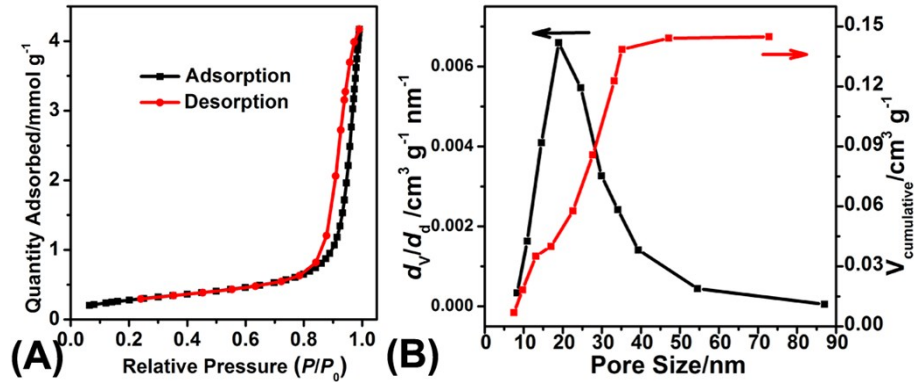


## Supporting Information

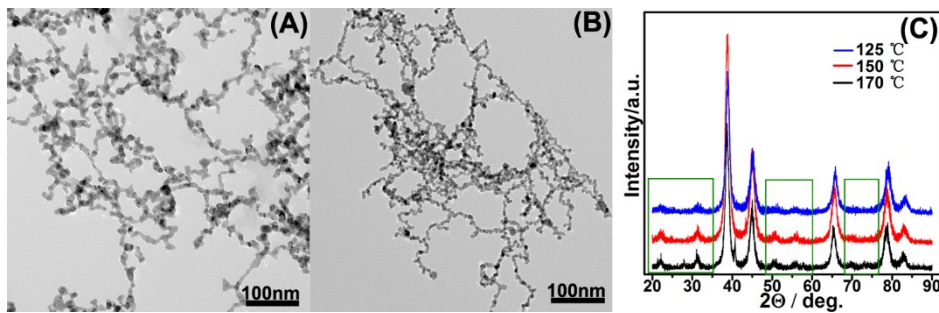
### **Intermetallic Pd<sub>3</sub>Pb Nanowire Networks Boost Ethanol Oxidation and Oxygen Reduction Reaction with Significantly Improved Methanol Tolerance**

Qirong Shi, Chengzhou Zhu,\* Cuixia Bi, Haibing Xia, Mark H. Engelhard, Dan Du, Yuehe Lin\*

