

## **Reinforced poly(hydroxyurethane) thermosets as high performance adhesives for aluminum substrates**

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### Synthesis and characterizations of TMPTC

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Figure S1: <sup>1</sup>H NMR spectra of TMPTE and TMPTC

Figure S2: FTIR spectra of TMPTE and TMPTC

Figure S3: <sup>1</sup>H NMR spectra of GPTMS and C-GPTMS

Figure S4: FTIR spectra of GPTMS and C-GPTMS

Figure S5: TGA of

- a) PHUs reinforced by 3wt% of ZnO, GPTMS-ZnO and C-GPTMS-ZnO fillers
- b) PHUs reinforced by 5wt% of ZnO, GPTMS-ZnO and C-GPTMS-ZnO fillers

Figure S6: TEM micrographs of

- a) reinforced non-functionalized (PHU 5N),
- b) epoxy functionalized (PHU-5E) and c) cyclic carbonate functionalized (PHU 5C) formulations. (scale bar = 1 μm).

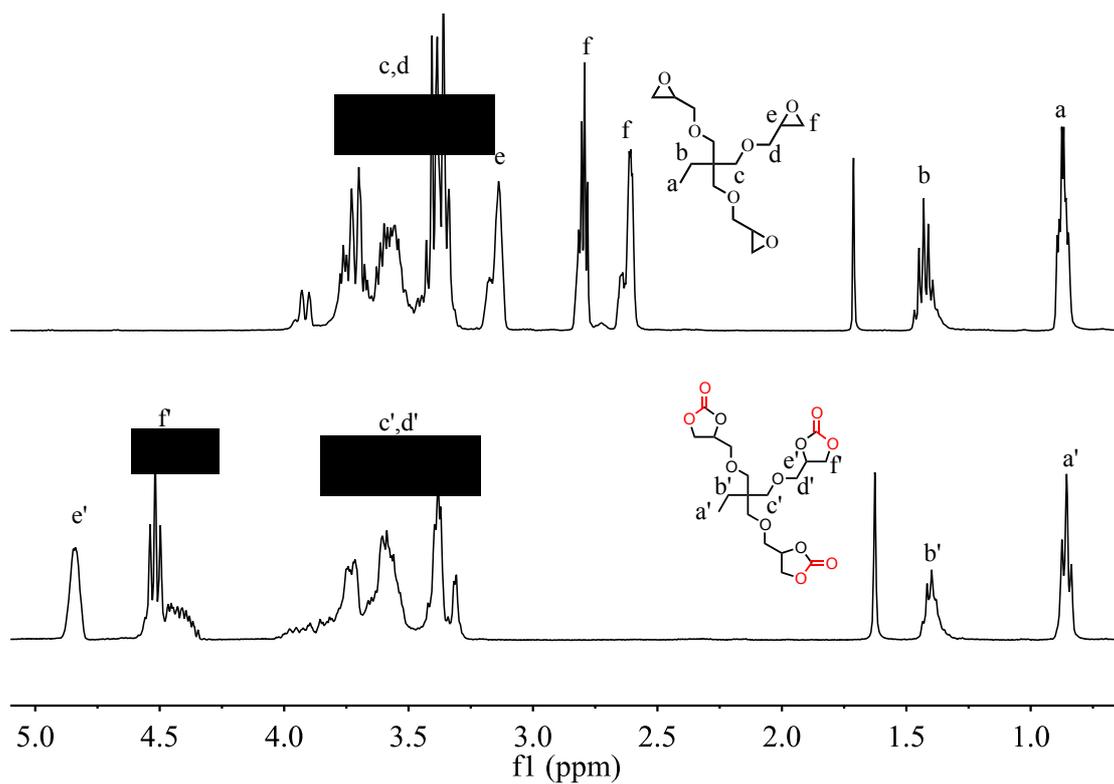
Figure S7: DSC of PHUs reinforced by 3, 5wt% of ZnO, GPTMS-ZnO and C-GPTMS-ZnO fillers.

