

Supporting Information to

Title Bulbous gold-carbon nanodot hybrid nanoclusters

*Dominik H. Hasenöhrl, Avishek Saha, Volker Strauss, Leonie Wibmer, Stefanie Klein,
Dirk. M. Guldi, Andreas Hirsch*

Contents

General	S3
Absorption spectroscopy	S4
Dynamic light scattering analysis	S5
Raman spectroscopy	S6
Time-correlated single-photon counting	S7
Transient absorption characterization	S8
Transmission electron microscopy	S11
Dynamic light scattering and Fluorescence measurements in Serum	S15
Additional Synthesis attempts	S17
Radiation therapy experiments	S18

General

All chemicals were used as received from commercial sources. Average hydrodynamic radii of nanoparticles were measured with a Zetasizer Nano ZS from Malvern Instruments Ltd. featuring a dynamic light scattering analyzer. Absorption spectra were recorded with a Lambda 2 UV-vis spectrometer from Perkin Elmer in 10 mm quartz cuvettes. Fluorescence spectra were recorded with a Fluoromax 3 spectrofluorometer from Horiba in 10 mm quartz cuvettes. Bright field and dark field TEM analysis was performed with a TEM 912 Omega from Zeiss. Specimen grids were prepared by drop casting of the sample solution onto the TEM grids and subsequent drying in a desiccator. For dark field imaging the direct transmitted beam was blocked with the objective aperture and only the beam deflected by carbon or gold atoms was recorded. TEM images of the cells loaded with the NPs were received using a Zeiss 906 TEM. Transient absorption spectroscopy was performed with either 387, 505, or 550 nm laser pump pulses (1 kHz, 150 fs pulse width) from an amplified Ti:sapphire laser system CPA 2110 from Clark-MXR Inc. The viability assays were measured with a Synergy HT microplate reader from BioTek.

Absorption spectroscopy

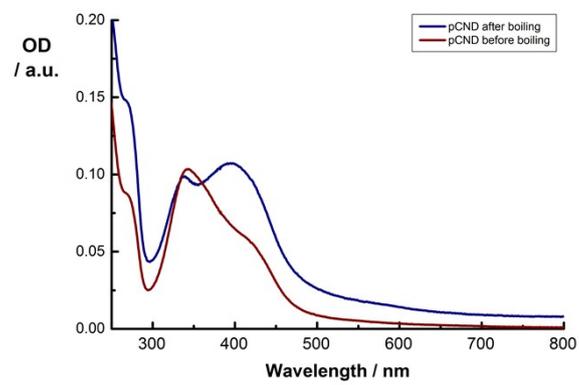


Figure S1. Absorption spectra of pCND in H₂O before boiling (red) and after boiling (blue).

Dynamic light scattering analysis

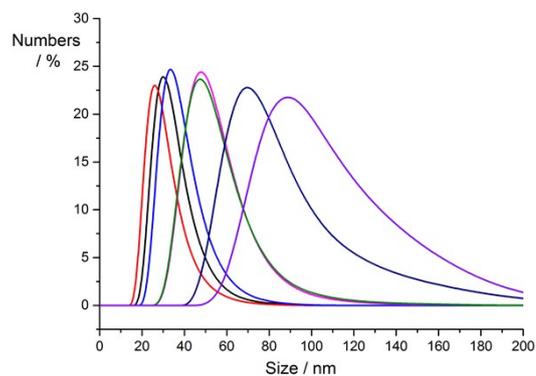


Figure S2. Dynamic light scattering plots of various HAuCl₄ (red) 0.75 mM (black) 1.00 mM (blue) 1.25 mM (green) 2.00 mM (magenta) 2.50 mM (navy blue) 3.00 mM (violet) 4.00 mM