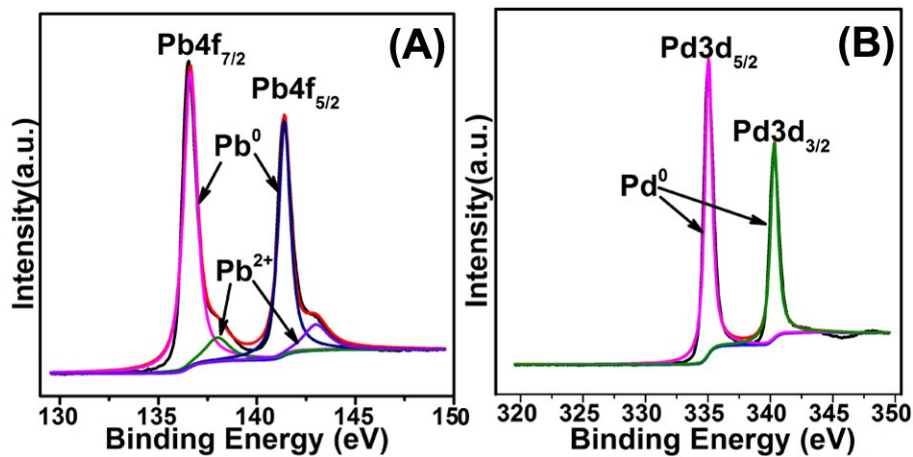
## Supporting Figures



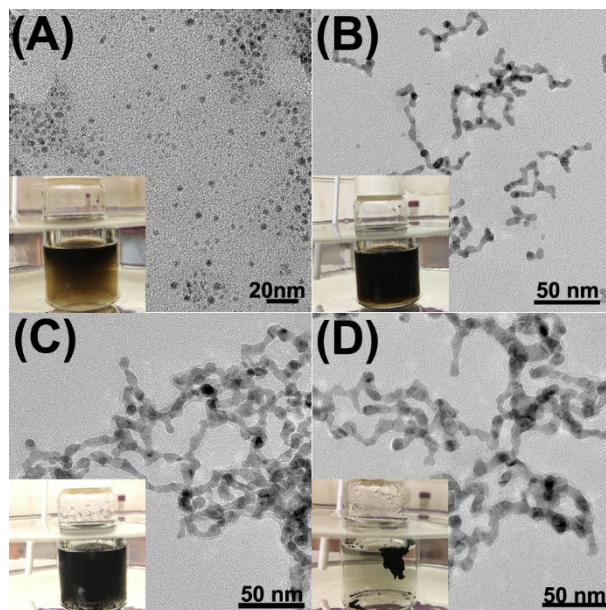
**Figure S1.** (A) Nitrogen physisorption isotherm, and (B) pore size distributions (black curve) and cumulative pore volumes ( $V_{\text{cumulative}}$ , red curve) of as-fabricated IM-Pd<sub>3</sub>Pb NNs.



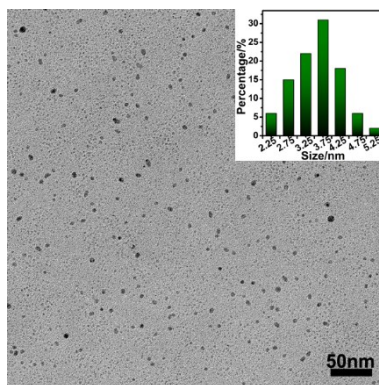
**Figure S2.** TEM images and size distribution (insets) of Pd<sub>3</sub>Pb NNs synthesized at 125 °C (A) and 150 °C (B), respectively. XRD pattern of Pd<sub>3</sub>Pb NNs obtained at 125 °C, 150 °C and 170 °C, respectively.



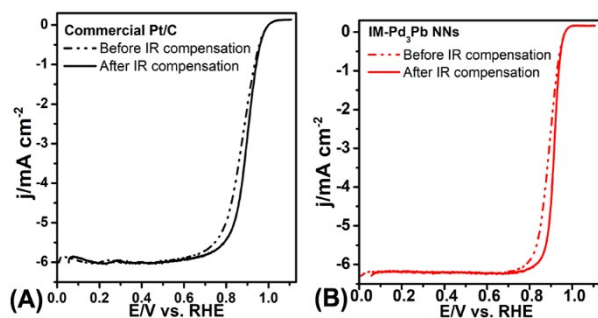
**Figure S3.** XPS of the (A) Pb 4f and (B) Pd 3d peaks of IM-Pd<sub>3</sub>Pb NNs.



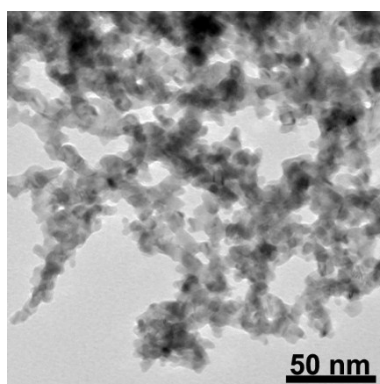
**Figure S4.** Time-dependent TEM images of the IM-Pd<sub>3</sub>Pb NNs growth procedures tracked at 2 min (A), 5 min (B), 15 mins (C), and 30 min (D) after reaction begins.



