

Black Hybride Iodobismuthate Containing Linear Anionic Chains

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Electronic supplementary information

Synthesis of 1,1'-(1,n-alkanediyl)bis(4-methylpyridinium) bromides. A slightly heated mixture containing 5 ml of α,ω -dibromoalkane (Aldrich 97%) and 20 ml of 4-picoline (Aldrich 98%), was stirred with a magnetic stirrer for 24 h. The obtained PiC_nBr_2 precipitate was washed with acetonitrile (Component-Reaktiv dry) and ethanol (Component-Reaktiv 96%) and then dried in air. In the case of 1,5-dibromopentane, the thus-prepared product was hygroscopic. Therefore, after being washed with dry acetonitrile, it was immediately dissolved in water or concentrated HI (Sigma-Aldrich, 57 wt % in H_2O stabilized 99.95%) to obtain a 2M solution.

Synthesis of 1,1'-(1,2-ethanediyl)bis(4-methylpyridinium) iodobismuthate. The 3 ml of 2 M PiC_2Br_2 solution in concentrated HI was added to 15 ml of 0.1 M BiI_3 (Lankhit ultra dry 99.998%) solution in 15 ml of concentrated HI. A crystalline substance precipitated from red solutions in 1 hour. The red solid **1** (1.427 g, 92.0%) was separated from the solution by decantation, washed three times with dry acetonitrile and ethanol, and then dried in air.

Synthesis of 1,1'-(1,5-pentanediy)bis(4-methylpyridinium) iodobismuthates. The 3 ml of 2 M PiC_5Br_2 solution in concentrated HI was added to 15 ml of 0.1 M BiI_3 solution in 15 ml of concentrated HI. A crystalline solid precipitated from red solutions in 1 hour. The black solid **2** (1.488 g, 90.2%) was separated from the solution by decantation, washed three times with dry acetonitrile and ethanol, and then dried in air.

The 3 ml of 2 M PiC_5Br_2 aqueous solution was added to 15 ml of 0.1 M BiI_3 solution in 15 ml of concentrated HI. A crystalline solid precipitated from red solutions in 1 hour. A mixture of black and red crystals was separated from the solution by decantation, washed three times with dry acetonitrile and ethanol, and then dried in air. Red crystals of **3** were mechanically separated from the black ones.

The 3 ml of 2 M PiC_5Br_2 aqueous solution was added to 15 ml of 0.1 M BiI_3 solution in 15 ml of saturated KI (Reakhim 99%) aqueous solution. A red non-crystalline solid precipitated from red solution in 1 hour. This red solid transformed to black one (**2a**) in several hours. Black

solid **2a** was separated from the solution by decantation, washed three times with water and ethanol, and then dried in air.

