

SUPPLEMENTARY INFORMATION

Chlorogenic acids and the acyl-quinic acids: discovery, biosynthesis, bioavailability and bioactivity

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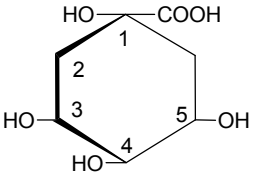
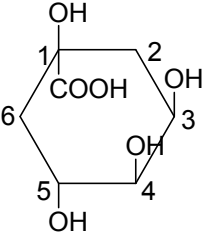
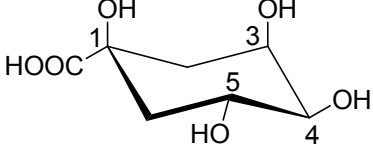
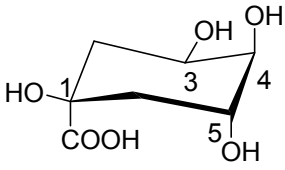
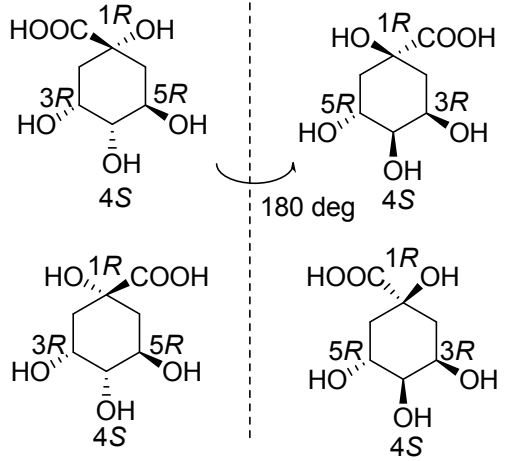
Supplementary Table 1. Trivial names associated with acyl-quinic acids and the associated systematic name using IUPAC numbering.^{a1}

