multidimensional scaling (nMDS) plot. 138 Tracer test. To examine the tortuosity factor of the packed columns, 10 mL of SF₆ gas were 139 injected into the column center at port 4. The soil air was sampled through the injection point and 140 at a sampling port at 15 and 30 cm distance on either side (ports 3 and 5), after 20 min, 40 min, 141 142 60 min, 80 min, 100 min, 130 min and 160 min. Headspace purging was stopped for the duration 143 of the tracer experiment. To avoid artifacts due to the spatial limitation of the experimental 144 system, the duration of the experiments was no longer than 160 min. Samples of 60 μ L of soil 145 gas were taken with a gas-tight syringe and injected directly into the gas chromatograph for SF_6 analysis by mass spectrometry. The measured concentration of SF₆ was compared with the 146 147 modelled concentration which was obtained by interpreting the diffusive transport with an 148 equation for an instantaneous plane source of sulphur hexafluoride (SF_6) in a porous medium as

- 149 described by Werner and Hohener ⁹. Simulated SF₆ concentration when using τ =0.4, 0.6, and 0.8
- 150 were compared with measured concentrations, and $\tau=0.8$ provided the best fit. Therefore, $\tau=0.8$
- 151 was used as the tortuosity value to simulate the volatilization and migration of VPHs or CO₂.
- 152 Such a high value of the tortuosity factor may be due to the fact that the columns were relatively
- 153 loosely packed in comparison with natural soils.

154 Modelling

155

156 Underpinning assumptions:

The VPH concentrations in soil air above the NAPL source are described by Raoult's law. The 157 158 VPH concentration on the atmospheric boundary side of the columns is zero, and this results in a net diffusive VPH flux from the NAPL source towards the atmospheric boundary. The VPHs 159 migrate through the column by Fickian diffusion in soil air. Advective gas transport is neglected. 160 161 The distribution of VPHs between soil air and water is described by Henry's law. VPH sorption to the soil and biochar or activated carbon solid matrix is linear and reversible. The kinetics of 162 the VPH uptake or release by biochar or activated carbon is described by sorption-retarded 163 164 intraparticle diffusion of VPHs in the biochar or activated carbon pores, which are water-filled. The VPH degrading biomass is present in the water in between, but not within particles, and can 165 simultaneously grow on all the VPH compounds. This biomass can only degrade VPHs which 166 167 are present in the water in between soil particles. Sorbed VPHs and VPHs in soil air are not directly bioaccessible. The growth rate of VPH degrading biomass is dependent on the VPH 168 concentration in the water in between particles, and limited by a maximum biomass 169 concentration. 170

171

- 172 Coordinates, time:
- 173 z (cm) Distance from the pollution source zone
- Time 174 t(s)
- 175

176 Soil and carbonaceous sorbent parameters:

177	$\theta_{a}, \theta_{w}, \theta_{s}, \theta_{c}$ (-)	Air, water, soil solid and carbonaceous sorbent-filled soil volume fractions
178	$\rho_{\rm s} ({\rm gcm}^{-3})$	Density of the soil solids
179	$\tau_{a}(-)$	Tortuosity factor for the air-filled soil pore space

180	ρ_{c} (gcm ⁻³)	Skeletal, solid density of the carbonaceous sorbent
181	p _{w,c} (-)	Water-filled porosity of the carbonaceous sorbent
182	$R_{c}(cm)$	Particle radius of the carbonaceous sorbent
183		
184	Chemical parameters	for compound k:
185	$C_{a,k}(gcm^{-3})$	Concentration of compound k in soil air
186	$S_{c,k}$ (gcm ⁻³)	Volumetric concentration of compound k in the carbonaceous sorbent
187	H _k (-)	Dimensionless air-to-water partitioning coefficient (Henry constant)
188	$K_{d,k}$ (cm ³ g ⁻¹)	Soil solid-to-water partitioning coefficient
189	$K_{c,k} (cm^3g^{-1})$	Carbonaceous sorbent solid-to-water partitioning coefficient
190	$D_{a,k}(cm^2s^{-1})$	Molecular diffusion coefficient of compound k in soil air
191	$D_{w,k}$ (cm ² s ⁻¹)	Molecular diffusion coefficient of compound k in water
192	$D_{c,k}(cm^2s^{-1})$	Apparent diffusion coefficient of compound k in the carbonaceous sorbent
193	x _k (-)	Mole fraction of compound k in the VPH source
194	$C_{pl,k}$ (gcm ⁻³)	Pure liquid vapor concentration of compound k
195		
196	Biological parameters	<u>S:</u>
197	C_b (gcm ⁻³)	Biomass concentration in soil water
198	C _{b,max} (gcm ⁻³)	Maximum biomass concentration in soil water
199	dec (s ⁻¹)	Biomass decay rate
200	$g_{k}(s^{-1})$	Biomass growth rate on compound k
201	$Y_k(gg^{-1})$	Biomass yield coefficient for compound k
202	$sg_k(cm^3g^{-1}s^{-1})$	Second order growth rate on compound k
203		
204	Column parameters:	
205		
206	r _{column} (cm)	Column radius
207	l _{column} (cm)	Column length
208	r _{neck} (cm)	Source neck radius
209	l_{neck} (cm)	Source neck length
210		
210		

