

## ***Electronic Supplementary Information (ESI)***

### **Quantum Tunneling Injection of Hot Electrons in Au/TiO<sub>2</sub> Plasmonic Photocatalysts**

**Yasuhiro Shiraishi,<sup>\*ab</sup> Naoki Yasumoto,<sup>a</sup> Jun Imai,<sup>a</sup> Hirokatsu Sakamoto,<sup>a</sup> Shunsuke Tanaka,<sup>c</sup> Satoshi Ichikawa,<sup>d</sup> Bunsho Ohtani,<sup>e</sup> and Takayuki Hirai<sup>a</sup>**

<sup>a</sup> Research Center for Solar Energy Chemistry, and Division of Chemical Engineering, Graduate School of Engineering Science, Osaka University, Toyonaka 560-8531, Japan

<sup>b</sup> Precursory Research for Embryonic Science and Technology (PRESTO), Japan Science and Technology Agency (JST), Saitama 332-0012, Japan

<sup>c</sup> Department of Chemical, Energy and Environmental Engineering, Kansai University, Suita 564-8680, Japan

<sup>d</sup> Institute for NanoScience Design, Osaka University, Toyonaka 560-8531, Japan

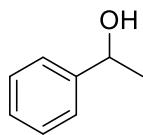
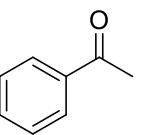
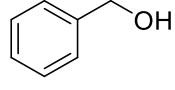
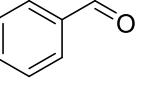
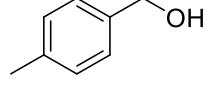
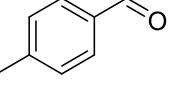
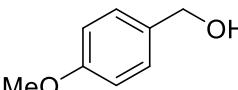
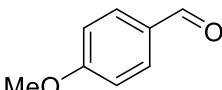
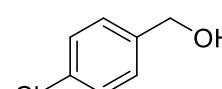
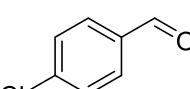
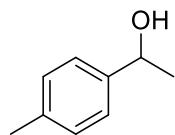
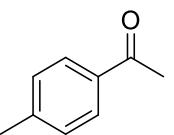
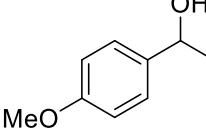
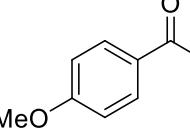
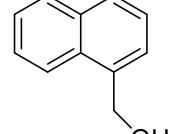
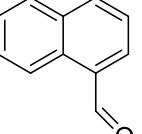
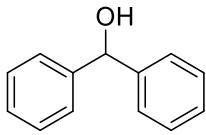
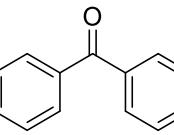
<sup>e</sup> Institute for Catalysis, Hokkaido University, Sapporo 001-0021, Japan

E-mail: shiraish@cheng.es.osaka-u.ac.jp

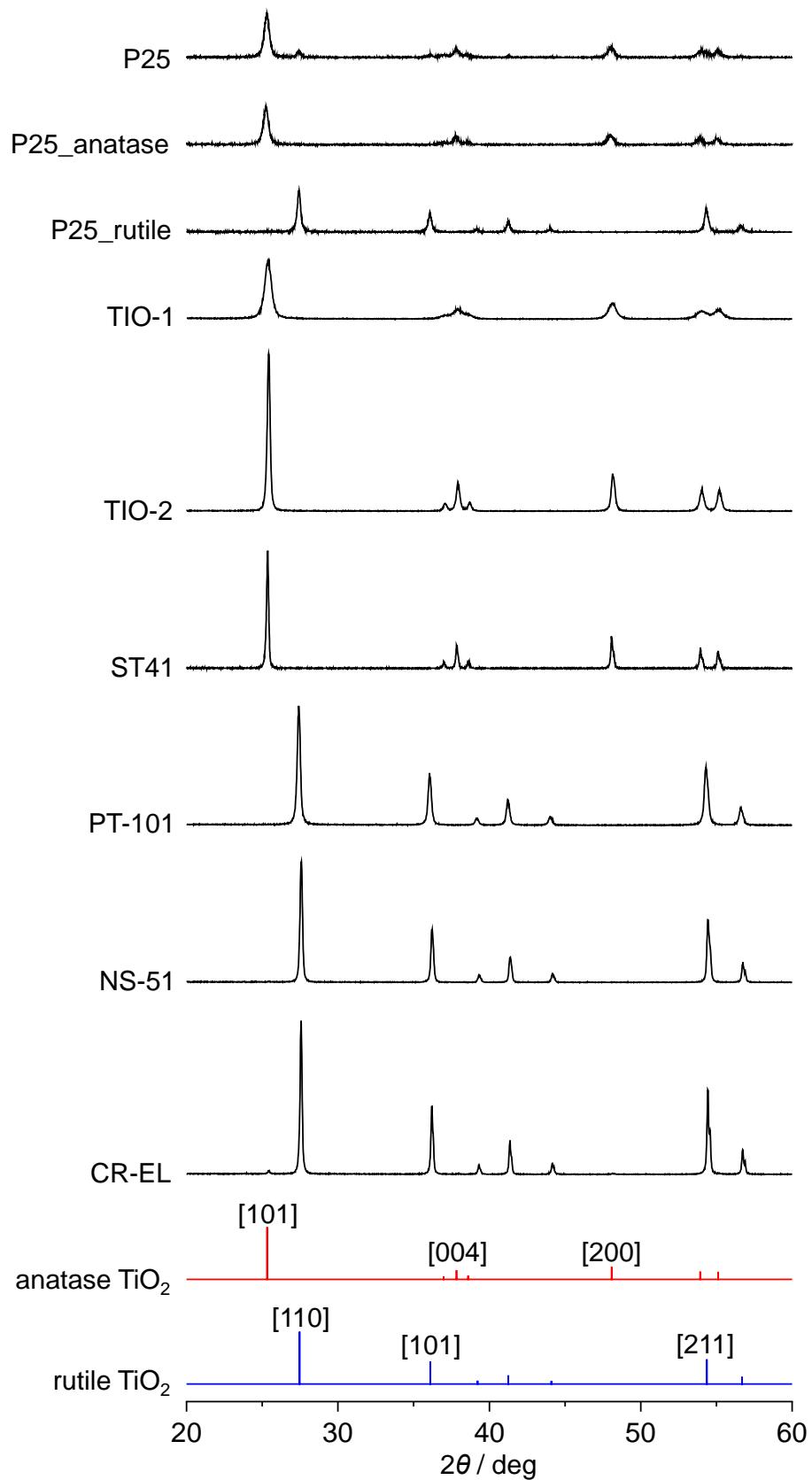
### **Table of Contents**

	Pages
<b>Table S1.</b> Aerobic oxidation of various alcohols.....	S2
<b>Fig. S1</b> XRD patterns for TiO <sub>2</sub> supports.....	S3
<b>Fig. S2</b> Size distribution of TiO <sub>2</sub> supports.....	S4
<b>Fig. S3</b> TEM images of Au <sub>2</sub> /P25 and Au <sub>2</sub> /P25_rutile.....	S5
<b>Fig. S4</b> TEM images and size distribution of Au particles on Au <sub>2</sub> /TiO <sub>2</sub> catalysts.....	S6
<b>Fig. S5</b> XPS chart of Au <sub>2</sub> /TiO <sub>2</sub> .....	S8
<b>Fig. S6</b> Effect of surface oxygen vacancies.....	S9
<b>Fig. S7</b> TEM images and size distribution of Au particle on Au/P25_anatase catalysts.....	S10
<b>Fig. S8</b> DR UV-vis spectra of the catalysts.....	S13
<b>Fig. S9</b> DRIFT spectra of CO-adsorbed on Au/P25_anatase catalysts.....	S14
<b>Fig. S10</b> Relationship between $N_{\text{interface}}$ and photocatalytic activity.....	S15
<b>Fig. S11</b> Photoresponse measurements at different temperatures.....	S16
<b>Fig. S12</b> EIS analysis at different temperatures.....	S16
<b>Fig. S13</b> Light emission spectra of a light source.....	S17

