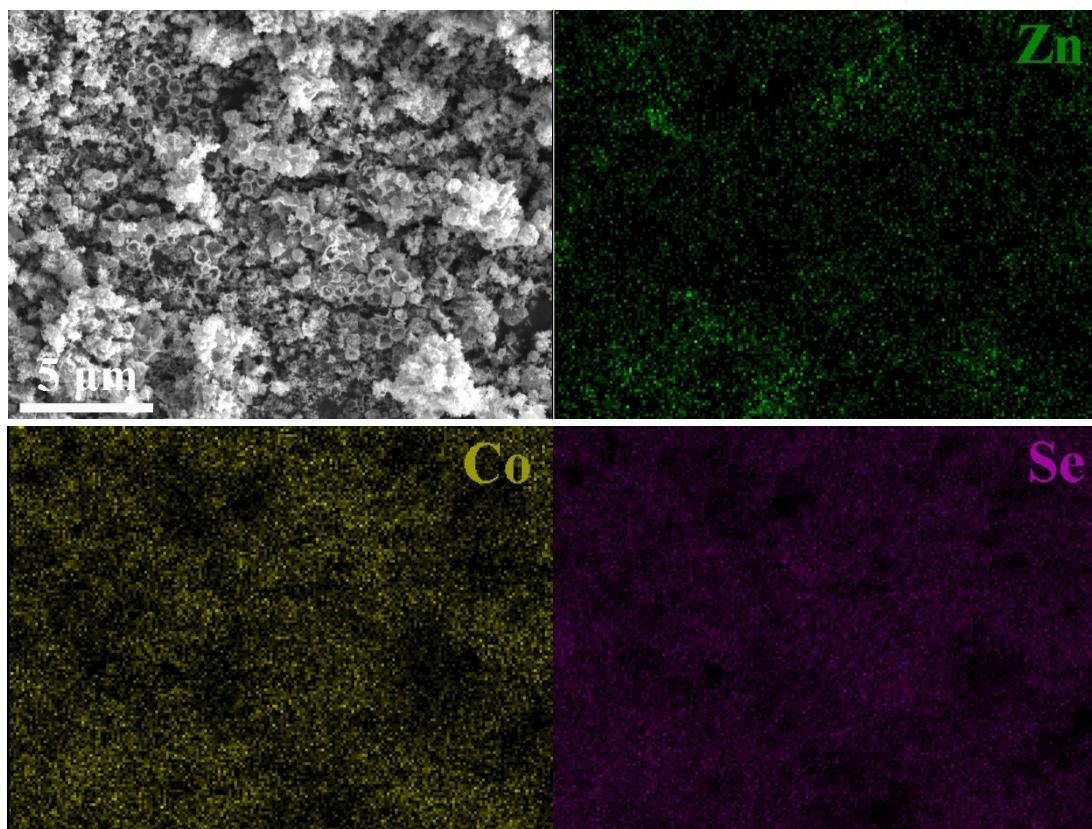


Supporting Information

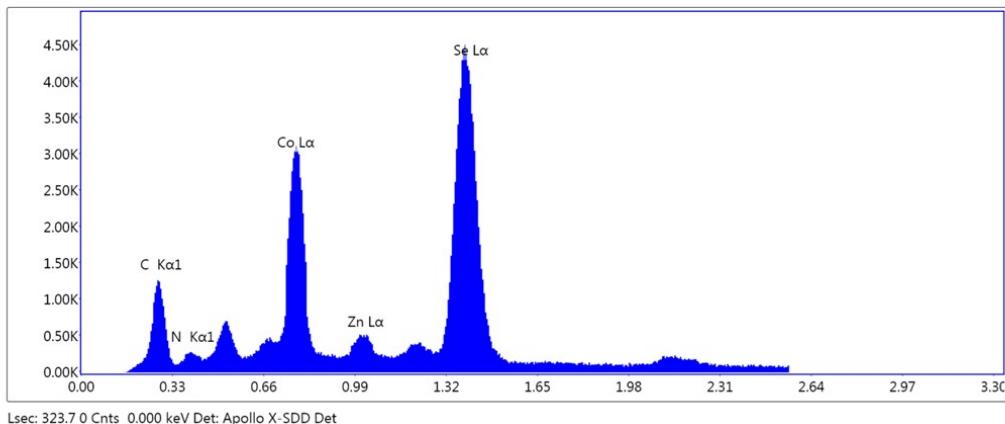
**Hollow Bimetallic Cobalt-based Selenide Polyhedrons Derived from Metal–organic Framework: An Efficient Bifunctional Electrocatalyst for Overall Water Splitting**

