

Supplementary Information

Electron-stimulated reactions in nanoscale water films adsorbed on α - $\text{Al}_2\text{O}_3(0001)$

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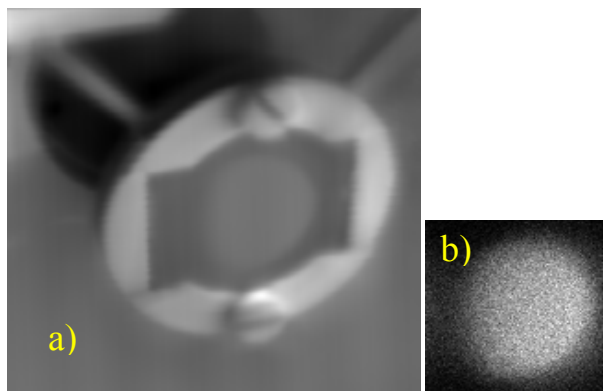


Fig. S1. a) Secondary electron image of a 40 ML H_2O film adsorbed on the $\text{Al}_2\text{O}_3(0001)$ sample (mounted on the sample holder). The image is obtained by rastering a 500 eV electron beam over the sample (at 30 K) and collecting the emitted secondary electrons versus the beam X-Y deflection. The $10\times 10\times 1$ mm crystal is attached to the resistively-heated $\varnothing 15$ mm and 1 mm thick Ta base plate using high-temperature Aremco 865 cement adhesive and covered from the front with a Mo retaining ring and 2 Mo screws (seen in the image). The 7 mm diameter 40 ML thick water spot dosed on the Al_2O_3 surface with a molecular beam is also seen in this image (lighter grey). The good resolution of the image, especially for the water spot, indicates that charging is not a significant issue here. A K-type thermocouple is spot-welded to the back side of the Ta base plate (not seen here). The Ta resistive heating leads are seen in the upper left and right corners. b) O_2 ESD image from a 40 ML thick water spot dosed on the Al_2O_3 surface obtained by rastering

the 100 eV electron beam over the sample and measuring the O₂ signal versus the beam X-Y deflection. The electron beam is smaller than the molecular beam spot size on the sample (~1.5 mm and 7.0 mm, respectively). The ESD signals presented in this study are an average of 80,000 data points (400 pixels/scan \times 200 points/pixel) per each 0.4 s scan.

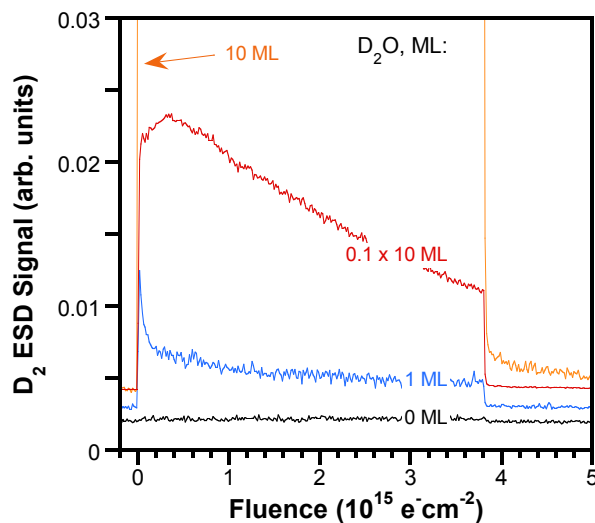


Fig. S2. D₂ ESD signals at 100 K versus electron fluence from 0, 1 and 10 ML of D₂O on Al₂O₃(0001). The traces are displaced for the sake of comparison. Radiation-induced change in the D₂ background signal is not significant (0 ML, black trace).