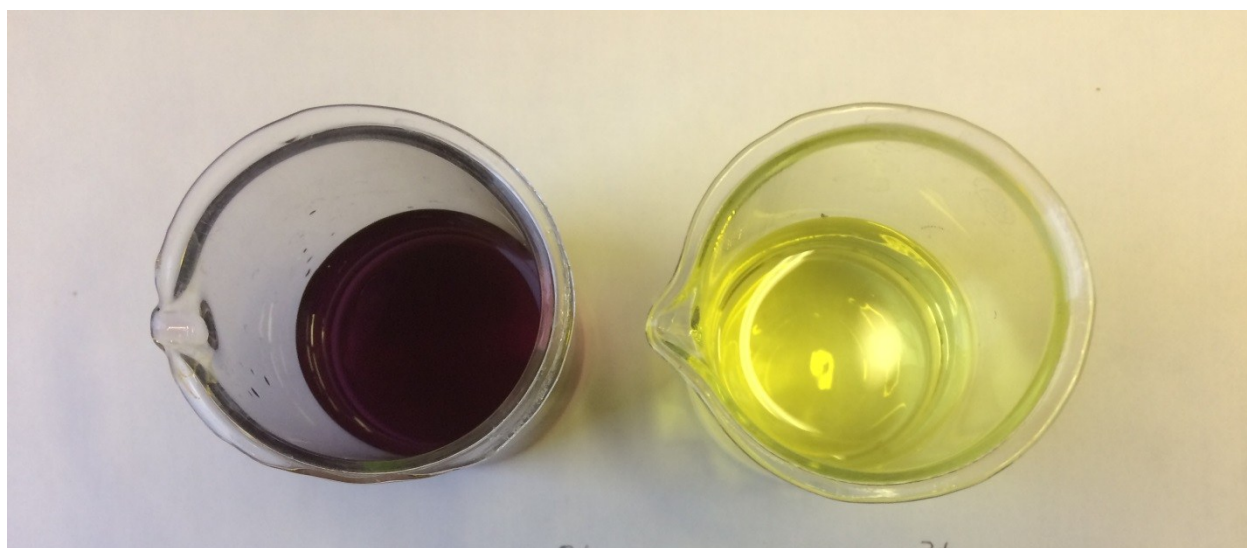
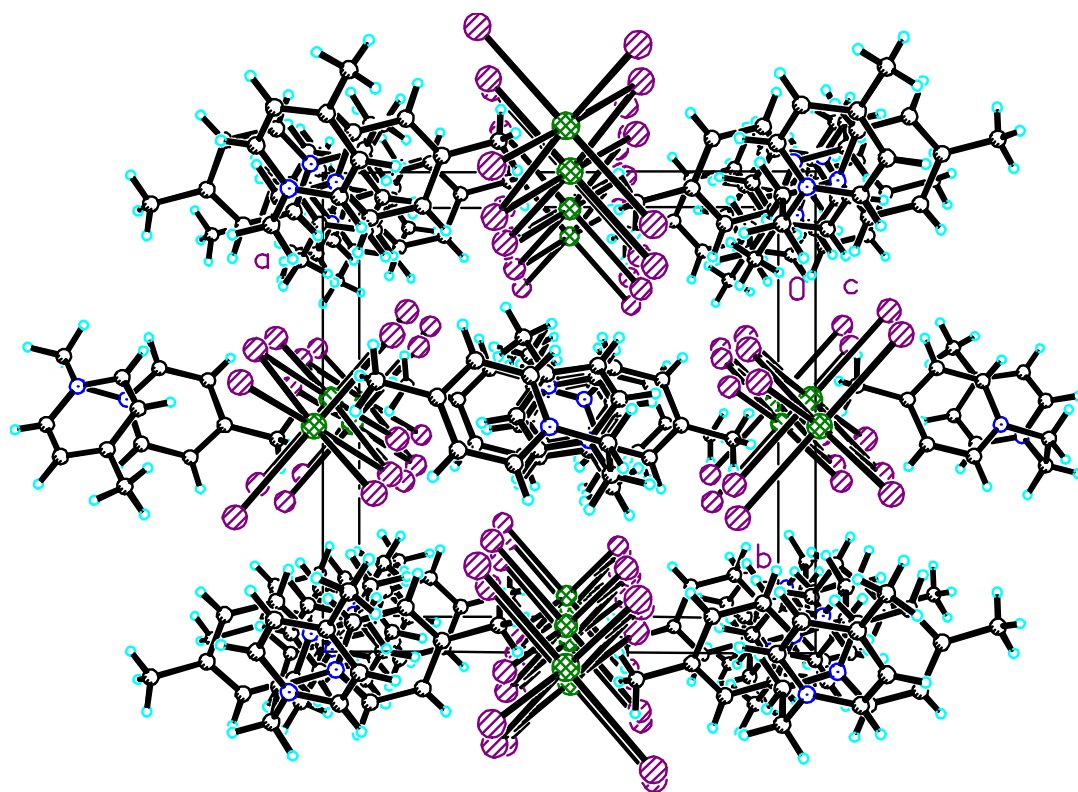
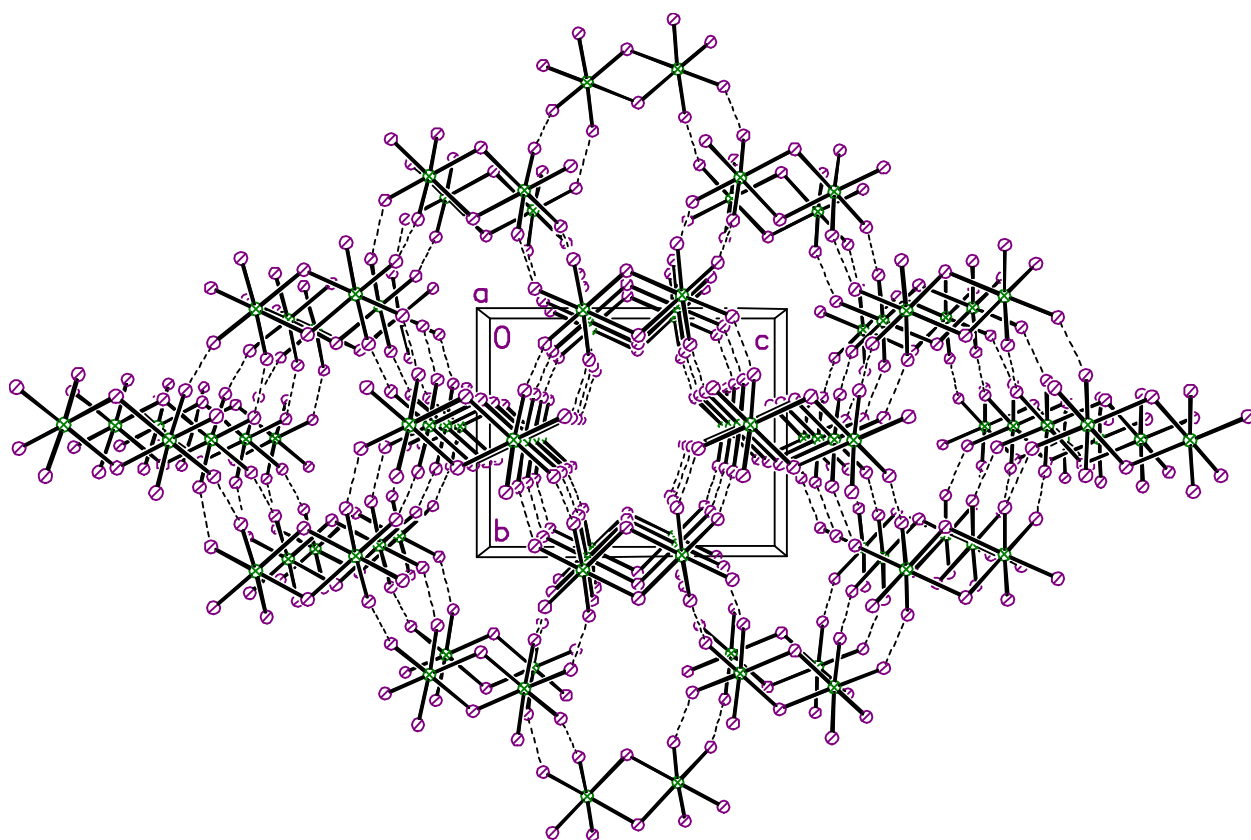