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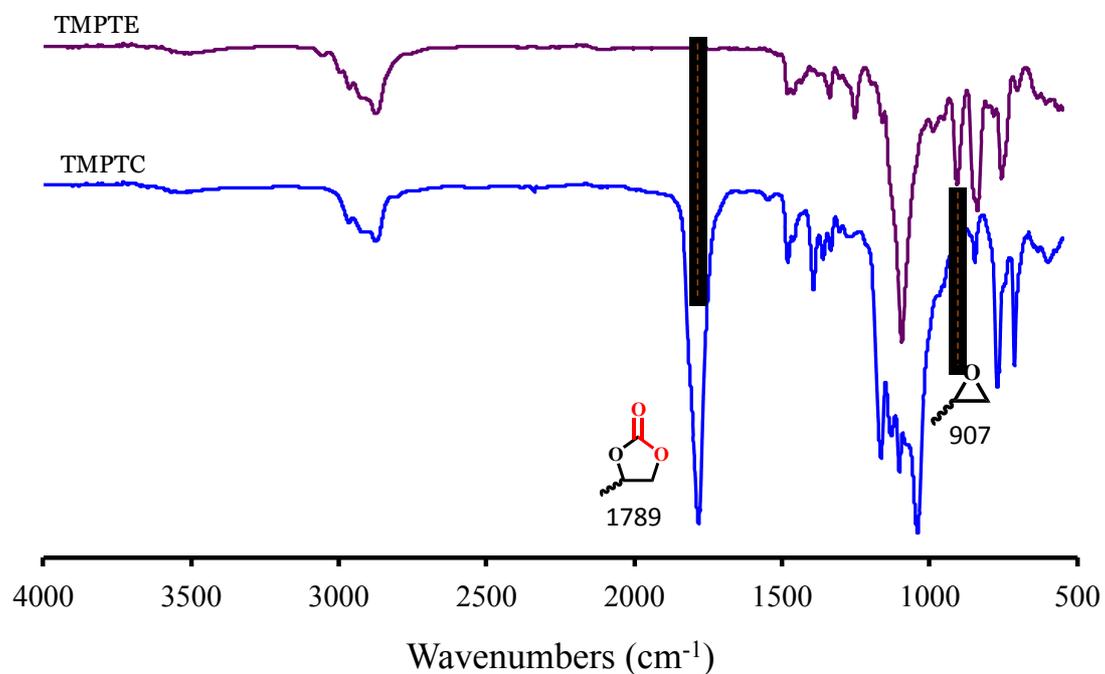
**Synthesis of 4,4'-(((2-ethyl-2-(((2-oxo-1,3-dioxolan-4-yl)methoxy) methyl) propane-1,3-diyl) bis(oxy))bis(methylene))bis(1,3-dioxolan-2-one) (TMPTC).**

TMPTC was synthesized by coupling CO<sub>2</sub> with epoxide, using a bicomponent organocatalyst that combined tetrabutylammonium iodide (TBAI) as catalyst with 1,3-bis(2-hydroxyhexafluoroisopropyl)benzene as activator. 34g (0.112 mol) of trimethylolpropane triglycidyl ether (TMPTE) was introduced in a 80 ml high pressure cell equipped with a mechanical stirrer and prior addition of 2.5 mol% (with respect to the epoxy content) of bicomponent organocatalyst combining TBAI (2.81mmol, 1.0384 g) and 1,3-bis(2-hydroxyhexafluoroisopropyl)benzene (2.81mmol, 0.698 mL). Then the reaction mixture was heated to 80°C prior equilibration of the cell at a CO<sub>2</sub> pressure of 100 bar for 24 h. The reactor was depressurized slowly to release unreacted CO<sub>2</sub>. The resulting product was collected as a viscous liquid that was used without any further purification.