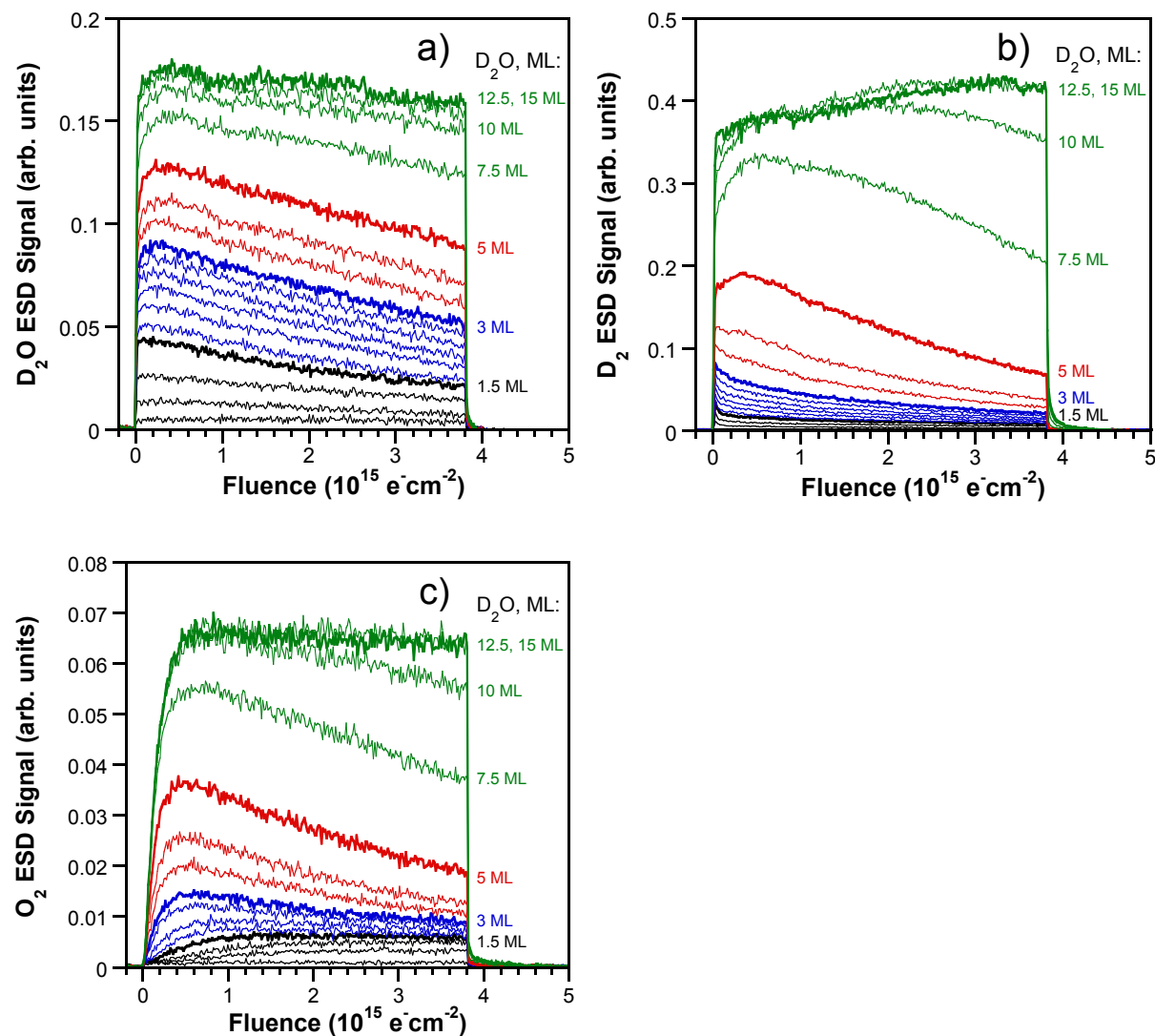


Fig. S3. D_2O , D_2 and O_2 ESD signals at 100 K versus electron fluence for various coverages of D_2O on $Al_2O_3(0001)$. (1 ML = 10^{15} cm^{-2}).