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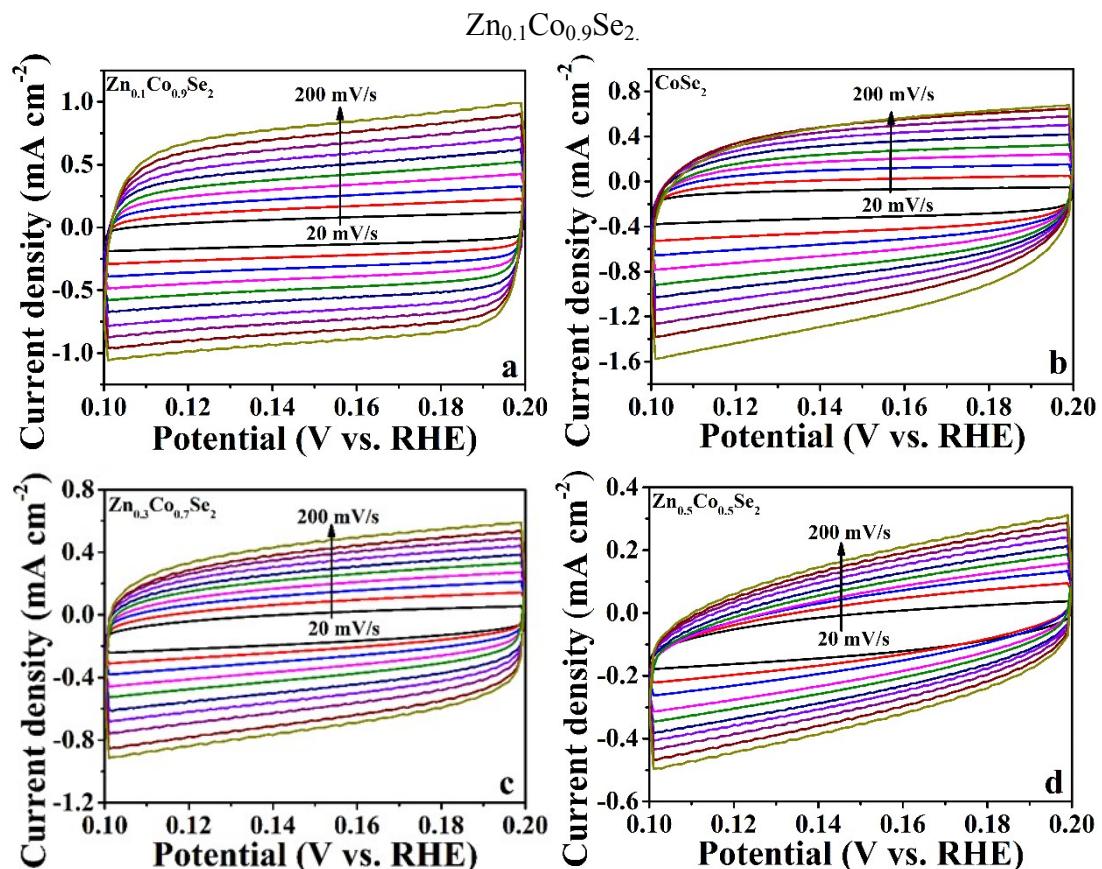