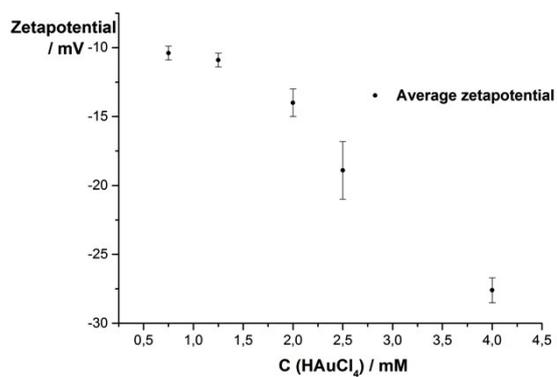


Figure S3. Zetapotential plot versus the conc. of HAuCl₄

Raman spectroscopy

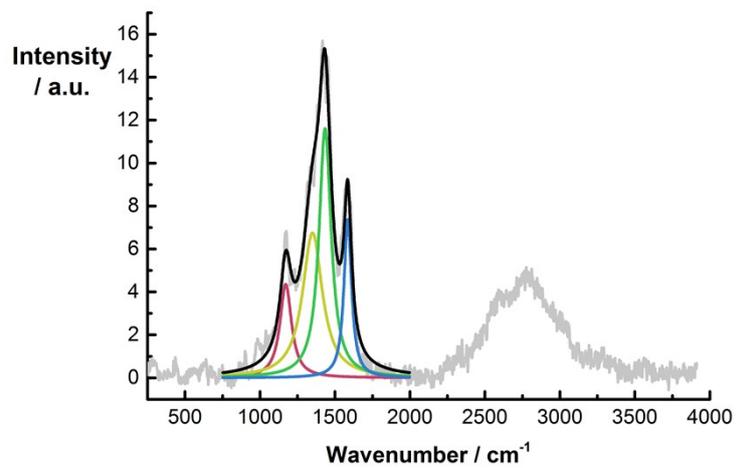


Figure S4. Raman spectrum of Au@pCND (gray) obtained upon 457 nm laser excitation with multiple Lorentzian peak fitting of the D- and G-band region.

Time-correlated single-photon counting

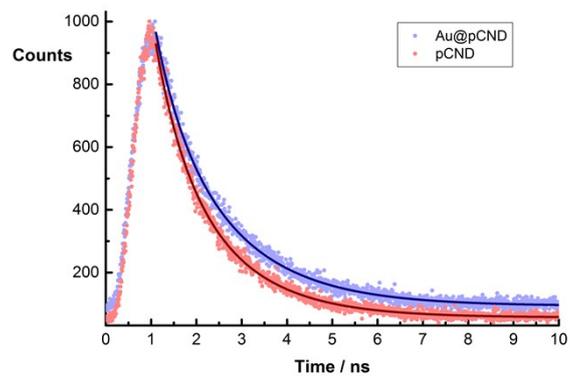


Figure S5. Time-emission profiles of pCND (red) and Au@pCND (blue) at 500 nm obtained by means of TCSPC upon excitation at 410 nm.

Transient absorption characterization

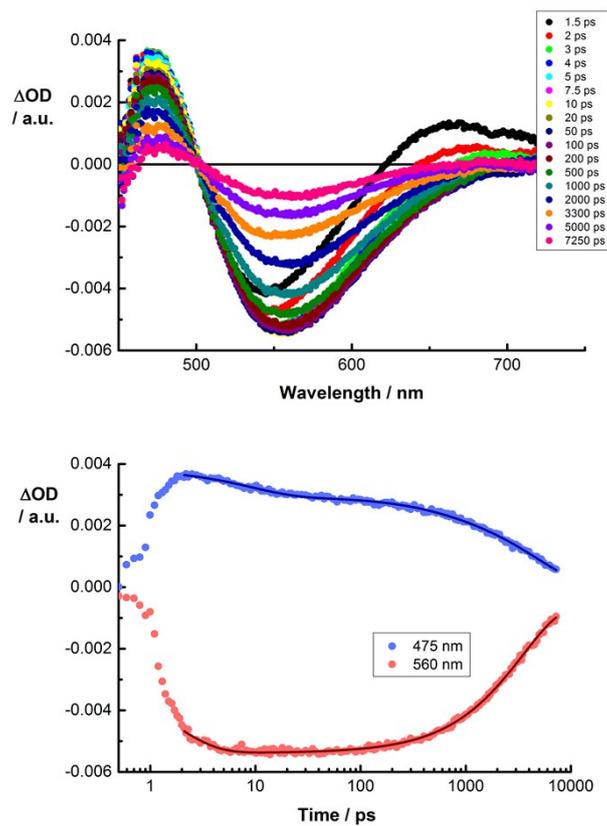


Figure S6. Left: Transient absorption spectra of 30 min boiled pCND obtained upon laser flash photolysis excited with 387 nm laser pulses in H₂O with time delays between 1.5 and 7250 ps. Right: Corresponding time-absorption profiles at 475 nm (blue) and 560 nm (red).

