

Electronic Supplementary Information

Polyaniline coated Fe₃O₄ hollow nanospheres as anode materials for lithium ion batteries

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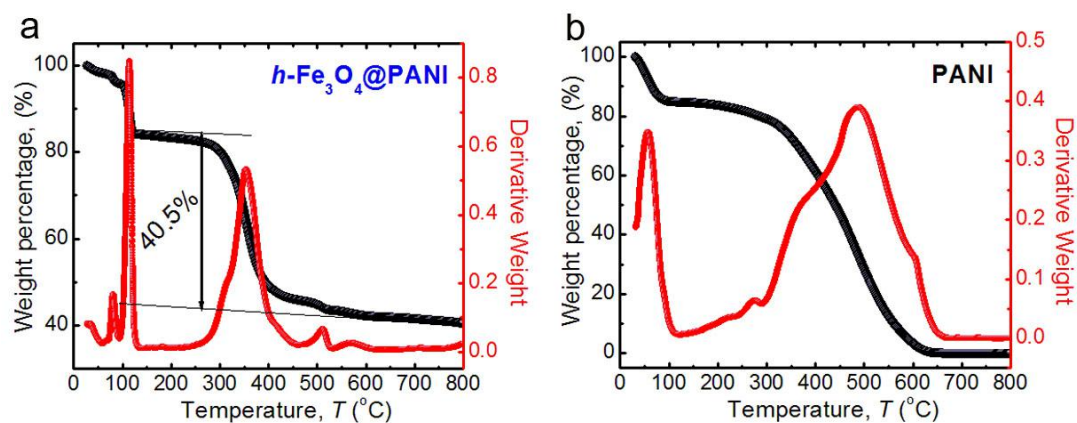


Fig. S1 Thermal gravimetric analysis (TGA, black lines) and derivative thermogravimetric (DTG, red lines) plots of (a) $h\text{-Fe}_3\text{O}_4\text{@PANI}$ composites and (b) the pure PANI measured by using TG 2050 thermogravimetric analyzer under an air atmosphere at the temperature range of 25-800 °C with a heating rate of 10 °C min⁻¹.

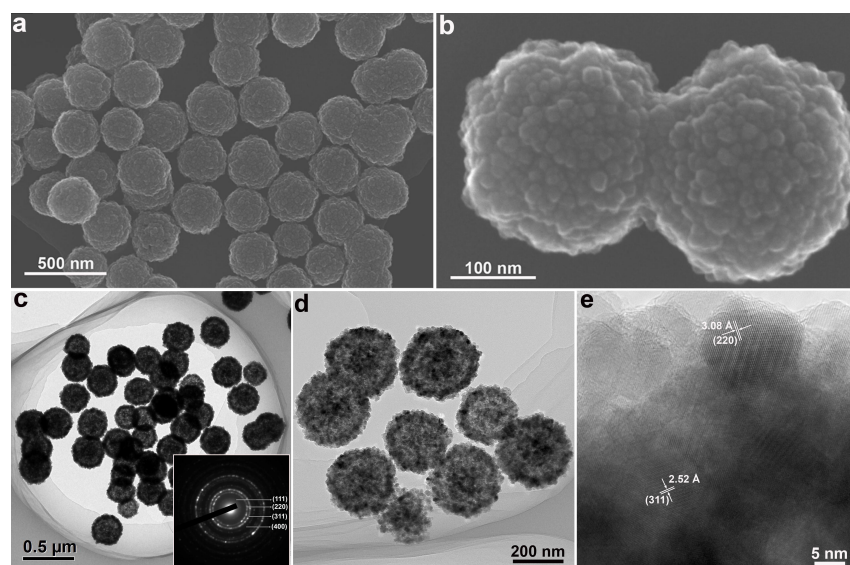


Fig. S2 Typical FESEM, TEM and HRTEM images of the h -Fe₃O₄ sample.