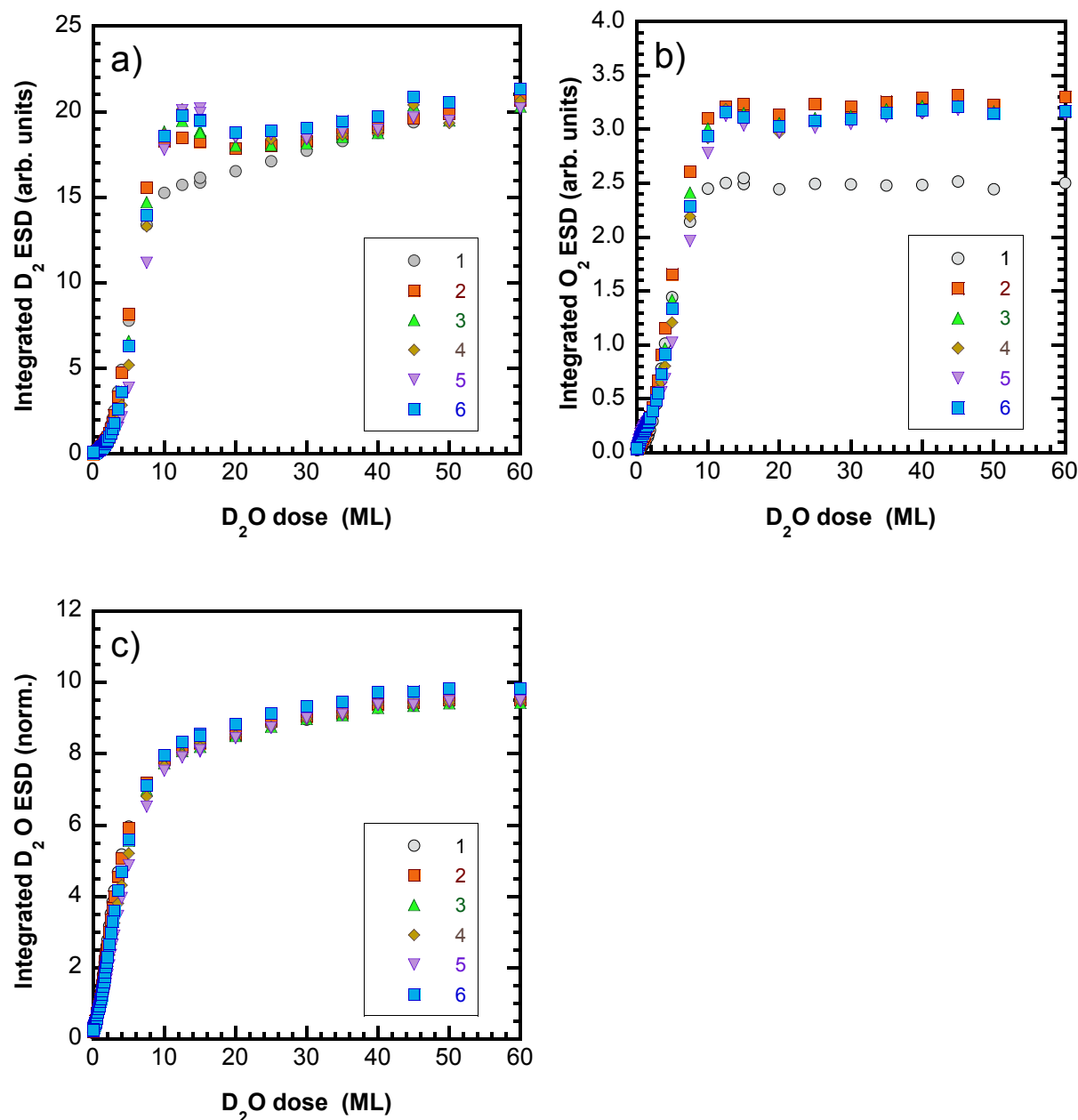


Fig. S4. a) D_2 , b) O_2 and c) D_2O ESD yields integrated in different electron fluence ranges versus initial D_2O coverage on $Al_2O_3(0001)$ for the data presented in Fig. 3. The integrations start at $\phi_e =$ (1) 0, (2) 0.46, (3) 1.22, (4) 1.98, and (5) $2.74 \times 10^{15} \text{ cm}^{-2}$, and the same integration width ($\Delta\phi_e = 0.76 \times 10^{15} \text{ cm}^{-2}$) is used for each. (6) The total ESD integrals from Fig. 3, arbitrarily scaled to facilitate comparison, are also shown. The initial D_2 ESD integral ((a), 1, grey circles) is also multiplied by 0.9 for the sake of comparison. A small peak in D_2 ESD data (a) in the 8 – 20 ML coverage range, which is initially absent, develops at higher

