

Supporting information

Highly active, durable and pH-universal hybrid oxide nanocrystals for efficient oxygen evolution

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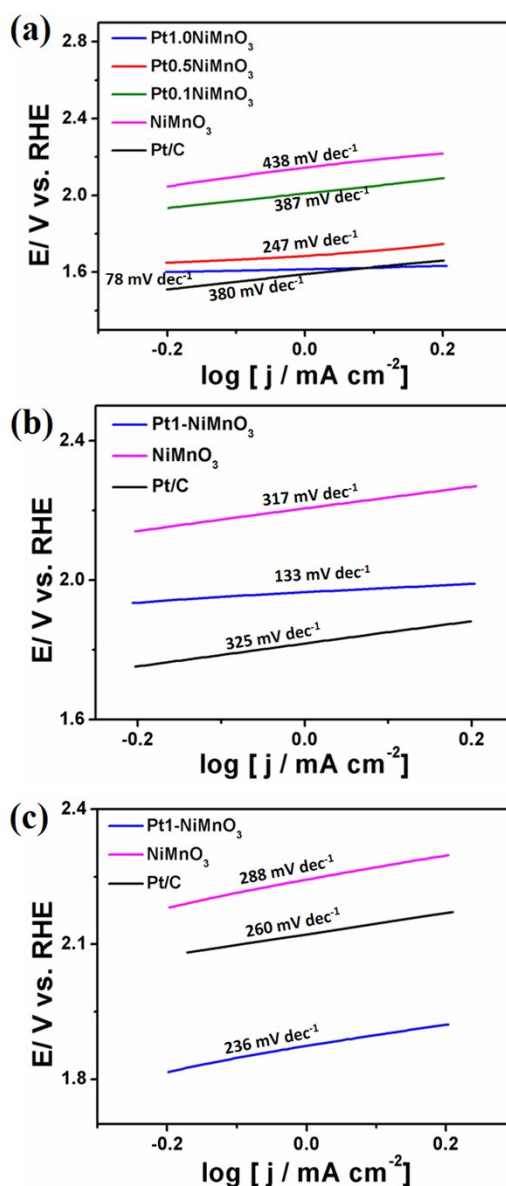


Figure S1. Tafel plot of the as-synthesized nanocatalysts relative to commercial Pt/C measured in (A) 1 M KOH, (b) PBS (pH 7.4), and (c) 0.5 H₂SO₄.

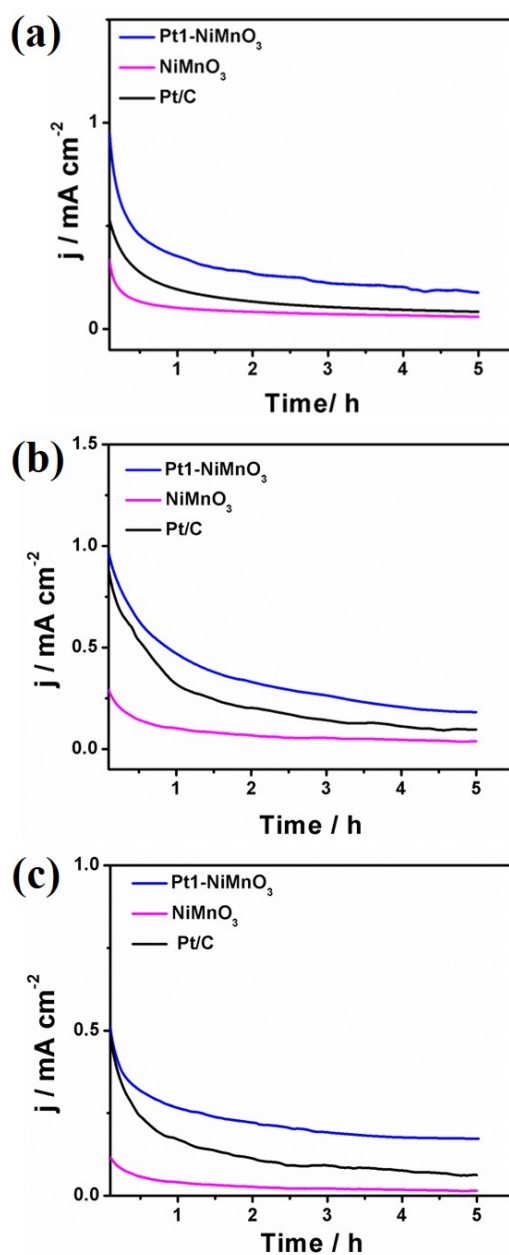


Figure S2. Chronoamperometric tests of the as-synthesized nanocatalysts relative to commercial Pt/C measured for 5 h in (A) 1 M KOH at 1.6 V, (b) PBS (pH 7.4) at 1.82 V, and (c) 0.5 H₂SO₄ at 1.8 V vs. RHE.