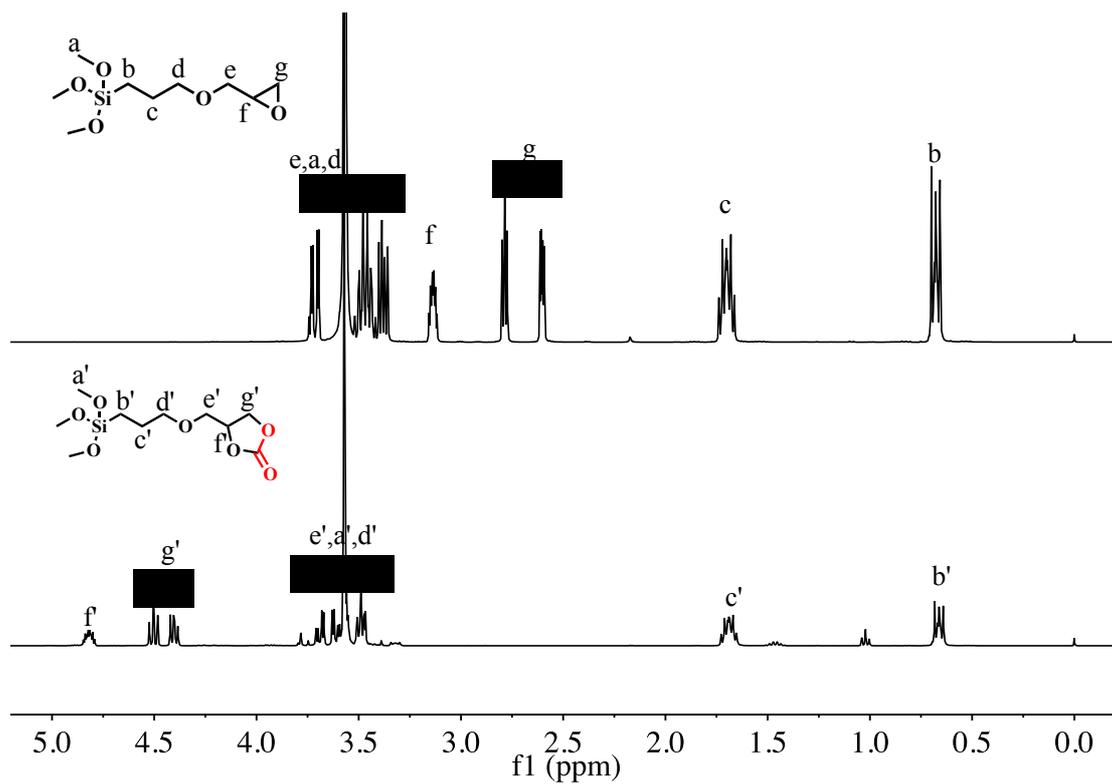
The complete conversion of TMPTE into TMPTC was confirmed by <sup>1</sup>H NMR spectroscopy with the disappearance of the peaks characteristic of -CH-O- and -CH<sub>2</sub>-O- of epoxides at 3.15 and 2.5-2.8 ppm, respectively, and the appearance of new signals of -CH-OC(=O)O- and -CH<sub>2</sub>-OC(=O)O- of cyclic carbonates at 4.8 and 4.25-4.6 ppm, respectively (Figure S1). Formation of the cyclic carbonate was further confirmed by FT-IR spectroscopy highlighting the presence of a strong signal at 1789 cm<sup>-1</sup> corresponding to the elongation of the C=O group (Figure S2).



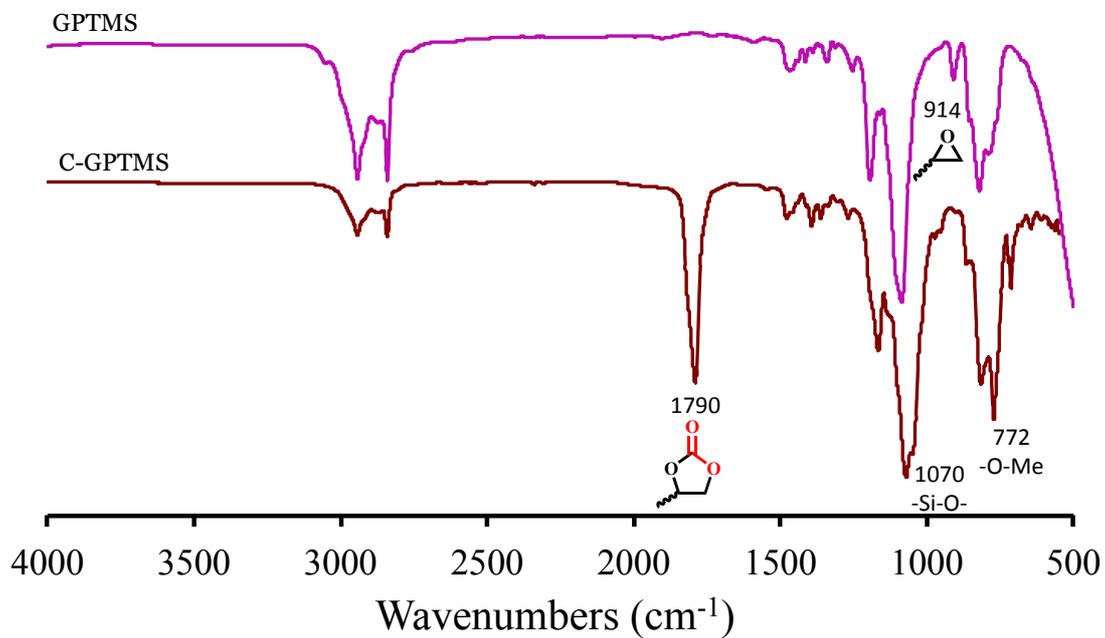
**Figure S1:**  $^1\text{H}$  NMR spectra of TMPTE (top) and TMPTC (bottom).