fluences. This peak is associated with D₂ molecules produced at the Al₂O₃/D₂O interface from precursors accumulating on the alumina surface with irradiation time. A similar peak can be seen in the O₂ ESD data (b), but not in the D₂O ESD data (c). D₂O ESD occurs from the D₂O/vacuum interface and it is not associated with reactions at the Al₂O₃/D₂O interface.¹ O₂ is also produced at the D₂O/vacuum interface but it is correlated with D₂ production and associated with reactions at the Al₂O₃/D₂O interface (see more details in our earlier publications^{2,3}).

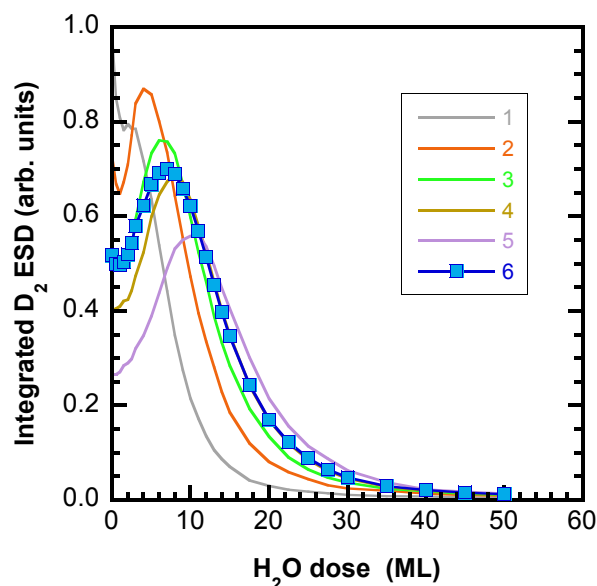


Fig. S5. D₂ ESD yields integrated in different electron fluence ranges versus the initial H₂O cap coverage in isotopically layered D₂O/H₂O films on Al₂O₃(0001) for the data presented in Fig. 5. The integrations start at $\phi_e = (1) 0, (2) 0.30, (3) 1.07, (4) 1.83, \text{ and } (5) 3.36 \times 10^{15} \text{ cm}^{-2}$ and the same integration width ($\Delta\phi_e = 0.30 \times 10^{15} \text{ cm}^{-2}$) is used for each. (6) The total ESD integral from Fig. 5, arbitrarily scaled to facilitate comparison, is also shown. At higher fluence, the maximum in the D₂ ESD shifts to the higher H₂O coverage due to the film sputtering (initial coverage is higher than the actual one) and due to the greater contribution of the reactions at the Al₂O₃/D₂O interface.