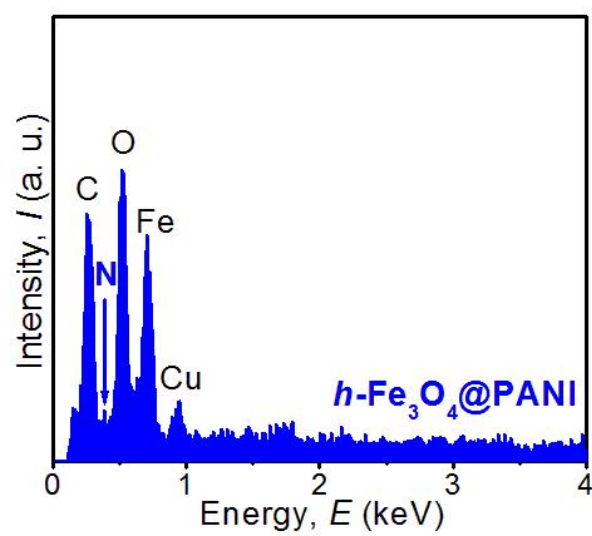


Fig. S3 EDX pattern of the $h\text{-Fe}_3\text{O}_4\text{@PANI}$ sample.

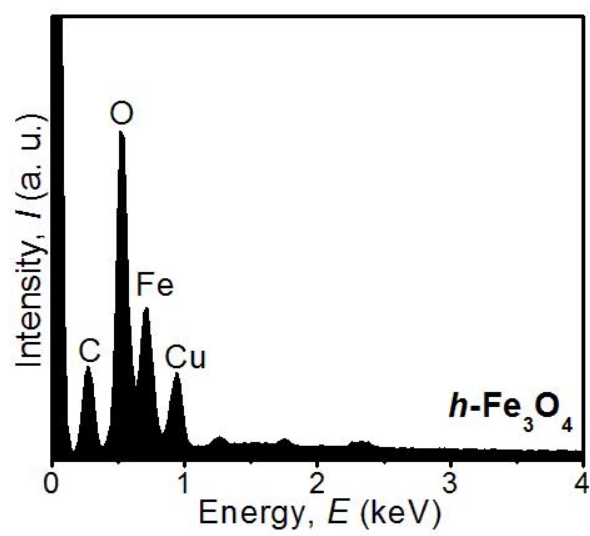


Fig. S4 EDX pattern of the $h\text{-Fe}_3\text{O}_4$ sample.

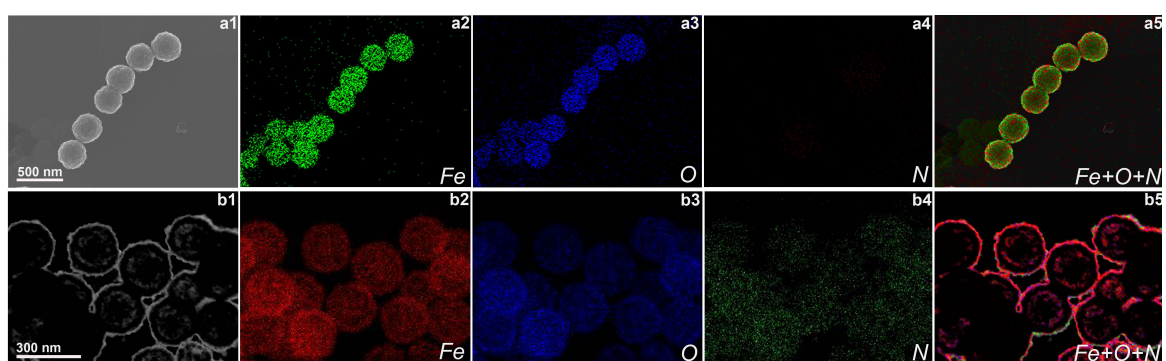


Fig. S5 The elemental maps of Fe, O and N for (a) $h\text{-Fe}_3\text{O}_4$ and (b) $h\text{-Fe}_3\text{O}_4\text{@PANI}$ samples.

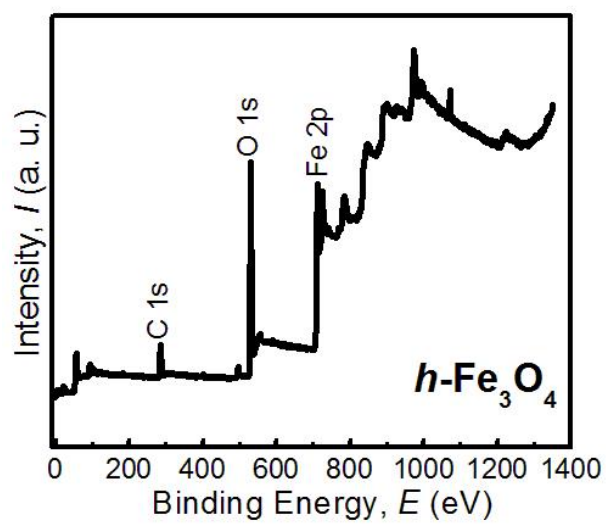


Fig. S6 XPS survey spectrum of the *h*-Fe₃O₄ sample.