Figure S1. Solutions of 1,1'-(1,5-pentanediy)bis(4-cyanopyridinium) bromide (left) and 1,1'-(1,5-pentanediy)bis(4-methylpyridinium) bromide (right) after addition of potassium hexacyanoferrate solution. The difference in the color of solutions is due to the presence and absence of outer-sphere charge transfer bands in the visible region of spectra and indicates a greater electron affinity energy of the 1,1'-(1,5-pentanediy) bis (4-cyanopyridinium) cation compared to (1,5-pentanediy) bis (4-methylpyridinium) cation [1].

1. S.I. Gorelsky, V.Yu. Kotov, and A.B.P. Lever. *Inorg. Chem.*, 1988, **37**, 4548



a



b

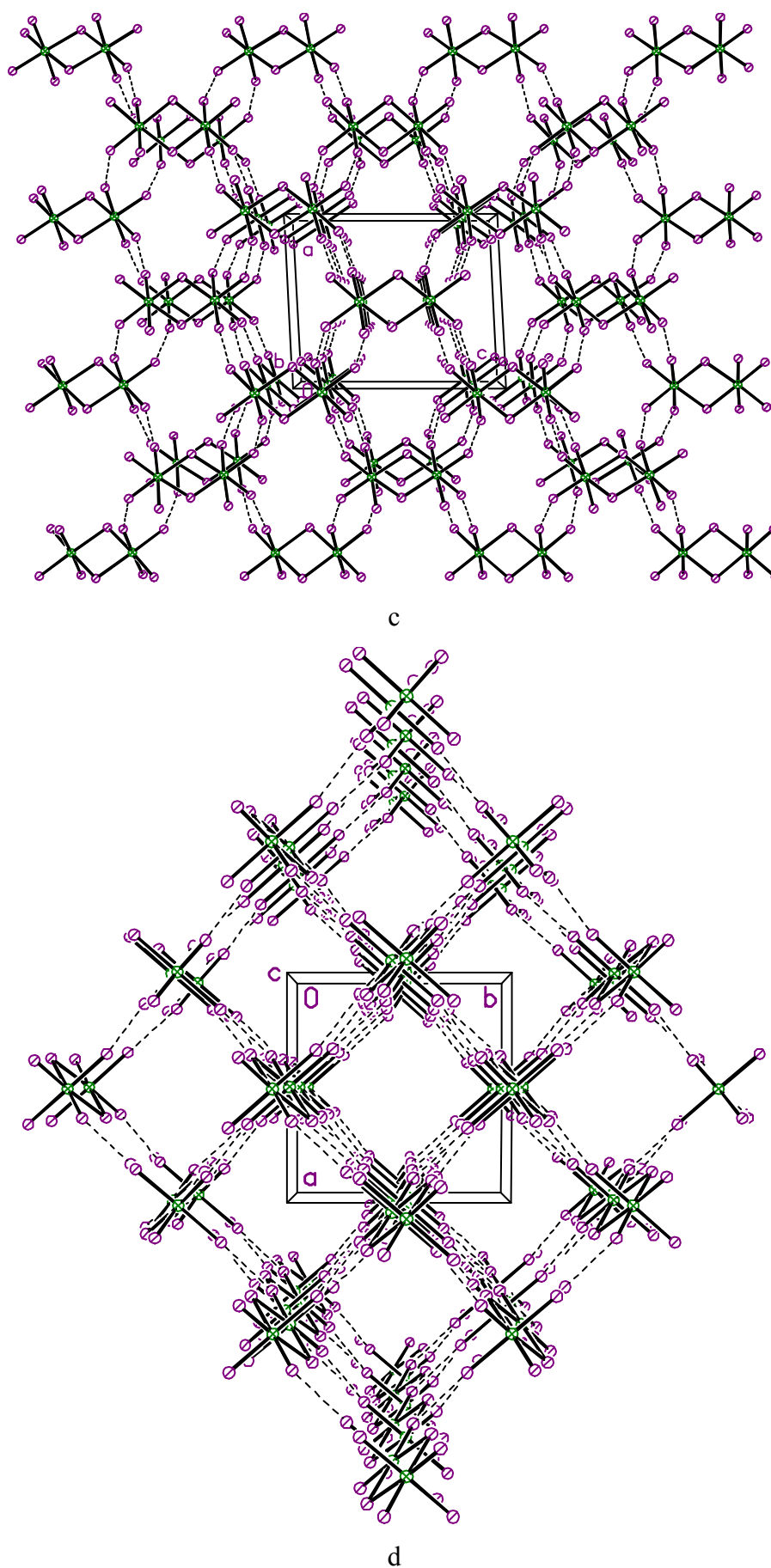


Figure S3. Projection of the structure of **1** along axis *c* (a). A schematic representation of short contacts I...I in the structure of **1** (b-d).

