

Electronic Supplementary Information

Supramolecular Assembly of Phosphole-Based Moiety into Nanostructures Dictated by Alkynylplatinum(II) Terpyridine Complexes Through Non-Covalent Pt···Pt and π - π Stacking Interactions: Synthesis, Characterization, Photophysics and Self-Assembly Behaviors

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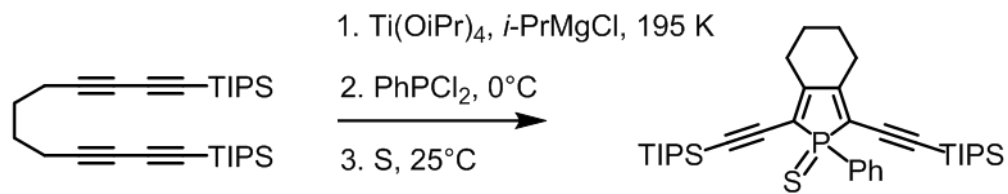
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Table S1. Crystal data and structure refinement at 150 K for **2** after this ‘squeeze’ treatment and before ‘squeeze’ treatment.

	2	2·2OTf.
Molecular formula	C ₇₂ H ₈₃ Pt ₂ S ₁ N ₆ P ₁	C ₈₀ H ₈₃ Cl ₁₂ Pt ₂ S ₃ N ₆ O ₆ P ₁
Molecular weight	1485.65	2281.25
<i>a</i> (Å)	13.7722(6)	13.7722(6)
<i>b</i> (Å)	39.9563(16)	39.9563(16)
<i>c</i> (Å)	17.7098(7)	17.7098(7)
α (°)	90	90
β (°)	97.536(3)	97.536(3)
γ (°)	90	90
<i>V</i> (Å ³)	9661.3(7)	9661.3(7)
<i>Z</i>	4	4
<i>D_c</i> (g cm ⁻³)	1.021	1.569
Crystal system	Monoclinic	Monoclinic
Space group	P21/c	P21/c
Temperature (K)	150(2)	150(2)
Wavelength Mo- <i>K</i> α (Å)	0.71069	0.71069
Crystal size (mm)	0.18 * 0.05 * 0.04	0.18 * 0.05 * 0.04
μ (mm ⁻¹)	2.962	3.367
<i>F</i> (000)	2976	4520
θ limit (°)	1.02–27.50	1.02–27.50
Index ranges <i>hkl</i>	-16 ≤ <i>h</i> ≤ 17, -45 ≤ <i>k</i> ≤ 51, -22 ≤ <i>l</i> ≤ 14	-16 ≤ <i>h</i> ≤ 17, -45 ≤ <i>k</i> ≤ 51, -22 ≤ <i>l</i> ≤ 14
Reflections collected	44633	44633

Independent reflections	21914	21914
Reflections [$I > 2\sigma(I)$]	9727	10760
Data/restraints/parameters	21914 / 0 / 717	21914 / 0 / 1021
Goodness-of-fit on F^2	0.893	1.110
Final R indices [$I > 2\sigma(I)$]	$R1 = 0.0780$	$R1 = 0.01182$
	$wR2 = 0.2031$	$wR2 = 0.2794$
R indices (all data)	$R1 = 0.1478$	$R1 = 0.2259$
	$wR2 = 0.2237$	$wR2 = 0.3411$
Largest diff peak and hole ($e \text{ \AA}^{-3}$)	1.616 and -2.872	2.265 and -3.273



Scheme S1. The detailed synthetic route for the phosphole-based bridging alkynyl ligand.

