

ESI for Journal of Materials Chemistry C: C7TC01197K

Table S1. Phase transition temperatures and enthalpy changes of (*m*-C_nOPhO)₈PcCu (n = 1-20: **2a-o**).

Compound (<i>m</i> -C _{<i>n</i>} OPhO) ₈ PcCu	Phase	T (°C) [ΔH (kJmol ⁻¹)]	Phase	$\rightsquigarrow$ relaxation		
<i>n</i> = 1: 2a ^{*1}	Glassy Col _{ro} (P2 ₁ /a)	Tg=69.9				
		K	191.3 [13.4] Col _{ro1} (P2 ₁ /a)	255.1 [2.69] Col _{ro2} (P2 ₁ /a)	360.1 [10.7] M _x (1st dc.)	405 2nd dc.
				K ₁ $\xrightleftharpoons{229.4}$ Col _r (P2m)	$\xrightleftharpoons{354.1 [19.6]}$	I.L.
				274.7 [100.5] $\xrightarrow{197.0}$		
<i>n</i> = 2: 2b		K _{2v} $\xrightarrow{84.2}$ K ₃	$\xrightarrow{181.4}$ K ₄			
<i>n</i> = 3: 2c			142.8[41.5] K ₃ $\xrightleftharpoons{207.4[17.5]}$ K ₄	$\xrightleftharpoons{234.1[29.9]}$ Col _r (P2m)	$\xrightleftharpoons{300.1[17.3]}$	I.L.
		K ₁ $\xleftarrow{113.4 [5.0]}$ K _{2v}				
<i>n</i> = 4: 2d				Glassy Col _r (P2m) _(v)	Tg=115.7	
					K ₂ $\xrightarrow{166.1[6.4]}$	
<i>n</i> = 5: 2e			K ₁ $\xrightleftharpoons{147.5[10.6]}$ K ₃	$\xrightleftharpoons{174.2 [21.0]}$ Col _r (P2m)	$\xrightleftharpoons{249.5 [11.4]}$	I.L.
<i>n</i> = 6: 2f			K _{1v} $\xrightarrow{68.4[6.02]}$ K ₂	$\xrightarrow{131.0^i}$ Col _r (P2m)	$\xrightleftharpoons{222.3[12.6]}$	I.L.
<i>n</i> = 7: 2g			Col _{ho} $\xleftarrow{169.3}$			
<i>n</i> = 8: 2h						
<i>n</i> = 9: 2i						
<i>n</i> = 10: 2j						
<i>n</i> = 11: 2k						
<i>n</i> = 12: 2l						
<i>n</i> = 13: 2m						
<i>n</i> = 14: 2n						
<i>n</i> = 15: 2o						
<i>n</i> = 16: 2p						
<i>n</i> = 17: 2q						
<i>n</i> = 18: 2r						
<i>n</i> = 19: 2s						
<i>n</i> = 20: 2t						
<i>n</i> = 21: 2u						
<i>n</i> = 22: 2v						
<i>n</i> = 23: 2w						
<i>n</i> = 24: 2x						
<i>n</i> = 25: 2y						
<i>n</i> = 26: 2z						
<i>n</i> = 27: 2aa						
<i>n</i> = 28: 2ab						
<i>n</i> = 29: 2ac						
<i>n</i> = 30: 2ad						
<i>n</i> = 31: 2ae						
<i>n</i> = 32: 2af						
<i>n</i> = 33: 2ag						
<i>n</i> = 34: 2ah						
<i>n</i> = 35: 2ai						
<i>n</i> = 36: 2aj						
<i>n</i> = 37: 2ak						
<i>n</i> = 38: 2al						
<i>n</i> = 39: 2am						
<i>n</i> = 40: 2an						
<i>n</i> = 41: 2ao						
<i>n</i> = 42: 2ap						
<i>n</i> = 43: 2aq						
<i>n</i> = 44: 2ar						
<i>n</i> = 45: 2as						
<i>n</i> = 46: 2at						
<i>n</i> = 47: 2au						
<i>n</i> = 48: 2av						
<i>n</i> = 49: 2aw						
<i>n</i> = 50: 2ax						
<i>n</i> = 51: 2ay						
<i>n</i> = 52: 2az						
<i>n</i> = 53: 2ba						
<i>n</i> = 54: 2bb						
<i>n</i> = 55: 2bc						
<i>n</i> = 56: 2bd						
<i>n</i> = 57: 2be						
<i>n</i> = 58: 2bf						
<i>n</i> = 59: 2bg						
<i>n</i> = 60: 2bh						
<i>n</i> = 61: 2bi						
<i>n</i> = 62: 2bj						
<i>n</i> = 63: 2bk						
<i>n</i> = 64: 2bl						
<i>n</i> = 65: 2bm						
<i>n</i> = 66: 2bn						
<i>n</i> = 67: 2bo						
<i>n</i> = 68: 2bp						
<i>n</i> = 69: 2bq						
<i>n</i> = 70: 2br						
<i>n</i> = 71: 2bs						
<i>n</i> = 72: 2bt						
<i>n</i> = 73: 2bu						
<i>n</i> = 74: 2bv						
<i>n</i> = 75: 2bw						
<i>n</i> = 76: 2bx						
<i>n</i> = 77: 2by						
<i>n</i> = 78: 2bz						
<i>n</i> = 79: 2ca						
<i>n</i> = 80: 2cb						
<i>n</i> = 81: 2cc						
<i>n</i> = 82: 2cd						
<i>n</i> = 83: 2ce						
<i>n</i> = 84: 2cf						
<i>n</i> = 85: 2cg						
<i>n</i> = 86: 2ch						
<i>n</i> = 87: 2ci						
<i>n</i> = 88: 2cj						
<i>n</i> = 89: 2ck						
<i>n</i> = 90: 2cl						
<i>n</i> = 91: 2cm						
<i>n</i> = 92: 2cn						
<i>n</i> = 93: 2co						
<i>n</i> = 94: 2cp						
<i>n</i> = 95: 2cq						
<i>n</i> = 96: 2cr						
<i>n</i> = 97: 2cs						
<i>n</i> = 98: 2ct						
<i>n</i> = 99: 2cu						
<i>n</i> = 100: 2cv						
<i>n</i> = 101: 2cw						
<i>n</i> = 102: 2cx						
<i>n</i> = 103: 2cy						
<i>n</i> = 104: 2cz						
<i>n</i> = 105: 2da						
<i>n</i> = 106: 2db						
<i>n</i> = 107: 2dc						
<i>n</i> = 108: 2dd						
<i>n</i> = 109: 2de						
<i>n</i> = 110: 2df						
<i>n</i> = 111: 2dg						
<i>n</i> = 112: 2dh						
<i>n</i> = 113: 2di						
<i>n</i> = 114: 2dj						
<i>n</i> = 115: 2dk						
<i>n</i> = 116: 2dl						
<i>n</i> = 117: 2dm						
<i>n</i> = 118: 2dn						
<i>n</i> = 119: 2do						
<i>n</i> = 120: 2dp						
<i>n</i> = 121: 2dq						
<i>n</i> = 122: 2dr						
<i>n</i> = 123: 2ds						
<i>n</i> = 124: 2dt						
<i>n</i> = 125: 2du						
<i>n</i> = 126: 2dv						
<i>n</i> = 127: 2dw						
<i>n</i> = 128: 2dx						
<i>n</i> = 129: 2dy						
<i>n</i> = 130: 2dz						
<i>n</i> = 131: 2ea						
<i>n</i> = 132: 2eb						
<i>n</i> = 133: 2ec						
<i>n</i> = 134: 2ed						
<i>n</i> = 135: 2ee						
<i>n</i> = 136: 2ef						
<i>n</i> = 137: 2eg						
<i>n</i> = 138: 2eh						
<i>n</i> = 139: 2ei						
<i>n</i> = 140: 2ej						
<i>n</i> = 141: 2ek						
<i>n</i> = 142: 2el						
<i>n</i> = 143: 2em						