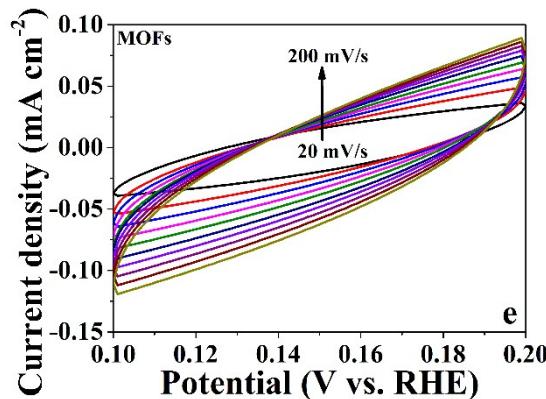
**Fig. S1** SEM image corresponding EDX mapping of  $\text{Zn}_{0.1}\text{Co}_{0.9}\text{Se}_2$



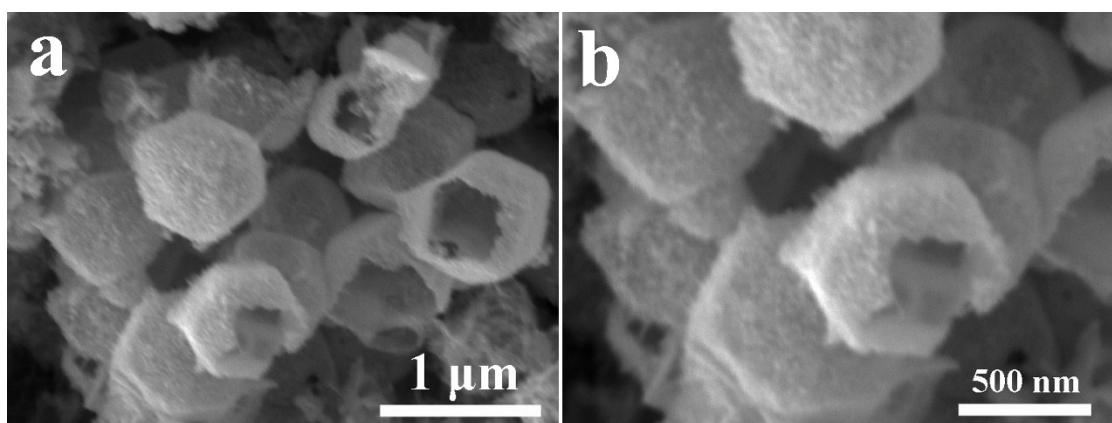
Element	Weight %	Atomic %	Net Int.	Error %	Kratio	Z	R	A
C-K	15.65	48.39	43.2	10.94	0.08	1.47	0.86	0.36
N-K	3.61	9.57	8.6	13.11	0.02	1.43	0.87	0.44
Co-L	23.89	15.06	72.7	5.02	0.22	0.96	1.02	0.97
Zn-L	2.47	1.40	7.4	10.61	0.02	0.94	1.03	0.91
Se-L	54.39	25.59	164.2	5.70	0.45	0.86	1.05	0.95