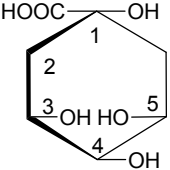
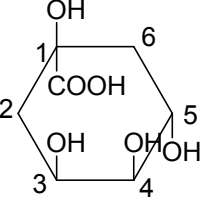
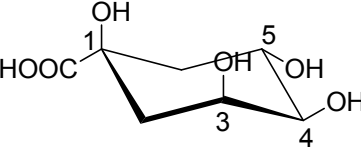
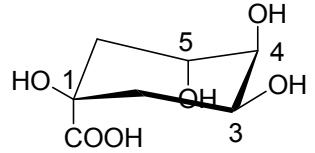
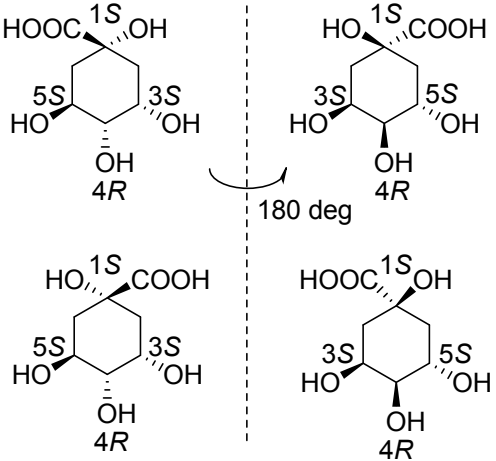
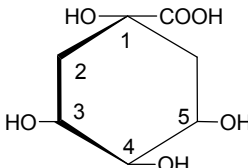
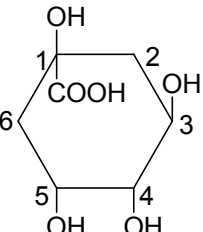
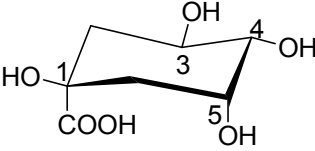
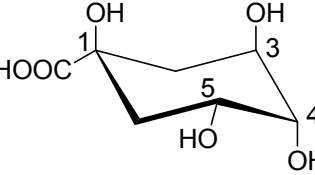
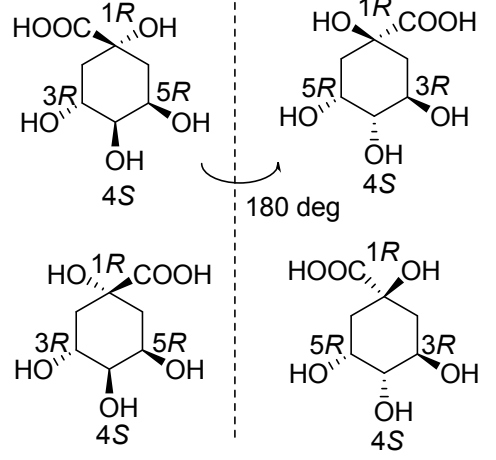
Trivial Name	Origin	Current Interpretation with IUPAC numbering	Notes
Band 510	Sondheimer (1958) ²	4- <i>O</i> -Caffeoylquinic acid	
Burkinabins	Ouattara <i>et al.</i> (2004) ³	Divanilloylquinic acids	
Castusic acid	Kirmizibekmev and Demir (2016) ⁴	4- <i>p</i> -Hydroxybenzoyl-5- <i>O</i> -caffeoylquinic acid	
Chlorogen acid/chlorogenic acid	Payen (1846) ^{5,6}	5- <i>O</i> -Caffeoylquinic acid	The original isolate subsequently described as 3-caffeoylquinic acid and now known as 5-caffeoylquinic acid IUPAC
Chlorogenic acids			Used in the plural to denote the extended family of structurally related compounds
Cryptochlorogenic acid	Unknown	4- <i>O</i> -Caffeoylquinic acid	
Cynarin(e)	Panizzi <i>et al.</i> (1954) ^{7,8}	1,3- <i>O</i> -Dicafeoylquinic acid	Rapidly formed from 1,5- <i>I</i> -dicafeoylquinic acid in aqueous media
Dactylifric acid	Maier <i>et al.</i> (1964) ⁹	5- <i>O</i> -Caffeoylshikimic acid	
Dattelic acid	Wada <i>et al.</i> (1988) ¹⁰	5- <i>O</i> -Caffeoylshikimic acid	
Hauschild's substance	Hauschild (1935) ¹¹	3-Caffeoylquinic-1,5- γ -lactone	Rapidly formed from 1,5-dicafeoylquinic acid in aqueous media
Irbic acid	Antognoni <i>et al.</i> (2011) ¹²	3,5- <i>O</i> -Dicafeoyl-4- <i>O</i> -malonoyl-quinic acid	This name was applied first to a compound isolated from cultured cells of <i>Centella asiatica</i> but subsequently found in the whole plant. ^{13,14} This compound had been known from at least 2007. ¹⁵
Isochlorogenic acid	Barnes <i>et al.</i> (1950) ¹⁶	A mixture of at least three dicafeoylquinic acids	Originally described as 5- <i>O</i> -caffeoylquinic acid using non-IUPAC numbering, contaminated with the associated lactone. Prefix 'iso' used to indicate 'isomer'. The two research groups used the descriptors a, b and c, or A, B and C, differently and it is thus very difficult to tell which letter applies to which regio-isomer in later usage. As originally published A = c = 3,4- <i>O</i> -diCQA, and C = b = 3,5- <i>O</i> -diCQA IUPAC. Logically B = a = 4,5- <i>O</i> -diCQA IUPAC but there remains an unexplained difference in specific rotation of the two isolates. ¹⁷
Isochlorogenic acid a, b and c, or A, B and C	Scarpati and Guiso (1964) ¹⁷ Corse <i>et al.</i> (1965) ¹⁸	Coffee bean dicafeoylquinic acids	