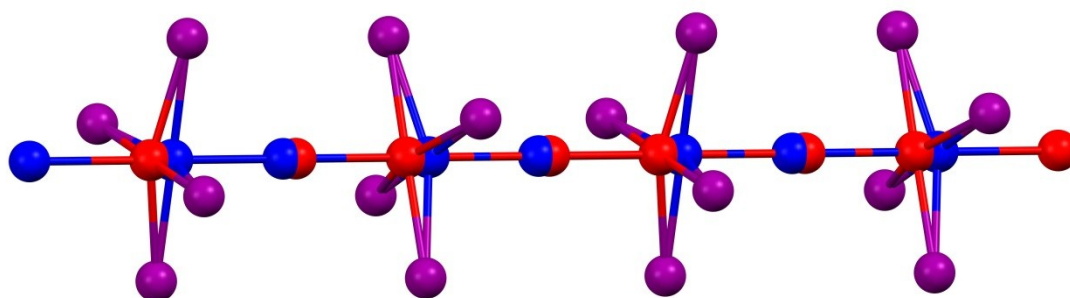
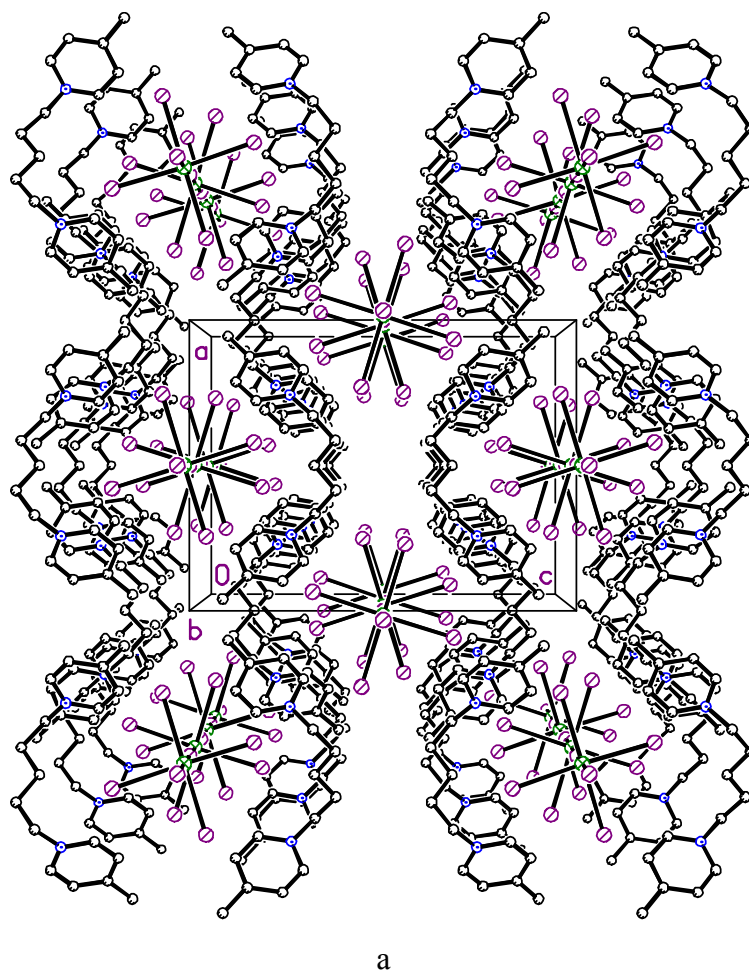


Figure S4. Projection of the structure of **2** along axis *b* (a). The structure of disordered 1D-chain $(\text{BiI}_5)_n^{2n-}$ in **2** (b). I(1,2) atoms identical for two orientations are violet, atoms of Bi and I different for two orientations are denoted by red and blue.

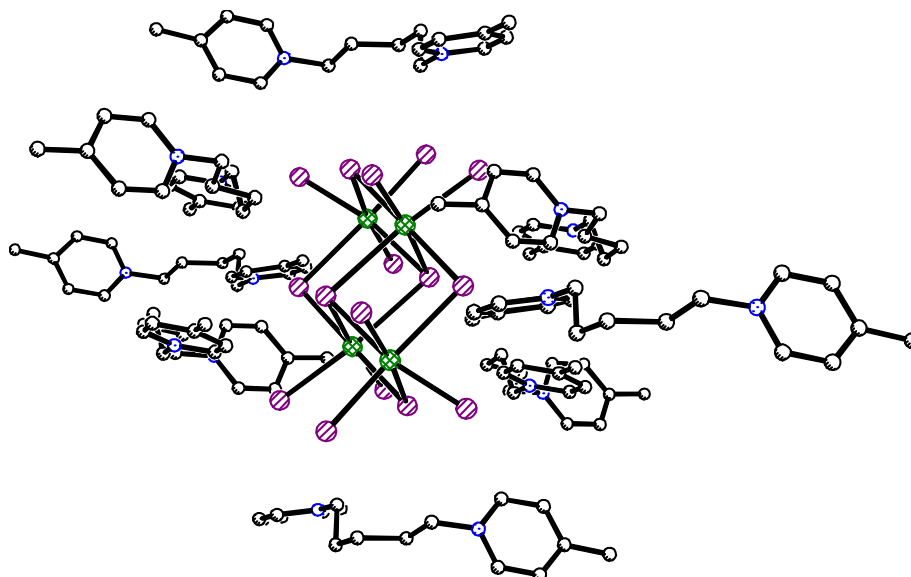


Figure S5. Immediate environment of the $\text{Bi}_4\text{I}_{16}^{4-}$ anion in **3**.