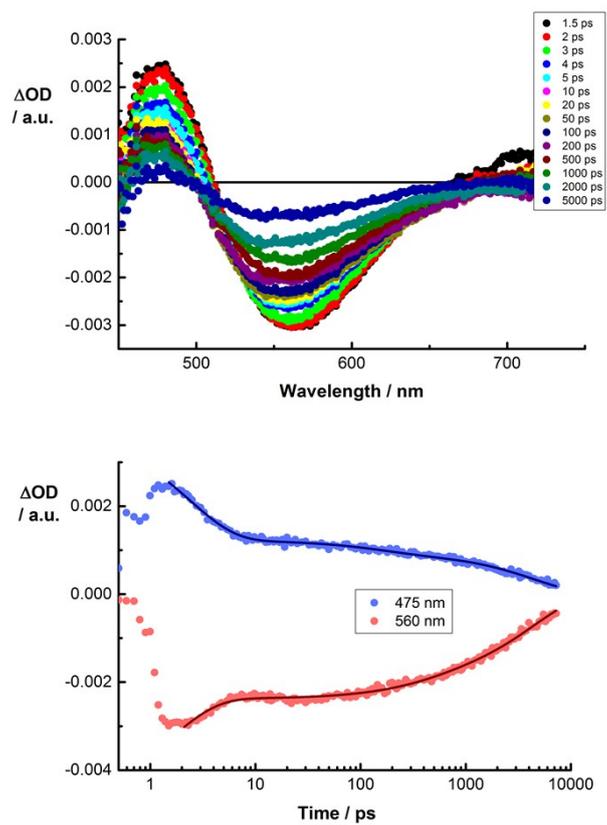


Figure S7. Left: Transient absorption spectra of Au@pCND obtained upon laser flash photolysis excited with 387 nm laser pulses in H₂O with time delays between 1.5 and 5000 ps. Right: Corresponding time-absorption profiles at 475 nm (blue) and 560 nm (red).

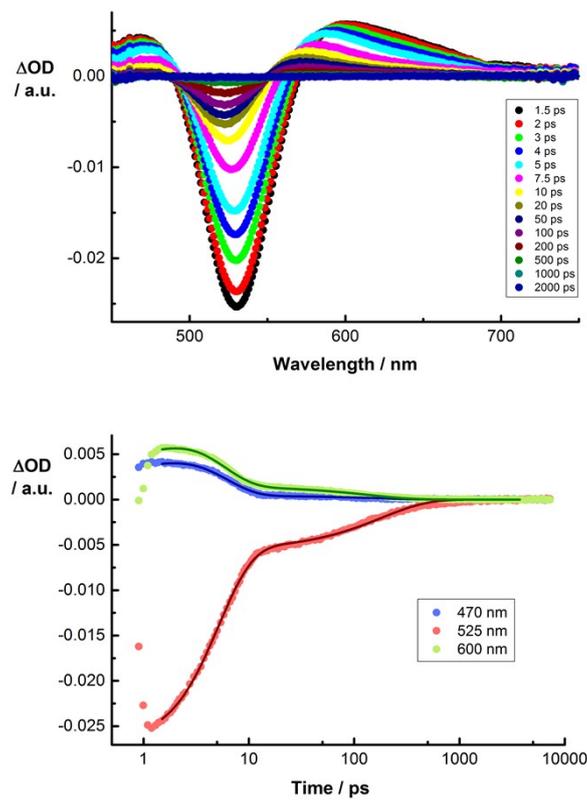


Figure S8. Left: Transient absorption spectra of citric acid stabilized AuNP obtained upon laser flash photolysis excited with 387 nm laser pulses in H₂O with time delays between 1.5 and 2000 ps. Right: Corresponding time-absorption profiles at 470 nm (blue), 525 nm (red) and 600 nm (green).

