Supporting Information

Intermetallic Pd₃Pb square nanoplates as highly efficient electrocatalysts for oxygen reduction reaction

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Figure S1. (a, c) Representative TEM images at low magnification, (b) edge length distribution and (d) thickness distribution of the Pd₃Pb square nanoplates prepared using the standard procedure.



Figure S2. (a) AFM image and (b) corresponding height profile of Pd₃Pb nanoplates.



Figure S3. EDX spectrum of the intermetallic Pd₃Pb square nanoplates, showing a Pd/Pb atomic ratio of 3.08:1.



Figure S4. (a) High-resolution TEM image of a single Pd₃Pb square nanoplate. (b) FFT pattern arising from high-resolution TEM image of the part marked by the red rectangle in (a).



Figure S5. EDX line-scan profiles of (a) a planar Pd_3Pb nanoplate and (b) a vertically upstanding Pd_3Pb nanoplate. The insets show the corresponding HADDF-STEM images.



Figure S6. XPS spectra of the Pd₃Pb intermetallic nanoplates for (a) Pd 3d and (b) Pb 4f orbitals, respectively.



Figure S7. (a, b) XRD pattern and its corresponding magnified one, (c, d) HRTEM images, (e) EDX-mapping and (f) EDX spectrum of the Pd₃Pb alloy dendritic nanocrystals.



Figure S8. Representative TEM images of the products prepared by using the starndard procedure except for different molar ratios of Pd to Pb precursors: (a) in the absence of Pb precursor, (b) 2:1, (c) 1:1, (d) 1:2, (e) 1:3 and (f) in the absence of Pd precursor.



Figure S9. The cyclic voltammetry (CV) curves of (a) the Pd_3Pb square nanoplates, (b) Pd_3Pb dendritic nanocrystals, (c) commercial Pt/C in Ar-purged 0.1 M KOH solutions at a sweep rate of 50 mV/s.



Figure S10. (a) Rotating-disk voltammograms of Pt/C in O_2 -saturated 0.1 M KOH at different rotation rates with a sweep rate of 10 mV s⁻¹. (a) The corresponding Koutechy-Levich plots of figure (a).



Figure S11. (a, c) ORR polarization curves of Pd₃Pb alloy dendritic nanocrystals/C and Pt/C in O₂-saturated 0.1 M KOH before and after different potential cycles between 0.6 and 1.0 V versus RHE, respectively. (b, d) The changes on specific and mass activities of the Pd₃Pb alloy dendritic nanocrystals/C and Pt/C after various potential cycles between 0.6 and 1.0 V versus RHE, respectively.



Figure S12. Representative TEM images of the Pd_3Pb alloy dendritic nanocrtstals/C (a) before and (b) after ADT. (c) HADDF-STEM-EDX mapping image and (d) EDX spectrum of the Pd_3Pb alloy dendritic nanocrtstals after 10000 cycles between 0.6 V and 1.0 V versus RHE.



Figure S13. Representative TEM images of the intermetallic Pd_3Pb square nanoplates/C (a) before and (b) after ADT. (c) HADDF-STEM-EDX mapping image and (d) EDX spectrum of the intermetallic Pd_3Pb square nanoplates after 10000 cycles between 0.6 V and 1.0 V versus RHE.



Figure S14. Representative TEM images of the commercial Pt/C (a) before and (b) after 10000 potential cycles between 0.6 V and 1.0 V versus RHE.