211 Model equations:

212

213 Compound k mass balance equation:

215
$$\left(\theta_{a} + \frac{\theta_{w}}{H_{k}} + \frac{\rho_{s}\theta_{s}K_{d,k}}{H_{k}} \right) \cdot \frac{\partial}{\partial t} C_{a,k}(z,t) = \theta_{a}\tau_{a}D_{a,k} \cdot \frac{\partial^{2}}{\partial z^{2}} C_{a,k}(z,t)$$

$$- \theta_{c} \cdot \frac{\partial}{\partial t} \left[\frac{3}{R_{c}^{3}} \int_{0}^{R_{c}} r^{2}S_{c,k}(z,r)dr \right] - \theta_{w} \cdot \frac{g_{k}}{Y_{k}} \cdot C_{b}(z,t)$$
217

218 if time > lag phase

219

220
$$g_k = sg_k \cdot \frac{C_{a,k}(z,t)}{H_k} \cdot \frac{C_{b,max} - C_b(z,t)}{C_{b,max}}$$

221 else

222

$$223 \quad g_k = 0$$

- 224
- 225 Carbonaceous sorbent intraparticle diffusion equation ¹⁰:

226

227
$$\frac{\partial}{\partial t} S_{c,k}(z,r,t) = \frac{D_{c,k}}{r^2} \cdot \frac{\partial}{\partial r} \left(r^{2\partial} / \frac{\partial r}{\partial r} S_{c,k}(z,r,t) \right)$$

228

229 Estimated intraparticle diffusion coefficient ¹⁰:

230

231
$$D_{c,k} = \frac{p_{w,c}^{2}}{p_{w,c} + (1 - p_{w,c}) \cdot \rho_{c} \cdot K_{c,k}} \cdot D_{w,k}$$

232

233 Biomass equation:

$$\frac{\partial}{\partial t} C_b(z,t) = \left(\sum_k g_k - dec\right) \cdot C_b(z,t)$$
235
236

237 Zero-concentration boundary condition at the column outlet:

$$238 \quad C_{a,k}(z = l_{column}, t) = 0$$

239

240 Boundary condition at the NAPL source:

241
$$C_{a,k}(NAPL \text{ source,}t) = x_k(t) \cdot C_{pl,k}$$

242

243 Mass conservation at the NAPL source – soil interface:

245
$$r_{column}^{2} \cdot \theta_{a} \cdot \tau_{a} \cdot \frac{\partial}{\partial z} C_{a,k}(z=0,t) \approx r_{neck}^{2} \frac{C_{a,k}(z=0,t) - C_{a,k}(NAPL \ source,t)}{l_{neck}}$$

246

247 Boundary condition at the soil water – carbonaceous sorbent interface:248

249
$$S_{c,k}(z,R_{c},t) = (p_{w,c} + (1-p_{w,c}) \cdot \rho_c \cdot K_{c,k}) \cdot \frac{C_{a,k}(z,t)}{H_k}$$