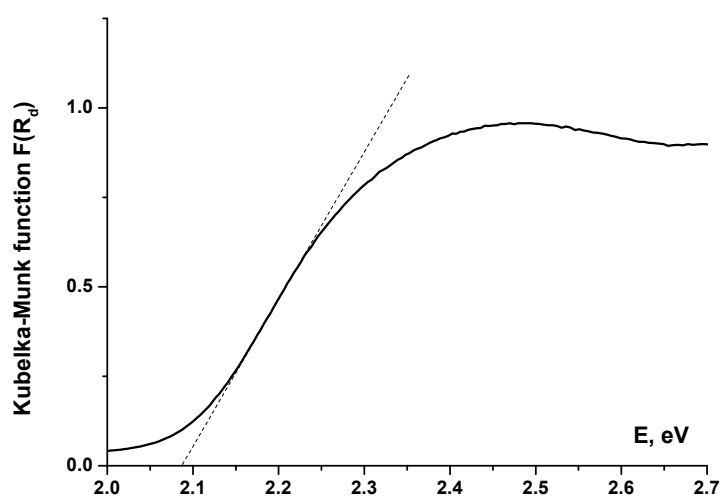


Figure S6. Reflectance spectra of **1** ($E_g = 2.08$ eV). The Kubelka-Munk function is defined as in

24.
$$F(R_d) = \frac{(1 - R_d)^2}{2R_d}$$
, where R_d is the absolute reflectance of the sample layer.

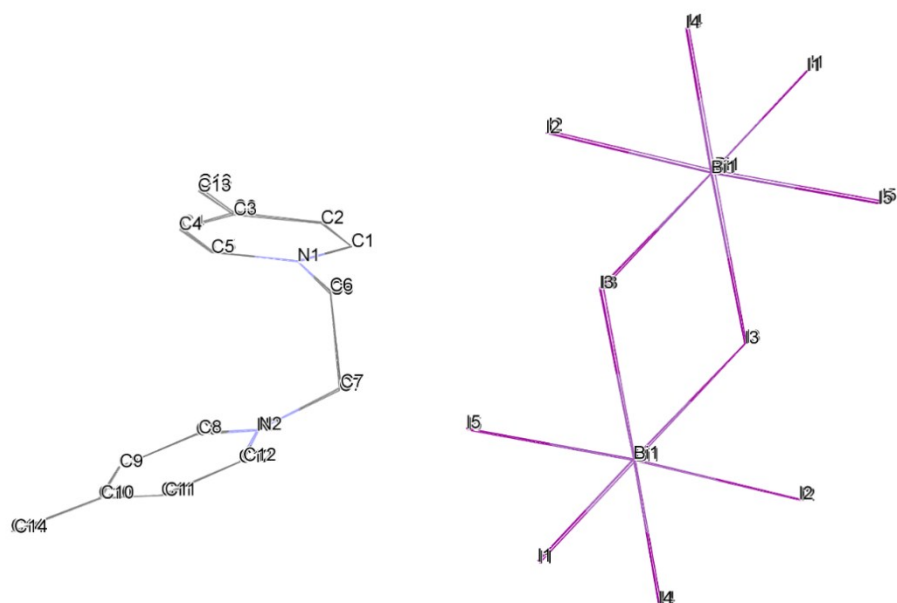
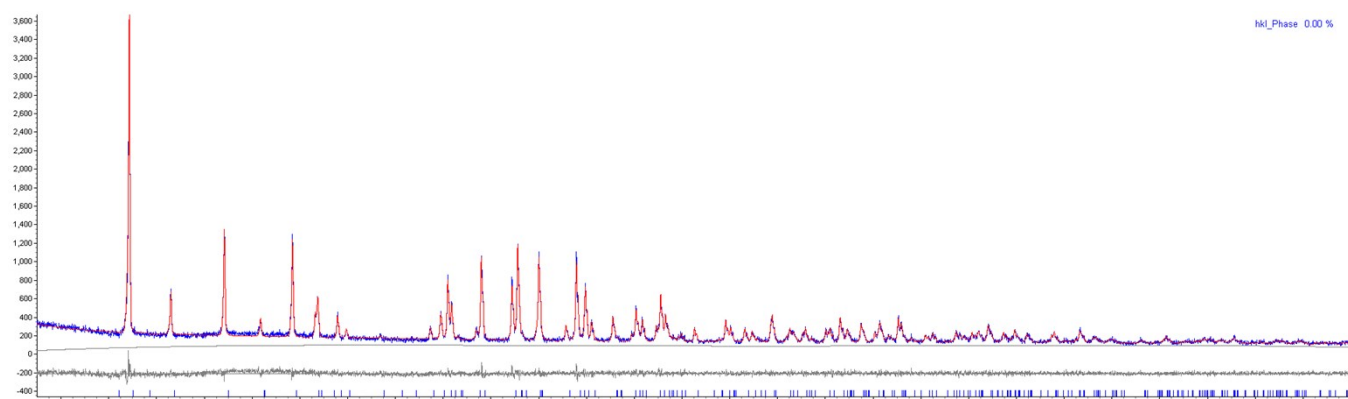
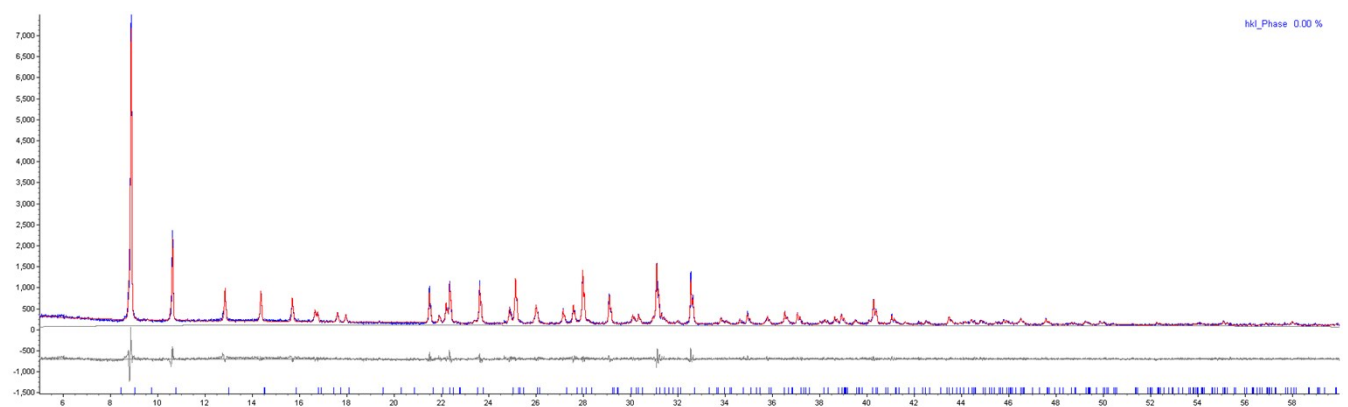


