

Supporting Information

Highly active and stable $\text{Er}_{0.4}\text{Bi}_{1.6}\text{O}_3$ decorated $\text{La}_{0.76}\text{Sr}_{0.19}\text{MnO}_{3+\delta}$ nanostructured oxygen electrodes of reversible solid oxide cells

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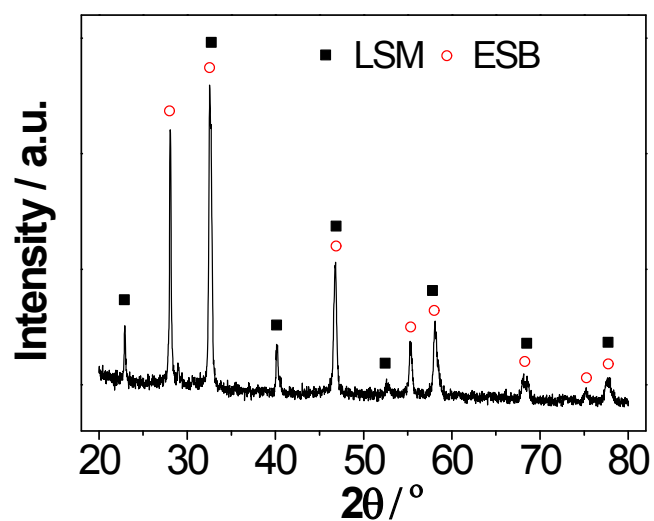


Fig. S1 XRD patterns of ESB decorated LSM electrode after heat-treatment under open circuit at 750 °C for 110 h.

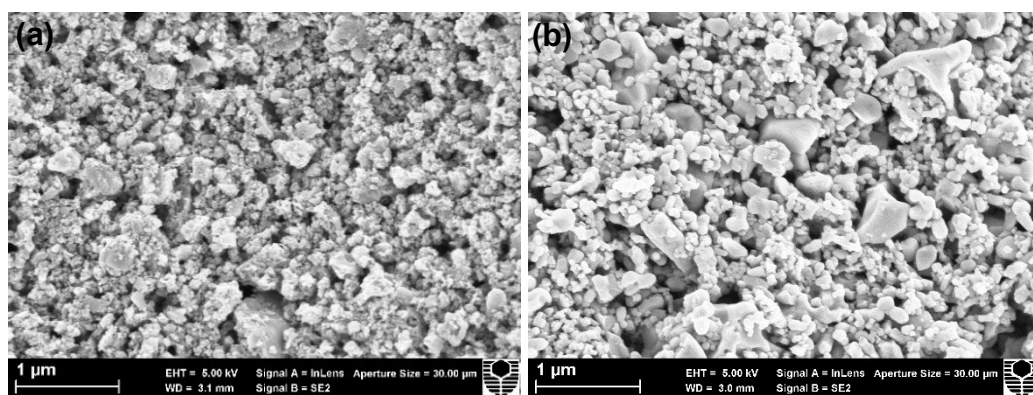


Fig. S2 SEM micrographs of directly assembled (a) decorated ESB-LSM and (b) mixed ESB-LSM composite electrodes.

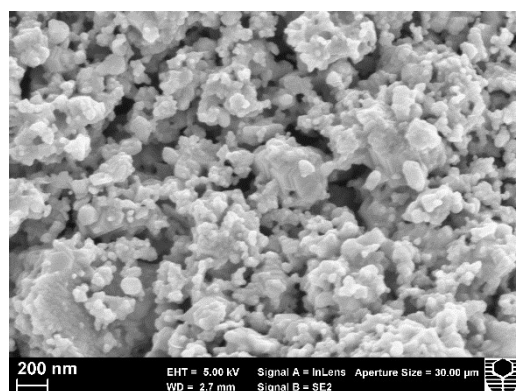


Fig. S3 SEM micrograph of electrode surface of ESB decorated LSM after heat-treatment under open circuit at 750 °C for 110 h.

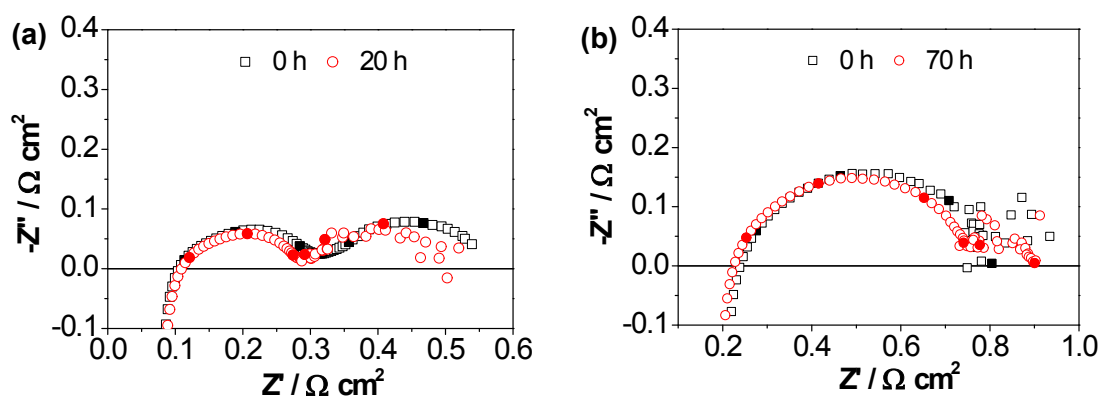


Fig. S4 Cell impedance spectra with a directly assembled ESB decorated LSM composite electrode before and after polarization at 500 mA cm⁻²: (a) 750 °C and (b) 650 °C.

