

## Electronic Supporting Information

### Probing the structural and electronic properties of Zirconium doped boron clusters: Zr distorted B<sub>12</sub> ligand framework

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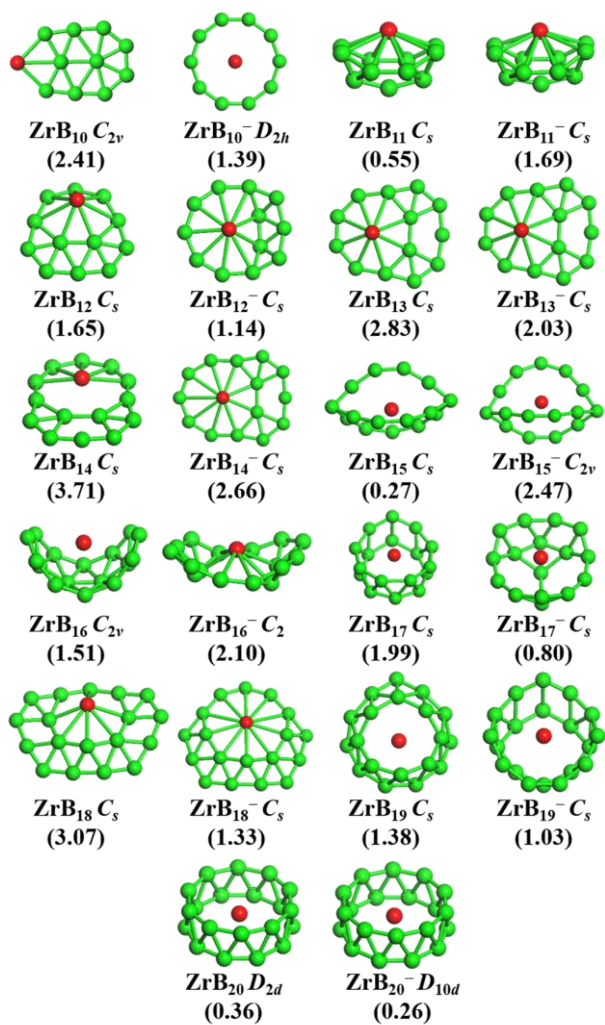


Figure S1. The metastable structures of ZrB<sub>n</sub><sup>Q</sup> ( $n = 10-20$ ,  $Q = 0, -1$ ) clusters along with the point group symmetry and the relative energy difference compared with the lowest-energy structures.

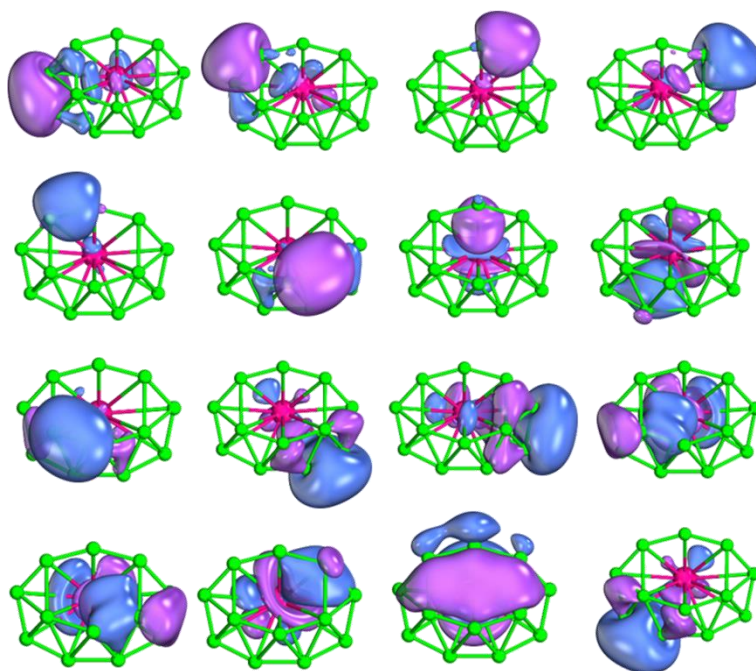


Figure S2. The intrinsic bond orbitals of anionic  $\text{ZrB}_{12}^-$  cluster visualized using the IBO localization method. The iso-surfaces enclose 80% of the orbital's electron density.

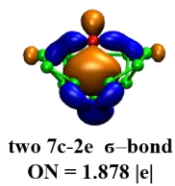
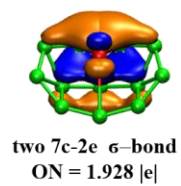
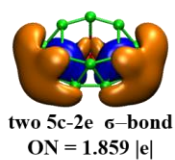
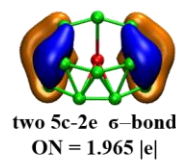
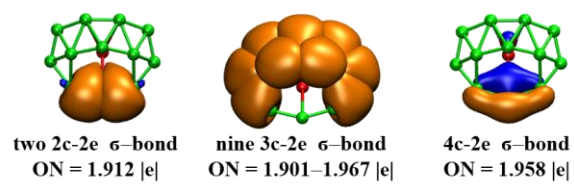


Figure S3. Chemical bonding pattern of anionic  $\text{ZrB}_{12}^-$  cluster shown by the AdNDP analysis. ON stands for occupation numbers.