Transmission electron microscopy

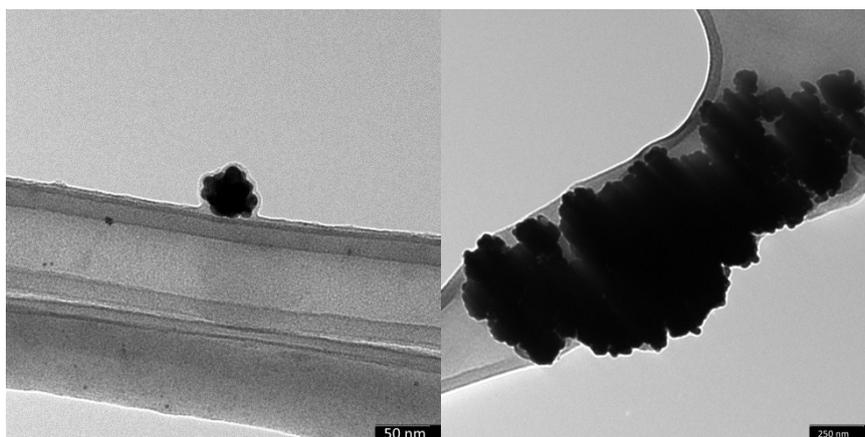


Figure S9. Representative TEM images of Au@pCNDs (left) and Au@pCNDs after addition of MPA (right).

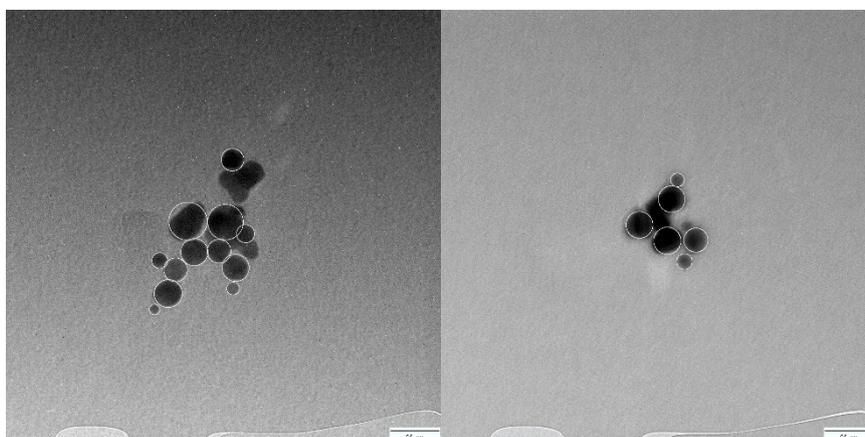


Figure S10. TEM images of Au@pCNDs synthesized by standard procedure and 0.75 mM HAuCl₄ conc.

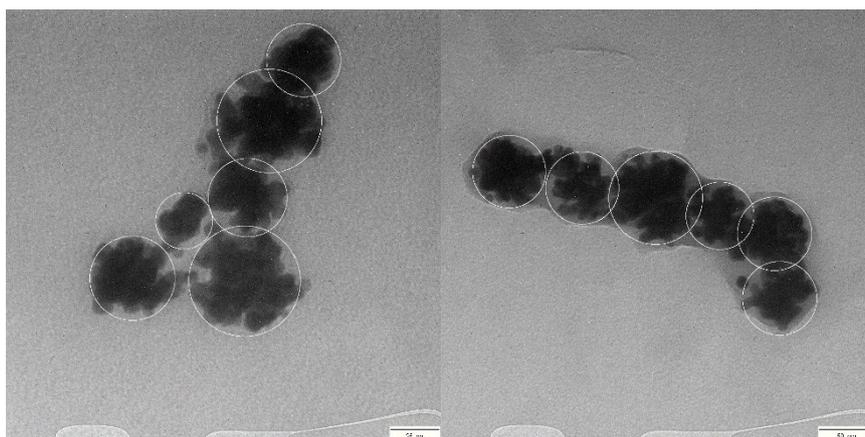


Figure S11. TEM images of Au@pCNDs synthesized by standard procedure and 1.00 mM HAuCl₄ conc.