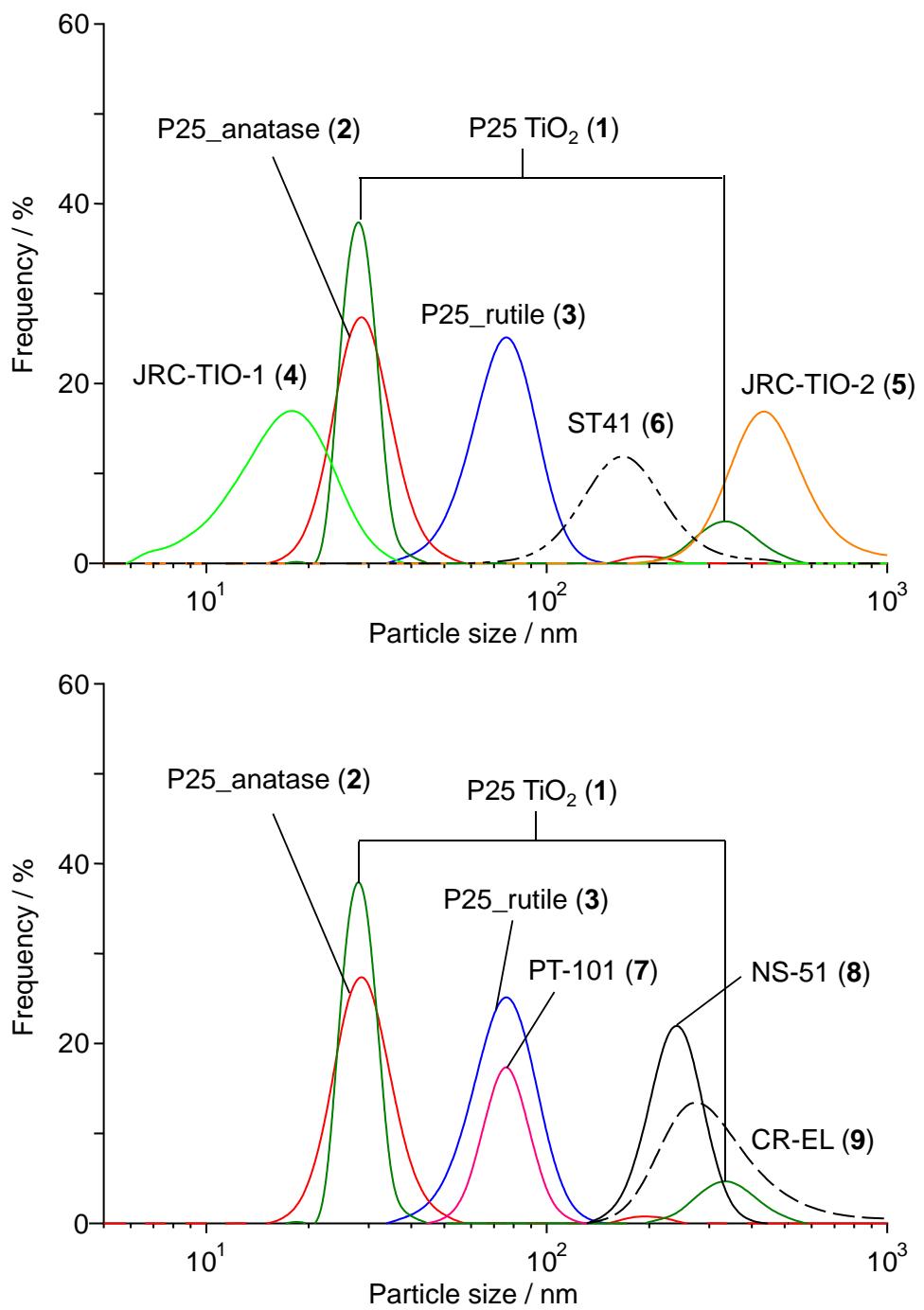
**Table S1.** Aerobic oxidation of various types of alcohols on Au<sub>2</sub>/P25\_anatase under visible light irradiation.<sup>a</sup>

run	Substrate	Time / h	Conv. / % <sup>b</sup>	Product	Yield / % <sup>c</sup>
1		8	>99		>99
2		6	>99		95
3		6	>99		99
4		6	>99		92
5		6	98		96
6		12	>99		97
7		12	96		95
8		12	>99		94
9		24	94		91

<sup>a</sup> Reaction conditions: toluene (5 mL), alcohol (5 mM), Au<sub>2</sub>/P25\_anatase catalyst (20 mg), O<sub>2</sub> (1 atm), Xe lamp ( $\lambda > 450$  nm), light intensity at 450–800 nm (16.8 mW cm<sup>-2</sup>), temperature (298 ± 0.5 K). <sup>b</sup> = (Alcohol converted) / (initial amount of alcohol) × 100. <sup>c</sup> = (Product formed) / (initial amount of alcohol) × 100.

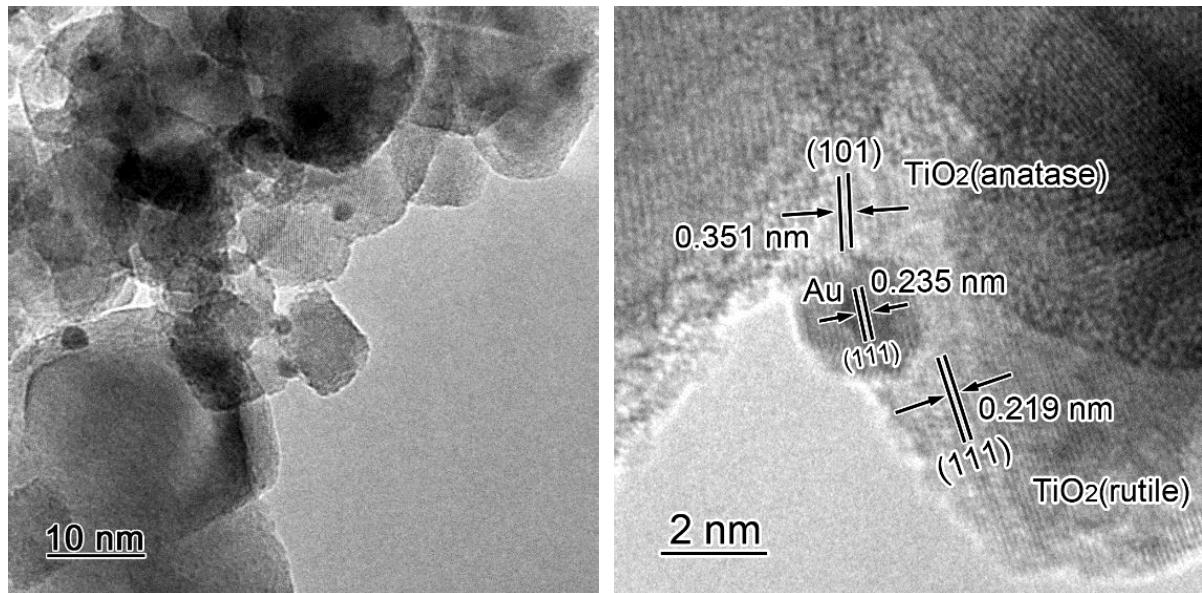


**Fig. S1** XRD patterns of  $\text{TiO}_2$  supports and standard patterns of anatase (JCPDS 21-1272) and rutile  $\text{TiO}_2$  (JCPDS 21-1276).