Figure S7. Superposition of structural fragments of **1**. The experiments were performed at 100, 150, and 298 K.



a



b

Figure S8. The results of refinements for **2** (a) and **2a**. (b)

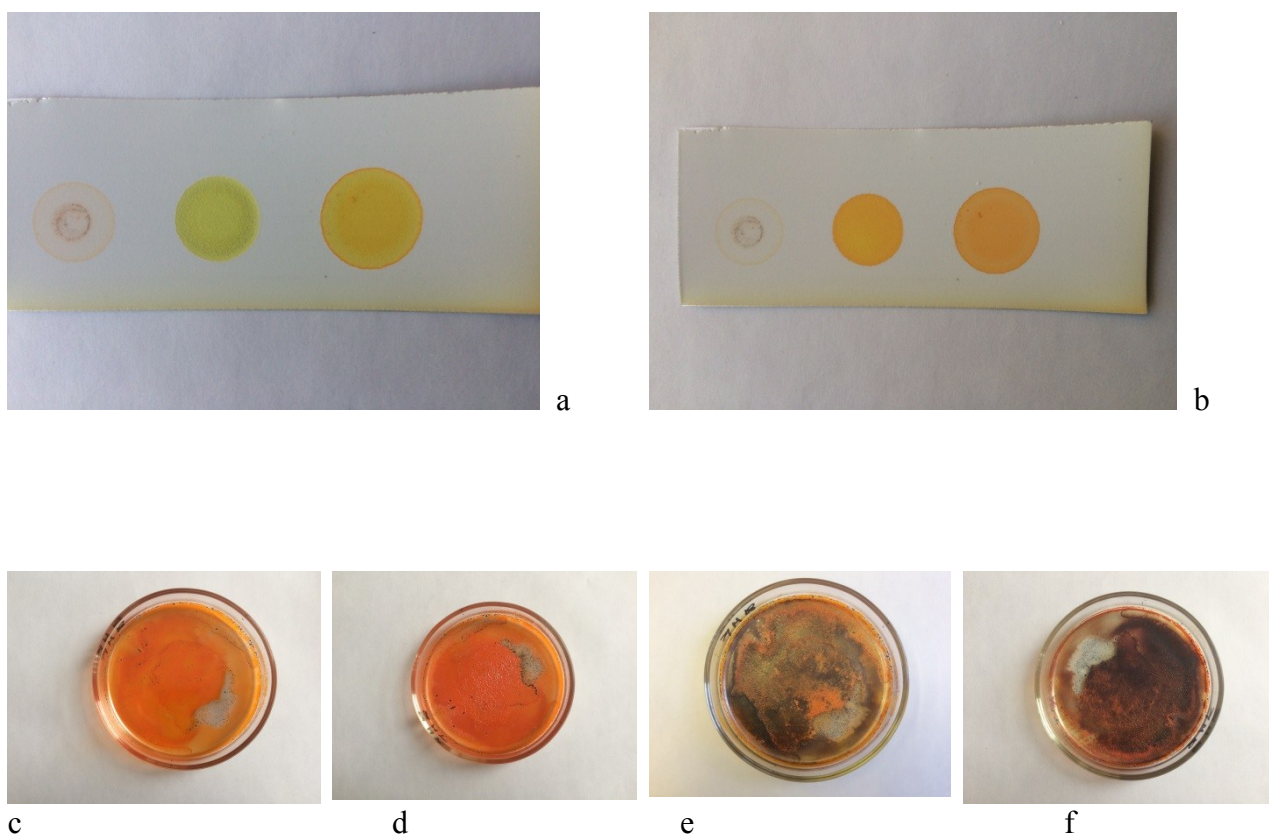


Figure S9. Drops of solutions of **2** in acetonitrile, dimethylformamide and dimethylsulfoxide on SiO_2 substrate before heating (a) and after heating to 160°C (b). The dried solution of **2** in dimethylformamide before (c) and after (d-f) treatment with water vapor during 1 minute. Time after treatment is 1 minute (d), one hour (e), and one day (f).