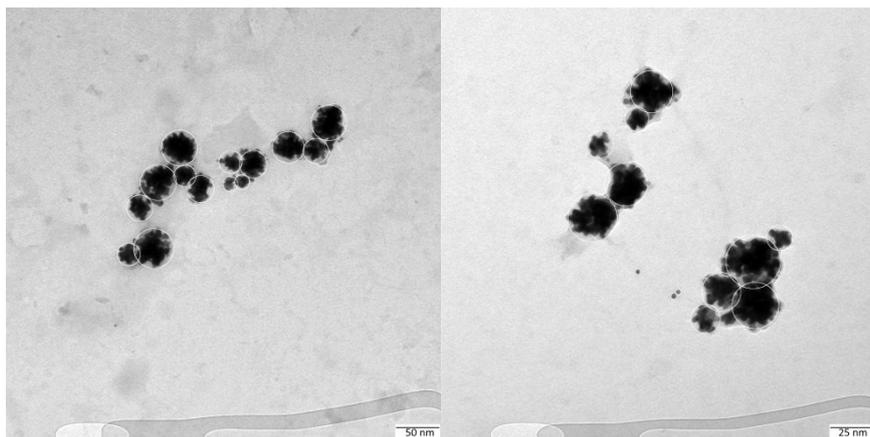


Figure S12. TEM images of Au@pCNDs synthesized by standard procedure and 1.25 mM HAuCl₄ conc.

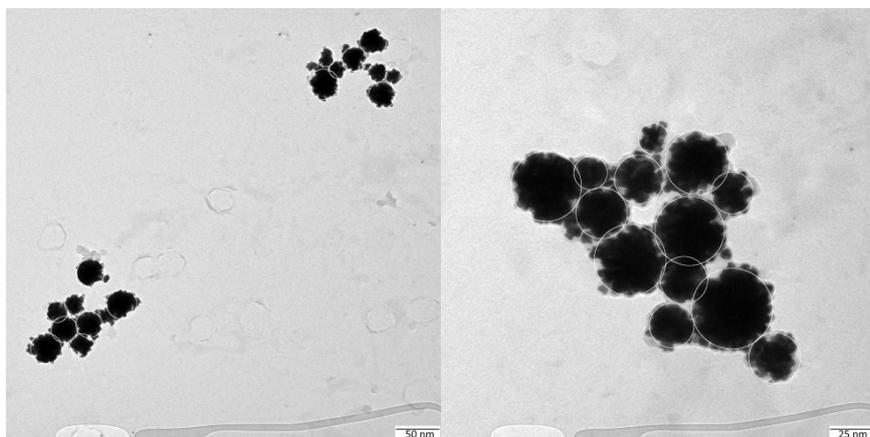


Figure S13. TEM images of Au@pCNDs synthesized by standard procedure and 2.00 mM HAuCl₄ conc.

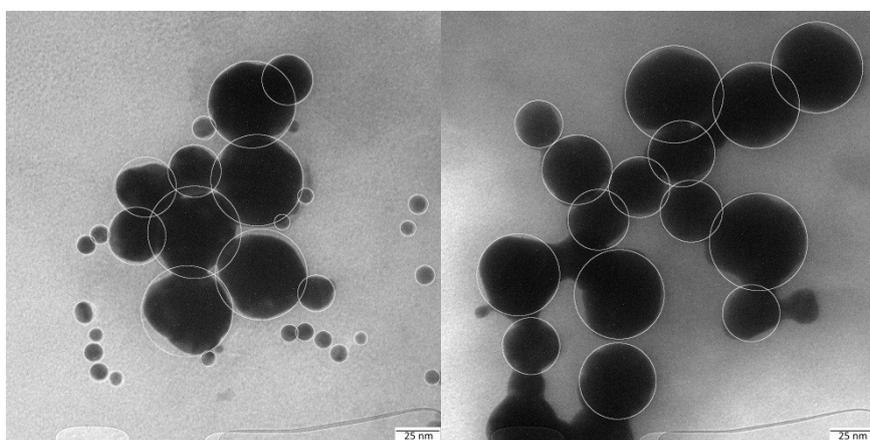


Figure S14. TEM images of Au@pCNDs synthesized by standard procedure and 2.50 mM HAuCl₄ conc.

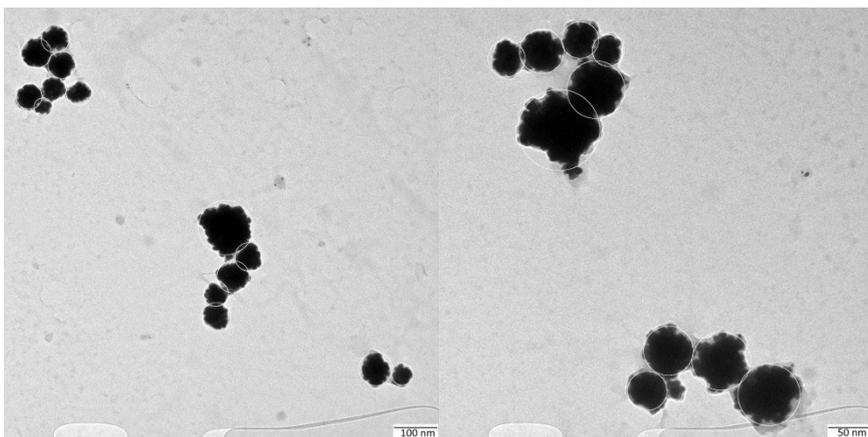


Figure S15. TEM images of Au@pCNDs synthesized by standard procedure and 3.00 mM HAuCL₄ conc.