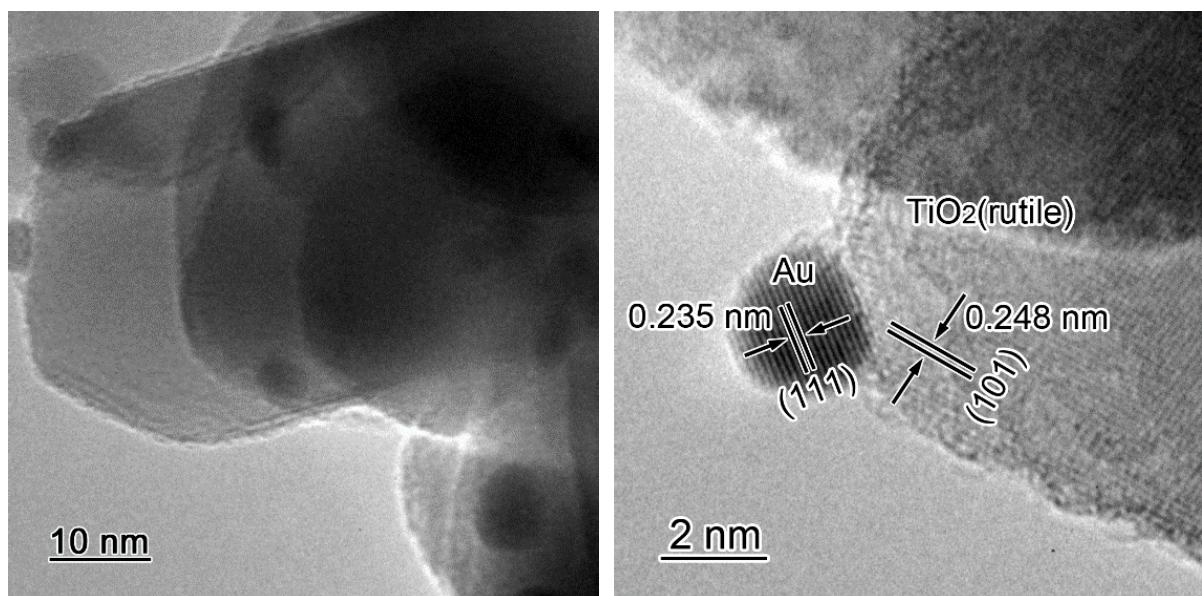


**Fig. S2** Size distribution of  $\text{TiO}_2$  supports determined by DLS analysis.

**Au<sub>2</sub>/P25 (1)**

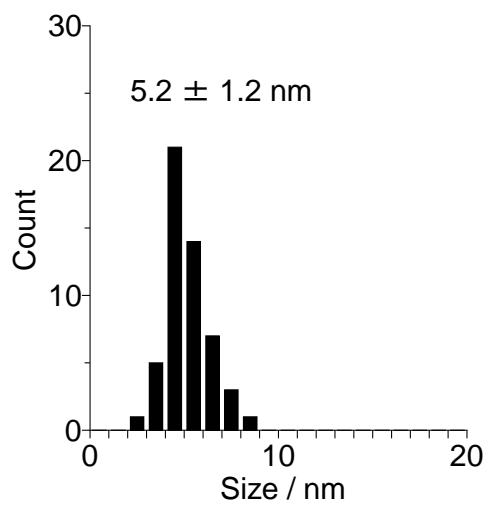
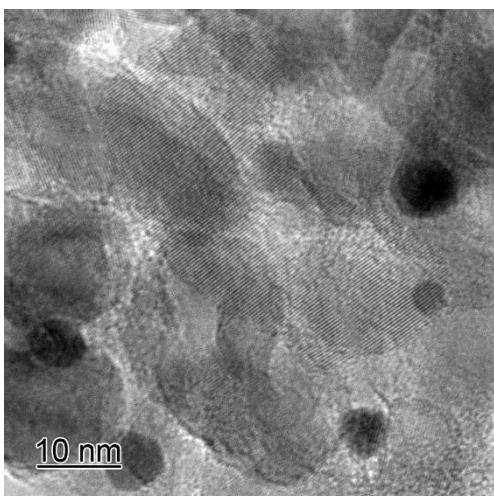


**Au<sub>2</sub>/P25\_rutile (3)**

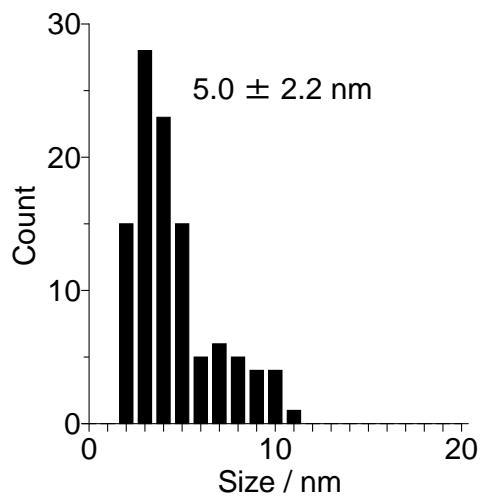
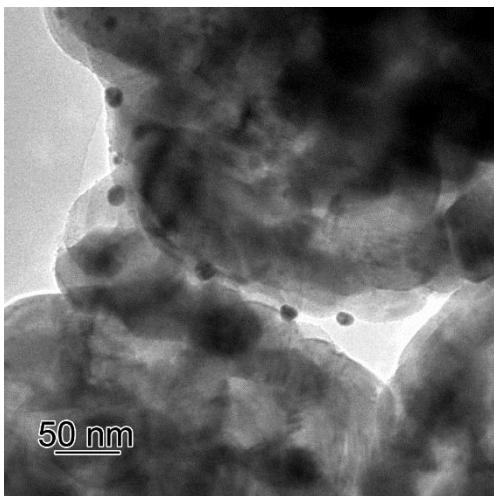


**Fig. S3** Typical TEM images of Au<sub>2</sub>/P25 and Au<sub>2</sub>/P25\_rutile catalysts.

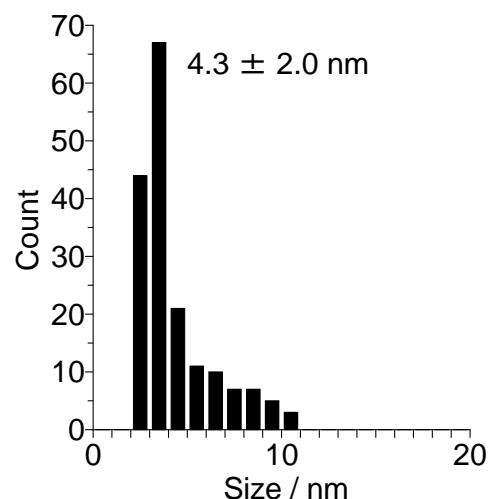
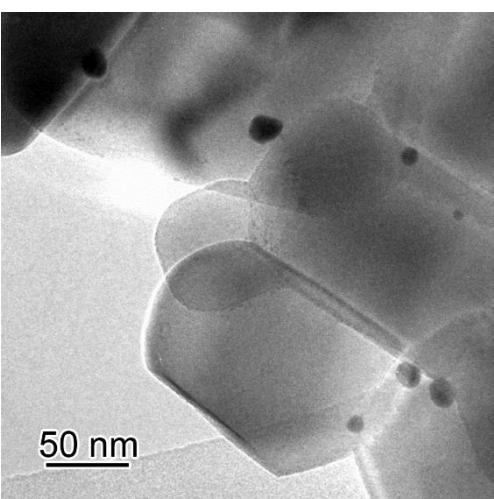
### **Au<sub>2</sub>/TIO-1 (4)**



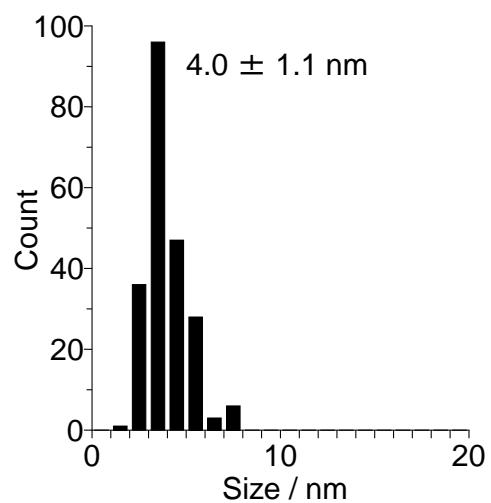
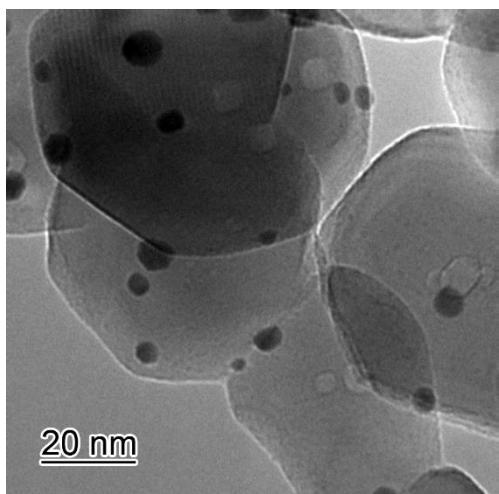
### **Au<sub>2</sub>/TIO-2 (5)**