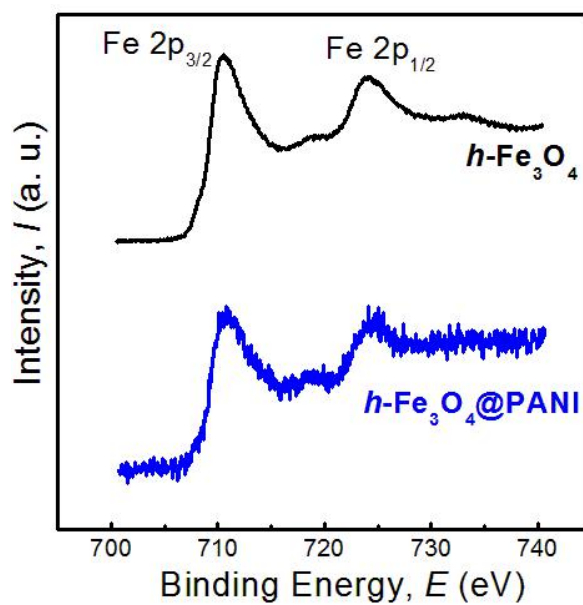


Fig. S7 (a) High-resolution XPS spectra of the Fe 2p region for the $h\text{-Fe}_3\text{O}_4$ (black) and $h\text{-Fe}_3\text{O}_4\text{@PANI}$ (blue) samples.

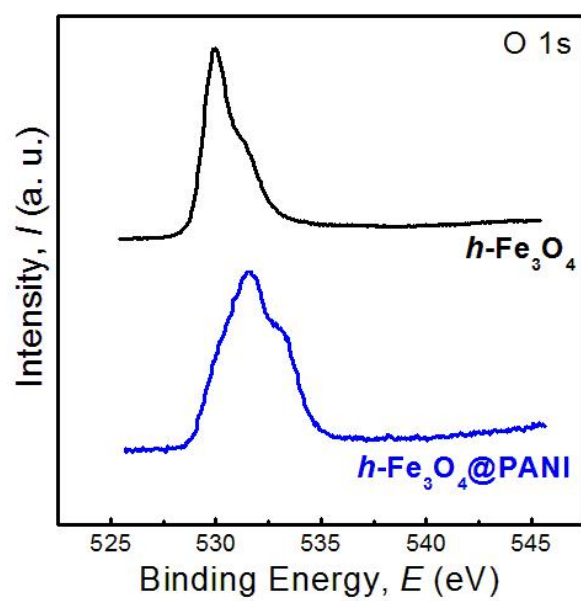


Fig. S8 High-resolution XPS spectra of the O 1s region for the $h\text{-Fe}_3\text{O}_4$ (black) and $h\text{-Fe}_3\text{O}_4\text{@PANI}$ (blue) samples.

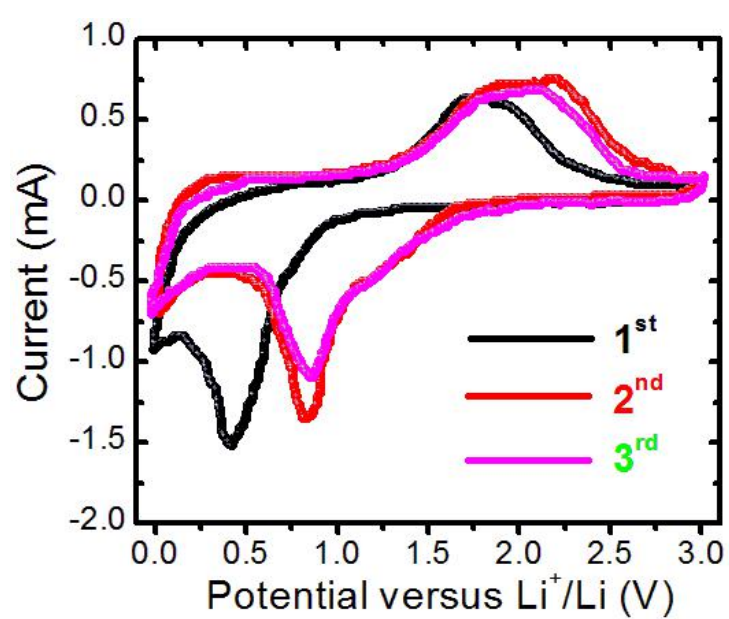


Fig. S9 CVs of *h*-Fe₃O₄@PANI sample at a scan rate of 0.5 mVs⁻¹ between 0.05 and 3 V.