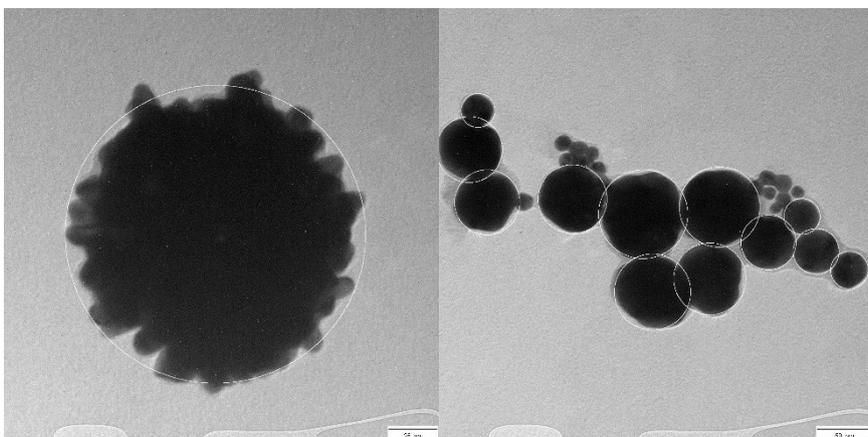


Figure S16. TEM images of Au@pCNDs synthesized by standard procedure and 4.00 mM HAuCL₄ conc.

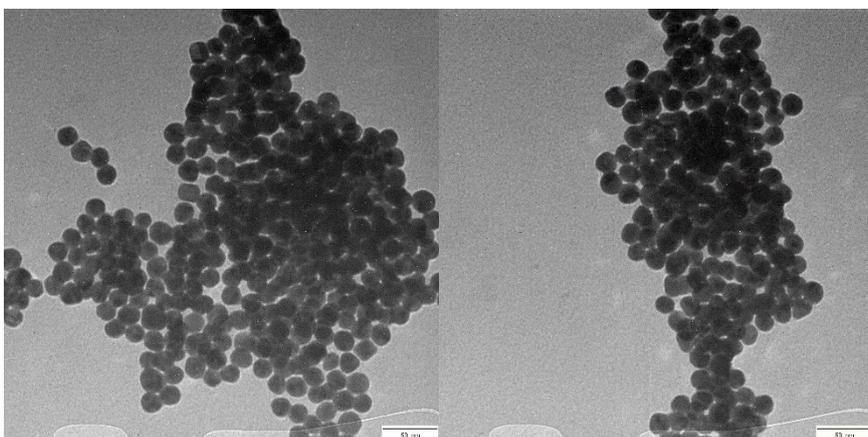


Figure S17. TEM images of Au@NPs synthesized by citrate reduction and 2.00 mM HAuCL₄ conc.

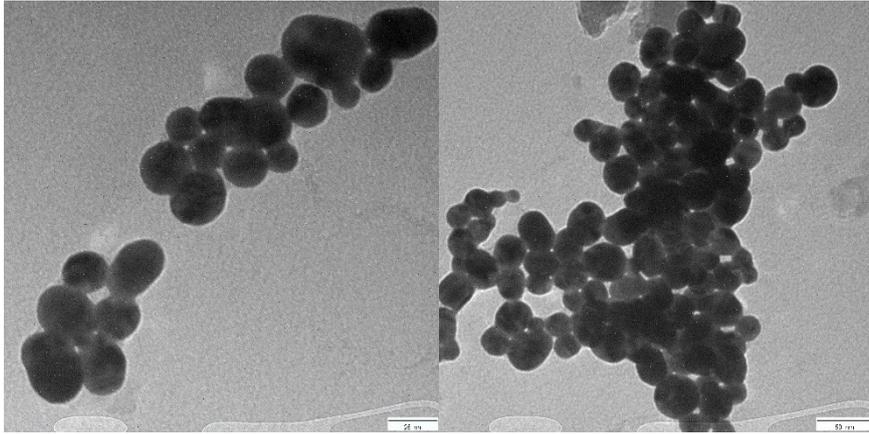


Figure S18. TEM images of Au@NPs synthesized by citrate reduction and 3.00 mM HAuCL₄ conc.