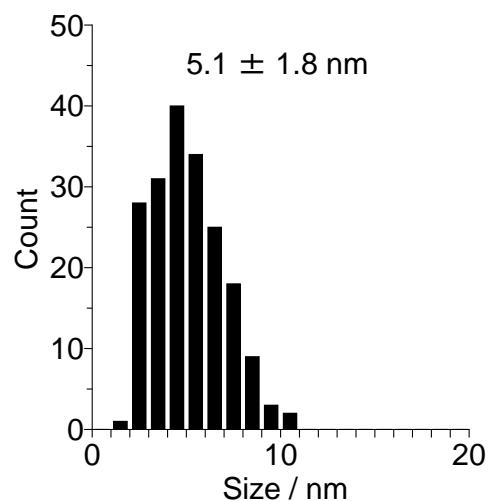
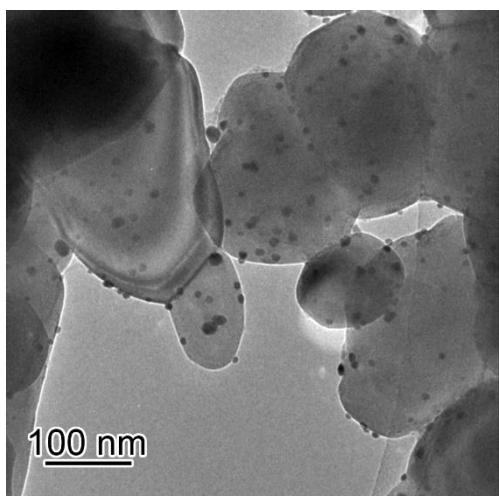
### **Au<sub>2</sub>/ST41 (6)**



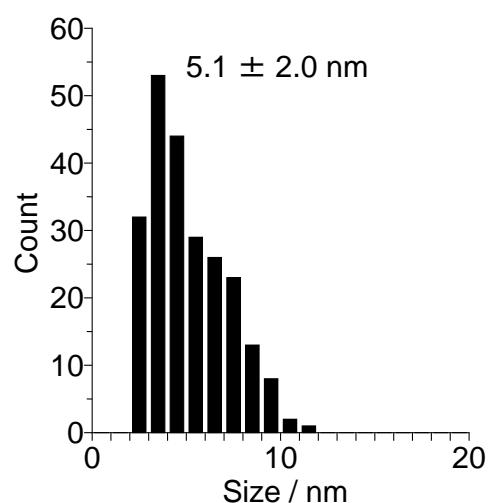
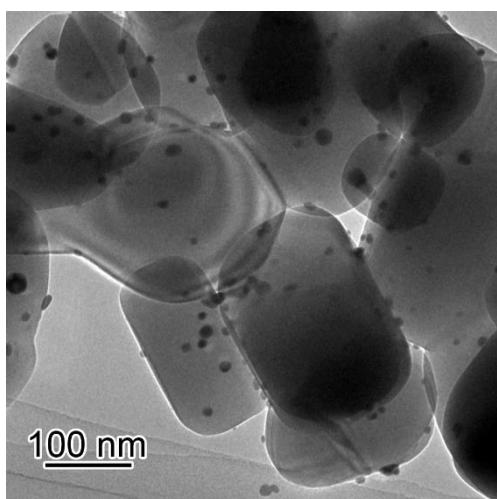
**Au<sub>2</sub>/PT-101 (7)**



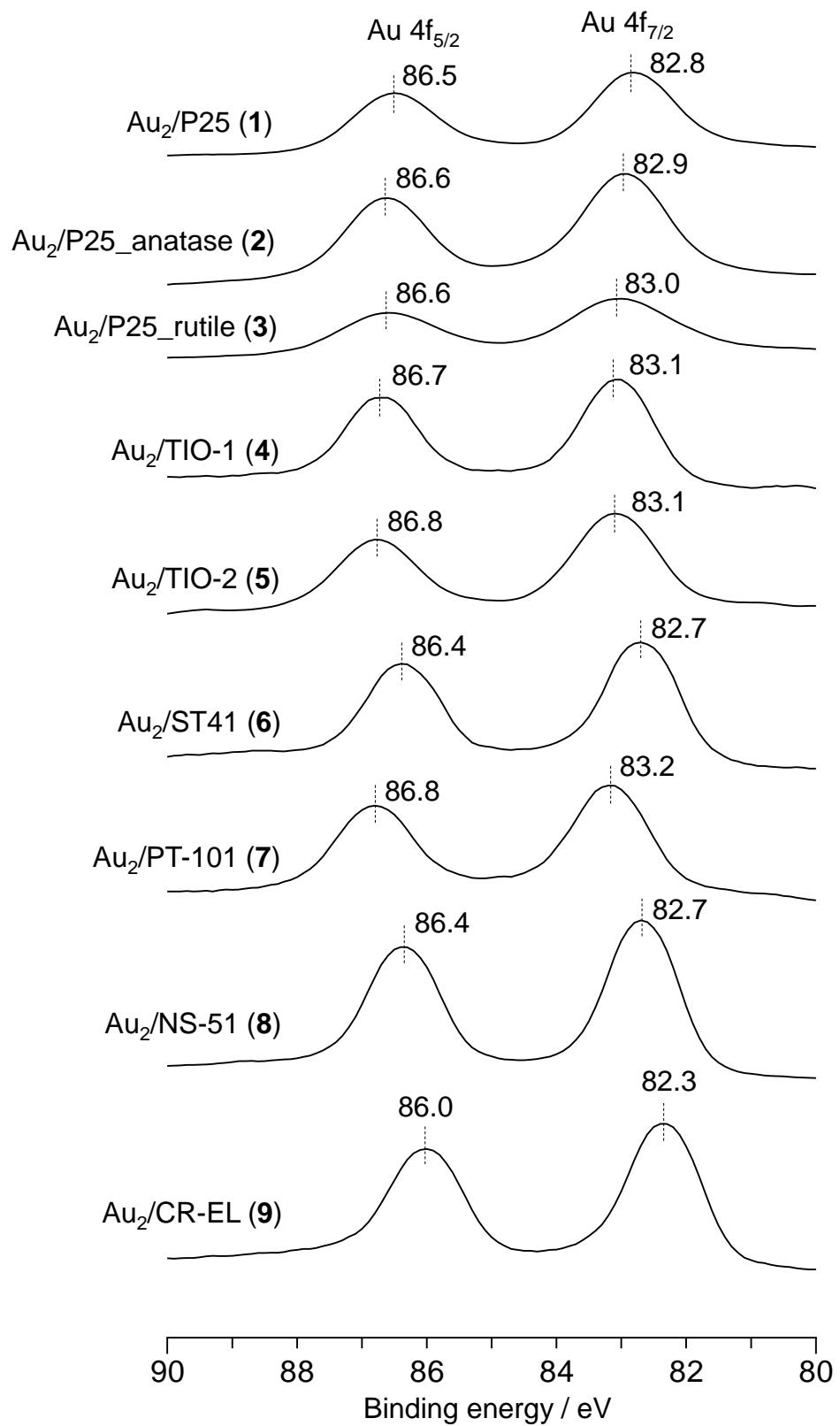
**Au<sub>2</sub>/NS-51 (8)**



**Au<sub>2</sub>/CR-EL (9)**



**Fig. S4** Typical TEM images and size distributions of Au/TiO<sub>2</sub> catalysts.

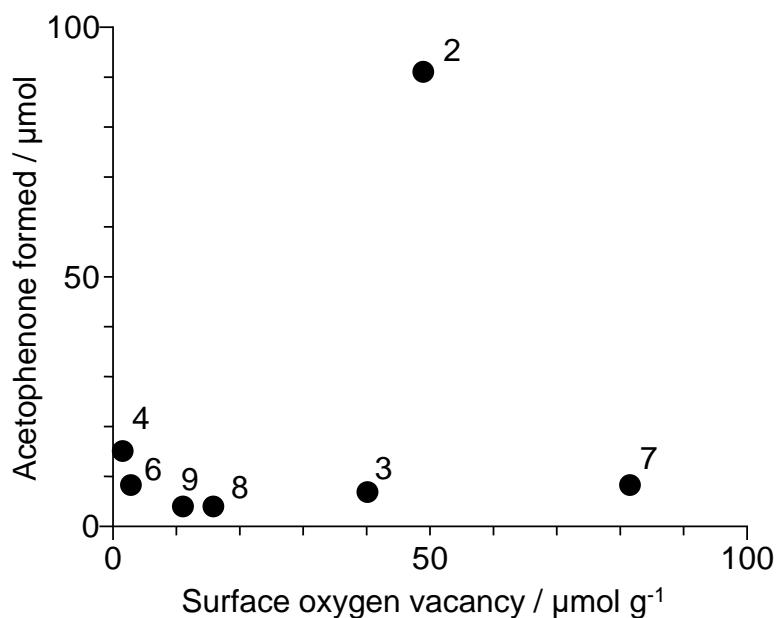


**Fig. S5** XPS chart (Au 4f level) of the respective  $\text{Au}_2/\text{TiO}_2$  catalysts.

The number of surface oxygen vacancies on the respective TiO<sub>2</sub> particles.<sup>a</sup>

