

**Electronic Supplementary Information (ESI)**

**To**

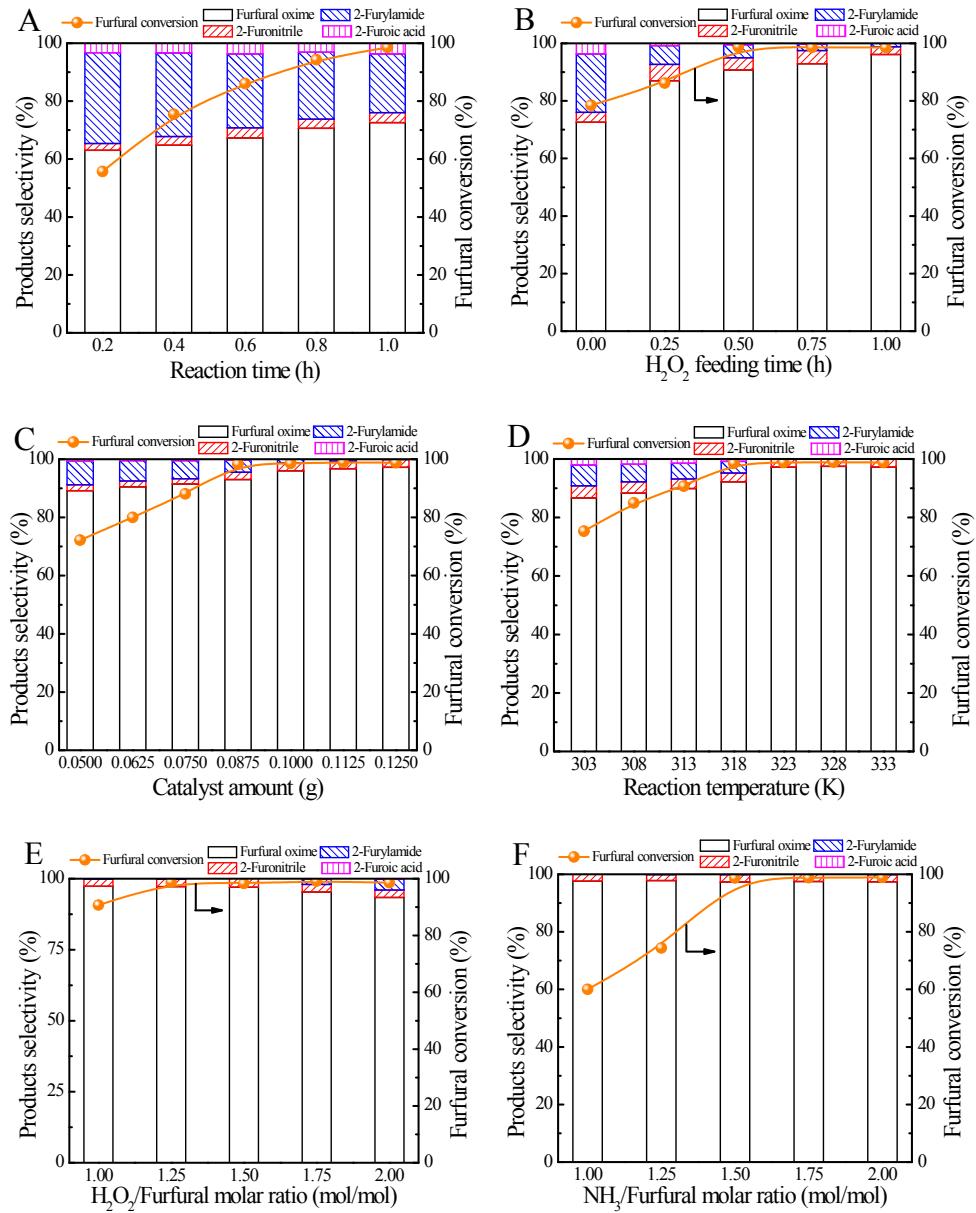
**Clean synthesis of furfural oxime through liquid-phase ammoximation of  
furfural over titanosilicate catalysts**