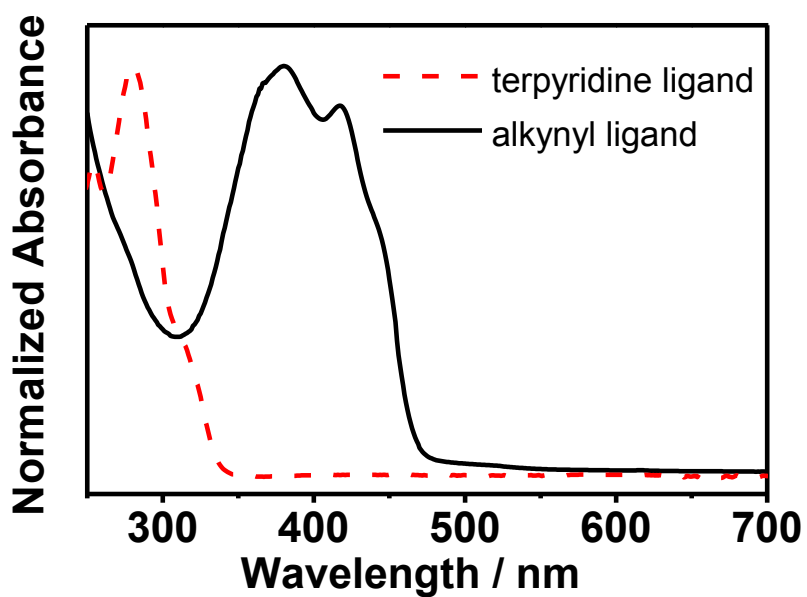


Figure S1. Normalized UV-Vis absorption spectra of the terpyridine (-----) and phosphole-based bridging alkynyl ligand (——) in CH_2Cl_2 .

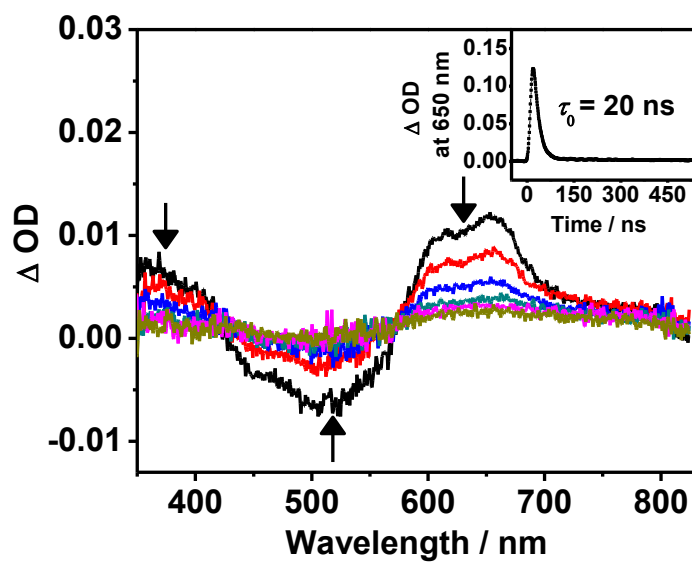


Figure S2. Transient absorption difference spectrum of **2** measured in degassed CH_2Cl_2 following a 355 nm laser pulse. The insert shows the decay trace of the absorption at 650 nm.

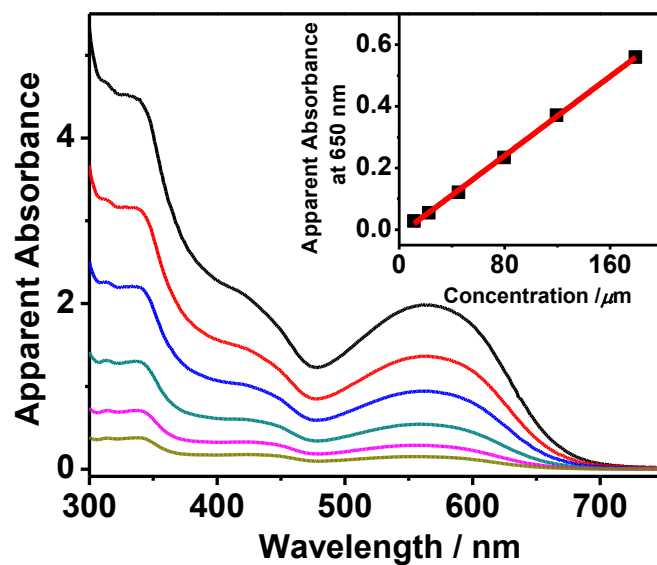


Figure S3. UV-Vis absorption spectra of **1** in CH₂Cl₂ at various concentrations. (Insert) The plot of the appearance absorbance at 650 nm at various concentrations. The apparent absorbance values have been obtained by correcting to a 1-cm path length equivalence.