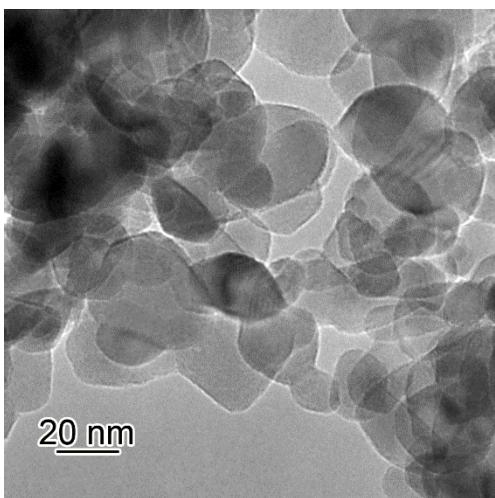
sample	TiO <sub>2</sub>	Surface oxygen vacancy / $\mu\text{mol g}^{-1}$
<b>2</b>	P25_anatase	49
<b>3</b>	P25_rutile	40.2
<b>4</b>	TIO-1	1.6
<b>6</b>	ST-41	2.9
<b>7</b>	PT101	81.6
<b>8</b>	NS-51	15.9
<b>9</b>	CR-EL	11.1

<sup>a</sup> The number of surface oxygen vacancies on TiO<sub>2</sub> was determined by DRIFT analysis with nitrobenzene as a probe molecule (Shiraishi, Y.; Hirakawa, H.; Togawa, Y.; Sugano, Y.; Ichikawa, S.; Hirai, T. *ACS Catal.* **2013**, *3*, 2318–2326).

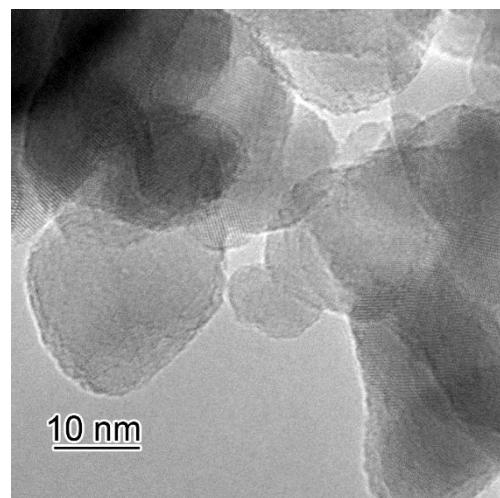


**Fig. S6** Relationship between the number of surface oxygen vacancies on the respective TiO<sub>2</sub> supports and (a) the amount of acetophenone formed on the respective Au/TiO<sub>2</sub> catalysts by aerobic oxidation of 1-phenylethanol under visible light ( $\lambda > 450$  nm, 12 h). The photoreaction conditions are identical to those in Fig. 3 (manuscript).

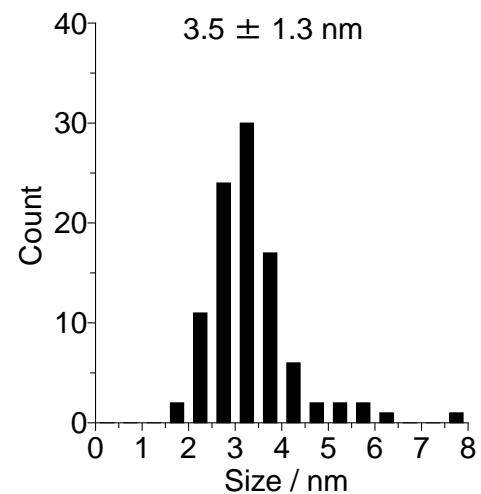
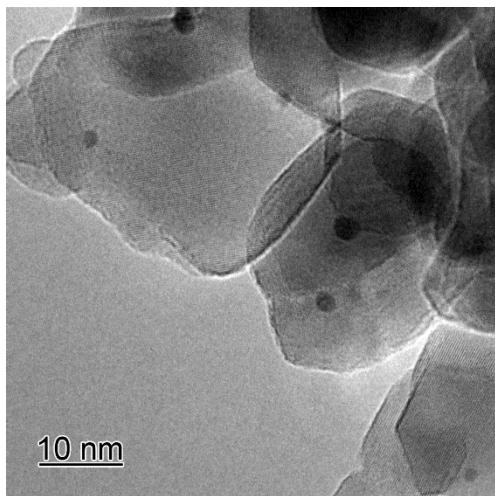
**Au<sub>0.5</sub>/P25\_anatase**



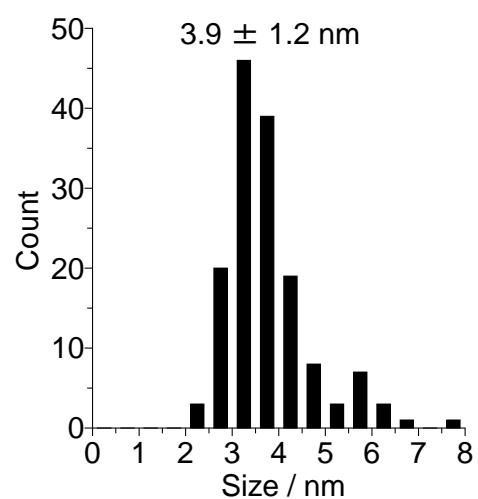
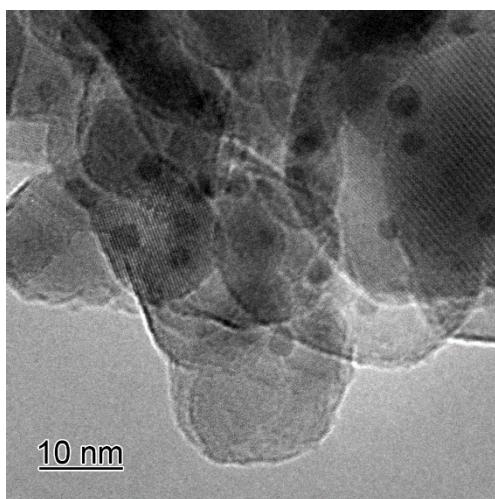
**Au<sub>1</sub>/P25\_anatase**



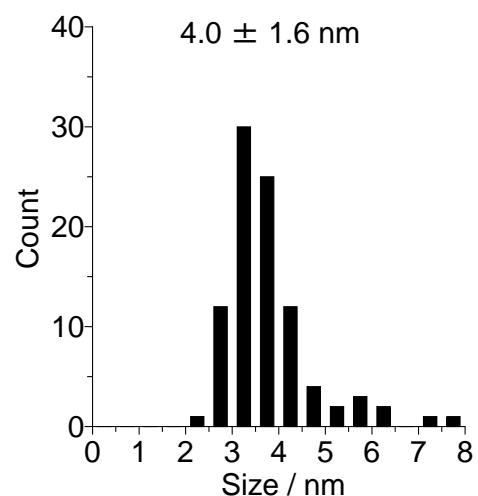
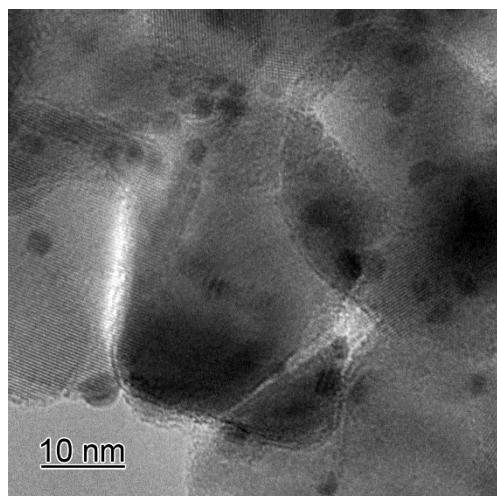
**Au<sub>1.5</sub>/P25\_anatase**



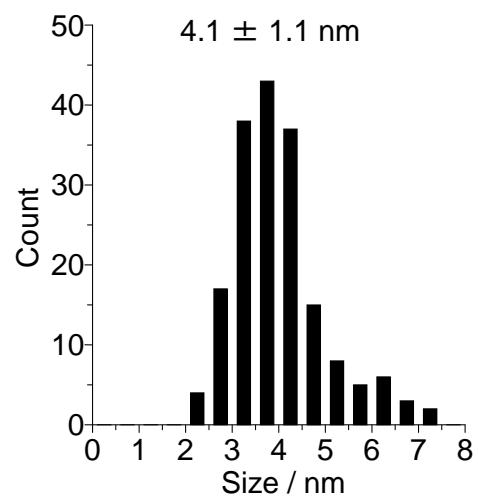
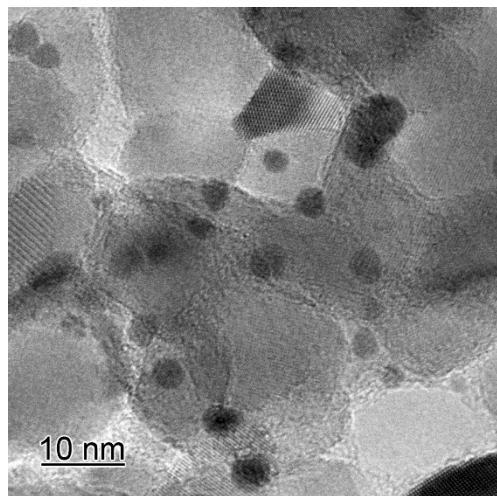
**Au<sub>3</sub>/P25\_anatase**