Table S1. Geometric parameters [Å and deg.] of **1-3**.**1**

	100K	150K	298K
Bi(1)-I(4)	2.9141(2)	2.9140(3)	2.9176(4)
Bi(1)-I(1)	2.9460(2)	2.9467(3)	2.9516(5)
Bi(1)-I(5)	3.0344(2)	3.0342(3)	3.0371(4)
Bi(1)-I(2)	3.0543(2)	3.0553(3)	3.0645(4)
Bi(1)-I(3)#1	3.3071(2)	3.3088(3)	3.3252(5)
Bi(1)-I(3)	3.3134(2)	3.3144(3)	3.3241(4)
	100K	150K	298K
I(4)-Bi(1)-I(1)	92.169(7)	92.348(8)	93.001(15)
I(4)-Bi(1)-I(5)	87.400(7)	87.448(8)	87.311(14)
I(1)-Bi(1)-I(5)	87.866(7)	87.796(8)	87.772(14)
I(4)-Bi(1)-I(2)	90.914(7)	90.972(8)	91.337(14)
I(1)-Bi(1)-I(2)	89.711(7)	89.808(8)	89.855(13)
I(5)-Bi(1)-I(2)	176.993(7)	177.073(9)	177.204(12)
I(4)-Bi(1)-I(3)#1	88.324(7)	88.424(8)	88.297(14)
I(1)-Bi(1)-I(3)#1	178.562(6)	178.481(8)	178.311(11)
I(5)-Bi(1)-I(3)#1	93.507(6)	93.547(8)	93.371(13)
I(2)-Bi(1)-I(3)#1	88.930(6)	88.871(7)	89.034(12)
I(4)-Bi(1)-I(3)	176.657(6)	176.525(8)	175.673(12)
I(1)-Bi(1)-I(3)	89.660(7)	89.817(8)	90.318(14)
I(5)-Bi(1)-I(3)	89.872(6)	89.921(8)	90.046(13)
I(2)-Bi(1)-I(3)	91.895(7)	91.753(8)	91.448(13)
I(3)#1-Bi(1)-I(3)	89.914(6)	89.475(7)	88.438(13)
Bi(1)#1-I(3)-Bi(1)	90.087(6)	90.524(7)	91.563(12)
	100K	150K	298K
C(5)-N(1)-C(1)-C(2)	-0.2(4)	-0.4(5)	-0.8(7)
C(6)-N(1)-C(1)-C(2)	176.5(3)	175.8(3)	175.7(5)
N(1)-C(1)-C(2)-C(3)	-0.3(4)	-0.4(6)	-0.2(8)
C(1)-C(2)-C(3)-C(4)	0.8(4)	1.2(6)	0.8(8)
C(1)-C(2)-C(3)-C(13)	-179.0(3)	-179.8(4)	-179.2(5)

C(2)-C(3)-C(4)-C(5)	-0.7(4)	-1.3(6)	-0.5(8)
C(13)-C(3)-C(4)-C(5)	179.1(3)	179.7(4)	179.5(6)
C(1)-N(1)-C(5)-C(4)	0.3(4)	0.4(5)	1.1(8)
C(6)-N(1)-C(5)-C(4)	-176.5(3)	-175.9(3)	-175.4(5)
C(3)-C(4)-C(5)-N(1)	0.2(5)	0.5(6)	-0.4(9)
C(1)-N(1)-C(6)-C(7)	-84.8(3)	-84.6(4)	-85.0(6)
C(5)-N(1)-C(6)-C(7)	91.9(3)	91.6(4)	91.4(6)
C(8)-N(2)-C(7)-C(6)	-75.7(3)	-76.6(4)	-76.8(6)
C(12)-N(2)-C(7)-C(6)	100.5(3)	100.6(4)	101.0(6)
N(1)-C(6)-C(7)-N(2)	-58.5(3)	-58.3(4)	-59.0(6)
C(12)-N(2)-C(8)-C(9)	-0.9(5)	-2.1(6)	-1.5(9)
C(7)-N(2)-C(8)-C(9)	175.2(3)	175.2(4)	176.3(6)
N(2)-C(8)-C(9)-C(10)	0.3(5)	1.0(6)	0.6(10)
C(8)-C(9)-C(10)-C(11)	0.5(5)	0.7(6)	0.1(10)
C(8)-C(9)-C(10)-C(14)	-177.4(3)	-178.2(4)	-177.6(6)
C(9)-C(10)-C(11)-C(12)	-0.9(5)	-1.4(6)	0.1(9)
C(14)-C(10)-C(11)-C(12)	177.0(3)	177.5(4)	177.8(6)
C(8)-N(2)-C(12)-C(11)	0.5(4)	1.4(6)	1.7(8)
C(7)-N(2)-C(12)-C(11)	-175.6(3)	-175.9(4)	-176.1(5)
C(10)-C(11)-C(12)-N(2)	0.4(5)	0.4(6)	-1.0(8)