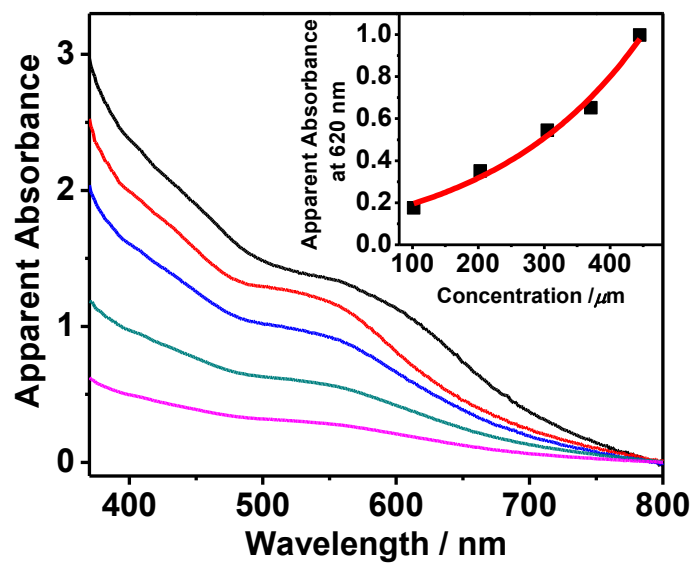


Figure S4. UV-Vis absorption spectra of **1** in DMSO at various concentrations. (Insert) The plot of the appearance absorbance at 620 nm at various concentrations. The apparent absorbance values have been obtained by correcting to a 1-cm path length equivalence.

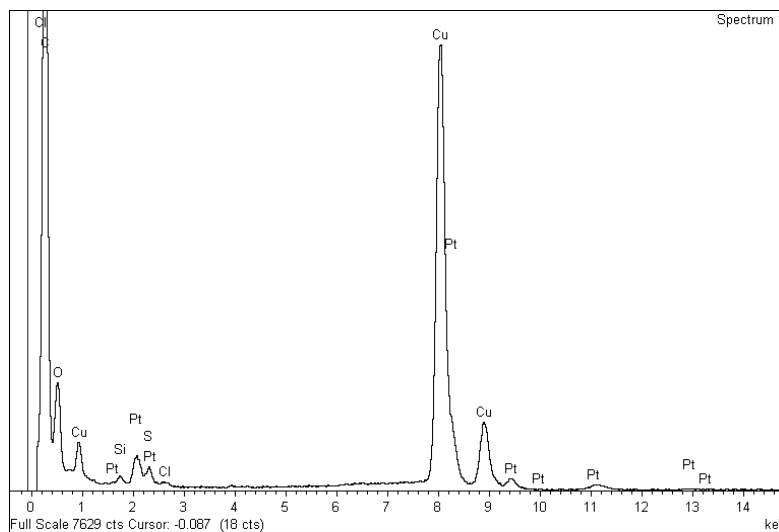


Figure S5. Energy-Dispersive X-ray (EDX) analysis on the spherical aggregates of **1**.

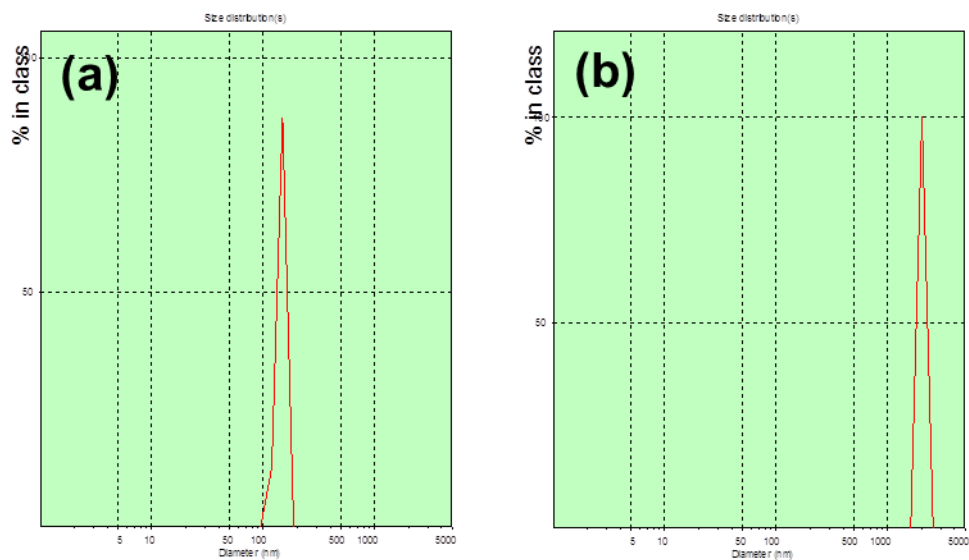


Figure S6. DLS measurement for **1** in (a) DMSO and (b) diethyl ether media.