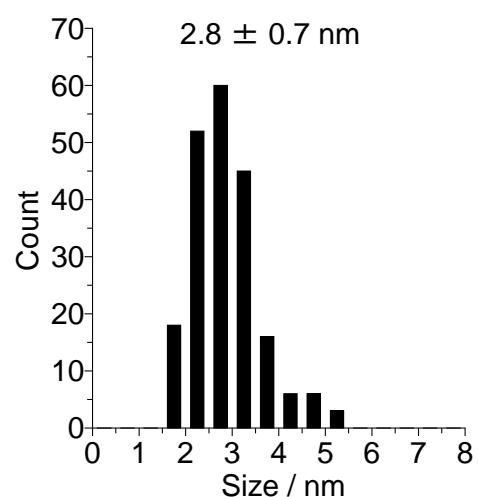
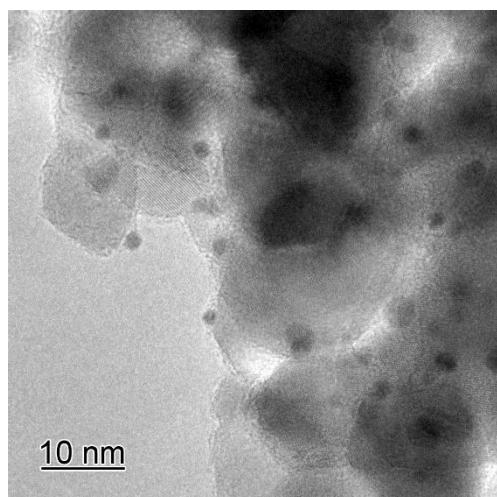
### Au<sub>4</sub>/P25\_anatase



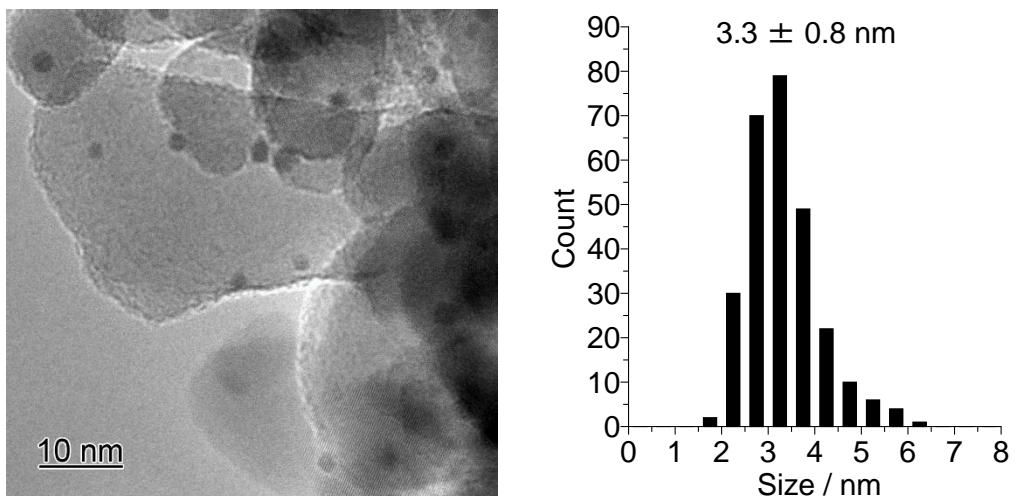
### Au<sub>5</sub>/P25\_anatase



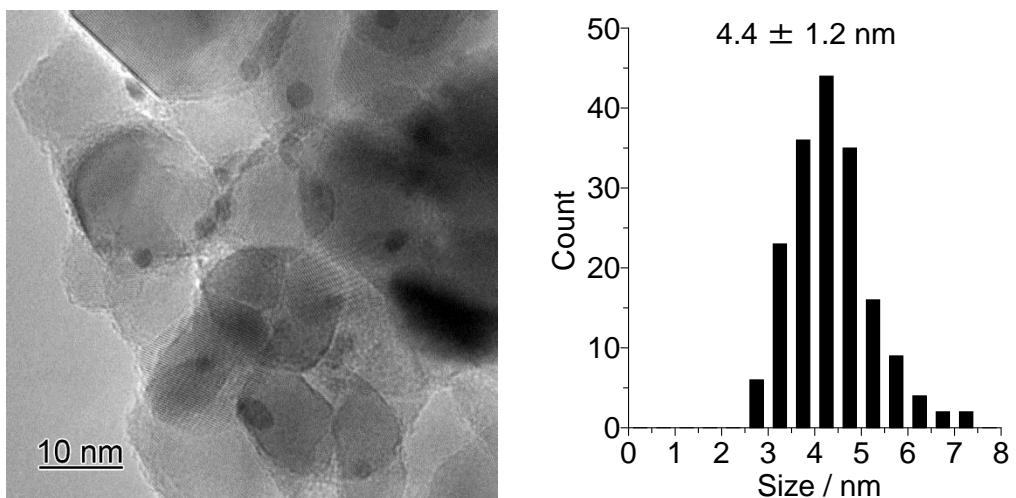
### Au<sub>2</sub>/P25\_anatase prepared by calcination at 473 K



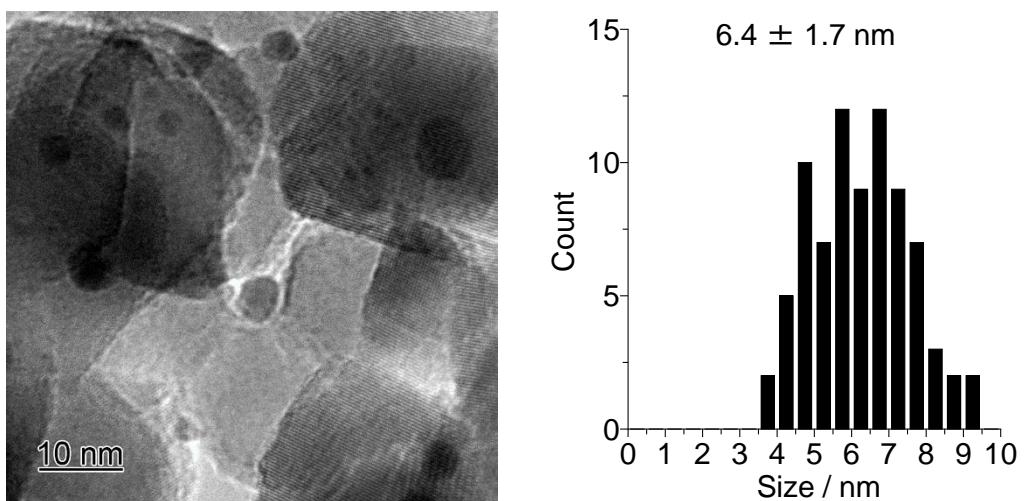
**Au<sub>2</sub>/P25\_anatase prepared by calcination at 573 K**