$n = 7$: **2g**

$$\begin{array}{ccccccc}
 X_{1v} & \xrightarrow{57.4} & X_2 & \xrightarrow{101.4[6.55]} & \text{Col}_{ro} & \xrightarrow{202.2[32.9]} & \text{I.L.} \\
 & & & & (P2_1/a) & & \\
 \text{Col}_{ho1} & \xleftarrow{175.8} & \text{Col}_{ho2} & \xleftarrow{197.9} & \text{Col}_{ho3} & \xleftarrow{202.8} & \uparrow \\
 & & & & & & 203.4[33.7]
 \end{array}$$

$n = 8$: **2h**

$$\begin{array}{ccccccc}
 X_{1v} & \xrightarrow{40.4[8.11]} & X_2 & \xrightarrow{83.4[3.41]} & \text{Col}_{ho1} & \xrightarrow{118.9[0.81]} & \text{Col}_{ro} & \xrightarrow{197.2[34.0]} & \text{I.L.} \\
 & & & & & & (P2_1/a) & & \\
 & & & & \text{Col}_{ho2} & \xleftarrow{170.9} & \text{Col}_{ho3} & \xleftarrow{200.4} & \uparrow \\
 & & & & & & & & 199.3[34.3]
 \end{array}$$

$n = 9$: **2i**

$$\begin{array}{ccccc}
 \text{Col}_{ho1(v)} & \xrightleftharpoons{74.1[1.64]} & \text{Col}_{ho2} & \xrightleftharpoons{193.0\ddot{r}} & \text{I.L.} \\
 & & & \nearrow 141-159 & \\
 & & & \text{Col}_{ro} & \nearrow 194.9\ddot{r} \\
 & & & (P2_1/a) &
 \end{array}$$

$n = 10$: **2j^{*2}**

$$\begin{array}{ccc}
 \text{Col}_{hd} & \xrightarrow{187.0[22.2]} & \text{I.L.} \\
 & \uparrow 188.2[19.3] & \\
 \text{Col}_{ho} & \xleftarrow{\quad} &
 \end{array}$$