**Fig. S2** SEM image corresponding EDS spot and the eZAF Smart Quant Results of

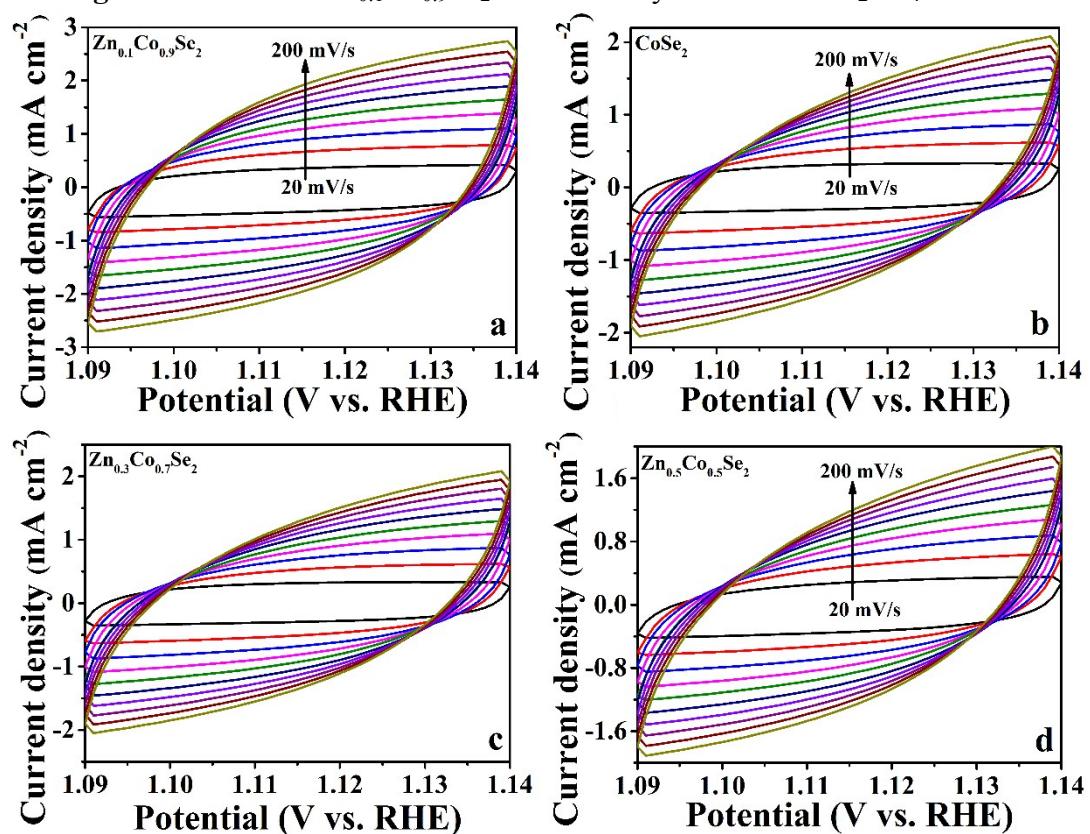


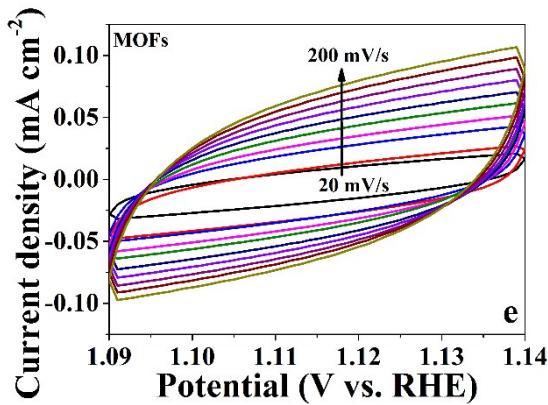


**Fig. S3** Voltammograms of the (a)  $\text{Zn}_{0.1}\text{Co}_{0.9}\text{Se}_2$  (b)  $\text{CoSe}_2$  (c)  $\text{Zn}_{0.3}\text{Co}_{0.7}\text{Se}_2$  (d)  $\text{Zn}_{0.5}\text{Co}_{0.5}\text{Se}_2$  (e) MOFs at various scan rate (20-200  $\text{mV s}^{-1}$ ) in 0.5 M  $\text{H}_2\text{SO}_4$  solution.