**Au<sub>2</sub>/P25\_anatase prepared by calcination at 723 K**

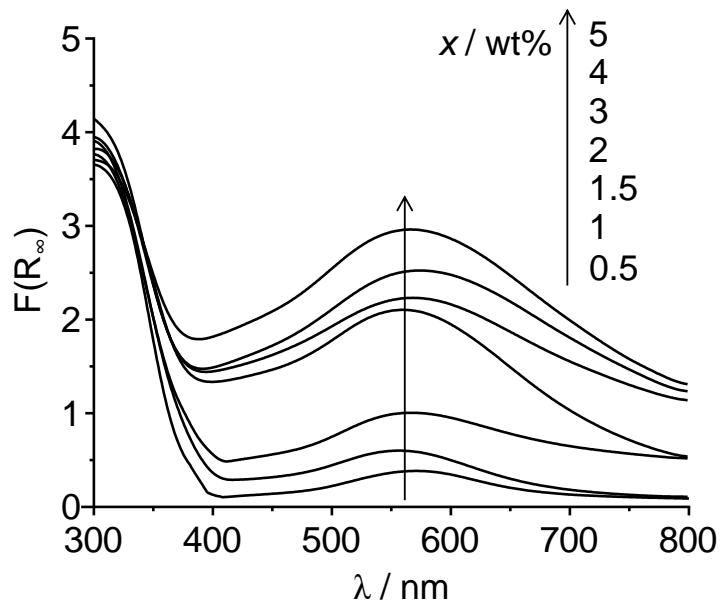


**Au<sub>2</sub>/P25\_anatase prepared by calcination at 773 K**

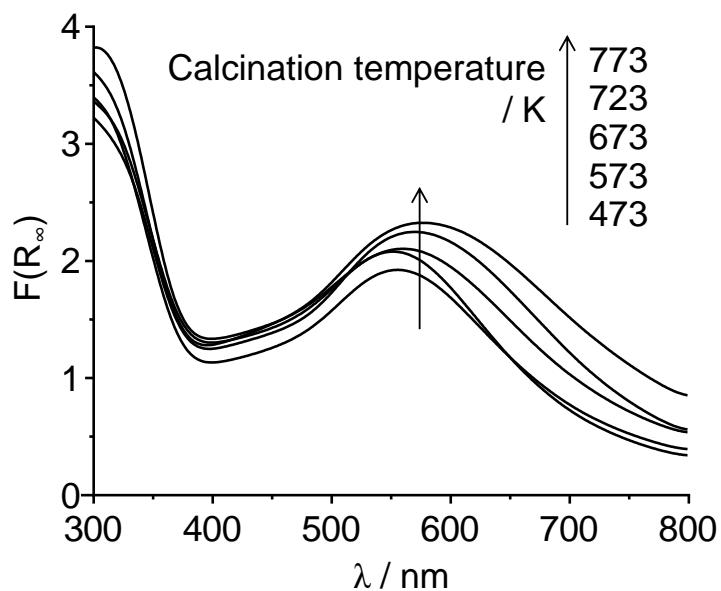


**Fig. S7** Typical TEM images and size distributions of the Au<sub>x</sub>/P25\_anatase and Au<sub>2</sub>/P25\_anatase catalysts prepared at different calcination temperature.

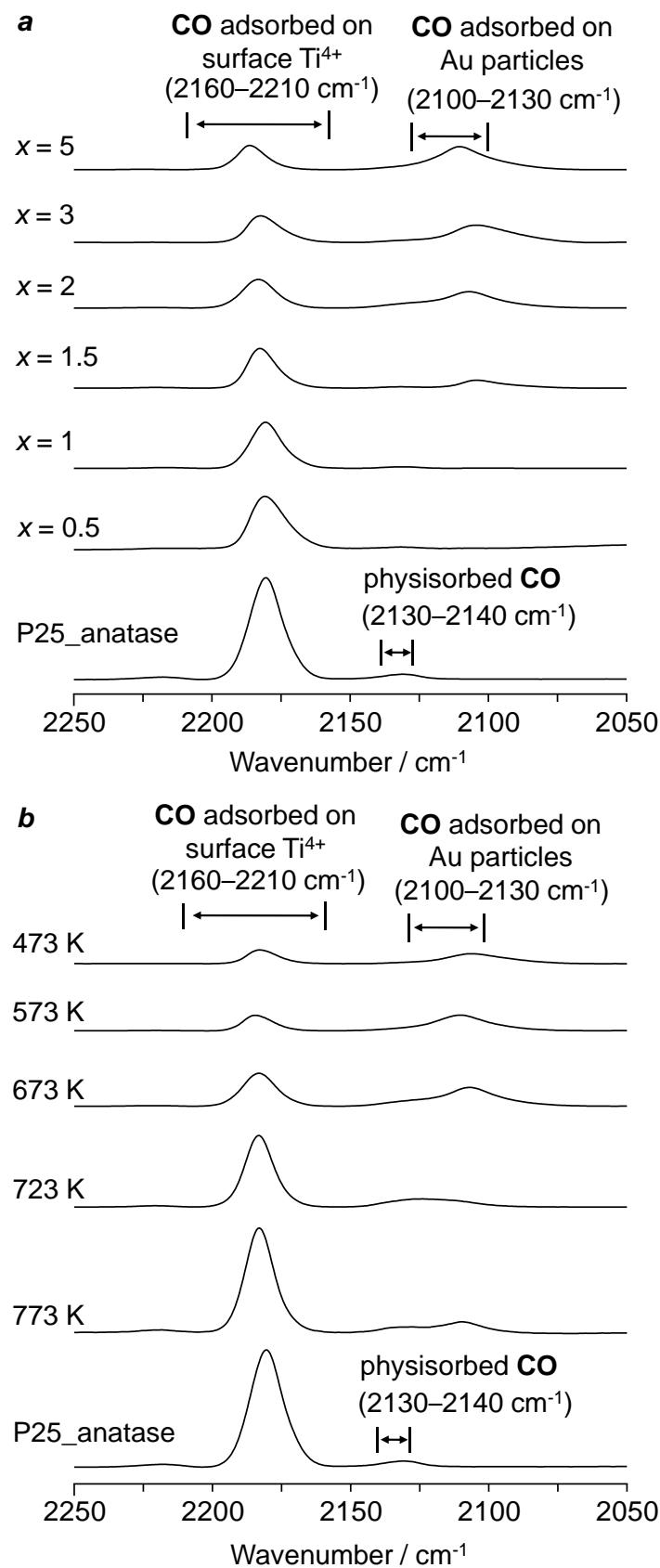
**Au<sub>x</sub>/P25\_anatase**



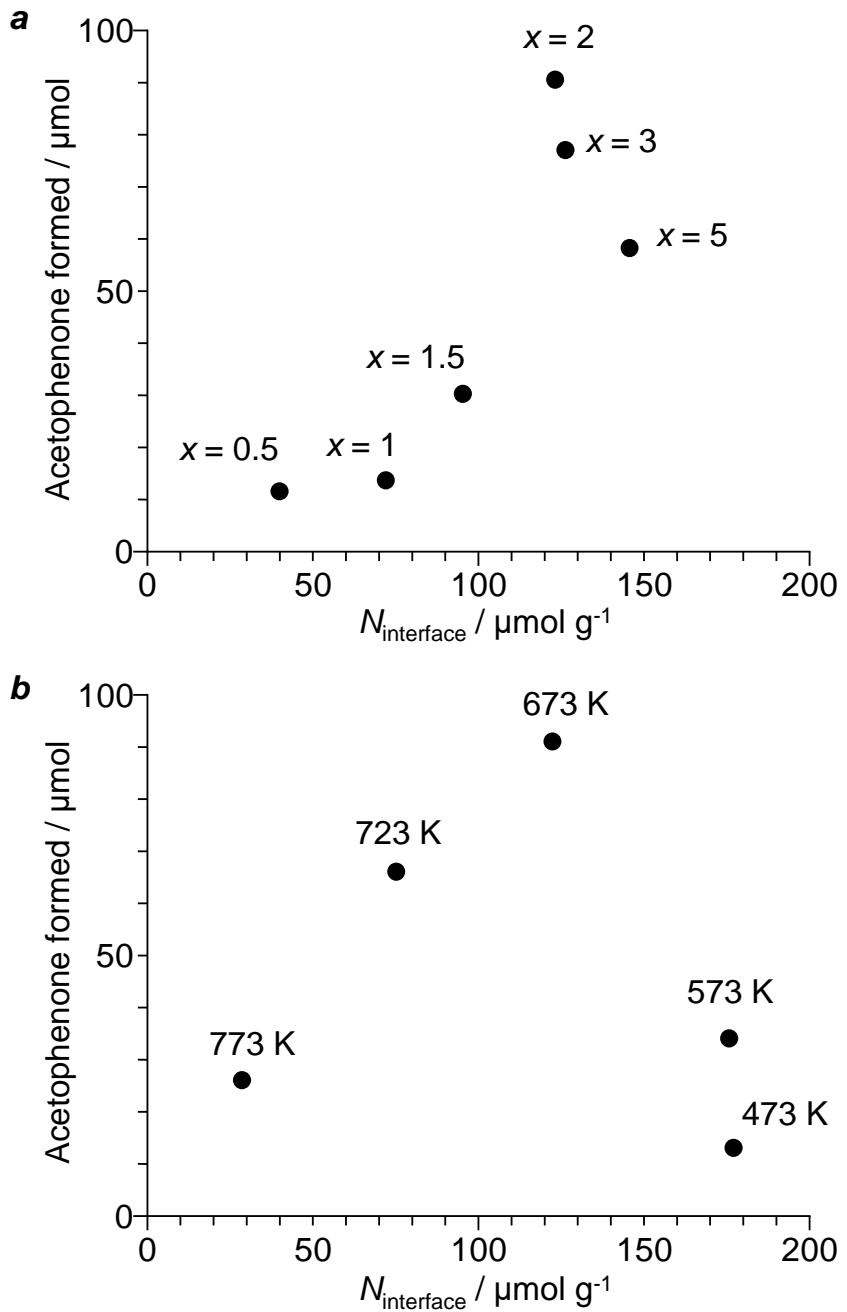
**Au<sub>2</sub>/P25\_anatase prepared at different calcination temperatures**