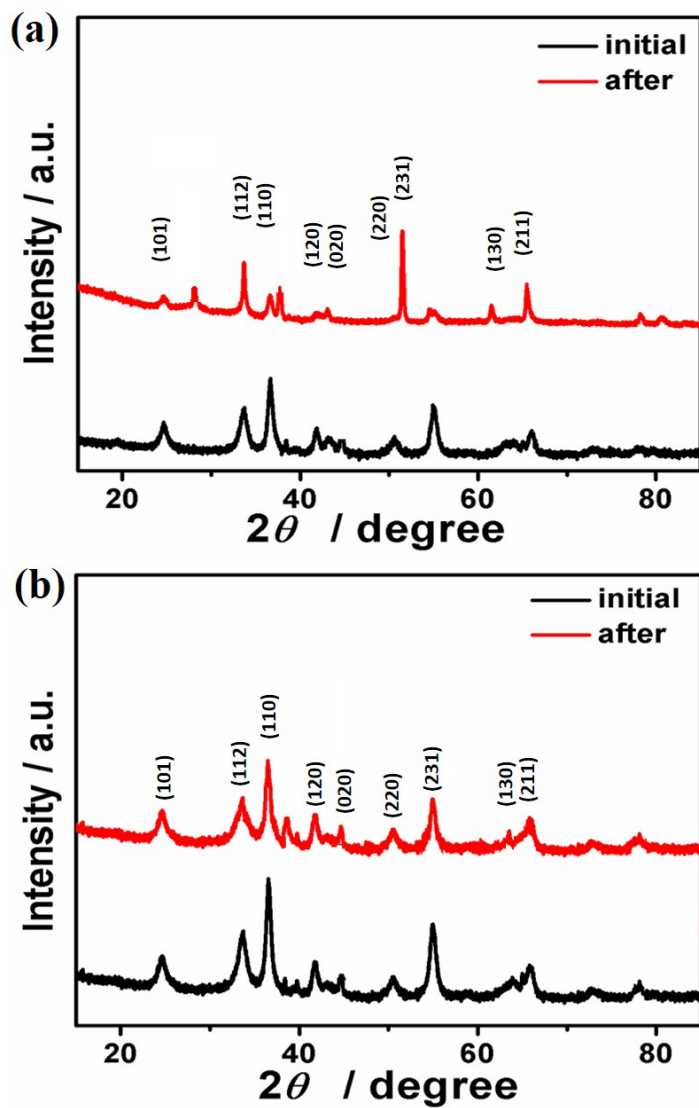


Figure S3. (a) XRD of NiMnO_3 and (b) of Pt1-NiMnO_3 before and after accelerated durability test for 5 hours in 1M KOH.

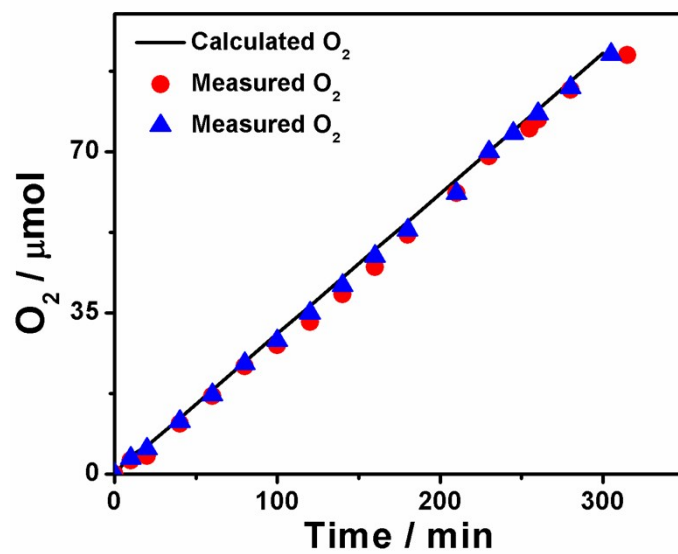


Figure S4. The amount of O₂ calculated theoretically (black-line) and measured experimentally by time on Pt1-NiMnO₃ (blue) and NiMnO₃ (red) in 1 M KOH.