$n = 12$: **2k^{*3}**

$$\begin{array}{ccc}
 \text{Col}_{hd} & \xrightarrow{178.4[27.4]} & \text{I.L.} \\
 & \uparrow 179.2[29.6] & \\
 \text{Col}_{ho} & \xleftarrow{\quad} &
 \end{array}$$

$n = 14$: **2l^{*3}**

$$\begin{array}{ccc}
 \text{Col}_{hd} & \xrightarrow{169.0[22.0]} & \text{I.L.} \\
 & \uparrow 169.6[25.5] & \\
 \text{Col}_{ho} & \xleftarrow{\quad} &
 \end{array}$$

$n = 16$: **2m^{*2}**

$$\begin{array}{ccc}
 K & \xrightleftharpoons{34.5[64.8]} & \text{Col}_{ho} & \xrightleftharpoons{160.9[18.0]} & \text{I.L.}
 \end{array}$$

$n = 18$: **2n^{*2}**

$$\begin{array}{ccc}
 K & \xrightleftharpoons{32.9[240]} & \text{Col}_{ho} & \xrightleftharpoons{152.3[27.8]} & \text{I.L.}
 \end{array}$$

$n = 20$: **2o^{*2}**

$$\begin{array}{ccc}
 K & \xrightleftharpoons{50.1[278]} & \text{Col}_{ho} & \xrightleftharpoons{156.3[20.6]} & \text{I.L.}
 \end{array}$$

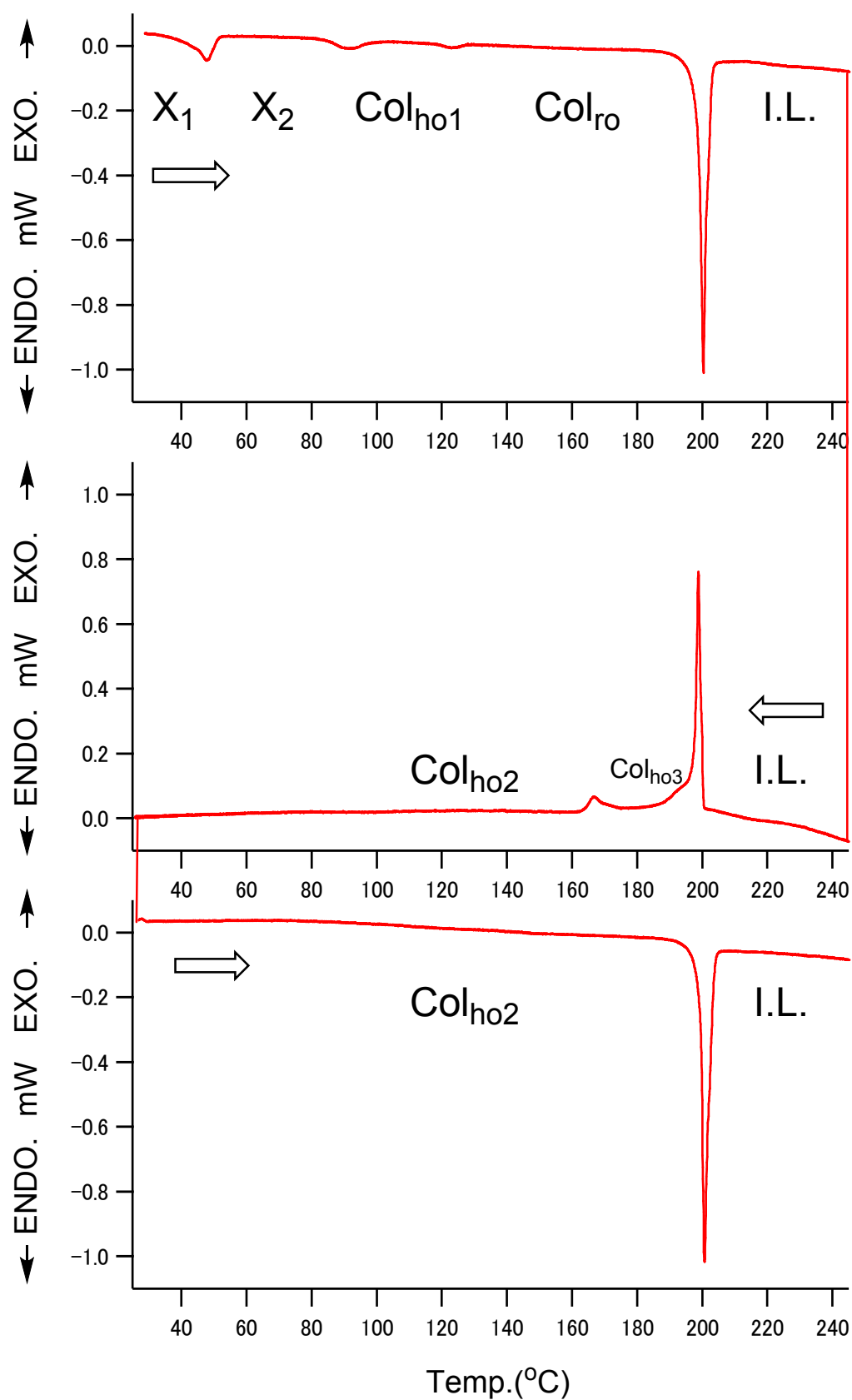


Fig. S1. DSC thermograms of $(m\text{-C}_8\text{OPhO})_8\text{PcCu}$ (**2h**). The heating and cooling rates were $2.5^{\circ}\text{C}/\text{min}$.