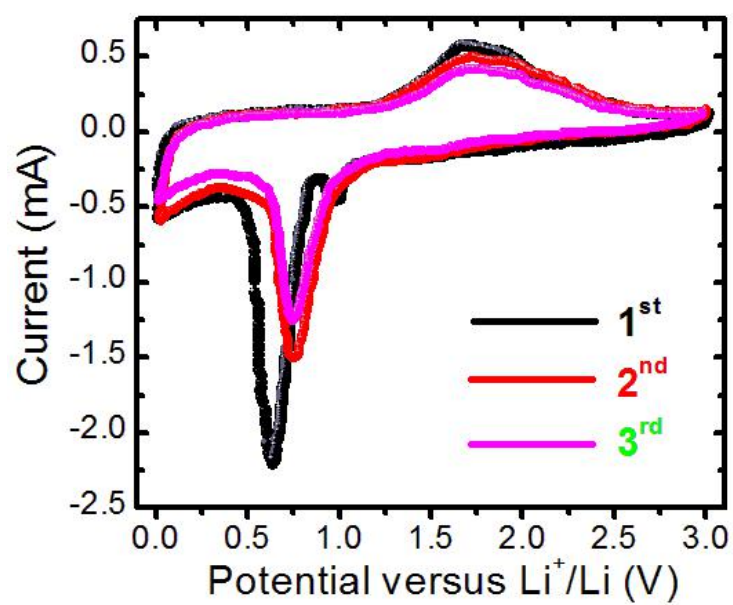


Fig. S10 CVs of $h\text{-Fe}_3\text{O}_4$ sample at a scan rate of 0.5 mVs^{-1} between 0.05 and 3 V.

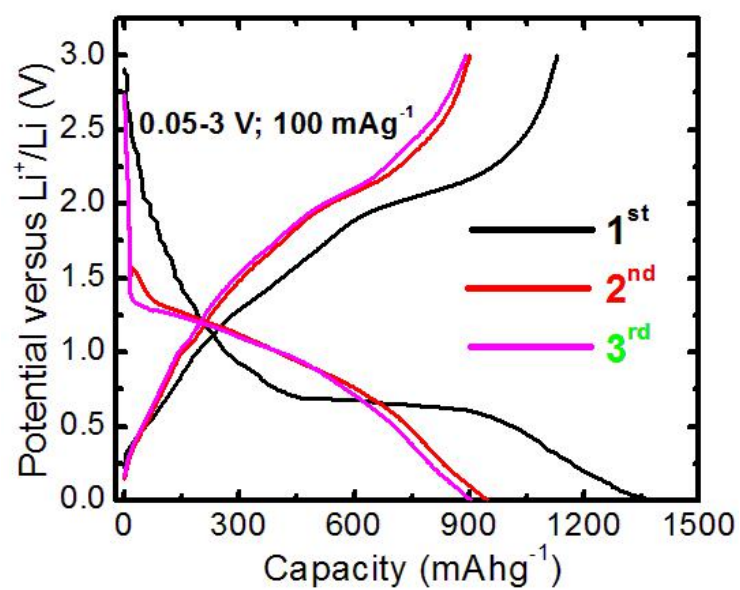


Fig. S11 Galvanostatic charge/discharge curves of *h*-Fe₃O₄ for the first three cycles between 0.05 V and 3.00 V (vs. Li⁺/Li) at a current density of 100 mA g⁻¹.