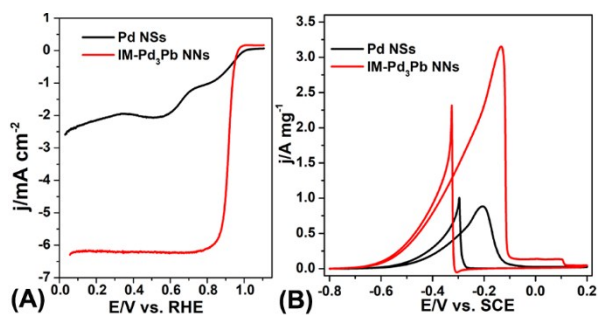
**Figure S5.** TEM image and size distribution (inset) of as-synthesized Pd NCs under the same condition with IM-Pd<sub>3</sub>Pb NNs only except the absence of Pb(acac)<sub>2</sub>.



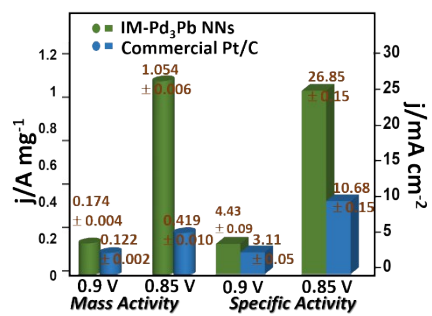
**Figure S6.** LSV of commercial Pt/C (A) and IM-Pd<sub>3</sub>Pb NNs (B) before and after 100 % IR compensation.



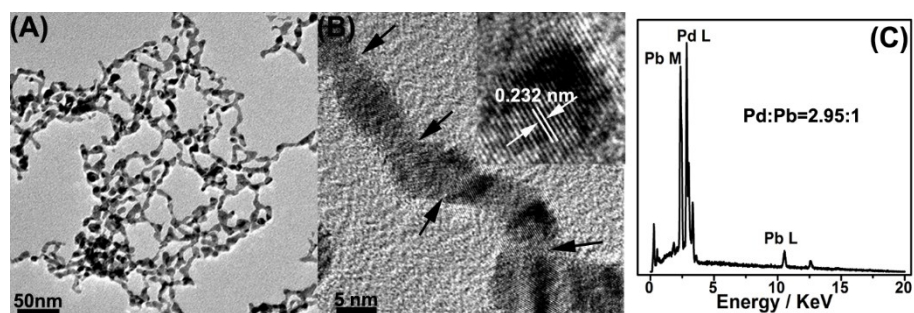
**Figure S7.** TEM image of as-synthesized Pd NSs.



**Figure S8.** (A) LSV curves of Pd NSs and IM-Pd<sub>3</sub>Pb NNs obtained in oxygen saturated 0.1 M KOH aqueous solution. (B) CV curves of Pd NSs and IM-Pd<sub>3</sub>Pb NNs in 1 M KOH aqueous solution containing 1 M ethanol.



**Figure S9.** MA and SA of commercial Pt/C and IM-Pd<sub>3</sub>Pb NNs at 0.85 V and 0.9 V with error correction, respectively.



**Figure S10.** TEM image (A), HRTEM image (B) and EDS (C) of IM-Pd<sub>3</sub>Pb NNs after ADT test.

## Supporting Tables

**Table S1.** The atomic ratios of Pd/Pb under different amount of Pd and Pb precursors.

Na <sub>2</sub> PdCl <sub>4</sub>	Pb(acac) <sub>2</sub>	Pd/Pb(precursor)	Pd/Pb(EDS)
7.1 mg	19.5 mg	1/2	2.95
7.1 mg	16.2 mg	1/6	3.05
7.1 mg	13.0 mg	3/4	3.30

**Table S2.** The reaction temperature and time for formation of intermetallic phases from previous reported works.

IM NCs	Solvent	temperature	time	reference
Pd <sub>3</sub> Pb NNs	EG	170 °C	1 h	our work
Pd <sub>3</sub> Pb NPs	annealing	600 °C	24 h	1
Pd <sub>3</sub> Pb NPs	TEG	180 °C	15 min	2
Pt <sub>3</sub> Co	OAm	160 °C	8 h	3
Pd <sub>2</sub> Sn	OAm, MAHC	300 °C	3 h	4
PtBi	OAm	200 °C	1.5 h	5
PtPb/Pt	OAm+ODE	160 °C	5.5 h	6
Pt <sub>3</sub> Zn	DMF	180 °C	9 h	7
PtCu <sub>3</sub>	OAm	170 °C	24 h	8

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