250 Boundary condition in the carbonaceous sorbent particle core:

251

$$252 \quad \frac{\partial}{\partial r} S_{c,k}(z,0,t) = 0$$

253

	Apparent first-order	Distribution coefficient K_d (cm ³ g ⁻¹)				
	biodegradation rate					
	in water, $k_w (s^{-1})$					
	Soil	Soil	Soil & 2% BC	Soil & 2% AC		
n-Pentane	6.5±0.9×10 ⁻³	53 ±16	86±19	23200±3000		
n-Hexane	5.3±0.2×10 ⁻³	68±6	112±24	133000±27000		
n-Octane	8.1±1.9×10 ⁻³	167±77	334±190	53300±19900		
n-Decane	8.9±2.9×10 ⁻³	805±355	1091±437	39800±6900		
n-Dodecane	2.5±0.6×10 ⁻²	1714±490	2151±295	206900±5800		
Methylcyclopentane	1.2±0.07×10 ⁻³	13±2	18±1	10600±700		
Methylcyclohexane	5.5±0.6×10 ⁻⁴	15 ±2	26±9	28800±8800		
Cyclohexane	6.0±0.3×10 ⁻⁴	6±0.8	8±1	3600±260		
Isooctane	1.1±0.06×10 ⁻²	83 ±14	124 ± 32	249000±76000		
Toluene	6.9±0.9×10 ⁻⁵	1.3±0.2	12±5	170±59		
m-Xylene	4.8±1.2×10 ⁻⁵	1.8±0.6	6±2	59±20		
1,2,4 TBM	1.6±0.7×10 ⁻⁴	4.4±2.0	10±3	59±10		

255 Table S1: Compound-specific parameter values determined in batch experiments.

Table S2: Summary of OTUs distinguished at 97% similarity that showed a minimum 16-fold increase in their relative abundance rank fifteen days after the addition of 0.3 mL of different VPH compound classes to soil batches.

Taxon (highest level	Relative abundance rank increase for different treatments compared to the soil only control							
classification)	+ 2-fold, ++	4-fold, +++ 8-fold, ++++ 16-fold, +-	++++ 32-fold, +++++ 64-fold, +++	-++++ 128-fold, , +++++++ 320-				
	fold or great	ter, equivalent decrease for minus sig	ns					
	Soil Rank	Soil RankSoil+monoaromatic hydrocarbonsSoil+straight chain alkanesSoil+branched and cyclic alkanes						
gRhodococcus	665	+	++++					
gRhodococcus	g_Rhodococcus 249 + ++++++ +							
Nocardioidaceae 107 + ++++								
s_Desulfosporosinus_meridiei	153	+	++++					
g_Tepidibacter	532		+++++					
gPhenylobacterium	782		+++++					
g_Polaromonas	185	+	++++++	+				
g_Perlucidibaca	749		++++	+				
f_Pseudomonadaceae 593 + +++++++		++++++	+++++					
f_Pseudomonadaceae	230		+++++	++++				
g_Pseudomonas	428	+	+++++++	+++++++				
g_Pseudomonas	232	++	+++++++	+++++++				
s_Pseudomonas_umsongensis 264 + +++++ +++++				+++++				
g_Pseudoxanthomonas	eudoxanthomonas 600 +++++							

Table S3: Summary of OTUs distinguished at 97% similarity which showed a minimum 8-fold increase in their relative abundance rank eight days after the addition of 0.3 mL of the VPH mixture to batches with soil, soil + biochar, or soil + activated carbon.

Taxon (highest level	Relative abundance rank increase for different treatments compared to the soil only control									
classification)	+ 2-fold, ++ 4-fold, +++ 8-fold, ++++ 16-fold, +++++ 32-fold, ++++++ 64-fold or greater, equivalent decrease for									
	minus signs	minus signs								
	Soil d0	oil d0 Soil+VPHs Soil+VPHs Soil+BC Soil+BC Soil+AC Soil+AC								
	Rank		+nutrients	+VPHs	+VPHs+	+VPHs	+VPHs+			
					nutrients		nutrients			
f_Nocardioidaceae	65			-		+++++	+++++			
g_Paenisporosarcina	615	+++++	+++++	++++	++++	+++				
gSphingopyxis	615	+	+++	+	+					
o_Burkholderiales	615	++	++	+++	+++	+++	+			
g_Achromobacter	93	+++++	+++++	++++++	++					
g_Acidovorax	615	++	+++++	++++	++++	++++	+			
gHydrogenophaga	222			++++	+++	++	+			
g_Cupriavidus	562	+	+	+		+	+++			
f_Pseudomonadaceae	615	+	++++	+++	+	+				
gPseudomonas	24	+++++	+++++	+	++	+	-			
s_Pseudomonas_umsongensis	139	+++++	++++	++	++	+	-			
g_Arenimonas	615	++	++	+++	+++	+				

Table S4: Summary of OTUs distinguished at 97% similarity which showed a minimum 8-fold increase in their relative abundance rank 430 days after the connection of a VPH NAPL source to columns filled with soil, soil & biochar, or soil & activated carbon.

