

Supplementary Information

Cocrystals of zonisamide: physicochemical characterization and sustained release solid forms

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1. List of coformers used in the cocrystal screening of ZNS

List of coformers used in the cocrystal screening of zonisamide. The coformers that resulted cocrystals were highlighted.

Methylparaben	3-Hydroxybenzoic acid
Ethylparaben	5-Methyl-2-pyridone (HMP)
Propylparaben	Citric acid
Piperazine	Piperine
Pyrogallol	Bisphenol A
Methyl gallate	Urea
Betaine monohydrate	2-Picolinic acid
Nicotinamide	Nicotinic acid
4,4'-Dipyridyl anhydrous	Isonicotinic acid
4-Aminobenzoic acid	2,5-Dihydroxybenzoic acid
EDTA	2-Picolinamide
L-(+)-Mandelic acid	Salicylamide
L-Proline (LP)	Salicylic acid
L-Tartaric acid	Pimelic acid
Acesulfame	Saccharin
Adipic acid	N-Acetyl-L-cysteine
Benzoic acid	Imidazole
Fumaric acid	Indole-3-acetic acid
Glutaric acid	Pamoic acid
Glycine	Orotic acid anhydrous
Isonicotinamide	4-Acetamidophenol
p-Coumaric acid	Caffeine (CAF)
Lactose	Catechol (CTC)
Maleic acid	Resorcinol
Malonic acid	Hydroquinone
4-Hydroxybenzoic acid	Succinic acid

2. Crystal structure parameters

Table S1 Crystal structure data for the cocrystals, ZNS-CAF and ZNS-HMP.

Compound reference	ZNS-CAF	ZNS-HMP
Chemical formula	$C_8H_8N_2O_3S \cdot C_8H_{10}N_4O_2$	$C_8H_8N_2O_3S \cdot C_6H_7NO$
Formula Mass	406.42	321.35
Crystal system	Monoclinic	Monoclinic
$a/\text{\AA}$	9.59020(10)	14.1373(3)
$b/\text{\AA}$	13.30840(10)	11.23707(18)
$c/\text{\AA}$	14.24360(10)	9.45521(17)
$\alpha/^\circ$	90	90.00
$\beta/^\circ$	91.8550(10)	106.2761(19)
$\gamma/^\circ$	90	90.00
Unit cell volume/ \AA^3	1816.96(3)	1441.87(4)
Temperature/K	297	100
Space group	$P2_1/n$	$P2_1/c$
No. of formula units per unit cell, Z	4	4
Radiation type	Cu-K α	Mo-K α
No. of reflections measured	27367	23512
No. of independent reflections	3217	3301
R_{int}	0.0209	0.0251
Final R_I values ($I > 2\sigma(I)$)	0.0504	0.0301
Final $wR(F^2)$ values ($I > 2\sigma(I)$)	0.1446	0.0775
Final R_I values (all data)	0.0504	0.0324
Final $wR(F^2)$ values (all data)	0.1446	0.0790
Goodness of fit on F^2	1.041	1.056
CCDC number	1832355	1816488

3. Table of hydrogen bonds

Table S2. Neutron normalized intermolecular interactions in the crystal structure of ZNS-CAF and ZNS-HMP cocrystals.

Cocrystal	D–H···A ^a	H···A/Å	D···A/Å	D–H···A/°	Symmetry code
ZNS-CAF	N2–H2B···O4	1.91	2.9047(1)	167	-
	N2–H2A···N5	2.02	2.9598(1)	155	3/2-x,-1/2+y,3/2-z
	C2–H2···O2	2.63	3.3390(1)	122	-1+x,y,z
	C3–H3···O4	2.45	3.4058(1)	146	1/2-x,1/2+y,3/2-z
	C8–H8A···O2	2.49	3.4627(3)	149	2-x,1-y,1-z
	C15–H15B···O3	2.44	3.4236(1)	151	-
	C15–H15C···O4	2.64	3.5293(1)	139	1-x,1-y,2-z
ZNS-HMP	C16–H16A···O4	2.56	3.1679(1)	115	-
	N2–H2A···O1	2.01	3.022(1)	176	x,3/2-y,-1/2+z
	N2–H2B···O4	1.83	2.828(1)	169	-
	N3–H3A···O4	1.78	2.784(1)	174	1-x,2-y,1-z
	C3–H3···O2	2.64	3.359(2)	123	-x,1/2+y,-1/2-z
	C8–H8A···O2	2.93	3.329(2)	102	x,3/2-y,1/2+z
	C8–H8B···O2	2.79	3.329(2)	111	x,3/2-y,1/2+z

^aD = Donor, A = Acceptor.