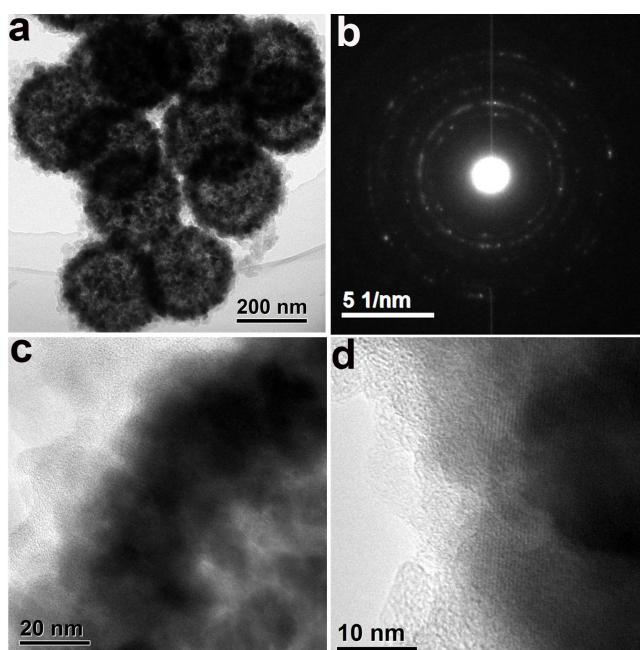


Fig. S12 (a) TEM image, (b) SAED pattern and (c, d) HRTEM images of the $h\text{-Fe}_3\text{O}_4\text{@PANI}$ electrode after cycling performance testing (50 cycles, current rate 0.1 C, 0.01-3 V versus Li/Li^+).

Table S1 Performance comparison of some LIB anode materials based on typical Fe₃O₄ structures (1 C = 1000 mA g⁻¹)

	Voltage Window /V (vs. Li ⁺ /Li)	Reversible Capacity(cycles) /mAh g ⁻¹	Rate Capability /mAh g ⁻¹	Ref
Fe ₃ O ₄ hollow microspheres	0.05 – 3.0	580@0.2 C (100)	40 @5C	1
Hollow ball-in-ball Co _x Fe _{3-x} O ₄	0.005 – 3.0	650.2@1C (100)	201.6 @10C	2
Fe ₃ O ₄ microspheres	0.01 – 3.0	450 @0.2C (110)	—	3
Fe ₃ O ₄ nanocubes	0.001 – 3.0	221.9@0.2 C (60)	51 @5C	4
Fe ₃ O ₄ /Fe nanocomposites	0.05 – 3.0	390@0.2C (50)	260 @2C	5
porous Fe ₃ O ₄ thin films	0.01 – 3.0	366@0.1C (100)	120 @5C	6
mesoporous Fe ₃ O ₄ nanorods	0.05 – 3.0	825.4 @0.1C (50)	715.7 @1C	7
TiO ₂ and Fe ₃ O ₄ with graphene	0.01 – 3.0	703@0.5 C (200)	169 @8C	8
Fe ₃ O ₄ @C/CNT nanostructures	0.01 – 3.0	693@0.3 C (200)	282 @1.2C	9
Fe ₃ O ₄ /graphene sheet composites	0.01– 3.0	1134 @0.1C	502 @5C	10
Fe ₃ O ₄ on aligned carbon nanotubes	0.1 – 3.0	1670 @0.1C	340 @9C	11
Fe ₃ O ₄ /Fe/Carbon composite	0.002 – 3.0	685 @0.05C	500 @0.5C	12
Fe ₃ O ₄ @polypyrrole nanocages	0.01– 3.0	950 @0.2C	490 @5C	13
Porous Fe ₃ O ₄ /C microbelts	0.01 – 3.5	710@0.1 C (50)	184 @3C	14
Carbon-encapsulated Fe ₃ O ₄ nanoparticles	0.005 – 3.0	998@1 C (100)	576 @10C	15
Yolk-shelled Fe ₃ O ₄ @carbon	0.01 – 3.0	1012@0.1 C (70)	900 @5C	16
Fe ₃ O ₄ -carbon nanocomposites	0.005 – 3.0	1409@0.2C (100)	414 @5C	17
Fe ₃ O ₄ nanoparticles in porous carbon	0.01– 3.0	702@0.35C (50)	—	18
Nitrogen-doped carbon encapsulated Fe ₃ O ₄	0.005 – 3.0	848@0.1C (50)	360 @2C	19
Fe ₃ O ₄ /C composite beads	0.05 – 3.0	698.8@0.1C (50)	573.1 @0.5C	20
Fe ₃ O ₄ @carbon nanorods	0.01 – 3.0	808.2@1C (100)	—	21
Hierarchical hollow Fe ₃ O ₄ microspheres	0.01– 3.0	851.9@1C (50)	654.5@5C	22
Hollow Fe ₃ O ₄ /C spheres	0.01– 3.0	984@0.2C (70)	460 @5C	23
Fe ₃ O ₄ @C core-shell nanorings	0.01– 3.0	923@0.2C (160)	632 @1C	24
C/ Fe ₃ O ₄ /C core-shell nanotubes	0.01– 3.0	700@0.1C (120)	—	25

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