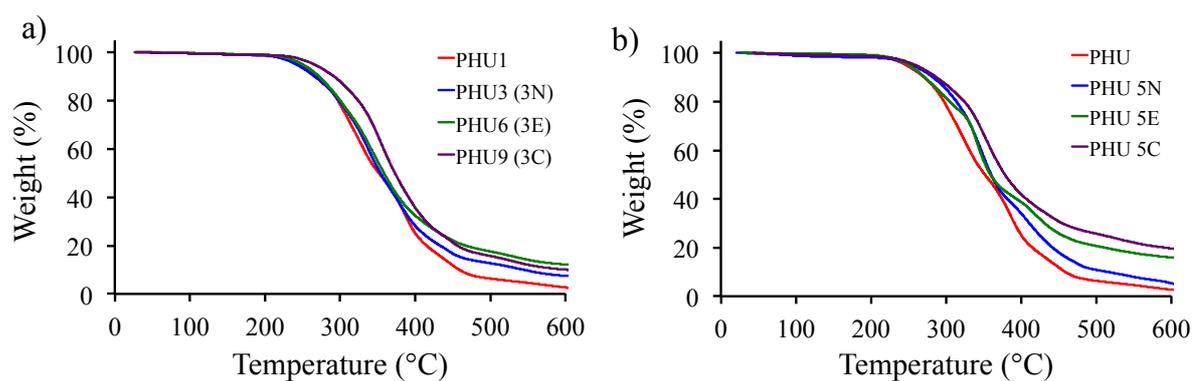
**Figure S2:** FTIR spectra of TMPTE and TMPTC



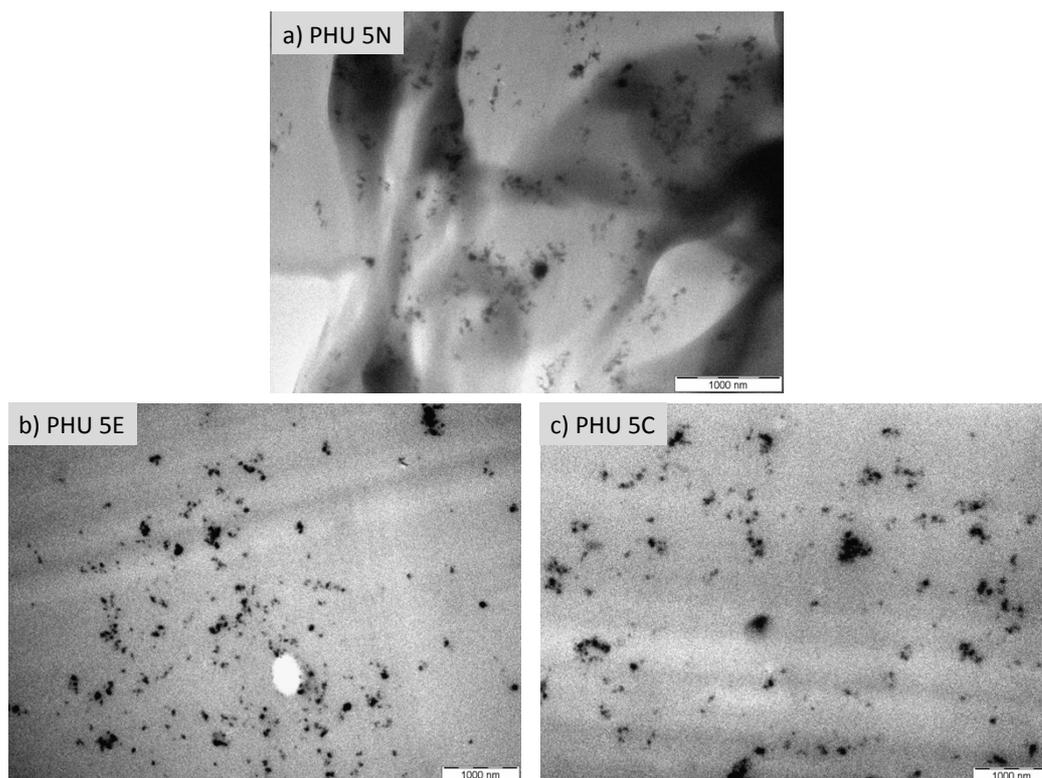
**Figure S3:** <sup>1</sup>H NMR spectra of GPTMS (top) and C-GPTMS (bottom)