Table S1. The calculated bond lengths R ( $\text{\AA}$ ) and lowest vibrational frequencies ( $\text{cm}^{-1}$ ) of  $\text{B}_2$ ,  $\text{Zr}_2$  and  $\text{ZrB}_2$  at different levels together with their corresponding experimental values.

Method	Cluster					
	$\text{B}_2$		$\text{Zr}_2$		$\text{ZrB}_2$	
	R/ $\text{\AA}$	$\omega$	R/ $\text{\AA}$	$\omega$	R/ $\text{\AA}$ (Zr-B)	$\omega$
HF	1.639	942.41	2.242	433.09	2.071	635.82
B3LYP	1.617	1002.11	2.252	425.86	2.078	597.02
PW91	1.616	1015.12	2.239	434.88	2.063	617.79
B3PW91	1.617	1012.13	2.243	431.45	2.067	611.03
PBE	1.617	1012.13	2.257	417.14	2.064	619.33
CCSD	1.639	940.69	2.288	394.53	2.204	586.53
<b>PBE0</b>	<b>1.607</b>	<b>1085.64</b>	<b>2.239</b>	<b>404.32</b>	<b>2.083</b>	<b>589.40</b>
<i>Exp.</i>	<b>1.590<sup>1</sup></b>	<b>1051.30<sup>1</sup></b>	<b>2.240<sup>2</sup></b>	<b>305.70<sup>2</sup></b>		
<i>Theor.</i> <sup>3</sup>	<b>1.653</b>				<b>2.102</b>	

Table S2. Detailed results of the calculated electronic states, symmetries, averaged binding energies ( $E_b$ ), and HOMO-LUMO energy gaps ( $E_{\text{gap}}$ ) for the ground state  $\text{ZrB}_n^Q$  ( $n = 10-20$ ,  $Q = 0, -1$ ) clusters.

$\text{ZrB}_n$					$\text{ZrB}_n^-$				
$n$	State	Sym.	$E_b$	$E_{\text{gap}}$	$n$	State	Sym.	$E_b$	$E_{\text{gap}}$
10	1-A <sub>1</sub>	C <sub>2v</sub>	7.92	2.20	10	2-A <sub>2</sub>	C <sub>2v</sub>	8.05	0.97
11	2-A''	C <sub>s</sub>	8.05	1.78	11	3-A''	C <sub>s</sub>	8.21	2.06
12	3-A''	C <sub>s</sub>	8.10	2.83	12	2-A''	C <sub>s</sub>	8.29	2.78
13	2-A'	C <sub>s</sub>	7.99	1.72	13	1-A'	C <sub>4v</sub>	8.23	2.17
14	1-A'	C <sub>s</sub>	8.24	2.64	14	2-A'	C <sub>s</sub>	8.38	1.21
15	2-A <sub>1</sub>	C <sub>2v</sub>	8.19	1.87	15	1-A <sub>1</sub>	C <sub>2v</sub>	8.46	1.89
16	1-A'	C <sub>s</sub>	8.28	1.91	16	2-A	C <sub>s</sub>	8.44	1.66
17	2-A''	C <sub>s</sub>	8.32	1.88	17	1-A'	C <sub>s</sub>	8.45	2.39
18	1-A <sub>1</sub>	C <sub>2v</sub>	8.41	1.53	18	2-B <sub>1</sub>	C <sub>2v</sub>	8.52	1.67
19	2-A''	C <sub>s</sub>	8.34	1.71	19	1-A'	C <sub>s</sub>	8.55	1.64
20	1-A'	C <sub>s</sub>	8.47	1.98	20	2-A'	C <sub>s</sub>	8.54	1.10

Table S3. The calculated VDE, ADE for the ground state  $\text{ZrB}_n^Q$  ( $n = 10-20$ ,  $Q = 0, -1$ ) clusters

Size	VDE	ADE
10	1.937	1.659
11	2.418	2.246
12	2.920	2.796
13	2.560	2.673
14	2.619	2.319
15	3.592	3.310
16	3.219	2.991
17	2.602	2.488
18	2.729	2.438
19	3.565	4.528
20	2.621	1.799

## References

1. K. P. Huber and G. Herzberg, *Molecular Spectra and Molecular Structure*, Vol. 4. Constants of Diatomic Molecules (*Van Nostrand Reinhold*, New York, 1979).
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3. J. G. Yao, X. W. Wang and Y. X. Wang, *Chem. Phys.* 2008, **351**, 1-6.