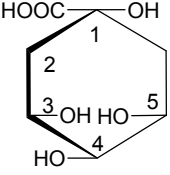
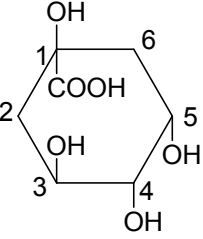
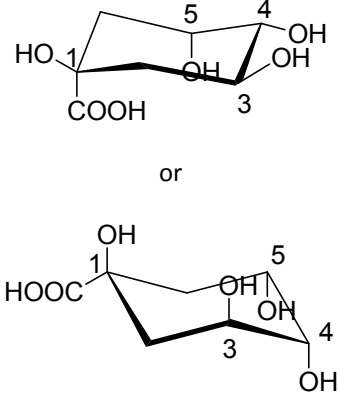
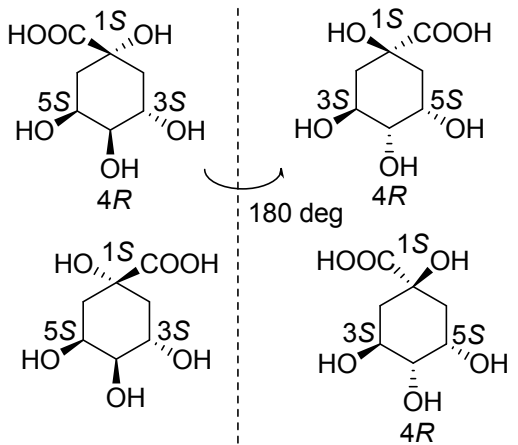
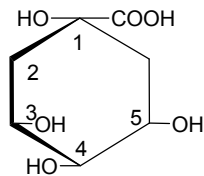
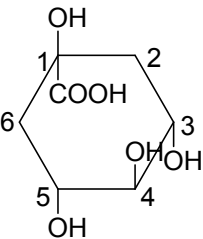
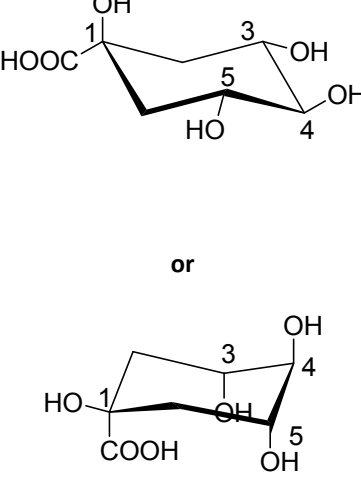
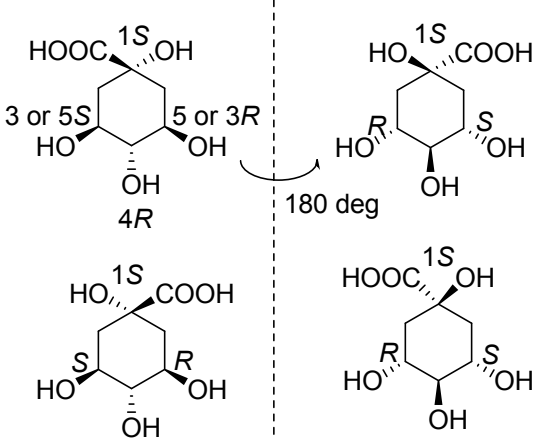
Macranthoin F and G	Chen <i>et al.</i> (1994) ¹⁹	Methyl-4,5- <i>O</i> -dicaffeoylquinic acid and methyl 3,5-dicaffeoylquinic acid, respectively	Note that the term 'macranthoin' refers to constituents of <i>Lonicera macranthoides</i> regardless of whether or not they are chlorogenic acids.
Mumeic acid	Nakamura <i>et al.</i> (2013) ²⁰	4-Benzoyl-5- <i>O</i> -caffeoylquinic acid	
<i>n</i> -Chlorogenic acid	Maier and Grimsehl (1982) ²¹	5- <i>O</i> -Caffeoylquinic acid	Prefix ' <i>n</i> ' used to distinguish 5- <i>O</i> -caffeoylquinic acid from total chlorogenic acids
Neochlorogenic acid	Corse (1953) ²²	3- <i>O</i> -Caffeoylquinic acid	Neochlorogenic acid
Origanine A-C	Liu <i>et al.</i> (2012) ²³	Derivatives of 1,3,4,5- and 1,3,5,6-tetra-carboxy-shikimic acid	Reported in <i>Origanum vulgare</i> L. Biosynthetic origin unknown and possibly different from (-)-quinic and (-)-shikimic acids.