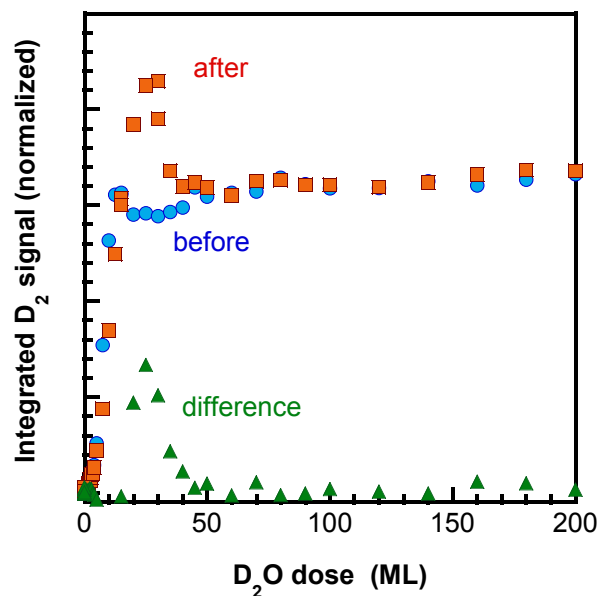


Fig. S6. Integrated D₂ ESD yields versus initial D₂O coverage on Al₂O₃(0001). The ESD yields are integrated during 1 s before the end of irradiation with a fluence of $3.8 \times 10^{15} \text{ cm}^{-2}$ (blue circles) and 20 s after the end of irradiation (red squares). The maximum in the D₂ ESD signal near 25 ML observed after irradiation is due to D₂ molecules produced at the Al₂O₃/D₂O interface and diffusing through the film to desorb. Above 25 ML, this signal decreases since the D₂ yield at the Al₂O₃/D₂O interface decreases with coverage (compare to the Fig. 5b and note that the actual coverage of the film is lower than initial due to sputtering). Above 50 ML, the post-radiation D₂ ESD yield stabilizes at a coverage-independent level proportional to the D₂ ESD signal at the end of irradiation (blue circles) and associated with a certain pumping out rate of our chamber. This component of the signal is associated mainly with the electron-induced processes at the D₂O/vacuum interface. Signal normalization and subtraction of this “pumping tail” yields the D₂ component produced at the Al₂O₃/D₂O interface (green triangles). Similar post-irradiation sample outgassing was also observed earlier for the Pt(111)/ASW systems and associated with the molecular hydrogen produced at the metal/water interface.^{4, 5}

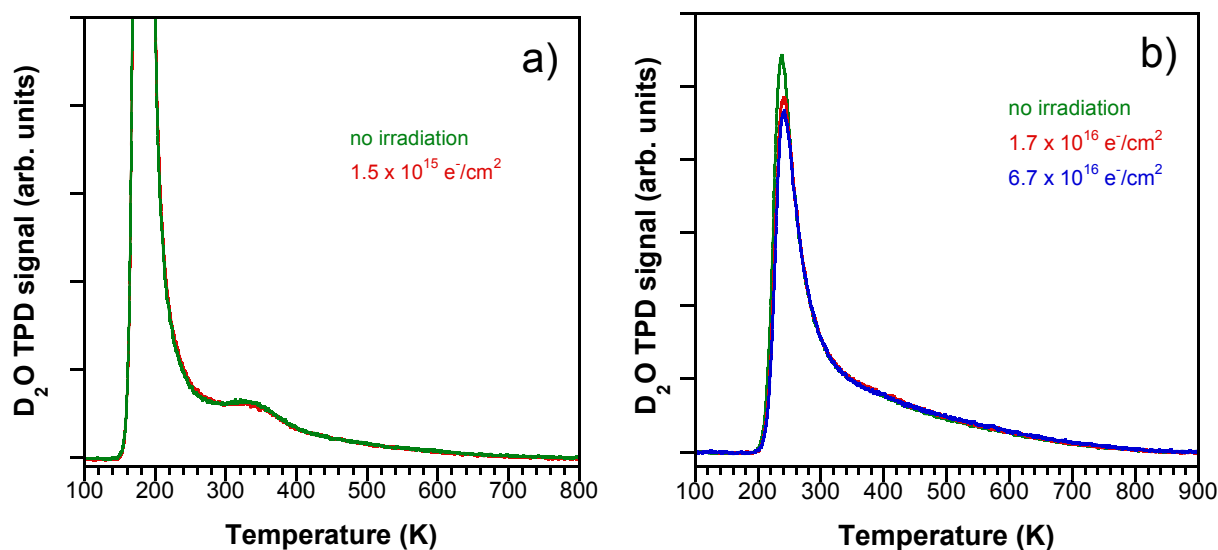


Fig. S7. a) TPD spectrum of 5 ML D₂O irradiated with $1.5 \times 10^{15} \text{ e}^-/\text{cm}^2$ electrons ($E_i = 100 \text{ eV}$) (red) and of similar coverage of non-irradiated D₂O (green). b) TPD spectrum of 0.5 ML D₂O dosed on Al₂O₃(0001) pre-irradiated with 500 eV electrons ($1.7 \times 10^{16} \text{ e}^-/\text{cm}^2$ and $6.7 \times 10^{16} \text{ e}^-/\text{cm}^2$) and not irradiated (red, blue and green respectively). $1 \text{ ML} = 10^{15} \text{ cm}^{-2}$.

References

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