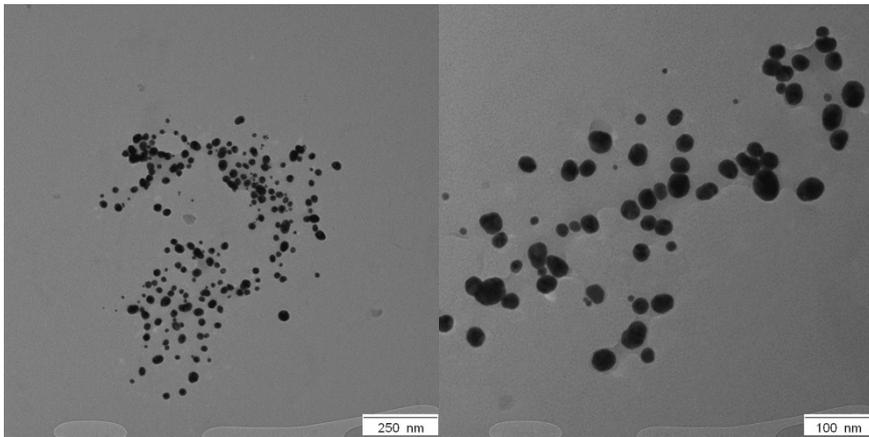


Figure S19. TEM images of Au@NPs synthesized by citrate reduction and 4.00 mM HAuCL₄ conc.

Dynamic light scattering and Fluorescence measurements in Serum

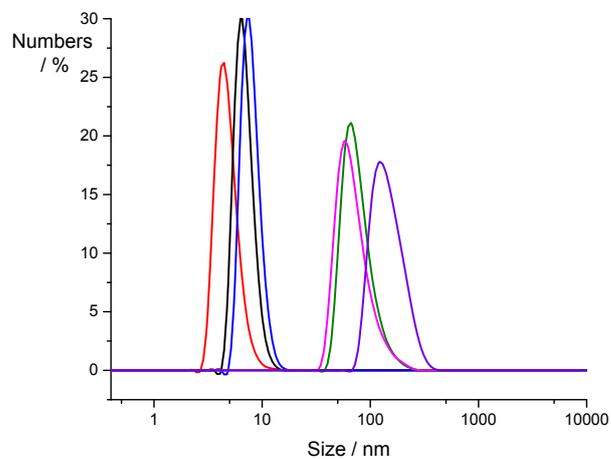


Figure S20. Dynamic light scattering plots of various HAuCl₄ (red) 0.75 mM (black) 1.00 mM (blue) 1.25 mM (green) 2.00 mM (magenta) 2.50 mM (violet) 4.00 mM in serum containing PBS

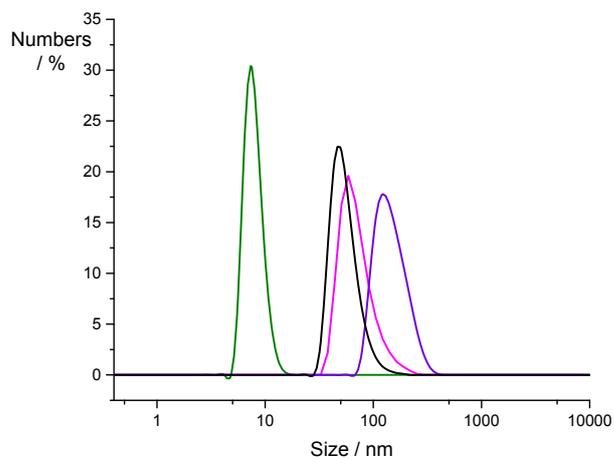


Figure S21. Comparison of dynamic light scattering sizes of various HAuCl₄ (green) 1.25 mM, (magenta) 2.50 mM and (violet) 4.00 mM to citric acid reduced/ stabilized AuNP in serum containing PBS that are used for the radiation experiments