Pistafolins	Hou <i>et al.</i> (2000) ²⁴	Galloylquinic acid depsides	Term applied to some galloylquinic acids of <i>Pistacia lentiscus</i>
Podospermic acid	Zidorn <i>et al.</i> (2005) ²⁵	1,3,5- <i>O</i> -tri-dihydrocaffeoylquinic acids	Some close relatives, such as di-dihydrocaffeoyl-feruloylquinic acids, may also be included in the trivial name. ²⁶
Pseudochlorogenic acid	Uritani and Miyano (1955) ²⁷	1- <i>O</i> -Caffeoylquinic acid	Probably the original isolate was a poorly defined mixture of caffeoylquinic and dicaffeoylquinic acids
Salicornate	Kim <i>et al.</i> (2011) ²⁸	Methyl-4- <i>O</i> -caffeoyl-3-dihydrocaffeoyl-quinic acid	
Theogallin	Roberts (1958) ²⁹	5-Galloylquinic acid	
Tuntungmadic acid	Chung <i>et al.</i> (2005) ³⁰	4- <i>O</i> -Dihydrocaffeoyl-5- <i>O</i> -caffeoylquinic acid	
Viarum acids	Wu <i>et al.</i> (2012) ³¹	5- <i>O</i> -Caffeoyl-[4-(1 β -[6-(5- <i>O</i> -caffeoyl)quinic acid]gluco-pyranosyl)]quinic acid and 3- <i>O</i> -malonyl-5- <i>O</i> -caffeoyl-[4-(1 β -[6-(5- <i>O</i> -caffeoyl)quinic acid]gluco-pyranosyl)]quinic acid	