Taxon (highest level classification)	n) Relative abundance rank increase for different treatments compared to the soil only control									
	+ 2-fold,	++ 4-fold, +++	8-fold, ++++ 16	5-fold, ++++ 3	2-fold, ++++++	64-fold or greater	r, equivalent deci	ease for minus	signs	
	Soil d0	Soil+VPHs	Soil+VPHs	Soil+VPHs	Soil+BC	Soil+BC	Soil+BC	Soil+AC+	Soil+AC	Soil+AC+
	Rank	d430	d430	d430	+VPHs d430	+VPHs d430	+VPHs d430	VPHs d430	+VPHs d430	VPHs d430
		source	middle	Atmos	source	middle	atmos	source	middle	atmos
gNitrosopumilus	72	++++					-	-		
gMicrobacterium	458			+++	+		+			
f_Nocardioidaceae	94			+			+	+++		+
g_Aeromicrobium	495			+	+	+	+		+	+++
g_Pseudonocardia	204	+	+	++	++	++	+++	+++++	+++++	+++++
o_Bacteroidales	629	+++++	+++	+++	+++	+++++	+++++	+	+	
f_Chitinophagaceae	86			++++		+				+
f_Flammeovirgaceae	9			+++	+	++	+	+		+
c_Chloroflexi	458	++	++	++	++	++	++	+++	+++	+++
f_Phyllobacteriaceae	131			+++			+		+	++
g_Oleomonas	629			++++						+
gSphingobium	80			++++			+			+
f_Comamonadaceae	79	++	++	+++	+	++	++	++	+++	+++
gHydrogenophaga	116	+	++	++++	++++	++++	+++++	++	+++	+++++
oMyxococcales	17			+++	-	-			++	++++
f_Haliangiaceae	36				+	+	++	++	++	+++
gNannocystis	420	+	+		-	-	-	++++	+++	++
f_Alteromonadaceae	629	+	+		+	++	+++		++	+
g_HB2-32-21	272	++	++++	++	+++	++++	++++	+	++++	+++++
f_Chromatiaceae	629	++++	+++	++	++	++++	++++	+		
o_PYR10d3	181			+++		+	+++++			
f_Pseudomonadaceae	166	+	+++	+	+++	+++++	++++	-	-	
f_Sinobacteraceae	105			++++	+	+		+	++	+
g_Pseudoxanthomonas	235			+++	+	+	++	+	++	++

Table S5: Measured column parameters, and *parameters obtained from the literature*¹¹

	Volume fraction air	Volume fraction	Tortuosity	Weight fraction
	(-)	water (-)	(-)	sorbent amendment
				(gg ⁻¹)
Soil	0.44	0.11	0.8	0
Soil & BC	0.48	0.12	0.8	0.02
Soil & AC	0.48	0.11	0.8	0.02

Soil-filled column length 93 (cm); soil-filled column radius 3.9 (cm); source neck length 10 (cm); source neck
radius 0.55 (cm); soil solids density 2.5 (gcm⁻³); *BC solid matrix density* 1.2 (gcm⁻³)¹¹; *AC solid matrix density*

radius 0.55 (cm); soil solids density 2.5 (gcm-3); BC solid matrix density 1.2 (gcm-3) 11; AC solid matrix density 1.96 (gcm⁻³)¹¹; BC and AC intraparticle porosity 0.5 (cm³cm⁻³); BC and AC particle radius 0.005 cm.

278 Table S6: Physicochemical model parameter values from the literature^a

Compound	Molecular	Pure	Henry's	Diffusion	Diffusion	Carbon	Initial
-	weight	liquid	constant	coefficient	coefficient	mass	moles in
	(g/mol)	vapour	(-)	in water	in air	fraction	source
		conc.		(gcm^{-2})	(gcm ⁻²)	(-)	(-)
		(g/cm^3)					
Pentane	72	0.002	50	0.000013	0.082	0.83	0.0086
Hexane	86	0.00065	70	0.000012	0.074	0.84	0.0131
Octane	114	0.000089	120	0.000009	0.064	0.84	0.0097
Decane	142	0.000014	293	0.000008	0.057	0.84	0.0142
Dodecane	170	0.000002	293	0.000007	0.051	0.85	0.0052
Methylcyclopentane	84	0.00060	15	0.000012	0.079	0.86	0.0131
Methylcyclohexane	98	0.00020	4.2	0.000011	0.073	0.86	0.0186
Cyclohexane	84	0.00042	7.3	0.000012	0.080	0.86	0.0147
Isooctane	114	0.00027	132	0.000009	0.064	0.83	0.0181
Toluene	92	0.00014	0.26	0.000011	0.078	0.91	0.0075
m-Xylene	106	0.000047	0.26	0.000010	0.072	0.91	0.0080
1,2,4-TMB	120	0.000015	0.27	0.000009	0.066	0.90	0.0087

²⁷⁹ ^aBased on ¹, pure liquid vapor pressures at 25°C reported in ³.