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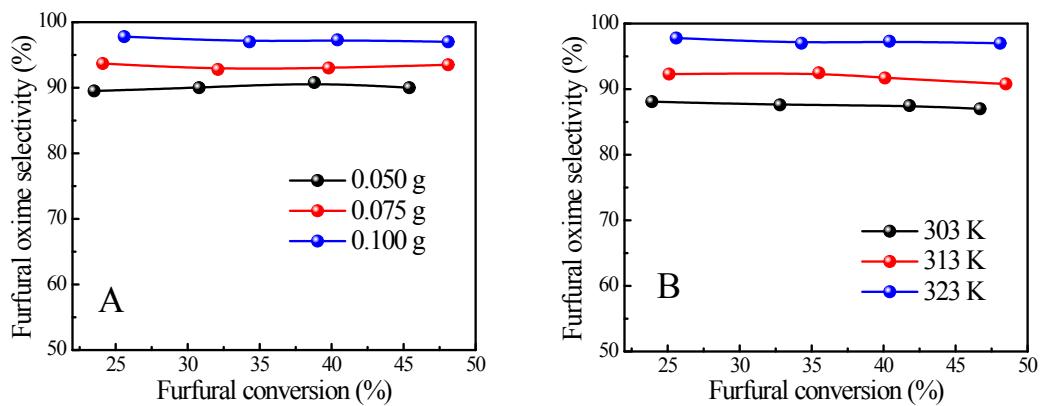
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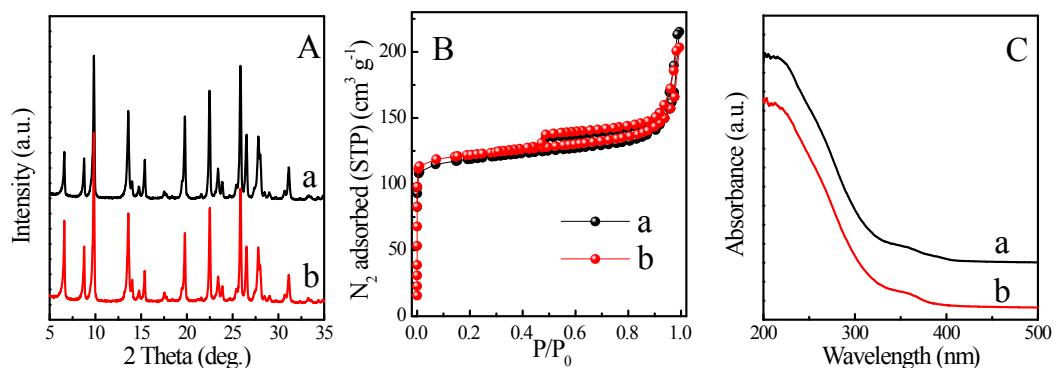
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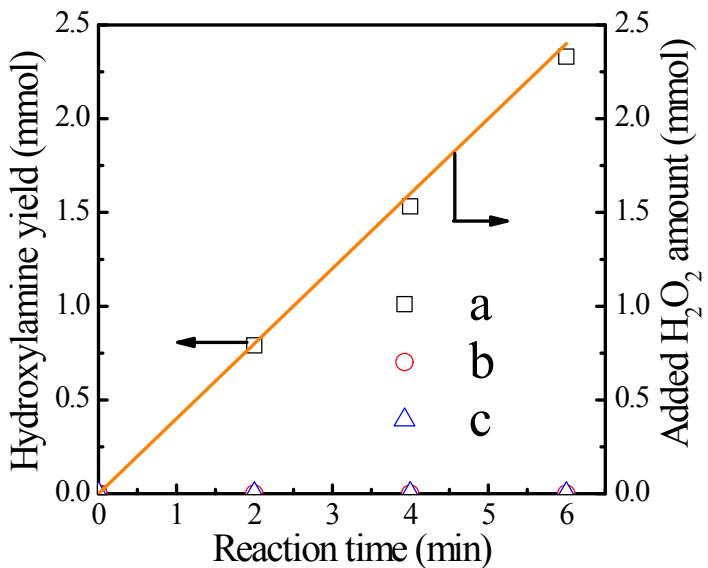
**Fig. S1.** Effect of reaction time (A), feeding time of H<sub>2</sub>O<sub>2</sub> (B), catalyst amount (C), reaction temperature (D), H<sub>2</sub>O<sub>2</sub>/furfural molar ratio (E) and NH<sub>3</sub>/furfural molar ratio (F) on furfural ammoniation over Ti-MOR. Reaction conditions: Ti-MOR, 0.1 g (A, B), 0.125 g (D, E, F); H<sub>2</sub>O, 12 g; furfural, 20 mmol; H<sub>2</sub>O<sub>2</sub> (31 wt. %), 30 mmol (A), 25 mmol (B, C, D, F); NH<sub>3</sub> (25 wt. %), 30 mmol; temp., 323 K; total time, 1 h. H<sub>2</sub>O<sub>2</sub> was fed at once (A); The feeding time of 0 h stands for that H<sub>2</sub>O<sub>2</sub> was fed at once, while the feeding time of 0.25, 0.50, 0.75 and 1.00 h stand for that H<sub>2</sub>O<sub>2</sub> was dropwise added at different feeding rate of 9.6, 4.8, 3.2 and 2.4 mL h<sup>-1</sup>, respectively (B); H<sub>2</sub>O<sub>2</sub> was dropwise added within 1 h (C, D, E, F).