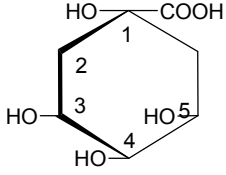
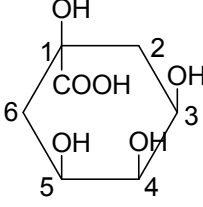
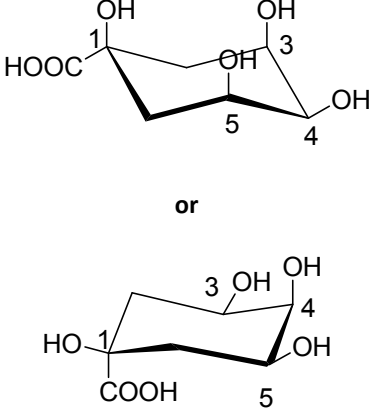
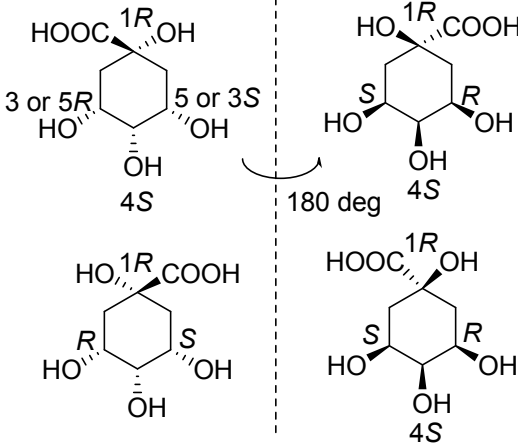
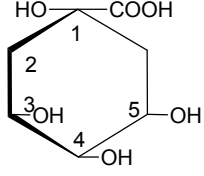
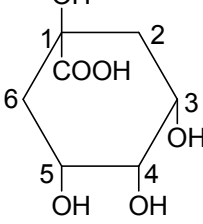
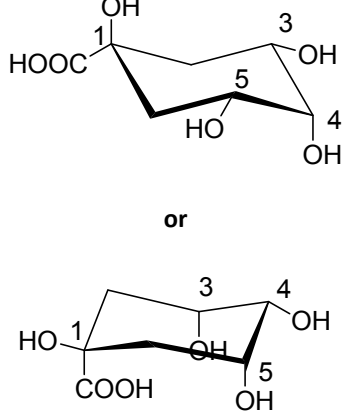
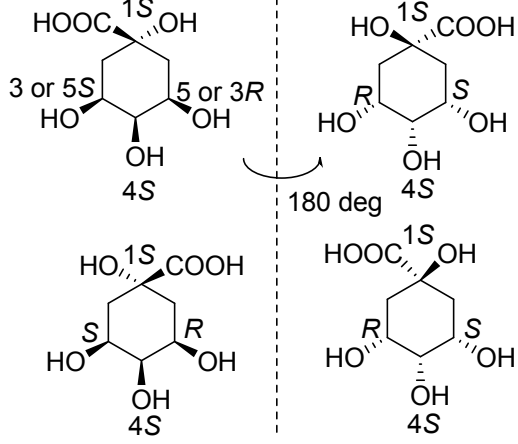
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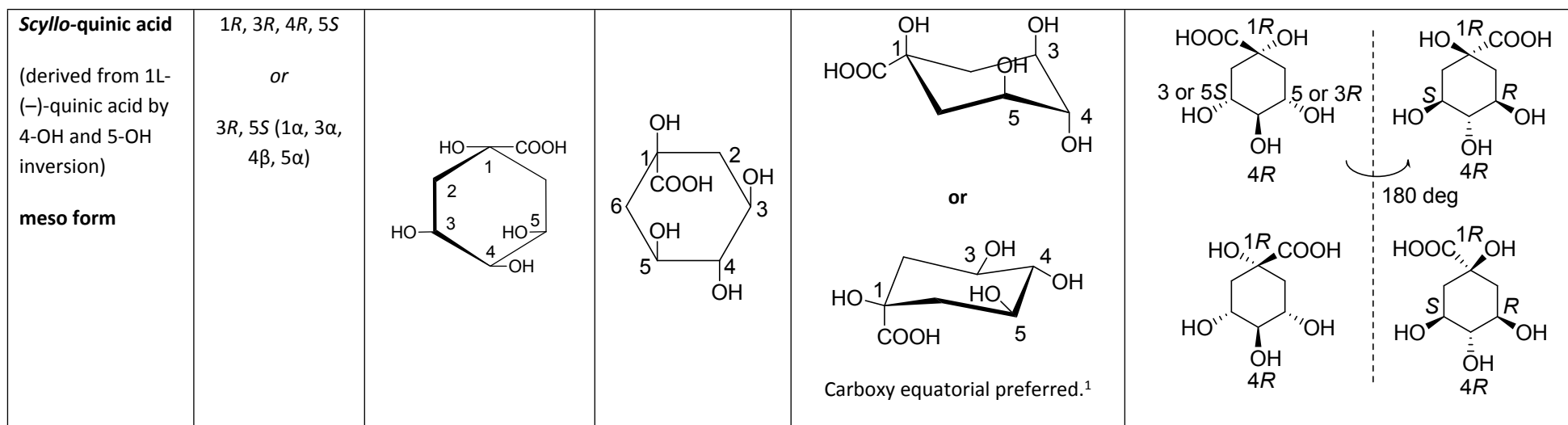
Supplementary Table S2 An unambiguous nomenclature for the acyl-quinic acids commonly known as chlorogenic acids.^{a1}