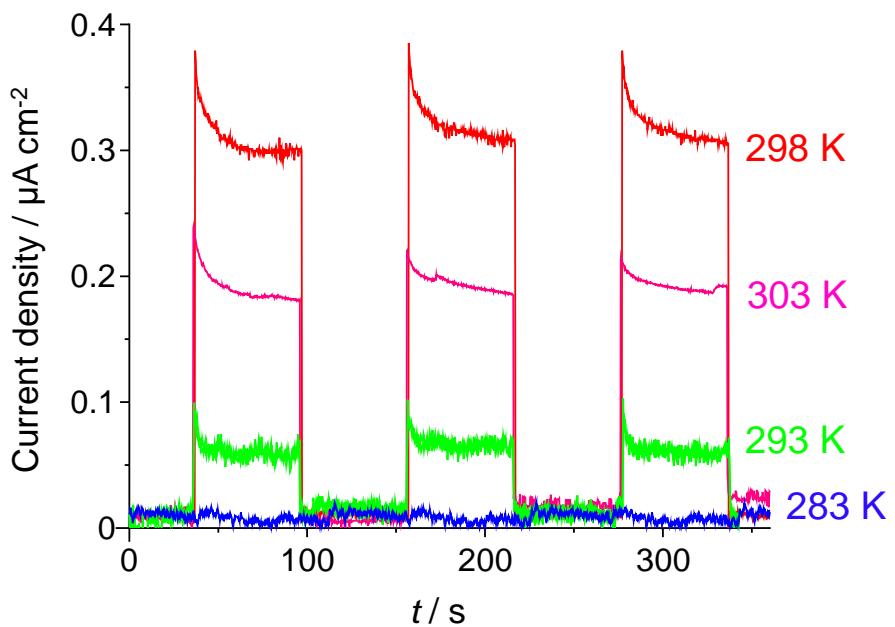
**Fig. S8** DR-UV-vis spectra of the catalysts.



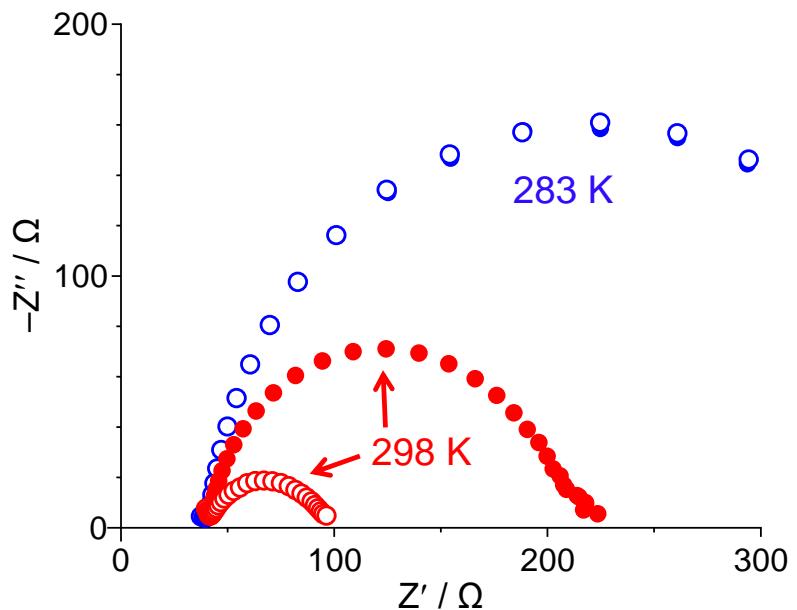
**Fig. S9** FTIR spectra of CO adsorbed on (a)  $\text{Au}_x/\text{P25\_anatase}$  catalysts and (b)  $\text{Au}_2/\text{P25\_anatase}$  catalysts prepared at different calcination temperatures.



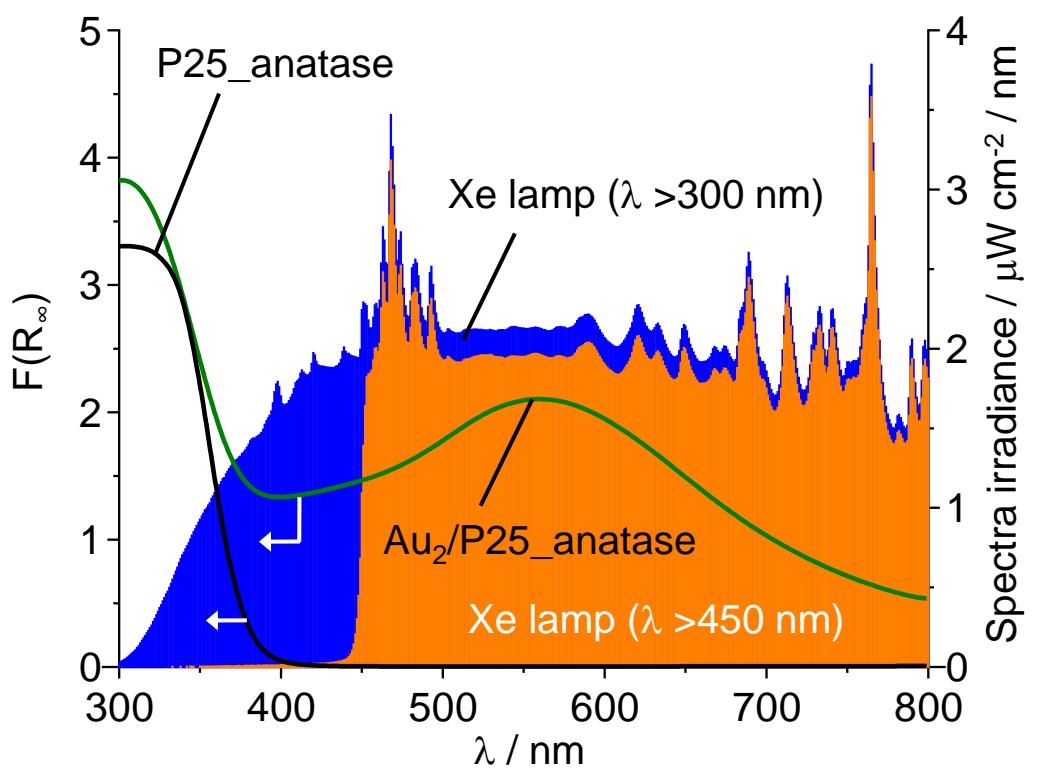
**Fig. S10** Relationship between  $N_{\text{interface}}$  and the amount of acetophenone formed by aerobic oxidation of 1-phenylethanol (12 h) performed under visible light irradiation, with (a)  $\text{Au}_x/\text{P25\_anatase}$  catalysts and (b)  $\text{Au}_2/\text{P25\_anatase}$  catalysts prepared at different calcination temperature. Reaction conditions are identical to those in Fig. 3 (manuscript).



**Fig. S11** Photocurrent response of  $\text{Au}_2/\text{P25\_anatase}$  measured on a FTO glass electrode at different temperatures. The measurements were carried out in 0.5 M  $\text{Na}_2\text{SO}_4$  under visible light ( $\lambda > 450 \text{ nm}$ ) at a bias of 0.1 V.



**Fig. S12** EIS Nyquist plots of  $\text{Au}_2/\text{P25\_anatase}$  in 0.5 M  $\text{Na}_2\text{SO}_4$  solution at a bias of 2 V (vs open circuit) in the frequency range of 10 mHz to 100 kHz, measured (closed symbols) in the dark or (open symbols) under visible light irradiation at different temperatures.



**Fig. S13** Light emission spectra of Xe lamp (blue) without filter, (orange) with  $\lambda > 450$  nm filter. DR UV-vis spectra of (green)  $\text{Au}_2/\text{P25\_anatase}$  and (black line)  $\text{P25\_anatase}$  are also shown here.