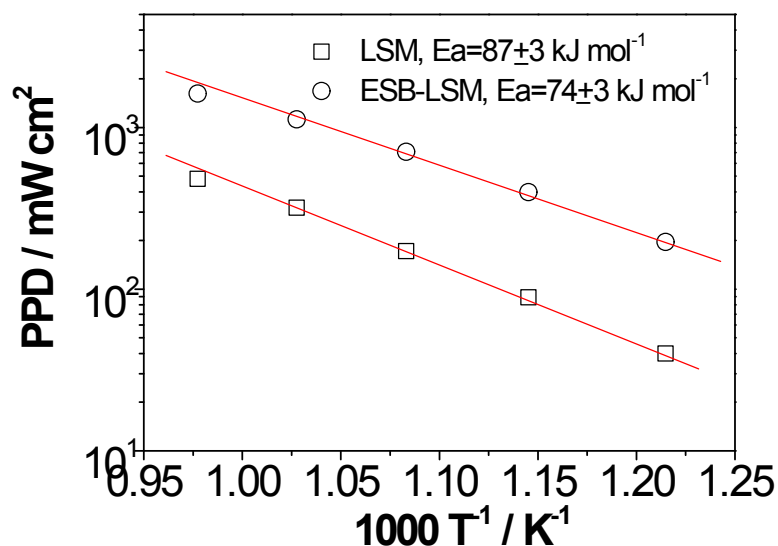


Fig. S5 Arrhenius plot of peak power densities (PPD) of the cells with directly assembled LSM and ESB decorated LSM electrodes.

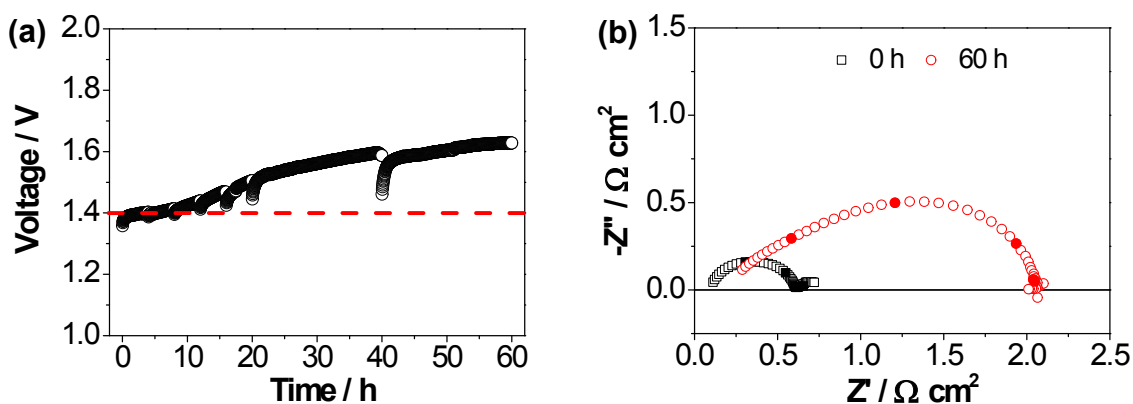


Fig. S6 Time dependent voltage curves (a) and impedance spectra (b) of the cell with a directly assembled LSM-YSZ (6/4, w/w) composite electrode at a constant electrolysis current of 500 mA cm^{-2} and 750°C . The hydrogen electrode was provided with $25 \text{ ml min}^{-1} \text{ H}_2 + 25 \text{ ml min}^{-1} \text{ CO}_2$. Prior to the stability test under the electrolysis mode, the cell was polarized under fuel cell mode 500 mA cm^{-2} and 750°C for 60 h.

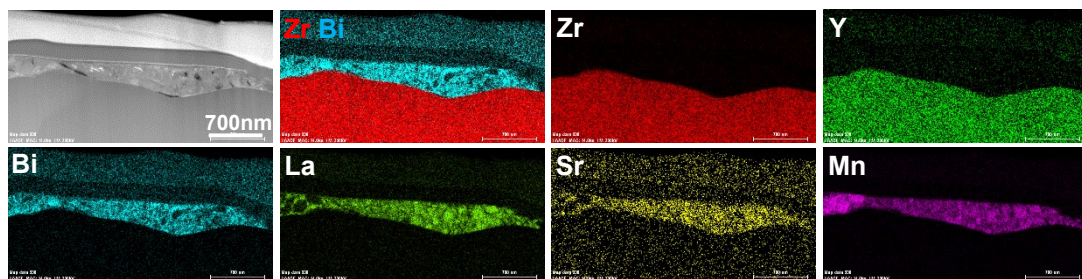


Fig. S7. HAADF micrograph and EDS maps on the electrode/electrolyte interface after the electrochemical stability test. The ESB decorated LSM electrode was peeled off using an adhesive tape approach.