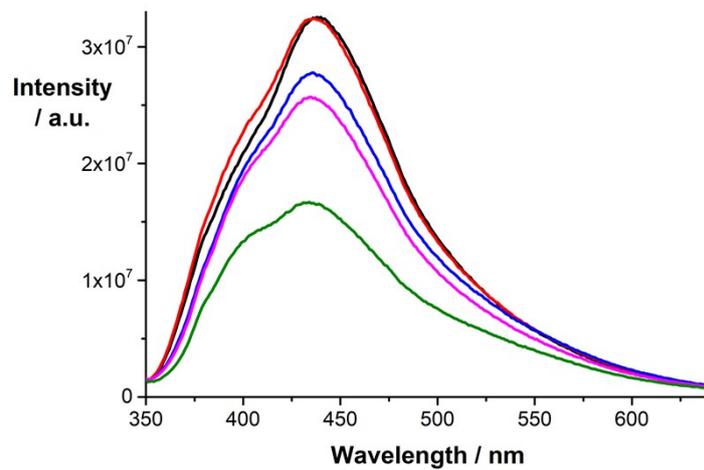


Figure S22. Steady state fluorescence spectra of Au@pCND in the cell medium obtained when reacting 0.06 mg/ml pCND with 1.00 (black), 1.25 (red), 2.00 (blue), 2.50 (magenta), and 4.00 (green) mM H₂AuCl₄.

Additional Synthesis attempts

Table S1. Results on high concentration of precursors, here the concentration of pCND is 0.385 mM. Note: These procedures were done using a sand bath. The reactions were quenched with H₂O (in batch 3 – 7 the)

batch	C _{pCND} [mg/mL]	C _{H₂AuCl₄} [mM]	ratio	R _{time} [min]	R _{temp.} [°C]	DLS size [nm]	TEM size [nm]	λ _{max} [nm]
1	1.20	25	1:2.5	1	100	70-75	7-14 particles, 50-80 agglomerates	~560
2 ^{a)}	1.20	50	1:5	10	100	80-85	57/75/92 agglomerates	~570
3 ^{b)}	1.20	50	1:5	10	100	80-85	no data	~550
4 ^{b)}	1.20	25	1:2.5	10	100	90	~7 small particles 90-100 agglomerates	just scattering
5 ^{b)c)}	1.20	100	1:10	10	100	400-600	2-5 a lot, 10-15 few	~580
6 ^{b)}	1.20	200	1:20	10	100	40	no data	~570
7 ^{b)}	1.20	12.5	1:1.25	10	100	58	9-16 particles, ~60 agglomerates	~560
8	1.20	50	1:5	30	100	90-150	no data	just scattering
9	1.20	50	1:5	10	60	polydispers	No particles observed	just scattering
10 ^{a)}	1.20	50	1:5	10	60	polydispers	no data	just scattering

^{a)}(MW heated) ^{b)}(CND preheated) ^{c)}(pH = 10)

Table S2. Results on low concentration of precursors. In this procedure we are able to produce smaller nanoparticles, for example, 10-20 nm, we do not observe any correlation between precursor concentration and particle sizes. Note: Reactions were quenched in an ice bath.

batch	C _{pCND} [mg/mL]	C _{H₂AuCl₄} [mM]	ratio	R _{time} [min]	R _{temp.} [°C]	DLS size [nm]	TEM size [nm]	λ _{max} [nm]
11	0.12	20	1:20	10	100	12	several homogenous particles, 60-70	broad ~560
12	0.12	10	1:10	10	100	57	50-72	~610
13	0.12	5	1:5	10	100	19	no data	just scattering
14	0.12	20	1:20	1	100	polydispers	no data	~640
15	0.12	10	1:10	1	100	48	86-90 agglomerates	~560
16	0.12	2	1:20	10	100	polydispers	no data	just scattering
17	0.006	2	1:40	10	100	polydispers	no data	just scattering
18	0.12	2	1:20	10	100	polydispers	no data	~600
19	0.24	2	1:10	10	100	polydispers	no data	~590
20	0.006	2	1:40	10	100	polydispers	no data	just scattering
21	0.12	2	1:20	5	100	polydispers	no data	just scattering
22	0.12	2	1:20	2	100	polydispers	no data	~600

Radiation therapy experiments

Table S3. Dose modifying factor of 6, 60 and 106 nm Au@pCND, citric acid reduced/stabilized AuNP and bare pCNDs

	DMF
pCND	0.72
Citric acid AuNP	0.62
1.25	0.85
2.5	0.53
4	0.52