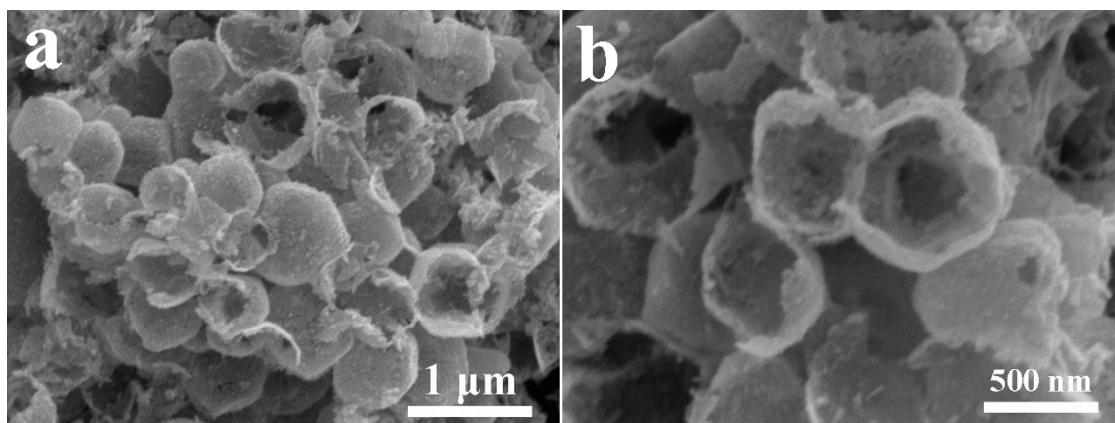


**Fig. S4** the SEM of  $\text{Zn}_{0.1}\text{Co}_{0.9}\text{Se}_2$  after 5000 cycles in 0.5M  $\text{H}_2\text{SO}_4$  solution.