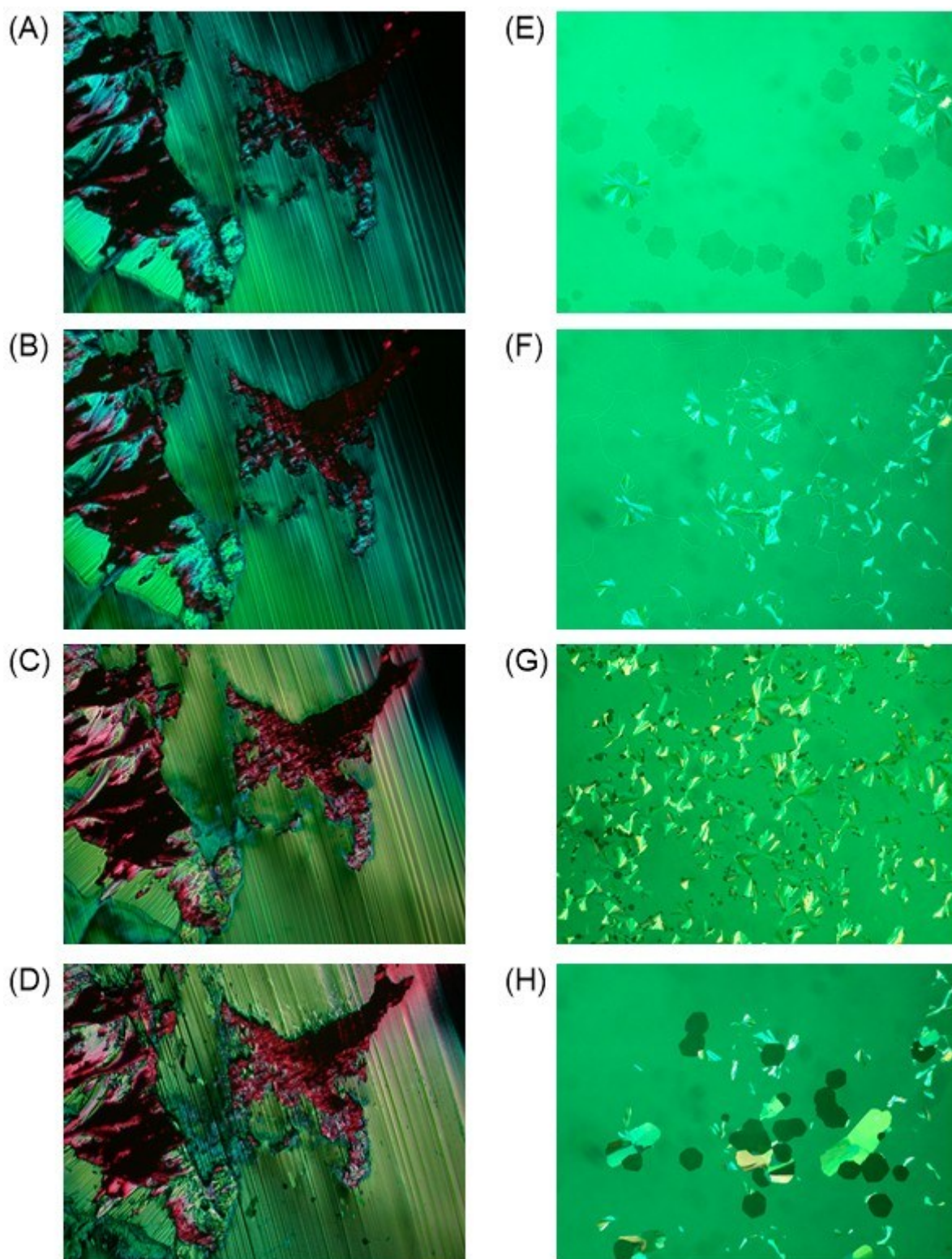


Fig. S2. Photomicrographs of $(m\text{-C}_8\text{OPhO})_8\text{PcCu}$ (**2h**) at various temperatures. (A) The virgin state of X_1 at 27.7°C; (B) X_2 at 60.0°C without discernible change; (C) Col_{h01} at 100.0°C with change of the birefringence; (D) Col_{ro} at 160.0°C; (E) when IL heated over 205.0°C was cooled down to 200.0°C, it was transforming into Col_{h03} ; (F) the IL was completely transformed into Col_{h03} ; (G) when the Col_{h03} in (F) was rapidly cooled down to RT, a mixture of supercooled Col_{h03} and partially resulted Col_{h02} (small dark dots); (H) only when the supercooled Col_{h03} was held at 165.0°C, the Col_{ro} (bright platelets) phase appeared with Col_{h02} (dark hexagons).

Table S1. X-ray data of **2a-o**.