Table S7: Biogeochemical model parameter values measured in batch experiments

Compound	Soil-water	Biochar-water	AC-water	Yield coefficient
	partitioning	partitioning	partitioning	(g biomass-C g ⁻¹ VPH)
	coefficient K _d	coefficient K _{bc}	coefficient Kac	
	$(cm^{3}g^{-1})$	$(cm^{3}g^{-1})$	$(cm^{3}g^{-1})$	
Pentane	53	1596	1157000	0.2183
Hexane	68	2131	6647000	0.2209
Octane	167	8180	2656000	0.2209
Decane	805	13479	1949000	0.2209
Dodecane	1714	20152	10258000	0.2236
Methylcyclopentane	13	237	529000	0.2262
Methylcyclohexane	15	535	1439000	0.2262
Cyclohexane	6	94	180000	0.2262
Isooctane	83	1965	12446000	0.2209
Toluene	1.3	533	8433	0.2393
m-Xylene	1.8	208	2858	0.2393
1,2,4-TMB	4.4	276	2726	0.2367

Table S8: Biogeochemical model parameter values measured in batch experiments or *obtained from the literature* ¹², and fitted from the column study data.

<i>from the iterature</i> ¹² , and fitted from the column study data.								
Compound	Second-order	Second-order	Second-order	Second-order				
	growth rate, soil	growth rate, soil	growth rate, soil &	growth rate, soil &				
	batch experiments	column, fitted	BC column, fitted	AC column, fitted				
	$(cm^3 g^{-1}s^{-1})$	$(cm^3 g^{-1}s^{-1})$	$(cm^3 g^{-1}s^{-1})$	$(cm^3 g^{-1}s^{-1})$				
Pentane	97	97	97	292				
Hexane	80	80	80	240				
Octane	122	122	122	367				
Decane	134	134	134	403				
Dodecane	382	382	382	1146				
Methylcyclopentane	18.5	5.6	3.7	55.6				
Methylcyclohexane	8.5	2.6	1.7	25.5				
Cyclohexane	9.3	2.8	1.9	27.8				
Isooctane	166	50	33	498				
Toluene	1.1	1.7	2.3	3.4				
m-Xylene	0.79	1.2	1.6	2.4				
1,2,4-TMB	2.6	3.9	5.2	7.8				

288 Initial VPH degrading biomass = 0.0000146 (g biomass C/cm³ water), based on the initial cell count and ¹². Soil

biomass carrying capacity derived from available nitrogen contents = 0.0011 (g biomass C/cm³ water); Soil biomass 290 carrying capacity fitted from the soil column study = 0.00058 (g biomass C/cm³ water); *biomass decay rate* from ¹²

 $291 = 0.00000025 (s^{-1})$; biomass decay rate fitted from the soil column study = $0.00000050 (s^{-1})$; lag phase all

292 compounds from batch experiments = 259200 (s); lag phase branched and cyclic alkanes fitted from the soil column 293 study = 518400 (s)

293 study =
$$518400$$
 (s)

294





Figure S2: Carbon-content normalized VPH concentrations measured in the middle of the soil column for the sum of the a) straight-chain alkanes, b) branched/cyclic alkanes, c) monoaromatic hydrocarbons, and d) total VPH-carbon emanation from the soil column. Inserts show the early data for the first two months. Measured data is shown as diamond symbols. Solid lines show the model predictions for the unamended soil column (solid blue line) based on parameter values which were independently measured or obtained from the literature (see Tables S5-8).



Figure S3: Carbon-content normalized VPH concentrations measured in the middle of the soil column for the sum of the a) straight-chain alkanes, b) branched/cyclic alkanes, c) monoaromatic hydrocarbons, and d) total VPH-carbon emanation from the soil column. Inserts show the early data for the first two months. Measured data is shown as diamond symbols. Solid lines show the model predictions for the unamended soil column (blue), and the soil & biochar column (red) and soil & AC column (green) based on microbiological parameter values optimized for the soil column.





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