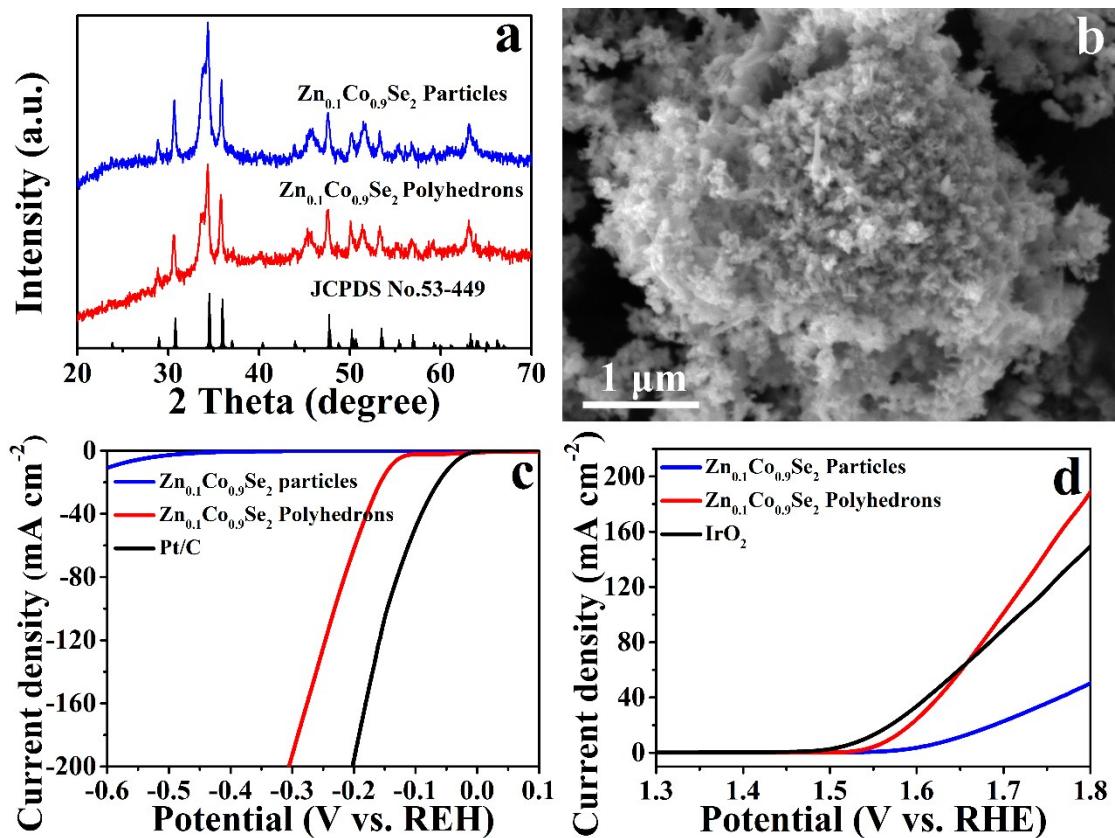




**Fig. S5** Voltammograms of the (a)  $\text{Zn}_{0.1}\text{Co}_{0.9}\text{Se}_2$  (b)  $\text{CoSe}_2$  (c)  $\text{Zn}_{0.3}\text{Co}_{0.7}\text{Se}_2$  (d)  $\text{Zn}_{0.5}\text{Co}_{0.5}\text{Se}_2$  (e) MOFs at various scan rate ( $20\text{-}200 \text{ mV s}^{-1}$ ) in 1.0 M KOH solution.



**Fig. S6** the SEM of  $\text{Zn}_{0.1}\text{Co}_{0.9}\text{Se}_2$  after 5000 cycles in 1.0 M KOH solution.



**Fig. S7** the XRD patterns (a), SEM (b) and the polarization curves for HER in 0.5M H<sub>2</sub>SO<sub>4</sub> solution (c) and OER in 1.0M KOH solution (d) of Zn<sub>0.1</sub>Co<sub>0.9</sub>Se<sub>2</sub> Particles.

**Table S1.** Element composition of various M<sub>x</sub>Co<sub>1-x</sub>Se<sub>2</sub> (a represent mass ratio of M/Co detected by ICP-AES and b represent the molar ratio after converting).

Samples	[Mass(M:Co)] <sup>a</sup>	[Molar(M:Co)] <sup>b</sup>
Zn <sub>0.1</sub> Co <sub>0.9</sub> Se <sub>2</sub>	0.1163	0.1048
Zn <sub>0.3</sub> Co <sub>0.7</sub> Se <sub>2</sub>	0.4588	0.4136
Zn <sub>0.5</sub> Co <sub>0.5</sub> Se <sub>2</sub>	0.8967	0.8084
Ni <sub>0.1</sub> Co <sub>0.9</sub> Se <sub>2</sub>	0.1128	0.1133
Cu <sub>0.1</sub> Co <sub>0.9</sub> Se <sub>2</sub>	0.1183	0.1097

**Table S2.** XPS results of Zn<sub>0.1</sub>Co<sub>0.3</sub>Se<sub>2</sub> on the atomic percentages of C, N, O, Zn, Co and the Se distributions.

element	C	N	O	Zn	Co	Se
percentage	79.76	1.88	7.87	0.31	2.51	7.67

**Table S3** The atomic percentage distributions of C, N, O, Zn, Co and Se in Zn<sub>0.1</sub>Co<sub>0.9</sub>Se<sub>2</sub> (C, N and O detected by Elemental Analyzer, Zn, Co and Se detected by ICP-AES)

element	C	N	O	Zn	Co	Se
atom percent	66.97	3.47	3.96	0.76	7.27	17.57

**Table S4.** Comparison of the HER activity of some reported transition metal selenide electrocatalysts in 0.5 M H<sub>2</sub>SO<sub>4</sub>.

