Electronic Supplementary Material (ESI) for Green Chemistry.

Facile synthesis of nitrogen-doped porous carbon spheres

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Table S1: Literature review on synthesis conditions employed to synthesize carbon spheres using polymers.

Precursor	Solvent	Catalyst	Temperature/Time	Refs.
Resorcinol/formaldehyde	Ethanol/water	NH_4OH	10-60°C, 24h 100°C, 24h (autoclave)	[1]
Resorcinol/formaldehyde, colloidal silica	Ethanol/water	NH_4OH	30° 24h 100°C 24h (autoclave)	[2;3]
Dopamine	Ethanol/water	NH_4OH	30h at room T	[4]
Dopamine Hydrochloride	Ethanol//water isopropyl alcohol/ethylene glycol	-	> 3 days	[5]
3-aminophenol/ formaldehyde	Ethanol/water	Urea	30°-6h	[6]
Melamine Resorcinol Formaldehyde	Water	-	40-80°C, 1h 80-180°C, 24h (autoclave)	[7]
Melamine Formaldehyde	Water	Formic acid	80°C	[8]
Phloroglucinol Terephthaladehyde Resorcinol Formaldehyde	Water	NH_4OH	70°C, 15-480 min 100°C, 24 (autoclave)	[9]
Resorcinol/ formaldehyde	Ethanol/water	Ethylenediamine	30°C, 24h 100°C, 24h (autoclave)	[10]
Resorcinol/ formaldehyde	Ethanol/water	NH_4OH, Cysteine	30-90°C, 24h 100°C, 24h (autoclave)	[11;12]
Tannin	Water	NH_4OH Para-toluenesulphonic acid	180-220°C, 24h (autoclave)	[13;14]
Resorcinol/ formaldehyde	Water	L-lysine NH_4OH	20°C, 1h 60°C, 10 min	[15]

Figure S1: Different steps for the synthesis of the polymer and carbon spheres.

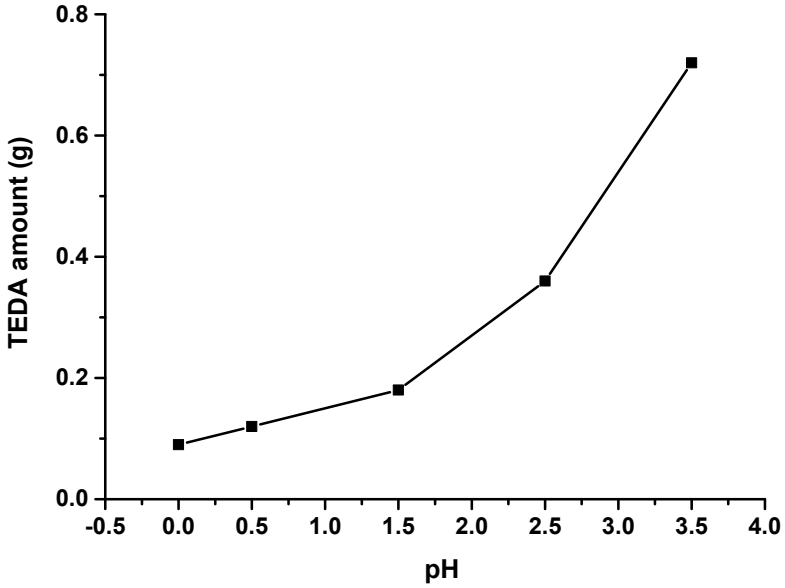

Figure S2: Evolution of the pH of solution with the amount of TEDA.

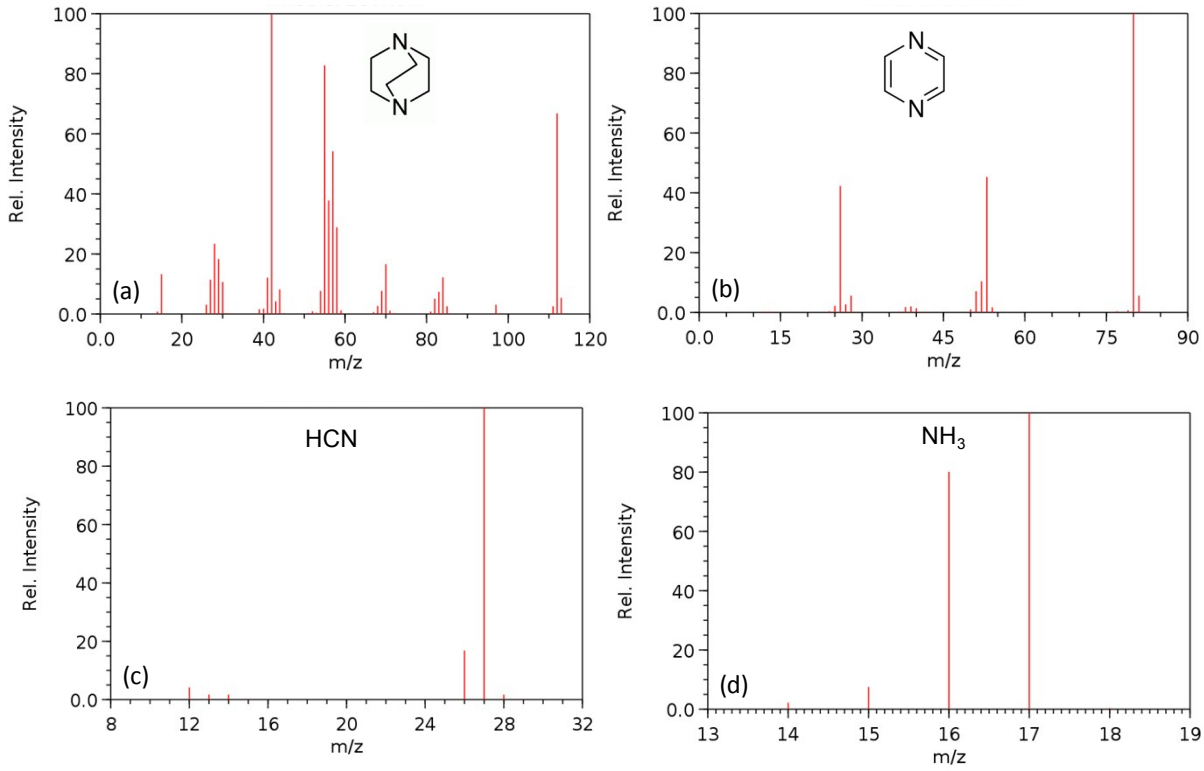

Figure S3: Mass spectra of (a) triethylenediamine (b) pyrazine and (c) hydrogen cyanide and (d) NH_3 (spectra obtained NIST Chemistry WebBook: http://webbook.nist.gov/chemistry).