Compound	Mesophase lattice constants(Å)	Peak No.	Spacing(Å)		Miller indices (<i>h k l</i>)
			Observed	Calculated	
2a: (<i>m</i> -C ₁ OPhO) ₈ PcCu [†]	Col _{ro1} (P2 ₁ /a) at 230 °C a = 29.4 b = 26.5 h = 6.09 Z = 2.0 for $\rho = 1.1$	1	20.1	19.7	(1 1 0)
		2	13.2	13.3	(0 2 0)
		3	11.8	12.1	(1 2 0)
		4	9.24	9.19	(3 1 0)
		5	8.46	8.46	(1 3 0)
		6	7.39	7.35	(4 0 0)
		7	6.51	6.47	(1 4 0)
		8	6.09	-	<i>h</i>
		9	5.17	5.22	(1 5 0)
		10	4.84	4.82	(6 1 0)
		11	4.54	4.60	(6 2 0)
		12	4.41	4.41	(0 6 0)
		13	4.15	4.15	(7 1 0)
	Col _{ro2} (P2 ₁ /a) at 310 °C a = 33.6 b = 27.1 h = 5.00 Z = 2.0 for $\rho = 1.1$	1	21.1	21.1	(1 1 0)
		2	16.8	16.8	(2 0 0)
		3	13.6	13.6	(0 2 0)
		4	8.61	8.64	(3 2 0)
		5	7.18	7.14	(4 2 0)
		6	6.48	6.52	(5 1 0)
		7	5.73	5.80	(3 4 0)
		8	5.33	5.36	(1 5 0)
		9	5.00	-	<i>h</i>
		10	4.57	4.56	(4 5 0)
		11	4.32	4.32	(6 4 0)
		12	3.97	3.98	(4 6 0)
		13	3.72	3.74	(5 6 0)
		14	3.55	3.55	(8 4 0)
		15	3.42	3.42	(0 8 0)
	M _x at 370 °C	1	19.5	-	-
		2	ca. 3.8	-	<i>h</i>
2b: (<i>m</i> -C ₂ OPhO) ₈ PcCu	Col _r (P2m) at 290°C a = 21.2 b = 17.3	1	21.2	21.2	(1 0 0)
		2	17.4	17.3	(0 1 0)
		3	13.4	13.4	(1 1 0)
		4	8.64	8.64	(0 2 0)
		5	6.71	6.71	(2 2 0)
		6	ca. 5.10	-	#1
2c: (<i>m</i> -C ₃ OPhO) ₈ PcCu	Col _r (P2m) at 265°C a = 22.5 b = 18.5	1	22.7	22.5	(1 0 0)
		2	18.5	18.5	(0 1 0)
		3	14.1	14.3	(1 1 0)
		4	9.10	9.22	(0 2 0)
		5	7.08	7.14	(2 2 0)
		6	ca. 5.06	-	#1
2d: (<i>m</i> -C ₄ OPhO) ₈ PcCu	Col _r (P2m) at 210°C a = 23.6 b = 20.3	1	23.6	23.6	(1 0 0)
		2	20.3	20.3	(0 1 0)
		3	9.43	9.43	(1 2 0)
		4	7.67	7.70	(2 2 0)
		5	ca. 4.90	-	#1
2e: (<i>m</i> -C ₅ OPhO) ₈ PcCu	Col _{ho} at 150°C a = 31.4 h = 3.41 Z = 1.0 for $\rho = 1.1$	1	27.2	27.2	(1 0 0)
		2	16.2	15.7	(1 1 0)
		3	10.2	10.3	(1 2 0)
		4	9.10	9.06	(3 0 0)
		5	ca. 4.82	-	#2
		6	3.41	-	<i>h</i>
	Col _r (P2m) at 210°C a = 24.8 b = 22.1	1	24.8	24.8	(1 0 0)
		2	22.1	22.1	(0 1 0)
		3	9.95	10.1	(1 2 0)
		4	8.38	8.27	(2 2 0)
		5	ca. 4.87	-	#1
2f: (<i>m</i> -C ₆ OPhO) ₈ PcCu	Col _{ro1} (P2 ₁ /a) at 178°C a = 47.6 b = 36.3 h = 3.50 Z = 2.0 for $\rho = 1.2$	1	28.9	28.1	(1 1 0)
		2	23.8	23.8	(2 0 0)
		3	17.3	17.4	(0 2 0)
		4	10.4	10.4	(2 3 0)
		5	9.47	9.53	(3 3 0)
		6	8.59	8.54	(1 4 0)
		7	7.38	7.36	(5 3 0)
		8	6.22	6.34	(7 2 0)
		9	ca. 5.08	-	#1
		10	3.50	-	<i>h</i>
	Col _{ro2} (P2 ₁ /a) at 203°C a = 47.8 b = 35.3 h = 3.51 Z = 2.0 for $\rho = 1.2$	1	28.8	28.4	(1 1 0)
		2	24.2	23.9	(2 0 0)
		3	10.5	10.6	(2 3 0)
		4	9.49	9.46	(3 3 0)
		5	8.61	8.67	(1 4 0)
		6	7.45	7.42	(5 3 0)
		7	6.76	6.77	(2 5 0)
		8	6.30	6.45	(3 5 0)
		9	ca. 5.34	-	#1
		10	4.70	4.73	(6 6 0)
		11	3.82	3.81	-
		12	3.51	-	<i>h</i>
		13	3.23	3.23	-

Table S1. (Continued).