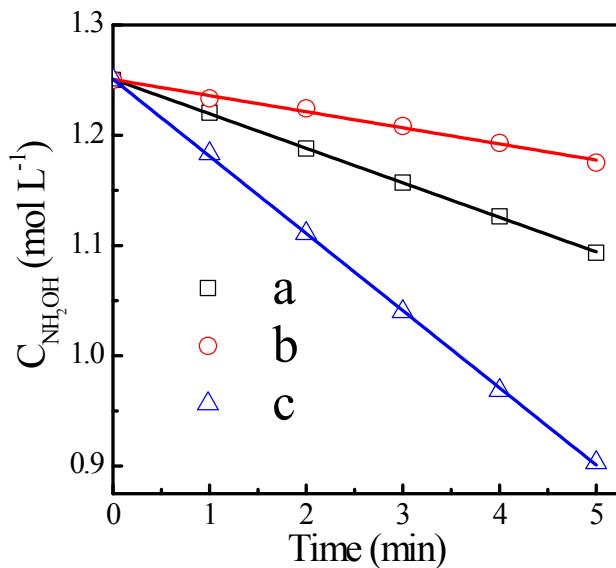
**Fig. S2.** Effect of catalyst amount (A) and reaction temperature (B) on the furfural oxime selectivity. Reaction conditions: Ti-MOR, 0.100 g (B); H<sub>2</sub>O, 12 g; furfural, 20 mmol; NH<sub>3</sub> (25 wt. %), 30 mmol; temp., 323 K (A). H<sub>2</sub>O<sub>2</sub> was fed at a constant rate of 0.5 mmol min<sup>-1</sup>.



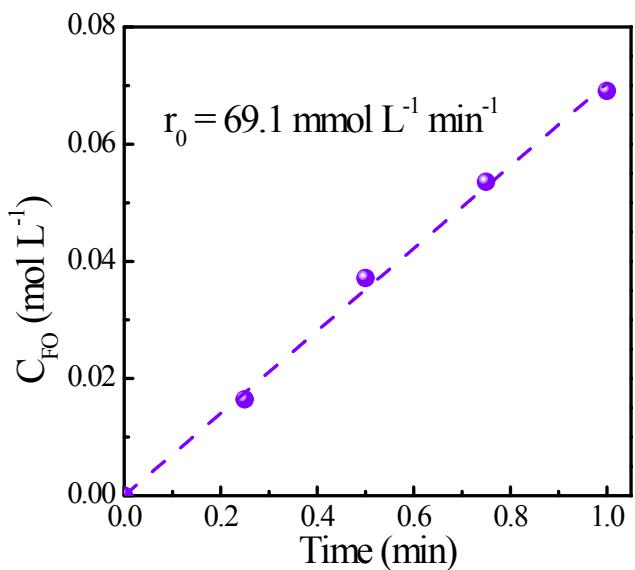
**Fig. S3.** XRD patterns (A), Nitrogen adsorption desorption isotherms at 77 K (B) and UV-vis spectra (C) of fresh Ti-MOR (a) and regenerated Ti-MOR (b) after five recycles of reaction-regeneration in the furfural ammoximation.



**Fig. S4.** Kinetic of hydroxylamine formation over Ti-MOR (a,  $\square$ ), TS-1 (b,  $\circ$ ) and Ti-MWW (c,  $\Delta$ ). Reaction conditions: cat., 0.1 g; NH<sub>3</sub>, 30 mmol; H<sub>2</sub>O<sub>2</sub>, 2.4 mmol; H<sub>2</sub>O, 12 g; temp., 323 K. H<sub>2</sub>O<sub>2</sub> was dropwise added at a constant rate within 6 min. The change of the amount of H<sub>2</sub>O<sub>2</sub> added with time is also shown.



**Fig. S5.** Kinetic of hydroxylamine decomposition over Ti-MOR (a), TS-1 (b) and Ti-MWW (c). Reaction conditions: cat., 0.1 g;  $\text{NH}_2\text{OH}\cdot\text{HCl}$ , 20 mmol;  $\text{H}_2\text{O}_2$ , 2.4 mmol;  $\text{H}_2\text{O}$ , 12 g; temp., 323 K.  $\text{H}_2\text{O}_2$  was dropwise added at a constant rate within 6 min.



**Fig. S6.** Kinetic of the furfural oximation with hydroxylamine. Reaction conditions:  
NH<sub>2</sub>OH·HCl, 20 mmol; H<sub>2</sub>O, 12 g; temp., 323 K.