Catalysts	Mass Loading (mg cm <sup>-2</sup> )	$\eta^{10}$ (mV vs. RHE)	Tafel slope (mV dec <sup>-1</sup> )	Reference
Zn <sub>0.1</sub> Co <sub>0.9</sub> Se <sub>2</sub>	0.285	140	49.9	<i>This work</i>
CoSe <sub>2</sub> @DC	0.357	132	82.0	<i>Nano Energy</i> <b>28</b> (2016) 143–150
CoSe <sub>2</sub>	1.0	160	40	<i>ACS Appl. Mater. Interfaces.</i>
NiSe <sub>2</sub>		190	44	<i>2016, 8, 5327-5334</i>
Fe <sub>0.7</sub> Co <sub>0.3</sub> Se <sub>2</sub> /RGO-12	1.1	166	36	<i>ACS Appl. Mater. Interfaces</i> <b>2016, 8, 18036-18042</b>
CoSe <sub>2</sub> NPs @ carbon fiber	3.0	137	42.1	<i>J. Am. Chem. Soc.,</i> <b>2014, 136, 4897</b>
CoSe <sub>2</sub> films@ Ti foil	2.8	135	62.0	<i>J. Mater. Chem. A,</i> <b>2014, 2, 13835</b>
CoSe <sub>2</sub> necklace-like NWs@CFP	2.8	165	34.0	<i>J. Mater. Chem. A,</i> <b>2015, 3, 9415</b>
Polymorphic CoSe <sub>2</sub>		150	31.2	<i>ACS Appl. Mater. Interfaces</i> <b>2015, 7, 1772</b>
Phase Pure CoSe <sub>2</sub>		272	61.0	<i>ACS Energy Lett.</i> <b>2016, 1, 607–61</b>
CoSe <sub>2</sub> NWs	1.3	130	32	<i>ACS Appl. Mater. Interfaces</i> <b>2015, 7, 3877-3881</b>
CoSe <sub>2</sub>		170	42.4	<i>Energy Environ. Sci.</i>
NiSe <sub>2</sub>	0.150	180	56.9	<i>2013, 6, 3553-3558</i>
CoSe <sub>2</sub> /C NPs		70	40.2	
MoS <sub>2</sub> /CoSe <sub>2</sub>	1.0	70	36.0	<i>Nat. Commun.</i> <b>2015, 6,</b>
CoSe <sub>2</sub> NSs		120	48.0	<b>5982-5989</b>
NiSe <sub>2</sub> NWs		170	31.1	<i>Nanoscale</i> <b>2015, 17, 14813-</b>
CoSe <sub>2</sub> NWs	1.0	196	39.6	<b>14816</b>
NiSe <sub>2</sub> NSs on Graphite disk		173	31	<i>ACS Catal.</i> <b>2015, 5, 6355-6361</b>

**Table S5.** Comparison of the OER activity of some reported transition metal selenide electrocatalysts in 1.0 M KOH.

Catalysts	Mass Loading (mg cm <sup>-2</sup> )	$\eta^{10}$ (mV vs. RHE)	Tafel slope (mV dec <sup>-1</sup> )	Reference
Zn <sub>0.1</sub> Co <sub>0.9</sub> Se <sub>2</sub>	0.285	340	43.2	<i>This work</i>
Zn-doped CoSe <sub>2</sub> /CFC		356	88	<i>ACS Appl. Mater. Interfaces</i> , 2016, 8, 26902-26907
CoSe <sub>2</sub> /CF	2.9	297	41	<i>Nano Research</i> , 2016, 9, 2234-2243
Co <sub>7</sub> Se <sub>8</sub>		260	32.6	<i>ACS Appl. Mater. Interfaces</i> , 2016, 8, 17292-17302
EG/CoSe-NiFe-LDH	4.0	250	57	<i>Energy Environ. Sci.</i> 2016, 9, 478-483
coral-like CoSe <sub>2</sub>		290	40	<i>ACS Catal.</i> 2015, 5, 3625-3637.
CoSe <sub>2</sub> /Ti mesh		292	69	<i>Chem. Commun.</i> 2015, 51, 16683-16686
Ni <sub>3</sub> Se <sub>2</sub>		290	82	<i>Catal. Sci. Technol.</i> 2015, 5, 4954-4958
Ni <sub>3</sub> Se <sub>2</sub> /CF		300	80	<i>ACS Appl. Mater. Interfaces</i> 2016, 8, 4718-4723
NiSe NWs	2.8	400	64	<i>Angew. Chem. Int. Ed.</i> 2015, 54, 9351-9355
2D FeSe <sub>2</sub>	0.01	330	48.1	<i>Nano Energy</i> 31 (2017) 90-95
CoSe <sub>2</sub> NiSe <sub>2</sub>	1.0	430 250	50 38	<i>ACS Appl. Mater. Interfaces.</i> 2016, 8, 5327-5334
NiSe <sub>2</sub>	1.4	323	83.6	<i>ACS Appl. Mater. Interfaces</i> 2016, 8, 19386-19392