2f: (m-C₆OPhO)₈PcCu	Col _{h01} at 31°C	1	28.0	28.0	(1 0 0)
	a = 32.4	2	16.2	16.2	(1 1 0)
	h = 3.34	3	10.4	10.6	(1 2 0)
	Z = 1.0 for $\rho = 1.2$	4	9.31	9.35	(3 0 0)
		5	ca. 4.67	-	#2
		6	3.99	4.01	(3 5 0)
		7	3.34	-	h
	Col _{h02} at 100°C	1	28.5	28.5	(1 0 0)
	a = 32.9	2	16.5	16.4	(1 1 0)
	h = 3.38	3	10.6	10.8	(1 2 0)
	Z = 1.0 for $\rho = 1.1$	4	9.37	9.49	(3 0 0)
		5	ca. 4.79	-	#2
		6	3.38	-	h
	Col _{h03} at 155°C	1	28.5	28.5	(1 0 0)
	a = 32.9	2	17.5	16.5	(1 1 0)
	h = 3.42	3	10.7	10.8	(1 2 0)
	Z = 1.0 for $\rho = 1.1$	4	ca. 4.84	-	#2
		5	3.42	-	h
	Col _{h04} at 190°C	1	26.7	26.7	(1 0 0)
	a = 30.9	2	9.92	10.1	(1 2 0)
	h = 3.61	3	ca. 4.85	-	#2
	Z = 1.0 for $\rho = 1.2$	4	3.61	-	h
2g: (m-C₇OPhO)₈PcCu	Col _{ro} (P2 ₁ /a) at 150°C	1	29.8	29.0	(1 1 0)
	a = 48.6 b = 36.1	2	24.0	24.3	(2 0 0)
	h = 3.48	3	16.8	16.9	(1 2 0)
	Z = 2.0 for $\rho = 1.2$	4	10.7	10.8	(2 3 0)
		5	9.70	9.66	(3 3 0)
		6	8.79	8.86	(1 4 0)
		7	7.63	7.56	(5 3 0)
		8	ca. 5.14	-	#1
		9	3.77	-	
		10	3.48	-	h
		11	3.20	-	
	Col _{h01} at r.t.	1	29.2	29.2	(1 0 0)
	a = 33.7	2	16.7	16.8	(1 1 0)
	h = 3.34	3	11.0	11.0	(1 2 0)
	Z = 1.0 for $\rho = 1.1$	4	9.69	9.72	(3 0 0)
		5	7.55	7.29	(4 0 0)
		6	ca. 4.69	-	#2
		7	3.34	-	h
	Col _{h02} at 179°C	1	30.1	30.1	(1 0 0)
	a = 34.7	2	18.3	17.4	(1 1 0)
	h = 3.45	3	11.2	11.4	(1 2 0)
	Z = 1.0 for $\rho = 1.0$	4	ca. 4.86	-	#2
		5	3.45	-	h
	Col _{h03} at 201.5°C	1	28.6	28.6	(1 0 0)
	a = 33.0	2	10.7	10.8	(1 2 0)
	h = 3.53	3	ca. 4.89	-	#2
	Z = 1.0 for $\rho = 1.1$	4	3.53	-	h
2h: (m-C₈OPhO)₈PcCu	Col _{h01} at 101°C	1	31.0	31.0	(1 0 0)
	a = 35.8	2	17.4	17.9	(1 1 0)
	h = 3.38	3	11.5	11.7	(1 2 0)
	Z = 1.0 for $\rho = 1.0$	4	ca. 4.74	-	#2
		5	3.38	-	h
	Col _{ro} (P2 ₁ /a) at 158°C	1	31.0	30.1	(1 1 0)
	a = 49.3 b = 38.1	2	25.0	24.7	(2 0 0)
	h = 3.48	3	17.4	17.8	(1 2 0)
	Z = 2.0 for $\rho = 1.2$	4	11.2	11.3	(2 3 0)
		5	10.0	10.0	(3 3 0)
		6	7.83	7.80	(5 3 0)
		7	6.55	6.61	(7 2 0)
		8	ca. 5.13	-	#1
		9	3.48	-	h
	Col _{h02} at 135°C	1	31.1	31.1	(1 0 0)
	a = 35.9	2	17.7	17.9	(1 1 0)
	h = 3.40	3	11.4	11.7	(1 2 0)
	Z = 1.0 for $\rho = 1.0$	4	ca. 4.76	-	#2
		5	3.40	-	h
	Col _{h03} at 175°C	1	31.1	31.1	(1 0 0)
	a = 35.9	2	18.1	17.9	(1 1 0)
	h = 3.44	3	11.5	11.7	(1 2 0)
	Z = 1.0 for $\rho = 1.0$	4	ca. 4.78	-	#2
		5	3.44	-	h

Table S1. (Continued).