Symmetry transformations used to generate equivalent atoms:

#1 -x,-y+1,-z

2

Bi(1)-I(3)#1	2.905(16)
Bi(1)-I(2)	3.0390(8)
Bi(1)-I(1)	3.0796(9)
Bi(1)-I(3)	3.188(16)
Bi(1)-I(4)	3.511(10)
Bi(1)-I(4)#3	3.845(11)
Bi(1B)-I(4)	2.929(11)
Bi(1B)-I(1)	3.0265(8)
Bi(1B)-I(2)	3.0921(9)
Bi(1B)-I(4)#3	3.263(10)

Bi(1B)-I(3)#1	3.487(16)
Bi(1B)-I(3)	3.771(16)
Bi(1)...Bi(1B)	0.5824(18)
I(3)...I(3)#1	0.28(3)
I(4)...I(4)#3	0.33(2)
C(5)-N(1)-C(1)-C(2)	-1(2)
C(6)-N(1)-C(1)-C(2)	-179.5(16)
N(1)-C(1)-C(2)-C(3)	2(3)
C(1)-C(2)-C(3)-C(4)	-2(3)
C(1)-C(2)-C(3)-C(9)	179.8(16)
C(2)-C(3)-C(4)-C(5)	1(3)
C(9)-C(3)-C(4)-C(5)	179.6(16)
C(1)-N(1)-C(5)-C(4)	0(2)
C(6)-N(1)-C(5)-C(4)	178.9(16)
C(3)-C(4)-C(5)-N(1)	-1(3)
C(5)-N(1)-C(6)-C(7)	107.9(16)
C(1)-N(1)-C(6)-C(7)	-73.7(17)
N(1)-C(6)-C(7)-C(8)	-60.9(16)
C(6)-C(7)-C(8)-C(7)#3	-176.9(15)

Symmetry transformations used to generate equivalent atoms:

#1 -x,-y+1,z #2 -x,y,-z+1 #3 -x,-y,z

3

Bi(1)-I(7)	2.8834(5)
Bi(1)-I(6)	2.8939(5)
Bi(1)-I(4)	3.0379(5)
Bi(1)-I(5)	3.1294(5)
Bi(1)-I(3)#1	3.3318(5)
Bi(1)-I(3)	3.3508(5)
Bi(2)-I(8)	2.9155(5)
Bi(2)-I(10)	2.9200(5)
Bi(2)-I(9)	2.9264(5)
Bi(2)-I(5)#1	3.2654(5)

Bi(2)-I(4)	3.2723(5)
Bi(2)-I(3)	3.3618(5)
C(5)-N(1)-C(1)-C(2)	-0.9(10)
C(6)-N(1)-C(1)-C(2)	-174.7(7)
N(1)-C(1)-C(2)-C(3)	1.1(11)
C(1)-C(2)-C(3)-C(4)	-1.0(10)
C(1)-C(2)-C(3)-C(16)	177.8(7)
C(2)-C(3)-C(4)-C(5)	0.9(10)
C(16)-C(3)-C(4)-C(5)	-177.9(6)
C(1)-N(1)-C(5)-C(4)	0.9(10)
C(6)-N(1)-C(5)-C(4)	174.6(6)
C(3)-C(4)-C(5)-N(1)	-0.9(10)
C(1)-N(1)-C(6)-C(7)	88.3(7)
C(5)-N(1)-C(6)-C(7)	-85.4(7)
N(1)-C(6)-C(7)-C(8)	-173.7(6)
C(6)-C(7)-C(8)-C(9)	-167.5(6)
C(7)-C(8)-C(9)-C(10)	77.4(7)
C(11)-N(2)-C(10)-C(9)	-69.4(8)
C(15)-N(2)-C(10)-C(9)	111.9(8)
C(8)-C(9)-C(10)-N(2)	-178.8(6)
C(15)-N(2)-C(11)-C(12)	-1.0(11)
C(10)-N(2)-C(11)-C(12)	-179.6(7)
N(2)-C(11)-C(12)-C(13)	2.5(12)
C(11)-C(12)-C(13)-C(14)	-2.1(11)
C(11)-C(12)-C(13)-C(17)	177.0(7)
C(12)-C(13)-C(14)-C(15)	0.3(11)
C(17)-C(13)-C(14)-C(15)	-178.8(7)
C(11)-N(2)-C(15)-C(14)	-0.9(11)
C(10)-N(2)-C(15)-C(14)	177.7(7)
C(13)-C(14)-C(15)-N(2)	1.2(12)

Symmetry transformations used to generate equivalent atoms:

#1 -x,-y+1,-z

Table S2. Optimized atomic coordinates of structure **2** (copy of VASP CONTCAR file)

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Energy = -1108.38709784 eV

Table S3. Optimized atomic coordinates of structure **3** (copy of VASP CONTCAR file)

kotov28 crystal

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16.0487159999999989 0.0000000000000000 -4.9967750000000004

0.0000000000000000 12.0002999999999993 0.0000000000000000

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Bi I N C H

8 32 8 68 96

Direct

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