The Crucial Role and an Effective Modulation Method of Intermolecular \( \pi-\pi \) Interaction of the Well-known A-D-A Type Electron Acceptor

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Experiment section

Instruments and Measurements

Bruker AVANCE 300 or 400 MHz spectrometer was used to measure the $^1$H and $^{13}$C NMR spectra of the compounds in the synthetic procedure at room temperature. Absorption spectra (in chloroform solution and as thin films) and molecular energy levels of the small molecular acceptors were measured on a Hitachi UH4150 UV-Vis spectrophotometer and a CHI650D Electrochemical Workstation, respectively. In the molecular energy level measurements, 0.1 M Bu$_4$NPF$_6$ acetonitrile solution was used as electrolyte, and ferrocene/ferrocenium was used as external standard and measured in parallel. The thickness of blend layers was measured via the surface profilometer Bruker Dektak XT. AFM and TEM images were obtained by Nanoscope V AFM and JEOL 2200FS instrument, respectively. The electron mobility was measured by SCLC method,$^1$ and the equations of $J = (9/8)\varepsilon_0\varepsilon_r\mu_e V_{app}^2/L^3$ was used to calculate the electron mobility, where $\varepsilon_0$ is the permittivity of free space, $\varepsilon_r$ is the relative permittivity of the organic material, $\mu$ is the electron mobility, $V_{app}$ is the effective applied voltage, and $L$ is the thickness of the film. Morphology Characterizations: GIWAXS measurements were conducted at the PLS-II 9A U-SAXS beam line of Pohang Accelerator Laboratory. The photo-CELIV measurements reported here were performed by the all-in-one characterization platform Paios developed and commercialized by Fluxim AG, Switzerland, the under a 0.32 V/us linearly increasing reverse bias pulse.$^2$

S1. NMR spectra of the target products
$^{13}$C NMR spectrum of IDT-C6

$^1$H NMR spectrum of IDT-PhC6
Figure S1. NMR spectra of the target products.

S2. TGA

Figure S2. TGA plots of IDT-C6 and IDT-PhC6.

S3. Films of acceptors spin coated by CB

500 nm
Figure S3. Photos of IDT-C6 and IDT-PhC6 films in chlorobenzene, captured by the camera of the surface profilometer Bruker Dektak XT.

S4. Theoretical study

![Figure S4. Optimized molecular geometries and frontier molecular orbitals for IDT-C6 and IDT-PhC6 calculated by DFT at the B3LYP/6-31G(d, p) level.](image)

S5. Optimized dimer geometries

![Figure S5. Optimized dimer geometries for IDT-C6 and IDT-PhC6.](image)

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<th>Devices</th>
<th>D:A</th>
<th>DIO (v/v)</th>
<th>$V_{oc}$ (V)</th>
<th>$J_{sc}$ (mA/cm$^2$)</th>
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Table S1. Summary of photovoltaic parameters of the PBDB-T:ITCC-based OSC devices with different D:A ratios and DIO contents.

S6. Absorption properties of acceptors.

![Absorption spectra of IDT-C6 and IDT-PhC6 films.](image1)

**Figure S6.** Absorption spectra of IDT-C6 and IDT-PhC6 films.

S7. Electron mobilities of pure acceptor films

![J-V plots of IDT-C6 and IDT-PhC6 based electron-only device.](image2)

**Figure S7.** $J^0.5-V$ plots of IDT-C6 and IDT-PhC6 based electron-only device.
S8. Charge recombination properties

**Figure S8.** a) $J_{ph}$ versus $V_{off}$ of the optimized devices. b) Photocurrent dependence on the light intensity for the IDT-C6- and IDT-PhC6-based OSC devices.

**S9. Hole mobilities of blended films**

**Figure S9.** Hole mobility plots for PBDB-TF:IDT-C6 and PBDB-TF:IDT-PhC6 devices.

**S10. Morphology of blended films**

**Figure S10.** TEM phase image of a) PBDB-TF:IDT-C6 blended films. b) PBDB-TF:IDT-PhC6 blended films.
Figure S11. Chemical structures of typical acceptors with hexyl or hexylphenyl side chains.

References