2i: (<i>m</i> -C ₉ OPhO) ₈ PcCu	Col _{ho1} at 35 °C a = 36.3 h = 3.33 Z = 1.0 for $\rho = 1.1$	1	31.4	31.4	(1 0 0)
		2	17.1	18.1	(1 1 0)
		3	11.7	11.9	(1 2 0)
		4	10.2	10.5	(3 0 0)
		5	7.87	7.86	(4 0 0)
		6	ca. 4.60	-	#2
		7	3.33	-	h
		1	31.8	31.8	(1 0 0)
		2	18.9	18.3	(1 1 0)
		3	11.9	12.0	(1 2 0)
	Col _{ho2} at 180 °C a = 36.7 h = 3.45 Z = 1.0 for $\rho = 1.0$	4	ca. 4.83	-	#2
		5	3.45	-	h
		1	31.7	31.0	(1 1 0)
		2	25.2	24.9	(2 0 0)
		3	18.7	18.4	(1 2 0)
		4	11.7	11.7	(2 3 0)
		5	10.2	10.3	(3 3 0)
		6	8.03	7.95	(5 3 0)
		7	ca. 5.16	-	#1
		8	3.79	-	-
	Col _{ro} (P2 ₇ /a) at 180 °C a = 49.8 b = 39.6 h = 3.48 Z = 2.0 for $\rho = 1.2$	9	3.48	-	h
		10	3.19	-	-
2j: (<i>m</i> -C ₁₀ OPhO) ₈ PcCu ^{II}	Col _{ho} at virgin r.t. a = 34.4 h = 3.5 ^{II} Z = ca. 1 for $\rho = 1.0$	1	29.8	29.8	(1 0 0)
		2	17.1	17.1	(1 1 0)
		3	11.2	11.3	(1 2 0)
		4	8.4	8.3	(1 3 0)
		5	6.7	6.8	(2 3 0)
		6	ca. 4.4	-	#2
		1	29.8	29.6	(1 0 0)
		2	17.1	17.1	(1 1 0)
		3	11.2	11.3	(1 2 0)
		4	8.4	8.3	(1 3 0)
	Col _{ho} at nonvirgin r.t. a = 34.4 h = 3.32 Z = ca. 1 for $\rho = 1.0$	5	6.7	6.8	(2 3 0)
		6	ca. 4.4	-	#2
		7	3.32	-	(0 0 1)
2k: (<i>m</i> -C ₁₂ OPhO) ₈ PcCu ^{III}	Col _{ho} at virgin r.t. a = 38.6 h = 3.5 ^{II} Z = ca. 1 for $\rho = 1.0$	1	33.4	33.4	(1 0 0)
		2	19.4	19.3	(1 1 0)
		3	17.0	16.7	(2 0 0)
		4	12.8	12.7	(2 1 0)
		5	11.0	10.8	(3 0 0)
		6	9.02	9.00	(3 1 0)
		7	ca. 4.5	-	#2
		1	32.4	32.4	(1 0 0)
		2	19.0	18.8	(1 1 0)
		3	16.4	16.3	(2 0 0)
	Col _{ho} at nonvirgin r.t. a = 37.5 h = 3.33 Z = ca. 1 for $\rho = 1.0$	4	12.4	12.3	(2 1 0)
		5	11.3	11.1	(3 0 0)
		6	9.48	9.65	(3 1 0)
		7	ca. 4.5	-	#2
		8	3.33	-	(0 0 1)
2l: (<i>m</i> -C ₁₄ OPhO) ₈ PcCu ^{III}	Col _{ho} at virgin r.t. a = 41.1 h = 3.5 ^{II} Z = ca. 1 for $\rho = 1.0$	1	35.6	35.6	(1 0 0)
		2	20.6	20.6	(1 1 0)
		3	18.0	17.8	(2 0 0)
		4	13.5	13.5	(2 1 0)
		5	ca. 4.5	-	#2
		1	34.6	34.6	(1 0 0)
		2	20.2	20.0	(1 1 0)
		3	17.4	17.3	(2 0 0)
		4	ca. 4.4	-	#2
		5	3.33	-	(0 0 1)
2m: (<i>m</i> -C ₁₆ OPhO) ₈ PcCu ^{II}	Col _{ho} at 50 °C a = 39.8 h = 3.36 Z = ca. 1 for $\rho = 1.0$	1	34.5	34.5	(1 0 0)
		2	20.4	19.9	(1 1 0)
		3	18.0	17.2	(2 1 0)
		4	ca. 4.5	-	#2
		5	3.36	-	(0 0 1)
2n: (<i>m</i> -C ₁₈ OPhO) ₈ PcCu ^{II}	Col _{ho} at 90 °C a = 42.5 h = 3.36 Z = ca. 1 for $\rho = 1.0$	1	36.8	36.8	(1 0 0)
		2	22.1	21.2	(1 1 0)
		3	18.9	18.4	(2 0 0)
		4	ca. 4.5	-	#2
		5	3.36	-	(0 0 1)
2o: (<i>m</i> -C ₂₀ OPhO) ₈ PcCu ^{II}	Col _{ho} at 100 °C a = 41.4 h = 3.35 Z = ca. 1 for $\rho = 1.0$	1	35.9	35.9	(1 0 0)
		2	22.4	22.4	(1 1 0)
		3	18.9	18.9	(2 0 0)
		4	ca. 4.6	-	#2
		5	3.35	-	(0 0 1)

h: Stacking distance = (0 0 1), #2: halos due to the molten alkyl groups, ρ : Assumed density (g/cm³) and ^{II}: Assumed value. ^{II}: Ref [21], ^{III}: Ref [20].