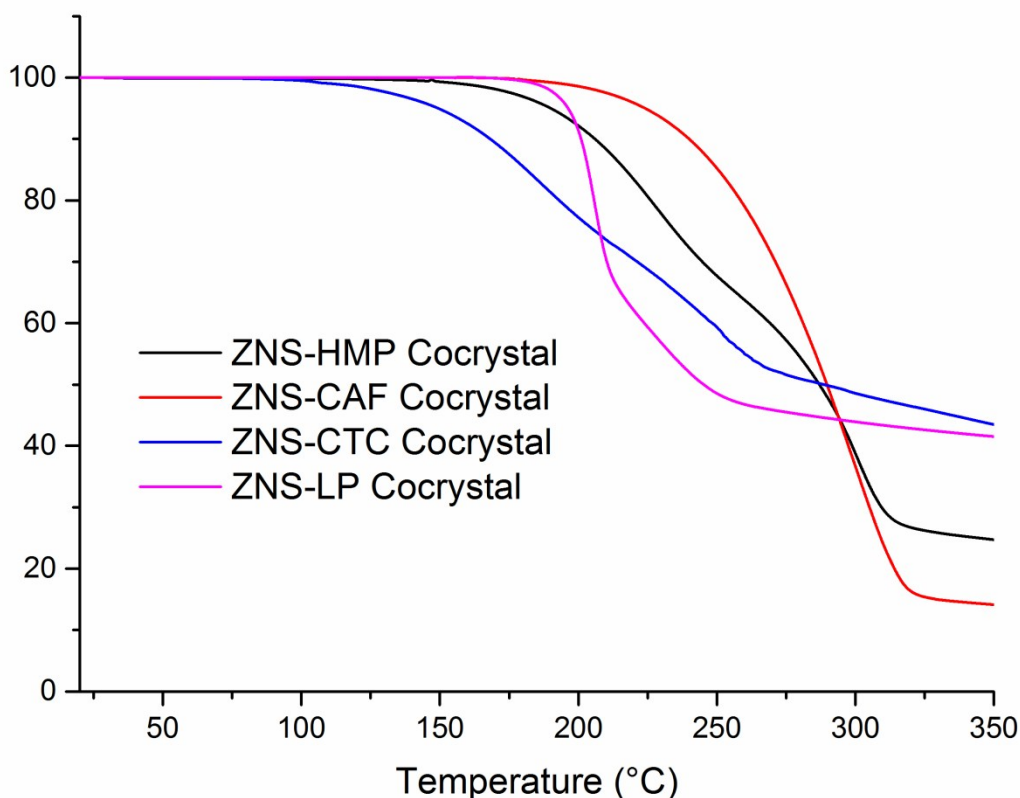


Figure S1. TGA thermograms of ZNS cocrystals.

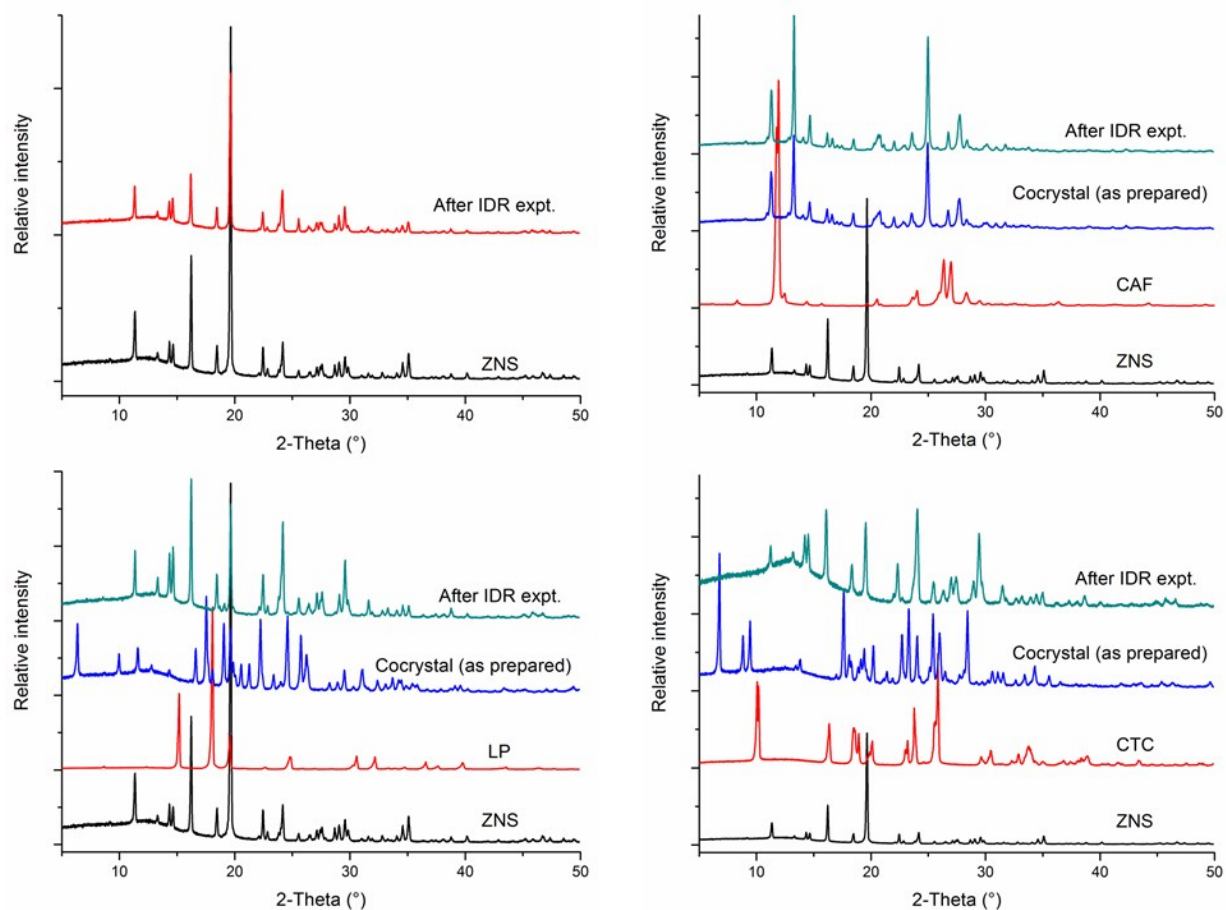


Figure S2. Comparison of the PXRD pattern of the samples remained after the IDR experiments with the respective reference PXRD patterns.