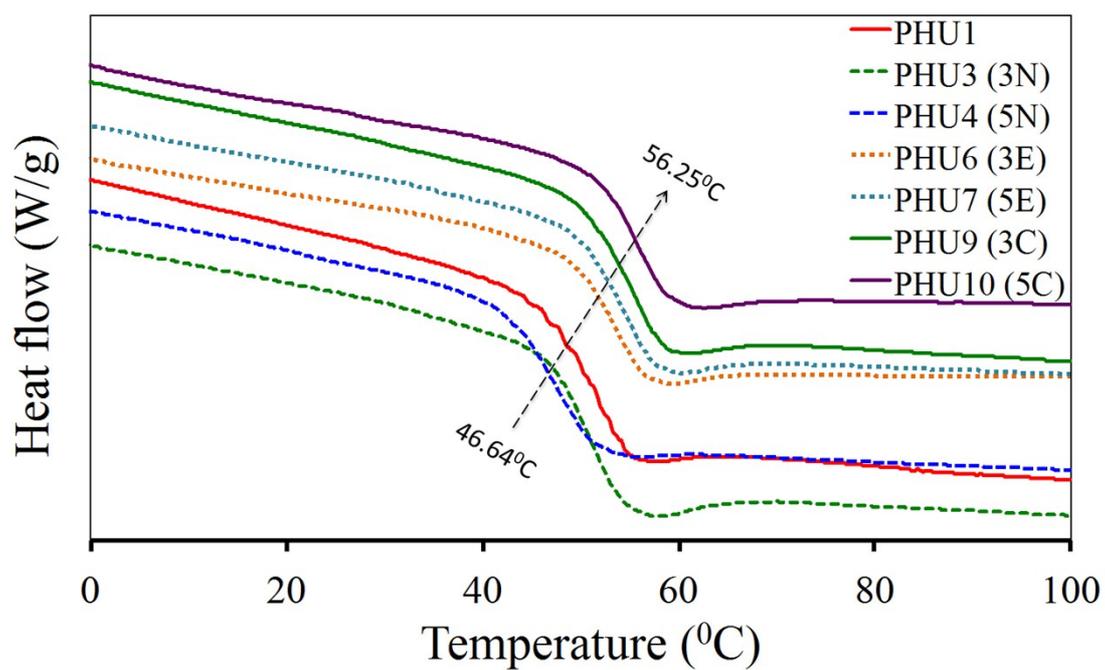
**Figure S4:** FTIR spectra of GPTMS and C-GPTMS



**Figure S5:** TGA of a) PHUs reinforced by 3wt% of ZnO, GPTMS-ZnO and C-GPTMS-ZnO fillers. b) PHUs reinforced by 5wt% of ZnO, GPTMS-ZnO and C-GPTMS-ZnO fillers.



**Figure S6.** TEM micrographs of a) reinforced non-functionalized (PHU 5N), b) epoxy functionalized (PHU-5E) and c) cyclic carbonate functionalized (PHU 5C) formulations. (scale bar = 1 $\mu$ m).



**Figure S7:** DSC of PHUs reinforced by 3, 5wt% of ZnO, GPTMS-ZnO and C-GPTMS-ZnO fillers.