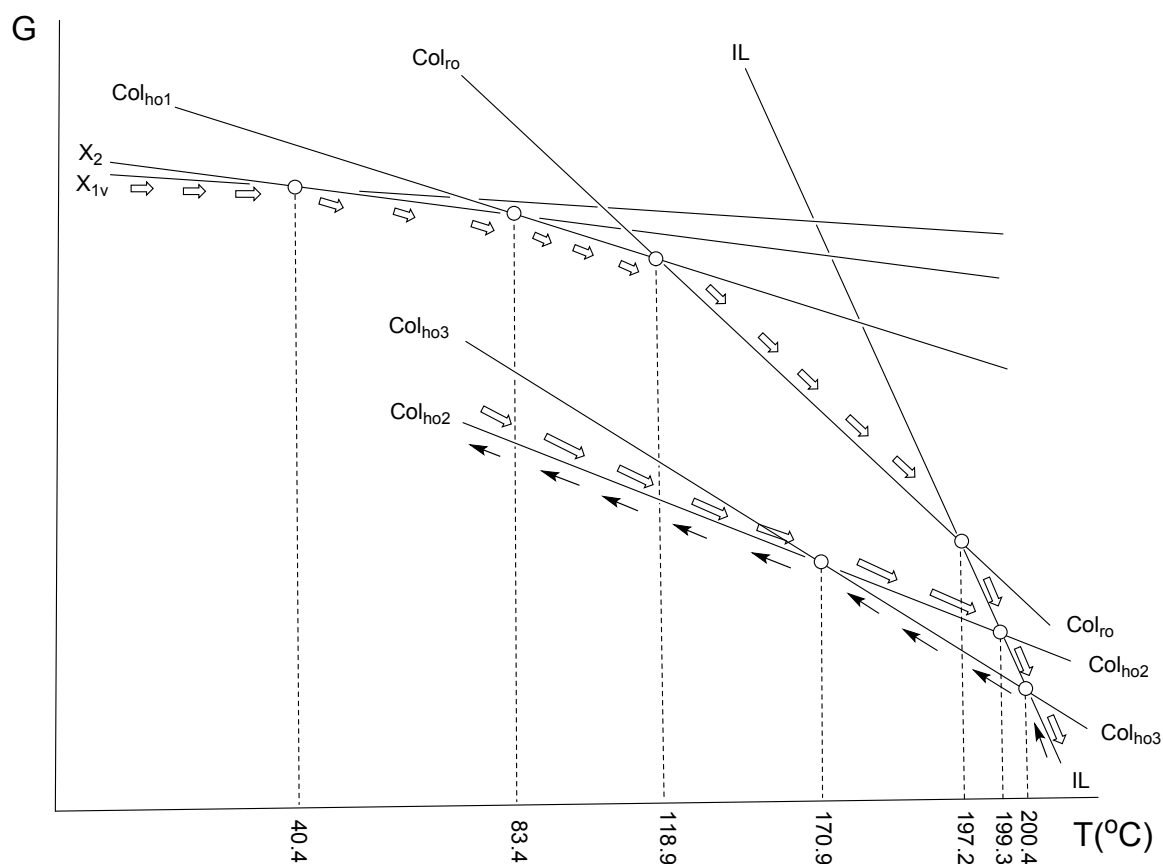


Fig. S3. Schematic free energy versus temperature (G-T) diagram of (*m*-C₈OPhO)₈PcCu (**2h**). On the 1st heating stage indicating small open arrows: when the virgin sample is heated from rt, an unidentifiable X_{1v} phase transforms into another unidentifiable X₂ phase at 40.4 °C of the intersection; the X₂ phase transforms into Col_{ho1} phase at 83.4 °C and then into Col_{ro}(P2_{1/a}) phase at 118.9 °C. On further heating, it clears into I.L. at 197.2 °C. On the 1st cooling stage indicating black arrows: when the IL over 210 °C is cooled, it transforms into Col_{ho3} phase at 200.4 °C of the intersection and then into Col_{ho2} at 170.9 °C. On the 2nd heating stage indicating long open arrows: when the Col_{ho2} phase is heated, it is superheated till 199.3 °C of the clearing point without transforming into Col_{ho3}. These complicated phase transition behaviour can rationally explain the DSC thermograms illustrated in Fig. S1, by using the G-T diagram. The phase appearance greatly depends on the thermal history and heating and cooling rates. Only when the supercooled Col_{ho3} was held at 165.0 °C for long hours, the Col_{ro} phase partially appeared in Col_{ho3} with Col_{ho2}, as shown in Fig. S2 Photo (H). At the present time, we cannot explain it by this G-T diagram. Further study is necessary.