Isomer	CIP ^b description and recommended description ¹	Fischer–Tollens structure	2D structure using the recommended convention	Conformers	2D structures for individual isomers drawn from four different perspectives
1L-(–)-quinic acid	<p>1<i>R</i>, 3<i>R</i>, 4<i>S</i>, 5<i>R</i></p> <p>or</p> <p>3<i>R</i>, 5<i>R</i>-(1<i>α</i>, 3<i>α</i>, 4<i>α</i>, 5<i>β</i>)</p>			 <p>Upper structure with equatorial COOH and axial 4H as shown in Corse and Lundin.³³</p> <p>The lower carboxy equatorial structure is preferred.¹</p> 	

<p>1D-(+)-quinic acid</p>	<p>1<i>S</i>, 3<i>S</i>, 4<i>R</i>, 5<i>S</i></p> <p>or</p> <p>3<i>S</i>, 5<i>S</i>-(1<i>α</i>, 3<i>α</i>, 4<i>α</i>, 5<i>β</i>)</p>			 <p>Upper structure with equatorial COOH and axial 4H.</p> <p>The lower carboxy equatorial structure is preferred.¹</p> 	
<p>1L-(-)-<i>epi</i>-quinic acid (derived from 1L-(-)-quinic acid by 4-OH inversion)</p>	<p>1<i>R</i>, 3<i>R</i>, 4<i>S</i>, 5<i>R</i></p> <p>or</p> <p>3<i>R</i>, 5<i>R</i>-(1<i>α</i>, 3<i>α</i>, 4<i>β</i>, 5<i>β</i>)</p>			 <p>Upper structure with equatorial COOH and axial 4H.</p> <p>The lower carboxy equatorial structure is preferred.¹</p> 	

<p>1D-(+)-<i>epi</i>-quinic acid (derived from 1D-(+)-quinic acid by 4-OH inversion)</p>	<p>1<i>S</i>, 3<i>S</i>, 4<i>R</i>, 5<i>S</i></p> <p>or</p> <p>3<i>S</i>, 5<i>S</i>-(1α, 3α, 4β, 5β)</p>			 <p>or</p> <p>Carboxy axial preferred.¹</p>	 <p>180 deg</p>
<p>Muco-quinic acid (derived from 1L-(-)-quinic acid by 3-OH inversion) <i>meso</i> form</p>	<p>1<i>S</i>, 3<i>S</i>, 4<i>R</i>, 5<i>R</i></p> <p>or</p> <p>3<i>S</i>, 5<i>R</i> (1α, 3β, 4α, 5β)</p>			 <p>or</p> <p>Carboxy equatorial preferred.¹</p>	 <p>180 deg</p>

<p>Cis-quinic acid</p> <p>(derived from 1L-(-)-quinic acid by 5-OH inversion) meso form</p>	<p>1R, 3R, 4S, 5S</p> <p>or</p> <p>3R, 5S (1α, 3α, 4α, 5α)</p>			 <p>Carboxy axial preferred.¹</p>	 <p>Carboxy axial preferred.¹</p>
<p>Neo-quinic acid</p> <p>(derived from 1L-(-)-quinic acid by 3-OH and 4-OH inversion) meso form</p>	<p>1S, 3S, 4S, 5R</p> <p>or</p> <p>3S, 5R (1α, 3β, 4β, 5β)</p>			 <p>Carboxy axial preferred.¹</p>	 <p>Carboxy axial preferred.¹</p>



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^bCIP - Cahn-Ingold-Prelog³⁴

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